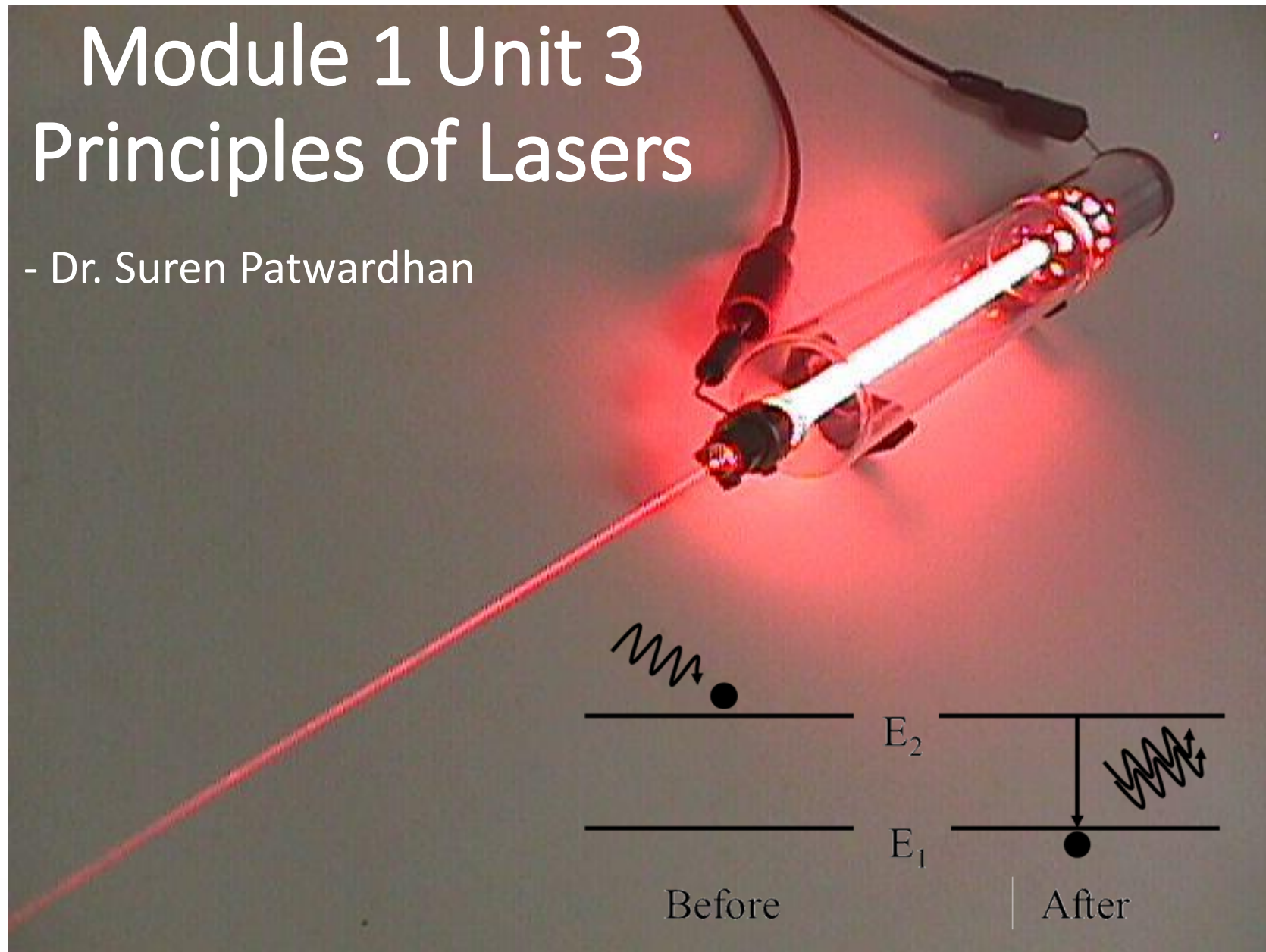


Module 1 Unit 3

Principles of Lasers

- Dr. Suren Patwardhan



LOSER Changed to LASER

- First discovered by Maiman 1960
- Once teased as discovery in search of applications
- Today, there are hundreds of applications:
 - CO₂ laser used in cosmetics
 - Nd:YAG laser used in machining
 - Ar laser used in surgery
 - Diode laser used in communications
 - Dye lasers used in spectroscopy
 - Nd:Glass laser used in nuclear fusion

Properties of LASER

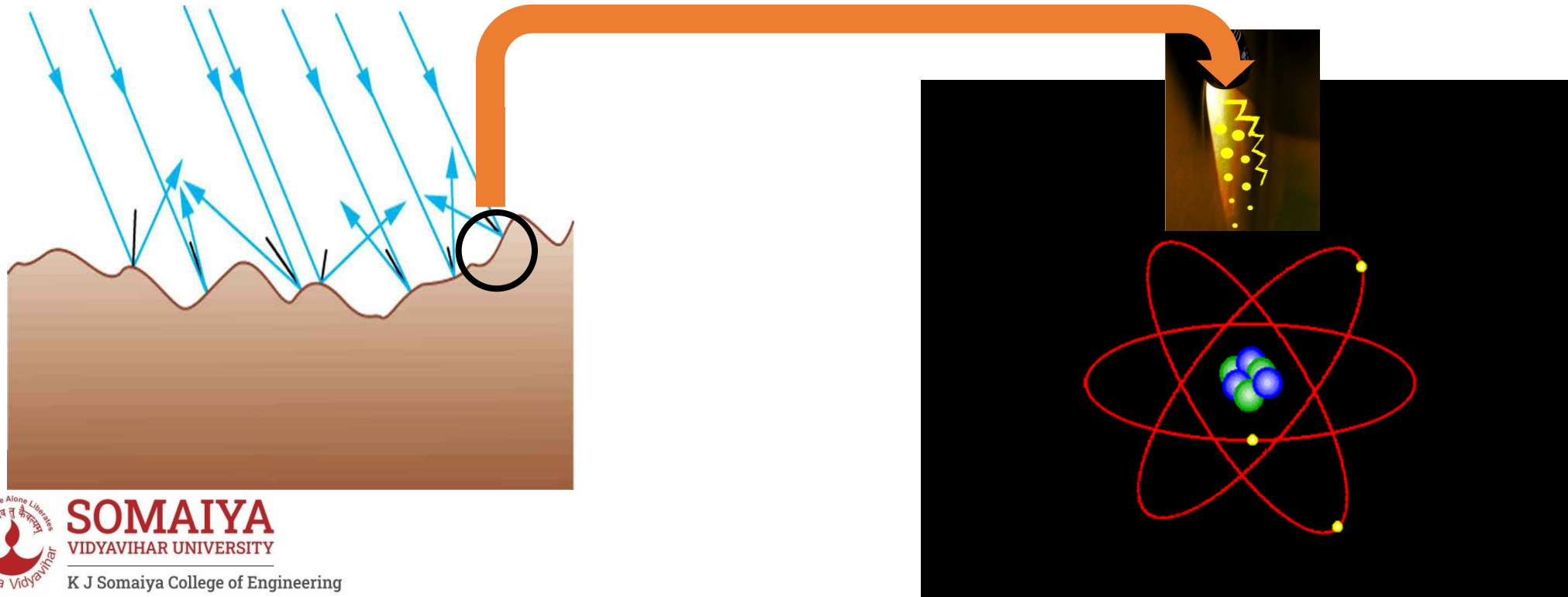
- Monochromatic
- Coherent
- Directional
- Focused
- Bright
- Polarized (optional)

