

# FLUORESCENT, AIEGEN AND CYTOTOXIC BEHAVIOUR OF ANNULATED POLYCYCLIC HETEROAROMATICS



**BHISMA K. PATEL**



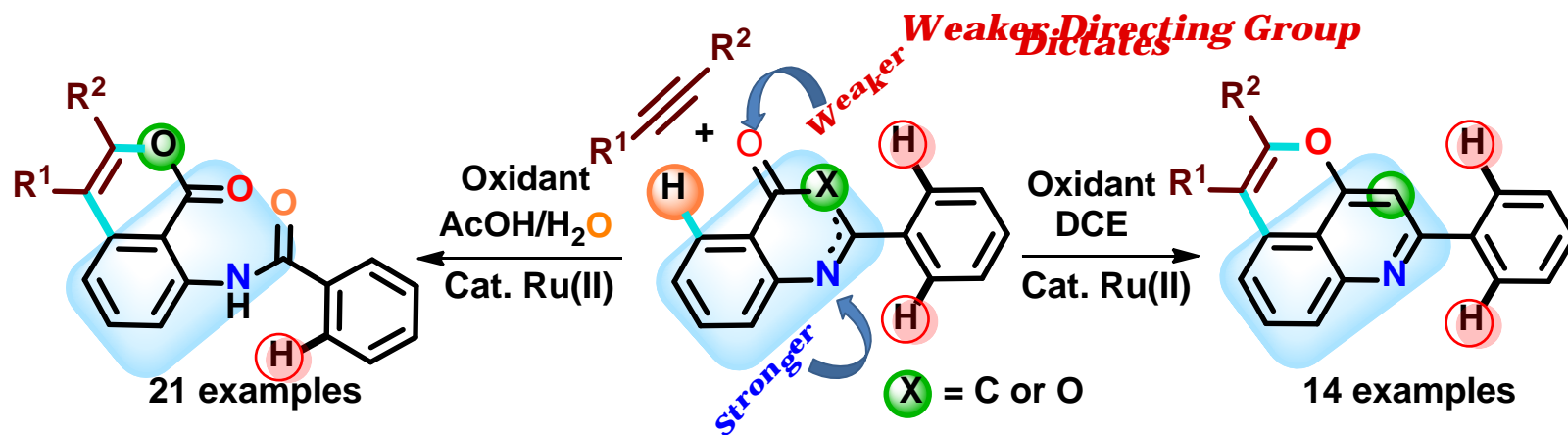
*Indian Academy of Sciences*  
**84th ANNUAL MEETING**

Light Emission: A useful yet mysterious process to the humankind.....



Harvesting energy using molecules.....

# C–H/O–H Annulations of Directing Arenes *via* Weak Coordination



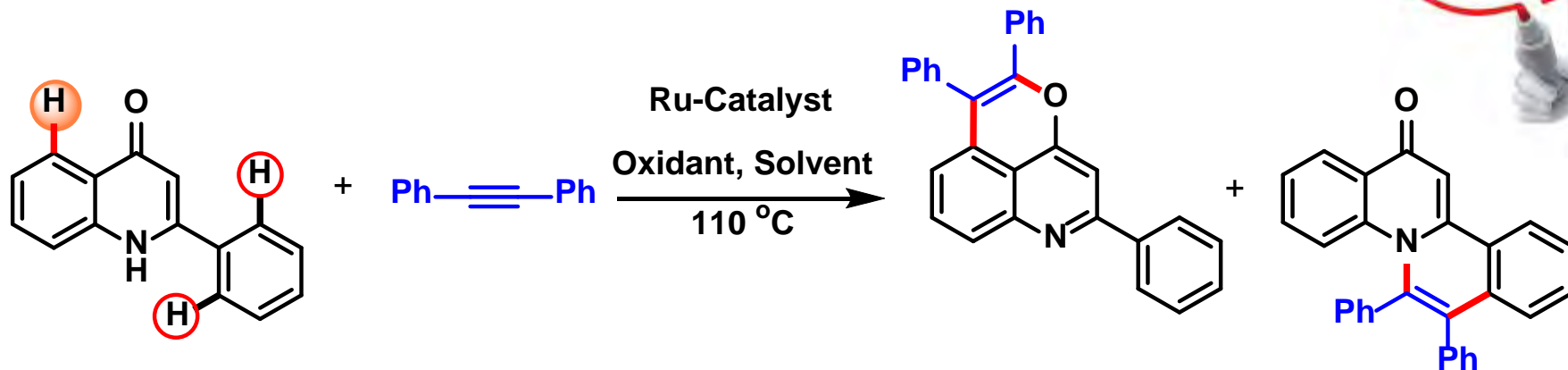
*Org. Lett.* **2015**, *17*, 5678

ACS Editors' Choice



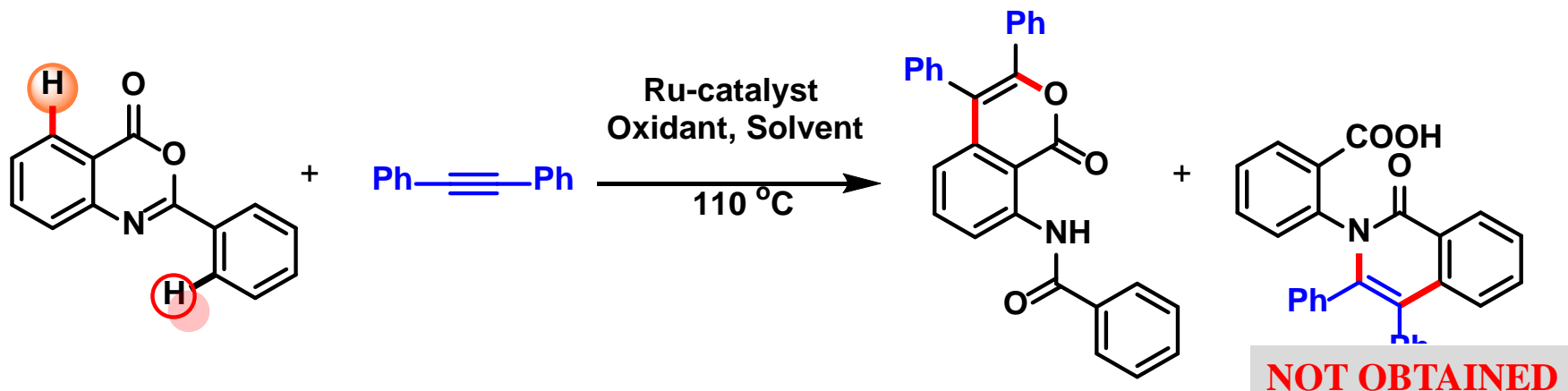
# Substrate Controlled Convergent Annulation

