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ARJUNA (NEET)

STATES OF MATTER

DPP-09

- **1.** Which of the following increases with the increase in temperature in case of a liquid?
 - (A) Vapour pressure
 - (B) Surface tension
 - (C) Viscosity
 - (D) All of these
- **2.** The condition of free vapourisation throughout the liquid is called
 - (A) Freezing
- (B) Melting
- (C) Evaporation
- (D) Boiling
- **3.** If pressure is 1 bar, then the boiling temperature is called
 - (A) Normal boiling point
 - (B) Standard boiling point
 - (C) Abnormal boiling point
 - (D) Simple boiling point
- **4.** Liquids at higher altitude boil at (in comparison to sea level)
 - (A) Lower temperature
 - (B) Higher temperature
 - (C) Same temperature
 - (D) Cannot say
- **5.** Choose the correct statement(s):
 - (A) Boiling does not occur when liquid is heated in a closed vessel
 - (B) On heating continuously vapour pressure of a liquid increases in a closed vessel
 - (C) Liquids may be considered as a continuation of gas phase into a region of small volume and very strong molecular attractions
 - (D) All of these

- **6.** Standard boiling point of the liquid is slightly
 - (A) Lower than the normal boiling point
 - (B) Higher than the normal boiling point
 - (C) Same as the normal boiling point
 - (D) May lower or higher than the normal boiling point
- 7. Liquids tend to rise or fall in capillary because of
 - (A) Boiling
- (B) Evaporation
- (C) Surface tension
- (D) Viscosity
- **8.** Which of the following is the unit of coefficient of viscosity?
 - (A) Ns m^{-2}
- (B) Pas
- (C) $kg m^{-1} s^{-1}$
- (D) All of these
- **9.** As the temperature rises viscosity of liquids
 - (A) Increases
 - (B) Decreases
 - (C) Remains same
 - (D) Increases or decreases depending on the nature of liquid
- **10.** Which of the following is the most viscous liquid?
 - (A) Water
 - (B) Alcohol
 - (C) Glass
 - (D) Ethylene glycol

ANSWERS

- **1.** (A)
- **2.** (D)
- **3.** (B)
- **4.** (A)
- **5.** (D)
- **6.** (A)
- **7.** (C)
- **8.** (D)
- **9.** (B)
- **10.** (C)





Note - If you have any query/issue

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