meaning

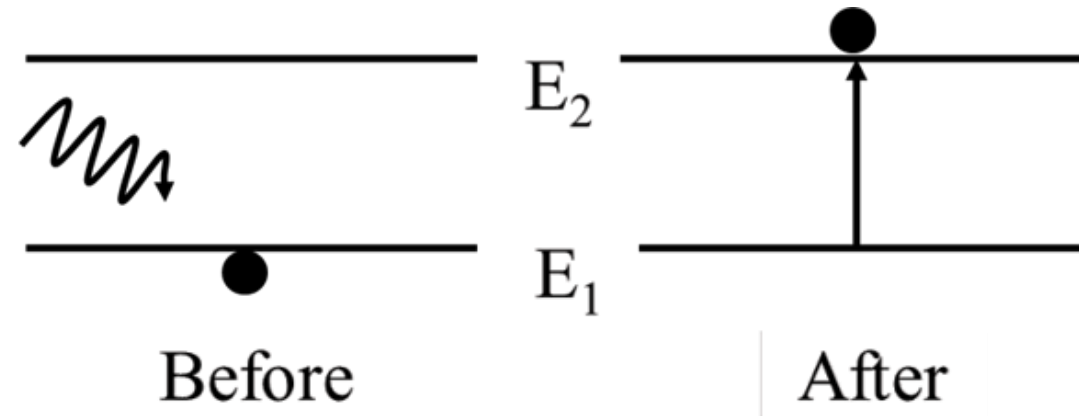
- Identical wavelength
- In phase
- Same path
- Low divergence
- High intensity
- E-M vibrations in same plane

Interaction of Radiation with Matter

1. Absorption
 - Energy supplied
2. Spontaneous emission
 - Energy emitted
3. Stimulated emission
 - Energy emitted



Absorption i.e. Stimulated Absorption



Rate of transition

$$\left. \frac{dN}{dt} \right|_{ab} = B_{12} N_1 Q$$

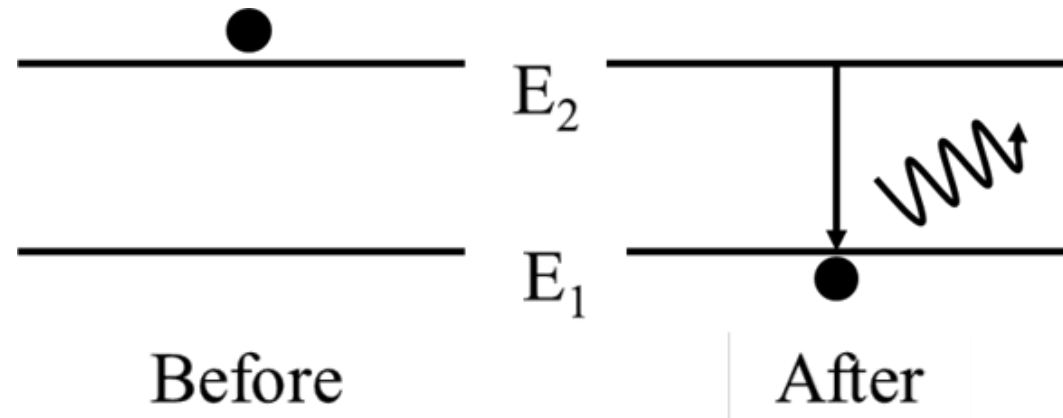
Where,

B_{12} : probability of absorption process

N_1 : number of atoms in lower energy level E_1

Q : energy density of incident radiation per unit frequency (J-s/m^3)

Spontaneous i.e. Natural Emission



Rate of transition

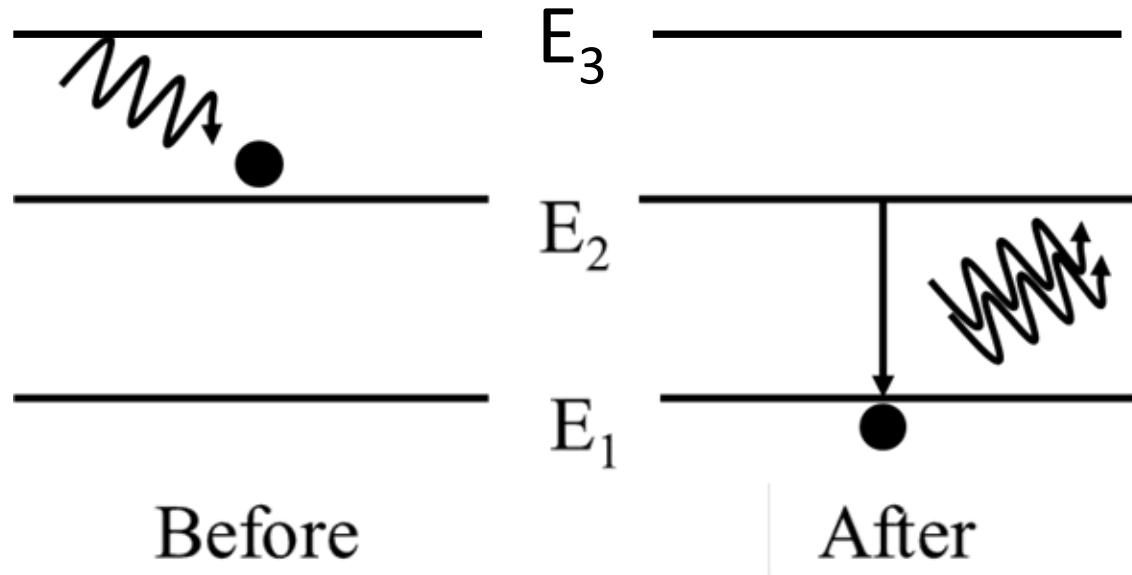
$$\left. \frac{dN}{dt} \right|_{sp} = A_{21} N_2$$