STRATEGY



## Optimized Condition:

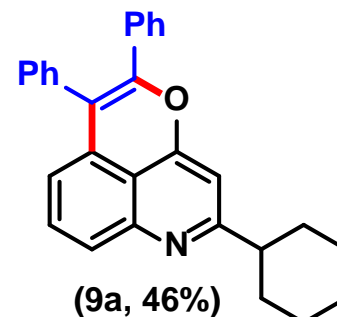
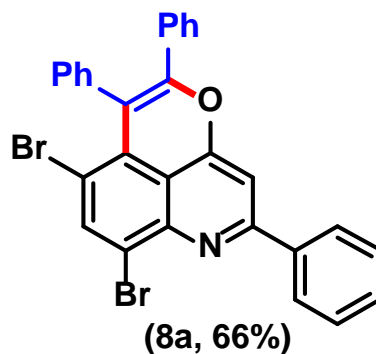
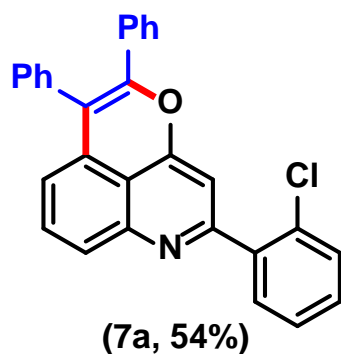
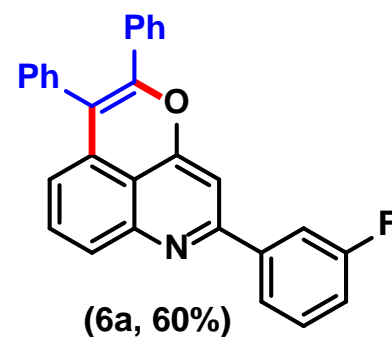
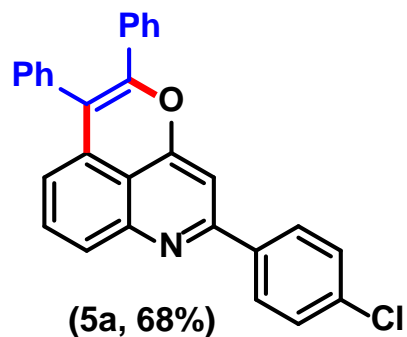
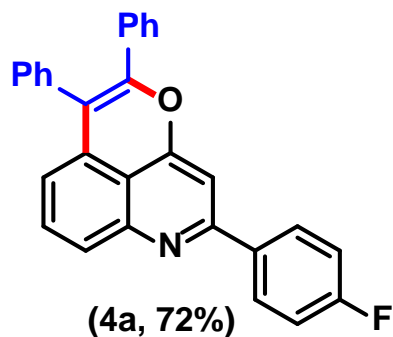
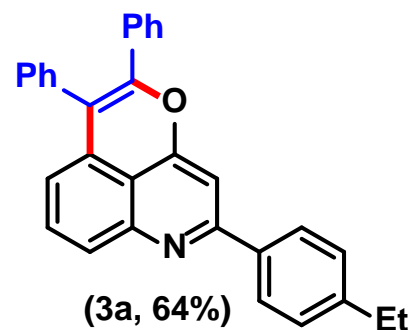
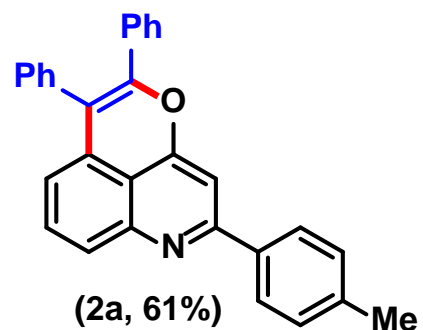
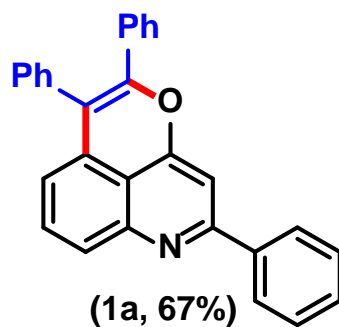
$[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$  (2.0 mol %),  $\text{Cu}(\text{OAc})_2$  (1.2 equiv), DCE,  $110\text{ }^\circ\text{C}$



## Optimized Condition:

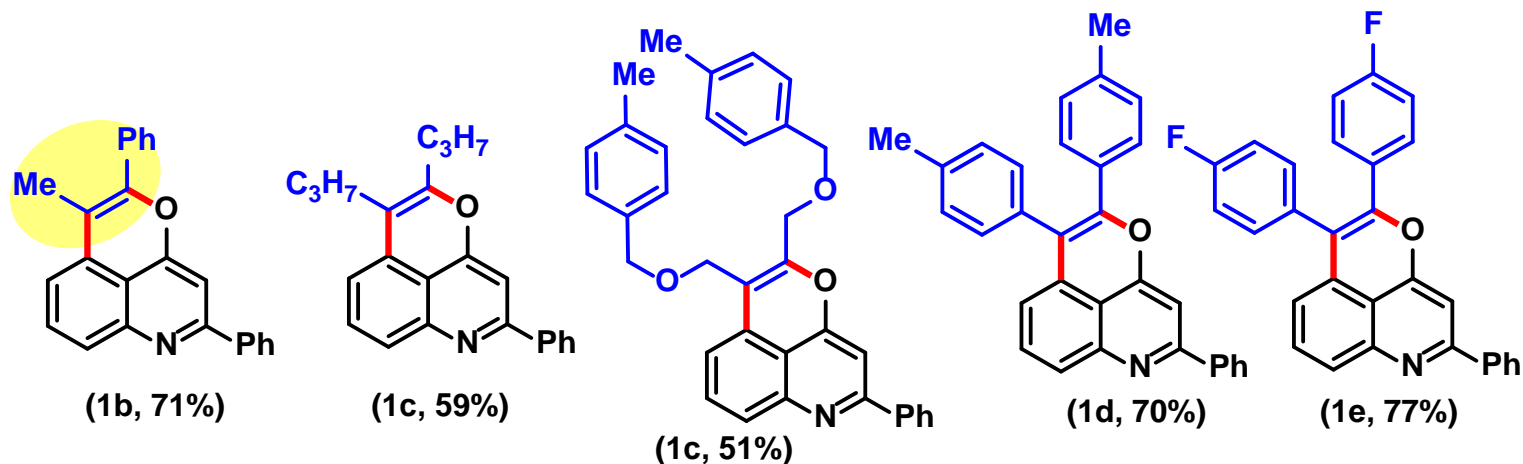
$[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$  (2.0 mol %),  $\text{AgOAc}$  (1.2 equiv),  $\text{AcOH}$ ,  $110\text{ }^\circ\text{C}$

## Scope of 2-Arylquinolinone for C–H/O–H Annulation

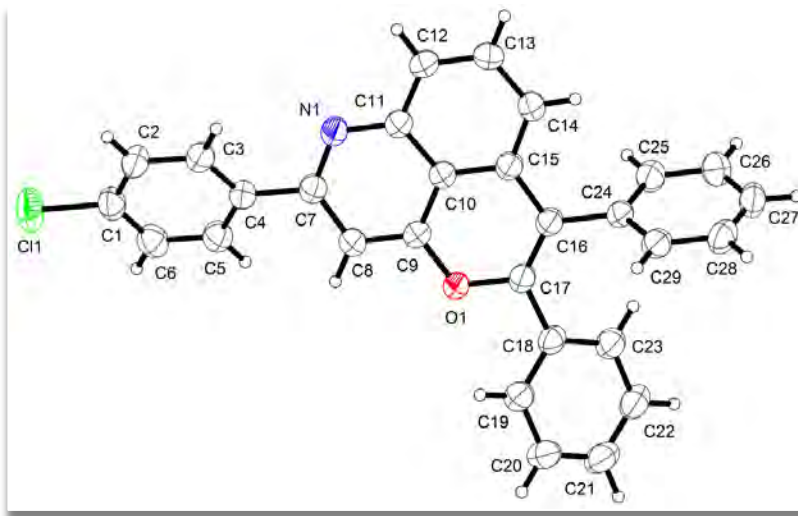




# Scope of Alkynes for Annulation with 2-Phenylquinolinone



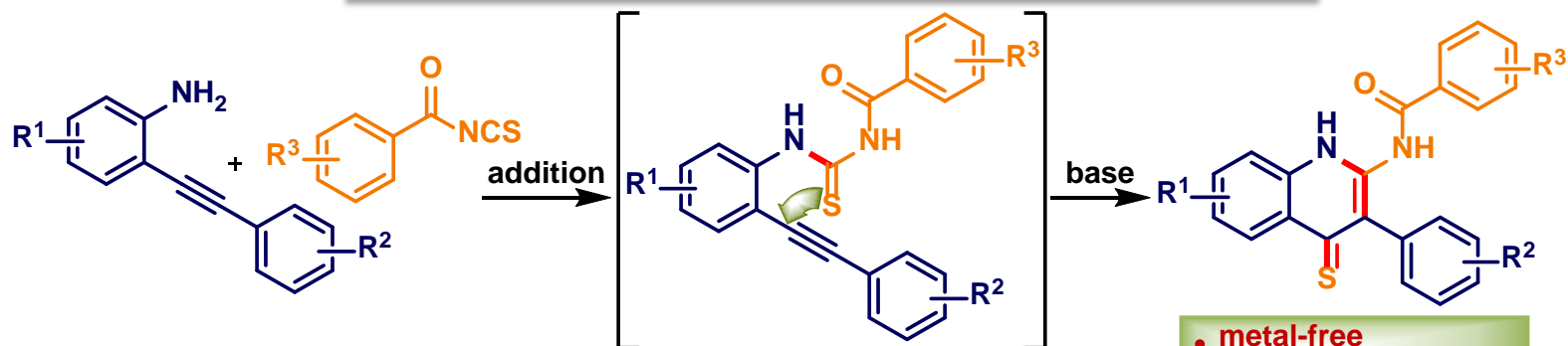
## Crystal structure:



