

Soil Dynamics

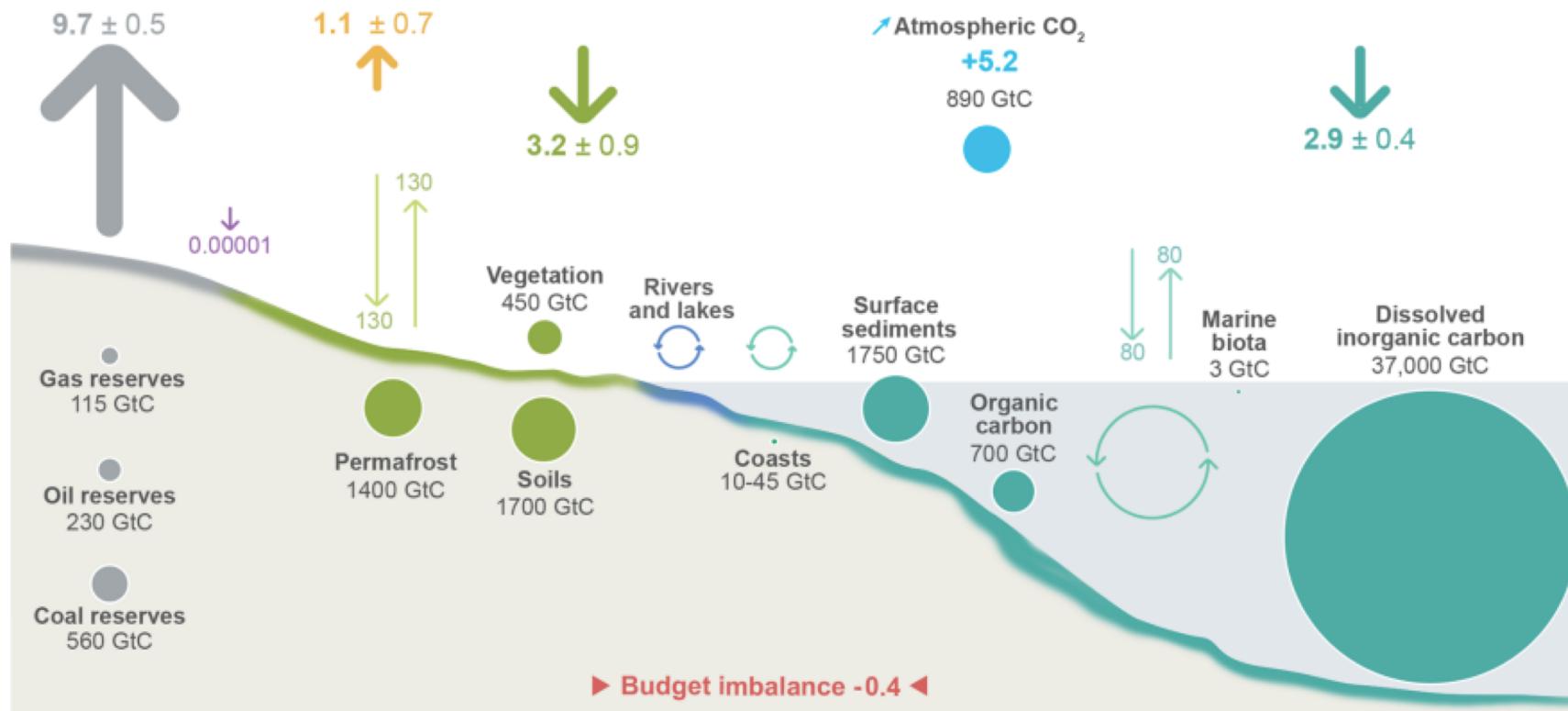
Slides based off of discussions in

Bolker et al. (1998). Linear analysis of soil decomposition: Insights from
the CENTURY model. *Ecological Applications* 8(2): 425-439.

Why soils?

Lots of C in soils!

The global carbon cycle

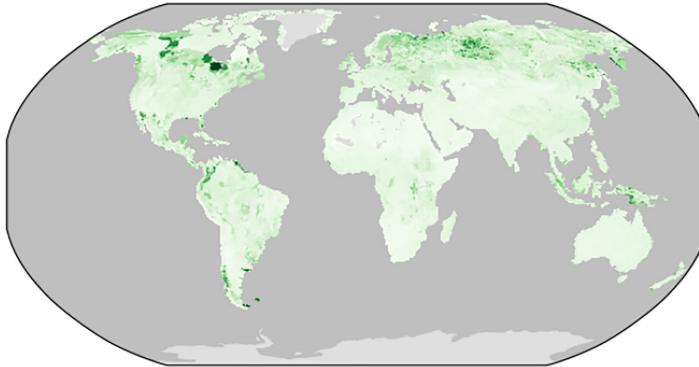


Anthropogenic fluxes
2014-2023 average
GtC per year

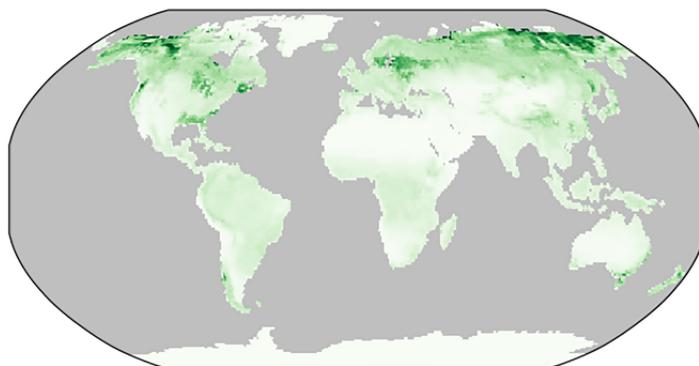
↑ Fossil CO₂ E_{FOS}
↑ Land-use change E_{LUC}
↓ CDR not included in E_{LUC}

↓ Land uptake S_{LAND}
↓ Ocean uptake S_{OCEAN}
+ Atmospheric increase G_{ATM}

Why soils?

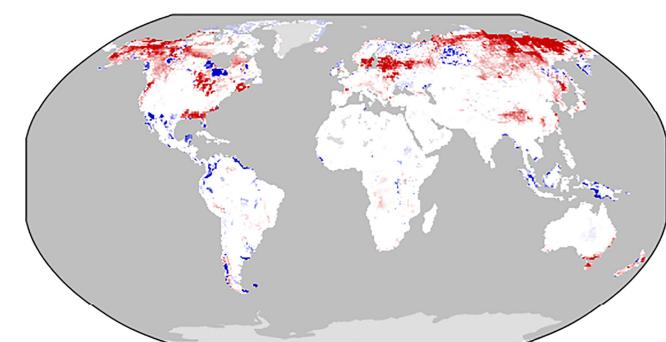
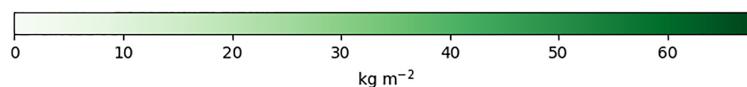


