

#### **Description**

#### **Image**







## Caption

1. Bubble wrap © PublicDomainPictures at Pixabay [Public domain] 2. Cable insulation © Byrev at Pixabay [Public domain] 3. PE bottles © HebiFot at Pixabay [Public domain]

#### The material

POLYETHYLENE, (-CH2-)n, first synthesized in 1933, looks like the simplest of molecules, but the number of ways in which the - CH2 - units can be linked is large. It is the first of the polyolefins, the bulk thermoplastic polymers that account for a dominant fraction of all polymer consumption. Polyethylene is inert, and extremely resistant to fresh and salt water, food, and most water-based solutions. Because of this it is widely used in household products, food containers like Tupperware and chopping boards. Polyethylene is cheap, and particularly easy to mold and fabricate. It accepts a wide range of colors, can be transparent, translucent or opaque, has a pleasant, slightly waxy feel, can be textured or metal coated, but is difficult to print on.

#### Composition (summary)

(-CH2-CH2-)n

#### **General properties**

Density	939	-	960	kg/m^3
Price	* 1,4	-	1,45	EUR/kg
Date first used	1936			

#### **Mechanical properties**

0,621	-	0,896	GPa
* 0,218	-	0,314	GPa
2,15	-	2,25	GPa
* 0,418	-	0,434	
17,9	-	29	MPa
20,7	-	44,8	MPa
19,7	-	31,9	MPa
200	-	800	% strain
5,4	-	8,7	HV
21	-	23	MPa
	* 0,218 2,15 * 0,418 17,9 20,7 19,7 200 5,4	* 0,218 - 2,15 - * 0,418 - 17,9 - 20,7 - 19,7 - 200 - 5,4 -	* 0,218 - 0,314 2,15 - 2,25 * 0,418 - 0,434 17,9 - 29 20,7 - 44,8 19,7 - 31,9 200 - 800 5,4 - 8,7

GRANTA EDUPACK					
Fracture toughness	*	1,44	-	1,72	MPa.m^0.5
Mechanical loss coefficient (tan delta)	*	0,0446	-	0,0644	
Thermal properties					
Melting point		125	_	132	°C
Glass temperature		-125	_	-90	°C
Maximum service temperature		90	_	110	°C
Minimum service temperature		-123	_	-73,2	°C
Thermal conductor or insulator?		Good ins	sulate		
Thermal conductivity		0,403	-	0,435	W/m.°C
Specific heat capacity		1,81e3	-	1,88e3	J/kg.°C
Thermal expansion coefficient		126	-	198	µstrain/°C
Electrical properties					
Electrical properties Electrical conductor or insulator?		Good ins	sulat	or	
Electrical conductor or insulator?		3,3e22		эг 3e24	uohm em
Dielectric constant (relative permittivity)		2,2	-		µohm.cm
		2,2 3e-4	-	2,4	
Dissipation factor (dielectric loss tangent)					NAV //ww
Dielectric strength (dielectric breakdown)		17,7	-	19,7	MV/m
Optical properties					
Transparency		Transluc	ent		
Refractive index		1,5	-	1,52	
Critical Materials Risk					
High critical material risk?		No			
Processability					
Castability		1	-	2	
Moldability		4	-	5	
Machinability		3	-	4	
Weldability		5			
Durability: water and aqueous solutions					
Water (fresh)		Excellen	t		
Water (salt)		Excellen			
Soils, acidic (peat)		Excellen			
Soils, alkaline (clay)		Excellen			
Wine		Excellen			
			•		
Durability: acids					
Acetic acid (10%)		Excellen	t		
Acetic acid (glacial)		Excellen	t		
Citric acid (10%)		Excellen	t		
Hydrochloric acid (10%)		Excellen	t		
Hydrochloric acid (36%)		Excellen	t		



# Polyethylene (PE)

Hydrofluoric acid (40%)	Excellent
Nitric acid (10%)	Excellent
Nitric acid (70%)	Limited use
Phosphoric acid (10%)	Excellent
Phosphoric acid (85%)	Excellent
Sulfuric acid (10%)	Excellent
Sulfuric acid (70%)	Excellent

## **Durability: alkalis**

Sodium hydroxide (10%)	Excellent
Sodium hydroxide (60%)	Excellent

# **Durability: fuels, oils and solvents**

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# Durability: alcohols, aldehydes, ketones

Acetaldehyde	Excellent
Acetone	Limited use
Ethyl alcohol (ethanol)	Excellent
Ethylene glycol	Excellent
Formaldehyde (40%)	Excellent
Glycerol	Excellent
Methyl alcohol (methanol)	Excellent

## **Durability: halogens and gases**

Chlorine gas (dry)	Limited use
Fluorine (gas)	Unacceptable
O2 (oxygen gas)	Unacceptable
Sulfur dioxide (gas)	Excellent

# **Durability: built environments**

Industrial atmosphere	Excellent
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## Polyethylene (PE)

Rural atmosphere	Excellent
Marine atmosphere	Excellent
UV radiation (sunlight)	Fair

## **Durability: flammability**

Flammability	Highly flammable
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### **Durability: thermal environments**