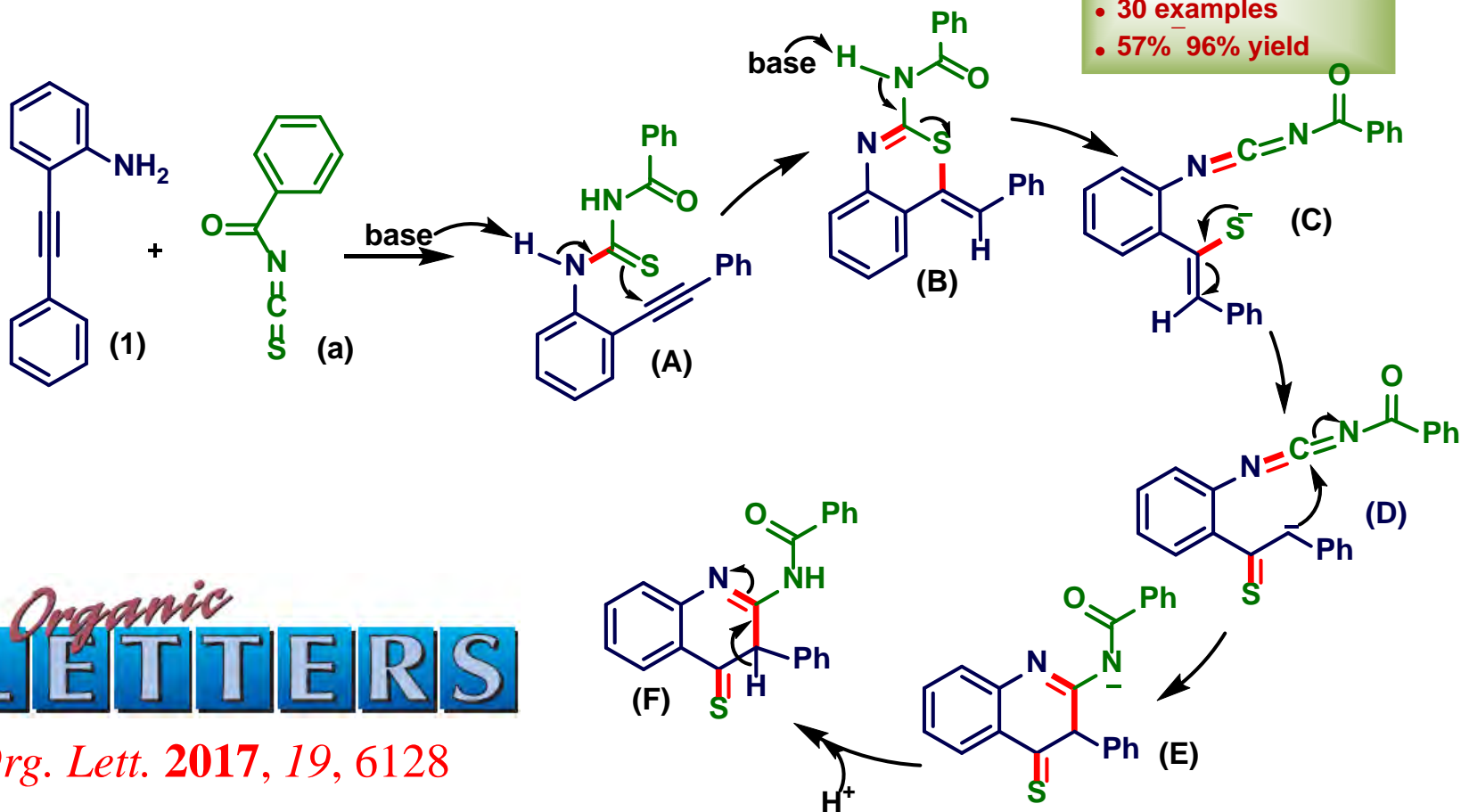
ORTEP view of 5a



# Synthesis of Quinoline-4(1*H*)-thiones

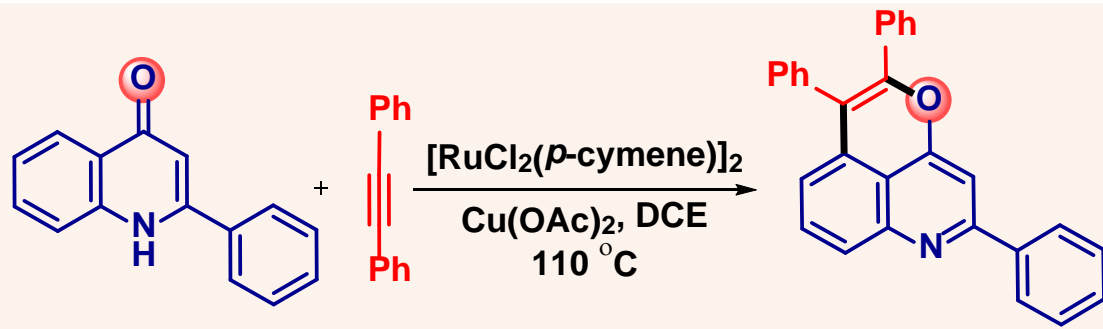


- metal-free
- 100% atom economy
- 30 examples
- 57%–96% yield



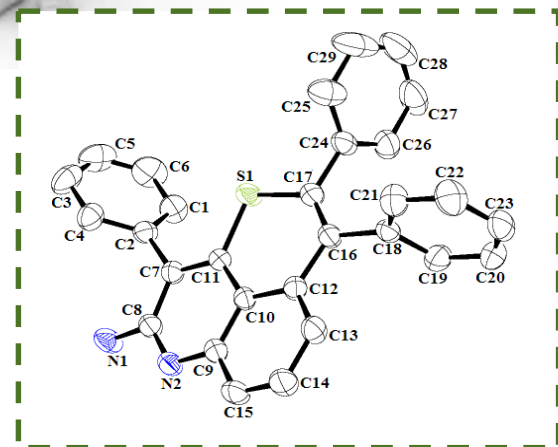
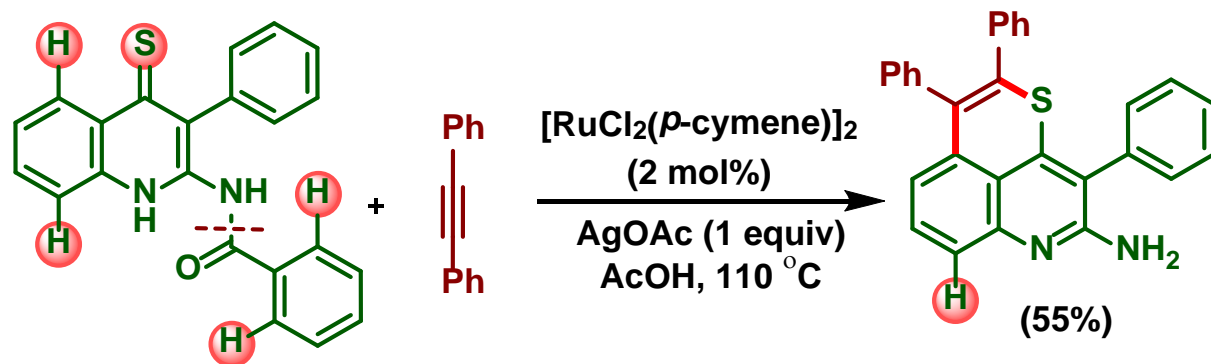
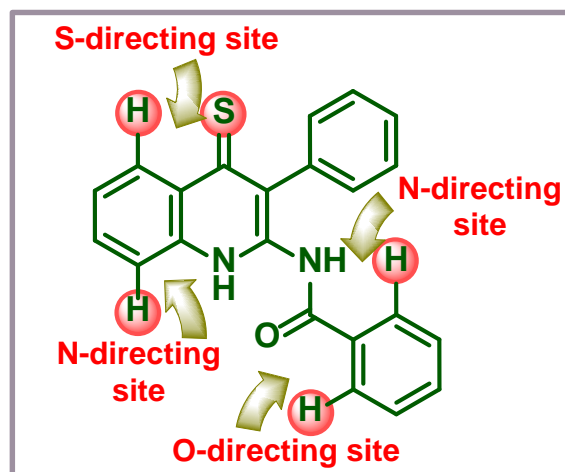
*Organic*  
**LETTERS**

