

Variational Monte Carlo (VMC) for AP3081 D

Bas NIJHOLT

April 5, 2014

Date started: March 6, 2014
Instructor: Dr. J.M. Thijssen

Abstract

The potential-energy curve for the electronic ground state of the hydrogen molecule has been calculated using the quantum Monte Carlo (QMC) method of solving the Schrödinger equation, with the use of the Born-Oppenheimer or any other adiabatic approximations. The wavefunction sampling was carried out in a 6-dimensional configuration space of the four particles (two electrons and two protons).

1 Introduction

The determination of energies for molecular systems is a problem of general interest in chemistry and physics. We report here calculations of the ground state energy for the simplest molecular system containing a electron pair bond using the Variational Monte Carlo (VMC) method using the Born-Oppenheimer approximation. Accurate potential energy curves as functions of the inter-atomic distance have been computed for the ground state of the hydrogen molecule. These calculations also provide us with an example of a computationally demanding problem which is well suited for parallel computing. There is a long history of increasingly accurate theoretical calculations of the energy of the hydrogen molecule and increasingly accurate experimental measurements of the dissociation energy. The history of accurate calculations of energies for H_2 begins with the 1933 work of James and Coolidge [2] which represented one of the first successes in solving the Schrödinger equation for molecules. In the 1960's, more accurate results for the hydrogen molecule were obtained by Kolos and Roothaan [6] and by Kolos and Wolniewicz [4, 5, 7, 8] which established the foundation for future calculations.

The effectiveness of the VMC method is demonstrated by applying it to a very simple quantum mechanical system, the harmonic oscillator from which we know the exact results.

2 Variational Monte Carlo

Variational Monte Carlo (VMC) is based on a direct application of Monte Carlo integration to explicitly correlated many-body wavefunctions. The variational principle of quantum mechanics, derived in the following section, states that the energy of a trial wavefunction will be greater than or equal to the energy of the exact wavefunction. Optimised forms for many-body wavefunctions enable the accurate determination of expectation values.

2.1 Variational principle

In quantum mechanics, the variational method is one way of finding approximations to the lowest energy eigenstate or ground state, and some excited states. The basis for this method is the variational principle [1, 9]. The method consists in choosing a "trial wavefunction" depending on one or more parameters, and finding the values of these parameters for which the expectation value of the energy is the lowest possible. The wavefunction obtained by fixing the parameters to such values is then an approximation to the ground state wavefunction, and the expectation value of the energy in that state is an upper bound to the ground state energy.

The variational principle of quantum mechanics may be derived by expanding a normalized trial wavefunction, ψ_T , in terms of the exact normalized eigenstates of the Hamiltonian

$$\psi_T = \sum_{i=0}^{\infty} c_i \psi_i \quad , \quad (1)$$

where the expansion coefficients, c_i , are normalised

$$\sum_{i=0}^{\infty} |c_i|^2 = 1 \quad . \quad (2)$$

The expectation of the many-body Hamiltonian, \hat{H} , may be evaluated

$$\langle \psi_T | \hat{H} | \psi_T \rangle = \langle \sum_i c_i \psi_i | \hat{H} | \sum_j c_j \psi_j \rangle = \sum_i \sum_j c_i^* c_j \langle \psi_i | \hat{H} | \psi_j \rangle = \sum_i |c_i|^2 \epsilon_i \quad (3)$$

where

$$\epsilon_i = \langle \psi_i | \hat{H} | \psi_i \rangle \quad (4)$$

The expectation value of a trial wavefunction with the Hamiltonian must therefore be greater than or equal to the true ground state energy.

