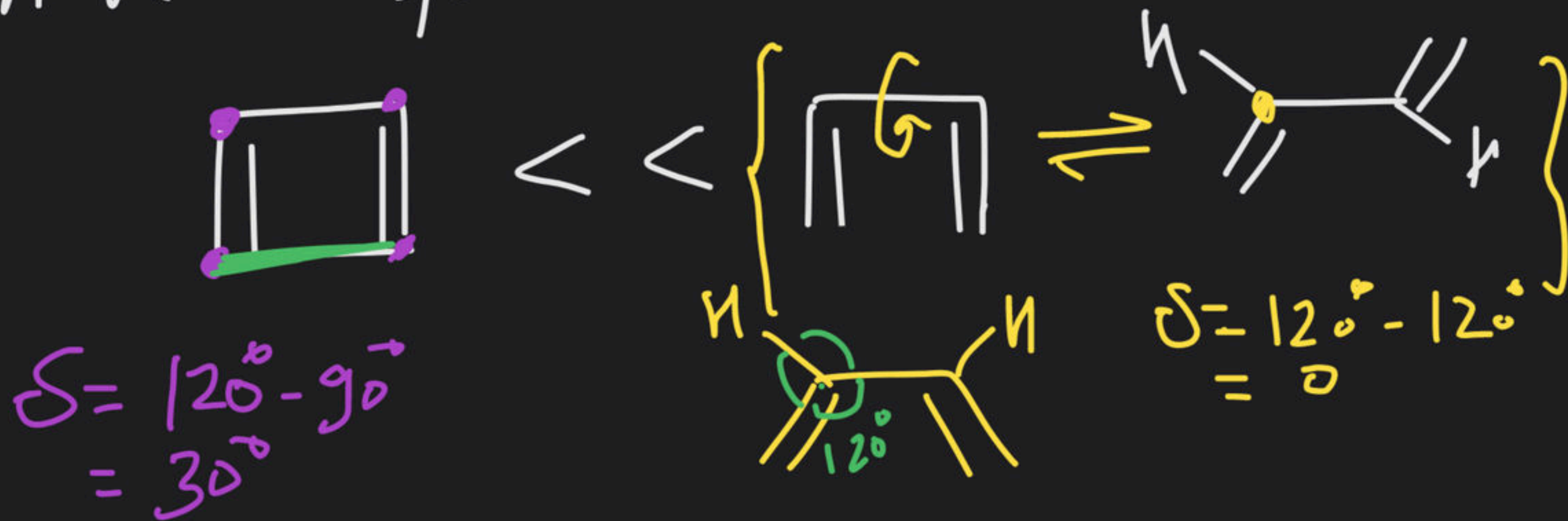


Resonance Energy, Sigma Resonance, SIR Effect

Course on General Organic Chemistry (GOC) for Dropper 13th students

(#) Anti Aromatic Compound

⇒ Cyclic compounds which are highly unstable than its open chain analogous form is known as antiaromatic compound.



Condition for a Compound to be Antiaromatic

Compound must

(a) be cyclic

(b) be planar

(c) be conjugated

(d) have $(4n)$ no. of πe^- , where $n = 1, 2, 3, \dots$

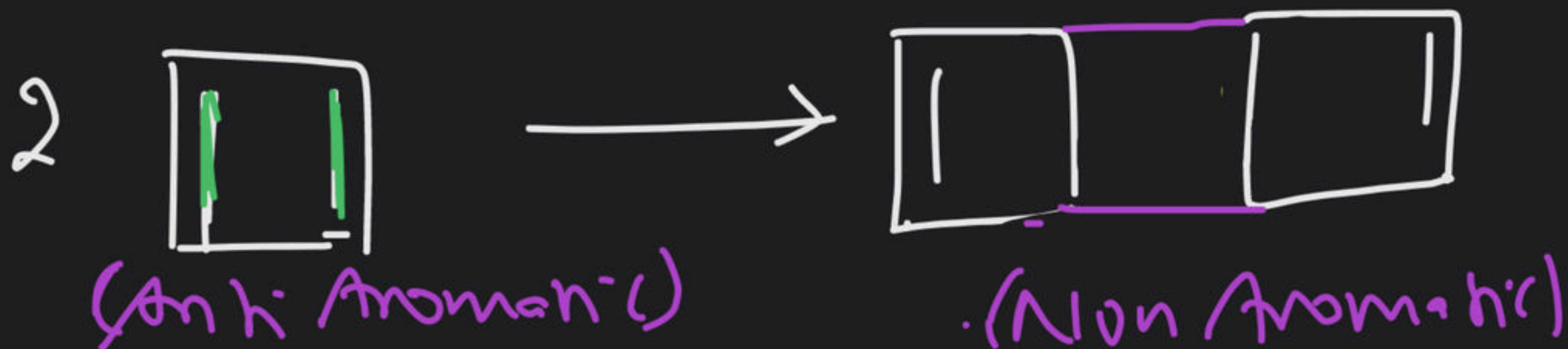
Note

(i) Cyclic compounds containing more than 7 carbon atoms are

100%

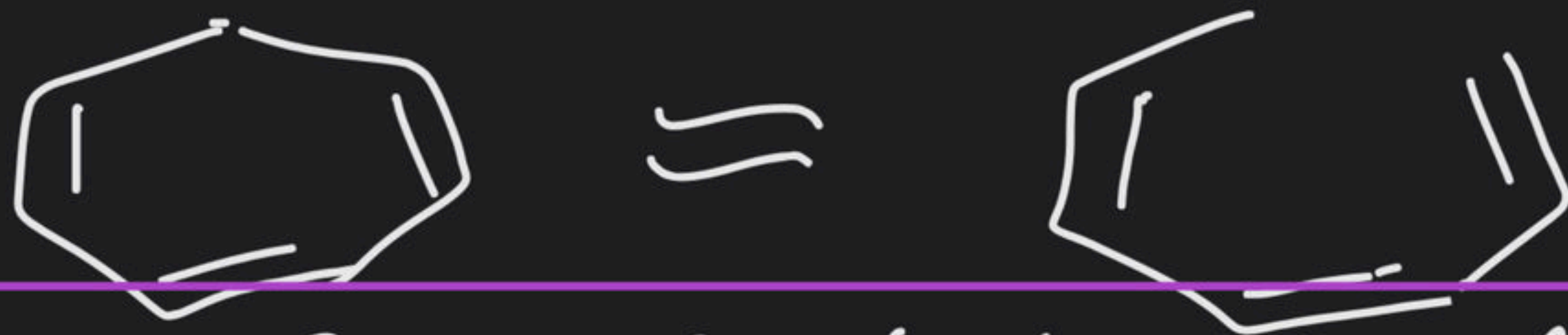
Either Aromatic ($4n+2$) or Non Aromatic (Rest all cases).

MA Imp
Anti aromatic compounds are highly unstable at Room Temperature & don't exist. AA compound easily dimerises.



(#) Non Aromatic

Cyclic compounds which are comparatively stable than its open chain analog form.