Where,

A_{21} : probability of spontaneous emission process

N_2 : number of atoms in higher energy level

Stimulated i.e. triggered emission



Emitted radiation has same

- i. Wavelength
- ii. Direction
- iii. Phase

Rate of transition

$$\left. \frac{dN}{dt} \right|_{st} = B_{21} N_2 Q$$

Where,

B_{21} : probability of stimulated emission process

N_2 : number of atoms in higher energy level E_2

Q : energy density of incident radiation per unit frequency

Important Laser Physics Terms

- Population

The number of active atoms occupying a particular energy state

- Population Inversion

Creating a non-equilibrium state with more atoms in excited states

- Pumping

Process of supplying trigger/input energy to achieve population inversion

- Metastable State

Special energy levels having unusually high lifetime

- Active medium

Region of the laser source where population inversion is achieved

Lasing Condition

- At equilibrium, $N_1 = \frac{g_1}{g_2} N_2 e^{(E_2 - E_1)/kT}$
- Since $E_2 > E_1$, at equilibrium, $N_1 \gg N_2$
- Lower states E_1 are usually more *populated* than upper states E_2
- For laser emission, we need $N_2 > N_1$
- This is a non-equilibrium state
- Achieved by using pumping and metastable state/s

Types of Pumping

- Optical pumping

A broad and bright source of light (photons) is used to supply energy

Example: Xenon flash lamp in Ruby laser, Nd:YAG laser, Dye lasers

- Electrical Pumping

High electric field is set up by a pair of electrodes

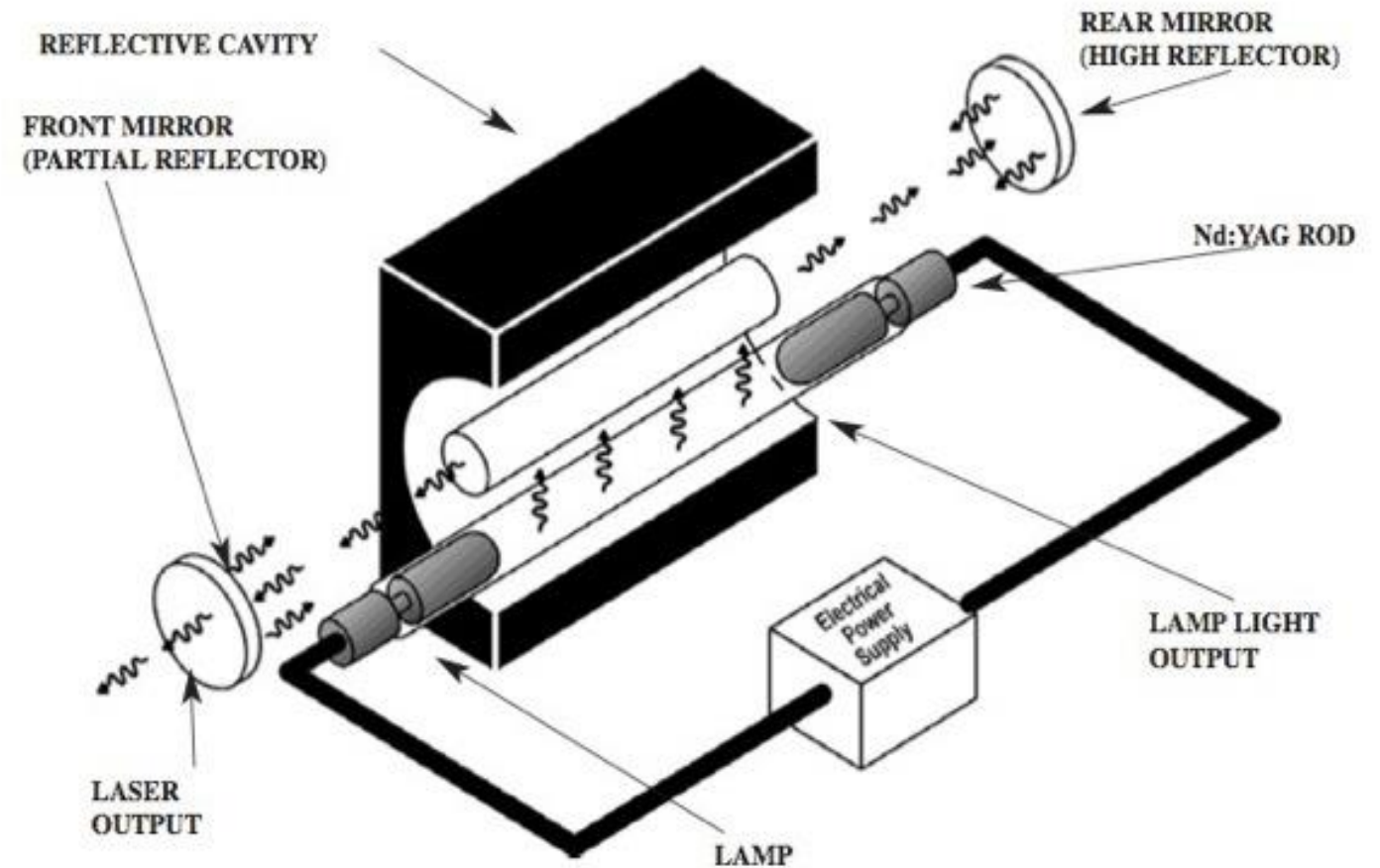
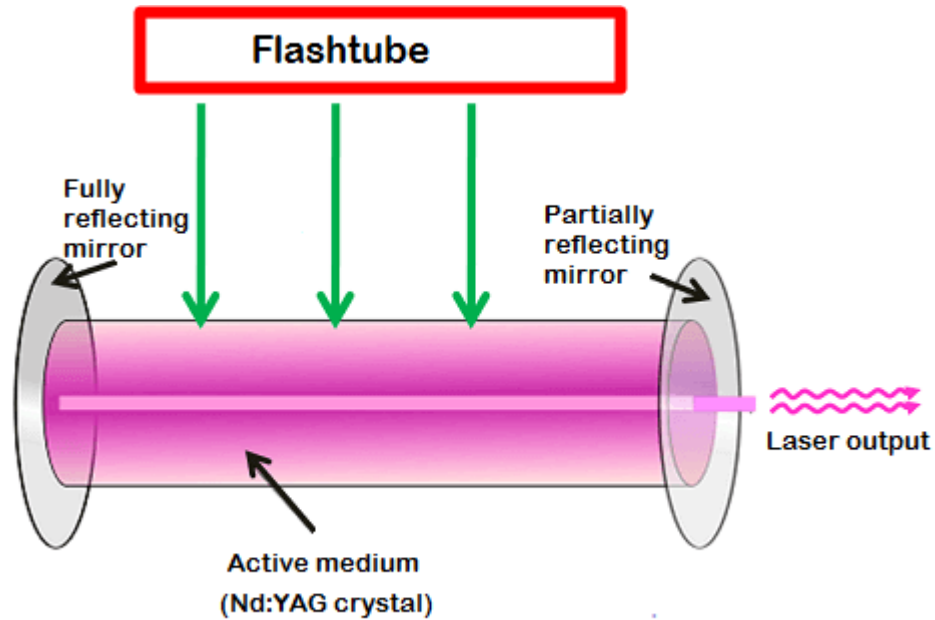
Example: He-Ne laser, CO₂ laser, Ar laser

