## 1 | Entropy

#flo #disorganized

Startistical measure of randomness in a reaction of systems.

Greater numbers of microstates means that there is more entropy

To think about this, think about states of matter:

- Gas => Most Entropy
- Water => Meh Entropy
- Solids => Least Entropy

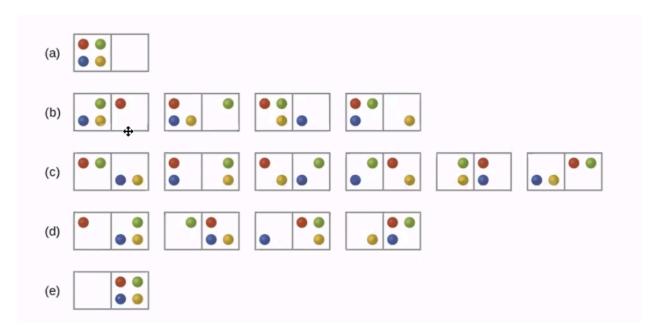


Figure 1: Screen Shot 2020-10-02 at 2.29.24 PM.png

In this image, states (a) and (e) are least likely. This is because the greater the spread, the greater the entropy; systems like to have an increase of entropic state as much as it is possible.

Definition 1 · Second Law of Thermodynamics In the universe, the

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