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Laboratory Module for Master Programs  
**Biophysics Lab Course**

Experiment:  
**Ionchannel**

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Group 8

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We hereby confirm that we have elaborated the present work independently and have detailed knowledge of the entire contents.

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# 1 Introduction

## 1.1 Gramicidin A

Gramicidin A is a linear peptid consisting of 15 alternating L- and D- amino acids with hydrophobic side chains. Its left hand  $\beta$  helix formation projects all hydrophobic side chains outwards, making it a high affinity inside the lipid membrane. Meanwhile, a hydrophilic channel is formed in the centre by the carbonyl groups of the peptide backbone. In a linear dimer, two gramicidin molecules could form a hydrophilic transmembrane pore and enables the ions to travel through.

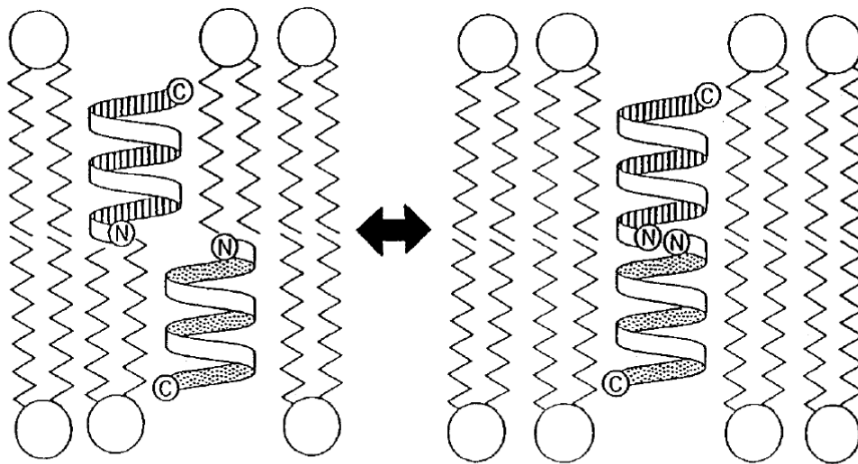


Figure 1.1: The sketch of the gramicidin monomer and dimer in the membrane

## 1.2 Ion transport

Pure lipid membrane seldom permeable for ions. In cells, the ions are transported via the transmembrane channels. For gramicidin channels, the negatively polarized carbonyl groups form an binding sites array for the cations. An kinetic model could be derived by assuming there are only two symmetry binding sites in the channel:

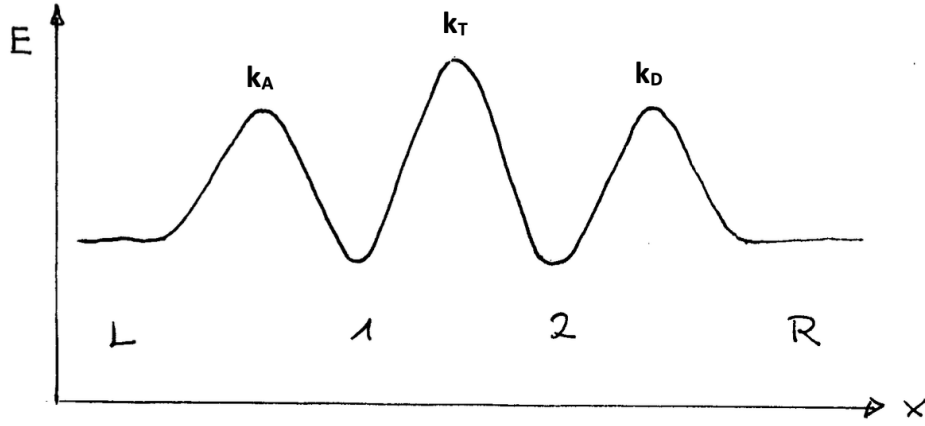


Figure 1.2: Potential plot of the two site binding model

Here the  $k_A$ ,  $k_T$  and  $k_D$  are the rates of binding, transport and dissociation process of the ions, and the barrier height  $E^*$  determined the possibility of the ion hopping to another state, where:

$$k = k_0 \cdot e^{-\frac{E^*}{k_B T}} \quad (1.1)$$

If a voltage  $V_M$  was applied on the membrane, the energy barrier now is  $E^* = V_M q$ . And when the voltage is small, we could acquire the following formulas describe the current through a single ion pore:

$$i_0 = \frac{k_A \cdot c}{2 + \frac{k_D}{k_T}} \cdot \frac{q^2}{k_B T} \cdot V_M \quad (1.2)$$

