Source: KBhBIO101Carbs

# 1 | Structures of Carbohydrates

Each carbohydrate could be a monomer (6 carbons, simple structure). A carbohydrate monomer (simple sugar) is called a "monosacharide"

- · Two monomers could be chained to build a more complicated structure named Disachoride
- · Monomers could be chained to build "polymers"
- · Complicated polymers is what forms the energy builds of life
- · The same atoms, with different bonds and hence a different species, result in "isomers"

General chemical formula:  $C_nH_{2n}O$ 

- Monosacharride => a monomer of carbohydrates
- Disachoride => a dinomer (?) of carbohydrates
- Polysachride => a polymer of carbohydrates

#### 1.1 | Basic Monomers

Glucose: ring of 6 carbonsFructose: ring of 5 carbons

## 1.2 | The mer-library

| Name      | Note                      | Composition                                  |
|-----------|---------------------------|--|
| Sucrose   | Common Sugar              | Disachoride: Glucose + Fructose              |
| Lactose   | The thing that's in milk  | Disachoride: Glucose + Galactose             |
| Cellose   | We can't digest this, but | Polysacharides: beta-Glucose + beta-Glucose+ |
|           | plants use it             | beta-beta-Glucose + + Glucose                |
| Glucose   | Bulding block of sugar    | Monomer                                      |
| Galactose |                           | Monomer                                      |
| Fructose  | Controvercial             | Monomer                                      |

## 1.3 | Making and Breaking -mers

#### Creating a polymer ("dehydration")

- · Take monomers
- · Remove water molecules
- · Fill the now-gaping hole with the next monomers

#### Breaking a polymer ("rehydration")

- Take polymers
- · Add water
- · Get Glucose
- Profit!

Hence, you get thirsty after around 45mins whenever you eat lots of sugar — ye gotta get that water to rehydrate and break down those polymers.

Bonds are called "glycocidic" bonds

## 1.4 | Alpha vs Beta glucose

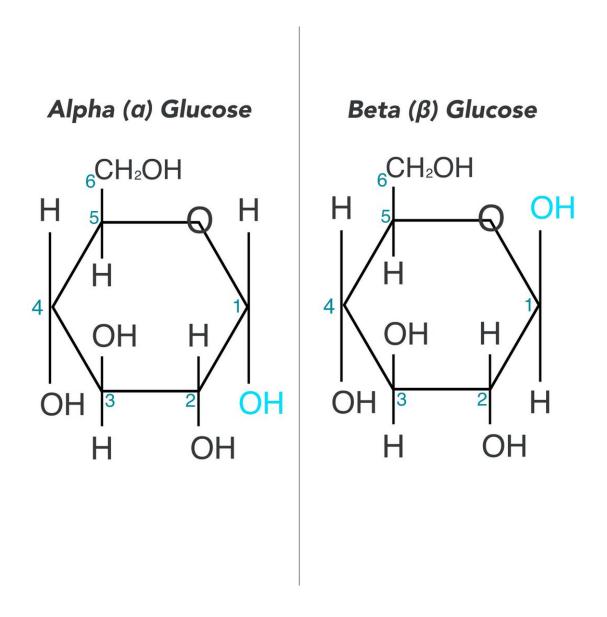


Figure 1: CrLHc0-WEAAe12C.jpg

And now, a note on energy.

[KBhBIO101Enthalpy]

You could add even more monosachrides/disacharides up to get polysacharides (starch, fiber, glycogen)

- We get energy for lots of glucose (the alpha variant of which's polysacharide is starch), but we can't get any from cellulose (whose polysacratide is fiber)
- We eat fiber to maintain gut health + poop goodly. Cellulose is hydrophillic, meaning that fiber makes your guts lubricated.
- · Polysaccharides linked together by glycosidic bonds.

NOTE! Whichever carbohydrates you are using, you get energy from breaking its bonds.