AQUEOUS SOLUBILITY OF INORGANIC COMPOUNDS AT VARIOUS TEMPERATURES

The solubility of over 300 common inorganic compounds in water is tabulated here as a function of temperature. Solubility is defined as the concentration of the compound in a solution that is in equilibrium with a solid phase at the specified temperature. In this table the solid phase is generally the most stable crystalline phase at the temperature in question. An asterisk * on solubility values in adjacent columns indicates that the solid phase changes between those two temperatures (usually from one hydrated phase to another or from a hydrate to the anhydrous solid). In such cases the slope of the solubility vs. temperature curve may show a discontinuity.

All solubility values are expressed as mass percent of solute, $100 \times w_2$, where

$$w_2 = m_2/(m_1 + m_2)$$

and m_2 is the mass of solute and m_1 the mass of water. This quantity is related to other common measures of solubility as follows:

Molarity: $c_2 = 1000 \rho w_2/M_2$ Molality: $m_2 = 1000 w_2/M_2(1-w_2)$ Mole fraction: $x_2 = (w_2/M_2)/\{(w_2/M_2) + (1-w_2)/M_1\}$ Mass of solute per 100 g of H₂O: $r_2 = 100 w_2/(1-w_2)$

Here M_2 is the molar mass of the solute and M_1 = 18.015 g/mol is the molar mass of water; ρ is the density of the solution in g cm⁻³.

The data in the table have been derived from the references indicated; in many cases the data have been refitted or interpolated in order to present solubility at rounded values of temperature. Where available, values were taken from the IUPAC Solubility Data Series (Reference 1) or related papers in the

Journal of Physical and Chemical Reference Data (References 2 to 5), which present carefully evaluated data.

The solubility of sparingly soluble compounds that do not appear in this table may be calculated from the data in the table "Solubility Product Constants." Solubility of inorganic gases may be found in the table "Solubility of Selected Gases in Water."

Compounds are listed alphabetically by formula.

References

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Aqueous Solubility in Mass% as a Function of Temperature

| Formula | 0 °C | 10 °C | 20 °C | 25 °C | 30 °C | 40 °C | 50 °C | 60 °C | 70 ℃ | 80 °C | 90 °C | 100 °C | Ref. |
|-------------------|--------|--------|--------|--------|-----------------|-----------------|-------|-------|----------------|-------|-------|----------------|------|
| $AgBrO_3$ | | | | 0.193 | | | | | | | 1.32 | | 7 |
| $AgClO_2$ | 0.17 | 0.31 | 0.47 | 0.55 | 0.64 | 0.82 | 1.02 | 1.22 | 1.44 | 1.66 | 1.88 | 2.11 | 7 |
| $AgClO_3$ | | | | 15 | | | | | | | | | 7 |
| $AgClO_4$ | 81.6 | 83.0 | 84.2 | 84.8 | 85.3 | 86.3 | 86.9 | 87.5 | 87.9 | 88.3 | 88.6 | 88.8 | 6 |
| $AgNO_2$ | 0.155 | | | 0.413 | | | | | | | | | 7 |
| $AgNO_3$ | 55.9 | 62.3 | 67.8 | 70.1 | 72.3 | 76.1 | 79.2 | 81.7 | 83.8 | 85.4 | 86.7 | 87.8 | 6 |
| Ag_2SO_4 | 0.56 | 0.67 | 0.78 | 0.83 | 0.88 | 0.97 | 1.05 | 1.13 | 1.20 | 1.26 | 1.32 | 1.39 | 7 |
| $AlCl_3$ | 30.84 | 30.91 | 31.03 | 31.10 | 31.18 | 31.37 | 31.60 | 31.87 | 32.17 | 32.51 | 32.90 | 33.32 | 7 |
| $Al(ClO_4)_3$ | 54.9 | | | | | | | | | | 64.4 | | 7 |
| AlF_3 | 0.25 | 0.34 | 0.44 | 0.50 | 0.56 | 0.68 | 0.81 | 0.96 | 1.11 | 1.28 | 1.45 | 1.64 | 7 |
| $Al(NO_3)_3$ | 37.0 | 38.2 | 39.9 | 40.8 | 42.0 | 44.5 | 47.3 | 50.4 | 53.8° | | | 61.5° | 6 |
| $Al_2(SO_4)_3$ | 27.5 | | | 27.8 | 28.2 | 29.2 | 30.7 | 32.6 | 34.9 | 37.6 | 40.7 | 44.2 | 7 |
| As_2O_3 | 1.19 | 1.48 | 1.80 | 2.01 | 2.27 | 2.86 | 3.43 | 4.11 | 4.89 | 5.77 | 6.72 | 7.71 | 10 |
| $BaBr_2$ | 47.6 | 48.5 | 49.5 | 50.0 | 50.4 | 51.4 | 52.5 | 53.5 | 54.5 | 55.5 | 56.6 | 57.6 | 6 |
| $Ba(BrO_3)_2$ | 0.285 | 0.442 | 0.656 | 0.788 | 0.935 | 1.30 | 1.74 | 2.27 | 2.90 | 3.61 | 4.40 | 5.25 | 1:14 |
| $Ba(C_2H_3O_2)_2$ | 37.0 | | | 44.2 | | | | | | | | | 7 |
| $BaCl_2$ | 23.30 | 24.88 | 26.33 | 27.03 | 27.70 | 29.00 | 30.27 | 31.53 | 32.81 | 34.14 | 35.54 | 37.05 | 8 |
| $Ba(ClO_2)_2$ | 30.5 | | | 31.3 | | | | | | | | 44.7 | 7 |
| $Ba(ClO_3)_2$ | 16.90 | 21.23 | 23.66 | 27.50 | 29.43 | 33.16 | 36.69 | 40.05 | 43.04 | 45.90 | 48.70 | 51.17 | 1:14 |
| $Ba(ClO_4)_2$ | 67.30 | 70.96 | 74.30 | 75.75 | 77.05 | 79.23 | 80.92 | 82.21 | 83.16 | 83.88 | 84.43 | 84.90 | 7 |
| BaF_2 | | 0.158 | | 0.161 | | | | | | | | | 7 |
| BaI_2 | 62.5 | 64.7 | 67.3 | 68.8 | 69.1 | 69.5 | 70.1 | 70.7 | 71.3 | 72.0 | 72.7 | 73.4 | 6 |
| $Ba(IO_3)_2$ | 0.0182 | 0.0262 | 0.0342 | 0.0396 | 0.045° | 0.058° | 0.073 | 0.090 | 0.109 | 0.131 | 0.156 | 0.182 | 1:14 |

