

Lecture 17, 18, 19

CH-4114

Molecular Simulation

“Everything that living things do can be understood in terms of the jigglings and wigglings of atoms.”

- Richard P. Feynman

By
Dr. Susmita Roy
IISER-Kolkata



Thermostat: Velocity Rescaling, Berendsen, Nosé-Hoover

Velocity rescaling method

Instantaneous Temperature from Kinetic Energy

In molecular dynamics, the instantaneous temperature of a system with N particles is obtained from the kinetic energy (KE):

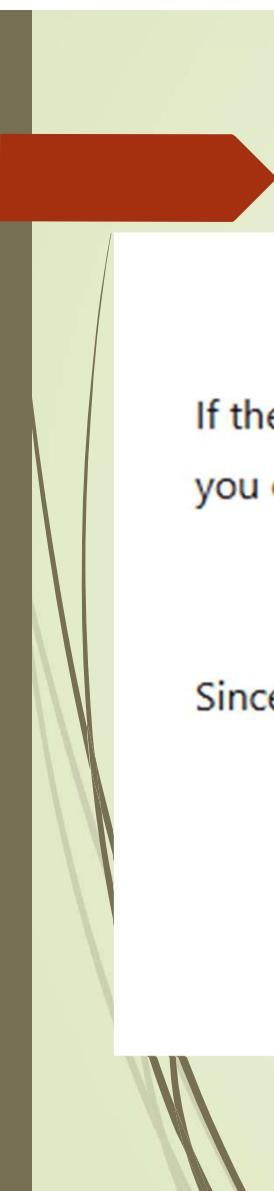
$$E_k = \frac{1}{2} \sum_{i=1}^N m_i v_i^2 = \frac{3}{2} N k_B T$$

So,

$$T = \frac{2E_k}{3Nk_B}$$

At any time t , you can compute the temperature $T(t)$ from the velocities.

[Michele Parrinello](#) and coworkers, J Chem Phys, 2007 Jan 7;126(1):014101. doi:10.1063/1.2408420.



Idea: Maintain constant temperature by rescaling velocities

If the current temperature $T(t)$ deviates from the desired temperature T_0 , you can **rescale all velocities** by a constant factor λ :

$$\mathbf{v}'_i = \lambda \mathbf{v}_i$$

Since kinetic energy depends on v^2 , the new kinetic energy (and hence the new temperature) changes as:

$$\begin{aligned} E'_k &= \frac{1}{2} \sum_i m_i (\lambda v_i)^2 = \lambda^2 E_k \\ \Rightarrow T' &= \lambda^2 T \end{aligned}$$

Determine the scaling factor λ

We want the new temperature T' to equal the desired target temperature T_0 .

So:

$$T_0 = \lambda^2 T(t)$$

$$\lambda = \sqrt{\frac{T_0}{T(t)}}$$

That's it — the simplest possible way to control temperature:
multiply all particle velocities by this factor λ at every MD step.



What happens during the simulation

At each time step:

1. Compute instantaneous $T(t)$ from velocities.
2. Compute scaling factor $\lambda = \sqrt{T_0/T(t)}$.
3. Multiply all velocities by λ .
4. Continue to the next step.

This ensures the **instantaneous temperature** matches T_0 exactly after every rescaling.

Limitations to be aware of

Mathematical effect on equations of motion

However, this procedure **overrides** the natural dynamics — you are continuously forcing the system to have the exact temperature.

Therefore, the **equations of motion become irreversible**.

That means:

- Energy and momentum conservation are broken.
- The system **does not** represent a physical ensemble (not canonical or microcanonical).
- “Transition probabilities” do not satisfy **detailed balance**, so it’s not correct for statistical sampling.

Temperature control — Langevin as a thermostat

Molecular World: Imagine a particle in a “noisy soup”

Imagine a single particle (say, a dust grain or an atom) moving in a liquid.

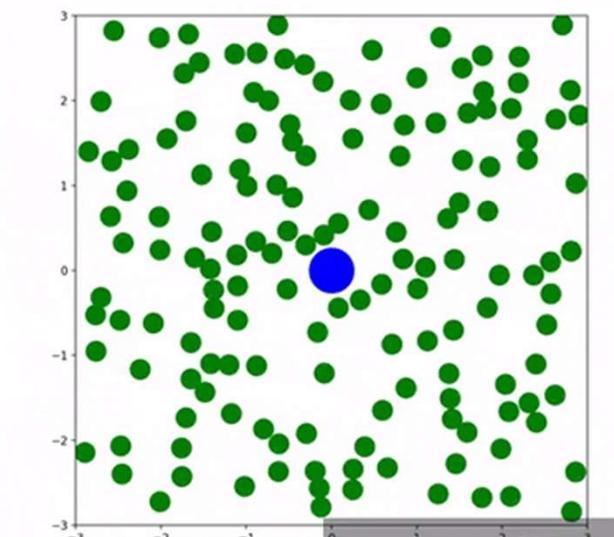
It is constantly bombarded by solvent molecules:

- Each hit changes its velocity a little, randomly.
- At the same time, friction from the fluid slows it down.

