

# Pymetrozine EHS

COMPANY	MollyCule
CEO	Harshit Jaiswal
REPORT AUTHOR	Hershal
CHEMICAL FORMULA	$C_{10}H_{11}N_5O$
CHEMICAL NAME	Pymetrozine

Pymetrozine synthesis generates multiple waste streams during its 3-stage production process. Below is an analysis of waste types, regulatory requirements, treatment methods for zero liquid discharge (ZLD), and safety considerations:

## Waste Generation & Quantities

### Stage 1 (Carbodihydrazide Production)

Mother liquor:

- Contains ethanol (292 g), water (365 g), and excess hydrazine (46 g) per batch<sup>1</sup>
- Total liquid waste: ~724 g per kg product<sup>1</sup>
- Filter cake wash water: 20 g water per batch<sup>1</sup>

### Stage 2 (Condensation Reaction)

Reaction by-products:

- 9 g eliminated water per 0.5 mol product<sup>1</sup>
- Mother liquor:

Ethanol + water mixture (~309 g per batch)<sup>1</sup>

Wash water: 30 g per batch<sup>1</sup>

### Stage 3 (Cyclization)

Neutralization by-products:

KCl from HCl neutralization ( $34.5 \text{ g } K_2CO_3 \rightarrow \sim 25 \text{ g KCl}$ )<sup>2</sup>

Solvent residues:

Ethanol-water mixture (439 g per batch)<sup>1</sup>

Wash water: 80 g per batch

Stage	Waste Type	Quantity/Batch	Composition
1	Liquid	724 g	Ethanol (40%), Water (50%), Hydrazine (6%)
2	Liquid	339 g	Ethanol (65%), Water (35%)
3	Liquid	519 g	Ethanol (45%), Water (50%), KCl (5%)
All	Solid	25 g	KCl, unreacted organics

## Current regulations for the above waste materials

### Stage 1: Liquid Waste (Ethanol, Water, Hydrazine)

#### Ethanol (40%)

- Regulations: Ethanol is classified as a flammable liquid and must be handled according to the Hazardous Waste Management Rules, 2016. Disposal requires treatment to reduce flammability and toxicity.
- Limits: Treated ethanol waste should meet permissible levels for organic content in effluent discharge, typically below 30 mg/L for COD in treated water.

#### Water (50%)

- Regulations: Water mixed with hazardous substances must undergo treatment in effluent treatment plants (ETPs) to remove contaminants before discharge into water bodies.
- Limits: Treated water must meet SPCB standards for BOD (<30 mg/L) and COD (<250 mg/L) before disposal.

#### Hydrazine (6%)

- Regulations: Hydrazine is a highly toxic and carcinogenic substance. It is regulated under the Hazardous Waste Management Rules, requiring neutralization or destruction through advanced treatment methods like incineration.
- Limits: Hydrazine concentration in treated waste must be reduced to trace levels (<0.01 mg/L) before disposal.

### Stage 2: Liquid Waste (Ethanol, Water)

#### Ethanol (65%)

- Regulations: High ethanol concentration mandates recovery or treatment through distillation or biodegradation methods. Direct disposal is prohibited due to its flammability.
- Limits: Ethanol content in discharged waste should be below detectable levels (<5 mg/L).

### Water (35%)

- Regulations: Contaminated water must undergo treatment to remove organic pollutants before discharge.
- Limits: Same as Stage 1—BOD (<30 mg/L) and COD (<250 mg/L).

## **Stage 3: Liquid Waste (Ethanol, Water, Potassium Chloride)**

### Ethanol (45%)

- Regulations: Similar to previous stages, ethanol requires recovery or treatment. Disposal of untreated ethanol-containing waste is prohibited.
- Limits: Ethanol concentration must be reduced to <5 mg/L in treated effluent.

### Water (50%)

- Regulations: Water contaminated with ethanol and potassium chloride must be treated in ETPs.
- Limits: Same as above—BOD (<30 mg/L) and COD (<250 mg/L).