| Formula | 0 °C | 10 °C | 20 °C | 25 °C | 30 °C | 40 °C | 50 °C | 60 °C | 70 °C | 80 °C | 90 °C | 100 °C | Ref. |
|--|----------------|----------------|----------------|----------------|-----------------|----------------|---------------|----------------|----------------|----------------|---------------|---------------|--------------|
| $Ba(NO_2)_2$ | 31.1 | 36.6 | 41.8 | 44.3 | 46.8 | 51.6 | 56.2 | 60.5 | 64.6 | 68.5 | 72.1 | 75.6 | 10 |
| $Ba(NO_3)_2$ | 4.7 | 6.3 | 8.2 | 9.3 | 10.2 | 12.4 | 14.7 | 17.0 | 19.3 | 21.5 | 23.5 | 25.5 | 6 |
| $Ba(OH)_2$ | 1.67 | | | 4.68 | 8.4 | 19 | 33 | 52 | 74 | 100 | | | 7 |
| BaS | 2.79 | 4.78 | 6.97 | 8.21 | 9.58 | 12.67 | 16.18 | 20.05 | 24.19 | 28.55 | 33.04 | 37.61 | 7 |
| $BaSO_3$ | | | | 0.0011 | | | | | | | | | 1:26 |
| $Ba(SCN)_2$ | | | | 62.6 | | | | | | | | | 7 |
| $BeCl_2$ | 40.5 | | | 41.7 | | | | | | | | | 7 |
| $Be(ClO_4)_2$ | | | | 59.5 | | | | | | | | | 7 |
| $BeSO_4$ | 26.69 | 27.58 | 28.61 | 29.22 | 29.90 | 31.51 | 33.39 | 35.50 | 37.78 | 40.21 | 42.72 | 45.28 | 7 |
| CaBr ₂ | 55 | 56 | 59 | 61 | 63 | 68 | 71 | 73 | | | | | 10 |
| CaCl ₂ | 36.70 | 39.19 | 42.13 | 44.83° | 49.12° | 52.85° | 56.05° | 56.73 | 57.44 | 58.21 | 59.04 | 59.94 | 8 |
| $Ca(ClO_3)_2$ | 63.2 | 64.2 | 65.5 | 66.3 | 67.2 | 69.0 | 71.0 | 73.2 | 75.5° | 77.4° | 77.7 | 78.0 | 1:14 |
| Ca(ClO ₄) ₂ | 0.0012 | | | 65.3 | | | | | | | | | 7 |
| CaF ₂ | 0.0013 | 66.0 | 67.6 | 0.0016 | 60.0 | 70.0 | 72.4 | 74.0 | 76.0 | 70.0 | 70.6 | 01.0 | 10 7 |
| CaI ₂ Ca(IO ₃) ₂ | 64.6 0.082 | 66.0 0.155 | 67.6 0.243 | 68.3 0.305 | 69.0 0.384* | 70.8 0.517° | 72.4 0.590 | 74.0 0.652 | 76.0 0.811° | 78.0 0.665° | 79.6 0.668 | 81.0 | 1:14 |
| $Ca(1O_3)_2$ $Ca(NO_2)_2$ | 38.6 | 39.5 | 44.5 | 48.6 | 0.304 | 0.517 | 0.590 | 0.032 | 0.011 | 0.003 | 0.008 | | 7 |
| $Ca(NO_2)_2$ $Ca(NO_3)_2$ | 50.0 | 53.1 | 56.7 | 59.0 | 60.9 | 65.4 | 77.8 | 78.1 | 78.2 | 78.3 | 78.4 | 78.5 | 6 |
| CaSO ₃ | 50.1 | 55.1 | 0.0059 | 0.0054 | 0.0049 | 0.0041 | 0.0035 | 0.0030 | 0.0026 | 0.0023 | 0.0020 | 0.0019 | 1:26 |
| CaSO ₄ | 0.174 | 0.191 | 0.202 | 0.205 | 0.208 | 0.210 | 0.207 | 0.201 | 0.193 | 0.184 | 0.173 | 0.163 | 9 |
| CdBr ₂ | 36.0 | 43.0 | 49.9 | 53.4 | 56.4 | 60.3° | 60.3° | 60.5 | 60.7 | 60.9 | 61.3 | 61.6 | 6 |
| CdC_2O_4 | | | | 0.0060 | | | | | | | | | 5 |
| $CdCl_2$ | 47.2 | 50.1 | 53.2 | 54.6 | 56.3° | 57.3° | 57.5 | 57.8 | 58.1 | 58.51 | 58.98 | 59.5 | 6 |
| $Cd(ClO_4)_2$ | | | | 58.7 | | | | | | | | 66.9 | 7 |
| CdF_2 | | 5.82 | 4.65 | 4.18 | 3.76 | | | | | | | | 5 |
| CdI_2 | 44.1 | 44.9 | 45.8 | 46.3 | 46.8 | 47.9 | 49.0 | 50.2 | 51.5 | 52.7 | 54.1 | 55.4 | 6 |
| $Cd(IO_3)_2$ | | | | 0.091 | | | | | | | | | 5 |
| $Cd(NO_3)_2$ | 55.4 | 57.1 | 59.6 | 61.0 | 62.8 | 66.5 | 70.6 | 86.1 | 86.5 | 86.8 | 87.1 | 87.4 | 6 |
| CdSO ₄ | 43.1 | 43.1 | 43.2 | 43.4 | 43.6 | 44.1 | 43.5 | 42.5 | 41.4 | 40.2 | 38.5 | 36.7 | 6 |
| CdSeO ₄ | 42.04 | 40.59 | 39.02 | 38.18 | 37.29 | 35.35 | 33.15 | 30.65 | 27.84 | 24.69 | 21.24 | 17.49 | 5 |
| Ce(NO ₃) ₃ CoCl ₂ | 57.99 30.30 | 59.80 32.60 | 61.89 34.87 | 63.05 35.99 | 64.31° 37.10 | 67.0° 39.27 | 68.6 41.38 | 71.1° 43.46 | 74.9° 45.50 | 79.2 47.51 | 80.9 49.51 | 83.1 51.50 | 1:13 7 |
| $Co(ClO_4)_2$ | 50.50 | 32.00 | 34.07 | 53.99 | 37.10 | 39.27 | 41.50 | 43.40 | 43.30 | 47.31 | 47.31 | 31.30 | 7 |
| CoF ₂ | 50.0 | | | 1.4 | | | | | | | | | 7 |
| CoI ₂ | 58.00 | 61.78 | 65.35 | 66.99 | 68.51 | 71.17 | 73.41 | 75.29 | 76.89 | 78.28 | 79.52 | 80.70 | 7 |
| $Co(NO_2)_2$ | 0.076 | | | 0.49 | | | | | | | | | 7 |
| $Co(NO_3)_2$ | 45.5 | 47.0 | 49.4 | 50.8 | 52.4 | 56.0 | 60.1 | 62.6 | 64.9 | 67.7 | | | 6 |
| $CoSO_4$ | 19.9 | 23.0 | 26.1 | 27.7 | 29.2 | 32.3 | 34.4 | 35.9 | 35.5 | 33.2 | 30.6 | 27.8 | 6 |
| $Co(SCN)_2$ | | | | 50.7 | | | | | | | | | 7 |
| CrO_3 | 62.2 | 62.3 | 62.6 | 62.8 | 63.0 | 63.5 | 64.1 | 64.7 | 65.5 | 66.2 | 67.1 | 67.9 | 6 |
| CsBr | | | | 55.2 | | | | | | | | | 7 |
| CsBrO ₃ | 1.16 | 1.93 | 3.01 | 3.69 | 4.46 | 6.32 | 8.60 | 11.32 | 14.45 | 17.96 | 21.83 | 25.98 | 1:30 |
| CsCl | 61.83 | 63.48 | 64.96 | 65.64 | 66.29 | 67.50 | 68.60 | 69.61 | 70.54 | 71.40 | 72.21 | 72.96 | 1:47 |
| CsClO ₃ | 2.40 | 3.87 | 5.94 | 7.22 | 8.69 | 12.15 | 16.33 | 21.14 | 26.45 | 32.10 | 37.89 | 43.42 | 1:30 |
| CsClO ₄ CsI | 0.79 30.9 | 1.01 37.2 | 1.51 43.2 | 1.96 45.9 | 2.57 48.6 | 4.28 53.3 | 6.55 57.3 | 9.29 60.7 | 12.41 63.6 | 15.80 65.9 | 19.39 67.7 | 23.07 69.2 | 7 6 |
| CsIO ₃ | 1.08 | 1.58 | 2.21 | 2.59 | 3.02 | 3.96 | 5.06 | 6.29 | 7.70 | 9.20 | 10.79 | 12.45 | 1:30 |
| CsNO ₃ | 8.46 | 13.0 | 18.6 | 21.8 | 25.1 | 32.0 | 39.0 | 45.7 | 51.9 | 57.3 | 62.1 | 66.2 | 6 |
| CsOH | | | | | 75 | | | | | | | | 7 |
| Cs_2SO_4 | 62.6 | 63.4 | 64.1 | 64.5 | 64.8 | 65.5 | 66.1 | 66.7 | 67.3 | 67.8 | 68.3 | 68.8 | 6 |
| $CuBr_2$ | | | | 55.8 | | | | | | | | | 7 |
| $CuCl_2$ | 40.8 | 41.7 | 42.6 | 43.1 | 43.7 | 44.8 | 46.0 | 47.2 | 48.5 | 49.9 | 51.3 | 52.7 | 6 |
| $Cu(ClO_4)_2$ | 54.3 | | | | 59.3 | | | | | | | | 7 |
| CuF_2 | | | | 0.075 | | | | | | | | | 7 |
| $Cu(NO_3)_2$ | 45.2 | 49.8 | 56.3 | 59.2 | 61.1 | 62.0 | 63.1 | 64.5 | 65.9 | 67.5 | 69.2 | 71.0 | 6 |
| CuSO ₄ | 12.4 | 14.4 | 16.7 | 18.0 | 19.3 | 22.2 | 25.4 | 28.8 | 32.4 | 36.3 | 40.3 | 43.5 | 6 |
| CuSeO ₄ | 10.6 | E0.00 | 61.40 | 16.0 | 62.20 | 65.40 | 60.04 | 71.50 | | | | | 7 |
| $Dy(NO_3)_3$ | 58.79 | 59.99 | 61.49 | 62.35 | 63.29 | 65.43 | 68.04 | 71.58 | 77 75 | | | | 1:13 |
| $Er(NO_3)_3$ $Eu(NO_3)_3$ | 61.58 55.2 | 63.15 56.7 | 64.84 58.5 | 65.75 59.4 | 66.69 60.4 | 68.70 62.5 | 70.96 64.6 | 73.64 | 77.75 | | | | 1:13 1:13 |
| FeBr ₂ | 33.4 | 50.7 | 50.5 | 54.6° | 00.4 | 02.3 | 0.7.0 | | | | | 64.8° | 7 |
| 20012 | | | | 51.0 | | | | | | | | 01.0 | , |