Observation 1



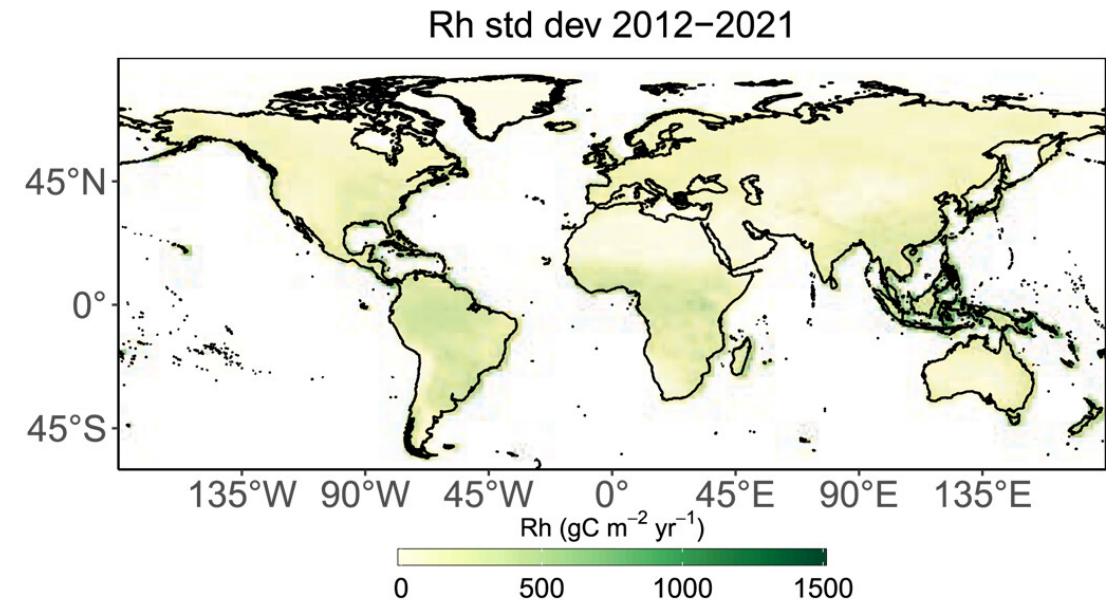
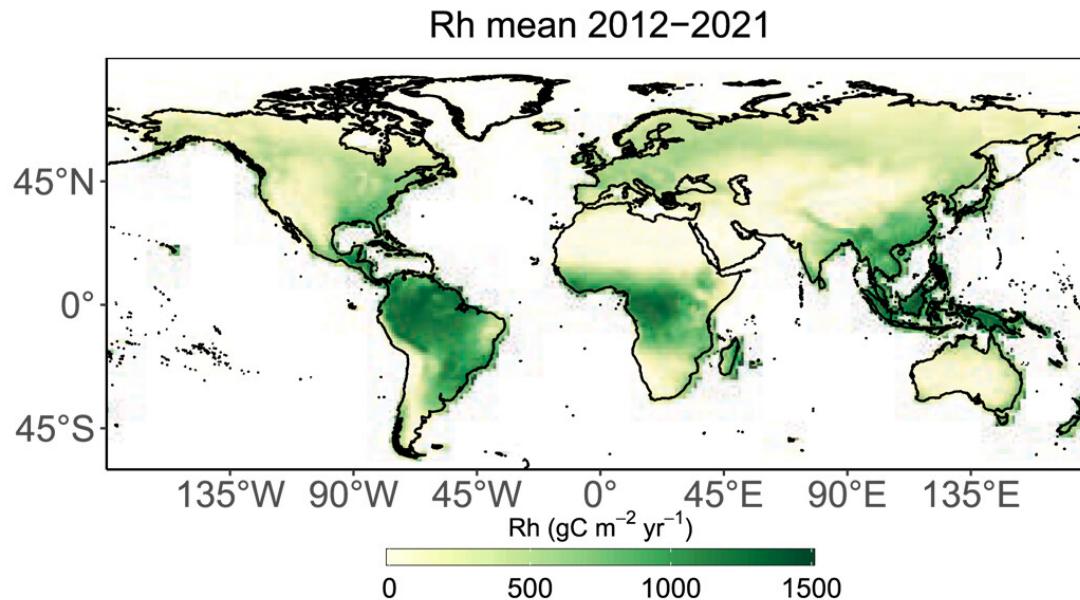
Observation 2

Although it's hard to know how much!



Difference

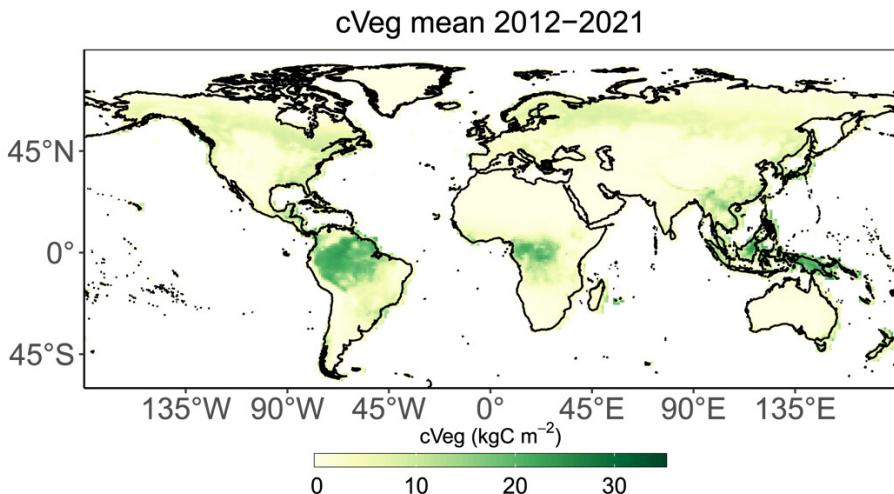
Why soils?



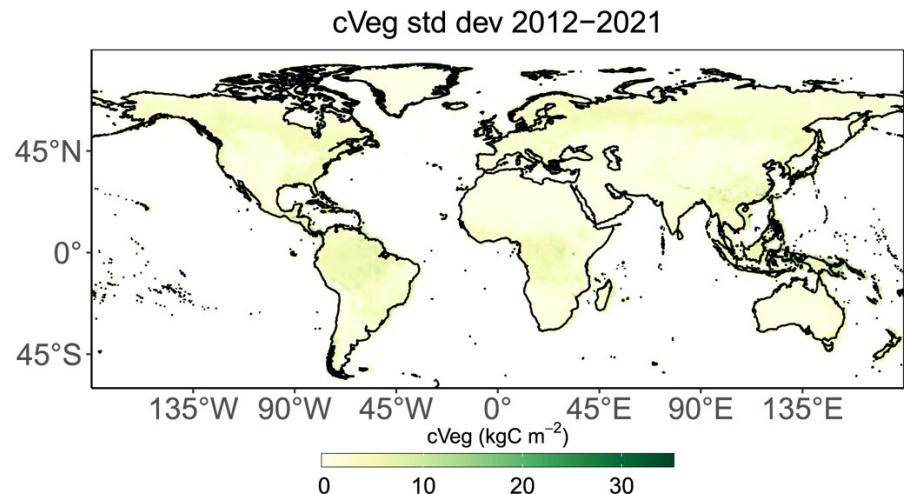
Soils are alive! But how much so?

Why soils?