⇒ Condⁿ for a Compound to be Non Aromatic

Compound should not be

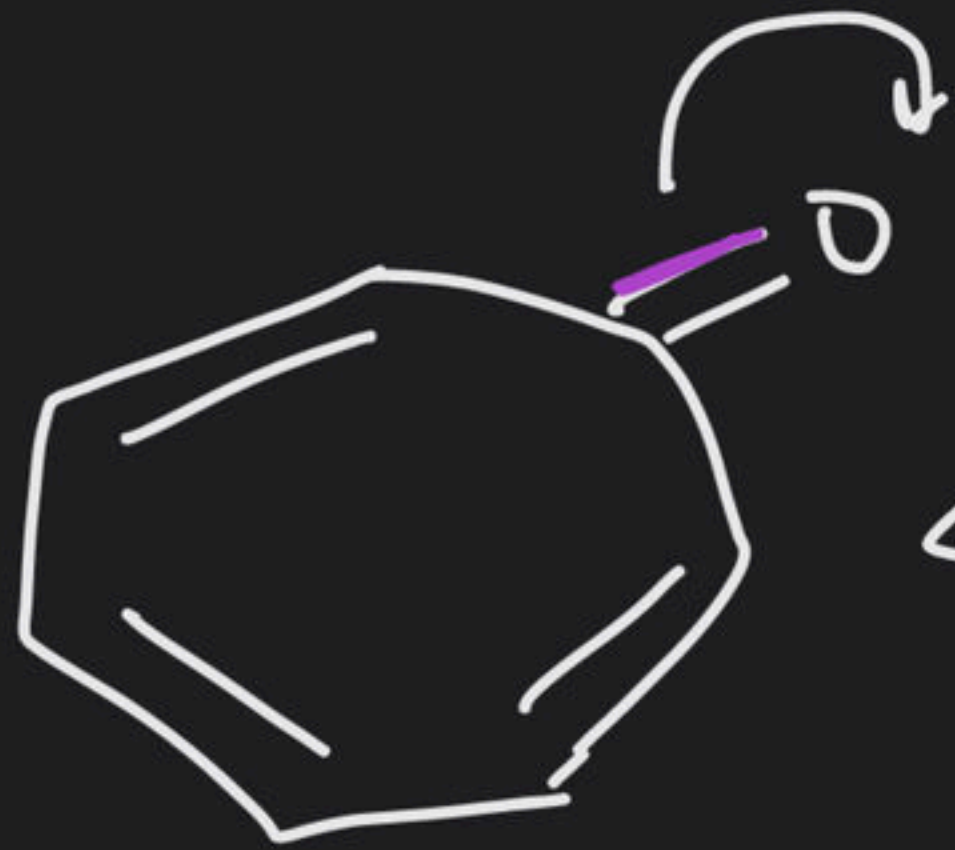
(a) Aromatic

(b) Anti Aromatic

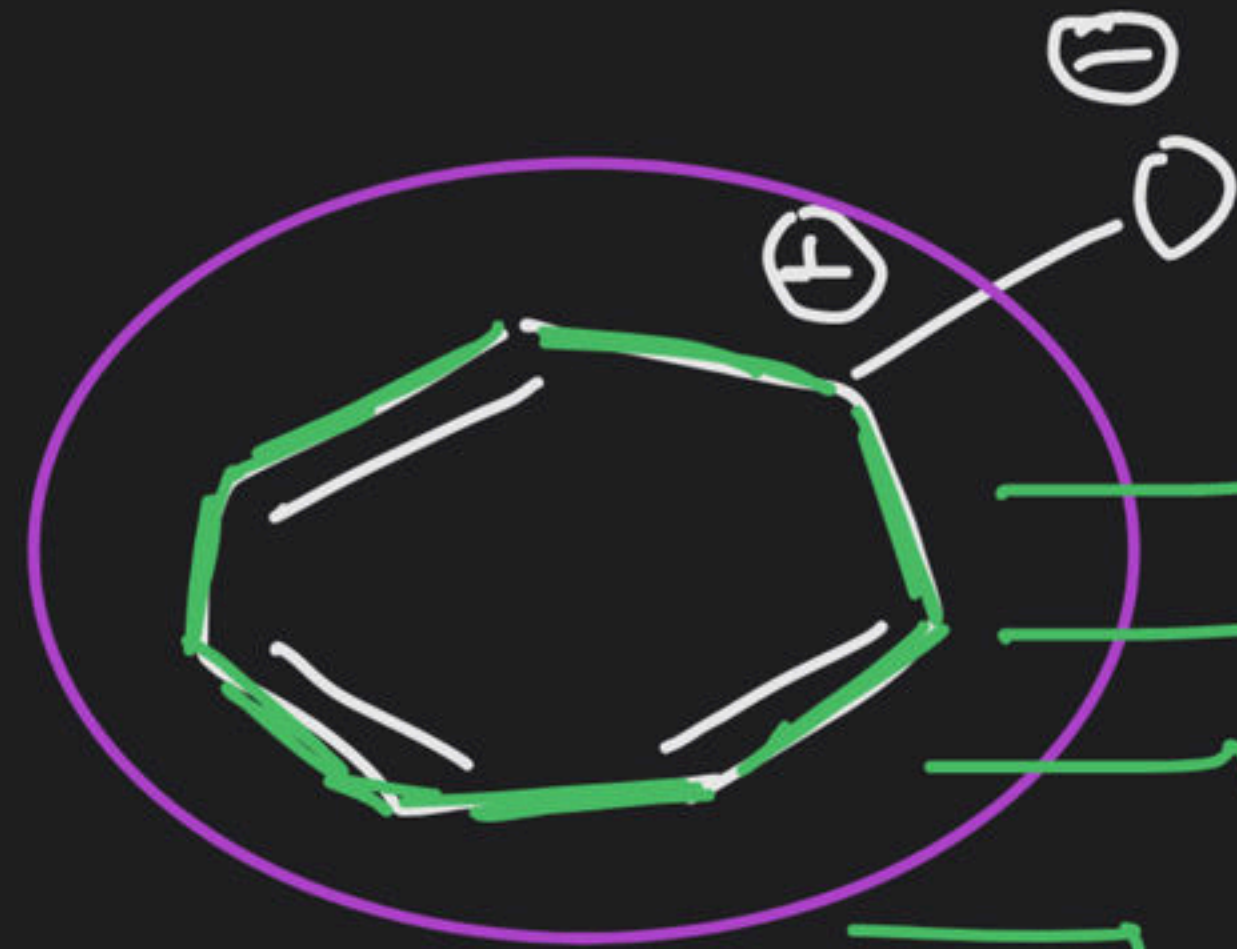
(#) Quasi Aromatic Compound

Aromatic Compounds having Charge Dispersion are known as Quasi Aromatic Compound.

Ex:



Aromatic



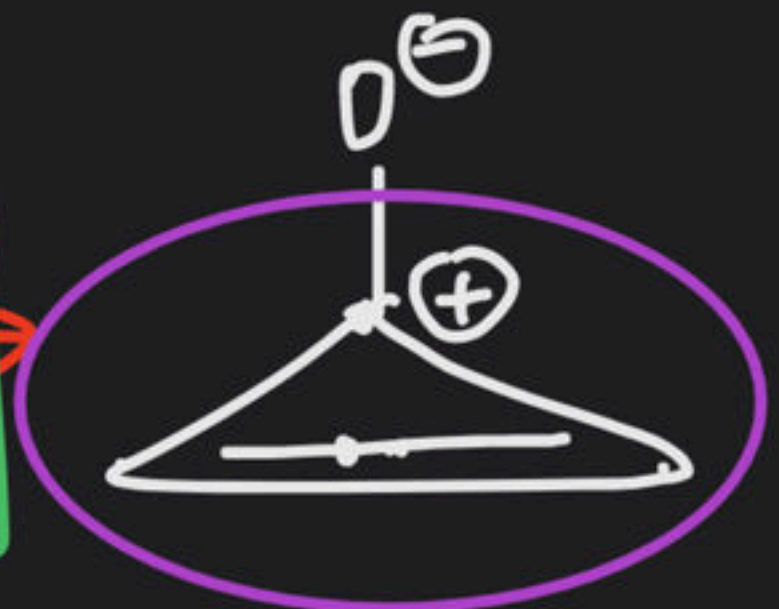
Aromatic

Quasi Aromatic

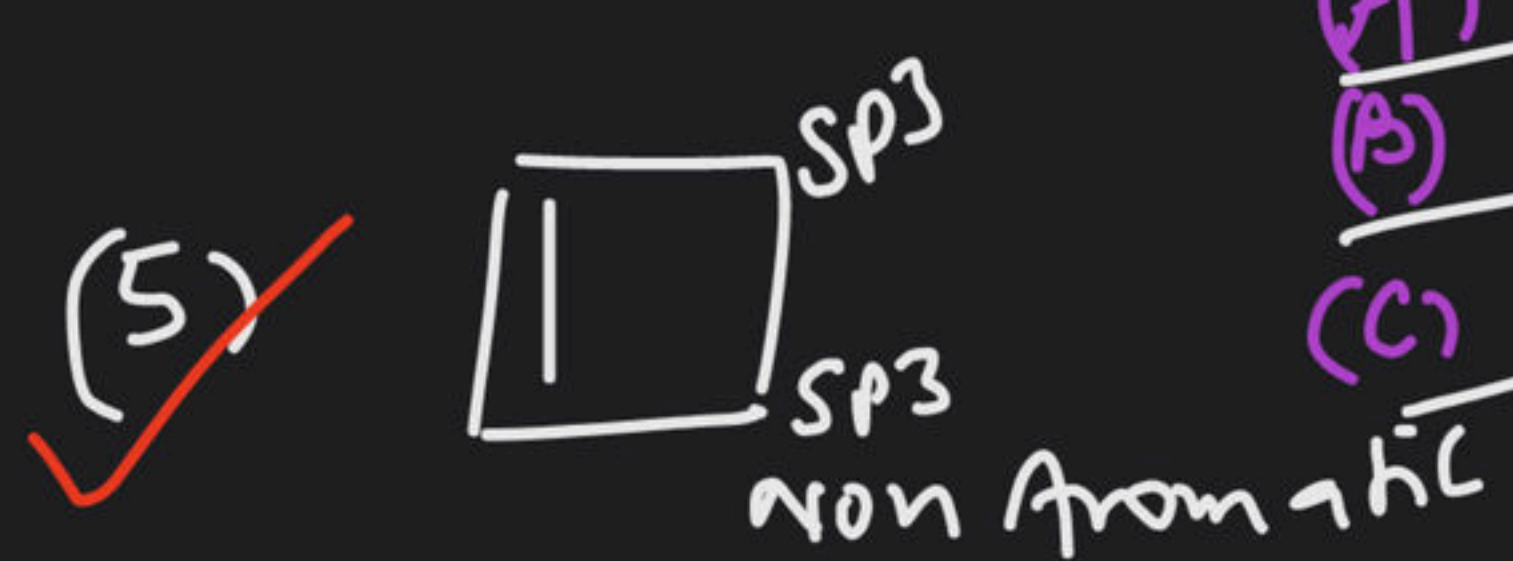
Cyclic ✓
planar ✓
Conjugated ✓
6 π e⁻s ✓



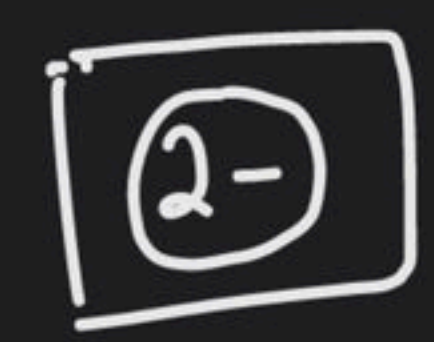
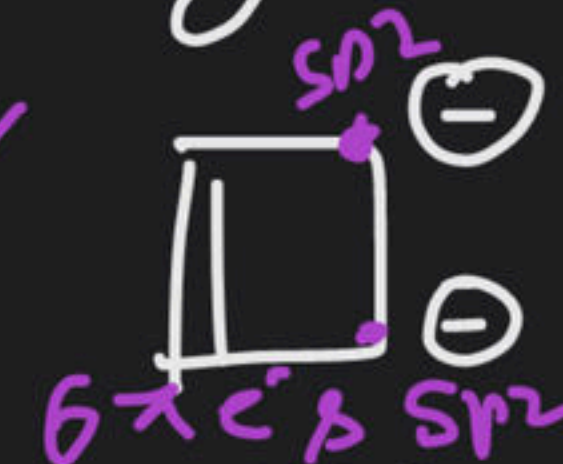
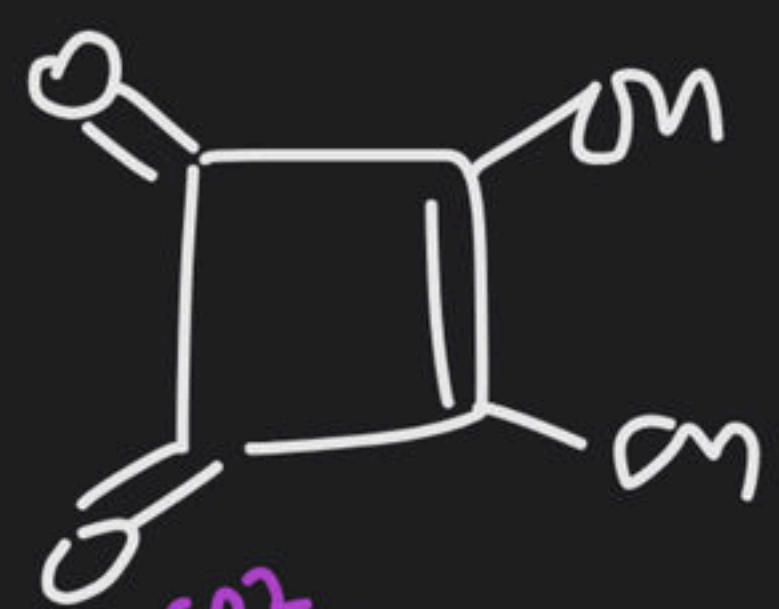
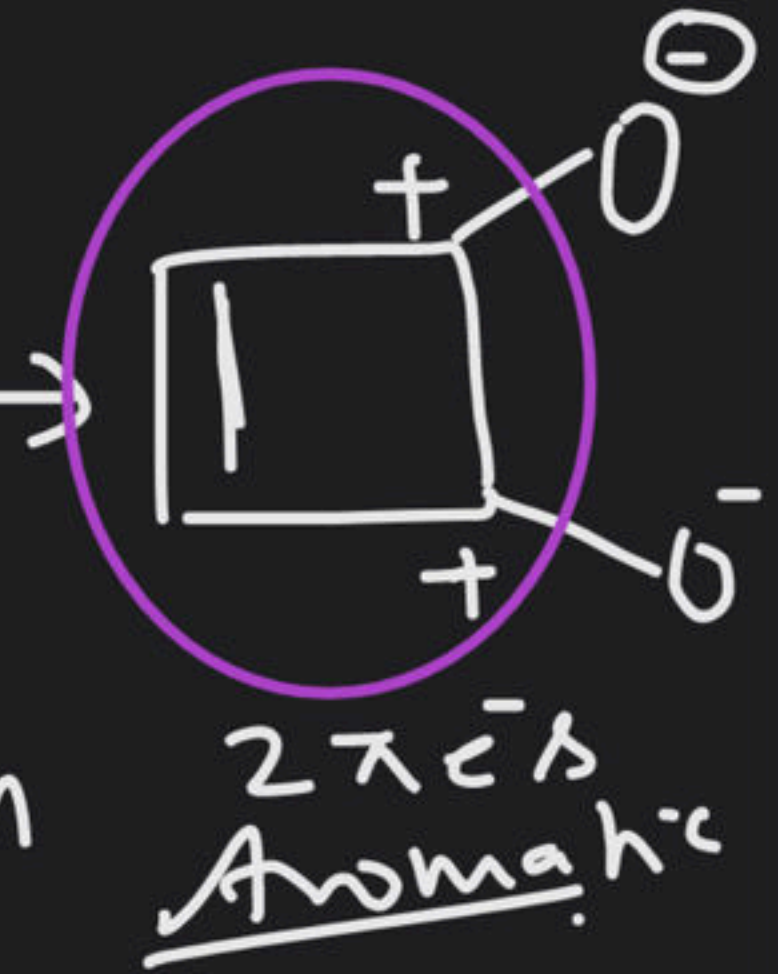
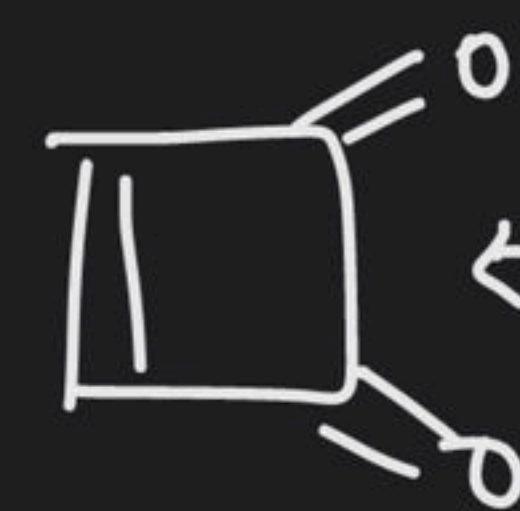
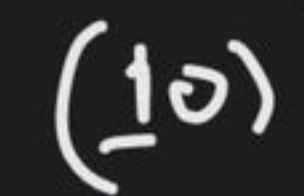
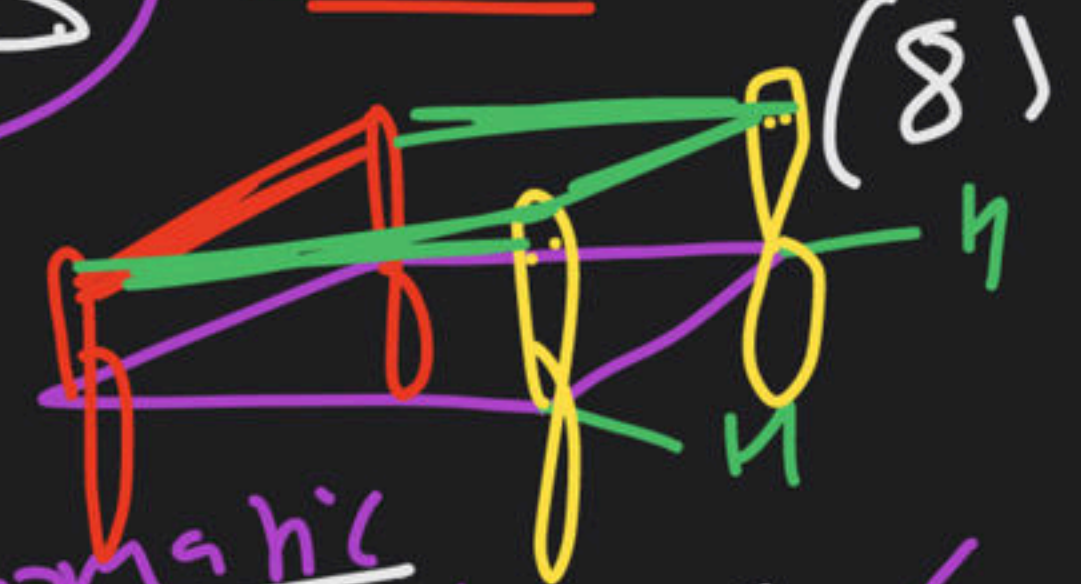
cyclic ✓
 planar ✓
 conjugated ✗
 (Non Aromatic) (6)