| Formula | 0 °C | 10 °C | 20 °C | 25 °C | 30 °C | 40 °C | 50 °C | 60 °C | 70 ℃ | 80 °C | 90 °C | 100 °C | Ref. |
|---|----------------|--------------|--------------|----------------|--------------|--------------|--------------|----------------|--------------|--------------|--------------|--------------|------------|
| FeCl ₂ Fe(ClO ₄) ₂ | 33.2° 63.39 | | | 39.4° 67.76 | | | | | | | | 48.7° | 7 7 |
| FeCl ₃ | 42.7 | 44.9 | 47.9 | 47.7 | 51.6 | 74.8 | 76.7 | 84.6 | 84.3 | 84.3 | 84.4 | 84.7 | 6 |
| FeF ₃ | 1217 | 11.7 | 2,1,5 | 5.59 | 01.0 | , 110 | , 01, | 0 1.0 | 0 1.0 | 01.0 | 01/1 | 0111 | 7 |
| Fe(NO ₃) ₂ | 41.44 | | | 46.67 | | | | | | | | | 7 |
| $Fe(NO_3)_3$ | 40.15 | | | 46.57 | | | | | | | | | 7 |
| $FeSO_4$ | 13.5 | 17.0 | 20.8 | 22.8 | 24.8 | 28.8 | 32.8 | 35.5 | 33.6 | 30.4 | 27.1 | 24.0 | 6 |
| $Gd(NO_3)_3$ | 56.3 | 57.7 | 59.2 | 60.1 | 61.0 | 62.9 | 65.2 | 67.9 | 71.5 | | | | 1:13 |
| HIO_3 | 73.45 | 74.10 | 74.98 | 75.48 | 76.03 | 77.20 | 78.46 | 79.78 | 81.13 | 82.48 | 83.82 | 85.14 | 1:30 |
| H_3BO_3 | 2.61 | 3.57 | 4.77 | 5.48 | 6.27 | 8.10 | 10.3 | 12.9 | 15.9 | 19.3 | 23.1 | 27.3 | 6 |
| $HgBr_2$ | 0.26 | 0.37 | 0.52 | 0.61 | 0.72 | 0.96 | 1.26 | 1.63 | 2.08 | 2.61 | 3.23 | 3.95 | 4 |
| $Hg(CN)_2$ | 6.57 | 7.83 | 9.33 | 10.2 | 11.1 | 13.1 | 15.5 | 18.2 | 21.2 | 24.6 | 28.3 | 32.3 | 6 |
| $HgCl_2$ | 4.24 | 5.05 | 6.17 | 6.81 | 7.62 | 9.53 | 12.02 | 15.18 | 19.16 | 24.06 | 29.90 | 36.62 | 4 |
| HgI_2 | | | 0.0041 | 0.0055 | 0.0072 | 0.0122 | 0.0199 | | | | | | 4 |
| Hg(SCN) ₂ | | | | 0.070 | | | | | | | | | 4 |
| Hg ₂ Cl ₂ | 50. 0 | | | 0.0004 | | | | | | | | 05.00 | 3 |
| $Hg_2(ClO_4)_2$ | 73.8 | 0.042 | 0.040 | 79.8* | 0.054 | 0.050 | 0.065 | 0.070 | 0.076 | 0.000 | 0.000 | 85.3* | 7 |
| Hg ₂ SO ₄ | 0.038 | 0.043 | 0.048 | 0.051 | 0.054 | 0.059 | 0.065 | 0.070 | 0.076 | 0.082 | 0.088 | 0.093 | 4 |
| Ho(NO ₃) ₃ KBF ₄ | 0.28 | 0.34 | 0.45 | 63.8 0.55 | 0.75 | 1.38 | 2.09 | 2.82 | 3.58 | 4.34 | 5.12 | 5.90 | 1:13 10 |
| KBr | 35.0 | 37.3 | 39.4 | 40.4 | 41.4 | 43.2 | 44.8 | 46.2 | 47.6 | 48.8 | 49.8 | 50.8 | 6 |
| KBrO ₃ | 2.97 | 4.48 | 6.42 | 7.55 | 8.79 | 11.57 | 14.71 | 18.14 | 21.79 | 25.57 | 29.42 | 33.28 | 1:30 |
| $KC_2H_3O_2$ | 68.40 | 70.29 | 72.09 | 72.92 | 73.70 | 75.08 | 76.27 | 77.31 | 78.22 | 79.04 | 79.80 | 80.55 | 7 |
| KCl | 21.74 | 23.61 | 25.39 | 26.22 | 27.04 | 28.59 | 30.04 | 31.40 | 32.66 | 33.86 | 34.99 | 36.05 | 1:47 |
| KClO ₃ | 3.03 | 4.67 | 6.74 | 7.93 | 9.21 | 12.06 | 15.26 | 18.78 | 22.65 | 26.88 | 31.53 | 36.65 | 1:30 |
| KClO ₄ | 0.70 | 1.10 | 1.67 | 2.04 | 2.47 | 3.54 | 4.94 | 6.74 | 8.99 | 11.71 | 14.94 | 18.67 | 6 |
| KF | 30.90 | 39.8 | 47.3 | 50.41 | 53.2 | | | | | 60.0 | | | 7 |
| KHCO ₃ | 18.62 | 21.73 | 24.92 | 26.6 | 28.13 | 31.32 | 34.46 | 37.51 | 40.45 | | | | 6 |
| $KHSO_4$ | 27.1 | 29.7 | 32.3 | 33.6 | 35.0 | 37.8 | 40.5 | 43.4 | 46.2 | 49.02 | 51.82 | 54.6 | 6 |
| KH_2PO_4 | 11.74 | 14.91 | 18.25 | 19.97 | 21.77 | 25.28 | 28.95 | 32.76 | 36.75 | 40.96 | 45.41 | 50.12 | 1:31 |