2.2 Monte Carlo integration

Before the expectation value of a trial wavefunction with the many-body Hamiltonian may be computed, the integral must be transformed into a form suitable

for Monte Carlo integration. Trial wavefunctions, ψ_T , are dependent on the set of N electron positions, $\mathbf{R} = \{\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N\}$. The expectation value is given by

$$E = \frac{\int \psi_T^* \hat{H} \psi_T d\mathbf{R}}{\int \psi_T^* \psi_T d\mathbf{R}} \quad (5)$$

which may be rewritten in an importance sampled form in terms of the probability density $|\psi_T|^2$ as

$$E = \frac{\int |\psi_T|^2 \frac{\hat{H} \psi_T}{\psi_T} d\mathbf{R}}{\int |\psi_T|^2 d\mathbf{R}} \quad (6)$$

The Metropolis algorithm samples configurations - sets of electron positions \mathbf{R} - from the probability distribution $|\psi_T|^2$, and the variational energy is obtained by averaging the "local energy" E_L over the set of configurations $\{\mathbf{R}\}$ as

$$E_L(\mathbf{R}) = \frac{\hat{H} \psi_T(\mathbf{R})}{\psi_T(\mathbf{R})} \quad (7)$$

and

$$E = \frac{1}{N} \sum E_L(\mathbf{R}_i) \quad (8)$$

2.3 The local energy

The local energy, $E_L(\mathbf{R})$, equation 7 is one of the central quantities in quantum Monte Carlo (QMC) methods. It occurs in both the variational and diffusion Monte Carlo algorithms and its properties are exploited to optimise trial wavefunctions. The local energy has the useful property that for an exact eigenstate of the Hamiltonian, the local energy is constant. For a general trial wavefunction the local energy is not constant and the variance of the local energy is a measure of how well the trial wavefunction approximates an eigenstate.

Determination of the local energy is one of the most computationally costly operations performed in QMC calculations. Application of the Hamiltonian to the trial wavefunction requires computation of the second derivatives of the wavefunction and the calculation of the electron-electron and electron-ion potentials.

2.4 Trial wavefunction

The choice of trial wavefunction is critical in VMC calculations. All observables are evaluated with respect to the probability distribution $|\Psi_T(\mathbf{R})|^2$. The trial wavefunction, $\Psi_T(\mathbf{R})$, must well approximate an exact eigenstate for all \mathbf{R} in order that accurate results are obtained. Improved trial wavefunctions also improve the importance sampling, reducing the cost of obtaining a certain statistical accuracy. Quantum Monte Carlo methods are able to exploit trial wavefunctions of arbitrary forms. Any wavefunction that is physical and for

which the value, gradient and Laplacian of the wavefunction may be efficiently computed can be used.

The power of Quantum Monte Carlo methods lies in the flexibility of the form of the trial wavefunction. In early studies the wavefunction was taken to be a Jastrow function [3] which is of the form

$$\psi = \exp \left[\sum_{i < j}^N -u(r_{ij}) \right] \quad . \quad (9)$$

2.5 The VMC algorithm

The VMC algorithm consists of two distinct phases. In the first a walker consisting of an initially random set of electron positions is propagated according to the Metropolis algorithm, in order to equilibrate it and begin sampling $|\Psi|^2$. In the second phase, the walker continues to be moved, but energies and other observables are also accumulated for later averaging and statistical analysis. The procedure is as follows [10].

1. Generate initial configuration using random positions for the electrons
2. For every electron in the configuration:
 - (a) Propose a move from \mathbf{r} to \mathbf{r}'
 - (b) Compute $w = |\Psi(\mathbf{r}')/\Psi(\mathbf{r})|^2$
 - (c) Accept or reject move according to Metropolis probability $\min(1, w)$
 - (d) After equilibration phase accumulate the contribution to the local energy
3. Repeat step 2 until sufficient data accumulated