*Org. Lett.* **2017**, *19*, 6128



**Weak co-ordinating site  
dominates over the strong**

*Org. Lett.* **2015**, *17*, 5678

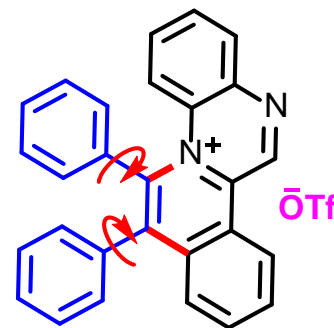
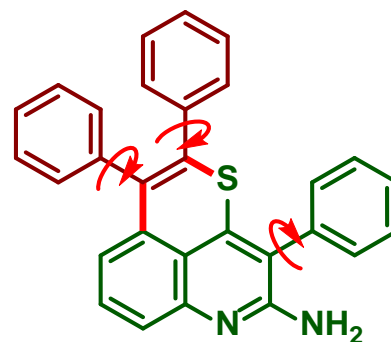
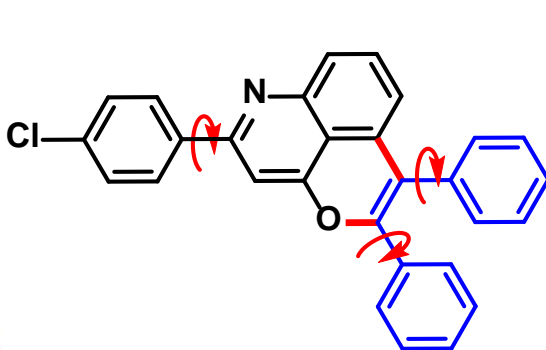
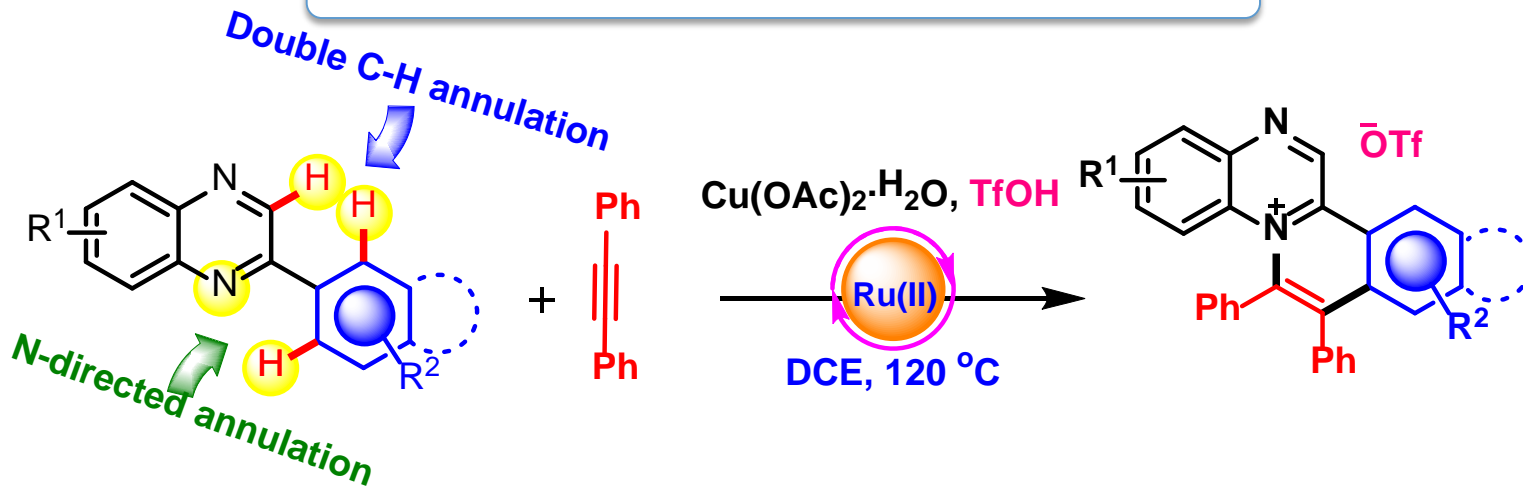


**Again Weak Coordination Dictates.**

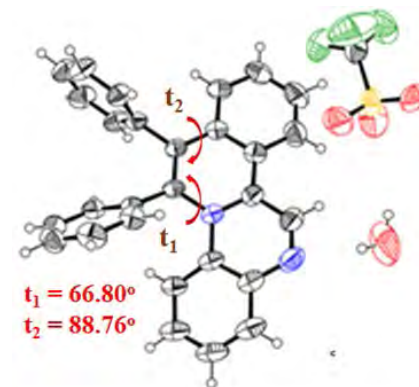
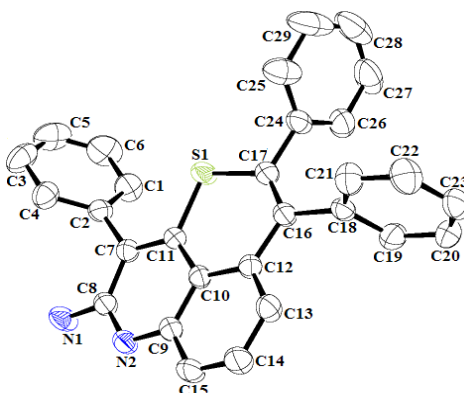
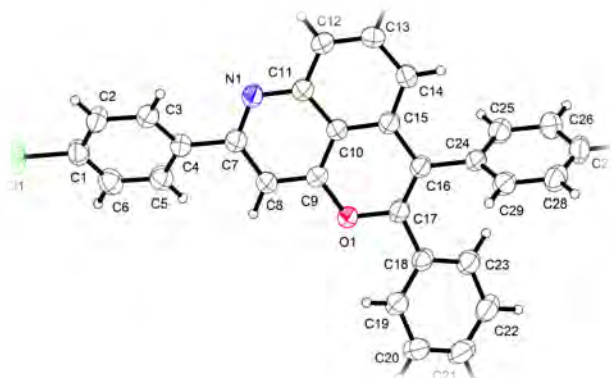
**The first example of C-H, S-H annulation**