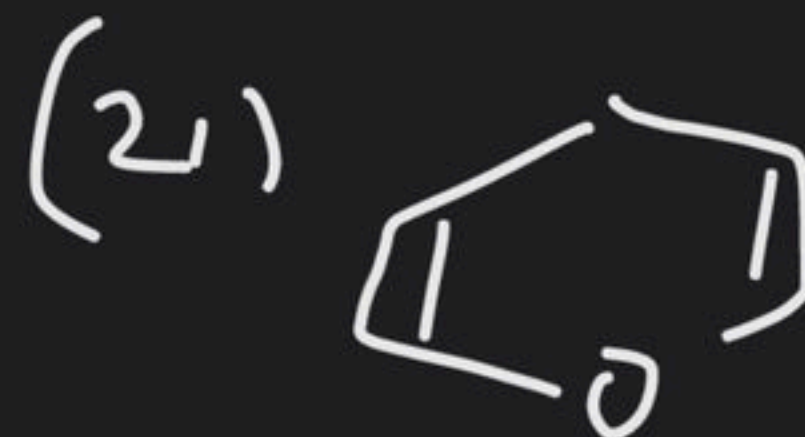
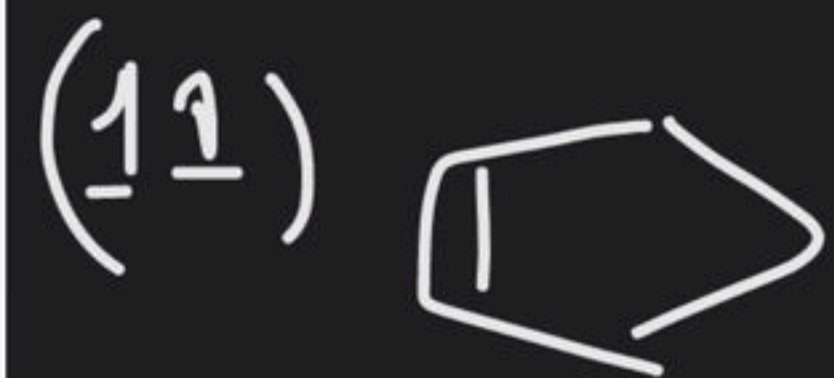


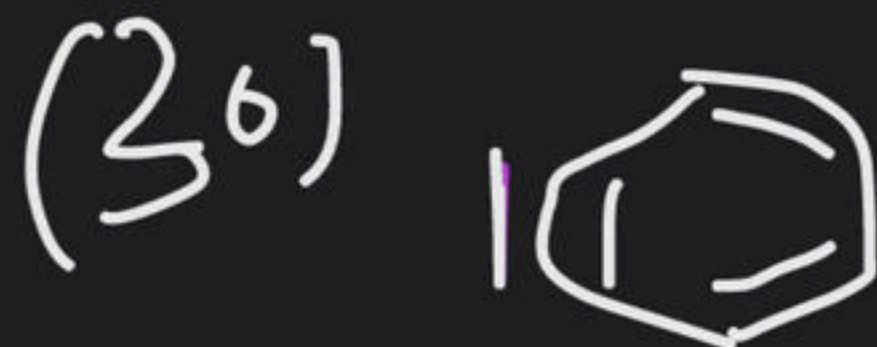
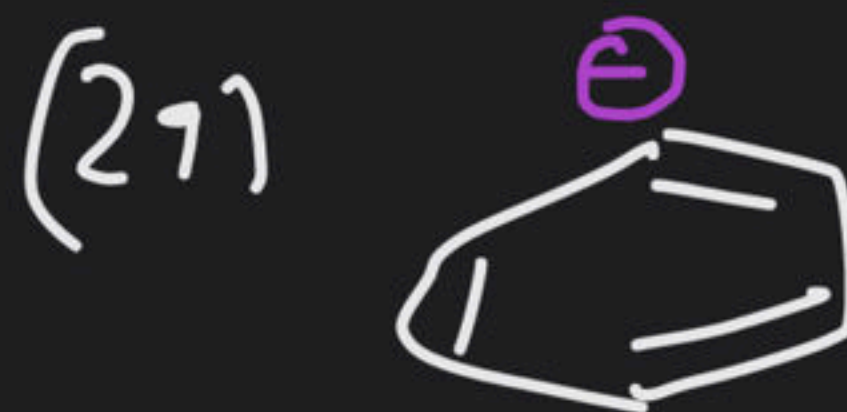
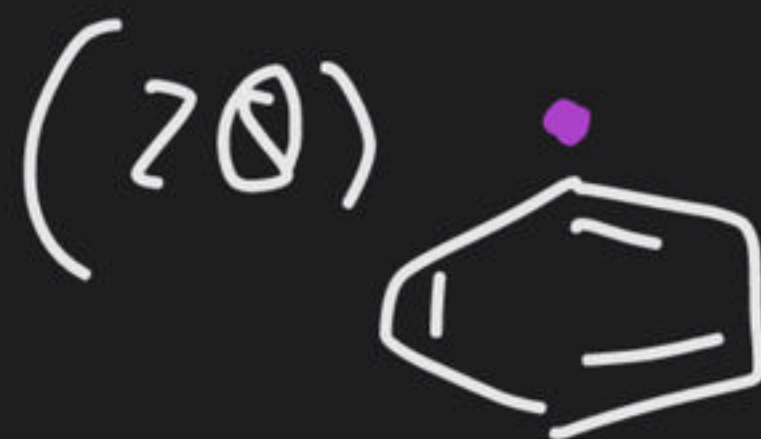
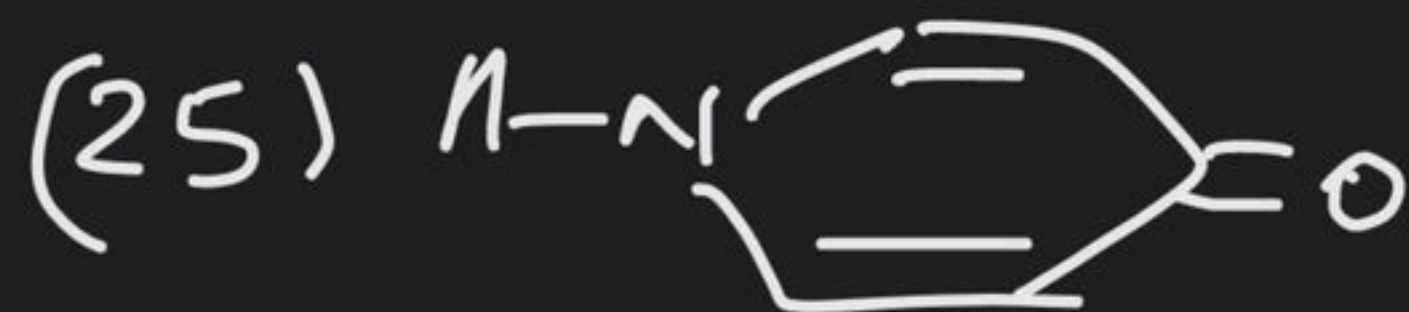
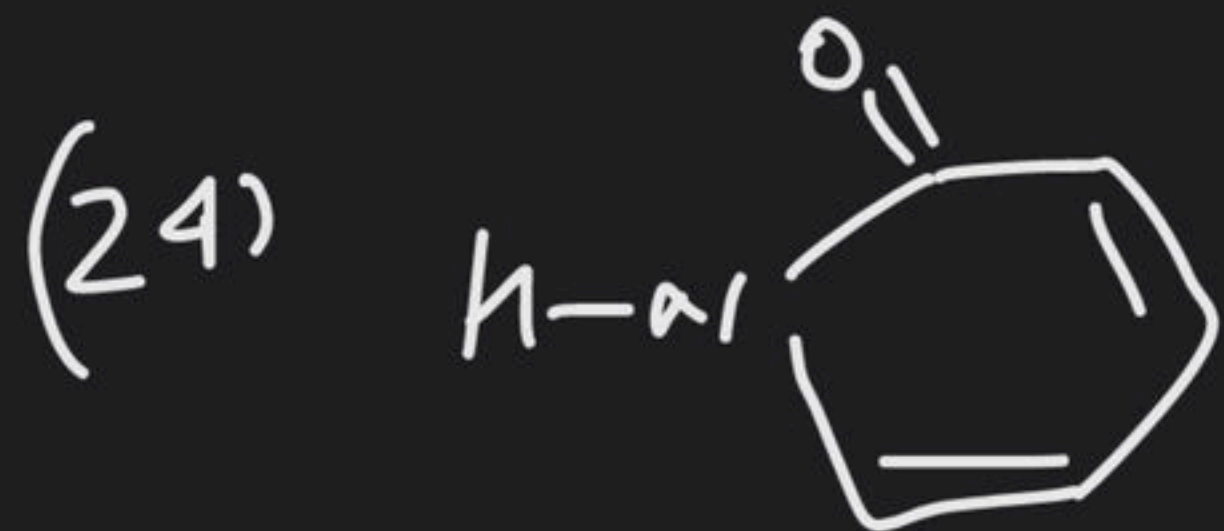
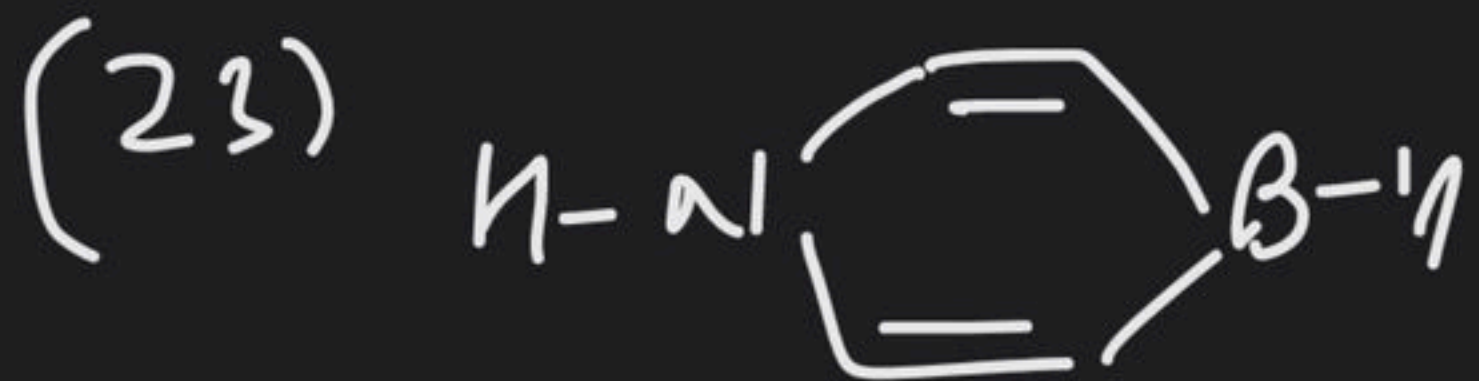
Aromatic