2.6 Harmonic oscillator

2.7 Hydrogen atom

2.8 Helium atom

2.9 Hydrogen molecule H_2

3 Computer Simulation

4 Results

5 Conclusions

References

- [1] David Jeffrey Griffiths and Edward G Harris. *Introduction to quantum mechanics*, volume 2. Prentice Hall New Jersey, 1995.
- [2] Hubert M James and Albert Sprague Coolidge. The ground state of the hydrogen molecule. *The Journal of Chemical Physics*, 1(12):825–835, 1933.
- [3] Robert Jastrow. Many-body problem with strong forces. *Physical Review*, 98(5):1479, 1955.
- [4] W Kol, L Wolniewicz, et al. Accurate adiabatic treatment of the ground state of the hydrogen molecule. *The Journal of Chemical Physics*, 41(12):3663–3673, 1964.
- [5] W Kol, L Wolniewicz, et al. Potential-energy curves for the $x^1\sigma_g^+$, $b^3\sigma_u^+$, and $c^1\pi_u$ states of the hydrogen molecule. *The Journal of Chemical Physics*, 43(7):2429–2441, 1965.
- [6] W. Kolos and C. C. J. Roothaan. Accurate electronic wave functions for the h_2 molecule. *Reviews of Modern Physics*, 32:219–232, Apr 1960.
- [7] W Kolos and L Wolniewicz. Nonadiabatic theory for diatomic molecules and its application to the hydrogen molecule. *Reviews of Modern Physics*, 35(3):473, 1963.
- [8] W Kolos and L Wolniewicz. Improved theoretical ground-state energy of the hydrogen molecule. *The Journal of Chemical Physics*, 49(1):404–410, 1968.
- [9] Jun John Sakurai and San Fu Tuan. *Modern quantum mechanics*, volume 104. Addison-Wesley Reading (Mass.), 1994.
- [10] J.M. Thijssen. *Computational Physics*. Cambridge University Press, 2007.

6 Appendix

6.1 Derivation of local energies

We will present the derivations for the atoms studied in this report To evaluate the local energy we must work out E_L in

$$\langle E \rangle = \frac{\int dR \Psi_T^2(R) E_L(R)}{\int dR \Psi_T^2(R)} \quad (10)$$

in which E_L is

$$E_L(R) = \frac{\hat{H}\Psi_T(R)}{\Psi_T(R)} \quad (11)$$

$$\hat{H} = \sum_{i=1}^N \frac{p_i^2}{2m} + \frac{1}{2} \sum_{i \neq j=1}^N \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|} + \sum_i V_{e-b}(\mathbf{r}_i) \quad (12)$$

6.2 Hydrogen molecule H_2

$$\hat{H} = -\frac{\hbar}{2m} (\nabla_1^2 + \nabla_2^2) - \left[\frac{ke^2}{|\vec{r}_1 + \frac{s}{2}\hat{i}|} + \frac{ke^2}{|\vec{r}_1 - \frac{s}{2}\hat{i}|} + \frac{ke^2}{|\vec{r}_2 + \frac{s}{2}\hat{i}|} + \frac{ke^2}{|\vec{r}_2 - \frac{s}{2}\hat{i}|} \right] + \frac{ke^2}{|\vec{r}_1 + \vec{r}_2|} \quad (13)$$

in which the first term is the kinetic energy of the two electrons. The part in the square brackets is the attraction between the nucleus and the electrons and the last term is the Coulomb interaction between the electrons. Setting the constants to unity, and setting $\vec{r}_{1L} = \vec{r}_1 + \frac{s}{2}\hat{i}$, $\vec{r}_{1R} = \vec{r}_1 - \frac{s}{2}\hat{i}$, $\vec{r}_{2L} = \vec{r}_2 + \frac{s}{2}\hat{i}$, $\vec{r}_{2R} = \vec{r}_2 - \frac{s}{2}\hat{i}$ and $\vec{r}_{12} = \vec{r}_1 - \vec{r}_2$ we obtain

$$\hat{H} = -\frac{1}{2} (\nabla_1^2 + \nabla_2^2) - \left[\frac{1}{|\vec{r}_{1L}|} + \frac{1}{|\vec{r}_{1R}|} + \frac{1}{|\vec{r}_{2L}|} + \frac{1}{|\vec{r}_{2R}|} \right] + \frac{1}{|\vec{r}_{12}|} \quad (14)$$