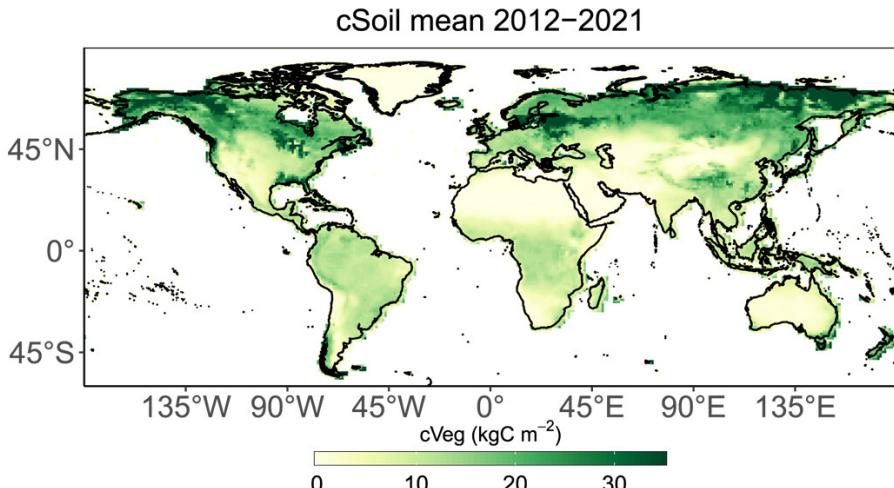
a



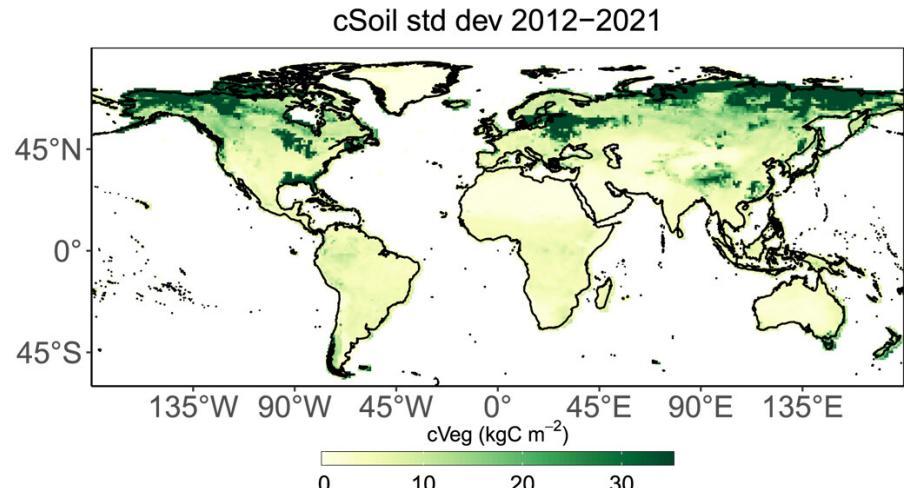
b



c



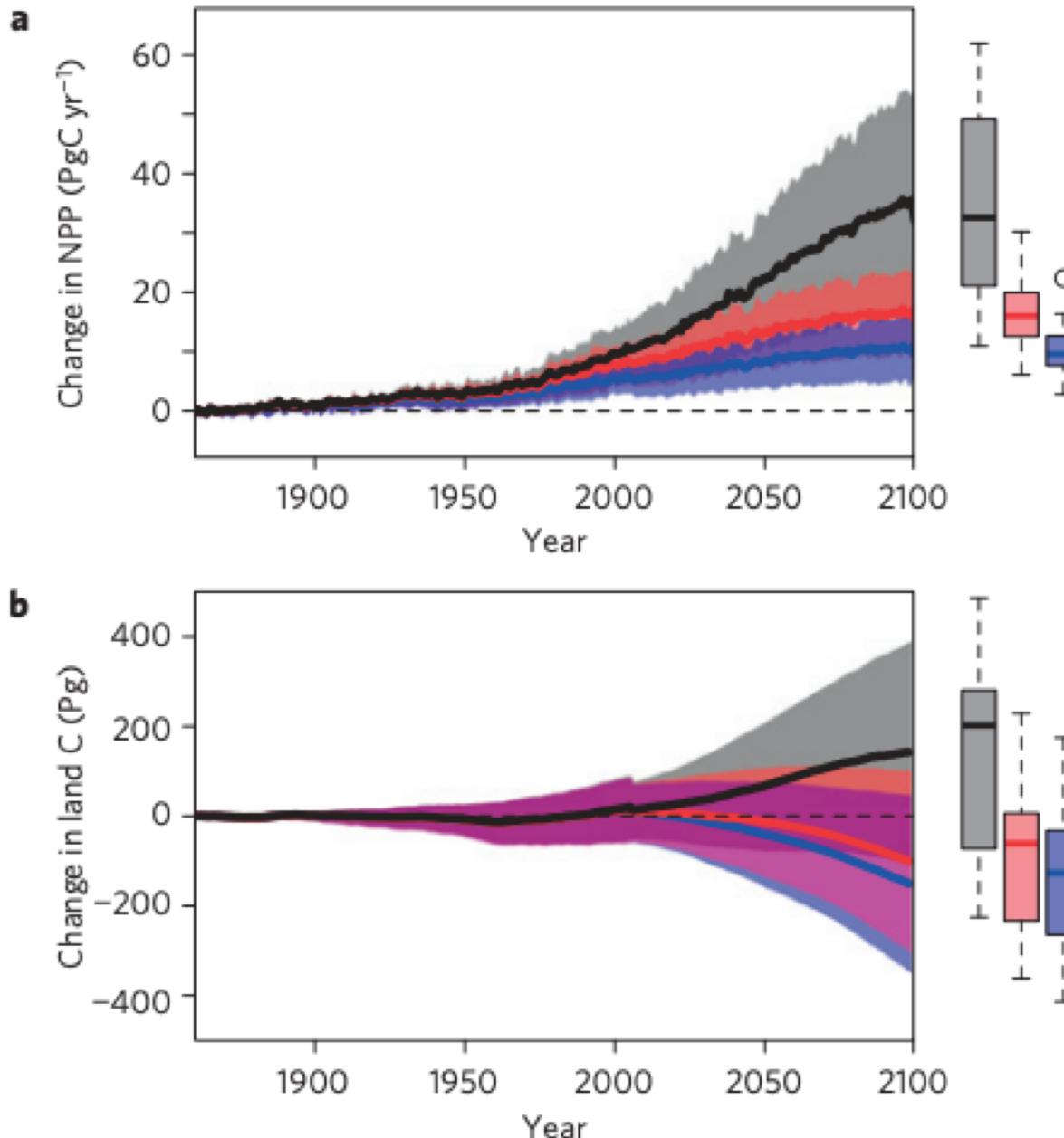
d



Models are uncertain!

Why soils?

They are also drivers of C-N dynamics



How do we model soils? CENTURY as an example

**Soil Science Society
of America Journal**



Division S-3—Soil Microbiology and Biochemistry

Analysis of Factors Controlling Soil Organic Matter Levels in Great Plains Grasslands[†]

W. J. Parton, D. S. Schimel, C. V. Cole, D. S. Ojima

First published: 01 September 1987 |

<https://doi.org/10.2136/sssaj1987.03615995005100050015x> | Citations: 326

[†] Contribution from the Natural Resource Ecology Lab., Colorado State Univ., Fort Collins, CO 80523.



PDF



TOOLS



SHARE

Outline of Bolker et al. (1998)

- Structure of CENTURY
- Modeling soil C decay in CENTURY
 - Carbon
 - Carbon + Nutrients
- Simplification of ecosystem Models

Ecological Applications, 8(2), 1998, pp. 425–439
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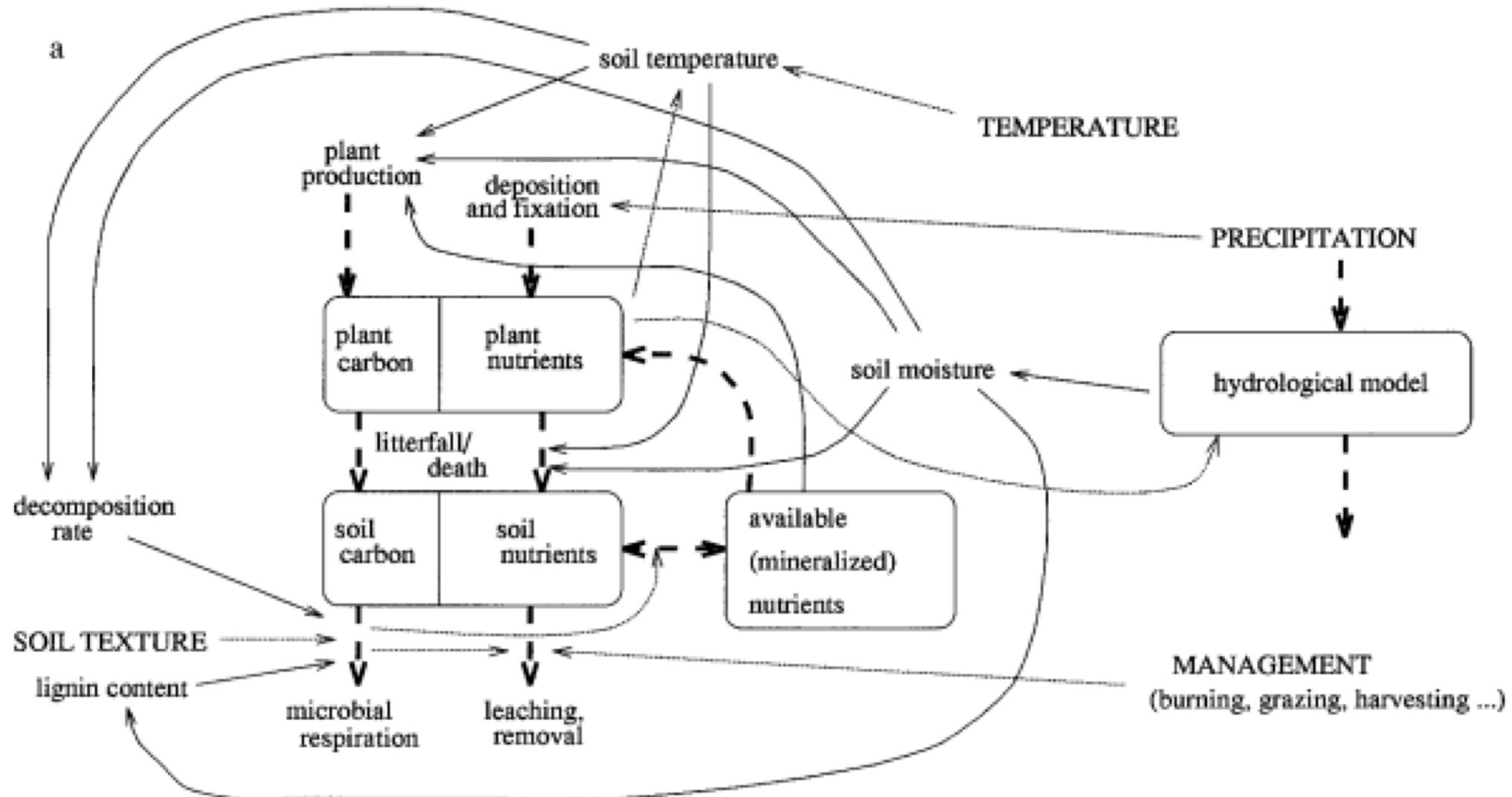
LINEAR ANALYSIS OF SOIL DECOMPOSITION: INSIGHTS FROM THE CENTURY MODEL