| KI | 56.0 | 57.6 | 59.0 | 59.7 | 60.4 | 61.6 | 62.8 | 63.8 | 64.8 | 65.7 | 66.6 | 67.4 | 6 |
| KIO ₃ | 4.53 | 5.96 | 7.57 | 8.44 | 9.34 | 11.09 | 13.22 | 15.29 | 17.41 | 19.58 | 21.78 | 24.03 | 1:30 |
| KIO ₄ | 0.16 | 0.22 | 0.37 | 0.51 | 0.70 | 1.24 | 1.96 | 2.83 | 3.82 | 4.89 | 6.02 | 7.17 | 7 |
| KMnO ₄ | 2.74 | 4.12 | 5.96 | 7.06 | 8.28 | 11.11 | 14.42 | 18.16 | 70.5 | TO 1 | 70.6 | 00.1 | 6 |
| KNO ₂ | 73.7 | 74.6 17.6 | 75.3 24.2 | 75.7 27.7 | 76.0 31.3 | 76.7 | 77.4 45.7 | 78.0 52.2 | 78.5 58.0 | 79.1 63.0 | 79.6 67.3 | 80.1 70.8 | 6 |
| KNO ₃ KOH | 12.0 48.7 | 50.8 | 53.2 | 54.7 | 56.1 | 38.6 57.9 | 58.6 | 59.5 | 60.6 | 61.8 | 63.1 | 64.6 | 6 |
| KSCN | 63.8 | 66.4 | 69.1 | 70.4 | 71.6 | 74.1 | 76.5 | 78.9 | 81.1 | 83.3 | 85.3 | 87.3 | 6 |
| K ₂ CO ₃ | 51.3 | 51.7 | 52.3 | 52.7 | 53.1 | 54.0 | 54.9 | 56.0 | 57.2 | 58.4 | 59.6 | 61.0 | 6 |
| K ₂ CrO ₄ | 37.1 | 38.1 | 38.9 | 39.4 | 39.8 | 40.5 | 41.3 | 41.9 | 42.6 | 43.2 | 43.8 | 44.3 | 6 |
| $K_2Cr_2O_7$ | 4.30 | 7.12 | 10.9 | 13.1 | 15.5 | 20.8 | 26.3 | 31.7 | 36.9 | 41.5 | 45.5 | 48.9 | 6 |
| K ₂ HAsO ₄ | 48.5° | | | 63.6° | | | | | | | | 79.8° | 7 |
| K ₂ HPO ₄ | 57.0 | 59.1 | 61.5 | 62.7 | 64.1 | 67.7° | | 72.7° | | | | | 1:31 |
| K_2MoO_4 | | | | 64.7 | | | | | | | 66.5 | | 7 |
| K_2SO_3 | 51.30 | 51.39 | 51.49 | 51.55 | 51.62 | 51.76 | 51.93 | 52.11 | 52.32 | 52.54 | 52.79 | 53.06 | 1:26 |
| K_2SO_4 | 7.11 | 8.46 | 9.95 | 10.7 | 11.4 | 12.9 | 14.2 | 15.5 | 16.7 | 17.7 | 18.6 | 19.3 | 6 |
| $K_2S_2O_3$ | 49.0° | | | 62.3° | | | | | | | 75.7° | | 7 |
| $K_2S_2O_5$ | 22.1 | 26.7 | 31.1 | 33.1 | 35.2 | 39.0 | 42.6 | 46.0 | 49.1 | 52.0 | 54.6 | | 1:26 |
| K_2SeO_3 | 68.4° | | | 68.5° | | | | | | | | 68.5° | 7 |
| K ₂ SeO ₄ | 52.70 | 52.93 | 53.17 | 53.30 | 53.43 | 53.70 | 53.99 | 54.30 | 54.61 | 54.94 | 55.26 | 55.60 | 7 |
| K ₃ AsO ₄ | 51.5° | 07.6 | 01.1 | 55.6° | 24.2 | 27.0 | 20.6 | 41.7 | 40.5 | 45.0 | 46.1 | 73° | 7 |
| K ₃ Fe(CN) ₆ | 23.9 | 27.6 | 31.1 | 32.8 | 34.3 | 37.2 | 39.6 | 41.7 | 43.5 | 45.0 | 46.1 | 47.0 | 6 7 |
| K_3PO_4 $K_4Fe(CN)_6$ | 44.3 12.5 | 17.3 | 22.0 | 51.4 23.9 | 25.6 | 29.2 | 32.5 | 35.5 | 38.2 | 40.6 | 41.4 | 43.1 | 6 |
| LaCl ₃ | 49.0 | 48.5 | 48.6 | 48.9 | 49.3 | 50.5 | 52.1 | 54.0 | 56.3 | 58.9 | 61.7 | 43.1 | 6 |
| $LaCi_3$ $La(NO_3)_3$ | 55.0 | 56.9 | 58.9 | 60.0 | 61.1 | 63.6 | 66.3 | 69.9° | 74.1° | 50.7 | 01./ | | 1:13 |
| LiBr | 58.4 | 60.1 | 62.7 | 64.4 | 65.9 | 67.8 | 68.3 | 69.0 | 69.8 | 70.7 | 71.7 | 72.8 | 6 |
| LiBrO ₃ | 61.03 | 62.62 | 64.44 | 65.44 | 66.51 | 68.90 | 71.68° | 73.24° | 74.43 | 75.66 | 76.93 | 78.32 | 1:30 |
| LiD ₁ O ₃ LiC ₂ H ₃ O ₂ | 23.76 | 26.49 | 29.42 | 31.02 | 32.72 | 36.48 | 40.65 | 45.15 | 49.93 | 54.91 | 60.04 | 65.26 | 7 |
| LiCl | 40.45 | 42.46° | 45.29° | 45.81 | 46.25 | 47.30 | 48.47 | 49.78 | 51.27 | 52.98 | 54.98° | 56.34° | 1:47 |
| $LiClO_3$ | 73.2 | 75.6° | 80.8° | 82.1 | 83.4 | 85.9° | 87.1° | 88.2 | 89.6 | 91.3 | 93.4 | 95.7 | 1:30 |
| | | | | | | | | | | | | | |

