## Thermodynamics - Dr. Susmita Roy

\* Mathematical Interlude; Some simple calculus are often used in thermodynan -al treatments. It may be worth to get familiar with those mathematical ingradients which provide us with some elementary comfort during owe study of thermodyne

## 1. Partial Derivatives o-

Let a quantity 'Z' is a function of two independent variables of 'k' and 'y' i.e. Z = f(x, y). If the coordinates of k and y charge by a small amounts dx f dy then the change in the value of 'Z' is given by

$$dz = \left(\frac{\partial z}{\partial R}\right)_{Y} dR + \left(\frac{\partial z}{\partial Y}\right)_{R} dY$$

where  $\left(\frac{\partial z}{\partial \kappa}\right)$  is the change of 'z' for unit change of 'k' at constanty; Similarly,  $\left(\frac{\partial z}{\partial y}\right)_{k}$  is the change of 'z' for unit change of 'y' at constant 'k'.

Let us suppose the three quantities x, y, z are scaleted as f(x,y,z) = 0; So, we have  $x = \psi(y,z) \neq y = \varphi(z,x)$ 

Therefore, we can write, dr= ( dr ) dy + ( dr ) dz . Dand

dy = ( dx ) dz + ( dx ) dx --- (2)

Substituting (2) in (1), we have,  $d\kappa = \left(\frac{\partial \kappa}{\partial y}\right)_z \left[\left(\frac{\partial y}{\partial z}\right)_R dz + \left(\frac{\partial y}{\partial \kappa}\right)_z d\kappa\right] + \left(\frac{\partial \kappa}{\partial z}\right)_R dz - \left(\frac{\partial \gamma}{\partial z}\right)_R dz + \left(\frac{\partial \gamma}{\partial \kappa}\right)_z d\kappa\right] + \left(\frac{\partial \kappa}{\partial z}\right)_R dz - \left(\frac{\partial \gamma}{\partial z}\right)_R d\kappa$  $= \left(\frac{\partial x}{\partial y}\right)^2 \left(\frac{\partial y}{\partial x}\right)^2 dx + \left[\left(\frac{\partial x}{\partial y}\right)^2 \left(\frac{\partial y}{\partial z}\right)^2 + \left(\frac{\partial x}{\partial z}\right)^2\right] dz$ 

Eq 3 and 3' are generally true for all values of x, y, 2. (3') Considering x and Z as independent variables in Eq. 3 &3'= a) Let's suppose dz = 0, but dx = 0 then

$$\frac{d\kappa = \left(\frac{\partial \kappa}{\partial y}\right)_{z} \left(\frac{\partial \gamma}{\partial x}\right)_{z}}{d\kappa} = 1. \Rightarrow \left(\frac{\partial \kappa}{\partial y}\right)_{z} = \left(\frac{\partial \gamma}{\partial x}\right)_{z} = \frac{1}{\left(\frac{\partial \gamma}{\partial x}\right)_{z}}$$

b) Let us take 
$$dz \neq 0$$
, but  $dk = 0$ , then

$$\left(\frac{\partial x}{\partial t}\right)_{z} \left(\frac{\partial y}{\partial x}\right)_{x} + \left(\frac{\partial z}{\partial z}\right)_{y} = 0$$
or,  $\left(\frac{\partial x}{\partial y}\right)_{z} \left(\frac{\partial y}{\partial z}\right)_{x} = -\left(\frac{\partial x}{\partial z}\right)_{y}$ 

As we can write,  $\left(\frac{\partial x}{\partial z}\right)_{x} = \frac{1}{2z}$ 

Hence,  $\left(\frac{\partial x}{\partial y}\right)_{z} \left(\frac{\partial z}{\partial z}\right)_{x} \left(\frac{\partial z}{\partial x}\right)_{y} = -1$ 

This is called cyclic rule.

A proof of the ryclic relation can be shown in the following on natural variables,  $p, y, T$ .

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we know  $py = RT$ 

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here,  $f(p, y, T) = 0$  and  $\left(\frac{\partial p}{\partial y}\right)_{T} = -\frac{RT}{y} \cdot \left(\frac{\partial y}{\partial T}\right)_{T} = \frac{R}{R}$ 

Thus,  $\left(\frac{\partial p}{\partial y}\right)_{T} \left(\frac{\partial y}{\partial T}\right)_{P} \left(\frac{\partial T}{\partial P}\right)_{Y} = -1$  (Proved)

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Let us consider a quantity 'z' cohose value is determined solely by two other variables x and y at any moment, in any given state, that is, if x fy are given a particular value of 'z' is though fixed. If x fy vary, then z will also the value of 'z' is though fixed. If x fy vary, then z will also the vary. Thus, mathematically, if we consider Z = f(x,y), the vary. Thus, mathematically, if we consider the derivative of change of z can be estimated provided the derivatives on the function z correct x and y care known. Formal derivatives on the function z correct x and y care known. Independent variable variable

If kindy both change simultaneously, the total change is expussion, and the desired and the simultaneously, the total change is expussion, and the desired and

 $\Rightarrow dZ = M(x, \frac{1}{2}) dx + N(x, \frac{1}{2}) d\frac{1}{2}$ where  $M(x, \frac{1}{2}) = \left(\frac{\partial Z}{\partial x}\right)_{x}$  and  $N(x, \frac{1}{2}) = \left(\frac{\partial Z}{\partial y}\right)_{x}$ 

For a differential M(n,y)dx + N(x,y) dy to be exect, it must satisfy the fall is \* Conditions for Exactness: it must satisfy the following relation (which is mathematically known as Schwarz's theorem; also called the cross partial derivative test):

\[
\frac{\partial}{\partial} = \frac{\partial}{\pa => This ensures M, N come from a common potential
function Z = f(x,y). Example: - Consider the differential expression: dz = (2xy) dx + (xx+3yx) dy Here, M = 2ky and H = x +3y Performing cross bantial derivative test: DM = D (20y) = 2K 3H = d (x+3y) = 2x Since,  $\frac{\partial M}{\partial Z} = \frac{\partial H}{\partial Z}$ , the differential 'dz' is exact.

This means there exists a function H Z (\*17) such that  $\frac{\partial Z}{\partial x} = 2xy \quad ; \quad \frac{\partial Z}{\partial y} = x^2 + 3y^2$ Integrating  $\frac{\partial u}{\partial x}$  co.v.t. x: Z(n,y)= \ \ 2 my dn = my + 9(y) To determine g(y), differentiate a.v.ty! 347 = x+ 9"(y) = x+3y Comparing L.H.S & R.H.S. we get get g(y)= y3 Therefore Z(x,y)= xy+y3 \* A differential is perfect (exact) if it derives from a function 107(x,y)

\* If the exactness condition is  $\frac{\partial M}{\partial t} = \frac{\partial N}{\partial x}$ \* If exact, we can find Z(x,y) by integration.