So the motion is not purely Newtonian anymore — it's a mix of:

- Systematic forces (e.g., spring force, potential gradient),
- Friction (drag) that resists motion, and
- Random kicks from solvent collisions.

This random-damped motion is exactly what Langevin dynamics models.



The Langevin equation (1D form first)

For a particle of mass m at position $x(t)$, the Langevin equation is:

$$m \frac{d^2x}{dt^2} = F(x) - \gamma m \frac{dx}{dt} + R(t)$$

The Langevin equation (1D form first)

For a particle of mass m at position $x(t)$, the Langevin equation is:

$$m \frac{d^2x}{dt^2} = F(x) - \gamma m \frac{dx}{dt} + R(t)$$

or, equivalently,

$$m\dot{v} = F(x) - \gamma mv + R(t)$$

where each term means:

Term	Meaning	Physical role
$F(x)$	Deterministic force (e.g., $-dU/dx$)	Drives the system along potential
$-\gamma mv$	Friction (drag)	Dissipates kinetic energy
$R(t)$	Random force ("noise")	Injects random energy from the environment

Here γ (gamma) is the **friction coefficient** — it measures how strongly the particle is coupled to the "bath".

Why the random and friction terms go together

If we only had drag (friction), the particle would just slow down and stop.

But in real life, solvent collisions also kick it around randomly, so that on average, its kinetic energy matches the temperature of the environment.

This balance between friction and random kicks leads to **thermal equilibrium** — the system temperature T .

Mathematically, the random force satisfies:

$$\langle R(t) \rangle = 0$$

$$\langle R(t)R(t') \rangle = 2\gamma m k_B T \delta(t - t')$$

This is called the **fluctuation-dissipation theorem** — it ensures that the amount of random “noise” exactly compensates for the frictional “dissipation,” maintaining the correct average thermal energy.

Physical meaning of γ (friction coefficient)

- If γ is very small → friction and noise are weak → dynamics are close to Newtonian.
The system behaves almost like microcanonical dynamics.
- If γ is very large → velocity changes almost instantaneously due to friction.
The motion becomes overdamped → inertial effects vanish.
Then we get Brownian dynamics:

$$\gamma \dot{x} = \frac{F(x)}{m} + \text{noise}$$

Temperature control — Langevin as a thermostat

In molecular dynamics simulations, the Langevin equation acts as a thermostat.

- The random term injects heat → keeps average kinetic energy $\approx \frac{3}{2} N k_B T$.
- The friction term removes excess energy → prevents runaway heating.

Together, they ensure the system samples the canonical (NVT) ensemble, i.e., correct temperature distribution.

Berendsen thermostat

Step 1. Start from the Langevin equation (heat bath coupling)