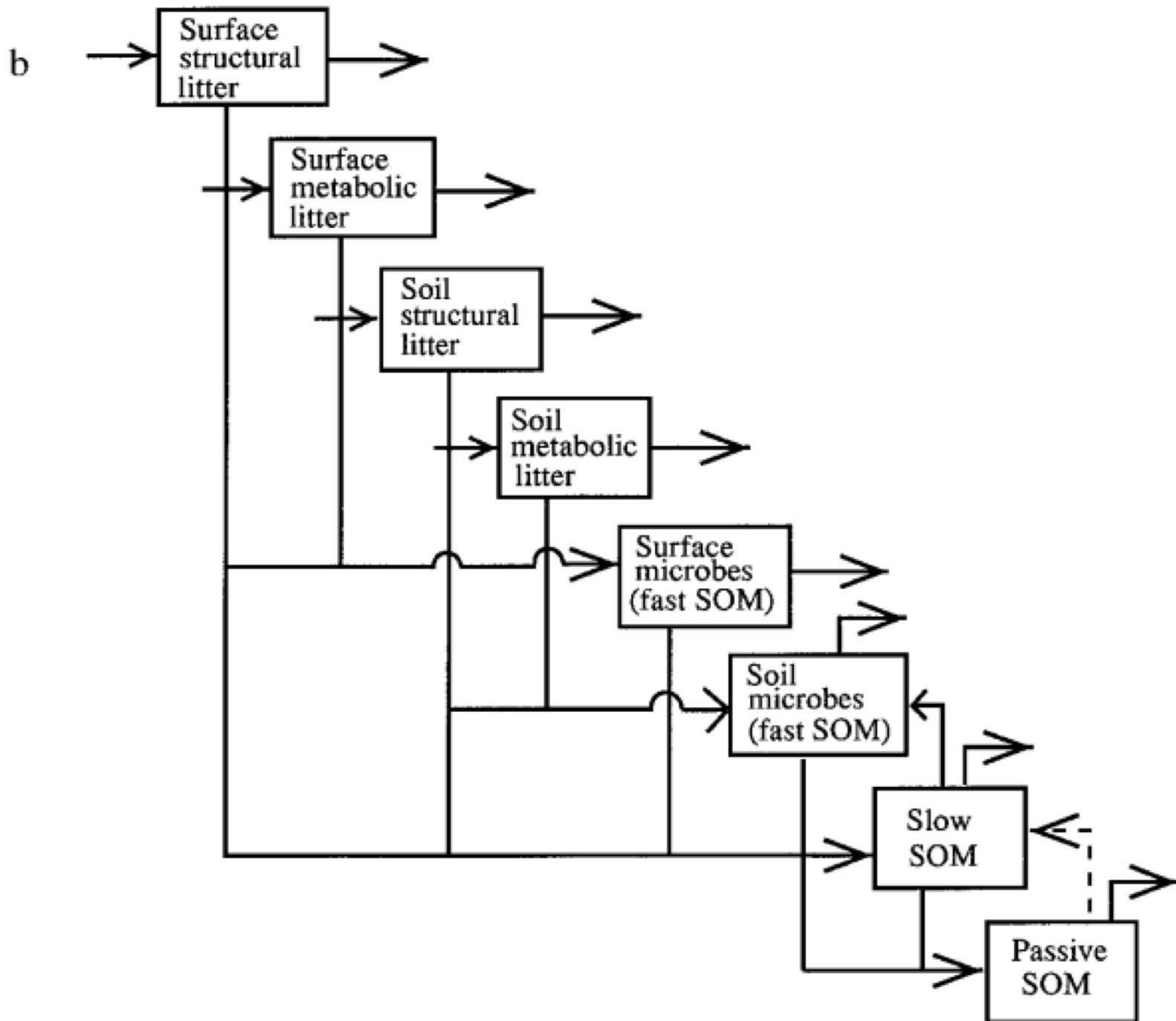
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¹Department of Ecology and Evolutionary Biology, Princeton University, Princeton, New Jersey 08544-1003 USA

²Natural Resources Laboratory, Colorado State University, Fort Collins, Colorado 80523 USA

Structure of CENTURY





Modeling soil C decay

$$\frac{dC}{dT} = \delta(t)MC$$

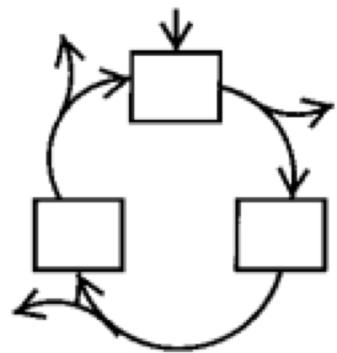
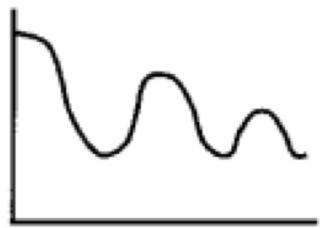
C = vector of C in each compartment (g)

T = time (yr)

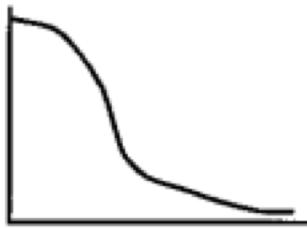
$\delta(t)$ = decomposition factor (unitless?)

M = m_{ij} or rate which material from j goes to i (yr^{-1})

a) Complex eigenvalues



b) Real eigenvalues,
positive and negative
coefficients



c) Real eigenvalues,
all positive coefficients

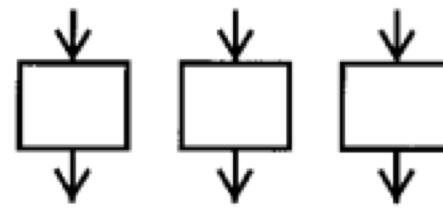
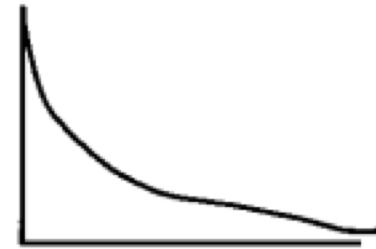
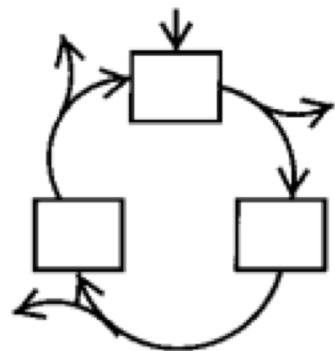


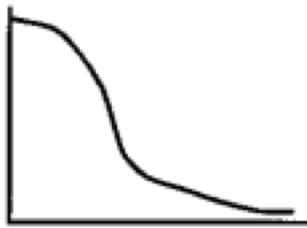
FIG. 2. Dynamical possibilities of a linear decay system: (a) damped oscillations; (b) serial transfer; (c) pure decay.