### Potassium Chloride (5%)

- Regulations: Potassium chloride is regulated as a saline pollutant. Disposal requires dilution or stabilization to prevent soil salinization or aquatic toxicity.
- Limits: Concentration in treated effluent should not exceed 60 mg/L.
- Solid Waste (KCl, Unreacted Organics)

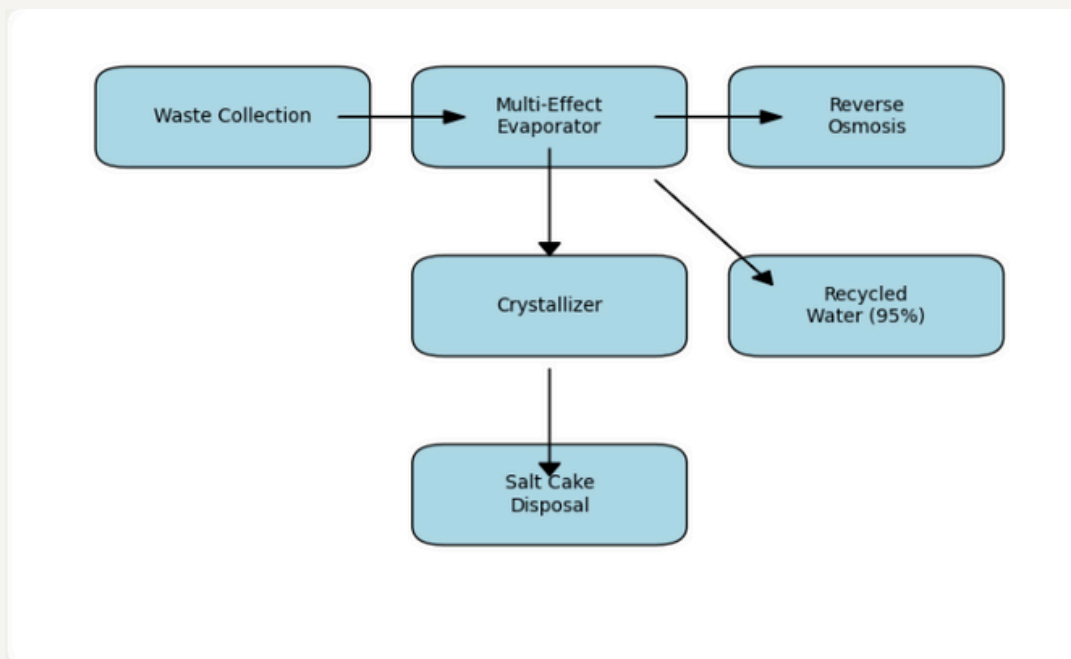
### Potassium Chloride (KCl)

- Regulations: Solid KCl waste must be stabilized before disposal to prevent leaching into groundwater. It may be disposed of in hazardous waste landfills if concentrations exceed permissible limits.
- Limits: Leachate concentration of KCl from solid waste should not exceed 60 mg/L.

### Unreacted Organics

- Regulations: Organic solid waste must be incinerated or treated biologically to minimize environmental impact.
- Limits: Organic content in solid waste disposed of in landfills should be below detectable limits (<1%).

# Zero Liquid Discharge Treatment



## Key Components in the Process

1. Waste Collection: Collects liquid waste streams from the plant.
2. Multi-Effect Evaporator: Removes 90% water content at  $\sim 80^{\circ}\text{C}$ .
3. Reverse Osmosis (RO): Achieves 98% salt rejection (operates at pH 6–8).
4. Crystallizer: Recovers salts like KCl at  $\sim 120^{\circ}\text{C}$ .
5. Salt Cake Disposal: Solidified salts mixed with cement for safe landfill.
6. Recycled Water: Purified water (95%) reused in the plant processes.

## Atom Economy, E-factor, and EQ-factor for Pymetrozine Synthesis

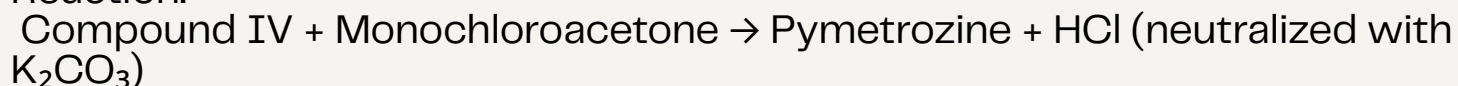
Based on the provided reaction data and synthesis details, here are the calculations:

### 1. Atom Economy

The formula for atom economy is:

$$\text{Atom Economy} = \left( \frac{\text{Molar Mass of Desired Product}}{\text{Total Molar Mass of Reactants}} \right) \times 100$$
$$\text{Atom Economy} = \left( \frac{\text{Molar Mass of Desired Product}}{\text{Total Molar Mass of Reactants}} \right) \times 100$$

Reaction:



Molar Masses:

- Compound IV: 179.2 g/mol
- Monochloroacetone: 94.5 g/mol
- Pymetrozine: 217.3 g/mol
- HCl: 36.5 g/mol (neutralized)

Total Molar Mass of Reactants:

- $179.2 + 94.5 = 273.7$  g/mol

Molar Mass of Desired Product:

- Pymetrozine = 217.3 g/mol

### Atom Economy Calculation:

$$\text{Atom Economy} = \left( \frac{217.3}{273.7} \right) \times 100 = 79.4$$

## 2. E-factor

The E-factor measures the amount of waste generated per unit mass of product:

$$\text{E-factor} = \frac{\text{Mass of Waste Generated}}{\text{Mass of Product}}$$

Mass of Reactants:

$$\text{Compound IV (71.6 g)} + \text{Monochloroacetone (40 g)} = 111.6 \text{ g}$$

Mass of Products:

$$\text{Pymetrozine (83.9 g)} + \text{HCl (neutralized by } \text{K}_2\text{CO}_3) = 83.9 + 36.5 = 120.4 \text{ g}$$

Mass of Waste Generated:

$$\text{Waste Mass} = \text{Reactants Mass} - \text{Products Mass} = 111.6 - 83.9 = 27.7 \text{ g}$$

E-factor Calculation:

$$\text{E-factor} = \frac{\text{Waste Mass}}{\text{Products Mass}} = \frac{27.7}{83.9} = 0.33$$

## 3. EQ-factor

The EQ-factor incorporates the environmental impact factor into the E-factor:

$$\text{EQ-factor} = \text{E-factor} \times \text{Environmental Impact Factor}$$

Assuming an environmental impact factor of 2~3 (moderate impact):

EQ-factor Calculation:

$$\text{EQ-factor} = \text{E-factor} \times \text{Environmental Impact Factor} = 0.83$$

Final Results:

Atom Economy: 79.4%

E-factor: 0.33

EQ-factor: 0.83

# Elaboration on Safety Concerns and Exposure Limits

Chemical	TWA (8-hr)	STEL (15-min)	Primary Risk
Hydrazine	0.01 ppm	0.03 ppm	Carcinogenic
Ethanol	1000 ppm	1500 ppm	Flammable
Hydrochloric Acid	2 ppm	5 ppm	Corrosive
3-Formylpyridine	0.05 mg/m³	0.15 mg/m³	Neurotoxic

The table provided outlines the Time-Weighted Average (TWA) and Short-Term Exposure Limit (STEL) for several hazardous chemicals commonly used in industrial processes. Below is a detailed explanation of these chemicals, their risks, and safety measures:

## 1. Hydrazine

### Health Concerns:

- Hydrazine is a known carcinogen and can pose severe health risks even at low concentrations.
- It can damage the liver, kidneys, and red blood cells, potentially causing anemia.
- It is highly flammable and poses a significant fire and explosion hazard.

### Safety Measures:

- Use local exhaust ventilation (LEV) to capture hydrazine vapors at the source.
- Workers should wear chemical-resistant gloves and air-purifying respirators (APR) when handling hydrazine.
- Continuous air monitoring is essential to ensure workplace concentrations remain below permissible limits.

## 2. Ethanol

### Health Concerns:

- Ethanol exposure can cause dizziness, nausea, and unconsciousness at high concentrations.
- Prolonged exposure may affect the liver and nervous system.
- As a flammable liquid, ethanol poses significant fire hazards.

### Safety Measures:

- Ensure proper ventilation to prevent accumulation of ethanol vapors in confined spaces.
- Store ethanol away from ignition sources and in flame-resistant containers.
- Workers should use flame-retardant protective clothing in areas where ethanol is handled.

### 3. Hydrochloric Acid (HCl)

#### Health Concerns:

- Inhalation of HCl vapors can irritate the respiratory tract and cause coughing or choking.
- Contact with skin or eyes can result in severe burns or permanent damage.