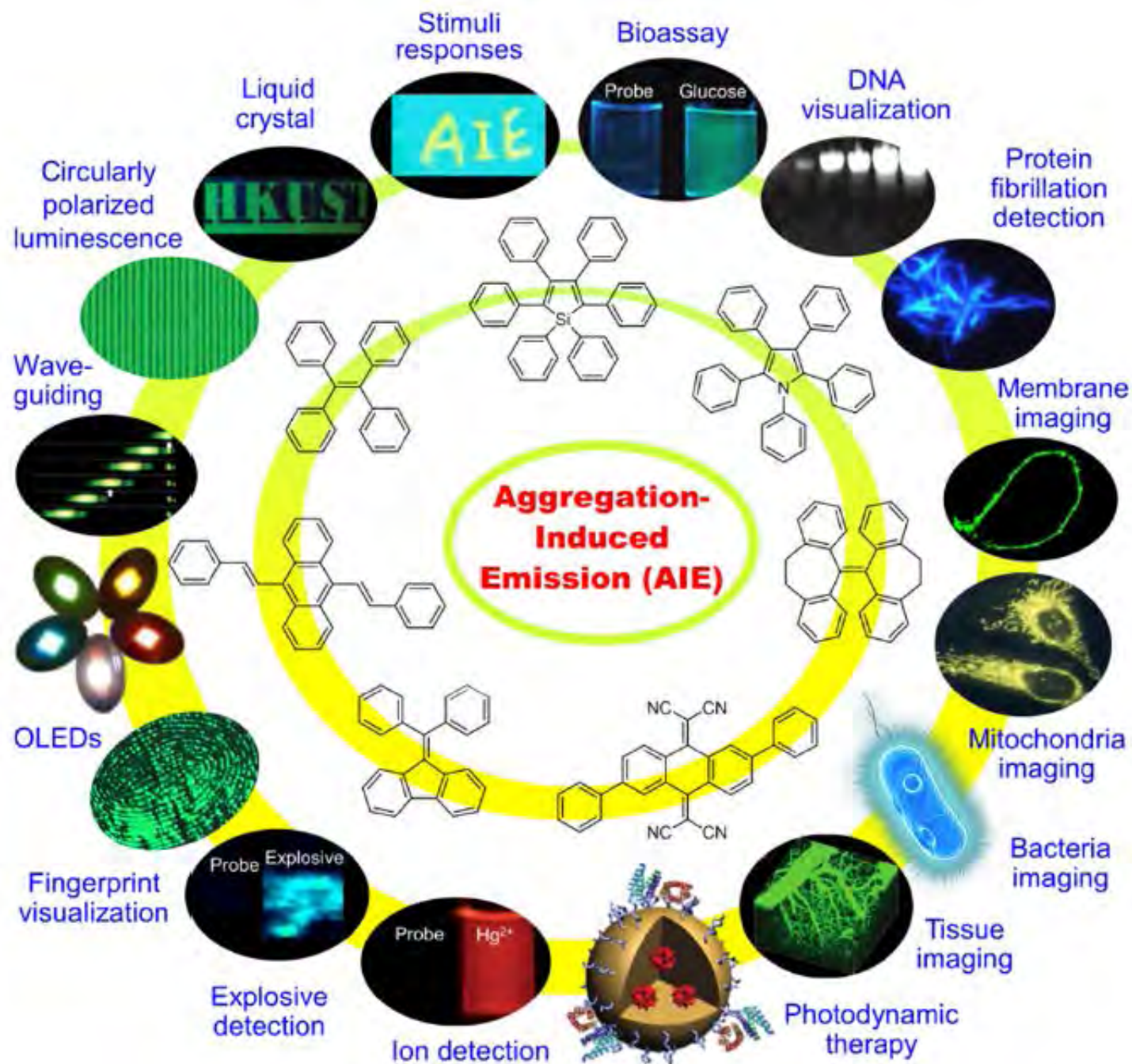


# Heteroarene Directed Annulations



HIGHLY FLUORESCENT MOLECULES

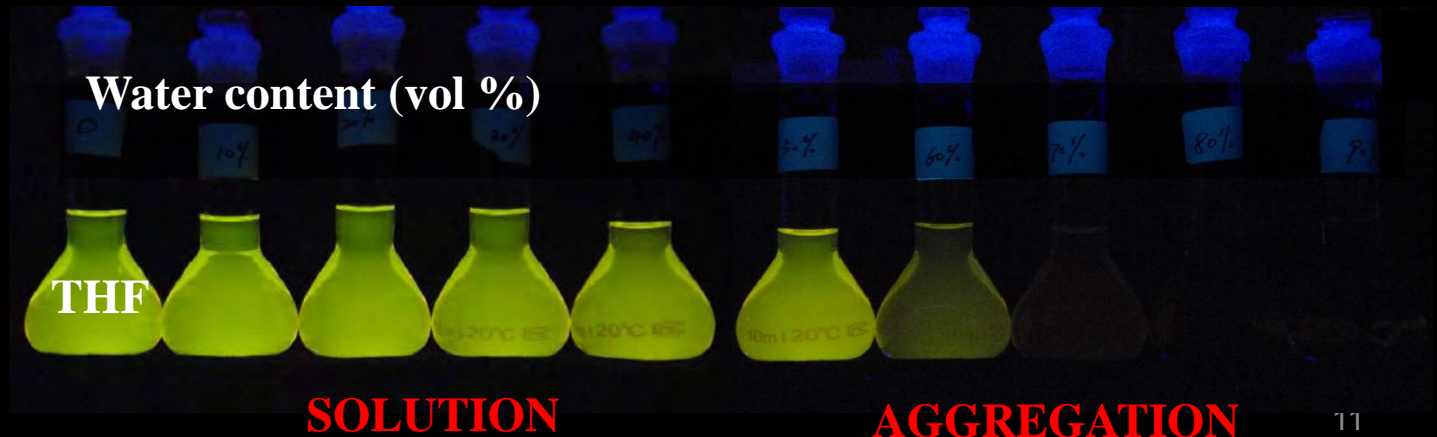




# An Issue in the Field

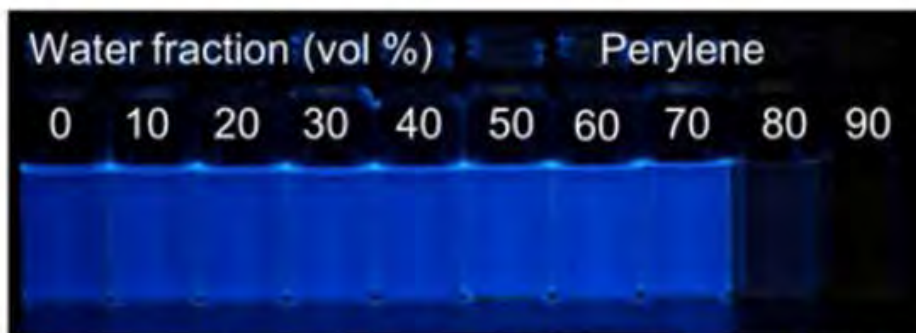
## Concentration Cause Quenching

- First discovered by Förster in 1954
- Now a general belief in the area
- A molecule quenches its own fluorescence at high concentration, partially due to **aggregate formation** (hence aggregation-caused quenching or ACQ).
- **Common organic dyes** show marked **concentration quenching** effects.