- Direct conversion

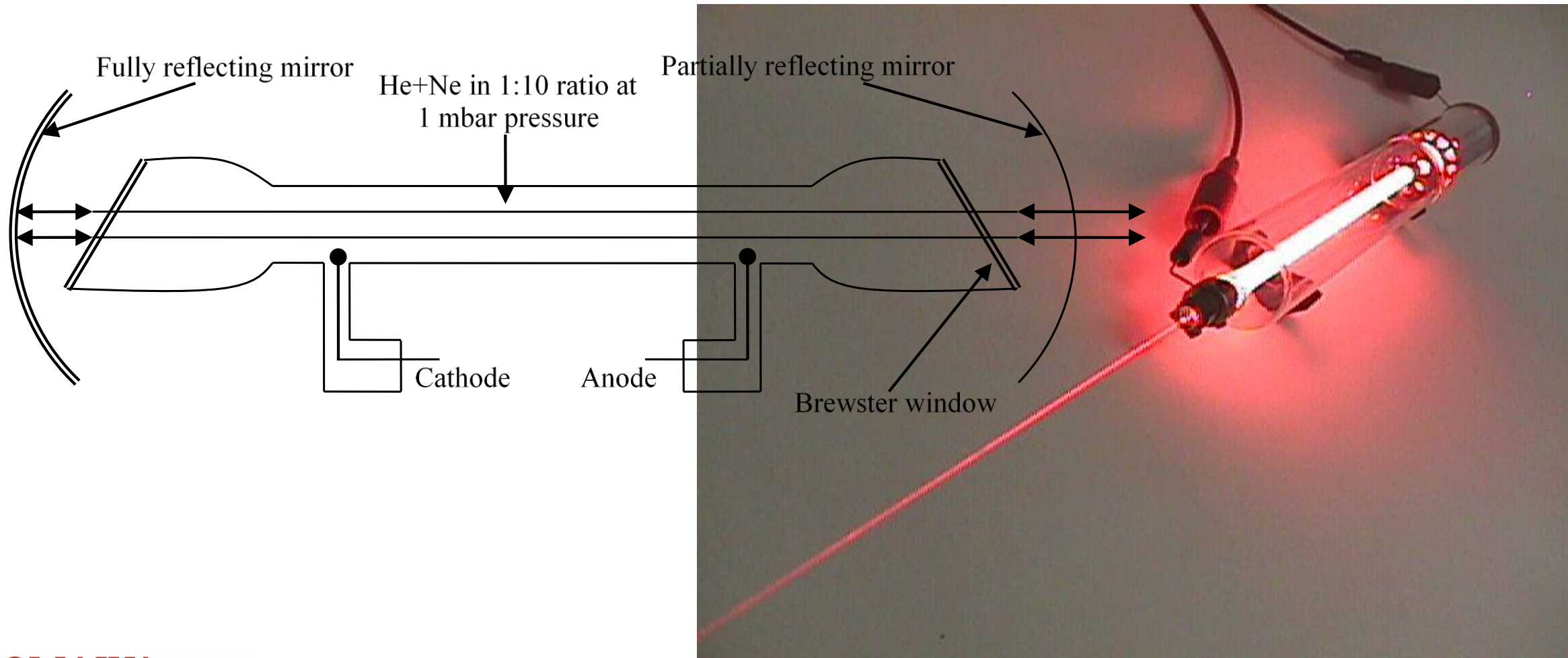
By passing an electric current

Example: Semiconductor diode lasers

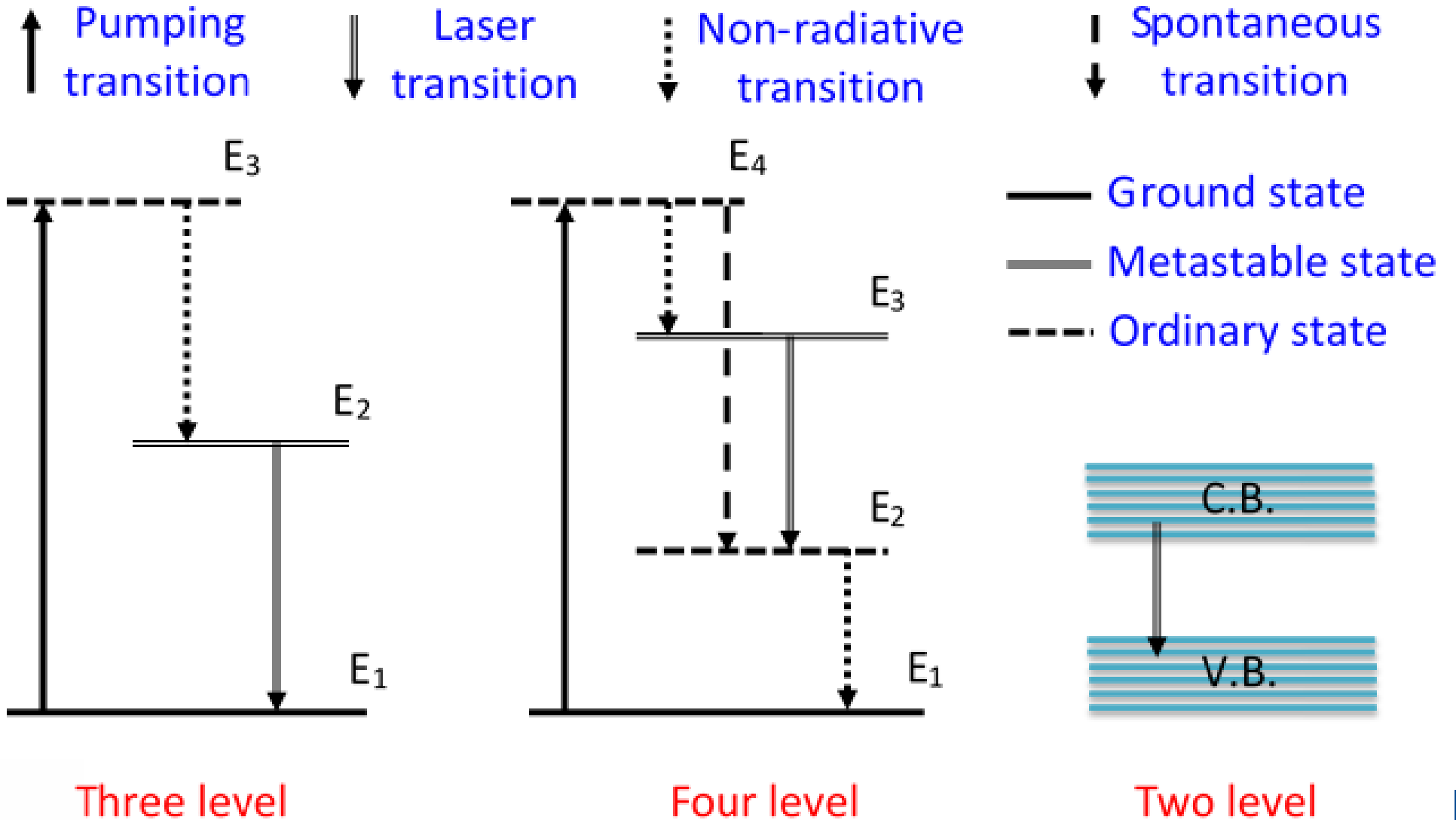
