



Aalto University  
School of Engineering

# COE-C2004 - Materials Science and Engineering

*Prof. Junhe Lian*

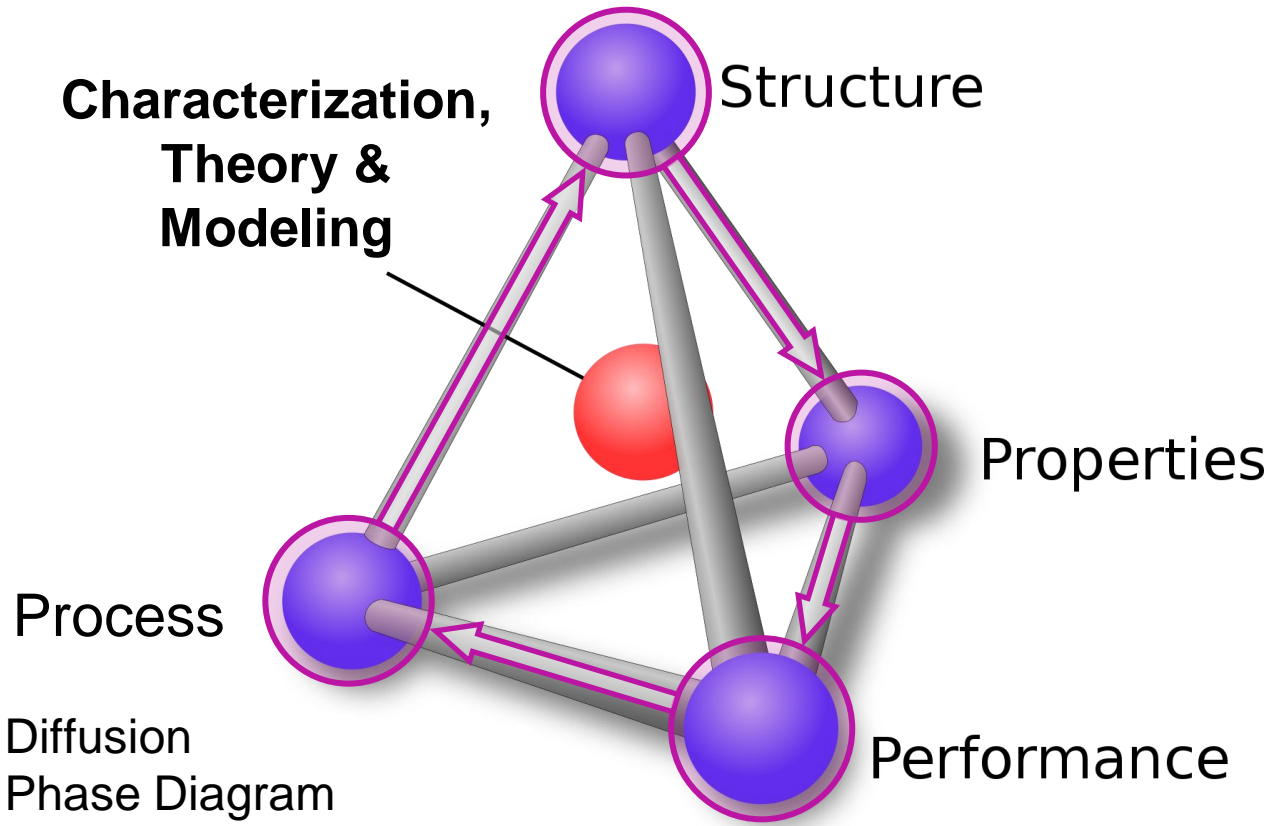
*Wenqi Liu (Primary teaching Assistant)*

*Rongfei Juan (Teaching Assistant)*

# Processes



# Previously



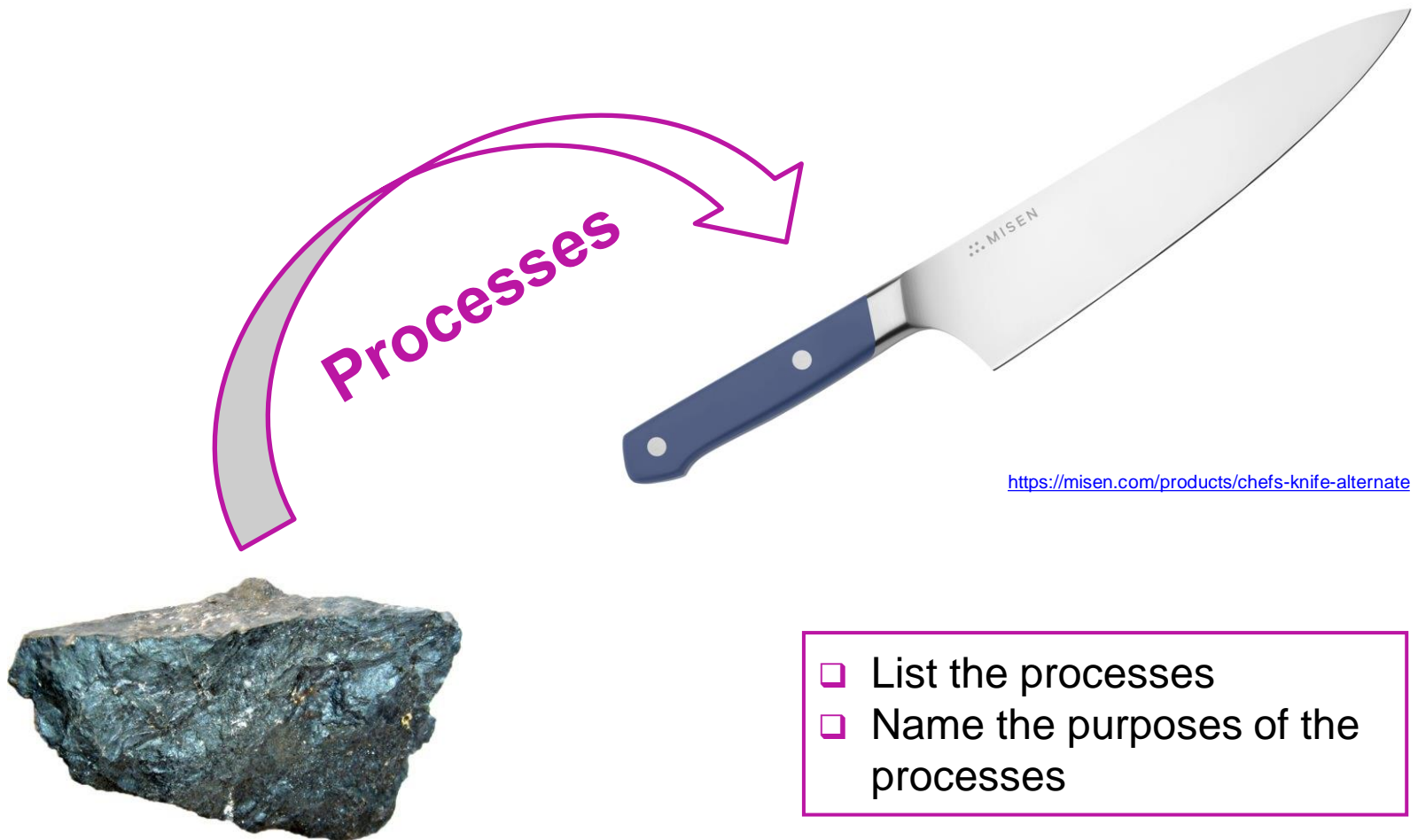
- ☐ Diffusion
- ☐ Phase Diagram
- ☐ Phase Transformation
- ☐ **Processes**

# Learning Objectives

After studying this chapter you should be able to do the following:

- ❑ Name and describe four casting techniques.
- ❑ Name and describe four forming operations that are used to shape metal alloys.
- ❑ State the purposes of and describe procedures for the four heat treatments: process annealing, stress relief annealing, full annealing, and spheroidizing.
- ❑ Correlate the processes with structure of materials.