This represents Ohm's law with the single channel conductance:

$$\Lambda = \frac{k_A \cdot c}{2 + \frac{k_D}{k_T}} \cdot \frac{q^2}{k_B T} \quad (1.3)$$

which depends on the linear relationship of the ion concentration to the channel numbers  $N_p(t)$ :

$$I_M(t) = N_p(t) \cdot \Lambda \cdot V_M \quad (1.4)$$

The current steps could be observed if there are one or a few of channels open together. With the chemical dynamic model of the dimerization:



We could analyse the dissociation rate of gramicidin dimer channel with the single channel open and close signal, since it should follow the poisson interval distribution. For multi channels opening signal, the changing rate is related to both association and dissociation rate, thus we could not calculate the dissociation rate. However, we could still observe the stepwise current distribution.

### 1.3 Autocorrelation analysis

When there are many channels open and close in short time period, it is impossible to analyze the curve by distinguish the explicit gating values. In this case, we need autocorrelation analysis. The basic idea of autocorrelation analysis is by comparing the similarity of the curve at time  $t$  and  $(t+T)$  to acquire the properties of the target. In our experiment, we dealt with the current fluctuation around the average current:

$$x(t) = I(t) - \langle I \rangle \quad (1.6)$$

Thus, we had the following correlation function:

$$C(T) = \langle x(t) \cdot x(t+T) \rangle = \langle x^2 \rangle \cdot e^{-T/\tau} = \langle (I(t) - \langle I \rangle)^2 \rangle e^{-T/\tau} \quad (1.7)$$

If the gramicidin concentration is relatively low, the relaxation time  $\tau$  should equal to the  $1/k_d$ , where  $k_d$  is the dissociation rate of the gramicidin dimer. In this way, even we could not identify the individual channel in the signals, we can also acquire the dissociation rate.

### 1.4 Black lipid membrane

In order to embed the gramicidin ion channels, we used the 'Müller-Rudin' method to construct an artificial membrane. The lipid( mono-olein) is dissolved in the long chain hydrocarbon solvent(tetra-decane) at a concentration of 3,5 mg/ml . In the practice, a drop of lipid oil is attached to the opening of a coated hydrophobic glass capillary. The oil would slowly immerse along the hydrophobic glass capillary, leaves the membrane at the entrance becoming thinner. At the beginning, a newton ring could be observed due to the interference of the light reflected from the front and back surface of the membrane. When the membrane is much thinner than the wavelength, interference between rays reflected on the 1° and 2° surface is no longer possible, which means the lipid will be black, which is called black lipid membrane.

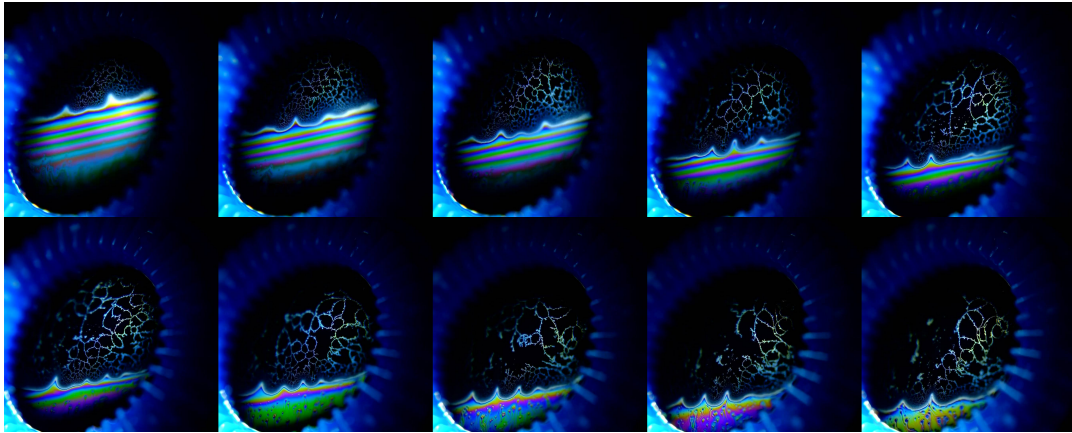


Figure 1.3: Timeseries of the membrane thinning process of a soap bubble membrane, photographed by Nicolae Turcan.  $t = 20s$