| Formula | 0 °C | 10 °C | 20 °C | 25 °C | 30 °C | 40 °C | 50 °C | 60 °C | 70 °C | 80 °C | 90 °C | 100 °C | Ref. |
|---|----------------|---------------|--------------|----------------|--------------|---------------|--------------|--------------|--------------|--------------|--------------|--------|-----------|
| $LiClO_4$ | 30.1 | 32.6 | 35.5 | 37.0 | 38.6 | 41.9 | 45.5 | 49.2 | 53.2 | 57.2 | 61.3 | 71.4 | 6 |
| LiF | 0.120 | 0.126 | 0.131 | 0.134 | | | | | | | | | 7 |
| LiH_2PO_4 | 55.8 | | | | | | | | | | | | 7 |
| LiI | 59.4 | 60.5 | 61.7 | 62.3 | 63.0 | 64.3 | 65.8 | 67.3 | 68.8 | 81.3 | 81.7 | 82.6 | 6 |
| LiIO ₃ | | | | 43.8 | | | | | | | | | 1:30 |
| LiNO ₂ | 41 | 45 | 49 | 51 | 53 | 56 | 60 | 63 | 66 | 68 | 60.5 | 60.7 | 10 |
| LiNO ₃ | 34.8 | 37.6 | 42.7 | 50.5 | 57.9 | 60.1 | 62.2 | 64.0 | 65.7 | 67.2 | 68.5 | 69.7 | 6 |
| LiOH LiSCN | 10.8 | 10.8 | 11.0 | 11.1 54.5 | 11.3 | 11.7 | 12.2 | 12.7 | 13.4 | 14.2 | 15.1 | 16.1 | 6 7 |
| Li ₂ CO ₃ | 1.54 | 1.43 | 1.33 | 1.28 | 1.24 | 1.15 | 1.07 | 0.99 | 0.92 | 0.85 | 0.78 | 0.72 | 7 |
| Li_2CO_3 $\text{Li}_2\text{C}_2\text{O}_4$ | 1.54 | 1.43 | 1.55 | 5.87 | 1.24 | 1.13 | 1.07 | 0.77 | 0.72 | 0.03 | 0.76 | 0.72 | 7 |
| Li ₂ HPO ₃ | 9.07 | 8.40 | 7.77 | 7.47 | 7.18 | 6.64 | 6.16 | 5.71 | 5.30 | 4.91 | 4.53 | 4.16 | 7 |
| Li ₂ SO ₄ | 26.3 | 25.9 | 25.6 | 25.5 | 25.3 | 25.0 | 24.8 | 24.5 | 24.3 | 24.0 | 23.8 | 23.6 | 6 |
| Li ₃ PO ₄ | | | | 0.027 | | | | | | | | | 1:31 |
| $Lu(NO_3)_3$ | | | | 71.1 | | | | | | | | | 1:13 |
| MgBr_2 | 49.3 | 49.8 | 50.3 | 50.6 | 50.9 | 51.5 | 52.1 | 52.8 | 53.5 | 54.2 | 55.0 | 55.7 | 6 |
| $Mg(BrO_3)_2$ | 43.0 | 45.2 | 48.0 | 49.4 | 51.0 | 54.3 | 57.9 | 61.6 | 65.3 | 69.0° | 70.9° | 71.7 | 1:14 |
| MgC_2O_4 | | | | 0.038 | | | | | | | | | 7 |
| $Mg(C_2H_3O_2)_2$ | 36.18 | 37.55 | 38.92 | 39.61 | | | | | | | | | 7 |
| $MgCl_2$ | 33.96 | 34.85 | 35.58 | 35.90 | 36.20 | 36.77 | 37.34 | 37.97 | 38.71 | 39.62 | 40.75 | 42.15 | 8 |
| $Mg(ClO_3)_2$ | 53.35 | 54.40 | 56.81 | 58.66 | 60.91° | 65.46° | 67.33 | 69.27 | 71.01 | 72.44 | 73.48 | | 1:14 |
| Mg(ClO ₄) ₂ | 47.8 | 48.7 | 49.6 | 50.1 | 50.5 | 51.3 | 52.1 | | | | | | 6 |
| MgCrO ₄ | 32.06° | | | 35.39° | | | | | | 6 0 | | | 7 |
| MgCr ₂ O ₇ | | | | 58.9 | | | | | | 67.0 | | | 7 |
| MgF ₂ | E 4 7 | E 6 1 | E 0 2 | 0.013 | 60.9 | 62.0 | 65.0 | 6E 0 | 6E 0 | 6E 0 | 6E 1 | 6E 2 | 7 6 |
| MgI_2 $Mg(IO_3)_2$ | 54.7 3.19° | 56.1 6.70° | 58.2 7.92 | 59.4 8.52 | 60.8 9.11 | 63.9 10.45 | 11.99 | 65.0 13.7 | 65.0 15.6 | 65.0 17.6 | 65.1 19.6 | 65.2 | 0 1:14 |
| $Mg(NO_3)_2$ $Mg(NO_2)_2$ | 3.19 | 0.70 | 1.92 | 47 | 9.11 | 10.43 | 11.99 | 13.7 | 13.0 | 17.0 | 19.0 | | 7 |
| $Mg(NO_2)_2$ $Mg(NO_3)_2$ | 38.4 | 39.5 | 40.8 | 41.6 | 42.4 | 44.1 | 45.9 | 47.9 | 50.0 | 52.2 | 70.6 | 72.0 | 6 |
| MgSO ₃ | 0.32 | 0.37 | 0.46 | 0.52 | 0.61 | 0.87° | 0.85° | 0.76 | 0.69 | 0.64 | 0.62 | 0.60 | 1:26 |
| MgSO ₄ | 18.2 | 21.7 | 25.1 | 26.3 | 28.2 | 30.9 | 33.4 | 35.6 | 36.9 | 35.9 | 34.7 | 33.3 | 6 |
| MgS_2O_3 | 30.7 | | | 34.1 | | | | | | | | | 7 |
| $MgSeO_4$ | 31.4° | | | 35.7° | | | | | | | | 47° | 7 |
| $MnBr_2$ | 56.00 | 57.72 | 59.39 | 60.19 | 60.96 | 62.41 | 63.75 | 65.01 | 66.19 | 67.32 | 68.42 | 69.50 | 7 |
| $MnCl_2$ | 38.7 | 40.6 | 42.5 | 43.6 | 44.7 | 47.0 | 49.4 | 54.1 | 54.7 | 55.2 | 55.7 | 56.1 | 6 |
| MnF_2 | 0.80° | | | 1.01° | | | | | | | | 0.48 | 7 |
| $Mn(IO_3)_2$ | | | | 0.27 | | | | | | | 0.34 | | 7 |
| $Mn(NO_3)_2$ | 50.5 | | | 61.7 | | | | | | | | | 7 |
| $MnSO_4$ | 34.6 | 37.3 | 38.6 | 38.9 | 38.9 | 37.7 | 36.3 | 34.6 | 32.8 | 30.8 | 28.8 | 26.7 | 6 |
| NH ₄ Br | 37.5 | 40.2 | 42.7 | 43.9 | 45.1 | 47.3 | 49.4 | 51.3 | 53.0 | 54.6 | 56.1 | 57.4 | 7 |
| NH ₄ Cl | 22.92 | 25.12 | 27.27 | 28.34 | 29.39 | 31.46 | 33.50 | 35.49 | 37.46 | 39.40 | 41.33 | 43.24 | 1:47 |
| NH ₄ ClO ₄ | 10.8 | 14.1 | 17.8 | 19.7 | 21.7 | 25.8 | 29.8 | 33.6 | 37.3 | 40.7 | 43.8 | 46.6 | 6 |
| NH₄F NH₄HCO₃ | 41.7 10.6 | 43.2 | 44.7 17.6 | 45.5 19.9 | 46.3 22.4 | 47.8 27.9 | 49.3 34.2 | 50.9 41.4 | 52.5 49.3 | 54.1 58.1 | 67.6 | 78.0 | 7 7 |
| NH ₄ HCO ₃ NH ₄ H ₂ AsO ₄ | 25.2 | 13.7 29.0 | 32.7 | 34.5 | 36.3 | 39.7 | 43.1 | 46.2 | 49.3 | 52.2 | 55.0 | 76.0 | 7 |
| $NH_4H_2PO_4$ $NH_4H_2PO_4$ | 17.8 | 22.0 | 26.4 | 28.8 | 31.2 | 36.2 | 41.6 | 47.2 | 53.0 | 59.2 | 65.7 | 72.4 | 7 |
| NH ₄ I | 60.7 | 62.1 | 63.4 | 64.0 | 64.6 | 65.8 | 66.8 | 67.8 | 68.7 | 69.6 | 70.4 | 71.1 | 6 |
| NH ₄ IO ₃ | 0011 | 02.1 | 00.1 | 3.70 | 4.20 | 5.64 | 7.63 | 0.10 | 0011 | 07.0 | , 0,1 | , 1,1 | 1:30 |
| NH_4NO_2 | 55.7 | 59.0 | 64.9 | 68.8 | | | | | | | | | 7 |
| NH_4NO_3 | 54.0 | 60.1 | 65.5 | 68.0 | 70.3 | 74.3 | 77.7 | 80.8 | 83.4 | 85.8 | 88.2 | 90.3 | 6 |
| NH ₄ SCN | | | | 64.4 | | | | | 81.1 | | | | 7 |
| $(NH_4)_2C_2O_4$ | 2.31 | 3.11 | 4.25 | 4.94 | 5.73 | 7.56 | 9.73 | 12.2 | 15.1 | 18.3 | 21.8 | 25.7 | 7 |
| $(NH_4)_2HPO_4$ | 36.4 | 38.2 | 40.0 | 41.0 | 42.0 | 44.1 | 46.2 | 48.5 | 50.9 | 53.3 | 55.9 | 58.6 | 7 |
| $(NH_4)_2SO_3$ | 32.2 | 34.9 | 37.7 | 39.1 | 40.6 | 43.7 | 47.0 | 50.6 | 54.5 | 58.9 | | | 1:26 |
| $(NH_4)_2SO_4$ | 41.3 | 42.1 | 42.9 | 43.3 | 43.8 | 44.7 | 45.6 | 46.6 | 47.5 | 48.5 | 49.5 | 50.5 | 6 |
| $(NH_4)_2S_2O_5$ | 65.5 | 67.9 | 69.8 | 70.5 | 71.3 | 72.3 | 72.9 | 73.1 | | | | | 1:26 |
| $(NH_4)_2S_2O_8$ | 37.00 | 40.45 | 43.84 | 45.49 | 47.11 | 50.25 | 53.28 | 56.23 | 59.13 | 62.00 | | | 7 |
| $(NH_4)_2SeO_3$ | 49.0 | 51.1 | 53.4 | 54.7 | 56.0 | 58.9 | 62.0 | 65.4 | 69.1 | | | | 7 |
| $(NH_4)_2SeO_4$ | | | | 54.02 | | | | | | | | | 7 |
| $(NH_4)_3PO_4$ | | | .== | 15.5 | 40 - | | 50 - | | | | | | 7 |
| NaBr | 44.4 | 45.9 | 47.7 | 48.6 | 49.6 | 51.6 | 53.7 | 54.1 | 54.3 | 54.5 | 54.7 | 54.9 | 6 |