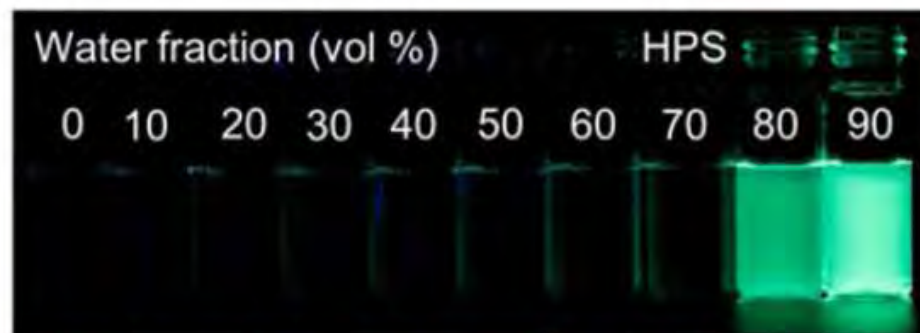




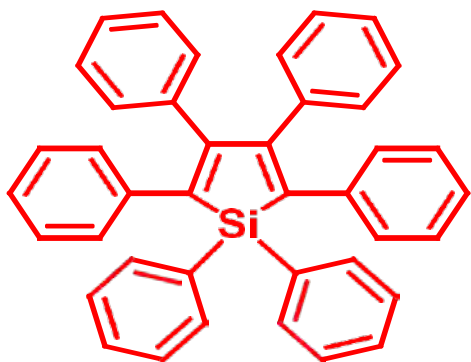
# An Anti-ACQ Photophysical Phenomenon



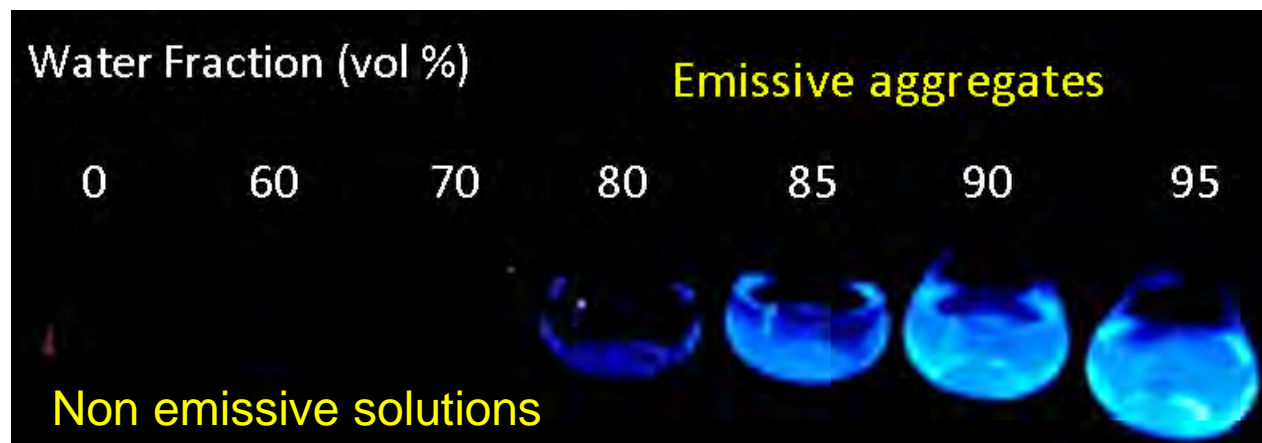
Aggregation-Caused Quenching (ACQ)



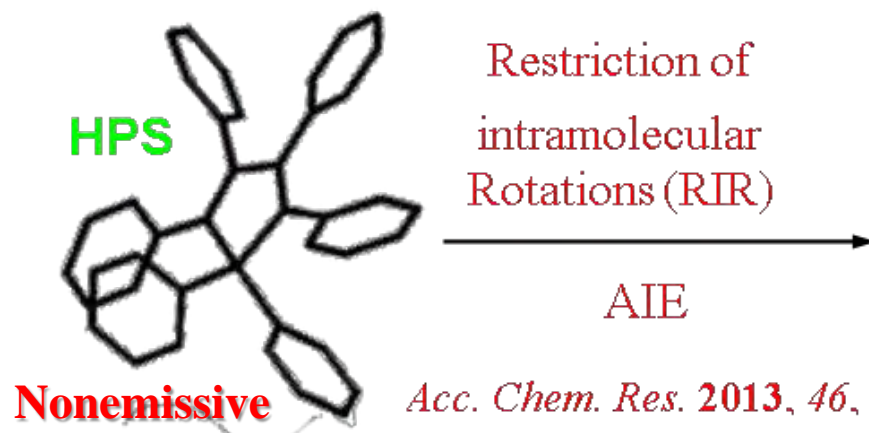
Aggregation-Induced Emission (AIE)



1,1,2,3,4,5-Hexaphenylsilole  
(HPS)

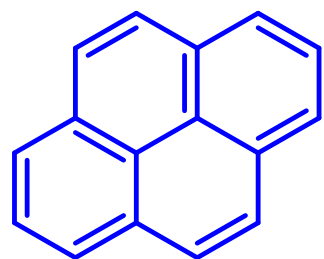


# Restriction of Intramolecular Rotations (RIR)



**Emissive (on)**

- Dynamic intramolecular rotations in the solution state
- Intramolecular motions restricted in the aggregate state
- Propeller-shape prevents  $\pi$ - $\pi$  interaction in the aggregates



**Emissive**

$\pi$ -stacking interactions

Aggregation caused quenching (ACQ)



**Nonemissive (off)**



**Emissive (on)**



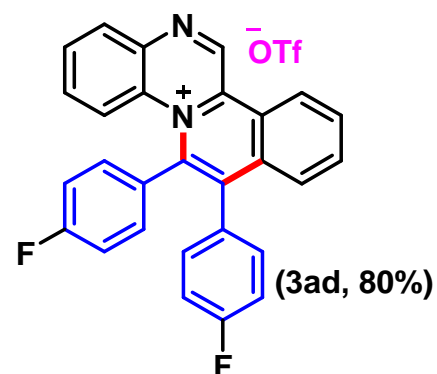
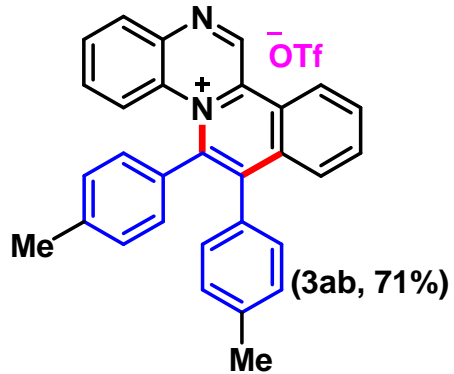
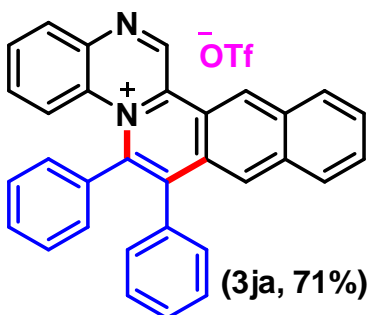
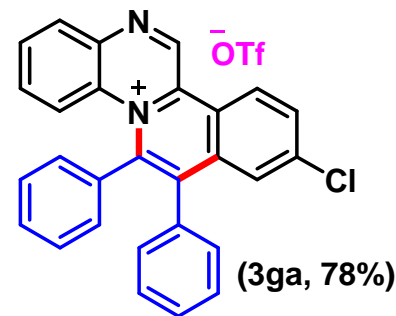
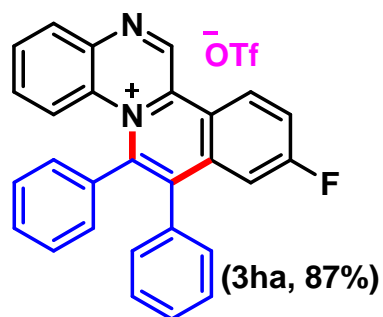
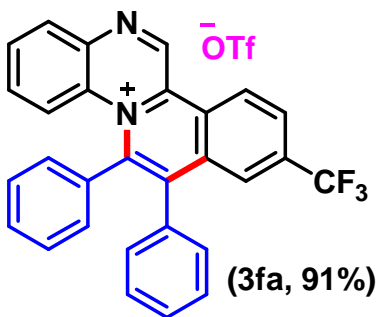
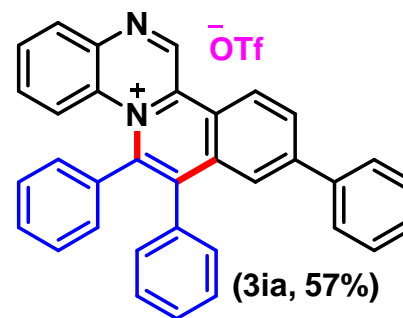
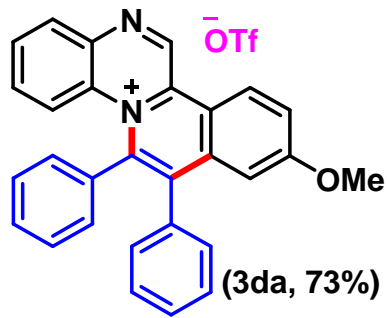
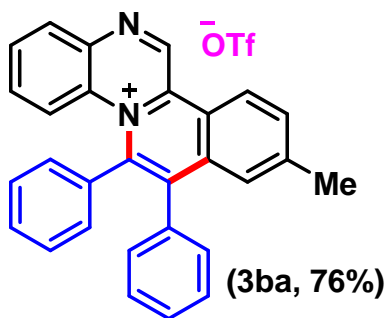
**Pol-AIE**