## 2 Experimental Results and Problems

### 2.1 Problems

#### 2.1.1 Preparation and characterisation of a lipid membrane

The preparation of the lipid membrane is described in the introductory part, and a more comprehensive description of the building process is provided in the publication by Müller and Rudin. [5]

**Electrical and optical observation of the membrane building process** The successful construction of a membrane can be confirmed in two independent manners.

The first is based on the change in electrical behaviour of the system( described in Figure 2.1a ), where an Wave signal current is applied trough the electrodes, and if no insulator is present at the end of the capillary, the applied frequency is basically the same as the measured one ( with a negligible shift in phase), On the other hand , if a lipid membrane is formed at the end of the capillary , acting as an insulator, we have basically constructed a capacitor and the measured frequency is changed as described in Fig 2.1b making the electrical setup essentially a rectifier.

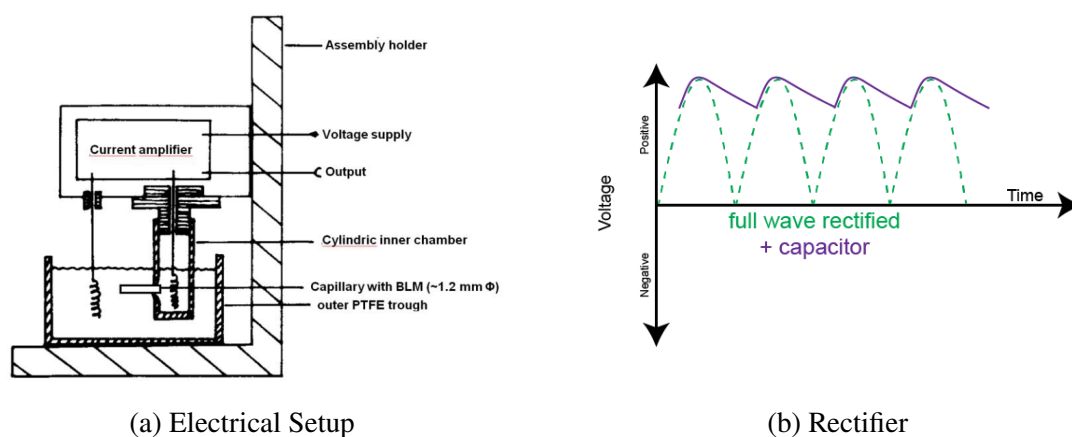


Figure 2.1: Electrical Observation

Another manner for determining if a membrane was successfully built is trough visual inspection of the edge of the capillary trough a microscope. The optical set-up consists in focusing the microscope to the edge of the capillary and setting it at a certain angle relative to the perpendicular axis of the membrane, and then illuminating the same sample from a

specular position relative to the perpendicular axis, this enables us to see the reflection of the light source on the sample trough the objective of the microscope.

If the optical setup is correctly tweaked and a membrane is formed at the tip of the capillary tube, then a characteristic optical phenomenon called *Newton's Rings* can be observed in the first minutes since the formation of the surface. After a few minutes the surface undergoes the Black Membrane phase and the effect presented in Figure 2.1 is no longer visible.