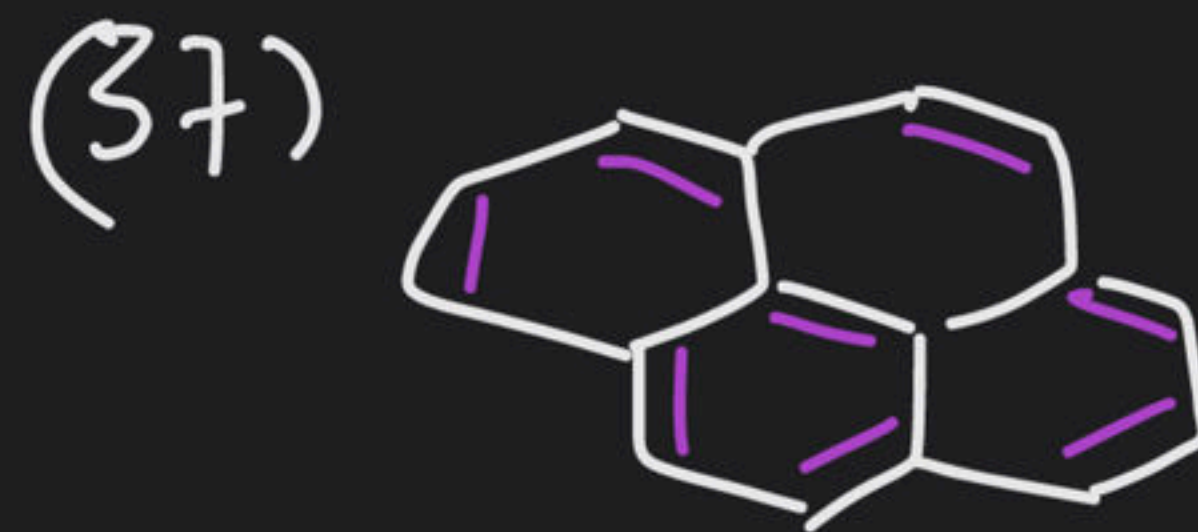
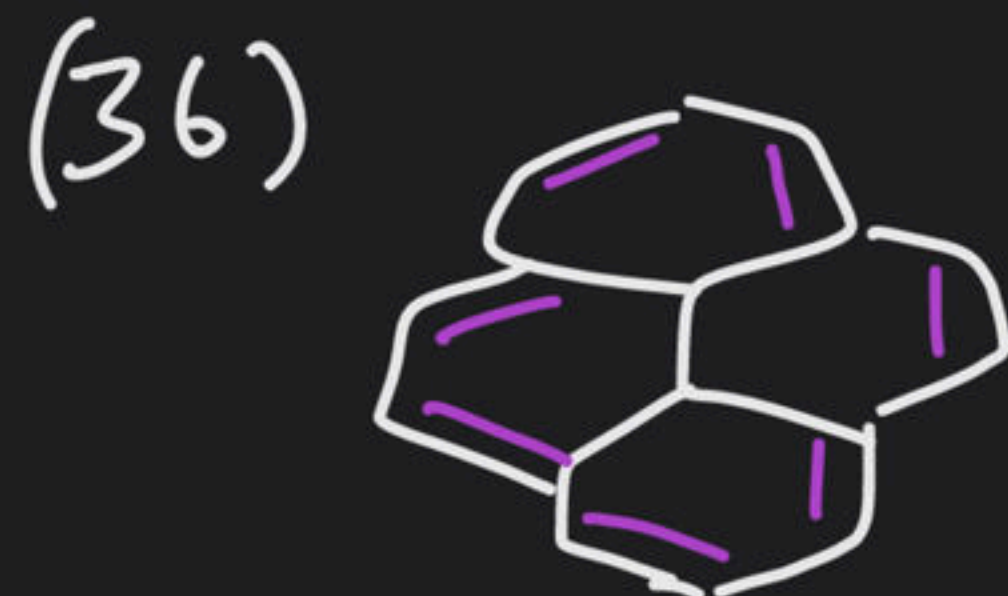
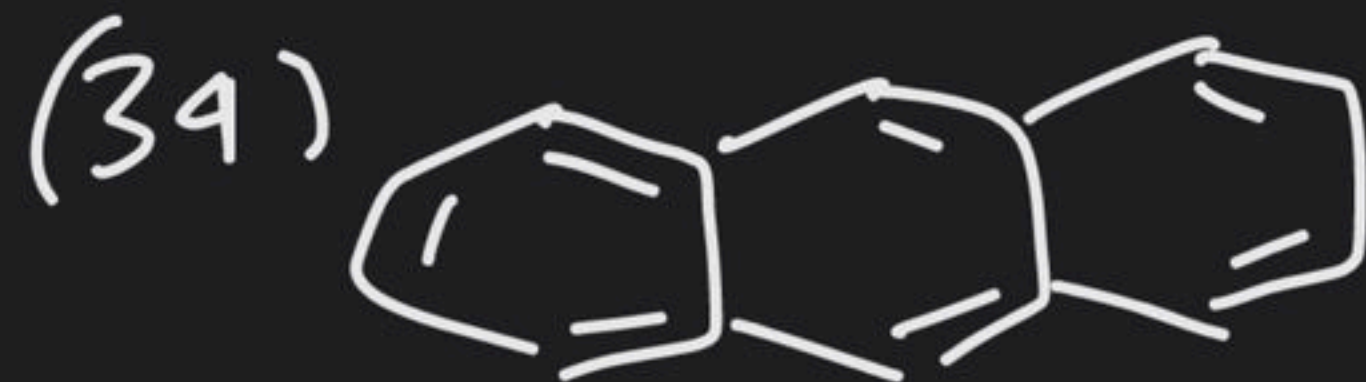
- (A) Aromatic
- (B) Non Aromatic
- (C) Anti Aromatic







(32) Inorganic Benzene / $B_3N_3H_6$ / Borazine or Borazole



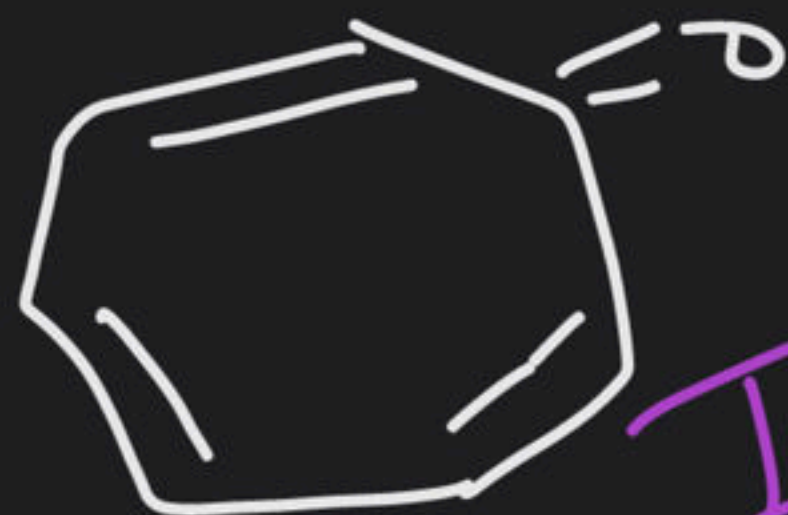
(42)



(43)



(44)



Imp
(45) Annulene [0]