Optical Pumping - Schematic



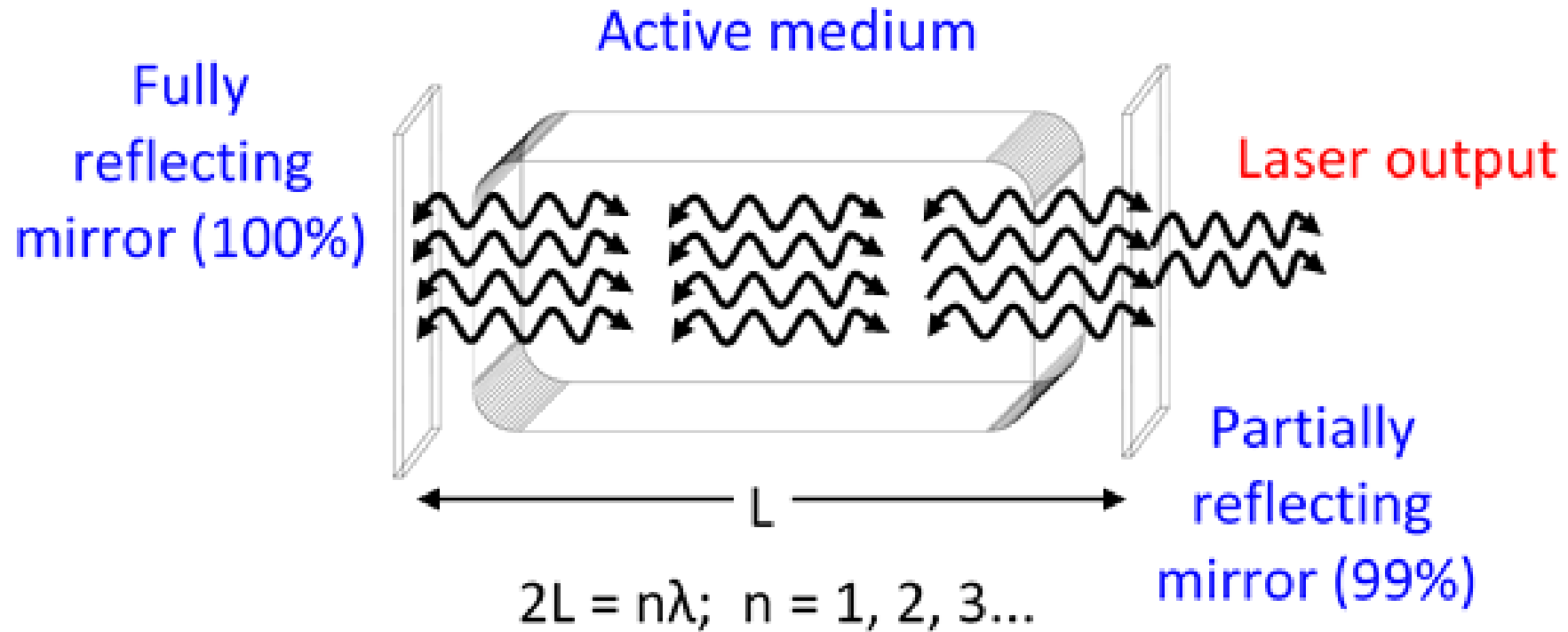
Electrical Pumping - Schematic



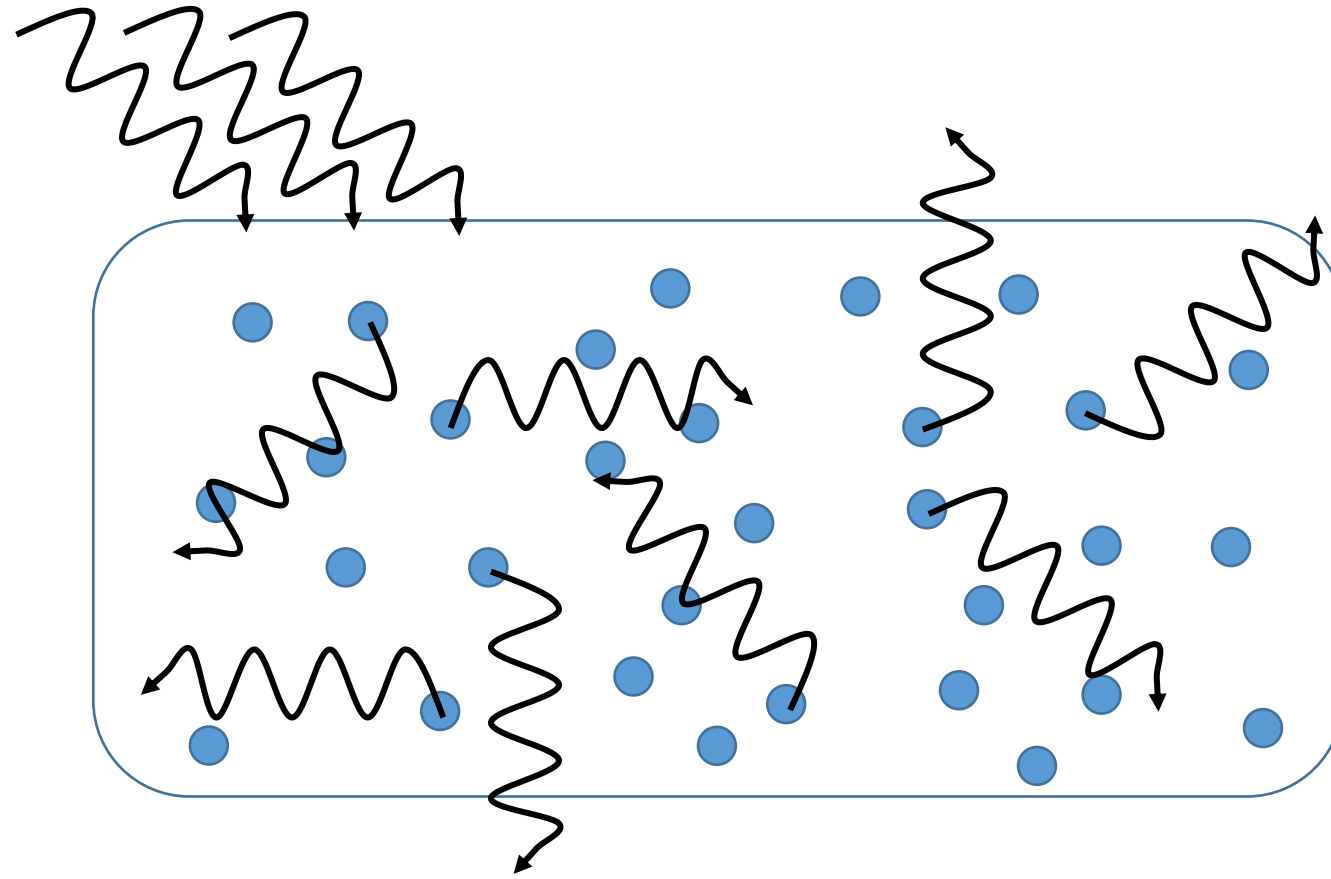