In the light of the solvability of this Hamiltonian we split it up in two non-interacting Hamiltonians and a interaction term.

$$\hat{H} = \hat{H}_1 + \hat{H}_2 + \hat{H}_{ee} \quad (15)$$

in which $\hat{H}_1 = -\frac{1}{2}\nabla_1^2 - \frac{1}{|\vec{r}_{1L}|} - \frac{1}{|\vec{r}_{1R}|}$, $\hat{H}_2 = -\frac{1}{2}\nabla_2^2 - \frac{2}{|\vec{r}_{1L}|} - \frac{2}{|\vec{r}_{2R}|}$ and $\hat{H}_{ee} = \frac{1}{|\vec{r}_{12}|}$. We have chosen our trail variational wave function $\Psi_{T,\alpha}$ to be

$$\Psi_T(\vec{r}_1, \vec{r}_2) = \phi(\vec{r}_1)\phi(\vec{r}_2)\psi(\vec{r}_1, \vec{r}_2) \quad (16)$$

where $\psi(\vec{r}_1, \vec{r}_2)$ is the Jastrow function

$$\psi(\vec{r}_1, \vec{r}_2) = \psi_{12} = e^{\frac{|\vec{r}_{12}|}{\alpha(1+\beta|\vec{r}_{12}|)}} \quad (17)$$

and $\phi(\vec{r}_1)$ and $\phi(\vec{r}_2)$ are

$$\phi(\vec{r}_1) = e^{-|\vec{r}_{1L}|/a} + e^{-|\vec{r}_{1R}|/a} = \phi_1 = \phi_{1L} + \phi_{1R} \quad (18)$$

$$\phi(\vec{r}_2) = e^{-|\vec{r}_{2L}|/a} + e^{-|\vec{r}_{2R}|/a} = \phi_2 = \phi_{2L} + \phi_{2R} \quad (19)$$

$$E_L(R) = \frac{\hat{H}\Psi_T(R)}{\Psi_T(R)} = \frac{1}{\phi_1\phi_2\psi_{12}} \hat{H}\phi_1\phi_2\psi_{12} \quad (20)$$

We let the Hamiltonians only work on correct terms, which results in

$$\frac{1}{\phi_1\phi_2\psi_{12}}(\hat{H}_1 + \hat{H}_2 + \hat{H}_{ee})\phi_1\phi_2\psi_{12} = \frac{1}{\phi_1\psi_{12}}\hat{H}_1\phi_1\psi_{12} + \frac{1}{\phi_2\psi_{12}}\hat{H}_2\phi_2\psi_{12} + \frac{1}{|\vec{r}_{12}|} \quad (21)$$

Now we first consider

$$\frac{1}{\phi_1\psi_{12}}\hat{H}_1\phi_1\psi_{12} = -\frac{1}{\phi_1\psi_{12}}\frac{1}{2}\nabla_1^2(\phi_1\psi_{12}) - \frac{1}{|\vec{r}_{1L}|} - \frac{1}{|\vec{r}_{1R}|} \quad (22)$$

from which we first consider $\nabla_1^2(\phi_1\psi_{12})$, using the chain rule

$$\nabla_1^2(\phi_1\psi_{12}) = \psi_{12}\nabla_1^2\phi_1 + 2\nabla_1\phi_1 \cdot \nabla_1\psi_{12} + \phi_1\nabla_1^2\psi_{12} \quad (23)$$

where

$$\nabla_1\phi_1 = -\frac{1}{a}\left(e^{-|\vec{r}_{1L}|/a} + e^{-|\vec{r}_{1R}|/a}\right) = -\frac{1}{a}\left(e^{-|\vec{r}_{1L}|/a}\hat{r}_{1L} + e^{-|\vec{r}_{1R}|/a}\hat{r}_{1R}\right) \quad (24)$$

and ¹

$$\nabla_1^2\phi_1 = \left[\frac{1}{a^2} - \frac{2}{ar_{1L}}\right]e^{-r_{1L}/a} + \left[\frac{1}{a^2} - \frac{2}{ar_{1R}}\right]e^{-r_{1R}/a} \quad (25)$$