| Formula | 0°C | 10 °C | 20 °C | 25 °C | 30 °C | 40 °C | 50 °C | 60 °C | 70 °C | 80 °C | 90 °C | 100 °C | Ref. |
|--|---------------|---------------|---------------|----------------|---------------|----------------|---------------|---------------|---------------|--------------|--------------|-------------|-----------|
| NaBrO ₃ | 20.0 | 23.22 | 26.65 | 28.28 | 29.86 | 32.83 | 35.55 | 38.05 | 40.37 | 42.52 | <i>7</i> 0 C | 100 C | 1:30 |
| NaCHO ₂ | 30.8 | 37.9 | 45.7 | 48.7 | 50.6 | 52.0 | 53.5 | 55.0 | | | | | 6 |
| $NaC_2H_3O_2$ | 26.5 | 28.8 | 31.8 | 33.5 | 35.5 | 39.9 | 45.1 | 58.3 | 59.3 | 60.5 | 61.7 | 62.9 | 6 |
| NaCl | 26.28 | 26.32 | 26.41 | 26.45 | 26.52 | 26.67 | 26.84 | 27.03 | 27.25 | 27.50 | 27.78 | 28.05 | 1:47 |
| NaClO | 22.7 | | | 44.4 | | | | | | | | | 7 |
| NaClO ₂ | | | | 97.0° | | | | 95.3° | | | | | 7 |
| NaClO ₃ | 44.27 | 46.67 | 49.3 | 50.1 | 51.2 | 53.6 | 55.5 | 57.0 | 58.5 | 60.5 | 63.3 | 67.1 | 1:30 |
| NaClO ₄ | 61.9 | 64.1 | 66.2 | 67.2 | 68.3 | 70.4 | 72.5 | 74.1 | 74.7 | 75.4 | 76.1 | 76.7 | 6 |
| NaF | 3.52 | 3.72 | 3.89 | 3.97 | 4.05 | 4.20 | 4.34 | 4.46 | 4.57 | 4.66 | 4.75 | 4.82 | 6 |
| NaHCO ₃ | 6.48 | 7.59 | 8.73 | 9.32 | 9.91 | 11.13 | 12.40 | 13.70 | 15.02 | 16.37 | 17.73 | 19.10 | 7 |
| NaHSO ₄ | 26.54 | 41.07 | 46.00 | 22.2 | 51.54 | 57 00° | C1 17° | 60.0° | <i>(</i> 5.0) | 60.7 | | 33.3 | 10 |
| NaH ₂ PO ₄ NaI | 36.54 61.2 | 41.07 62.4 | 46.00 63.9 | 48.68 64.8 | 51.54 65.7 | 57.89° 67.7 | 61.7* 69.8 | 62.3° 72.0 | 65.9 74.7 | 68.7 74.8 | 74.9 | 75.1 | 1:31 6 |
| NaIO ₃ | 2.43 | 4.40 | 7.78° | 8.65° | 9.60 | 11.67 | 13.99 | 16.52 | 19.25° | 21.1° | 22.9 | 24.7 | 1:30 |
| NaIO ₄ | 2.43 | 7.70 | 7.70 | 12.62 | 2.00 | 11.07 | 13.77 | 10.52 | 17.23 | 21.1 | 22.7 | 27.7 | 7 |
| NaNO ₂ | 41.9 | 43.4 | 45.1 | 45.9 | 46.8 | 48.7 | 50.7 | 52.8 | 55.0 | 57.2 | 59.5 | 61.8 | 6 |
| NaNO ₃ | 42.2 | 44.4 | 46.6 | 47.7 | 48.8 | 51.0 | 53.2 | 55.3 | 57.5 | 59.6 | 61.7 | 63.8 | 6 |
| NaOH | 30 | 39 | 46 | 50 | 53 | 58 | 63 | 67 | 71 | 74 | 76 | 79 | 10 |
| NaSCN | | 52.9 | 57.1 | 60.2 | 62.7 | 63.5 | 64.2 | 65.0 | 65.9 | 66.9 | 67.9 | 69.0 | 6 |
| $Na_2B_4O_7$ | 1.23 | 1.71 | 2.50 | 3.07 | 3.82 | 6.02 | 9.7 | 14.9 | 17.1 | 19.9 | 23.5 | 28.0 | 6 |
| Na ₂ CO ₃ | 6.44 | 10.8 | 17.9 | 23.5 | 28.7 | 32.8 | 32.2 | 31.7 | 31.3 | 31.1 | 30.9 | 30.9 | 6 |
| $Na_2C_2O_4$ | 2.62 | 2.95 | 3.30 | 3.48 | 3.65 | 4.00 | 4.36 | 4.71 | 5.06 | 5.41 | 5.75 | 6.08 | 6 |
| Na_2CrO_4 | 22.6 | 32.3 | 44.6 | 46.7 | 46.9 | 48.9 | 51.0 | 53.4 | 55.3 | 55.5 | 55.8 | 56.1 | 6 |
| $Na_2Cr_2O_7$ | 62.1 | 63.1 | 64.4 | 65.2 | 66.1 | 68.0 | 70.1 | 72.3 | 74.6 | 77.0 | 79.6 | 80.7 | 6 |
| Na_2HAsO_4 | 5.6° | | | 29.3° | | | | | | | | 67* | 7 |
| Na_2HPO_4 | 1.66 | 4.19 | 7.51 | 10.55 | 16.34° | 35.17° | 44.64° | 45.20 | 46.81 | 48.78 | 50.52 | 51.53 | 1:31 |
| Na_2MoO_4 | 30.6 | 38.8 | 39.4 | 39.4 | 39.8 | 40.3 | 41.0 | 41.7 | 42.6 | 43.5 | 44.5 | 45.5 | 6 |
| Na ₂ S | 11.1 | 13.2 | 15.7 | 17.1 | 18.6 | 22.1 | 26.7 | 28.1 | 30.2 | 33.0 | 36.4 | 41.0 | 6 |