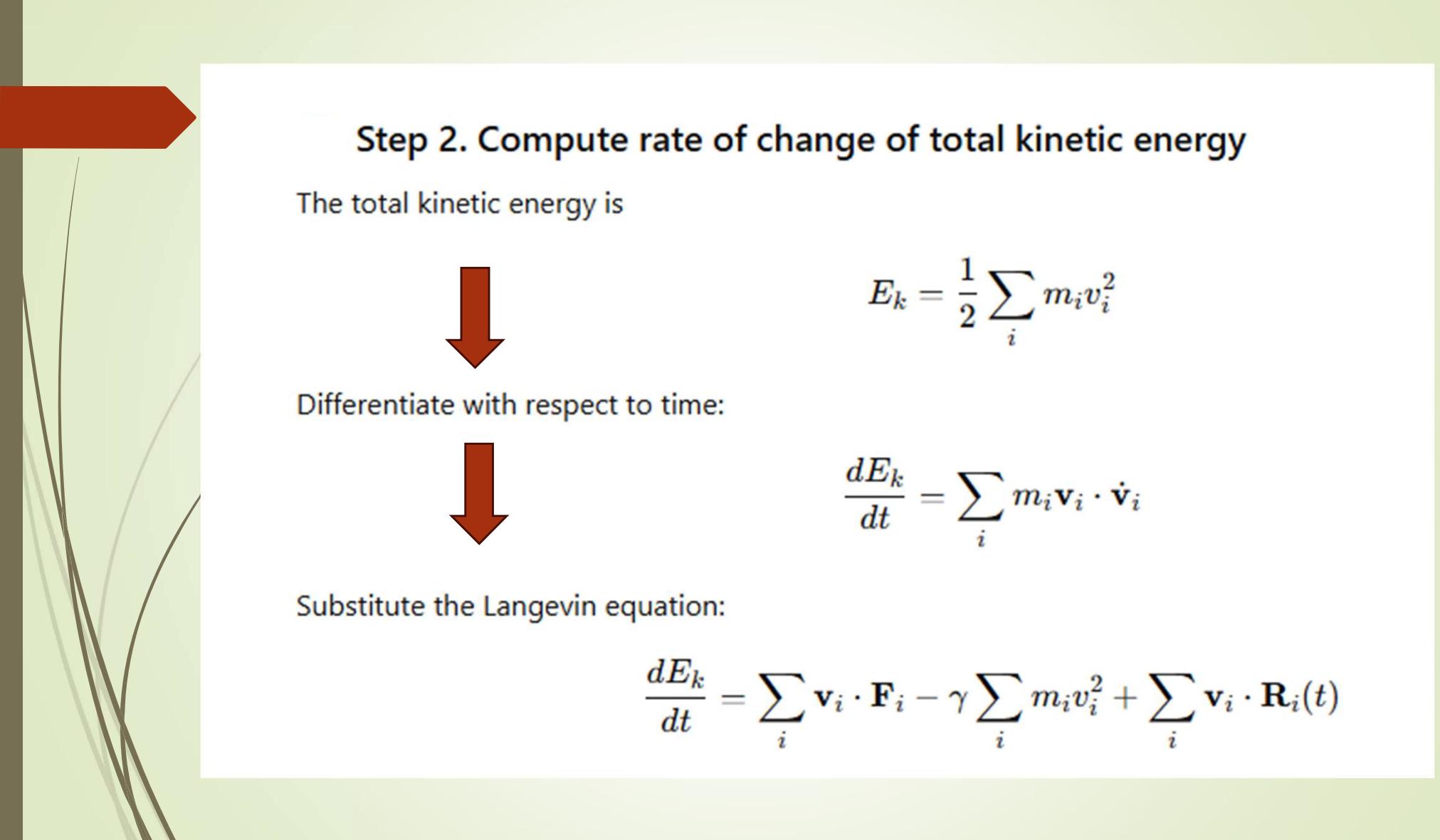
The **Berendsen thermostat** starts conceptually from the **Langevin equation**, where each particle is weakly coupled to a fictitious heat bath at temperature T_0 :

$$m_i \dot{\mathbf{v}}_i = \mathbf{F}_i - \gamma m_i \mathbf{v}_i + \mathbf{R}_i(t)$$

where:

- \mathbf{F}_i : systematic (deterministic) force on particle i
- γ : friction coefficient (coupling strength)
- $\mathbf{R}_i(t)$: random force (Gaussian noise)
- $\langle R_{i\alpha}(t) R_{j\beta}(t') \rangle = 2\gamma m_i k_B T_0 \delta_{ij} \delta_{\alpha\beta} \delta(t - t')$

This ensures that the system exchanges energy with the bath, keeping it near T_0 .



Step 2. Compute rate of change of total kinetic energy

The total kinetic energy is



$$E_k = \frac{1}{2} \sum_i m_i v_i^2$$

Differentiate with respect to time:



$$\frac{dE_k}{dt} = \sum_i m_i \mathbf{v}_i \cdot \dot{\mathbf{v}}_i$$

Substitute the Langevin equation:

$$\frac{dE_k}{dt} = \sum_i \mathbf{v}_i \cdot \mathbf{F}_i - \gamma \sum_i m_i v_i^2 + \sum_i \mathbf{v}_i \cdot \mathbf{R}_i(t)$$

Step 3. Average over stochastic noise

Since the random force $\mathbf{R}_i(t)$ is uncorrelated with the velocities and forces (white noise assumption), its average contribution vanishes:

$$\langle \mathbf{v}_i \cdot \mathbf{R}_i(t) \rangle = 0$$

So, the average rate of change of kinetic energy becomes:

$$\left\langle \frac{dE_k}{dt} \right\rangle = \sum_i \mathbf{v}_i \cdot \mathbf{F}_i - 2\gamma \left(E_k - \frac{3}{2} N k_B T_0 \right)$$

Here, $\frac{3}{2} N k_B T_0$ is the expected kinetic energy of the system if it were at temperature T_0 .