Similarly we find

$$\nabla_1\psi_{12} = \nabla_1\left(e^{\frac{r_{12}}{\alpha(1+\beta r_{12})}}\right) = \frac{e^{\frac{r_{12}}{\alpha(\beta r_{12}+1)}}}{\alpha(\beta r_{12}+1)^2}\hat{r}_{12} = \frac{\psi_{12}}{\alpha(\beta r_{12}+1)^2}\hat{r}_{12} \quad (26)$$

and

$$\nabla_1^2\psi_{12} = e^{\frac{r_{12}}{\alpha(1+\beta r_{12})}}\frac{2\alpha\beta r_{12} + 2\alpha + r_{12}}{r_{12}\alpha^2(\beta r_{12}+1)^4} = \frac{(1+2\alpha\beta)r_{12} + 2\alpha}{r_{12}\alpha^2(\beta r_{12}+1)^4}\psi_{12} \quad (27)$$

Now substituting equations 24, 25, 26 and 27 into 23.

$$\begin{aligned} \nabla_1^2(\phi_1\psi_{12}) &= \psi_{12}\left(\left[\frac{1}{a^2} - \frac{2}{ar_{1L}}\right]e^{-r_{1L}/a} + \left[\frac{1}{a^2} - \frac{2}{ar_{1R}}\right]e^{-r_{1R}/a}\right) + \\ &\quad \frac{2}{a}\left(e^{-r_{1L}/a}\hat{r}_{1L} + e^{-r_{1R}/a}\hat{r}_{1R}\right)\frac{\psi_{12}}{\alpha(\beta r_{12}+1)^2}\hat{r}_{12} \\ &\quad + \left(e^{-|\vec{r}_{1L}|/a} + e^{-|\vec{r}_{1R}|/a}\right)\left(\frac{(1+2\alpha\beta)r_{12} + 2\alpha}{r_{12}\alpha^2(\beta r_{12}+1)^4}\psi_{12}\right) \end{aligned} \quad (28)$$

shuffle to get

$$\begin{aligned} \frac{\nabla_1^2(\phi_1\psi_{12})}{\phi_1\psi_{12}} &= \frac{1}{\phi_1}\left(\left[\frac{1}{a^2} - \frac{2}{ar_{1L}}\right]e^{-r_{1L}/a} + \left[\frac{1}{a^2} - \frac{2}{ar_{1R}}\right]e^{-r_{1R}/a}\right) + \\ &\quad \frac{1}{\phi_1}\left(e^{-r_{1L}/a}\hat{r}_{1L} + e^{-r_{1R}/a}\hat{r}_{1R}\right) \cdot \frac{2\hat{r}_{12}}{\alpha a(\beta r_{12}+1)^2} + \frac{(1+2\alpha\beta)r_{12} + 2\alpha}{r_{12}\alpha^2(\beta r_{12}+1)^4} \end{aligned} \quad (29)$$

¹from now on we will omit the vector signs when it's not ambiguous

replace the exponents for the corresponding ϕ to obtain

$$\begin{aligned} \frac{\nabla_1^2(\phi_1\psi_{12})}{\phi_1\psi_{12}} = & \left(\left[\frac{1}{a^2} - \frac{2}{ar_{1L}} \right] \frac{\phi_{1L}}{\phi_1} + \left[\frac{1}{a^2} - \frac{2}{ar_{1R}} \right] \frac{\phi_{1R}}{\phi_1} \right) + \\ & \left(\frac{\phi_{1L}}{\phi_1} \hat{r}_{1L} + \frac{\phi_{1R}}{\phi_1} \hat{r}_{1R} \right) \cdot \frac{2\hat{r}_{12}}{\alpha a(\beta r_{12} + 1)^2} + \frac{(1 + 2\alpha\beta)r_{12} + 2\alpha}{r_{12}\alpha^2(\beta r_{12} + 1)^4} \end{aligned} \quad (30)$$