# How to produce a kitchen knife?



- ❑ List the processes
- ❑ Name the purposes of the processes

# Group work

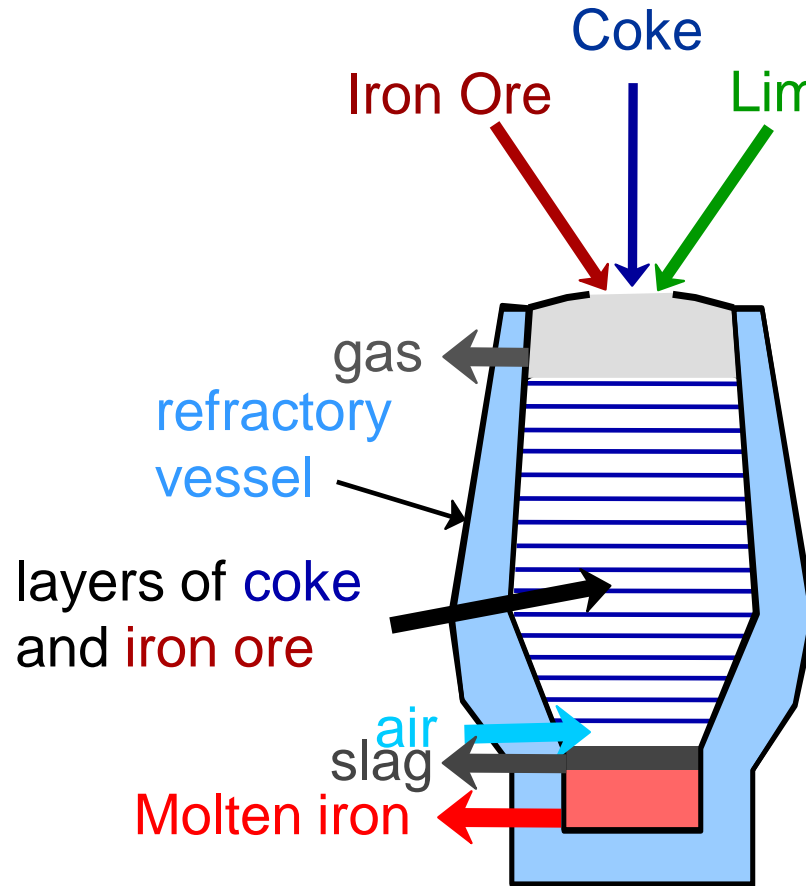
- ❑ **Four groups** are will be created by Zoom, leading you to Four separate rooms.
  - ❑ Each group will **decorate** your “**FLINGA Wall**”. (Post-it, Figures, and Drawings, etc.)
  - ❑ Following the **guiding question** on FLINGA, discuss and work together.
  - ❑ After the discussion, **present** your wall to all of us.
- 
- ❑ Discussion duration: ~10 mins.
  - ❑ Moderator:  $\max\{\text{Birth month}(\text{day})\}$
  - ❑ Presenter: decided by the group.

Group 1: <https://flinga.fi/s/FHDJCE2>

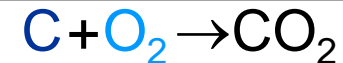
Group 2: <https://flinga.fi/s/FMKHY8B>

Group 3: <https://flinga.fi/s/FWP9EC6>

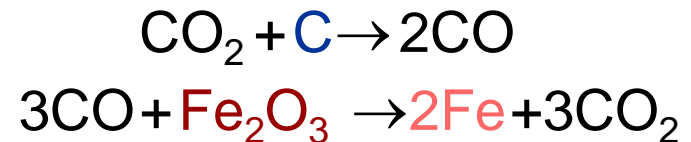
# Metal Making from Ore



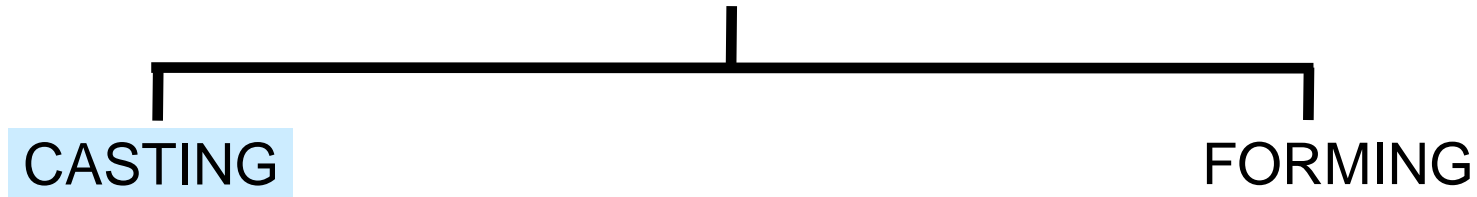
heat generation



reduction of iron ore to metal



# Metal Fabrication Methods (i)



## Casting

- mold is filled with molten metal
- metal melted in furnace, perhaps alloying elements added, then **cast** in a mold
- common and inexpensive
- gives good production of shapes
- good option for brittle materials without forming
- weaker products, internal defects



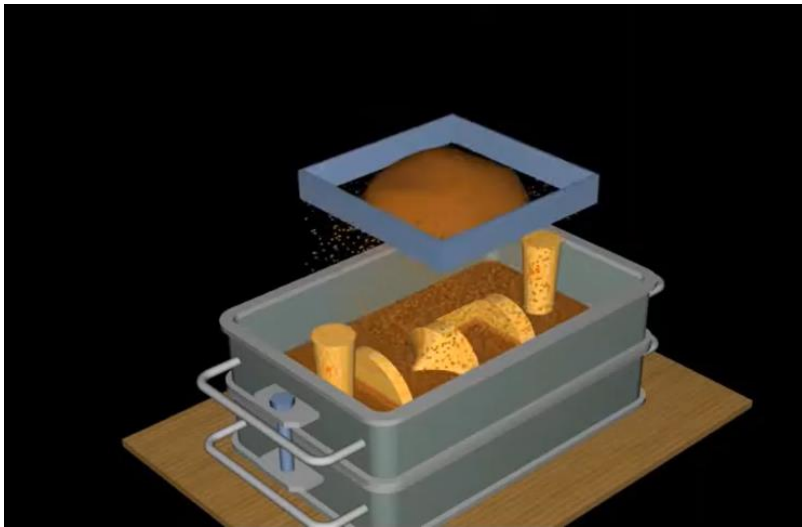
# Metal Fabrication Methods (ii)

## CASTING

## FORMING

- Sand Casting

(large parts, easy shape,  
e.g., auto engine blocks)



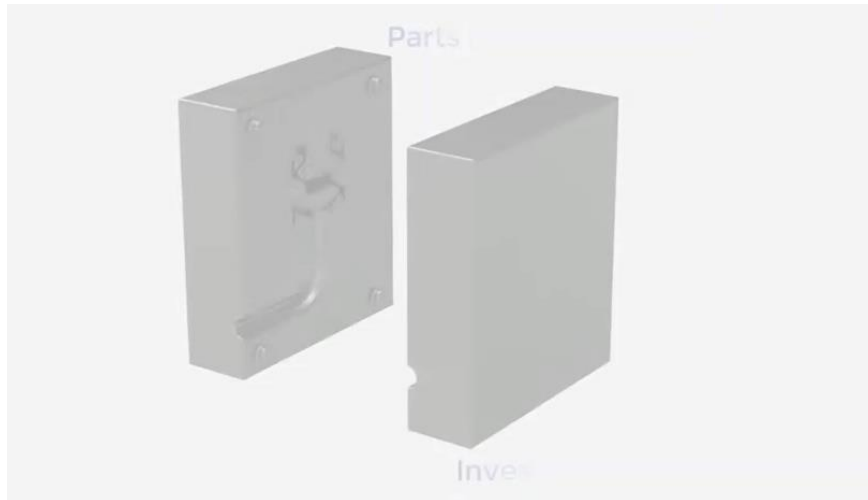
<https://www.youtube.com/watch?v=7BI4v0Gqc7k>

- What material will withstand  $T > 1600^{\circ}\text{C}$  and is inexpensive and easy to mold?
- Answer: **sand!!!**
- To create mold, pack sand around form (pattern) of desired shape

# Metal Fabrication Methods (iii)

## CASTING

- Investment Casting  
(low volume, complex shapes,  
e.g., jewelry, turbine blades)



[https://youtu.be/S4WuKJF\\_76c](https://youtu.be/S4WuKJF_76c)

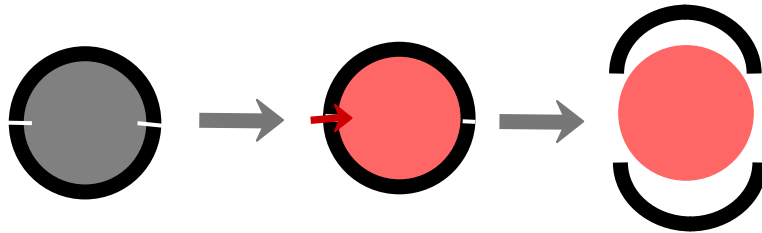
## FORMING

- Stage 0 — Preparing wax patterns.
- Stage I — Mold formed by pouring plaster of paris around wax pattern. Plaster allowed to harden.
- Stage II — Wax is melted and then poured from mold—hollow mold cavity remains.
- Stage III — Molten metal is poured into mold and allowed to solidify.

