

# Multi-Element Lunar Exosphere Simulation to Improve the Prediction of Ballistic Migration of Water.

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## Introduction

- New **numerical model** of the lunar exosphere based on the Monte-Carlo method, simulates individual, weighted particles in thermal equilibrium with the Moon surface (Smolka, 2022; Tucker et al. 2019).
- Simultaneously simulates **H, H<sub>2</sub>, OH, and H<sub>2</sub>O**, connected through conversion reactions like **photo-dissociation** and **geochemical surface reactions**.

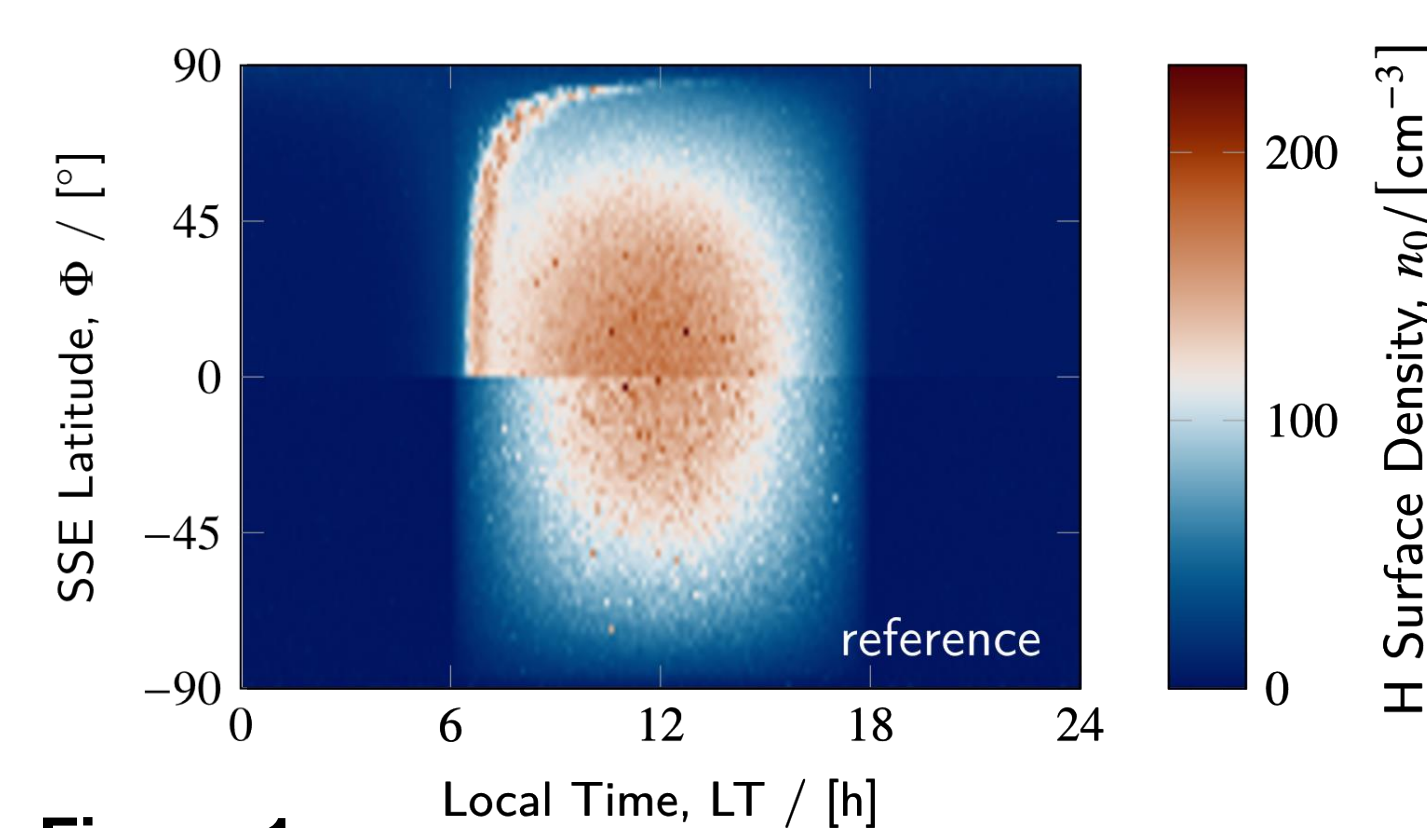
## Method

- The model takes advantage of a nested-loop architecture that iteratively solves for the **surface number density** while calculating particle conversions to be applied as source rates in the following iteration.
- Assumes pseudo-stationary **surface-bounded exosphere** in thermal equilibrium with a **Maxwell-Boltzmann flux** distribution of energies and full thermal accommodation (Brinkmann, 1970; Smith et al., 1978).
- The **loss and source mechanisms** of all four species are connected through particle conversion, based on **dissociation rates** (Huebner et al., 1992) and **surface reaction probabilities** (Crider and Vondrak, 2002).
- Uncertain parameters are modelled with probability distributions (like in Tucker et al., 2018), allowing a statistical analysis of the **uncertainty propagation**.