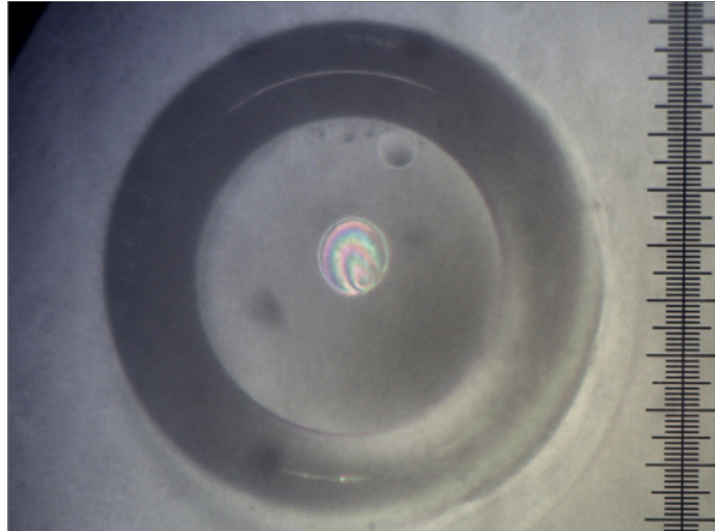


Figure 2.2: Optical Observation

**Determine the capacity, specific capacity and thickness of the membrane** To determine the capacitance of this newly formed capacitor consisting of the black lipid membrane and the 2 water cells, we used the oscilloscope in combination with a known value resistor (100k $\Omega$  and a piece of software allowing us to extract  $\tau$  ( RC time constant ) which is equal to the time required to discharge our capacitor to  $e^{-1}$  trough the known resistor. In Figure 3.2 the green cursor is set to 36.8% of the maximum current and we get the corresponding value for tau of 0.0480 ms.

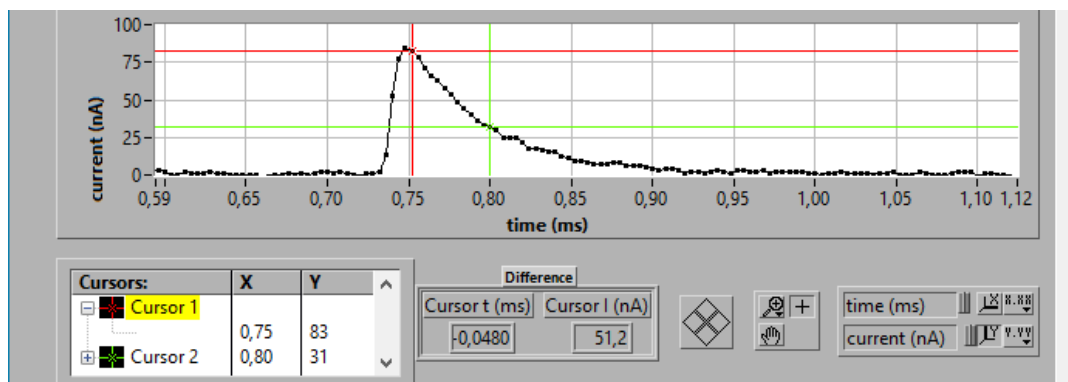


Figure 2.3: membrane Capacitance Software and  $\tau$  at 10V, 250kHz sampling rate, 1000 samples

Using formula 2.1, to solve for capacitance  $C$  with the values described above we get a final capacitance of 0.48 nF (nanoFarad)

$$C = \frac{\tau}{R} \quad (2.1)$$

By using the formula for a parallel plate capacitor (see formula 2.2), which well fits with the bi-dimensionality of our lipid bilayer, the calculated capacitance of 0.48nF , an Area (computed thanks to the scale in Figure 2.2) of 0.204mm<sup>2</sup> , the permittivity of vacuum  $\epsilon_0$  of 8.854 x 10<sup>-12</sup> F/m and the permittivity of oil  $\epsilon$  of 2 F/m ( since a reference value for mono-olein is not published, even tough the chemical constituents are oleic acid and glycerol with 3 and 46.5  $\epsilon$  respectively); we can therefore solve for  $d$  .

$$C = \epsilon * \epsilon_0 \frac{A}{d} \quad (2.2)$$

In conclusion the thickness ( $d$ ) of the membrane was established to be 7.52nm, the absolute error of this value was not computed but it should be noted that the order of magnitude corresponds with the one reported in other publications [2] and we believe this is a notable feat considering that only macroscopic properties were recorded during the experiment.

#### **Determine the membrane resistance and specific membrane conductivity**

To determine the membrane resistance and specific membrane conductivity, we applied a voltage of 10V across the membrane and measured the resulting current of the membrane to be 83nA . The current-voltage ( $I$ - $V$ ) relationship was recorded for the membrane.