# Metal Fabrication Methods (iv)

## CASTING

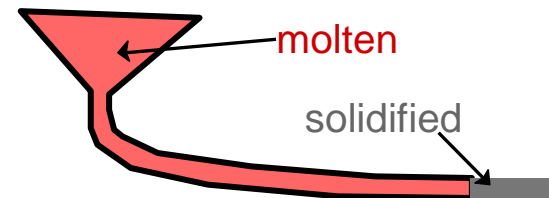
- Die Casting
  - high volume
  - for alloys having low melting temperatures (Al, Zn, Mg)



Recommended to watch:  
[What is die casting process?](#)

## FORMING

- Continuous Casting
  - simple shapes  
(e.g., rectangular slabs, cylinders)

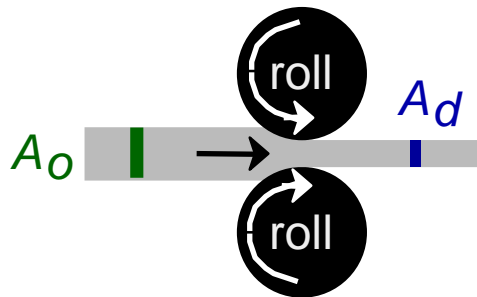


Recommended to watch:  
[Continuous Casting Process](#)

# Metal Fabrication Methods (v)

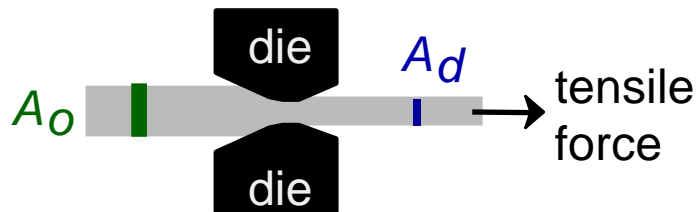
## CASTING

- Rolling (Hot or Cold Rolling)  
(I-beams, rails, sheet & plate)



Adapted from Fig. 11.9,  
Callister & Rethwisch 10e.

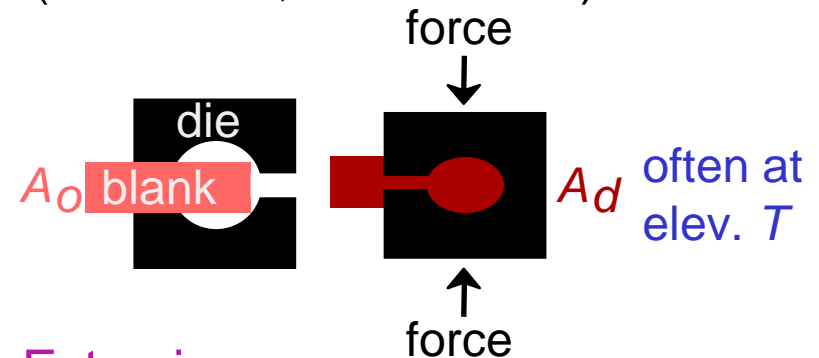
- Drawing  
(rods, wire, tubing)



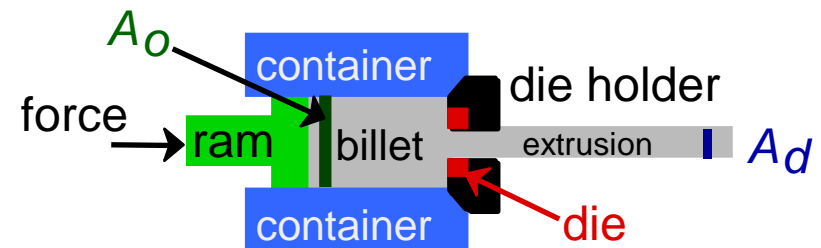
die must be well lubricated & clean

## FORMING

- Forging (Hammering; Stamping)  
(wrenches, crankshafts)

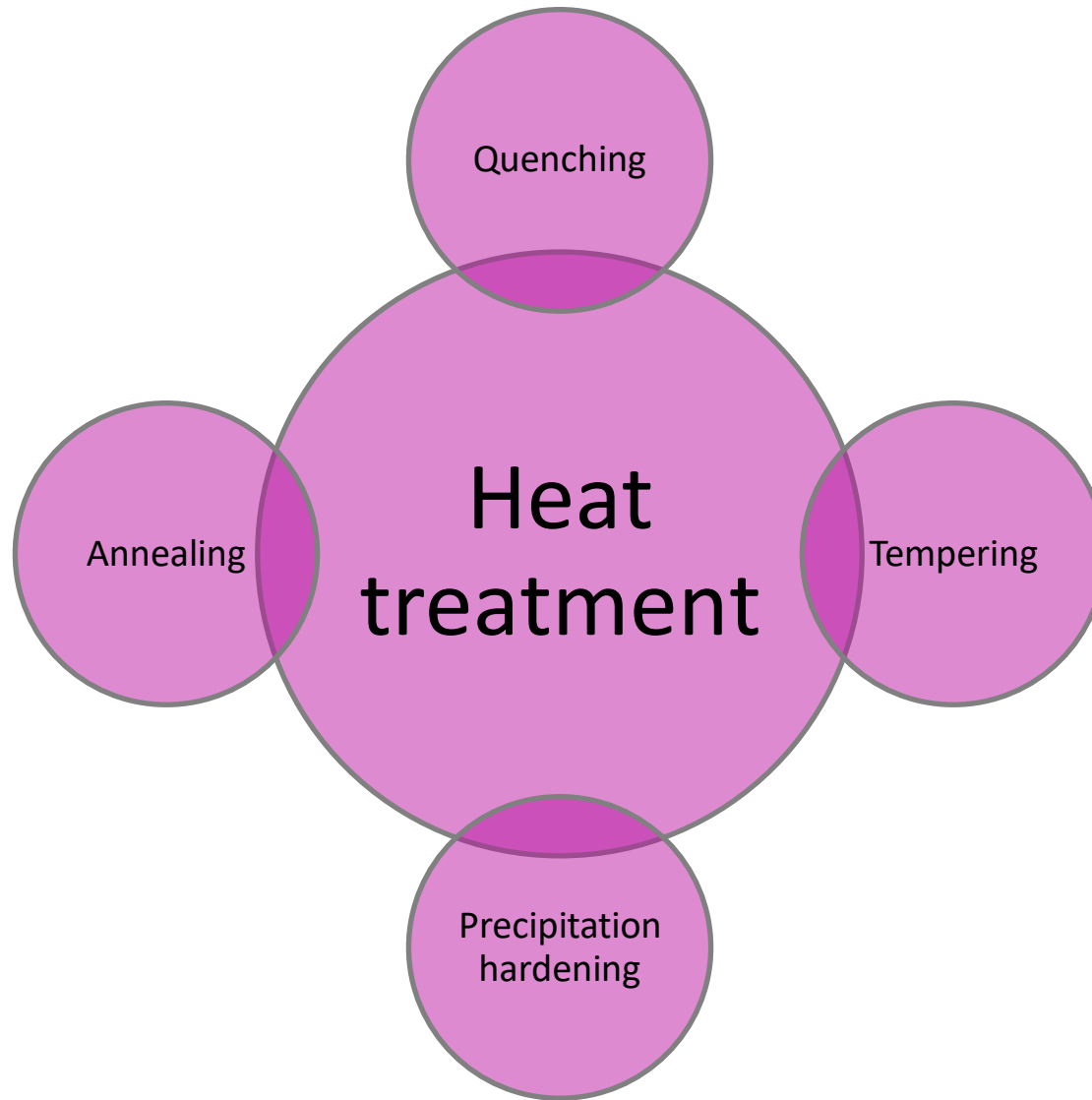


- Extrusion  
(rods, tubing)