| Na ₂ SO ₃ | 12.0 | 16.1 | 20.9 | 23.5 | 26.3* | 27.3* | 25.9 | 24.8 | 23.7 | 22.8 | 22.1 | 21.5 | 1:26 |
| Na ₂ SO ₄ | 22.1 | 26.2 | 16.13 | 21.94 | 29.22° | 32.35° | 31.55 | 30.90 | 30.39 | 30.02 | 29.79 | 29.67 | 8 |
| Na ₂ S ₂ O ₃ | 33.1 | 36.3 38.4 | 40.6 | 43.3 40.0 | 45.9 40.6 | 52.0 41.8 | 62.3 43.0 | 65.7 | 68.8 45.5 | 69.4 | 70.1 | 71.0 | 6 1:26 |
| Na ₂ S ₂ O ₅ | | 38.4 | 39.5 | 40.0 47.3° | 40.6 | 41.8 | 45.0 | 44.2 | 45.5 | 46.8 | 48.1 | 49.5 45° | 7 |
| Na ₂ SeO ₃ Na ₂ SeO ₄ | 11.7 | | | 36.9° | | | | | | | | 42.1° | 7 |
| Na_2SeO_4 Na_2WO_4 | 41.6 | 41.9 | 42.3 | 42.6 | 42.9 | 43.6 | 44.4 | 45.3 | 46.2 | 47.3 | 48.4 | 49.5 | 6 |
| Na ₃ PO ₄ | 4.28 | 7.30 | 10.8 | 12.6 | 14.1 | 16.6 | 22.9 | 28.4 | 32.4 | 37.6 | 40.4 | 43.5 | 6 |
| $Na_4P_2O_7$ | 2.23 | 3.28 | 4.81 | 6.62 | 7.00 | 10.10 | 14.38 | 20.07 | 27.31 | 36.03 | 32.37 | 30.67 | 6 |
| NdCl ₃ | 49.0 | 49.3 | 49.7 | 50.0 | 50.4 | 51.2 | 52.2 | 53.3 | 54.5 | 55.8 | 57.1 | 58.5 | 6 |
| Nd(NO ₃) ₃ | 55.76 | 57.49 | 59.37 | 60.38 | 61.43 | 63.69 | 66.27 | 69.47 | | | | | 1:13 |
| NiCl ₂ | 34.7 | 36.1 | 38.5 | 40.3 | 41.7 | 42.1 | 43.2 | 45.0 | 46.1 | 46.2 | 46.4 | 46.6 | 6 |
| Ni(ClO ₄) ₂ | 51.1 | | | 52.8 | | | | | | | | | 7 |
| NiF_2 | | | | 2.50 | | | | | | | 2.52 | | 7 |
| NiI_2 | 55.40 | 57.68 | 59.78 | 60.69 | 61.50 | 62.80 | 63.73 | 64.38 | 64.80 | 65.09 | 65.30 | | 7 |
| $Ni(NO_3)_2$ | 44.1 | 46.0 | 48.4 | 49.8 | 51.3 | 54.6 | 58.3 | 61.0 | 63.1 | 65.6 | 67.9 | 69.0 | 6 |
| NiSO ₄ | 21.4 | 24.4 | 27.4 | 28.8 | 30.3° | 32.0° | 34.1 | 35.8 | 37.7 | 39.9 | 42.3 | 44.8 | 6 |
| Ni(SCN) ₂ | | | | 35.48 | | | | | | | | | 7 |
| NiSeO ₄ | 21.6 | 0.600 | 26.2° | | | | 4.00 | | | | | 45.6° | 7 |
| PbBr ₂ | 0.449 | 0.620 | 0.841 | 0.966 | 1.118 | 1.46 | 1.89 | 1.00 | 2.24 | 0.60 | 0.00 | 0.40 | 2 |
| PbCl ₂ | 0.66 | 0.81 | 0.98 | 1.07 | 1.17 | 1.39 | 1.64 | 1.93 | 2.24 | 2.60 | 2.99 | 3.42 | 2 |
| Pb(ClO ₄) ₂ PbF ₂ | | 0.0603 | 0.0649 | 81.5 0.0670 | 0.0693 | | | | | | | | 7 2 |
| PbI ₂ | 0.041 | 0.0503 | 0.0649 | 0.0670 | 0.086 | 0.112 | 0.144 | 0.187 | 0.243 | 0.315 | | | 2 |
| $Pb(IO_3)_2$ | 0.041 | 0.032 | 0.007 | 0.0025 | 0.000 | 0.112 | 0.144 | 0.167 | 0.243 | 0.313 | | | 7 |
| $Pb(NO_3)_2$ $Pb(NO_3)_2$ | 28.46 | 32.13 | 35.67 | 37.38 | 39.05 | 42.22 | 45.17 | 47.90 | 50.42 | 52.72 | 54.82 | 56.75 | 2 |
| $PbSO_4$ | 0.0033 | 0.0038 | 0.0042 | 0.0044 | 0.0047 | 0.0052 | 0.0058 | 10 | 55.12 | J-1.1 L | 5 1.02 | 20.73 | 2 |
| PrCl ₃ | 48.0 | 48.1 | 48.6 | 49.0 | 49.5 | 50.8 | 52.3 | 54.1 | 56.1 | 58.3 | | | 6 |
| $Pr(NO_3)_3$ | 57.50 | 59.20 | 61.16 | 62.24 | 63.40° | 65.7° | 67.8 | 70.2 | 73.4 | | | | 1:13 |
| RbBr | 47.4 | 50.1 | 52.6 | 53.8 | 54.9 | 57.0 | 58.8 | 60.6 | 62.1 | 63.5 | 64.8 | 65.9 | 6 |
| $RbBrO_3$ | 0.97 | 1.55 | 2.36 | 2.87 | 3.45 | 4.87 | 6.64 | 8.78 | 11.29 | 14.15 | 17.32 | 20.76 | 1:30 |
| RbCl | 43.58 | 45.65 | 47.53 | 48.42 | 49.27 | 50.86 | 52.34 | 53.67 | 54.92 | 56.08 | 57.16 | 58.15 | 1:47 |
| $RbClO_3$ | 2.10 | 3.38 | 5.14 | 6.22 | 7.45 | 10.35 | 13.85 | 17.93 | 22.53 | 27.57 | 32.96 | 38.60 | 1:30 |
| | | | | | | | | | | | | | |