## Results

- Adsorption during the night and desorption at sunrise leads to **OH and H<sub>2</sub>O accumulating at the morning-terminator**, exhibiting a “snowplow” effect stretching from pole to pole.
- While H does not ad- or desorb, its peak must originate from the **connection to OH and H<sub>2</sub>O**.
- Conversions act as efficient recycling of particles, leading to **generally higher densities** compared to single-element studies.
- Confirms expected surface number densities of **H<sub>2</sub>: ~ 1200 ± 400 cm<sup>-3</sup>** (LAMP, Hurley et al., 2017) **OH: ~ 10<sup>4</sup> cm<sup>-3</sup>** (Wang et al., 2015) but **cannot** reproduce expectations for **H<sub>2</sub>O: ~ 60cm<sup>-3</sup>** (Jones et al., 2018).

Hydrogen bearing species **require multi-element models and simultaneous simulations** to predict the ballistic transportation of water. For accurate description of OH/H<sub>2</sub>O accumulations, a more sophisticated **geochemical lunar regolith model is needed**.

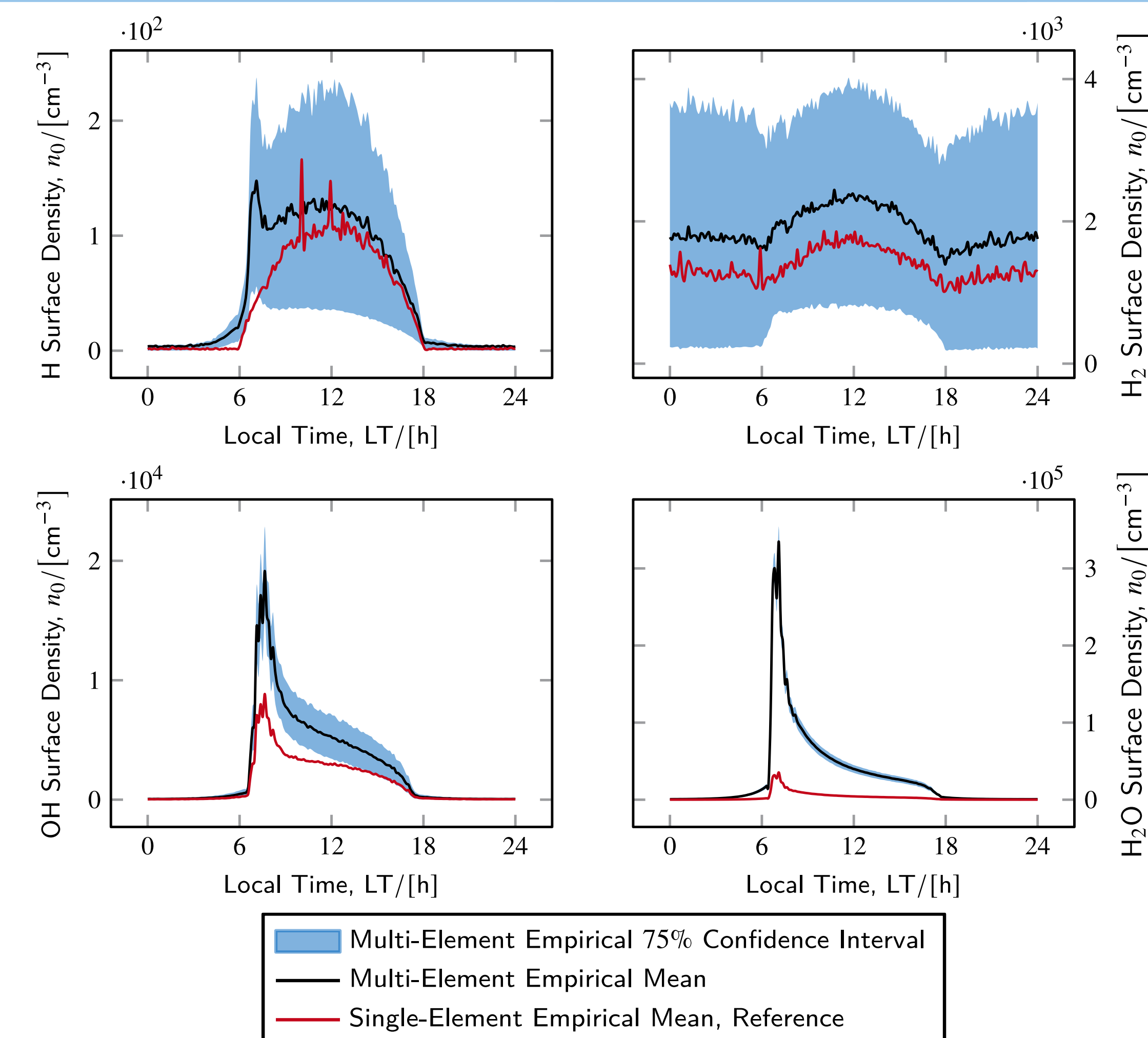


**Figure 1**  
Comparison of exospheric atomic hydrogen predictions of the multi-element study with a reference single-element simulation.

## Conclusion

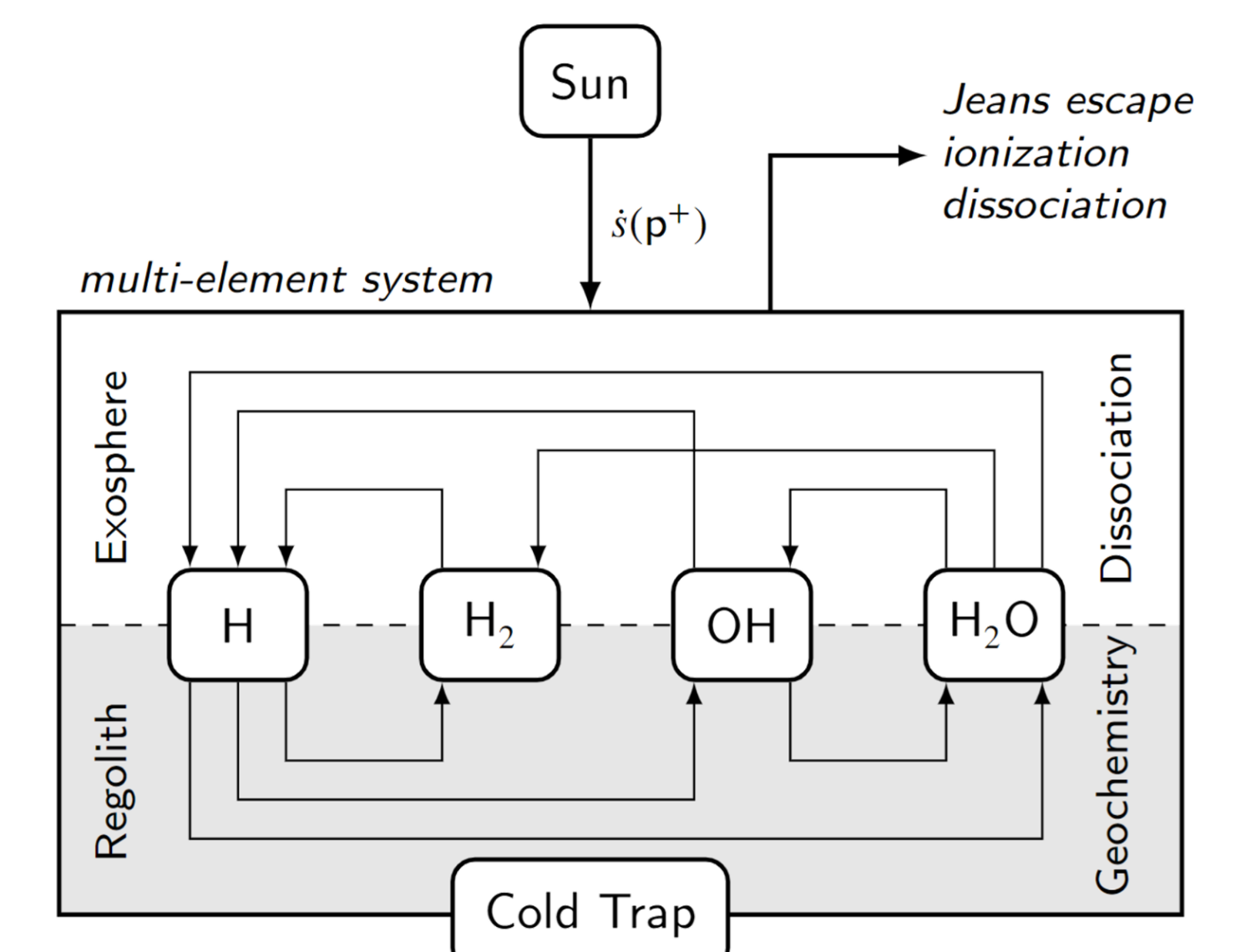
Clear changes in the surface number densities are visible once the species were connected in a multi-element model respecting particle conversions. Non-condensable species, like H, **develop a morning-terminator peak** due to heavy OH and H<sub>2</sub>O desorption and dissociation. **Globally increased densities** suggest that conversion acts as a recycling mechanism.