COT

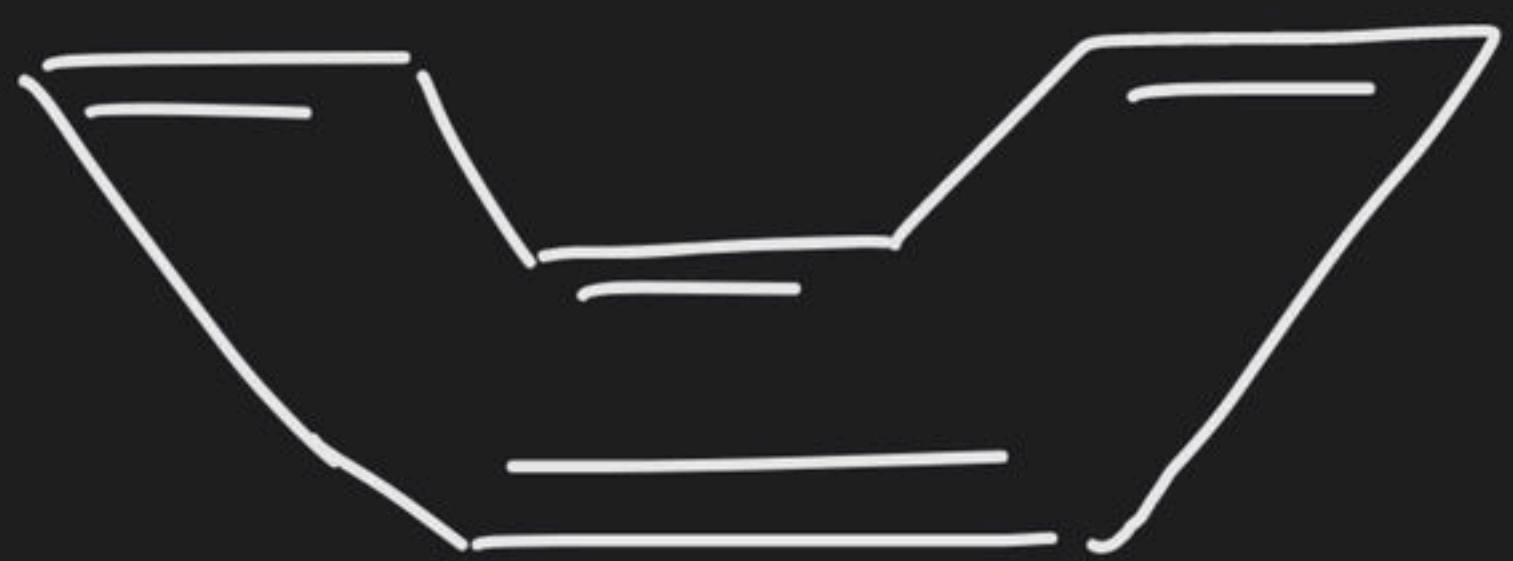
(4) Annulene [n]

\Rightarrow Cyclic polyene
atom & $\frac{n}{2}$
bonds.)

with n Carbon
no. of double

(Cyclo Octa
Tetraene)

Non Aromatic
non planar



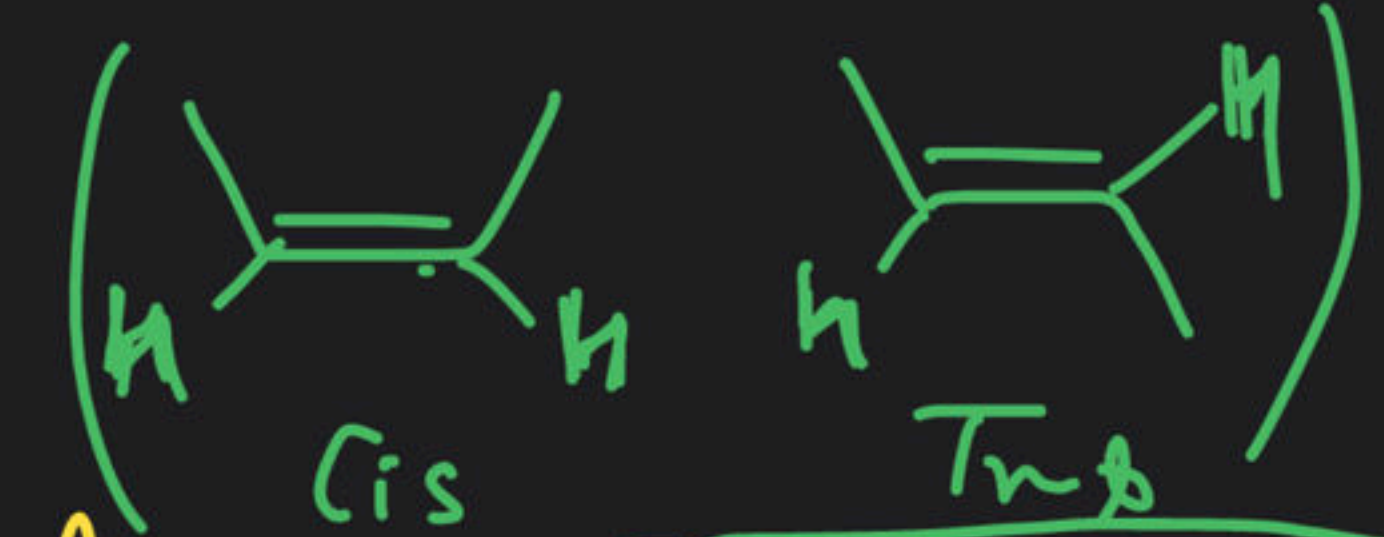
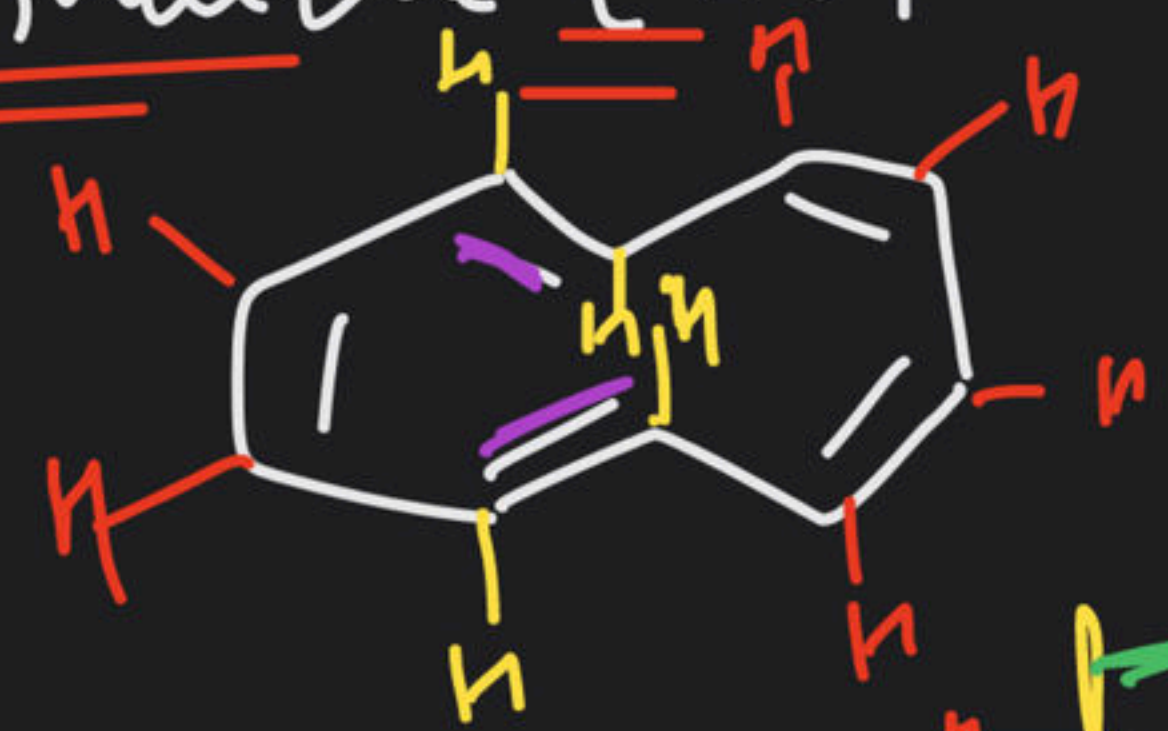
(Tub shaped)



(Tub shaped)

M. J. J.

(46) Annulene [10]

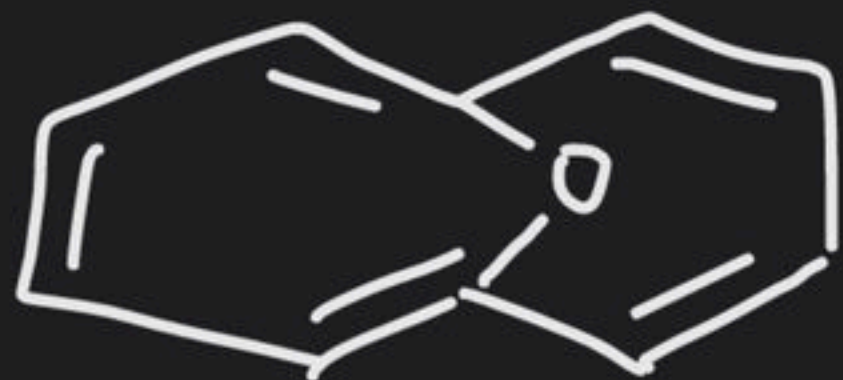


Non Aromatic

Non planar due to Repulsion b/w two trans inter-

vering H. atom.

(47)



Aromatic

(48)

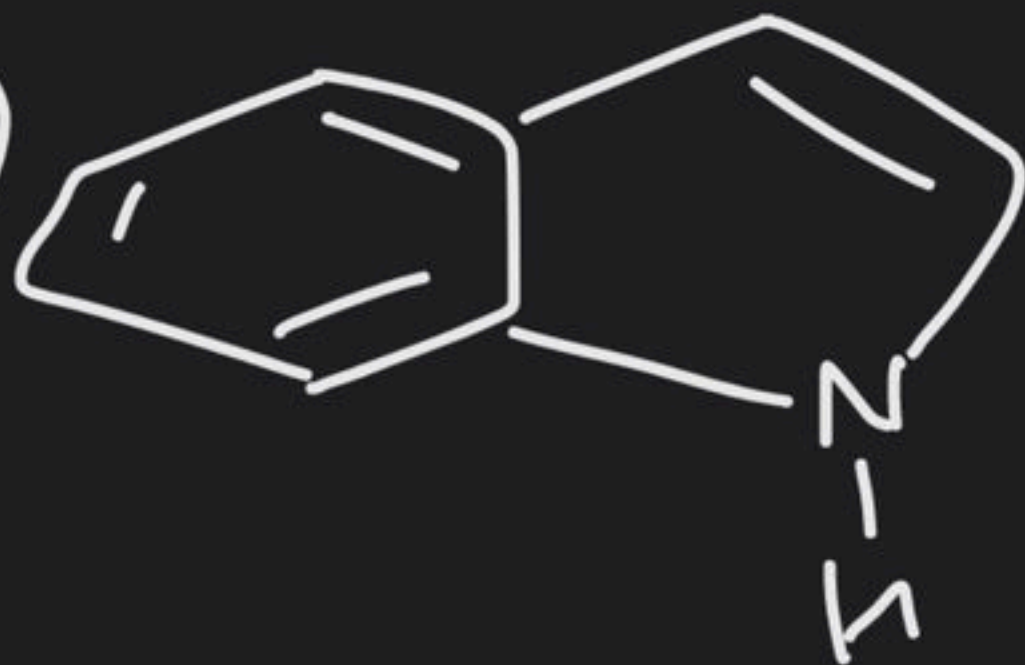


"

(49)

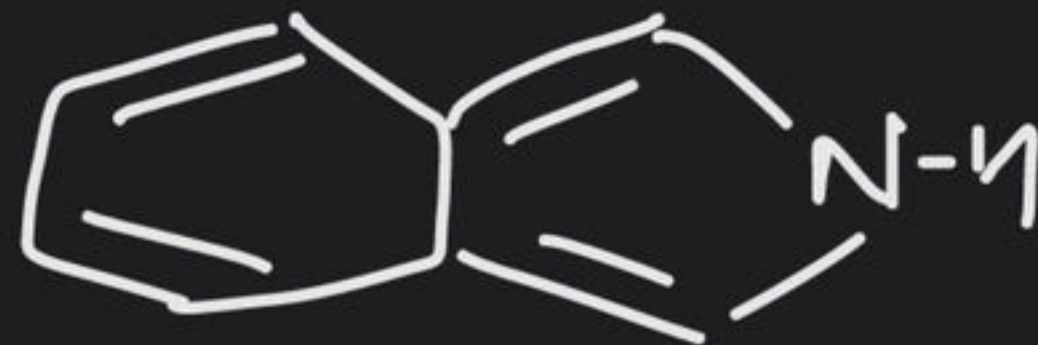


(50)