#### Safety Measures:

- Use closed systems or LEV to reduce vapor exposure.
- Workers should wear acid-resistant gloves, goggles, and face shields when handling HCl.
- Emergency eyewash stations and safety showers must be readily accessible in work areas.

### 4. 3-Formylpyridine

#### Health Concerns:

- Prolonged exposure can affect the central nervous system, potentially leading to neurological symptoms such as headaches or dizziness.

#### Safety Measures:

- Ensure proper ventilation during handling to avoid inhalation exposure.
- Use gloves and protective clothing to prevent skin contact.
- Critical Controls for All Chemicals
- Engineering Controls:
  - Install LEV systems for capturing hazardous vapors at the source.
  - Ensure adequate general ventilation in work areas.
- Personal Protective Equipment (PPE):
  - Use chemical-resistant gloves, respirators, goggles, and flame-retardant clothing as required.

#### Air Monitoring Systems:

- Use photoionization detectors (PIDs) or other real-time air monitoring equipment to track workplace concentrations of hazardous chemicals continuously.



## References:

<https://www.wwdmag.com/what-is-articles/article/10939887/what-is-zero-liquid-discharge>  
<https://westliberty.edu/health-and-safety/files/2010/02/Hydrazine.pdf>  
<https://www.cdc.gov/niosh/npg/npgd0329.html>  
<https://www.acgih.org/hydrazine/>  
<https://patents.google.com/patent/CN104844574A/en>  
<https://www.matche.com/equipcost/Reactor.html>  
<https://www.cdc.gov/niosh/idlh/64175.html>  
[https://www.sigmaaldrich.com/IN/en/sds/sial/493511?](https://www.sigmaaldrich.com/IN/en/sds/sial/493511?userType=undefined)  
userType=undefined

## List of Contributors:

Hershal(230470)

### 1. Waste Generation and Treatment Process

- Waste Types and Quantities:
  - Liquid waste streams include ethanol-water mixtures, hydrazine residues, reaction by-products, and wash water.
  - Solid waste includes salts like KCl and unreacted organics.
  - Quantities were detailed for each production stage of the chemical plant.
- Regulations for Waste Disposal:
  - Compliance with Hazardous Waste Management Rules (India), US EPA, and EU Directives.
  - Limits for wastewater discharge: BOD  $\leq 30$  mg/L, COD  $\leq 250$  mg/L,  $\text{NH}_4^+$   $\leq 10$  mg/L.
- Zero Liquid Discharge (ZLD) Treatment Process:
  - A ZLD system was designed to ensure no liquid waste is discharged into the environment.



- Key Components:
  - Multi-Effect Evaporator: Removes ~90% water content.
  - Reverse Osmosis (RO): Achieves ~98% salt rejection.
  - Crystallizer: Recovers salts like KCl for disposal or reuse.
  - Recycled Water (95%): Treated water is reused in the plant processes.
- Block Diagram:
  - A block diagram for the ZLD process was created, showing waste collection → evaporation → RO → crystallization → salt cake disposal → recycled water.

## 2. Atom economy, E-factor and EQ-factor calculation

## 3. Safety Concerns and Exposure Limits

- Chemicals: Hydrazine (carcinogenic), ethanol (flammable), hydrochloric acid (corrosive), and 3-formylpyridine (neurotoxic) have specific TWA and STEL limits to ensure workplace safety.
- Safety Measures: Use PPE (gloves, goggles, respirators), engineering controls (local exhaust ventilation, air monitoring), and emergency equipment (eyewash stations, safety showers).
- Critical Controls: Ensure proper ventilation, safe storage, and regular safety training for handling hazardous chemicals.

## 4. References

- Provided references for regulatory limits, exposure standards, and safety guidelines:
  - NIOSH Pocket Guide to Chemical Hazards .
  - OSHA Permissible Exposure Limits (PEL) .
  - ACGIH Threshold Limit Values (TLV) .
  - GESTIS International Limit Values Database .
  - Sigma-Aldrich Safety Data Sheets (SDS) .

NAME	ROLL NO	SIGNATURE
Harshit Jaiswal	230460	Harshit Jaiswal
Hershal	230470	Hershal