

# Matter in our Surrounding :

**DEFINITION : Every things  
which occupy space and  
has volumn called Matter.**

**Example : Book , Apple , etc.**

**PROPERTY OF MATTER**

- Have a intermolecular Space between them.
- Have a Kinetic energy in them  $K.E = \frac{1}{2} MV^2$ .
- Have an intermolecular Force of Attration.

## **TYPES OF MATTER**

**Mainly there are three types of Matter. But two more matter is included.**

$$3 + 2 = 5 ;$$

- Solid
- Liquid
- Gas
- Plasma
- B.E.S

**SOLID : Substance which has define shape and volumn.**

**Example : Ice , Box , etc...**

**LIQUID : Substance which has define volumn but not define shape.**

**Example : Water , Juice , etc.**

**Gas : Substance which has not define shape and not define volumn.**

**Example :  $\text{H}_2\text{O}$ ,  $\text{CO}_2$ , etc...**

**RELATIONSHIP BETWEEN**  
**STATE :-**

**On the basis of space :**

- Solid have minimum space, Liquid have much more space and gaseous have maximum space.

### On the basis of movement :

- Solids have minimum movement, Liquid have much more than Solid and Gaseous have maximum movement.

### On the basis of attrection :

- Solid have maximum force

of attraction, Liquid have much low than Solid and Gaseous have minimum force of attraction.

### **LATENT HEAT**

A amount of energy which is required to change a one state to another. Only for Solid-Liquid , Liquid-Gas.

**It has mainly two parts :-**

**1.Latent heat of fusion.**

**2.Latent heat of vapourisation.**

### **LATENT HEAT OF FUSION:**

**The amount of heat energy which is required to change a solid state into liquid.The value is  $3.34 * 10^5$  kg/J.**

**LATENT HEAT OF**



## **VAPURASATION :**

**The amount of heat energy which is required to change the liquid state into gaseous state. The value is  $22.5 \times 10^5$  kg/J.**

## **TEMPRETURE**

**•Solid-Liquid = The formations of solid state into liquid state is known as**

## **Fusion (Melting).**

- **Solid-Gas = The formation of solid state into gaseous state without changing its state into liquid state is known as Sublimation. The substance which undergoes in sublimation called sublimate. The solid state which is formed by the cooling the vapour of solid called sublimate.**

- **Liquid-Gas = The formations of liquid state into gaseous state is known vapourisation (Boiling).**
- **Gas-Liquid = The formations of gas state into liquid state is known condensation (Cooling).**
- **Liquid-Solid = The formations of liquid state into solid state is known solidification (Cooling).**

- **Gas-Solid = The formation of Gaseous state into Solid state without changing its state into liquid state is known as Depositions**

## **PRESSURE**

**By changing the pressure we can change the state of a matter.**

**By apply more pressure and decreasing the tempreature we can liquify the gas.**

**Example: L.P.G , C.N.G , etc..**

**Value of atmosphere pressure is  $1_{\text{atm}} = 0.01 * 10^5 \text{ Pa}$ . In costal area the atmosphere pressure is  $1_{\text{atm}}$  which is alos known as normal atmophere pressure.**

**Evapuration**

**The heat energy which is lost by particle of matter and formed into gas is known as evaporation.**

**OR**

**The formation of liquid state into gaseous state at any temperature.**

**Factor affecting are :-**

- Temperature.**
- Surface Area.**

- Humidity.
- Wind Speed.

Temperature : Increase in temperature means increase the rate of evaporation.

Surface Area : Increase in surface area means increase the rate of evaporation.

Humidity: Increase in humidity means increase the rate of evaporation.

**Wind Speed : Increase in wind speed means increase the rate of evaporation.**

### **Diffusion**

**When a gas goes from higher concentration to lower concentration is called diffusion.**

**The rate of diffusion increase**



**in gaseous state and decrease in solid state.**

**The liquid also under goes in diffusion because it have more intermolecular space and less force of attraction than Solid.**

## **PLASMA**

**It is known as the 4<sup>th</sup> state of matter. It can be seen in**

**flucent tube and neon tube.**  
**it is a mixture of free electron**  
**and ions. It is naturally present**  
**in stars and core. It can be**  
**created in earth by passing**  
**electricity in tube at very low**  
**pressure.**

**B.E.C**

**It is known as 5<sup>th</sup> state of**  
**matter and was discovered by**

**Albert Einstein and Indian physicist Satyendra Nath Bose in 1920s . And formed by Cornell , Ketterle , and Wieman of U.S.A by cooling a gas at extremely low density and at super low temperature and in 2001 they got Nobel Prize for that discovery.**

