

## Homework Two

Monday, October 3, 2022 6:26 PM

Savannah Gramze Due: 10/7/22

$$1. \rho_0 = \frac{3}{4\pi} \frac{M}{R^3} \quad (\text{uniform density})$$

$$\text{a) } \frac{dP(r)}{dr} = -\frac{GM(r)\rho(r)}{r^2}$$

$$\frac{dM(r)}{dr} = 4\pi r^2 \rho(r)$$

$$M(r) = \int_0^r 4\pi r^2 \rho_0 dr = \frac{4}{3}\pi r^3 \rho_0$$

$$\frac{dP(r)}{dr} = -\frac{4\pi}{3} \frac{G \rho_0^2 r^5}{R^2}$$

$$P(r) = -\frac{4\pi G}{3} \rho_0^2 \int_r^R r dr$$

$$\boxed{P(r) = \frac{2\pi}{3} G \rho_0^2 (R^2 - r^2)}$$

$$\text{b) } P(0) = \frac{2\pi}{3} G \rho_0^2 R^2$$

$$= \frac{2\pi}{3} G \frac{3M}{16\pi} \times \frac{M^2}{R^4} R^2$$

$$\boxed{P_c = \frac{3}{8\pi} \frac{M^2}{R^4}, k = \frac{3}{8\pi}}$$

$$\text{c) } P_c = \frac{3}{8\pi} G \frac{M^2}{R^4}$$

$$R_B = 1.38 \times 10^{-16}$$

$$M_{\odot} = 2 \times 10^{33} \text{ g}$$

$$R_{\odot} = 6.957 \times 10^{10} \text{ cm}$$

$$G = 6.67 \times 10^{-8} \text{ [cgs]}$$

$$\rho_c = \frac{3}{4\pi} \frac{M}{R^3}$$

$$\boxed{\rho_c = 1.42 \text{ g/cm}^3}$$

$$\boxed{P_c = 1.36 \times 10^{15} \text{ dyn/cm}^2}$$

$$P_c = \frac{\rho_c k T_c}{m_H m_p}$$

$$m_H = 1.67 \times 10^{-27} \text{ g}$$

$$M = 2X + \frac{3}{4}Y + \frac{1}{2}Z = 0 + \frac{3}{4}(0.36) + \frac{1}{2}(0.64)$$

$$\bar{M} = 0.59 \Rightarrow M = 1.695$$

$$\bar{\mu} = 0.59 \Rightarrow \mu = 1.695$$

$$T_c = \frac{\mu m_p P_c}{\beta c k_B} = \frac{\mu M_H}{k_B} \frac{3}{8\pi} G \frac{M^2}{R^4} \frac{3}{4\pi} \frac{M}{R^3}$$

$$\boxed{T_c = 19,664,78 \text{ K}}$$

vs  $1.3 \times 10^7 \text{ K}$  for hydrogen burning

Too far below burning temp

$$2. \quad P = K p^2 \quad P(z=0) = P_0 \\ p(z=0) = p_0$$

$$a) \text{ Find density in terms of } p_0 \quad P = K p^2$$

$$\frac{dP}{dz} = -g(z) g = 2Kp \frac{dp}{dz} \quad dp = \frac{-g}{2K} dz$$

$$\frac{dp}{dz} = \frac{-g}{2K} \quad g(z) = \frac{-g}{2K} z + C$$

$$p(z=0) = p_0 = 0 + C \Rightarrow p_0 = C$$

$$p(z_*) = 0 = \frac{-g}{2K} z_* + p_0 \quad \boxed{p(z) = \frac{-g}{2K} z + p_0}$$

$$\frac{2p_0 K}{g} = z_*$$

$$b) \text{ Given } \tilde{v} = 1 \text{ between } 0 < z < z_*$$

$$z = \int_0^{z_*} K_c p dz = K_c \int_0^{z_*} \left( \frac{-g}{2K} z + p_0 \right) dz$$

$$= K_c \left[ \frac{-g}{4K} z^2 + p_0 z \right] \Big|_0^{z_*}$$

$$\tilde{v} = K_c \left[ \frac{-g}{4K} z_*^2 + p_0 z_* \right] = 1$$

$$K_c = \left[ \frac{-g}{4K} z_*^2 + p_0 z_* \right]^{-1} \\ \tilde{v} = \int \frac{\frac{-g}{2K} z + p_0}{\frac{-g}{4K} z_*^2 + p_0 z} dz$$

$$c) P = \frac{p k_B T}{m} \quad PV = n kT \quad \text{assume hydrostatic eq.}$$

... A A ... A // A A A

$$\frac{dp}{dz} - \rho g = \frac{dp}{dz} = \frac{k_B}{m} \left( \frac{dp}{dz} + \rho \frac{dT}{dz} \right)$$

$$p = \frac{\rho k_B T}{m} = \rho g \Rightarrow T = \frac{m k_B p}{\rho k_B}$$

$$T(z) = \frac{m K}{k_B} \left( \frac{-g}{2K} z + p_0 \right) \Rightarrow T(z) = \frac{m}{k_B} \left( \frac{-g}{2} z + p_0 K \right)$$

$$B_\nu(\nu, t) = \frac{2 \nu^2 k_B T}{c^2} \text{ so } B_\nu = \frac{2 \nu^2}{c^2} (-g z + 2 p_0 K) \\ \text{so } a = 4 p_0 K \frac{\nu^2}{c^2} \text{ and } b = -2 g \frac{\nu^2}{c^2}$$

$$\beta = \frac{P_{\text{gas}}}{P_{\text{tot}}} = \text{const.}$$

$$3. U = \int (u_{\text{gas}} + u_{\text{rad}}) dm$$

$$E = U + \Omega = \frac{-\beta}{2-\beta} U = \frac{\beta}{2} \Omega$$

Implications for  $\beta = 0$  and  $\beta = 1$ ?  
 (rad. dominated) (gas dominated)

$E = \text{total energy}$   
 $\Omega = \text{total grav. potential}$   
 $U = \text{total thermal energy}$

a) Virial theorem

$$E = U + \Omega = \frac{-\beta}{2-\beta} U = \frac{\beta}{2} \Omega$$

$$\beta = 0$$

$$\frac{-0}{2-0} U = \frac{0}{2} \Omega$$

$$U = \Omega ?$$

No grav. potential, unstable

$$\beta = 1$$

$$\frac{-1}{2-1} U = \frac{1}{2} \Omega$$

$$-U = \frac{1}{2} \Omega$$

Star in equilibrium, Virial theorem holds

b) Stability of  $\star$ 's, contraction  $\xrightarrow{3}$  expansion

... will

b) Stability of  $\dot{s}$ 's, contraction  $\Rightarrow$  expansion

$\beta=0$  unstable  $\rightarrow$  no const. total energy, will  
expand  $\Rightarrow$  contract

$\beta=1$  stable  $\rightarrow$  const. energy, will not  
expand or contract