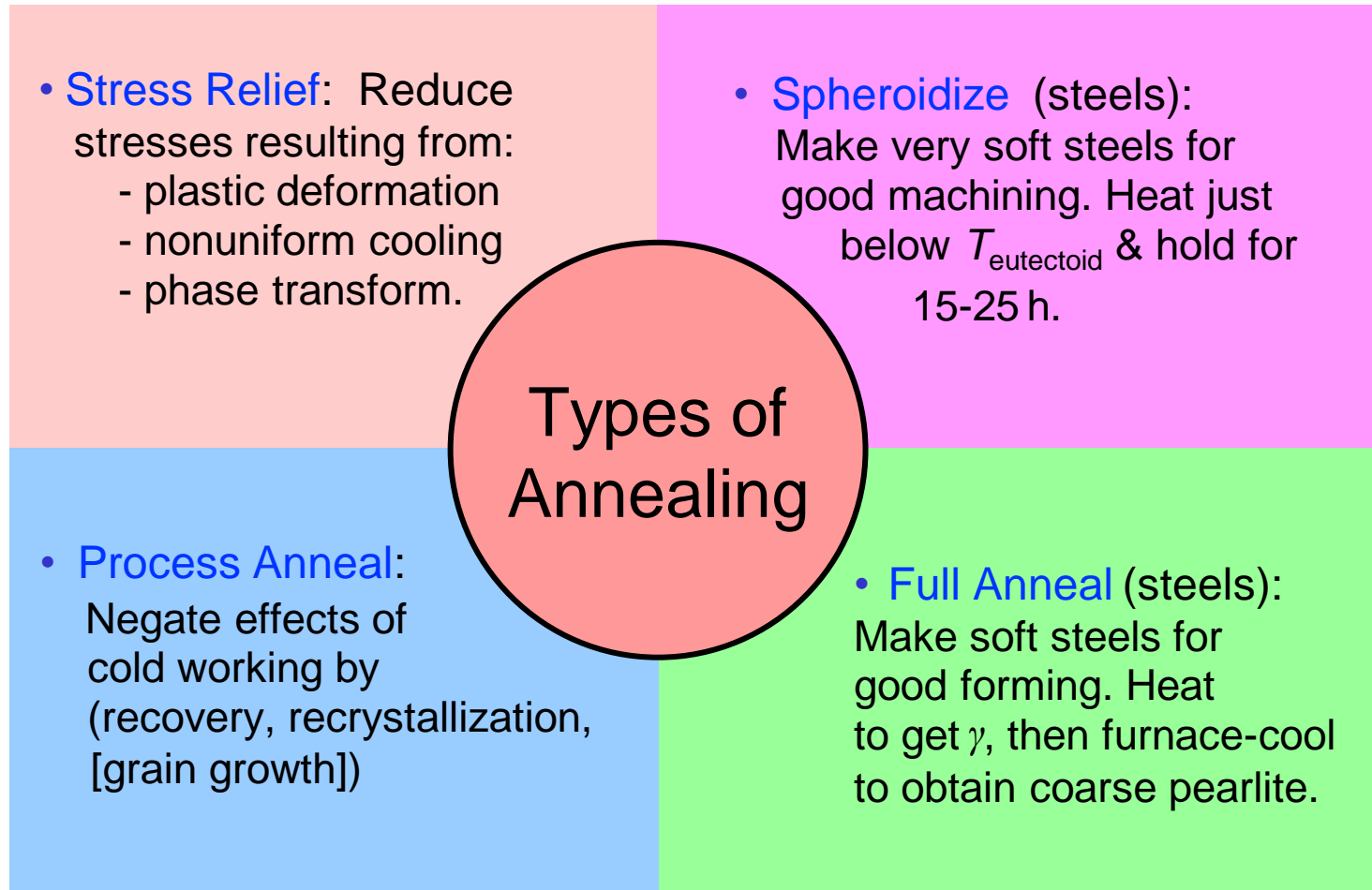
ductile metals, e.g. Cu, Al (hot)

# Heat treatment types



# Thermal Processing of Metals

**Annealing:** Heat to  $T_{\text{anneal}}$ , then cool slowly.



Based on discussion in Section 11.8, *Callister & Rethwisch 10e*.

# Spheroidite: Another Microstructure for the Fe-Fe<sub>3</sub>C System

- Spheroidite:
  - Fe<sub>3</sub>C particles within an  $\alpha$ -ferrite matrix
  - formation requires diffusion
  - heat bainite or pearlite at temperature just below eutectoid for long times
  - driving force – reduction of  $\alpha$ -ferrite/Fe<sub>3</sub>C interfacial area

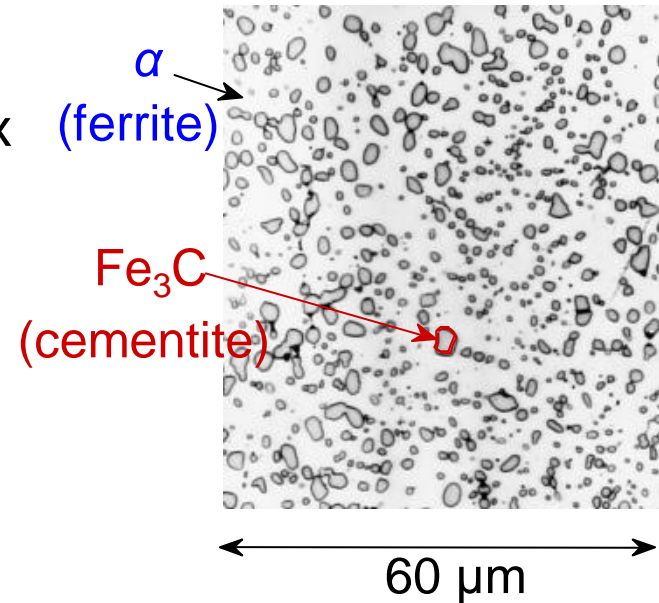
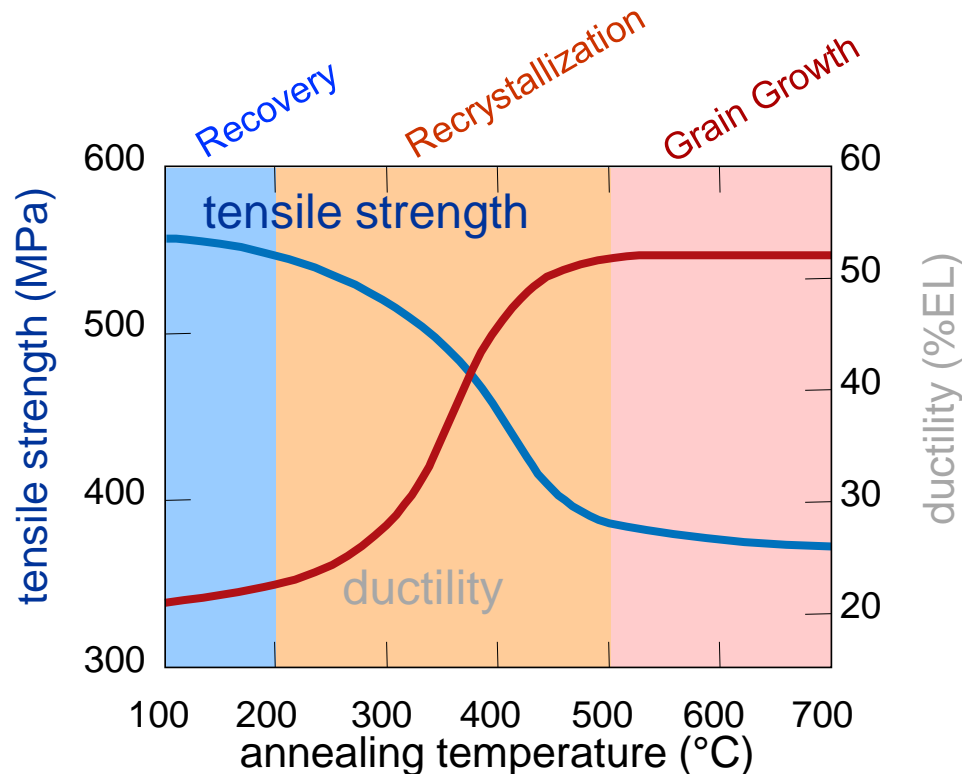


Fig. 10.19, Callister & Rethwisch 10e.  
(Copyright United States Steel Corporation, 1971.)

# Process Annealing of Cold-Worked Alloys

- Heat treating cold worked metals brings about changes in structure and properties
- As a result, effects of cold work are nullified!
- This type of heat treatment often termed “annealing”
- 1 hour treatment at  $T_{\text{anneal}}$  decreases tensile strength & increases %EL



Three Annealing stages:

1. Recovery (100-200°C)
2. Recrystallization (200-500°C)
3. Grain Growth (> 500°C)

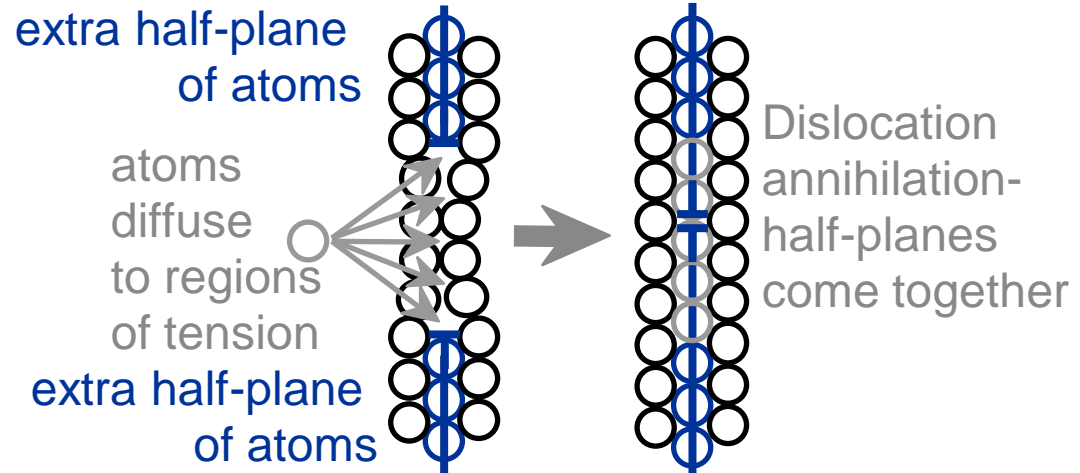
Fig. 7.22, Callister & Rethwisch 10e. (Adapted from G. Sachs and K. R. Van Horn, *Practical Metallurgy, Applied Metallurgy and the Industrial Processing of Ferrous and Nonferrous Metals and Alloys*, 1940. Reproduced by permission of ASM International, Materials Park, OH.)