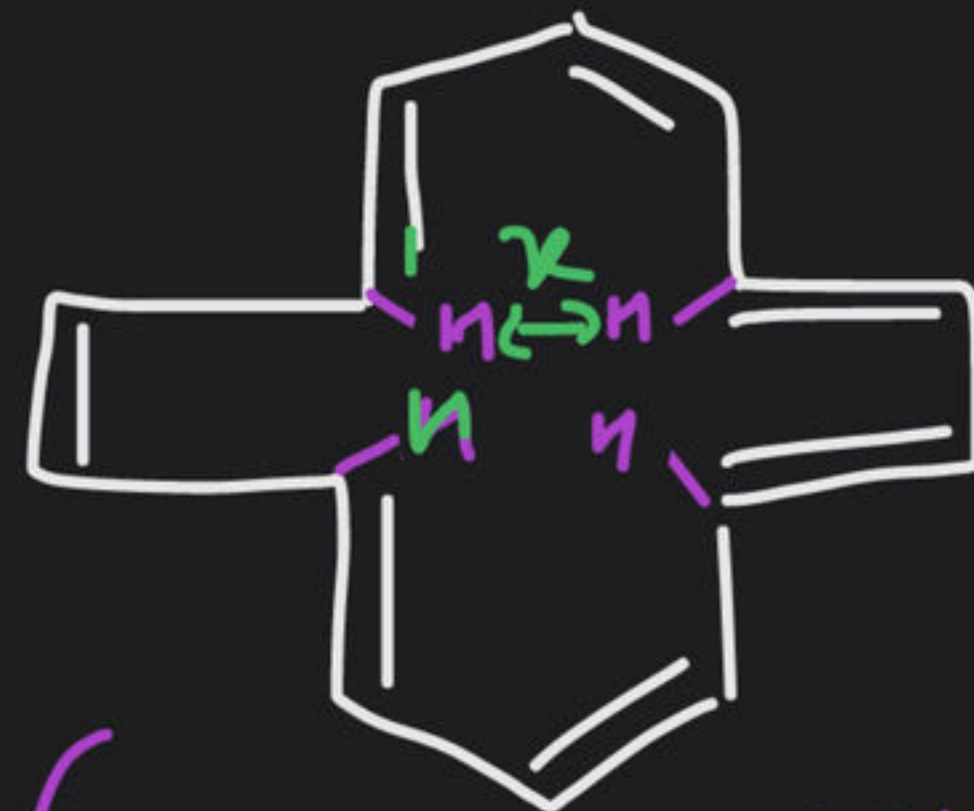


(Indole)

(51)

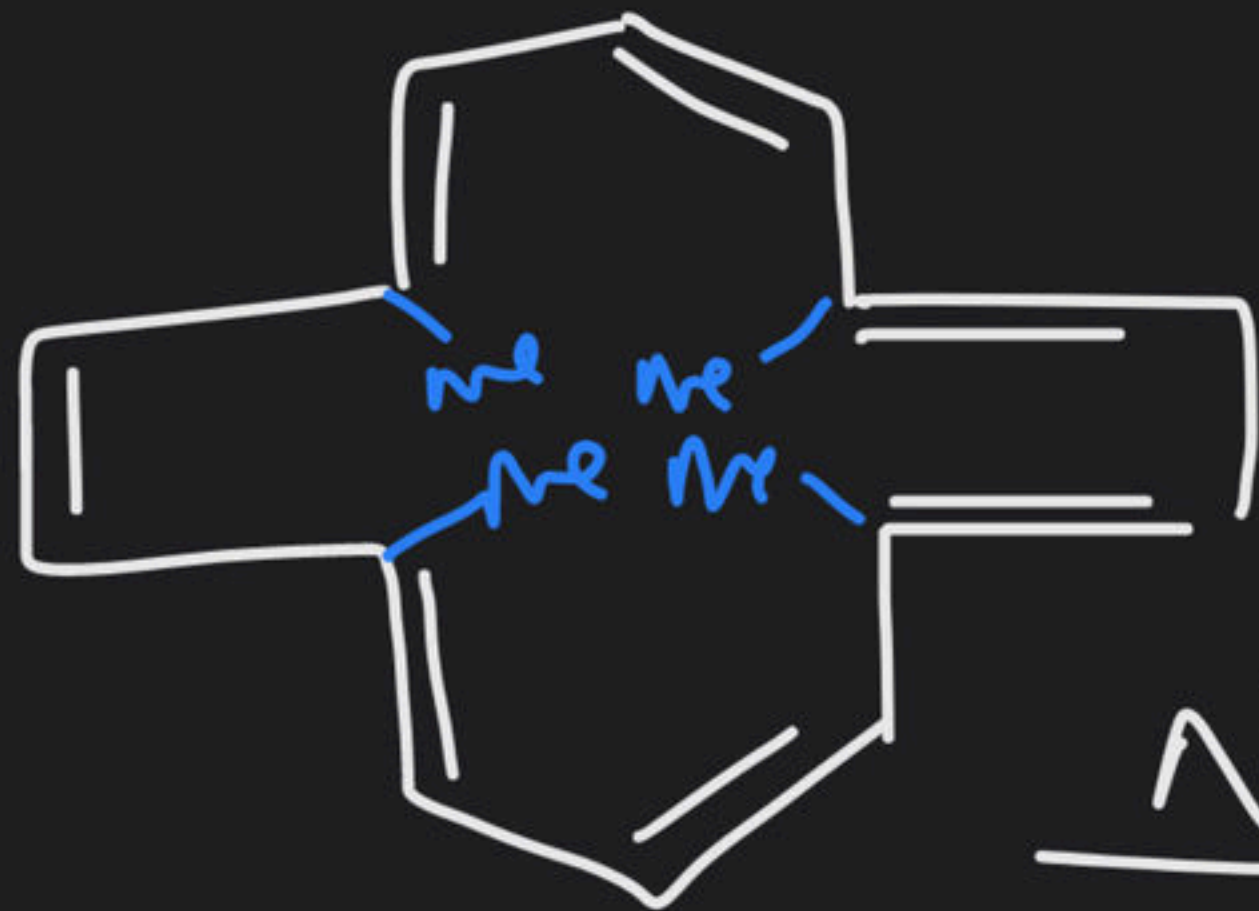


~~(52)~~ Annulene [14]



(Slight Repulsion
& Aromatic)

(53)