Tolerance to cryogenic temperatures	Unacceptable
Tolerance up to 150 C (302 F)	Acceptable
Tolerance up to 250 C (482 F)	Unacceptable
Tolerance up to 450 C (842 F)	Unacceptable
Tolerance up to 850 C (1562 F)	Unacceptable
Tolerance above 850 C (1562 F)	Unacceptable

## Geo-economic data for principal component

Annual world production, principal component	4,04e7			tonne/yr
Reserves, principal component	* 1,66e9	-	1,88e9	tonne

### Primary material production: energy, CO2 and water

Embodied energy, primary production	76,1	-	83,9	MJ/kg
CO2 footprint, primary production	1,77	-	1,95	kg/kg
Water usage	* 55,3	-	61,1	l/kg

# Material processing: energy

Polymer extrusion energy	* 5,9	-	6,52	MJ/kg
Polymer molding energy	* 20,8	-	23	MJ/kg
Coarse machining energy (per unit wt removed)	* 0,688	-	0,76	MJ/kg
Fine machining energy (per unit wt removed)	* 2,6	-	2,88	MJ/kg
Grinding energy (per unit wt removed)	* 4,73	-	5,23	MJ/kg

## **Material processing: CO2 footprint**

Polymer extrusion CO2	* 0,442	-	0,489	kg/kg
Polymer molding CO2	* 1,56	-	1,73	kg/kg
Coarse machining CO2 (per unit wt removed)	* 0,0516	-	0,057	kg/kg
Fine machining CO2 (per unit wt removed)	* 0,195	-	0,216	kg/kg
Grinding CO2 (per unit wt removed)	* 0,355	-	0,392	kg/kg

## Material recycling: energy, CO2 and recycle fraction

Recycle	<b>√</b>			
Embodied energy, recycling	* 25,4	-	28,1	MJ/kg
CO2 footprint, recycling	* 0,898	-	0,993	kg/kg
Recycle fraction in current supply	8,02	-	8,86	%
Downcycle	<b>√</b>			
Combust for energy recovery	<b>√</b>			
Heat of combustion (net)	* 44	-	46,2	MJ/kg



Combustion CO2	* 3,06 - 3,22 kg/kg
Landfill	✓
Biodegrade	×
Toxicity rating	Non-toxic

# A renewable resource? Environmental notes

PE is FDA compliant - indeed it is so non-toxic that it can be embedded in the human body (heart valves, hip-joint cups, artificial artery). PE, PP and PVC are made by processes that are relatively energy-efficient, making them the least energy-intensive of commodity polymers. The ethylene from which it is made at present is an oil derivative, but PE can be produced from renewable resources - from alcohol derived from the fermentation of sugar or starch, for instance. Its utility per kilogram far exceeds that of gasoline or fuel-oil (and its energy is stored and still accessible), so that production from oil will not disadvantage it in the near future. Polyethylene is readily recyclable if it has not been coated with other materials, and - if contaminated - it can be incinerated to recover the energy it contains.

#### Recycle mark





#### Supporting information

#### Design guidelines

PE is commercially produced as film, sheet, rod, foam and fiber. Drawn PE fiber has exceptional mechanical stiffness and strength, exploited in geo-textile and structural uses. PE is a good electrical insulator with low dielectric loss, so suitable for containers for microwave cooking. It has poor resistance to aromatics and chlorine; it is slow burning in fire. PE is cheap, easy to form, biologically inert and recyclable; it is one of the materials of the next 20 years.

#### Technical notes

Low density polyethylene (LDPE), used for film and packaging, has branched chains which do not pack well, making it less dense than water. Medium (MDPE) and High (HDPE) density polyethylenes have longer, less branched chains, making them stiffer and stronger; they are used for containers and pipes. Modern catalysis allows side-branching to be suppressed and molecular length to be controlled precisely, permitting precise tailoring both of the processing properties critical for drawing, blow molding, injection molding or extrusion and the use-properties of softening temperature, flexibility and toughness. Linear low-density polyethylene (LLPDE) is an example. In its pure form it is less resistant to organic solvents, but even this can be overcome by converting its surface to a fluoro-polymer by exposing it to fluorine gas. Treated in this way (when it known is known as 'Super PE') it can be used for petrol tanks in cars and copes with oil, cleaning fluid, cosmetics and that most corrosive of substances: cola concentrate. Very low density polyethylene (VDLPE) is similar to EVA and plasticized PVC.

#### Typical uses

Oil container, street bollards, milk bottles, toys, beer crate, food packaging, shrink wrap, squeeze tubes, disposable clothing, plastic bags, paper coatings, cable insulation, artificial joints, and as fibers - low cost ropes and packing tape reinforcement.

#### **Tradenames**

Alathon, Aquathene, Bapolene, Dowlex, Eltex, Empee, Eraclene, Ferrene, Fortiflex, HiVal, Hid, Kemcor, Lacqtene, Lupolen, Marlex, Nortuff, Novapol, Paxon, Petrothene, Polyfort, Rigidex, Sclair, Stamylyn, Statoil, Unival, Zemid



# Polyethylene (PE)

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Reference

ProcessUniverse
Producers