# Recovery

During recovery – reduction in disl. density – annihilation of disl.

- Scenario 1

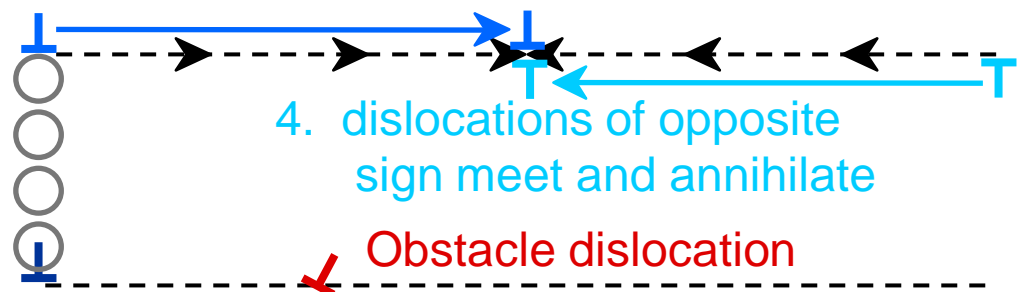


- Scenario 2

3. “Climbed” disl. can now move on new slip plane

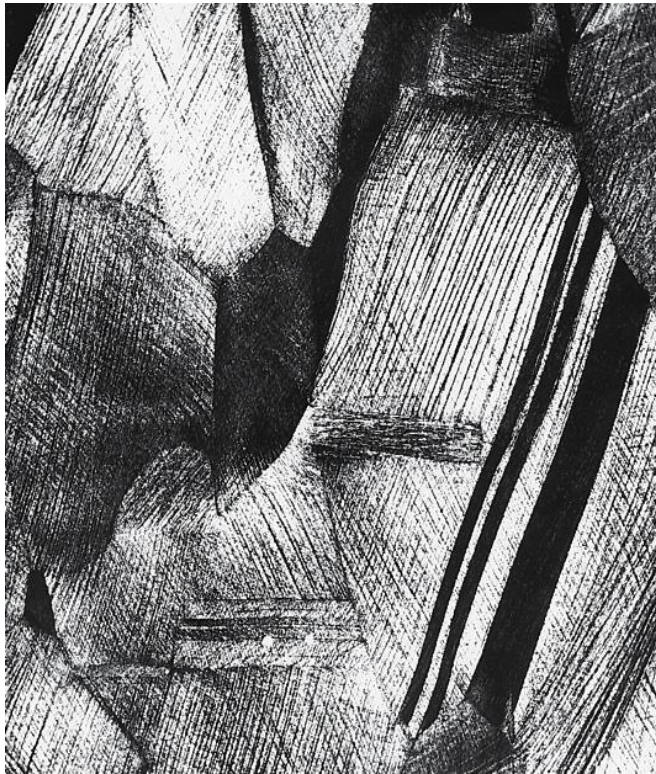
2. grey atoms leave by vacancy diffusion allowing disl. to “climb”

1. dislocation blocked; can’t move to the right

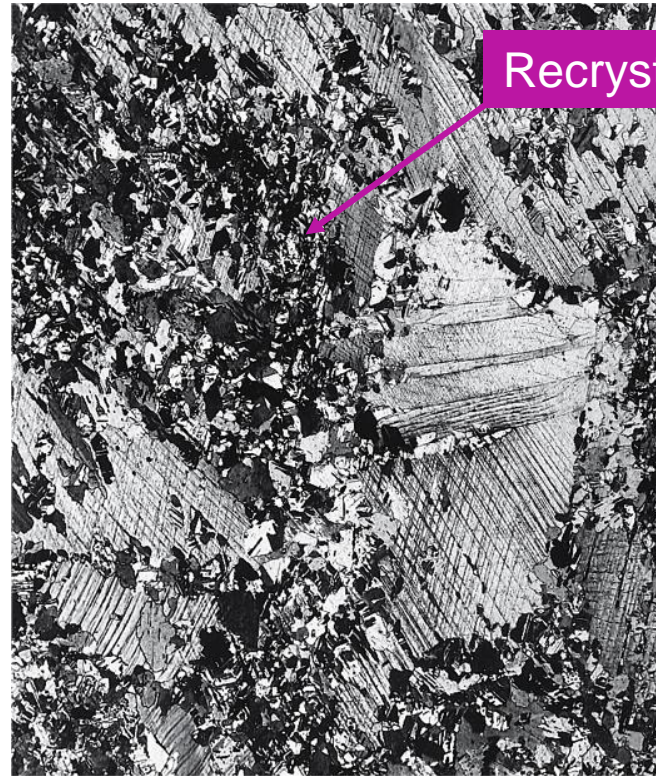


# Recrystallization

- New grains form that:
  - have low dislocation densities
  - are small in size
  - consume and replace parent cold-worked grains.



33%CW brass before heat treatment



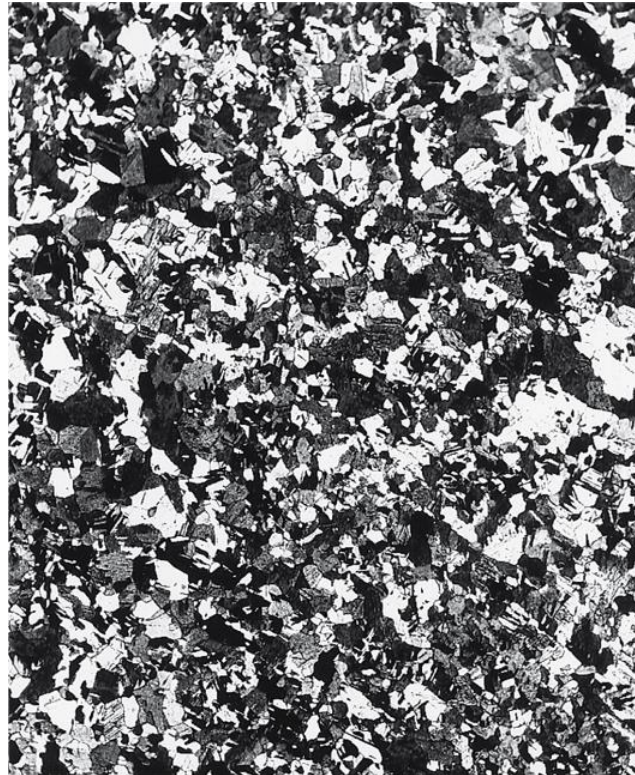
After 4 sec. at 580°C

Recrystallized grains

Adapted from Fig. 7.21  
(a),(c), Callister &  
Rethwisch 10e.  
(Photomicrographs  
courtesy of J.E. Burke,  
General Electric  
Company.)

# Recrystallization (cont.)

- All grains in cold-worked material have been consumed/replaced.



Adapted from Fig. 7.21 (d),  
*Callister & Rethwisch 10e.*  
(Photomicrographs courtesy  
of J.E. Burke, General  
Electric Company.)