(A) Aromatic

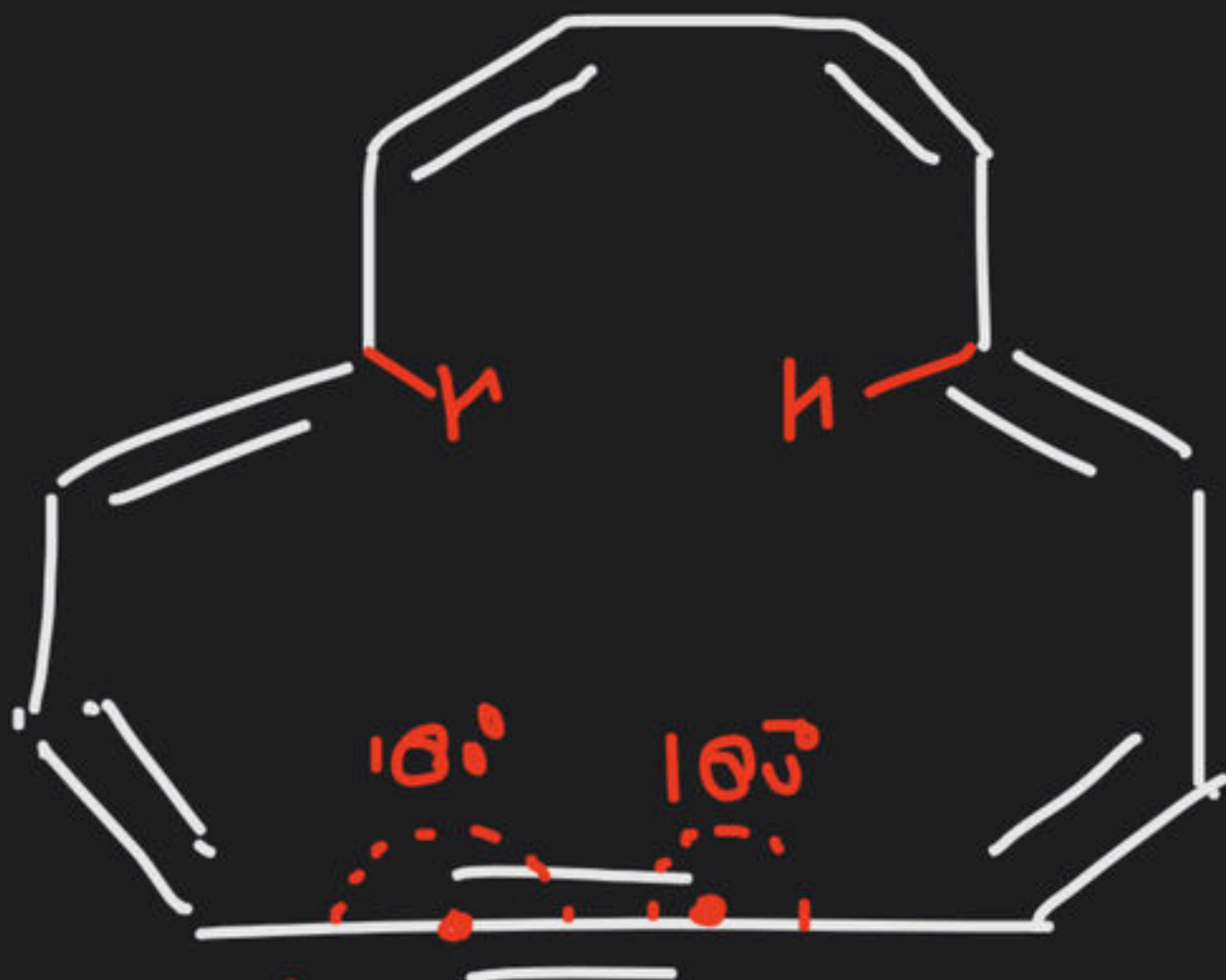
(B) Non Arom

(C) Anti

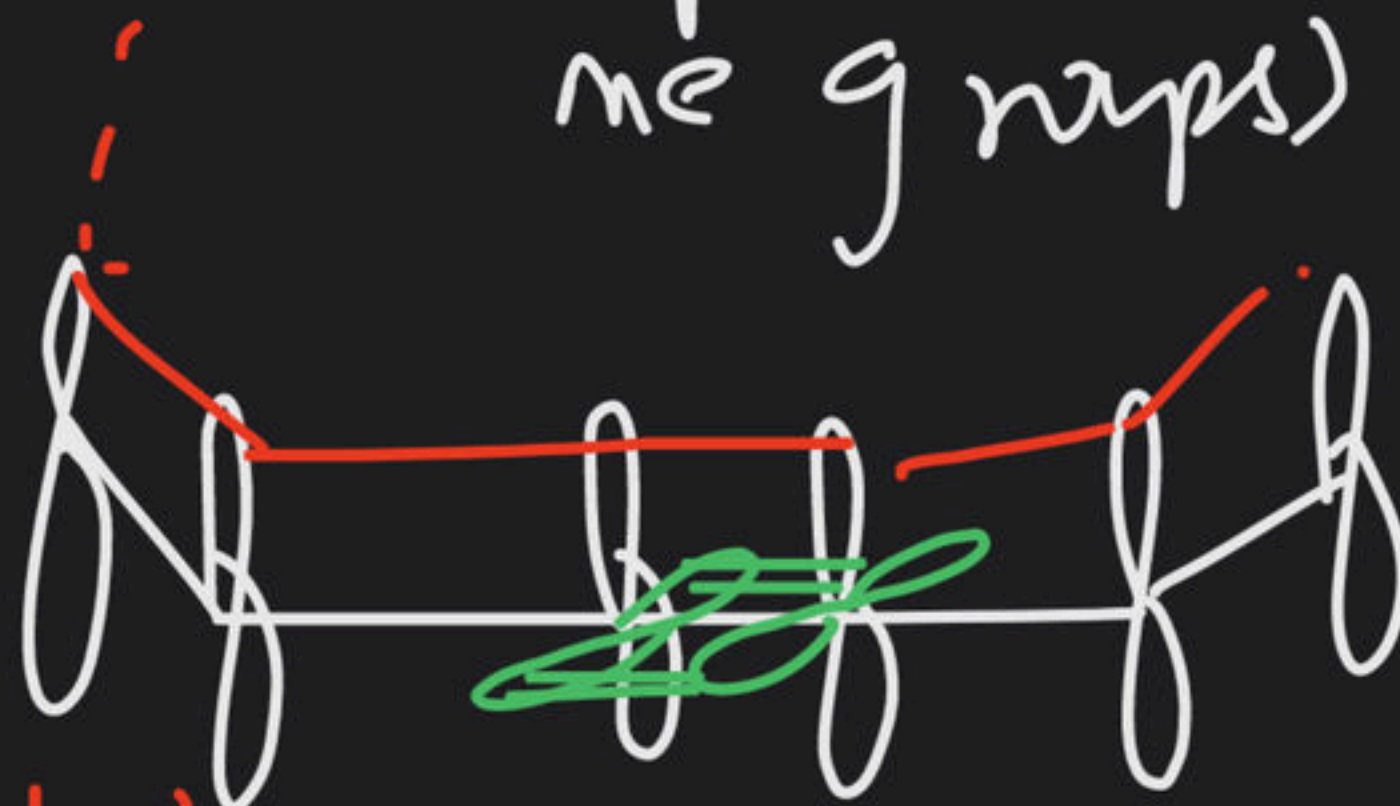
Non Aromatic (Non planar

due to huge
Repulsion b/w
me groups)

(54)



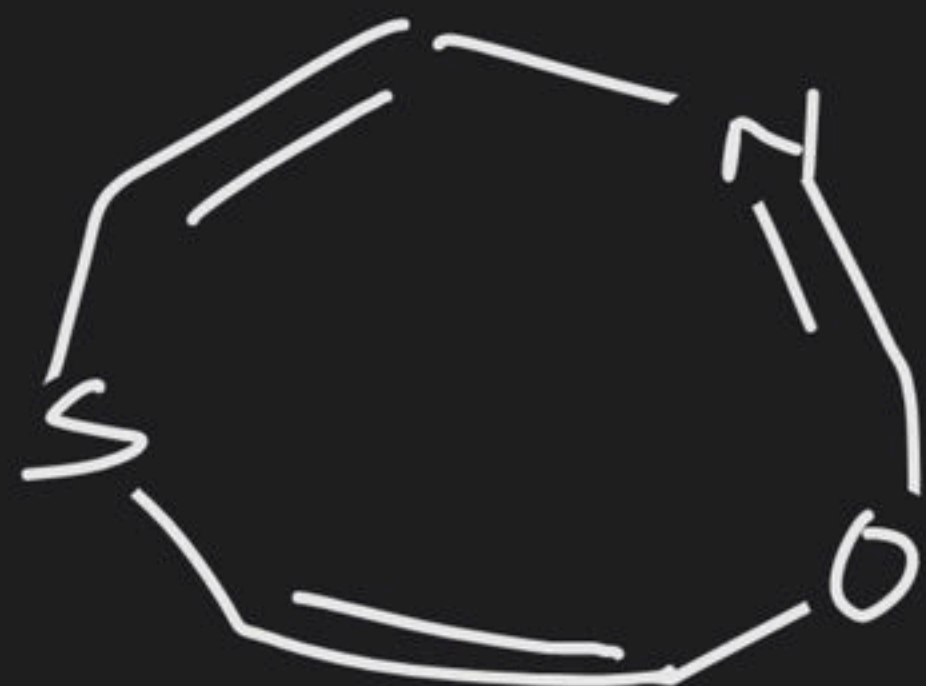
(14 πe^-) (Aromatic)



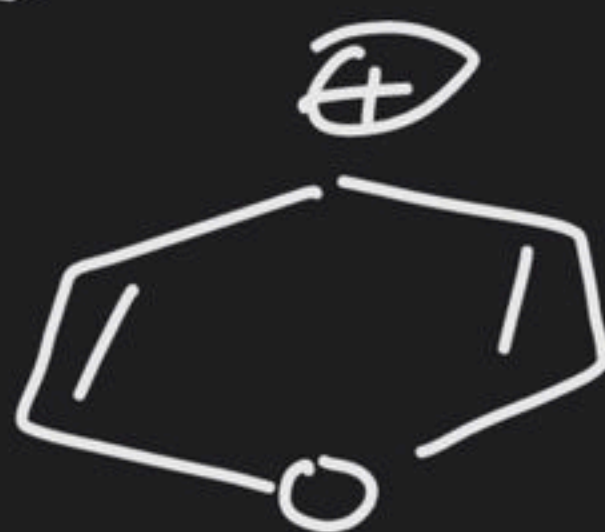
(55) Annulene [16]

(56) Annulene [18]

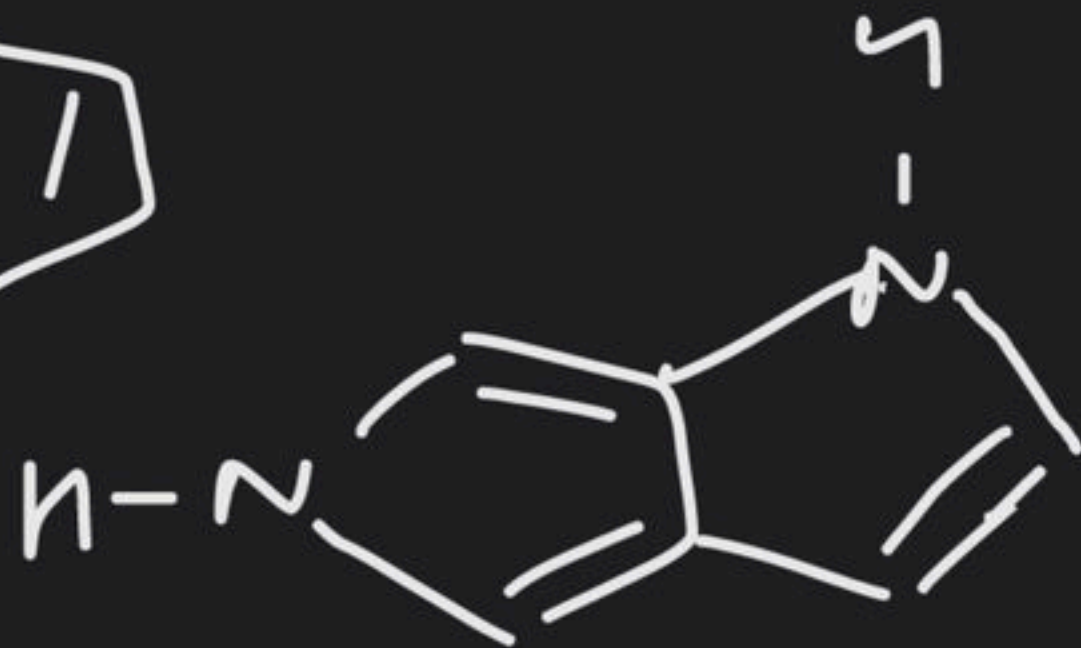
(57)



(58)



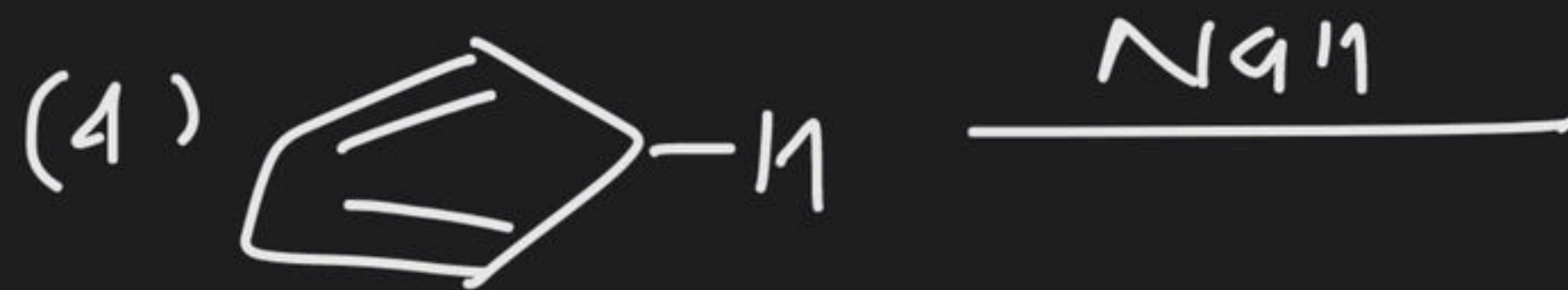
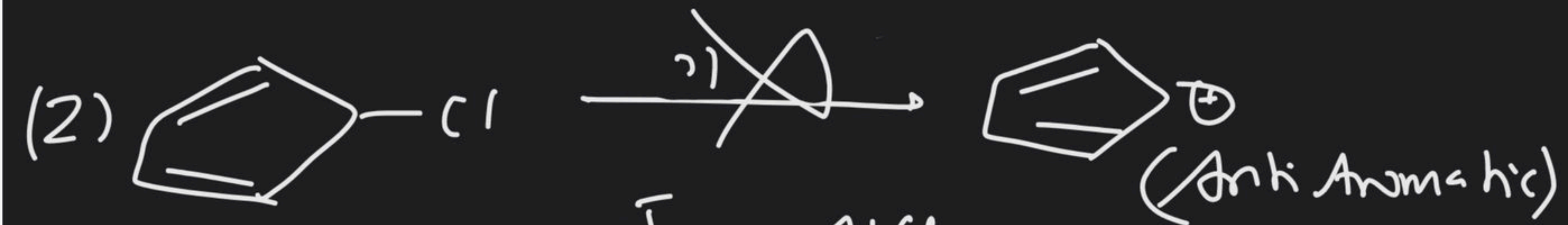
(59)