The first term \rightarrow rate of change of **potential energy**, i.e., mechanical energy exchange within the system (doesn't affect temperature directly).

Step 4. Connect kinetic energy to temperature

Define instantaneous temperature T from kinetic energy:

$$E_k = \frac{3}{2} N k_B T$$

Differentiate:

$$\frac{dT}{dt} = \frac{2}{3Nk_B} \frac{dE_k}{dt}$$

Plug the previous result in:

$$\frac{dT}{dt} = -\frac{2\gamma}{1}(T - T_0)$$

Step 5. Define coupling time constant

Let's define a temperature relaxation time τ_T as

$$\tau_T = \frac{1}{2\gamma}$$

Then the equation becomes:

$$\frac{dT}{dt} = \frac{1}{\tau_T} (T_0 - T)$$

Step 6. Solve for T(t)

This differential equation has an exponential relaxation solution:

$$T(t) = T_0 + [T(0) - T_0]e^{-t/\tau_T}$$

So the system temperature relaxes exponentially toward the bath temperature T_0 .

Step 7. Velocity rescaling form (used in MD)

Since $E_k \propto v^2 \propto T$, controlling T means rescaling velocities at each time step:

Let λ be the scaling factor for velocities at time t :

$$\mathbf{v}'_i = \lambda \mathbf{v}_i$$

Then $T' = \lambda^2 T$.

The change in temperature over one time step Δt according to the exponential relaxation is approximately:

$$T' = T + \frac{\Delta t}{\tau_T} (T_0 - T)$$

Replacing $T' = \lambda^2 T$, we get:

$$\lambda^2 T = T + \frac{\Delta t}{\tau_T} (T_0 - T)$$

$$\lambda = \sqrt{1 + \frac{\Delta t}{\tau_T} \left(\frac{T_0}{T} - 1 \right)}$$

Step 8. Implementing the Berendsen thermostat

Each MD step:

1. Compute instantaneous temperature T from current velocities.
2. Compute $\lambda = \sqrt{1 + \frac{\Delta t}{\tau_T} \left(\frac{T_0}{T} - 1 \right)}$
3. Rescale velocities: $\mathbf{v}_i \leftarrow \lambda \mathbf{v}_i$
 - If τ_T is large \rightarrow weak coupling \rightarrow system evolves almost naturally.
 - If τ_T is small \rightarrow strong coupling \rightarrow faster temperature adjustment (but less physical).

Advantages and limitations

Advantages:

- Simple and efficient implementation.
- Fast equilibration to desired temperature.
- Adjustable coupling strength τ_T .

Disadvantages:

- Does not generate a true canonical ensemble (distribution of kinetic energy is narrower).

Background: Nosé-Hoover Thermostat

What Nosé Tried to Do

Nosé wanted to derive canonical (constant T) dynamics from a purely Hamiltonian system — i.e., one that conserves energy but still samples a temperature-controlled ensemble.

To achieve that, he introduced an **extended system** with:

- Physical coordinates and momenta: $\mathbf{r}_i, \mathbf{p}_i$
- A *thermostat variable* s (which scales time)
- A conjugate momentum p_s for s
- A mass parameter Q for the thermostat

Nosé extended Hamiltonian

Nosé introduced an extra degree of freedom s (a scale factor for time/momenta) with conjugate momentum p_s and an associated mass parameter Q . The extended Hamiltonian is:

$$\mathcal{H}_N = \sum_i \frac{\mathbf{p}_i^2}{2m_i s^2} + U(\{\mathbf{q}\}) + \frac{p_s^2}{2Q} + gk_B T \ln s.$$

Remarks:

- The s factor rescales momenta, so the physical kinetic energy depends on s .
- The $gk_B T \ln s$ term ensures the correct weighting that leads to canonical sampling for the physical variables (after transforming to real time and integrating out the extended variables).

- q_i, p_i : coordinates and momenta of particle i (physical variables).
- m_i : mass of particle i .
- $U(\{\mathbf{q}\})$: potential energy.
- k_B : Boltzmann constant.
- Target (bath) temperature: T .
- g : number of degrees of freedom coupled to the thermostat (commonly $g = 3N$ minus constraints).
- Goal:** modify dynamics so that the *physical* coordinates sample the canonical ensemble at temperature T .