in which

$$\begin{aligned} & \left[\frac{1}{a^2} - \frac{2}{ar_{1L}} \right] \frac{\phi_{1L}}{\phi_1} + \left[\frac{1}{a^2} - \frac{2}{ar_{1R}} \right] \frac{\phi_{1R}}{\phi_1} \\ &= \frac{1}{a^2} \left[\frac{\phi_{1L}}{\phi_1} + \frac{\phi_{1R}}{\phi_1} \right] - \frac{2}{a\phi_1} \left[\frac{\phi_{1L}}{r_{1L}} + \frac{\phi_{1R}}{r_{1R}} \right] \\ &= \frac{1}{a^2} - \frac{2}{a\phi_1} \left[\frac{\phi_{1L}}{r_{1L}} + \frac{\phi_{1R}}{r_{1R}} \right] \end{aligned} \quad (31)$$

Now adding the equations for the second electron, which are the same ² because of the symmetry of the problem, multiplying with 1/2 and adding the Coulomb interaction terms to get

$$\begin{aligned} E_L = & -\frac{1}{a^2} + \frac{1}{a\phi_1} \left[\frac{\phi_{1L}}{r_{1L}} + \frac{\phi_{1R}}{r_{1R}} \right] + \frac{1}{a\phi_2} \left[\frac{\phi_{2L}}{r_{2L}} + \frac{\phi_{2R}}{r_{2R}} \right] + \\ & \left(\frac{\phi_{1L}}{\phi_1} \hat{r}_{1L} + \frac{\phi_{1R}}{\phi_1} \hat{r}_{1R} - \frac{\phi_{2L}}{\phi_2} \hat{r}_{2L} - \frac{\phi_{2R}}{\phi_2} \hat{r}_{2R} \right) \cdot \frac{\hat{r}_{12}}{\alpha a(\beta r_{12} + 1)^2} \\ & - \frac{(1 + 2\alpha\beta)r_{12} + 2\alpha}{r_{12}\alpha^2(\beta r_{12} + 1)^4} - \left[\frac{1}{|\vec{r}_{1L}|} + \frac{1}{|\vec{r}_{1R}|} + \frac{1}{|\vec{r}_{2L}|} + \frac{1}{|\vec{r}_{2R}|} \right] + \frac{1}{|\vec{r}_{12}|} \end{aligned} \quad (32)$$

with $\alpha = 2$

$$\begin{aligned} E_L = & -\frac{1}{a^2} + \frac{1}{a\phi_1} \left[\frac{\phi_{1L}}{r_{1L}} + \frac{\phi_{1R}}{r_{1R}} \right] + \frac{1}{a\phi_2} \left[\frac{\phi_{2L}}{r_{2L}} + \frac{\phi_{2R}}{r_{2R}} \right] + \\ & \left(\frac{\phi_{1L}\hat{r}_{1L} + \phi_{1R}\hat{r}_{1R}}{\phi_1} - \frac{\phi_{2L}\hat{r}_{2L} + \phi_{2R}\hat{r}_{2R}}{\phi_2} \right) \cdot \frac{\hat{r}_{12}}{2a(\beta r_{12} + 1)^2} \\ & - \frac{(1 + 4\beta)r_{12} + 4}{4r_{12}(\beta r_{12} + 1)^4} - \left[\frac{1}{|\vec{r}_{1L}|} + \frac{1}{|\vec{r}_{1R}|} + \frac{1}{|\vec{r}_{2L}|} + \frac{1}{|\vec{r}_{2R}|} \right] + \frac{1}{|\vec{r}_{12}|} \end{aligned} \quad (33)$$

7 The use of my program

8 Data

²up to one minus sign