Pumping Schemes and Metastable States



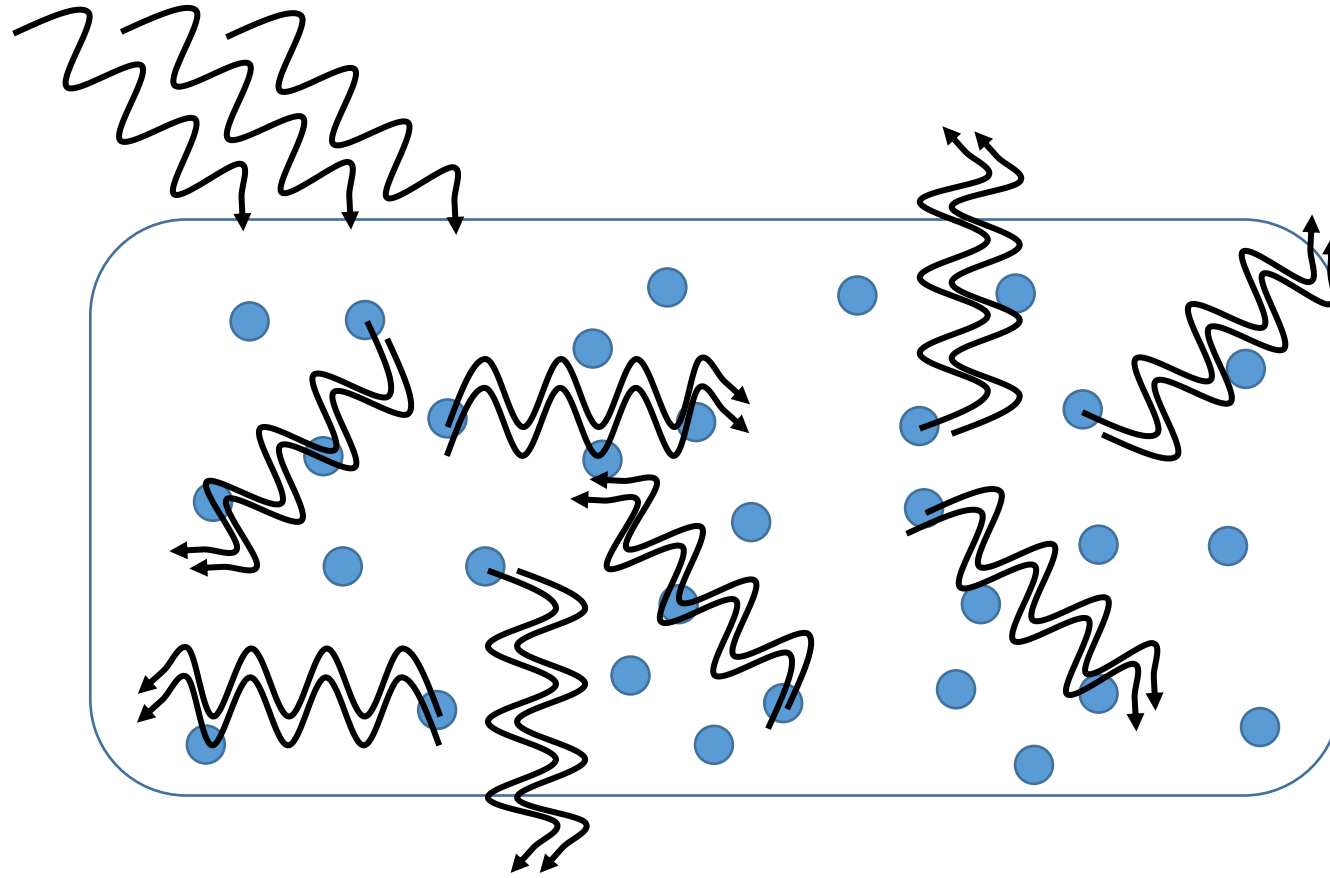
Optical Resonator



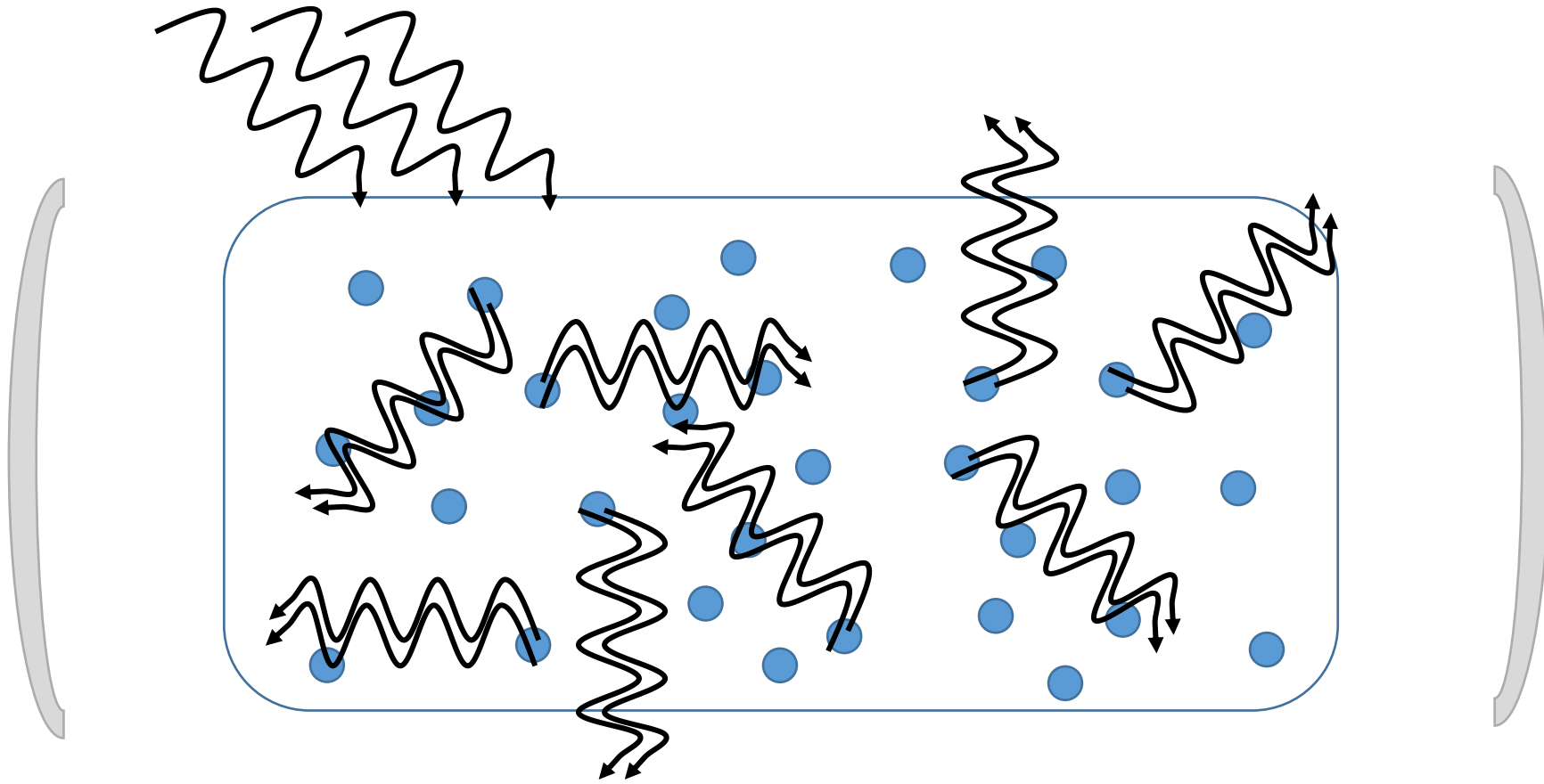
Action of Optical Resonator