Equations of motion in extended (Nosé) time τ

Treating \mathcal{H}_N as a Hamiltonian with time parameter τ , Hamilton's equations give:

$$\frac{d\mathbf{q}_i}{d\tau} = \frac{\partial \mathcal{H}_N}{\partial \mathbf{p}_i} = \frac{\mathbf{p}_i}{m_i s^2}$$

$$\frac{d\mathbf{p}_i}{d\tau} = -\frac{\partial \mathcal{H}_N}{\partial \mathbf{q}_i} = -\nabla_i U(\{\mathbf{q}\})$$

$$\frac{ds}{d\tau} = \frac{\partial \mathcal{H}_N}{\partial p_s} = \frac{p_s}{Q}$$

$$\frac{dp_s}{d\tau} = -\frac{\partial \mathcal{H}_N}{\partial s} = \sum_i \frac{\mathbf{p}_i^2}{m_i s^3} - \frac{gk_B T}{s}.$$

These are equations in the fictitious time τ . To connect to real MD time t we need a transformation.

Real-time transformation — Why we need it

In Nosé's extended system, the velocity of particle i is:

$$\frac{d\mathbf{r}_i}{d\tau} = \frac{\mathbf{p}_i}{m_i s^2}$$

But in *real dynamics*, we want:

$$\frac{d\mathbf{r}_i}{dt} = \frac{\mathbf{p}_i}{m_i s}$$

So, to fix this, Nosé defined:

$$dt = s d\tau \quad \Rightarrow \quad \frac{d}{dt} = \frac{1}{s} \frac{d}{d\tau}$$

That's the **real-time transformation**.

It makes physical velocities and accelerations meaningful again.

It is convenient to define a friction-like variable, $\zeta \equiv \frac{p_s}{\varrho}$.

Using the real-time derivative and expressing everything in physical variables leads to the Nosé–Hoover form.

4 — Nosé–Hoover equations (standard form)

Transforming to real time t and switching to momenta/velocities appropriate for MD gives the familiar Nosé–Hoover equations:

$$\dot{\mathbf{q}}_i = \frac{\mathbf{p}_i}{m_i} = \mathbf{v}_i$$

$$\dot{\mathbf{p}}_i = \mathbf{F}_i - \zeta \mathbf{p}_i \iff m_i \dot{\mathbf{v}}_i = \mathbf{F}_i - \zeta m_i \mathbf{v}_i$$

$$\dot{\zeta} = \frac{1}{Q} \left(\sum_i \frac{\mathbf{p}_i^2}{m_i} - gk_B T \right) = \frac{1}{Q} (2E_k - gk_B T)$$

where $\mathbf{F}_i = -\nabla_i U$, and $E_k = \frac{1}{2} \sum_i m_i v_i^2$ is the instantaneous kinetic energy.

These are deterministic ODEs in real physical time t .

A little work
around-
Homework

Temperature control interpretation

Differentiate the kinetic energy using Eqns (1–3):

$$\frac{dE_k}{dt} = \sum_i m_i \mathbf{v}_i \cdot \dot{\mathbf{v}}_i = \sum_i \mathbf{v}_i \cdot \mathbf{F}_i - 2\zeta E_k.$$

Ignoring the work term $\sum \mathbf{v}_i \cdot \mathbf{F}_i$ (it redistributes energy between kinetic and potential), the thermostat term $-2\zeta E_k$ controls the kinetic energy. From Eq. (3):

$$\dot{\zeta} = \frac{2E_k - gk_B T}{Q}.$$

Thus ζ adjusts so as to push E_k (hence temperature) toward the target value $gk_B T/2$. Linearizing around equilibrium yields a characteristic relaxation time scale

$$\tau_T \sim \sqrt{\frac{Q}{gk_B T}}.$$

So Q plays the role of a "mass" for the thermostat: larger $Q \Rightarrow$ slower thermostat; smaller $Q \Rightarrow$ tighter (faster) control — but very small Q can produce nonphysical oscillations.

Summary

1. Nosé introduced an extended Hamiltonian with a scaling variable s and mass Q .
2. Transforming to real time and eliminating p_s leads to the Nosé–Hoover equations:

$$\dot{\mathbf{q}}_i = \mathbf{v}_i,$$

$$m_i \dot{\mathbf{v}}_i = \mathbf{F}_i - \zeta m_i \mathbf{v}_i,$$

$$\dot{\zeta} = \frac{1}{Q} (2E_k - gk_B T).$$

3. Q controls the thermostat speed: $\tau_T \sim \sqrt{Q/(gk_B T)}$.
4. If the extended dynamics is ergodic, the physical variables sample the canonical ensemble; if not, chains or stochastic thermostats are used.