After 8 sec. at 580°C



# Recrystallization Temperature

$T_R$  = recrystallization temperature = temperature at which recrystallization just reaches completion in 1 h.

$$0.3T_m < T_R < 0.6T_m$$

For a specific metal/alloy,  $T_R$  depends on:

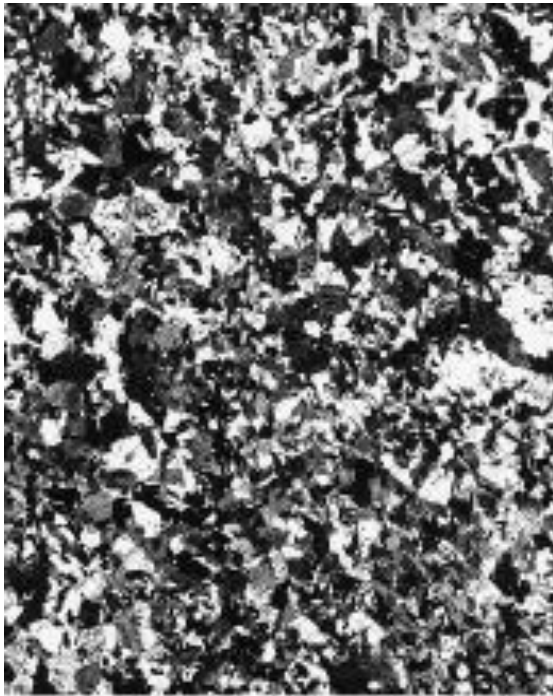
- %CW --  $T_R$  decreases with increasing %CW
- Purity of metal --  $T_R$  decreases with increasing purity

Cold work vs. Hot working

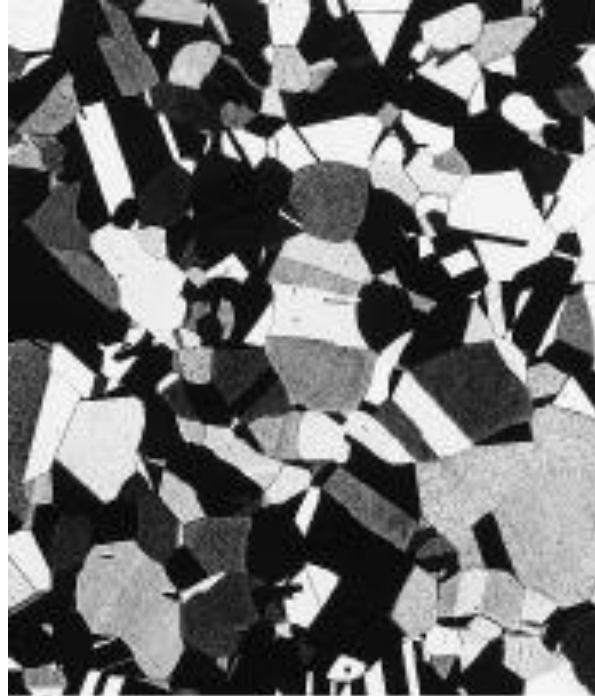
- **Hot working** → deformation above  $T_R$
- **Cold working** → deformation below  $T_R$

# Grain Growth

- Grain growth occurs as heat treatment continues.
  - Average grain size increases
  - Small grains shrink (and ultimately disappear)
  - Large grains continue to grow



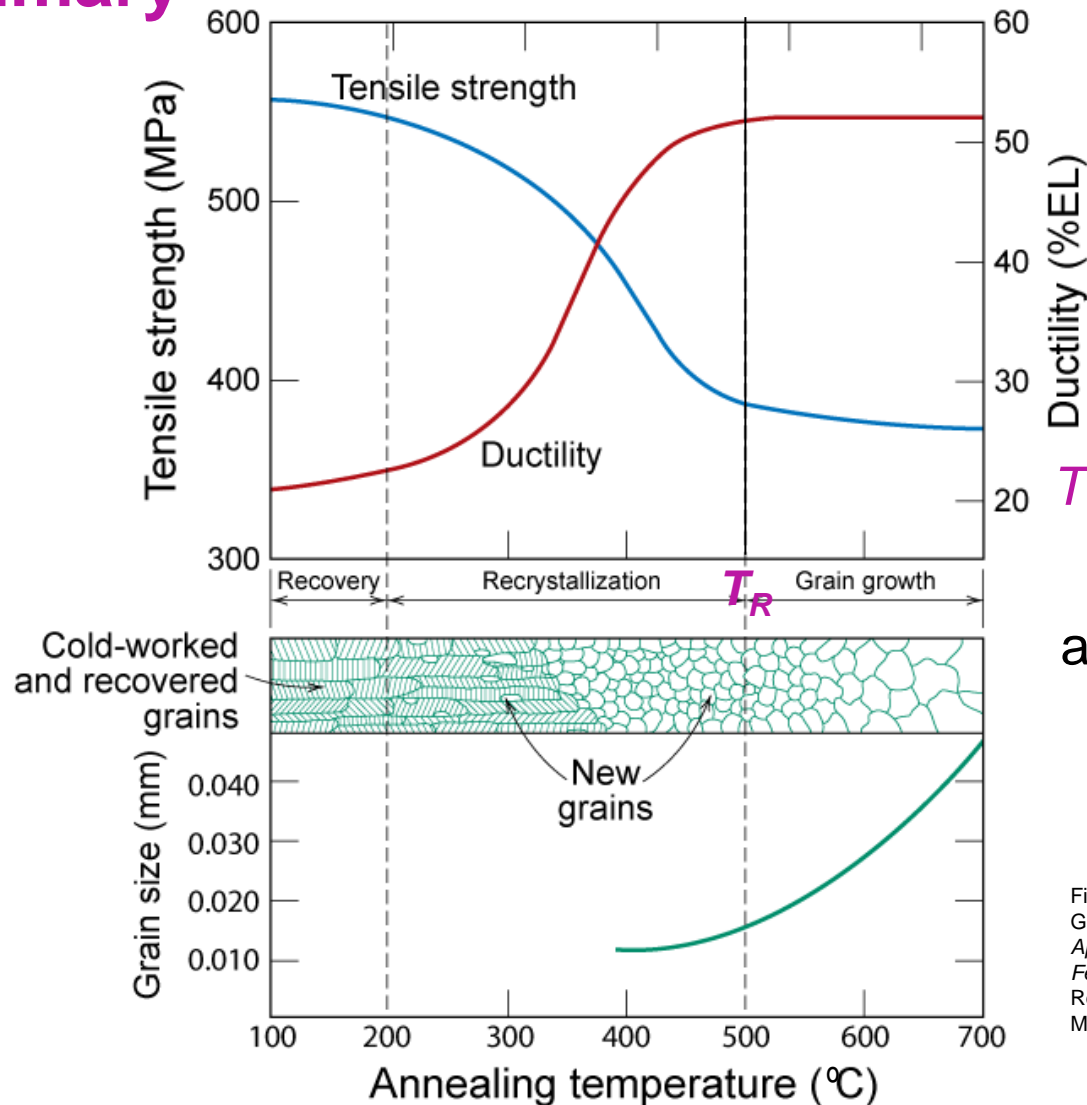
After 8 sec. at 580°C



After 15 min. at 580°C

Adapted from Fig. 9.21 (d),(e),  
*Callister & Rethwisch 10e.*  
(Photomicrographs courtesy of  
J.E. Burke, General Electric  
Company.)

# Recovery, Recrystallization, & Grain Growth Summary



$T_R$  = recrystallization temperature

annealing time = 1 h

Fig. 7.22, Callister & Rethwisch 10e. (Adapted from G. Sachs and K. R. Van Horn, *Practical Metallurgy, Applied Metallurgy and the Industrial Processing of Ferrous and Nonferrous Metals and Alloys*, 1940. Reproduced by permission of ASM International, Materials Park, OH.)

# Grain Size Influences Properties

## Recrystallization vs. Grain Growth

- ❑ **Metals having small grains** – relatively strong and tough at low temperatures
- ❑ **Metals having large grains** – good creep resistance at relatively high temperatures



<https://www.worldpipelines.com/project-news/29112019/balticconnectors-offshore-pipeline-filled-with-gas-and-ready-for-use/>

<https://www.solartronmetrology.jp/product-application/aerospace/profile-analysis-of-complex-curves-such-as-airfoils>

# Temperature – Time Paths

- a) Full Annealing
- b) Quenching
- c) Tempering  
(Tempered Martensite)