**Pol-AIEE**

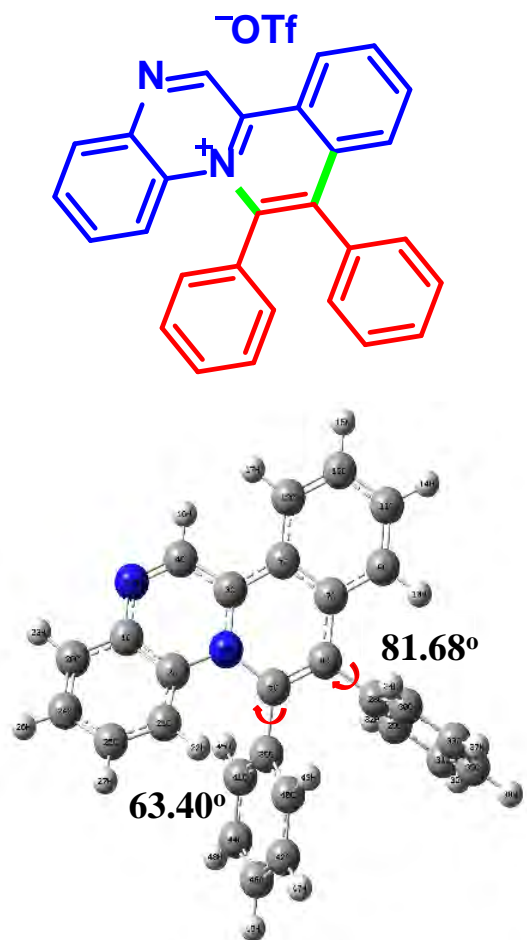
**Pol-ACQ**



# Representative Examples



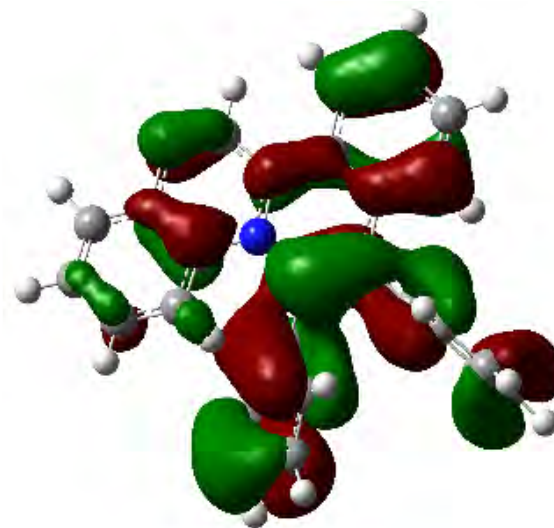
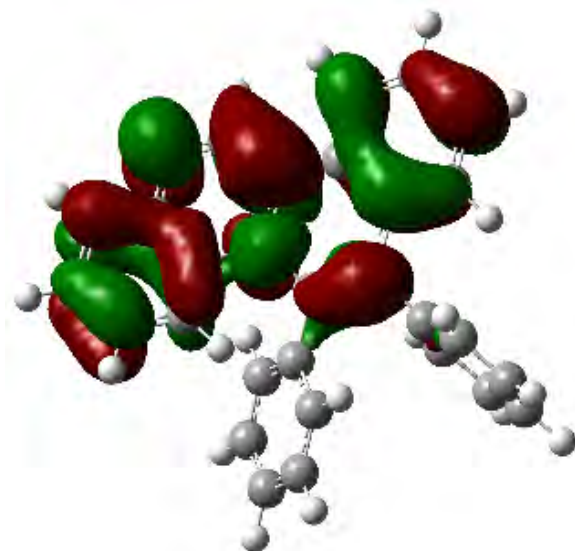
# Theoretical Investigations



$$E_{\text{LUMO}} = -3.48 \text{ eV}$$

$$\Delta E_{(\text{LUMO} - \text{HOMO})} = 3.30 \text{ eV}$$

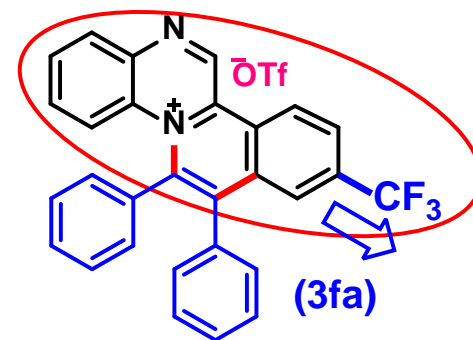
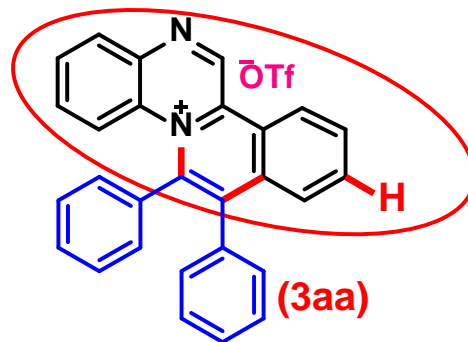
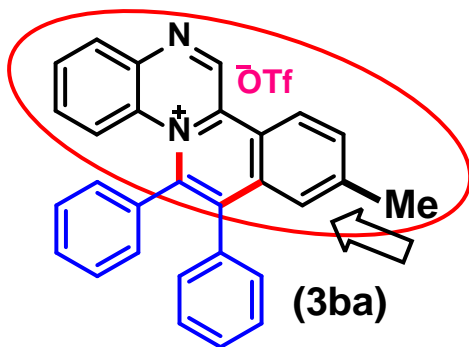
$$E_{\text{HOMO}} = -6.78 \text{ eV}$$



Geometry and electronic structure calculated using B3LYP/6-31G(d,p) level basic set using Gaussian 09