CENTURY tends to devolve to (c)

a) Complex eigenvalues



b) Real eigenvalues,
positive and negative
coefficients



c) Real eigenvalues,
all positive coefficients

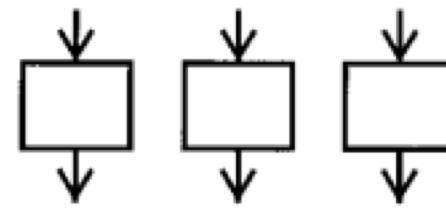
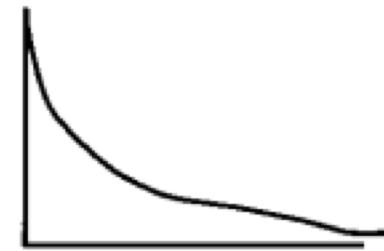
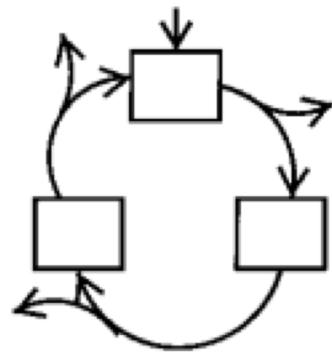


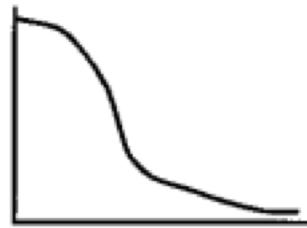
FIG. 2. Dynamical possibilities of a linear decay system: (a) damped oscillations; (b) serial transfer; (c) pure decay.

CENTURY tends to devolve to (c), meaning every C molecule is completely decomposed (returned to atmosphere or moved to more recalcitrant pool)

a) Complex eigenvalues



b) Real eigenvalues,
positive and negative
coefficients



c) Real eigenvalues,
all positive coefficients

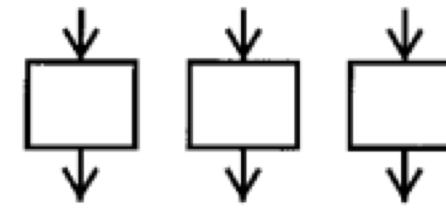
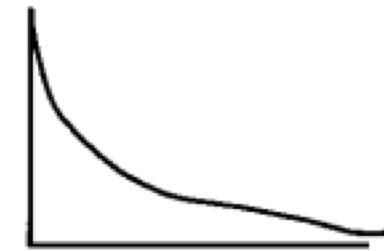
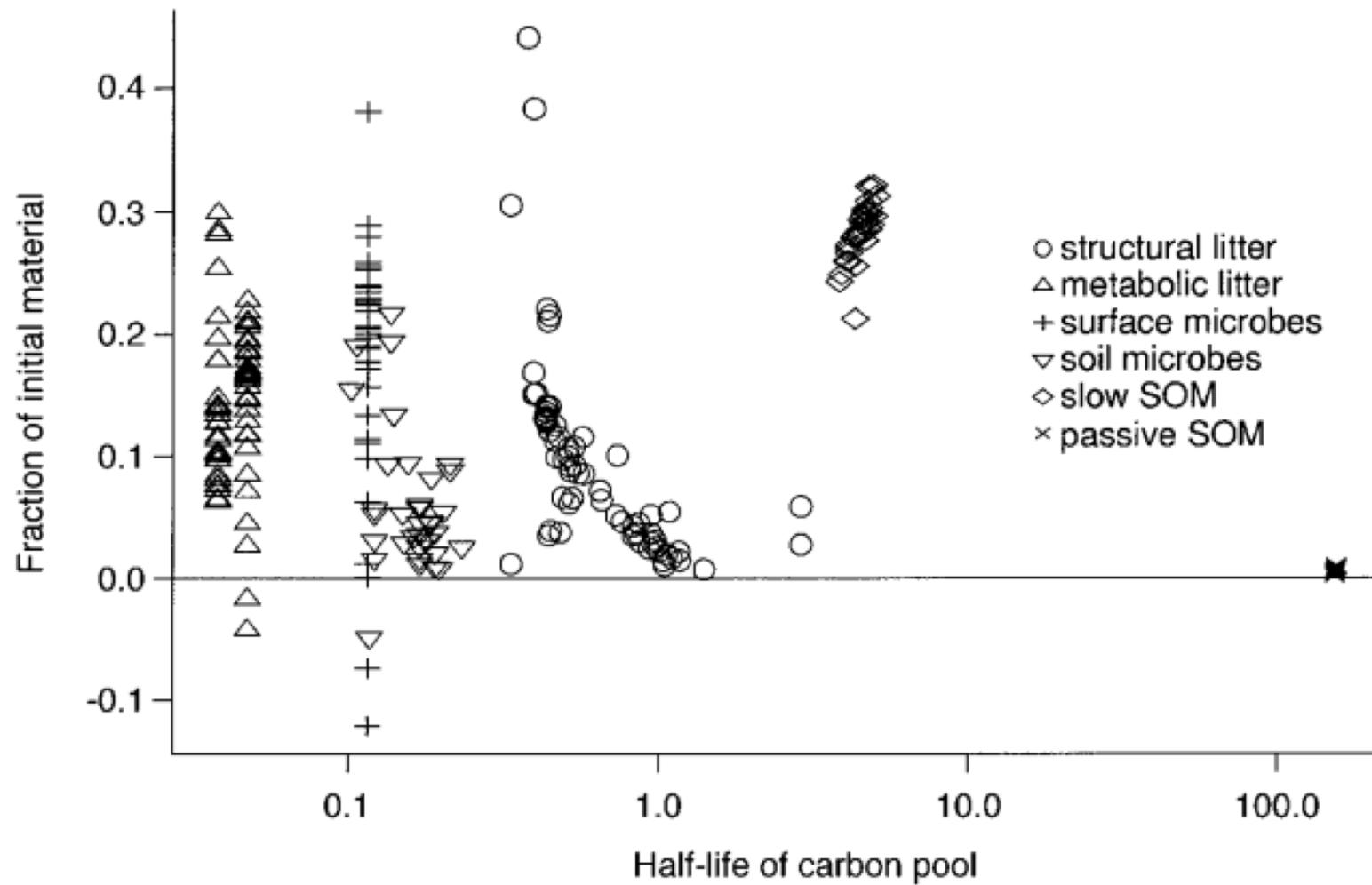


FIG. 2. Dynamical possibilities of a linear decay system: (a) damped oscillations; (b) serial transfer; (c) pure decay.

CENTURY tends to devolve to (c), meaning C can be partitioned
instantaneously depending on time scale of interest



Positive values
indicating continuous
flow through pools (but
note the passive pool)

FIG. 3. Decay rates (shown as half-lives) and loadings for 34 SCOPE sites around the world, determined by calculating the eigenvalues and eigenvectors for the soil texture, precipitation, and C:N ratios at each site. The horizontal axis shows the base decomposition half-life [$\log_2(e)/\lambda$; based on a decomposition factor of 1] for each carbon pool; the vertical axis shows the fraction of each incoming cohort of carbon in each pool (negative values represent serial transfers as in Fig. 1b). The passive pool appears as a tight cluster near (100, 0); the quantity and half-life of passive material vary little between sites.

CENTURY approach to nitrogen

- N model ignores explicit C dynamics
- N model is a closed system (recycling between plants and soil)
- N limits growth if water and temperature are not limiting