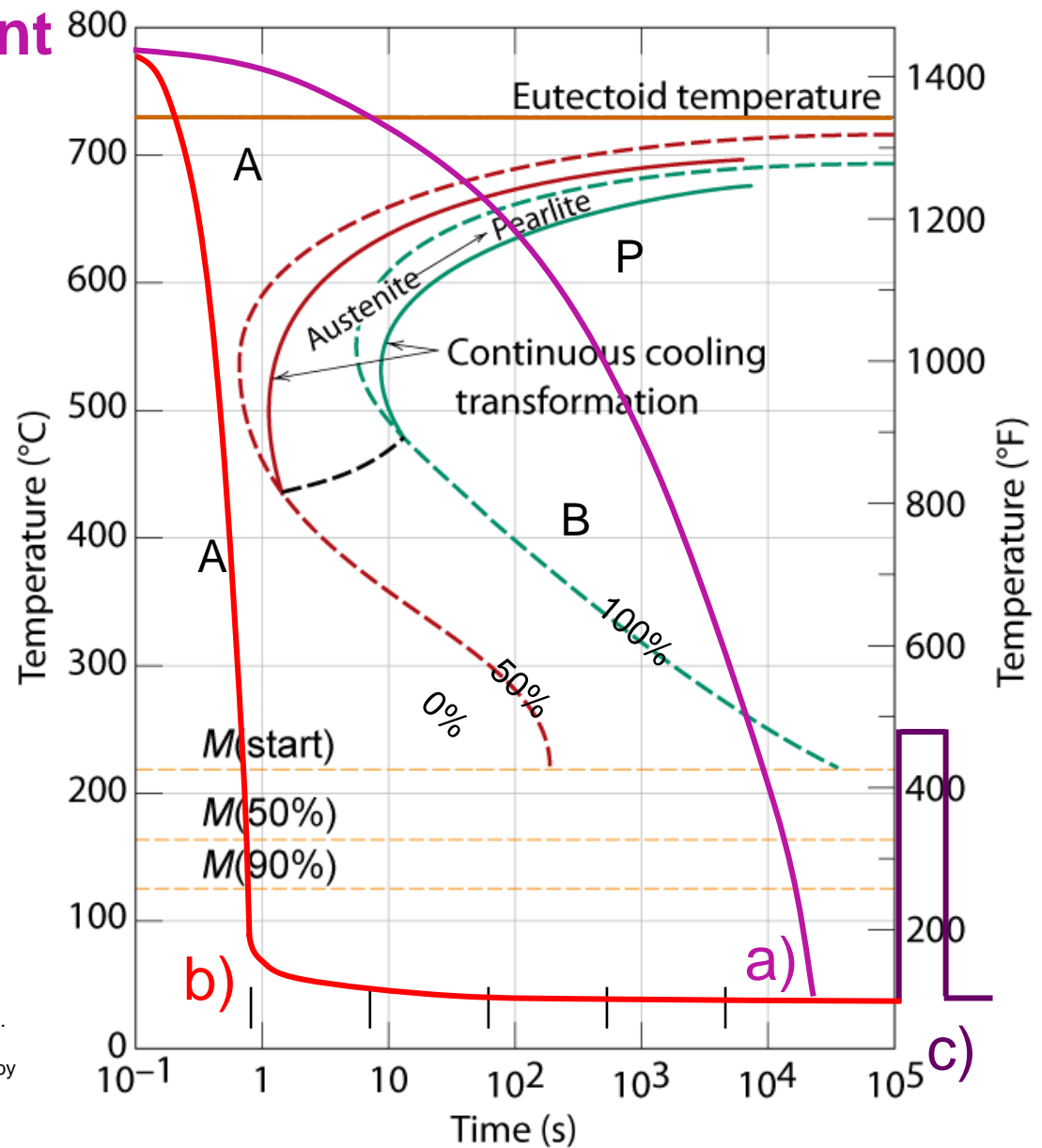


Fig. 10.25, *Callister & Rethwisch 10e*. [Adapted from H. Boyer (Editor), *Atlas of Isothermal Transformation and Cooling Transformation Diagrams*, 1977. Reproduced by permission of ASM International, Materials Park, OH.]



# Influences of Quenching Medium & Specimen Geometry

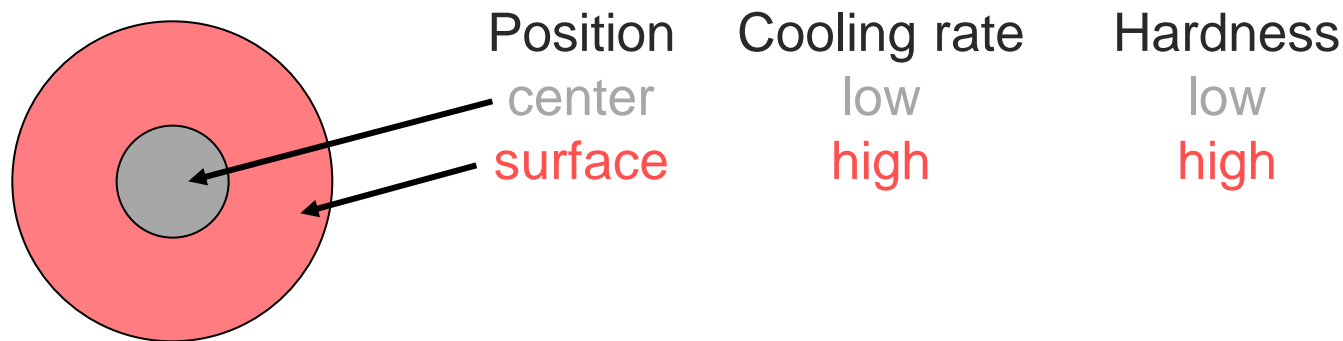
- Effect of quenching medium:

Medium	Severity of Quench	Hardness
air	low	low
oil	moderate	moderate
water	high	high

- Effect of specimen geometry:

When surface area-to-volume ratio increases:

- cooling rate throughout interior increases
- hardness throughout interior increases



# Precipitation Hardening

- Particles impede dislocation motion.

- Ex: Al-Cu system

- Procedure:

- Pt A: solution heat treat (get  $\alpha$  solid solution)
- Pt B: quench to room temp. (retain  $\alpha$  solid solution)
- Pt C: reheat to nucleate small  $\theta$  particles within  $\alpha$  phase.

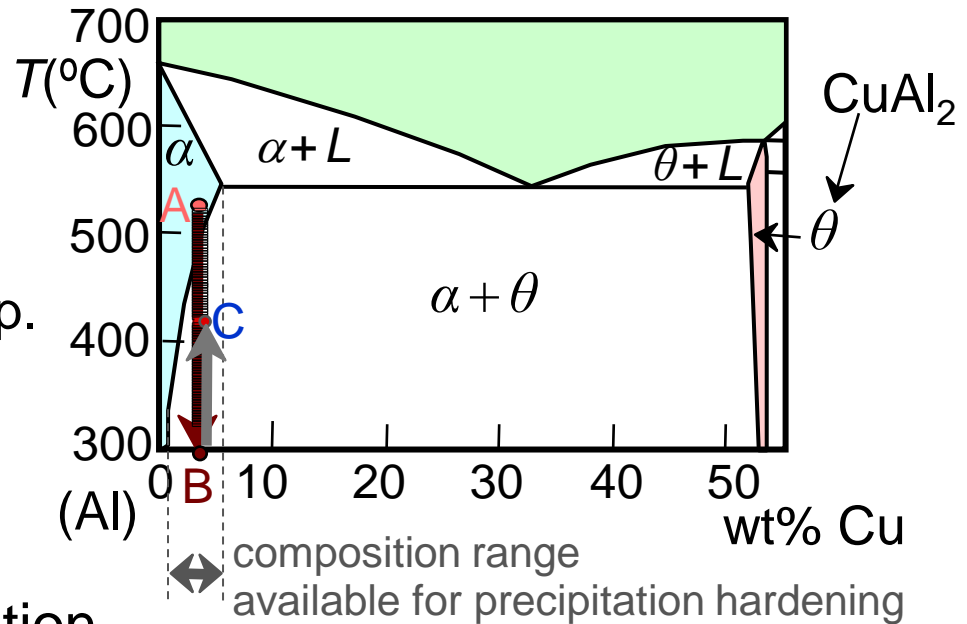
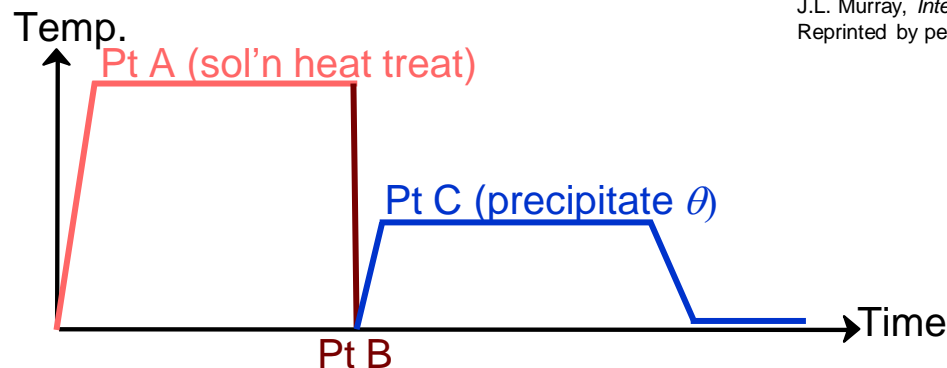


Fig. 11.29, Callister & Rethwisch 10e. (Adapted from J.L. Murray, *International Metals Review* 30, p.5, 1985. Reprinted by permission of ASM International.)

