
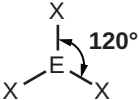
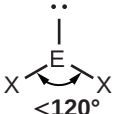
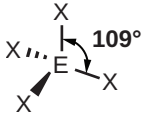
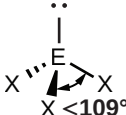
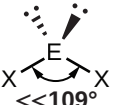
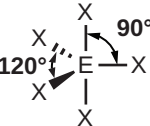
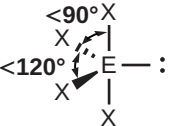
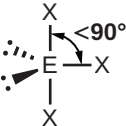
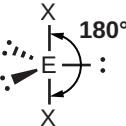

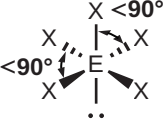
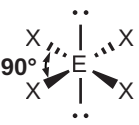

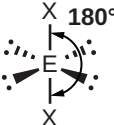


Number of electron pairs	Electron pair geometries: 0 lone pair	1 lone pair	2 lone pairs	3 lone pairs	4 lone pairs
2	 <p>Linear</p>				
3	 <p>Trigonal planar</p>	 <p>Bent or angular</p>			
4	 <p>Tetrahedral</p>	 <p>Trigonal pyramid</p>	 <p>Bent or angular</p>		
5	 <p>Trigonal bipyramid</p>	 <p>Sawhorse or seesaw</p>	 <p>T-shape</p>	 <p>Linear</p>	
6	 <p>Octahedral</p>	 <p>Square pyramid</p>	 <p>Square planar</p>	 <p>T-shape</p>	 <p>Linear</p>