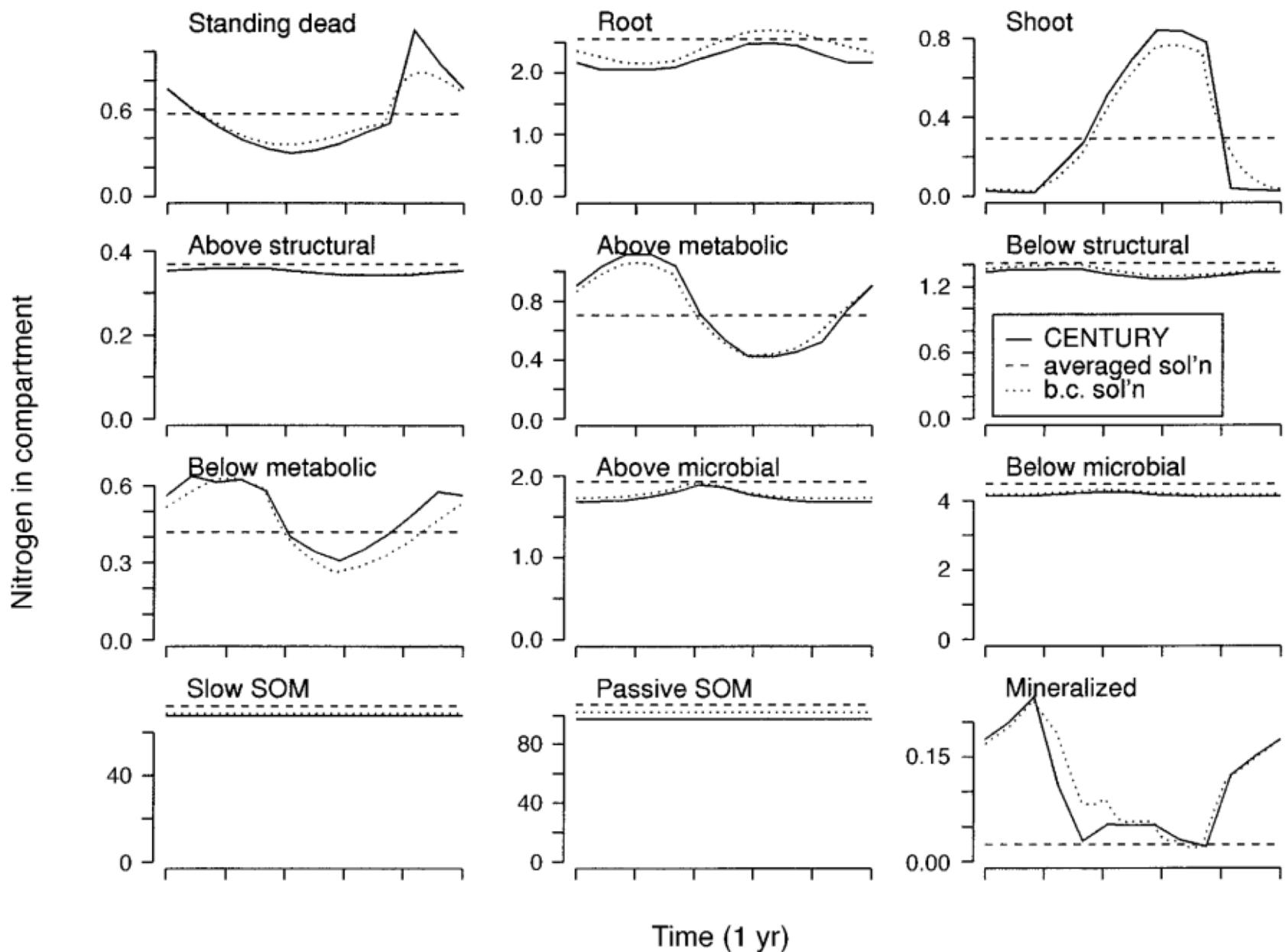


FIG. 4. Output of the CENTURY equilibrium equations (Appendix C) and boundary-condition solver compared with long-term stable seasonal variation in the real model (CENTURY model run for 5000 yr). Each subplot shows the nitrogen content of each nitrogen pool in the model over the course of 1 yr. Abbreviations: sol'n = solution; b.c. = boundary condition.

Simplification of ecosystem models

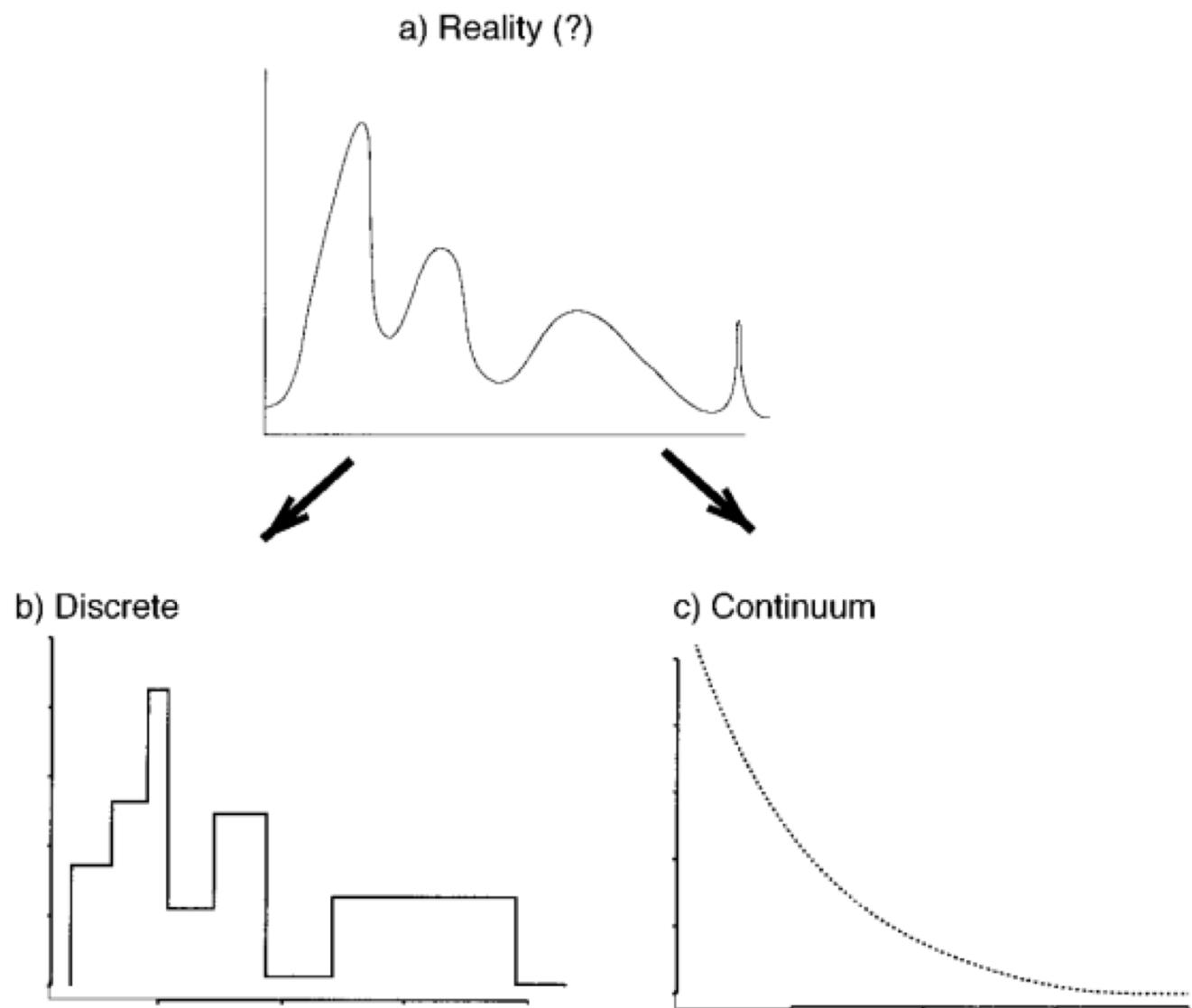
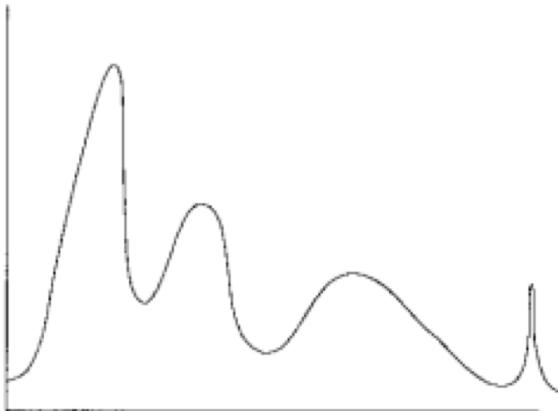
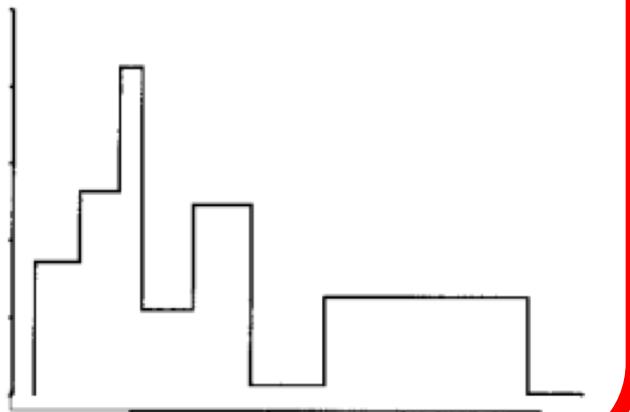


FIG. 5. Different approximations of distributions of decay times. The real (unknown) distribution of eventual decay times for the carbon compounds in a given sample of litter may look something like (a), with particular classes of compounds giving rise to particular peaks of different heights (total quantity) and widths (variability in chemical and physical properties). A model could represent this unknown distribution as (b) a series of discrete boxes, as in CENTURY, or as (c) a continuous distribution of decay times (see Fig. 7).

a) Reality (?)



b) Discrete



c) Continuum

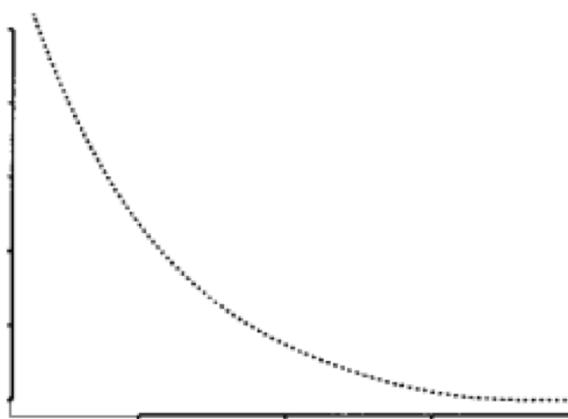


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Discrete: how many levels to use?