- Other alloys that precipitation harden:

- Cu-Be
- Cu-Sn
- Mg-Al



Adapted from Fig. 11.27, Callister & Rethwisch 10e.

# Influence of Precipitation Heat Treatment on $TS$ , % $EL$

- 2014 Al Alloy:
- Maxima on  $TS$  curves.
- Increasing  $T$  accelerates process.
- Minima on % $EL$  curves.

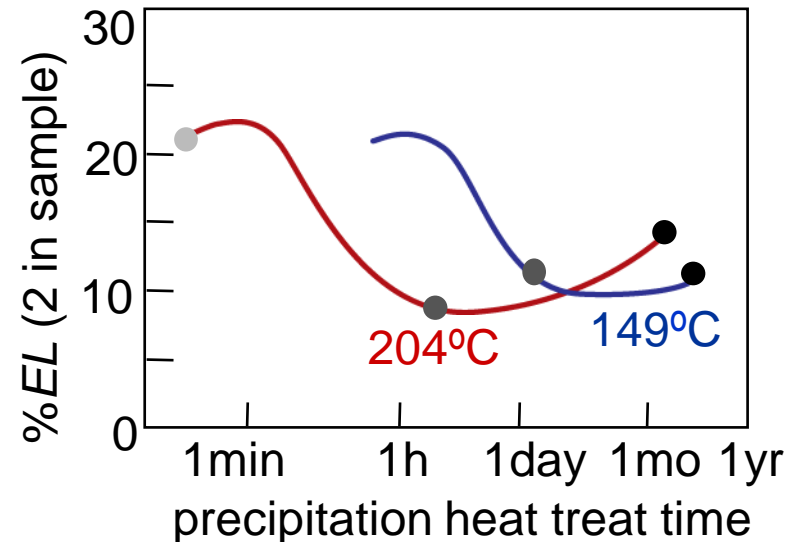
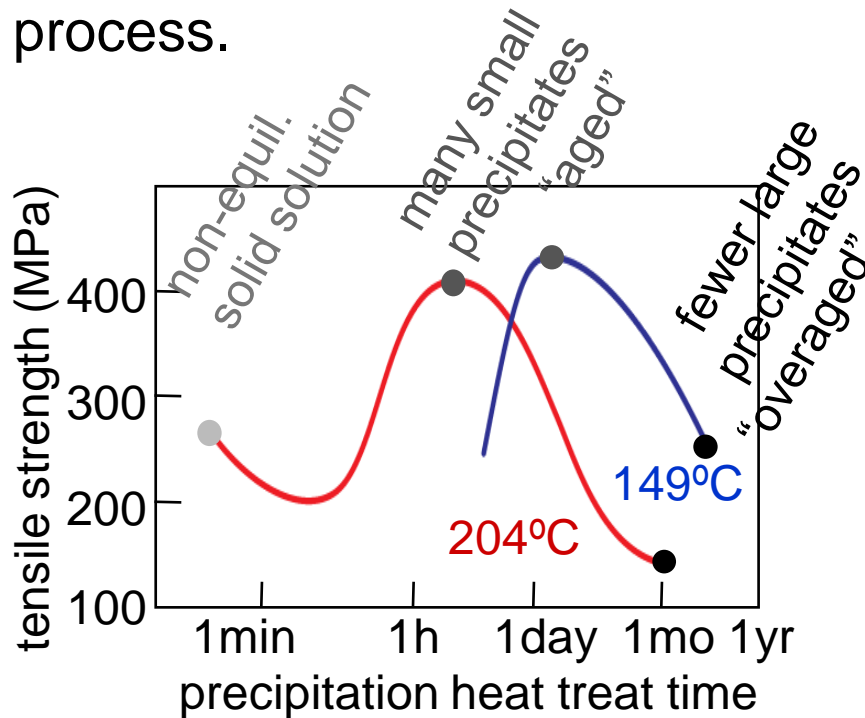
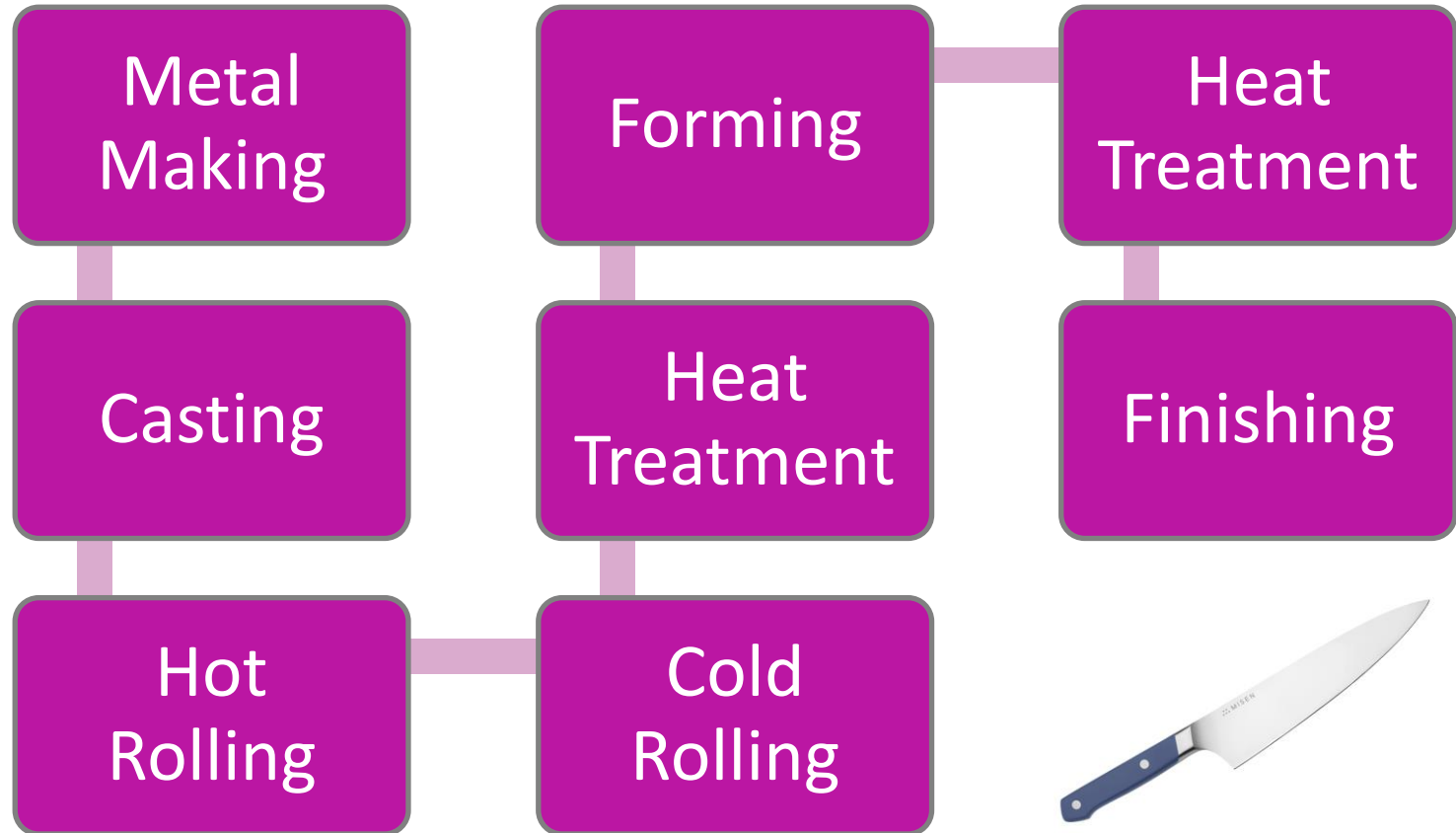


Fig. 11.32, Callister & Rethwisch 10e. [Adapted from *Metals Handbook: Properties and Selection: Nonferrous Alloys and Pure Metals*, Vol. 2, 9th ed., H. Baker (Managing Ed.), 1979. Reproduced by permission of ASM International, Materials Park, OH.]

# Summary

- ❑ Reduction action is applied to most metal making processes.
- ❑ Four casting methods: sand casting, investment casting, die casting, continuous casting.
- ❑ Four forming methods: forging, rolling, drawing and extrusion.
- ❑ Four important heat treatments: annealing, quenching, tempering, aging (precipitation hardening).
- ❑ Annealing includes stress relief, spheroidization, full annealing, process annealing.
- ❑ Quenching & Tempering are typically combined to produce high strength and balanced toughness of materials.
- ❑ Precipitation hardening or aging is strengthening materials due to formation of precipitate particles.

# How to produce a kitchen knife?



# Questions?