From the  $I$ - $V$  relationship, the membrane resistance ( $R_m$ ) was calculated using Ohm's Law:

$$R_m = \frac{V_m}{I_m} = 12.048M\Omega \quad (2.3)$$

where  $V_m$  is the applied membrane voltage and  $I_m$  is the measured current.

The specific membrane conductivity ( $\sigma_m$ ) is the reciprocal of resistivity. To find  $\sigma_m$ , we used the membrane area ( $A_m$ )of 0.204mm<sup>2</sup> and thickness ( $d_m$ ) of 7.52nm :

$$\sigma_m = \frac{1}{R_m} \cdot \frac{d_m}{A_m} = 2.25\mu S/m \quad (2.4)$$

#### **Determine the breakdown voltage and electric field in the membrane**

To determine the breakdown voltage ( $V_b$ ), we should have incrementally increased the voltage applied across the membrane. We should have recorded the voltage at which a sudden, significant increase in current was observed, indicating membrane breakdown. This voltage is identified as the breakdown voltage.

The electric field ( $E$ ) in the membrane is calculated using the measured breakdown voltage and the thickness of the membrane ( $d_m$ ). The electric field is given by:

$$E = \frac{V_b}{d_m} \quad (2.5)$$

where  $V_b$  is the breakdown voltage and  $d_m$  is the membrane thickness.



Our membrane broke down because of operation mistakes at a relatively low voltage ( 50mV) which is far away from typical voltages expected from literature (300 mV) [1], this led us to the conclusion that our membrane broke because of other reasons instead of reaching the actual break down voltage. For this reason we cannot provide an Experimental result.

### 2.1.2 Single channel measurements

After observing the black lipid membrane and the exponential wave decay on the monitor, we changed the resistors from 100 k $\Omega$  to 500 k $\Omega$  to 100 M $\Omega$  and 500 M $\Omega$  to observe the pA scale of tiny current. The voltage was also changed to DC, 50mV for the following experiment. Then 5 ml of gramicidin solution was suddenly injected around the membrane to form the ion channels. Then the software was used to record the current pattern.

#### Modify the conductivity of the membrane by very low doping with gramicidin

According to the equation 1.4, we could determine the conductivity under the single channel analysis ( $N_p(t) = 1$ ):

$$\Lambda = \frac{I_0}{V_M} \quad (2.6)$$

Where  $I_0$  is the current through the single channel. The current of the single channel is 1.7 pA (see next question), thus we could acquire the conductivity  $\Lambda = 3.4 \times 10^{-11} \Omega^{-1}$ , which is close to the published value  $3.0 \times 10^{-11} \Omega^{-1}$ . [3]

#### Determine the single-channel current and analyze the distribution of channel lifetimes

The single channel current could be observed from the single-channel signals under low gramicidin concentration. Here the leaking current of the membrane should be considered, since it has a clearly influence to the measurement. In Figure 2.1, we could calculate that the single channel current is 3.4 pA (total current) - 1.7 pA (leaking current) = 1.7 pA (single channel current).

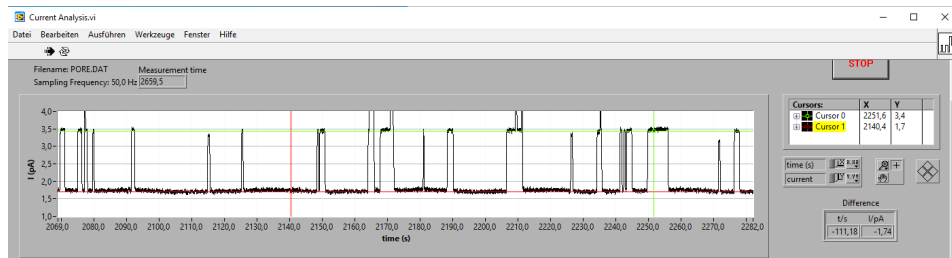


Figure 2.4: The current-time plot of the single channel signals.