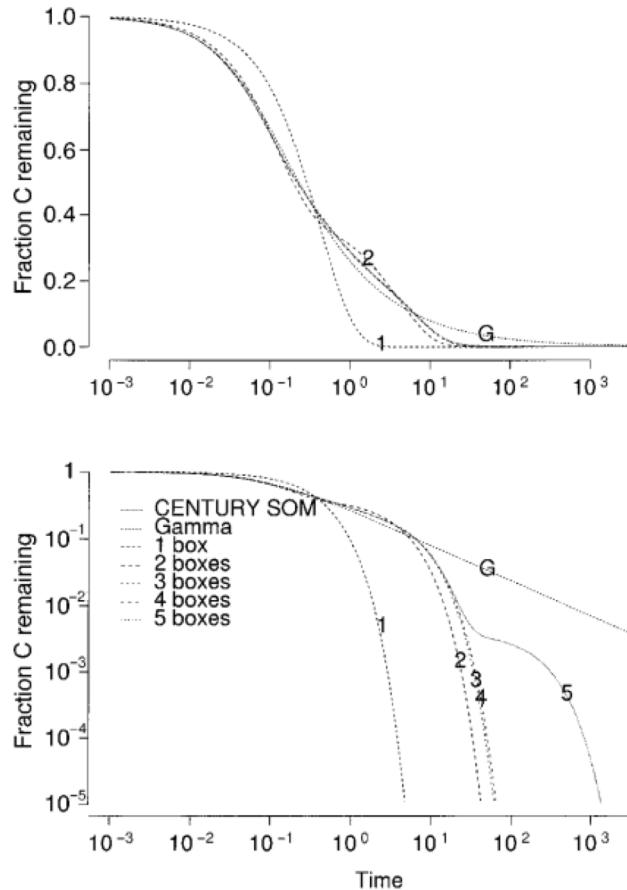
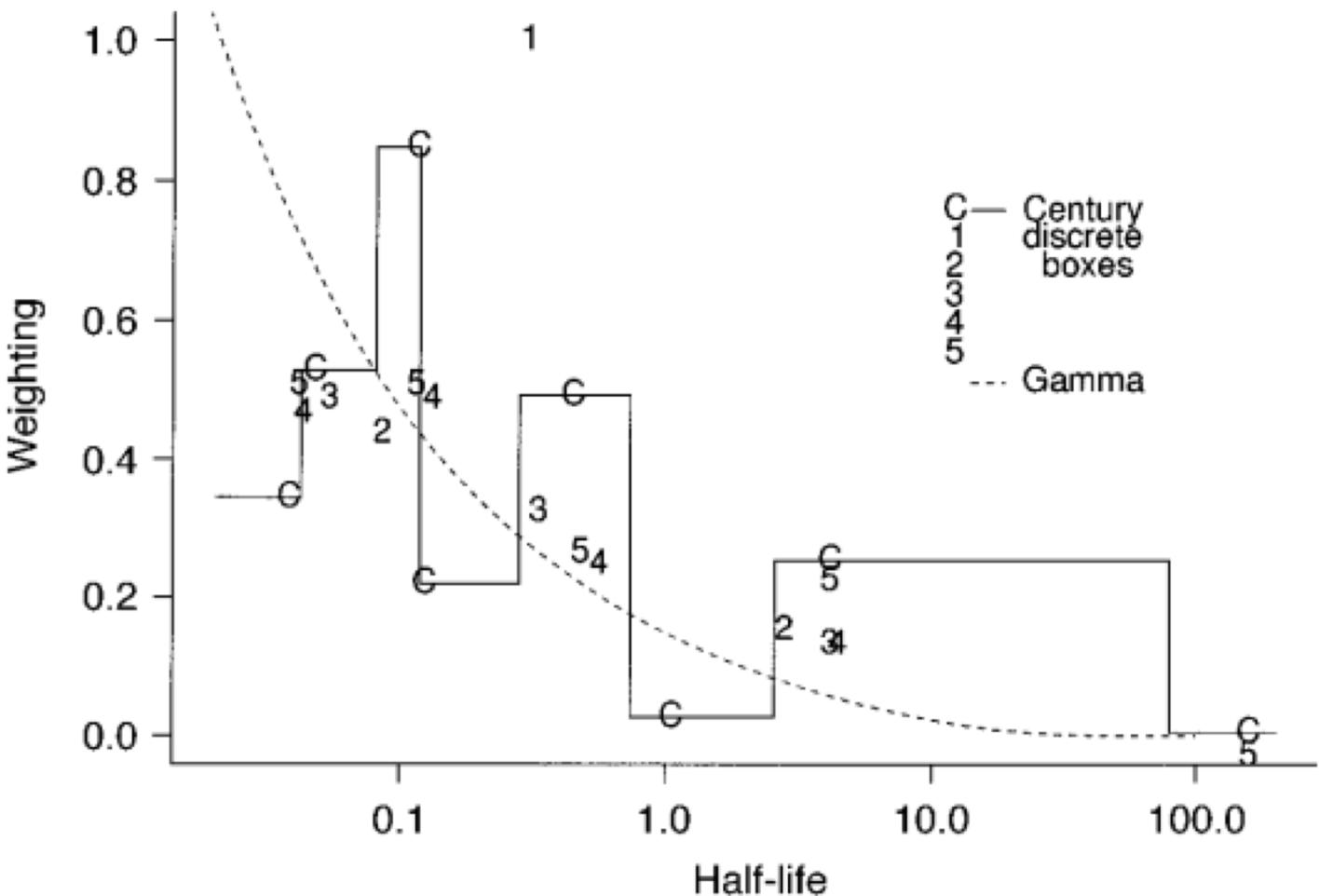


FIG. 6. Decay of one litter cohort over time for typical (CPER) parameters, as approximated by continuous (gamma distribution) and discrete models. The horizontal (time) axis is logarithmic. (a) Linear vertical scale: 3-, 4-, and 5-box models are indistinguishable from the full 8-box dynamics. (b) Logarithmic vertical scale: 5-box model is indistinguishable from the full 8-box dynamics.

FIG. 7. Distribution of decay rates and loadings for different fitted models. The horizontal axis shows half-life of carbon decay on a logarithmic scale, as in Fig. 3. The vertical axis shows the coefficients, normalized by the distance between boxes, so that the figure shows a rough frequency distribution of material with different half-lives. (The choice of normalization depends on the minimum and maximum possible decay rates and on the the horizontal axis scaling.) The solid line shows the distribution for the CENTURY model with CPER parameters; different numbers show n -box fit parameters; the dashed line shows the fit for the gamma distribution. Some loadings for the 5-box model are offset slightly for legibility.