Large uncertainty intervals and density increases strengthen the need for a **more sophisticated geochemical model** to better describe the **lunar water cycle**. Models of Jones et al. (2018), Tucker et al (2019), and Grumpe et al. (2019) will be used as guidelines in future iterations of the numerical model, especially focussing on **better descriptions of the OH/H<sub>2</sub>O behavior**.

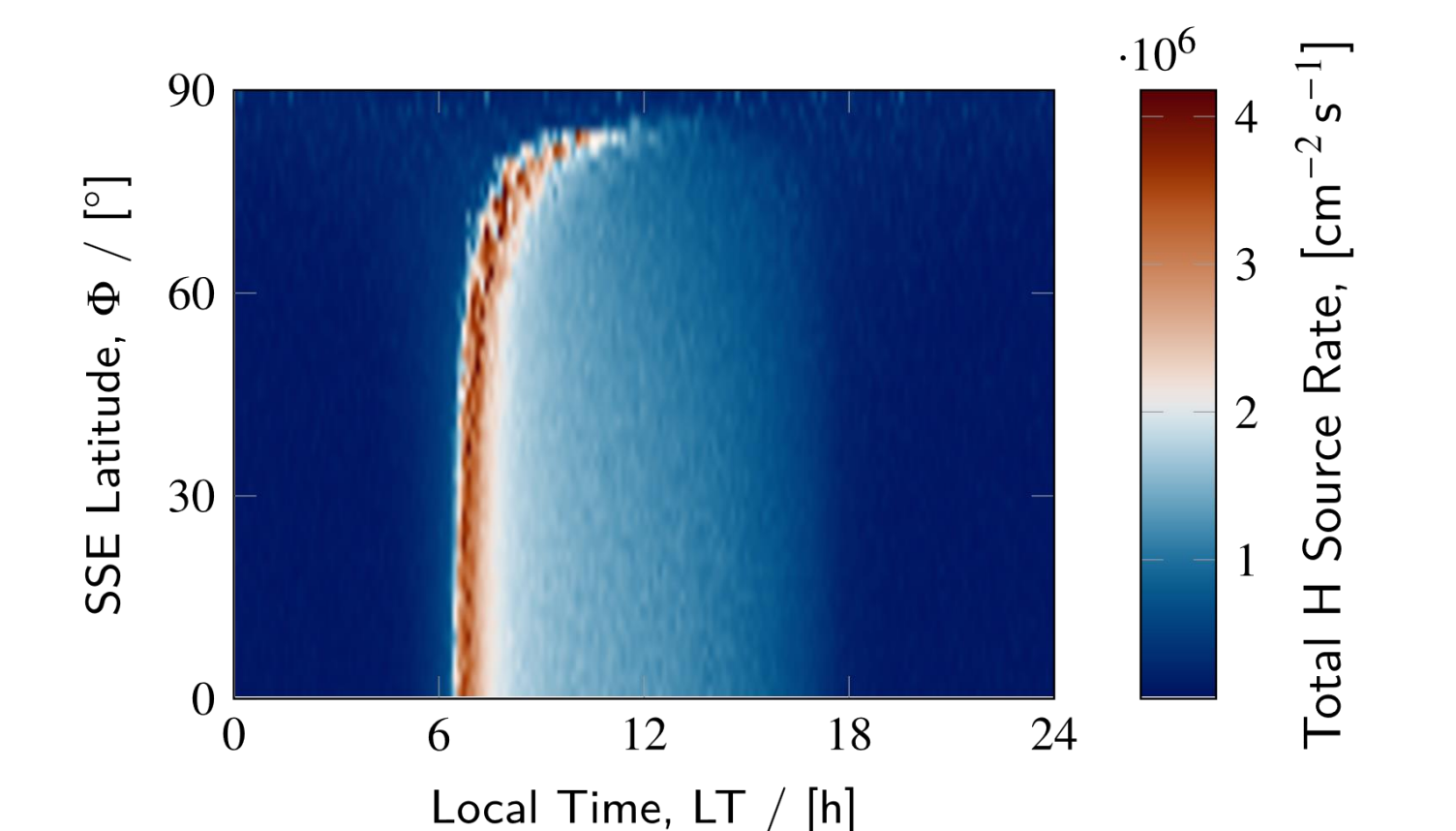


**Figure 2**  
Surface number densities of H, H<sub>2</sub>, OH, and H<sub>2</sub>O of the multi-element study including its empirical 75% confidence interval, and a reference density as a result of a single-element study with the same parameter setup.