Figure 2.2 tells the opening time distributions of the single channels. It follows the Poisson interval distribution. By renormalizing and fitting the curve, we could acquire the dissociation time of the gramicidin dimer ( $\lambda = \tau = 0.927s$ ,  $k_D = 1/\tau = 1.078$ ), which is close to the published value ( $\tau = 1.3s$ ) [4].

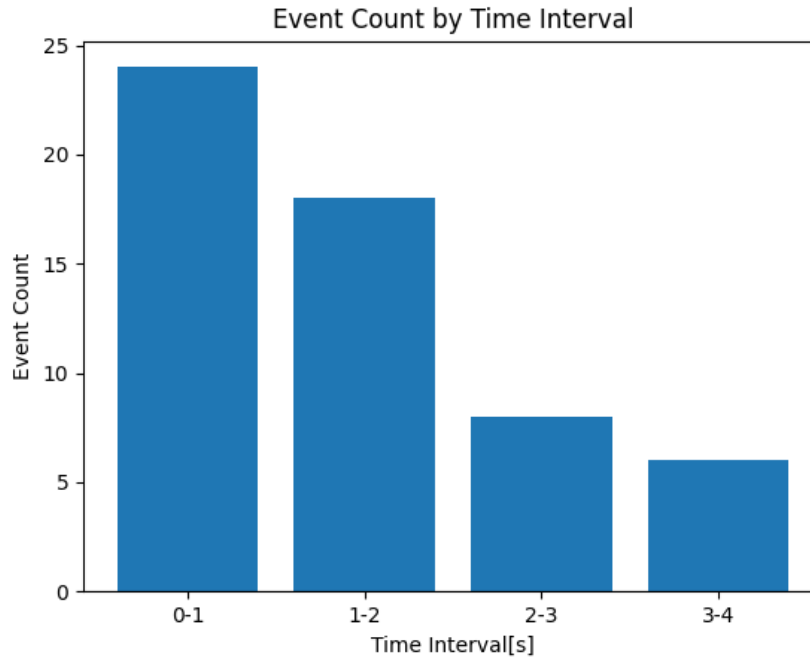


Figure 2.5: The opening time distributions of the single channels

With this dissociation time, we could estimate how many ions were transferred through the channel during the lifetime of the channel:

$$n = \frac{I \cdot \tau}{q} = 9.84 \times 10^6 = 1.63 \times 10^{-17} \text{ mol} \quad (2.7)$$

Here the  $q$  is the charge of a single  $\text{K}^+ = 1.6 \times 10^{-19} \text{ C}$ .

**Analyze the single-channel current at different membrane voltages. Is Ohm's law valid for single-channel currents?**

Unfortunately, in the experiment, we did not have time and chance to measure the single channel current at different membrane voltages.

### 2.1.3 Measurements with multiple ion channels

**Increase the gramicidin concentration in the membrane, so that several channels are open at any time and multiple current steps are visible in the signal**

In the experiment, after a short time of the gramicidin injection, we observed that the current curve (Figure 2.3) changed from single steps to multi-steps, which indicated that more gramicidin was embedded in the membrane and formed a dimer channel.

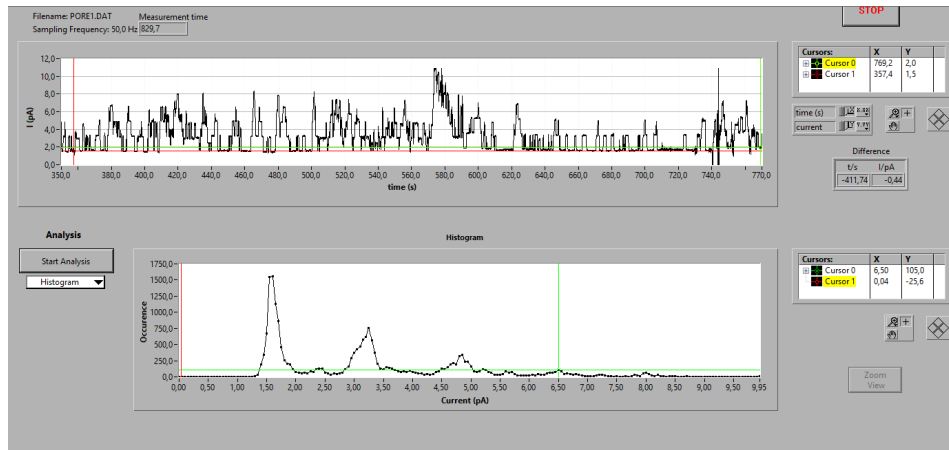


Figure 2.6: The upper side is the current-time plot under many channels open and close simultaneously. The lower side is the histogram of different currents.