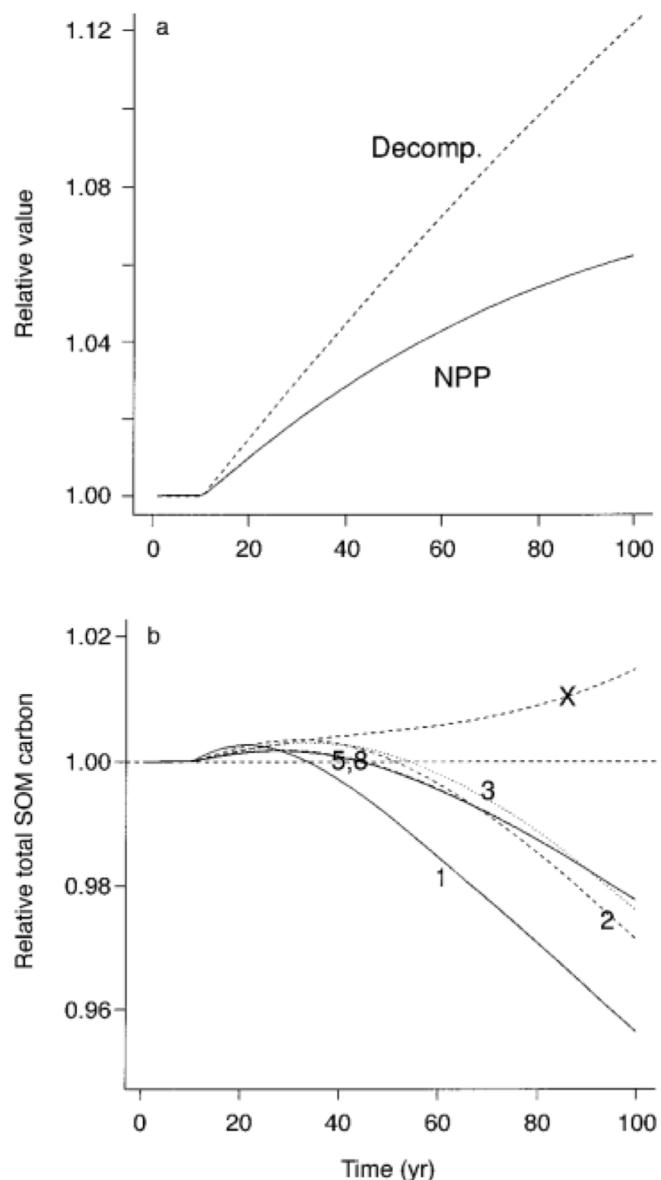


FIG. 8. Effects of SOM model simplification on a global-change scenario. (a) Relative changes in NPP (net primary productivity) and decomposition factor [$\delta(t)$] over time. Decomp. = decomposition. (b) Relative changes in total SOM carbon over time, for different numbers of compartments. Numbers label results from n -compartment models; \times labels results of increasing passive carbon. (The vertical axis is strongly expanded to show differences among models.)

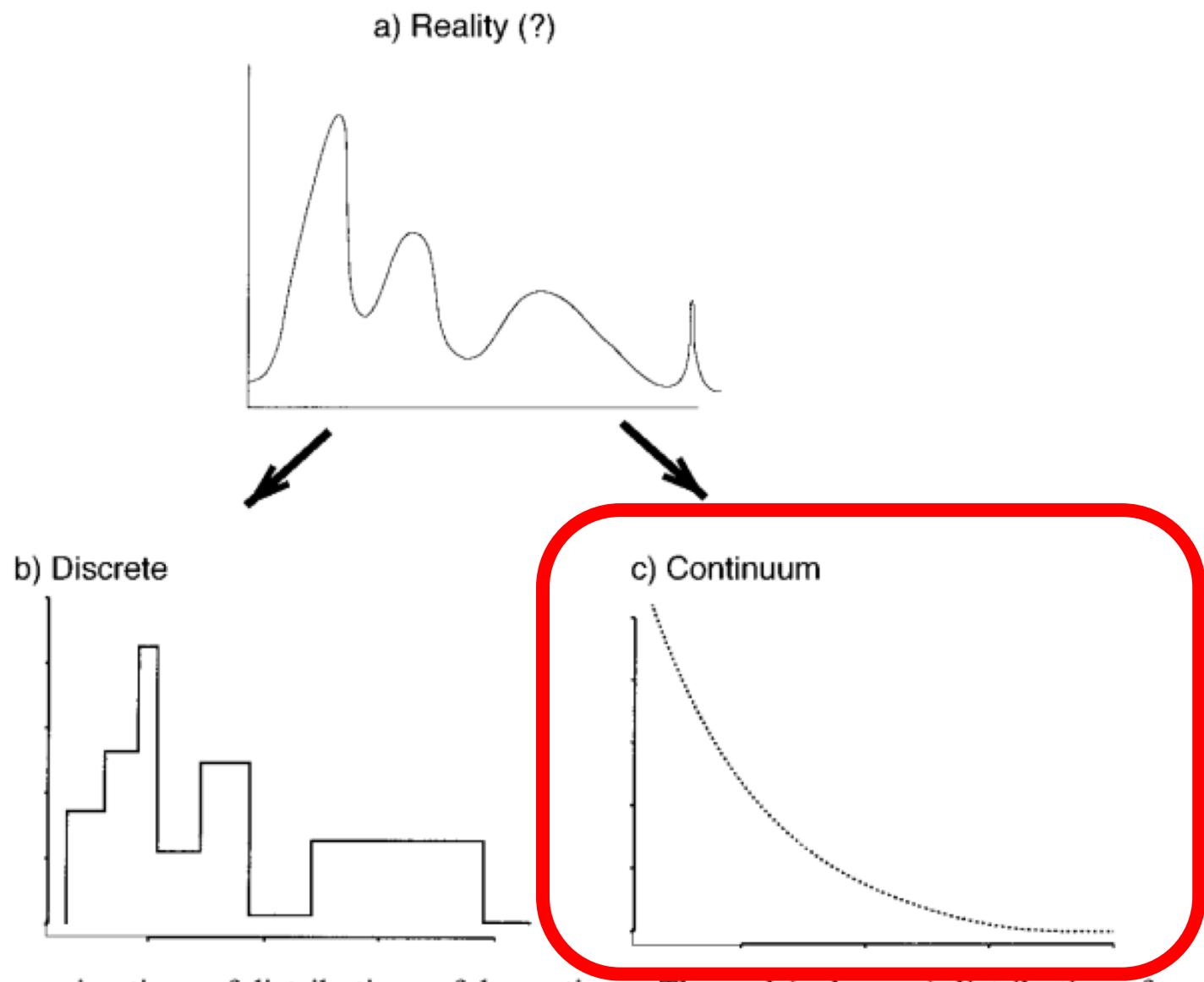
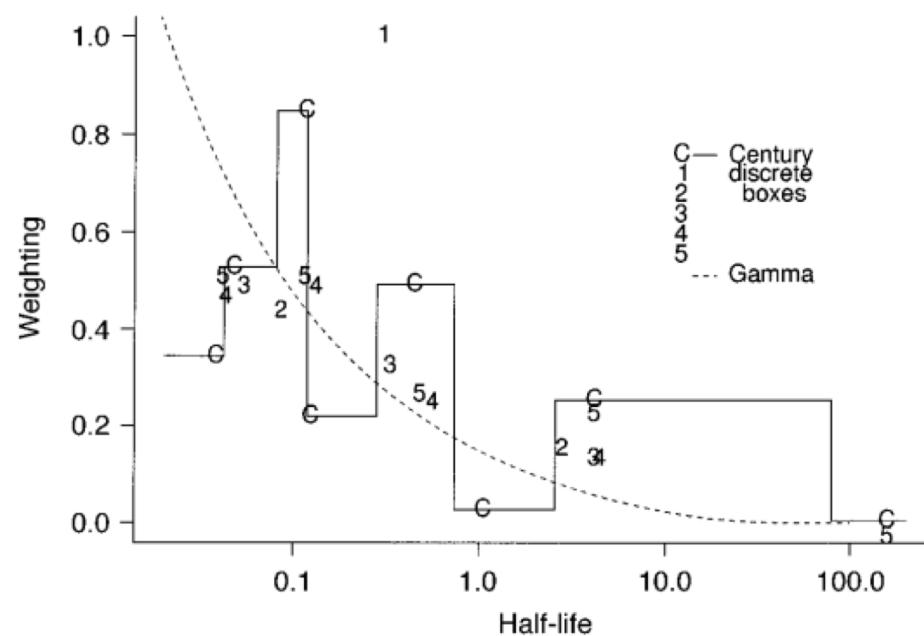


FIG. 5. Different approximations of distributions of decay times. The real (unknown) distribution of eventual decay times for the carbon compounds in a given sample of litter may look something like (a), with particular classes of compounds giving rise to particular peaks of different heights (total quantity) and widths (variability in chemical and physical properties). A model could represent this unknown distribution as (b) a series of discrete boxes, as in CENTURY, or as (c) a continuous distribution of decay times (see Fig. 7).

Is a continuous model distribution better?

- Need to pick a distribution
- And associate parameters
- Depends on availability of data and application for the model



Some general takehomes

- CENTURY can be simplified
- Simple models are easier to understand
 - E.g., parameter sensitivity
- Construct simplest model first and continuously test added complexity
 - More validation for the class approach!