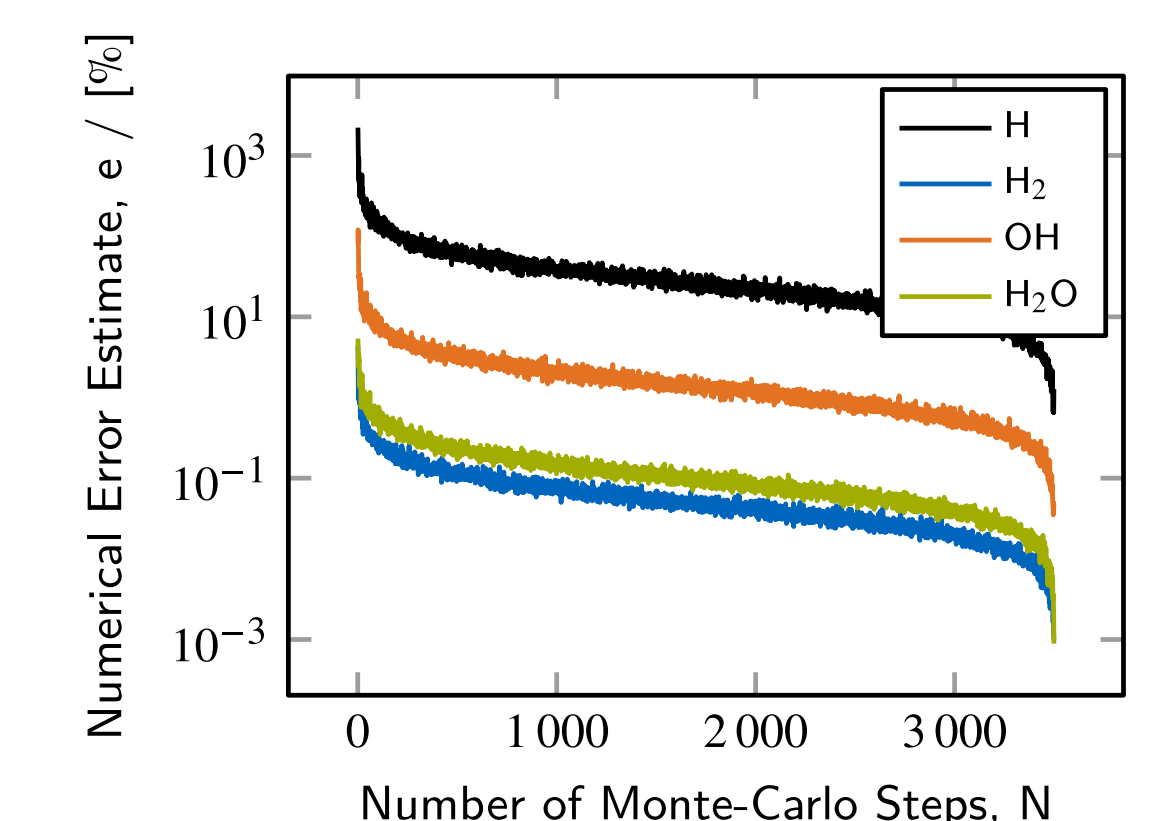
## Supplemental Materials



**Figure 3**  
Schematic of the implemented conversion processes in both the exosphere and the lunar surface.



**Figure 4**  
Total source for H particles based on conversions from H<sub>2</sub>, OH, and H<sub>2</sub>O through photo-dissociation.



**Figure 5**  
Numerical error estimation for all four constituents over the number of Monte-Carlo steps. Measured at a reference point on the day-side.

## Data availability

For data and model access, please contact Alexander Smolka ([alexander.smolka@tum.de](mailto:alexander.smolka@tum.de)).

Previous work: <https://mediatum.ub.tum.de/node?id=1658109>