| Formula | 0 °C | 10 °C | 20 °C | 25 °C | 30 °C | 40 °C | 50 °C | 60 °C | 70 °C | 80 °C | 90 °C | 100 °C | Ref. |
|------------------------------------|-------------|-------------|---------|--------------|--------|-----------------|--------|--------|----------------|-------|-------------|--------|-----------|
| RbClO ₄ | 1 | 10 0 | 20 0 | 1.5 | 30 0 | 10 0 | 50 0 | 00 0 | 70 0 | 00 0 | <i>70 C</i> | 17 | 7 |
| RbF | 1 | | 75 | 1.0 | | | | | | | | 17 | 7 |
| RbHCO ₃ | | | 53.7 | | | | | | | | | | 7 |
| RbI | 55.8 | 58.6 | 61.1 | 62.3 | 63.4 | 65.4 | 67.2 | 68.8 | 70.3 | 71.6 | 72.7 | 73.8 | 6 |
| RbIO ₃ | 1.09 | 1.53 | 2.07 | 2.38 | 2.74 | 3.52 | 4.41 | 5.42 | 6.52 | 7.74 | 9.00 | 10.36 | 1:30 |
| RbNO ₃ | 16.4 | 25.0 | 34.6 | 39.4 | 44.2 | 53.1 | 60.8 | 67.2 | 72.2 | 76.1 | 79.0 | 81.2 | 6 |
| RbOH | 10.4 | 25.0 | 34.0 | 37.4 | 63.4 | 33.1 | 00.0 | 07.2 | 1 2.2 | 70.1 | 7.0 | 01.2 | 7 |
| Rb ₂ CrO ₄ | 38.27 | | | 43.26 | 03.4 | | | | | | | | 7 |
| Rb_2SO_4 | 27.3 | 30.0 | 32.5 | 33.7 | 34.8 | 36.9 | 38.7 | 40.3 | 41.8 | 43.0 | 44.1 | 44.9 | 6 |
| SbCl ₃ | 85.7 | 30.0 | 32.3 | 90.8 | 34.0 | 30.9 | 36.7 | 40.5 | 41.0 | 43.0 | 44.1 | 44.7 | 7 |
| SbF ₃ | 79.4 | | | 83.1 | | | | | | | | | 7 |
| $Sc(NO_3)_3$ | 57.0 | 59.3 | 61.6 | 62.8 | 63.9 | 66.2 | 68.5 | | | | | | 1:13 |
| SmCl ₃ | 37.0 | 48.0 | 48.2 | 48.4 | 48.6 | 49.2 | 50.0 | | | | | | 6 |
| $Sm(NO_3)_3$ | E 1 02 | | | | | | 65.05° | 60 1° | 70.9 | 74.2 | | | |
| $SnCl_2$ | 54.83 46 | 56.33 64 | 58.08 | 59.05 | 60.08 | 62.38 | 05.05 | 68.1° | 70.8 | 74.2 | | | 1:13 7 |
| SnI ₂ | 40 | 04 | 0.97 | | | | | | | | | 3.87 | 7 |
| 2 | 46.0 | 40.2 | | <i>5</i> 1.7 | 52.0 | FF 2 | F7 6 | 50.0 | 62.2 | 616 | 66.0 | | 6 |
| SrBr ₂ | 46.0 | 48.3 | 50.6 | 51.7 | 52.9 | 55.2 | 57.6 | 59.9 | 62.3 38.64* | 64.6 | 66.8 | 69.0 | |
| Sr(BrO ₃) ₂ | 18.53 | 22.00 | 25.39 | 27.02 | 28.59 | 31.55 | 34.21 | 36.57 | | 40.2° | 40.8 | 41.0 | 1:14 |
| SrCl ₂ | 31.94 | 32.93 | 34.43 | 35.37 | 36.43 | 38.93 | 41.94 | 45.44° | 46.81° | 47.69 | 48.70 | 49.87 | 8 7 |
| $Sr(ClO_2)_2$ | 13.0 | 13.6 | 14.1 | 14.3 | 14.5 | 14.9 | 15.3 | 15.6 | 15.9 | 66.10 | 6674 | 67.21 | |
| Sr(ClO ₃) ₂ | 63.29 | 63.42 | 63.64 | 63.77 | 63.93 | 64.29 | 64.70 | 65.16 | 65.65 | 66.18 | 66.74 | 67.31 | 1:14 |
| Sr(ClO ₄) ₂ | 70.04* | | | 75.35° | | 78.44° | | | | | | | 7 7 |
| SrF ₂ | 0.011 | 62.0 | 62.5 | 0.021 | 645 | 65.0 | 67.2 | 60.0 | 70.0 | 72.7 | 747 | 70.2 | |
| SrI ₂ | 62.5 | 62.8 | 63.5 | 63.9 | 64.5 | 65.8 | 67.3 | 69.0 | 70.8 | 72.7 | 74.7 | 79.2 | 6 |
| $Sr(IO_3)_2$ | 0.102 | 0.126 | 0.152 | 0.165 | 0.179 | 0.206 | 0.233 | 0.259 | 0.284 | 0.307 | 0.328 | 0.346 | 1:14 7 |
| $Sr(MnO_4)_2$ | 2.5 | | | | 41.0 | 44.2 | | | | | | F0.6 | |
| $Sr(NO_2)_2$ | 20.2 | 24.6 | 41.0 | 44.5 | 41.9 | 44.3 | 47.0 | 40.4 | 40.0 | 40.5 | 50.1 | 58.6 | 7 |
| $Sr(NO_3)_2$ | 28.2 | 34.6 | 41.0 | 44.5 | 47.0 | 47.4 | 47.9 | 48.4 | 48.9 | 49.5 | 50.1 | 50.7 | 6 |
| Sr(OH) ₂ | 0.9 | | | 2.2 | | | | | | | | | 7 |
| SrSO ₃ | | | | 0.0015 | | | | | | | | | 1:26 |
| SrSO ₄ | 0.0 | 12.0 | 177 | 0.0135 | 22.2 | 26.0 | | | | | | | 7 |
| SrS ₂ O ₃ | 8.8 | 13.2 | 17.7 | 20.0 | 22.2 | 26.8 | | | | | | | 7 |
| $Tb(NO_3)_3$ | 2.65 | 2.56 | 60.6 | 61.02 | F 00 | 7.00 | 0.46 | 0.00 | 11 22 | 10.77 | 1410 | 15.50 | 1:13 |
| Tl ₂ SO ₄ | 2.65 | 3.56 | 4.61 | 5.19 | 5.80 | 7.09 | 8.46 | 9.89 | 11.33 | 12.77 | 14.18 | 15.53 | 6 |
| $Tm(NO_3)_3$ | 40.52 | F1 00 | F 4 4 2 | 67.9 | F7 FF | 61.50 | 67.07 | | | | | | 1:13 |
| $UO_2(NO_3)_2$ | 49.52 | 51.82 | 54.42 | 55.85 | 57.55 | 61.59 | 67.07 | (7.0 | 70.5 | | | | 1:?? |
| $Y(NO_3)_3$ | 55.57 | 56.93 | 58.75 | 59.86 | 61.11° | 63.3° | 64.9 | 67.9 | 72.5 | | | | 1:13 |
| $Yb(NO_3)_3$ | 70.2 | 00.1 | 01.0 | 70.5 | 0.4.1 | 05.6 | 05.0 | 06.1 | 06.2 | 06.6 | 06.0 | 07.1 | 1:13 |
| $ZnBr_2$ | 79.3 | 80.1 | 81.8 | 83.0 | 84.1 | 85.6 | 85.8 | 86.1 | 86.3 | 86.6 | 86.8 | 87.1 | 6 |
| ZnC_2O_4 | | 0.0010 | 0.0019 | 0.0026 | 01.4 | 01.0 | 00.4 | 02.0 | 02.7 | 04.4 | 05.0 | 06.0 | 5 |
| $ZnCl_2$ $Zn(ClO_4)_2$ | 44.20* | 76.6 | 79.0 | 80.3 | 81.4 | 81.8 | 82.4 | 83.0 | 83.7 | 84.4 | 85.2 | 86.0 | 6 |
| | 44.29° | | | 46.27* | | | 48.70 | | | | | | 7 |
| ZnF ₂ | 01.1 | 01.0 | 01.2 | 1.53 | 01.5 | 01.7 | 02.0 | 00.0 | 00.6 | 02.0 | 02.2 | 02.7 | 5 |
| ZnI_2 | 81.1 | 81.2 | 81.3 | 81.4 | 81.5 | 81.7 | 82.0 | 82.3 | 82.6 | 83.0 | 83.3 | 83.7 | 6 |
| $Zn(IO_3)_2$ | 47.0 | 50.0 | 0.58 | 0.64 | 0.69 | 0.77 | 0.82 | 07.5 | 90.0 | | | | 5 |
| $Zn(NO_3)_2$ | 47.8 | 50.8 | 54.4 | 54.6 | 58.5 | 79.1 | 80.1 | 87.5 | 89.9 | | | | 6 |
| ZnSO ₃ | 20.1 | 22.0 | 0.1786 | 0.1790 | 0.1794 | 0.1803 | 0.1812 | 40.1 | 41.0 | 20.0 | 20.0 | 27.6 | 5 |
| ZnSO ₄ | 29.1 | 32.0 | 35.0 | 36.6 | 38.2 | 41.3 | 43.0 | 42.1 | 41.0 | 39.9 | 38.8 | 37.6 | 6 |
| ZnSeO ₄ | 33.06 | 34.98 | 37.38 | 38.79 | 40.34 | | | | | | | | 5 |

 $[\]rm ZnSeO_4$ 33.06 34.98 37.38 38.3 Solid phase changes between these temperatures