ARJUNA (NEET)

STRUCTURE OF ATOM

DPP-1

- 1. Electron was discovered by
 - (A) J.J. Thomson
- (B) Rutherford
- (C) Madam Curie
- (D) E. Goldstein
- 2. Proton was discovered by
 - (A) J.J. Thomson
- (B) Rutherford
- (C) Madam Curie
- (D) E. Goldstein
- 3. Nucleus was discovered by
 - (A) J.J. Thomson
- (B) Rutherford
- (C) Madam Curie (D) E. Goldstein
- 4. What is the specific charge on e^{Θ} ?
 - (A) $1.76 \times \hat{10}^{8} \text{ c/gm}$
 - (B) $1.76 \times 10^{11} \text{ c/kg}$
 - (C) Both A & B
 - (D) $9.1 \times 10^{-31} \text{ c/kg}$
- 5. What is mass of one e^{Θ} in kg?
 - (A) $9.1 \times 10^{-31} \text{ kg}$
 - (B) $1.67 \times 10^{-27} \,\mathrm{kg}$
 - (C) $1.66 \times 10^{-27} \text{ kg}$
 - (D) $1.6 \times 10^{-19} \text{ kg}$

- 6. What is the charge of e^{Θ} ?
 - (A) $-1.6 \times 10^{-19} \,\mathrm{C}$ (B) $+1.6 \times 10^{-19} \,\mathrm{C}$
 - (C) Zero
- (D) None of these
- 7. What is the charge of 1 mole e^{Θ} ?
 - (A) $-1.6 \times 10^{-19} \,\mathrm{C}$ (B) 96500 C
 - (D) Both B & C (C) 1 Farraday
- 8. What is the mass of one proton?

 - (A) $9.1 \times 10^{-31} \text{ kg}$ (B) $1.66 \times 10^{-27} \text{ kg}$

 - (C) 96500 kg (D) $1.6 \times 10^{-19} \text{ kg}$
- 9. What is the charge on one proton?
 - (A) $+1.6 \times 10^{-19}$ C (B) Zero
 - (C) -1.6×10^{-19} C (D) 9.1×10^{-31} C
- 10. What is the charge on Neutron?
 - (A) 1.67×10^{-27} C (B) Zero
 - (C) $-1.6 \times 10^{-19} \,\mathrm{C}$ (D) $+1.6 \times 10^{-19} \,\mathrm{C}$

ANSWERS KEY

1. (A)

2. (D)

3. **(B)**

4. **(C)**

5. (A)

6. (A)

7. **(D)**

8. (B)

9. (A) 10. (B)





Note - If you have any query/issue



Mail us at support@physicswallah.org