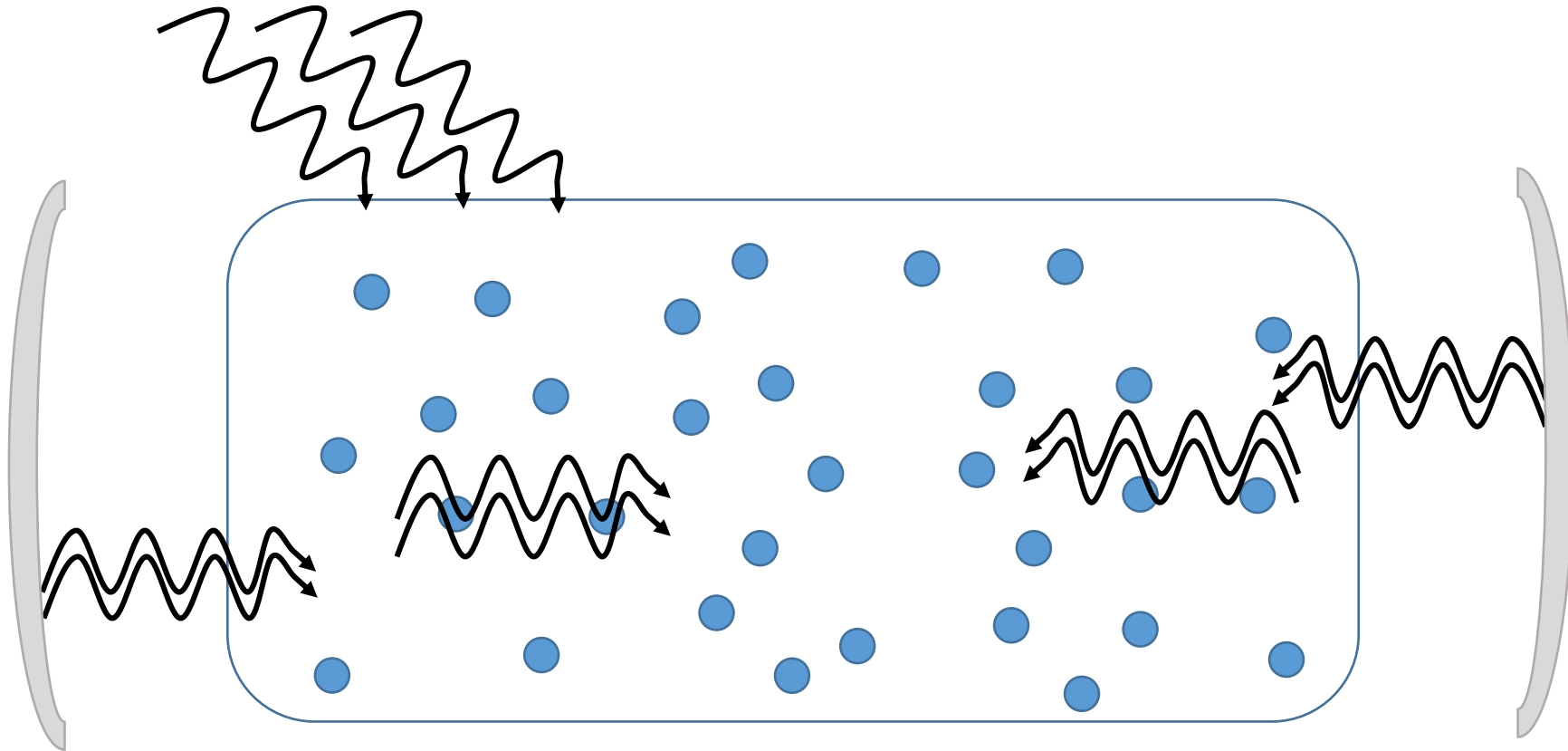
Action of Optical Resonator



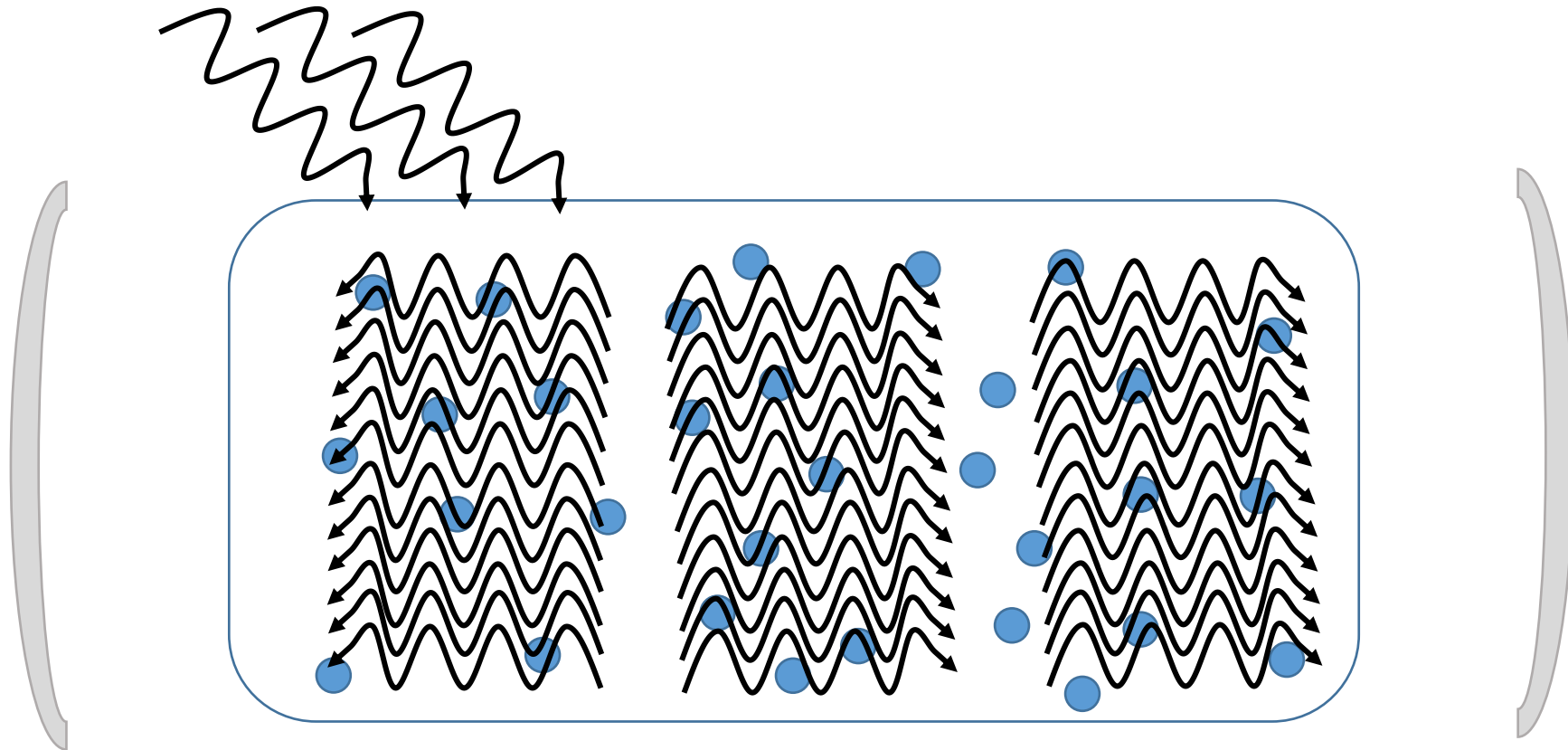
Action of Optical Resonator



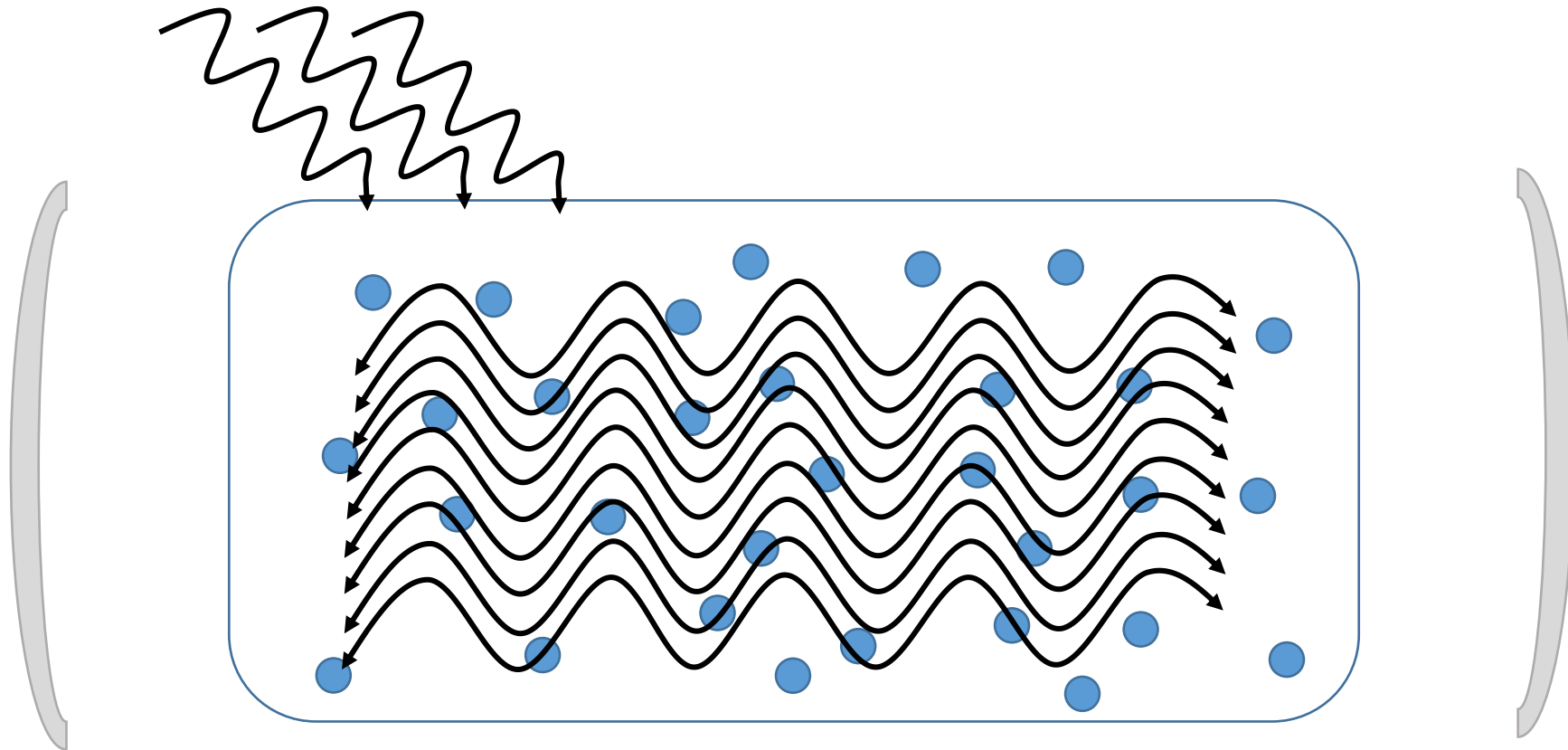
Action of Optical Resonator



Action of Optical Resonator



Action of Optical Resonator



Thanks!

Blackbody Radiation at Different Temperatures