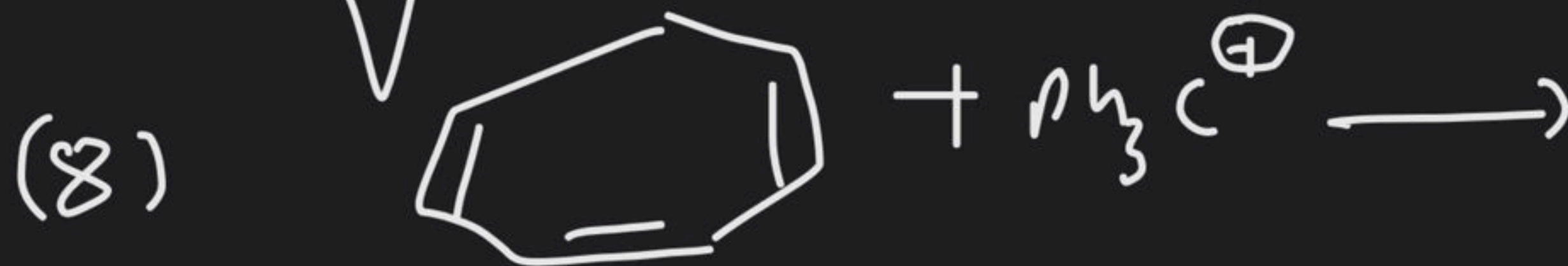


(60) Azulene

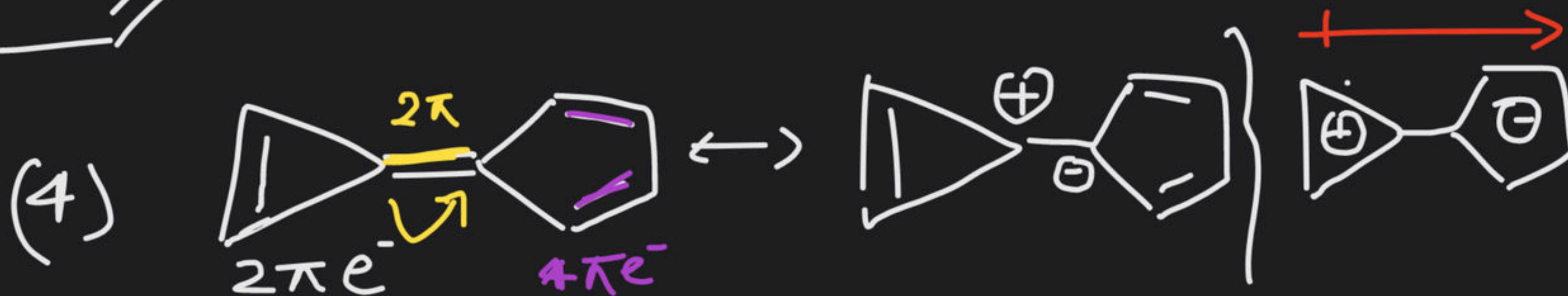


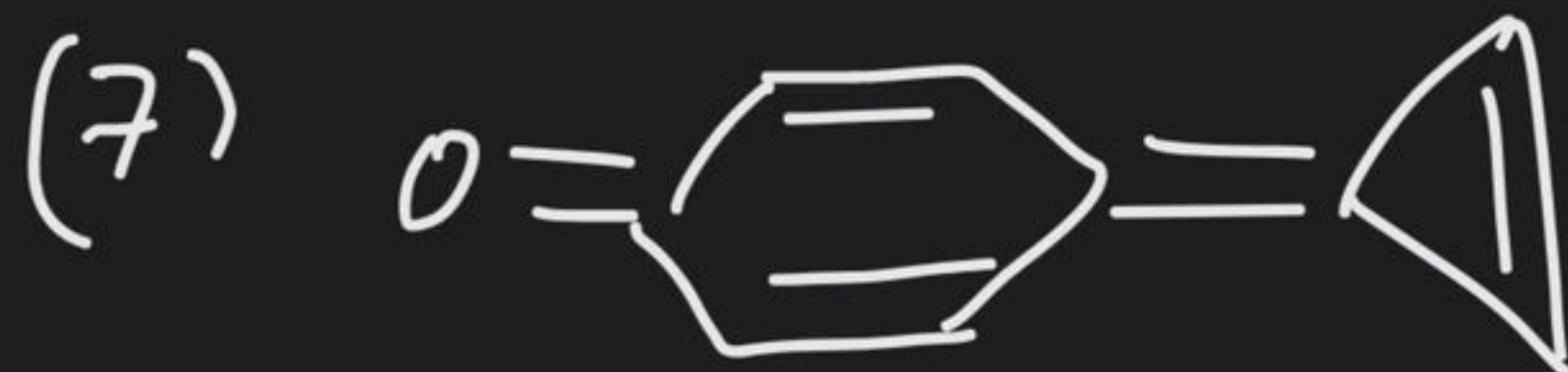
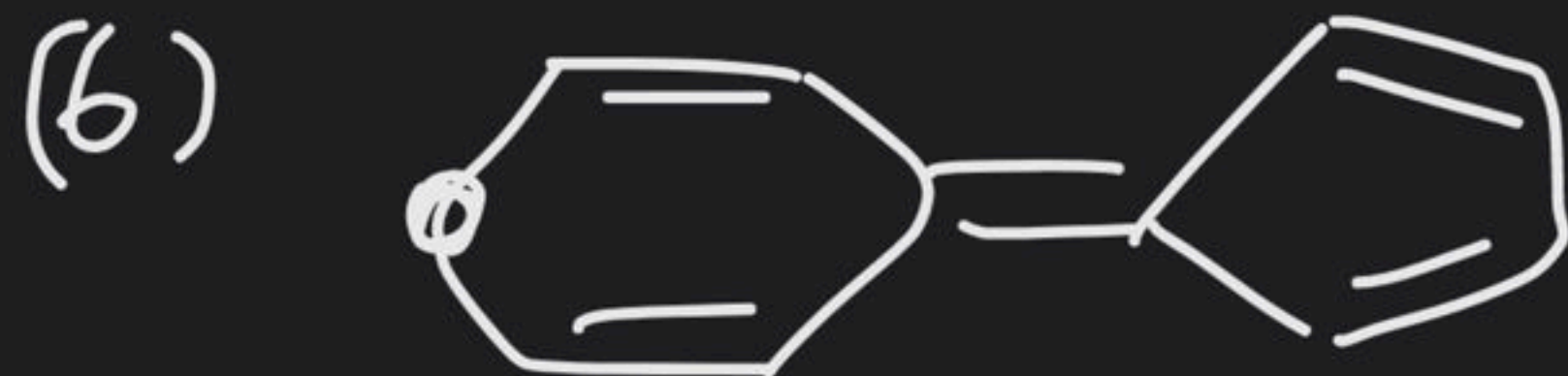
(61) Find Reactions
a) in which Aromatic product is obtained / which are feasible



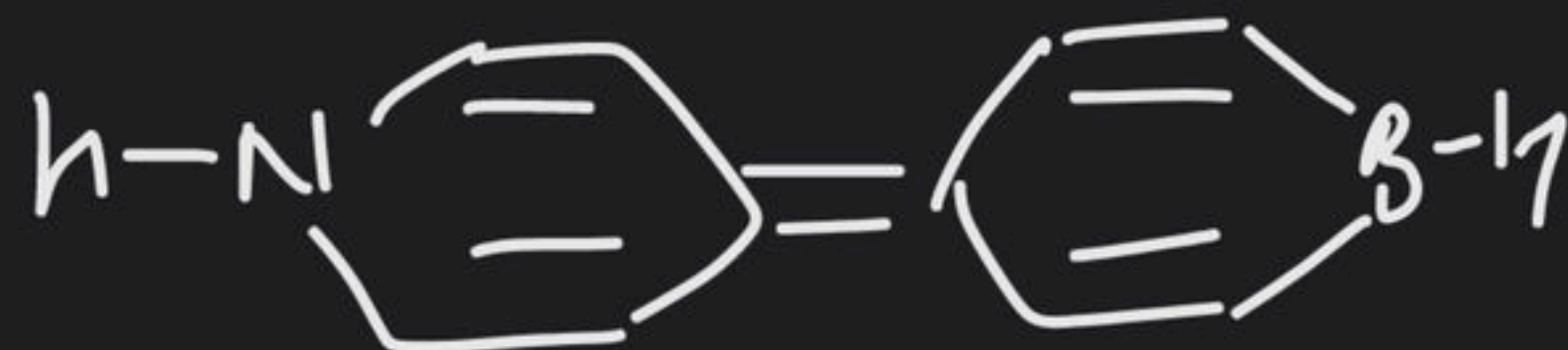


(62) Find direction of dipole moment 

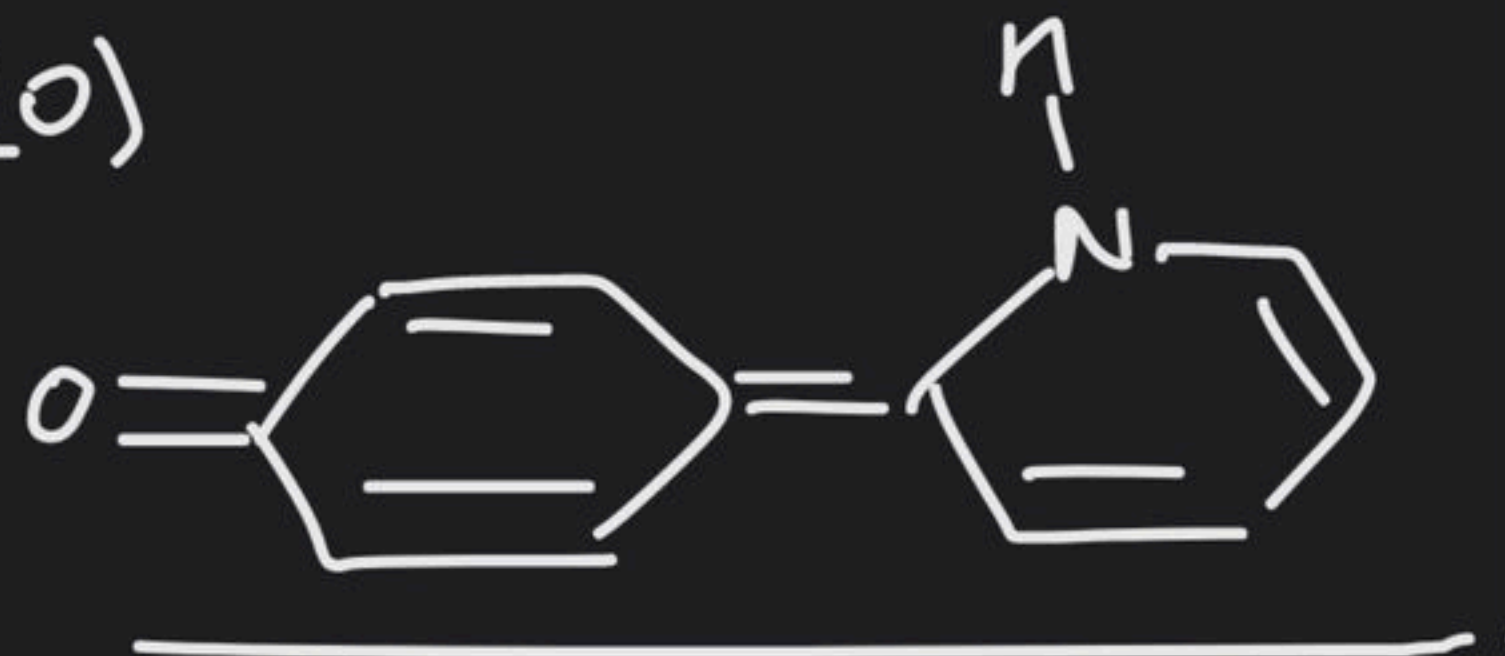




(9)



(10)



▲ 4 • Asked by Yuvrajvidh...

Sir ajj mera doubt bata dijiye ye question kese solve kare
Bulls eye ke batch test me aya tha.....

Question 60

YOU DIDN'T ATTEMPT

Your Time Taken: 1m 15s

Avg Time Taken By Others: 23s

Attempt Accuracy: 0%

Stereochemistry & Isomerism

Total number of structure isomers of molecular formula C_7H_{16} having at least one methyl substituent as per IUPAC rules :

7