### Determine the single-channel current by analysis of the current histogram

As is shown in the Figure 2.3, we could count the distribution of the different current levels. We could notice there are four major peaks, with current intensity 1.5 pA (leaking current), 3.3 pA (single channel current), 4.8 pA (double channel current) and 6.5 pA (triple channel current).

#### 2.1.4 Current fluctuation measurements

**Increase the gramicidin concentration even more, so that many channels are open at any time and current steps in the signal are not visible any more**

Theoretically, we should further increase the concentration of the gramicidin and observe a curve as the Figure 2.4.

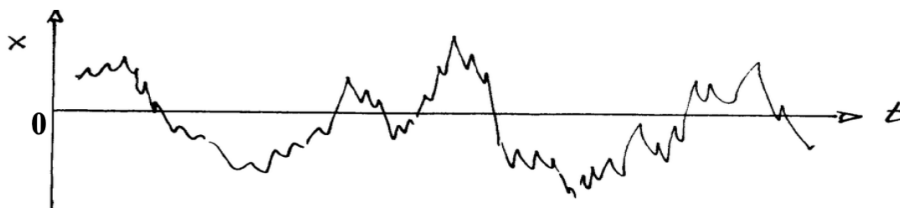


Figure 2.7: The sketch of the ideal curve shape for auto correlation analysis

But after multiple additions of gramicidin, we couldn't achieve a concentration of channels in the membrane high enough to produce a plot similar to the one presented above in figure 2.7, this resulted in insufficient data for the autocorrelation analysis which greatly limited the results presented in Figure 2.8.

### Determine the single-channel current and the average channel lifetime by autocorrelation analysis of the fluctuating current

During our experiment, the membrane broke before we could observe such a curve. As compensation, we applied the autocorrelation analysis on the curve so we could still distinguish multi-channels (Figure 2.5).

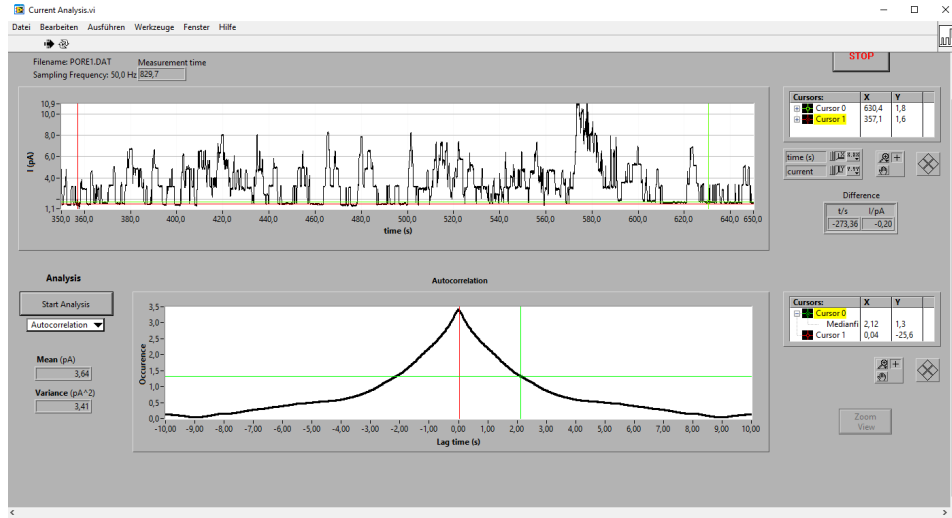


Figure 2.8: Autocorrelation results of the current-time curve

Using equation 1.7, we could acquire the relaxation time of the dimer  $\tau = 2.17$  s, and the dissociation time  $k_D = 1/\tau = 0.46/\text{s}$ .

# Bibliography

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