# Can the Colour be Tuned?

## Substituents on the LUMO



$$\Delta E_{\text{(LUMO-HOMO)}} = 3.36$$

$$\lambda_{\text{max,em}} = 529$$

>

$$3.30$$

>

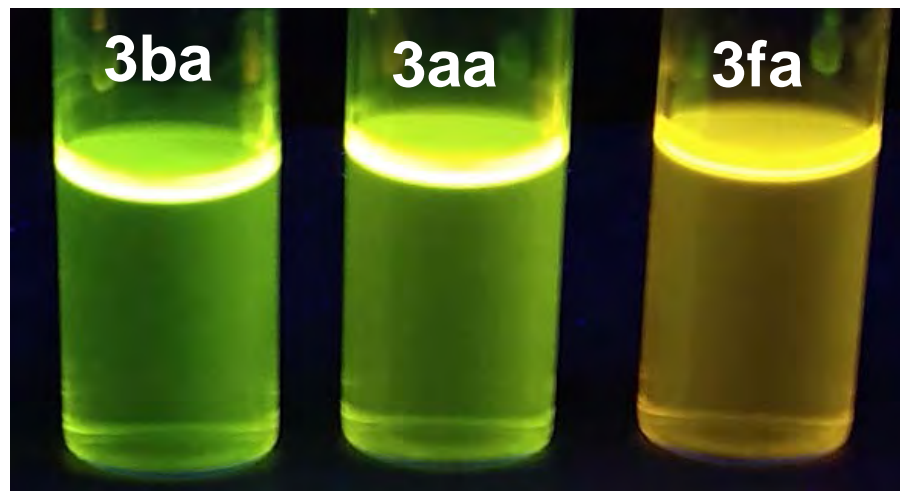
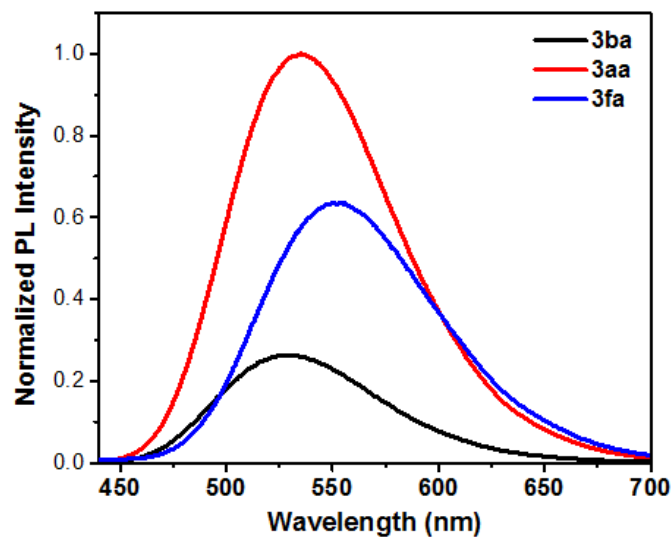
$$3.22 \text{ eV}$$

<

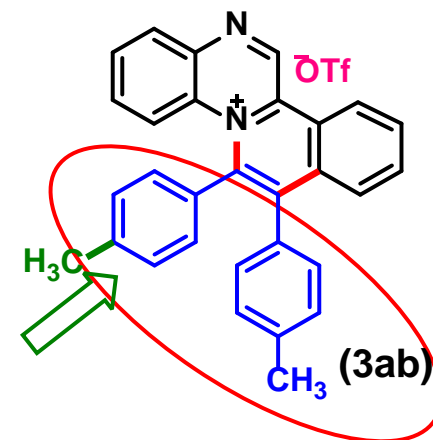
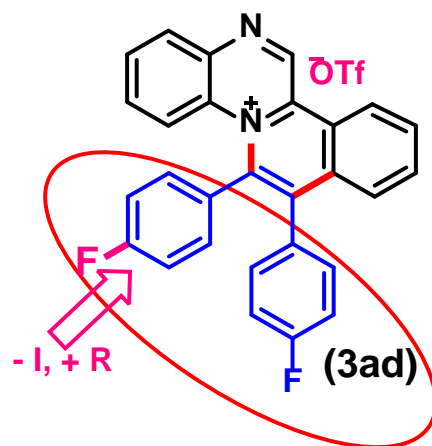
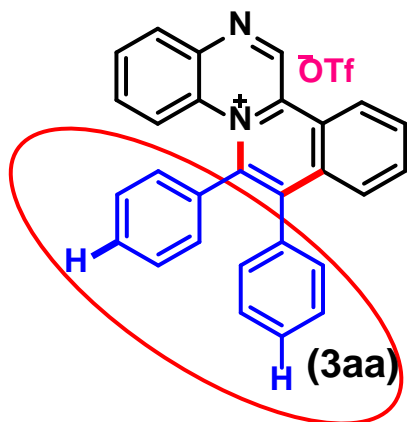
$$535$$

<

$$554 \text{ nm}$$

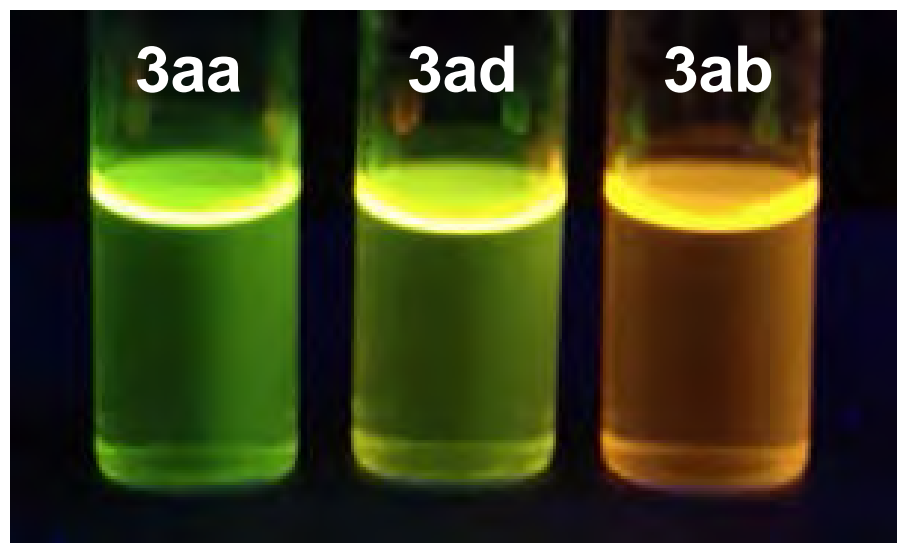
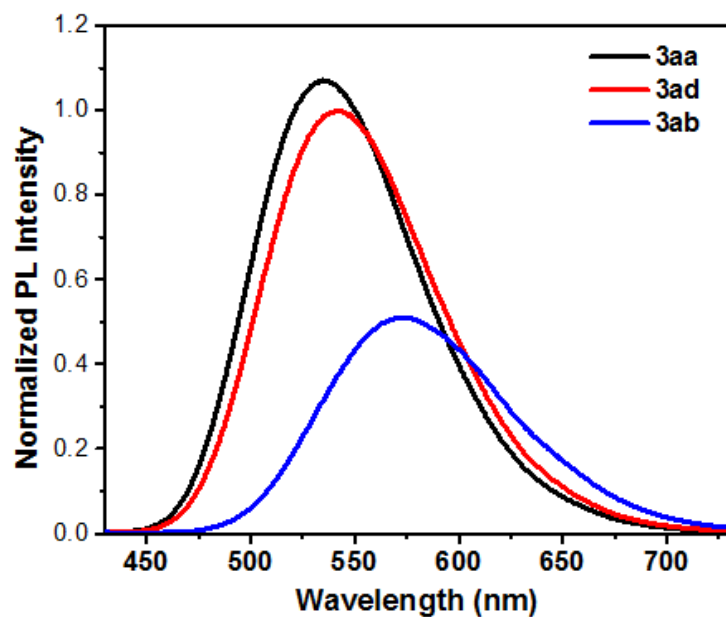


# Substituents on the HOMO

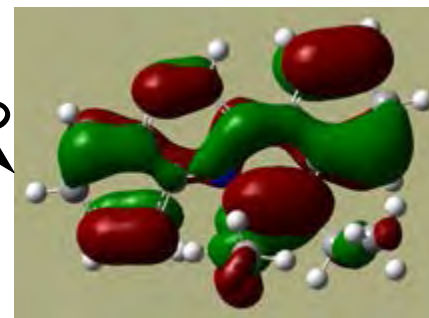
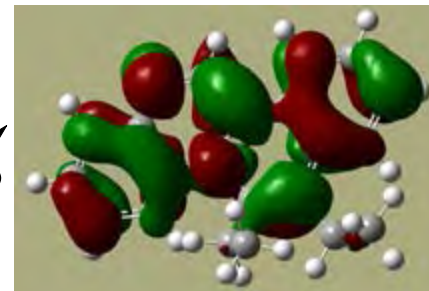
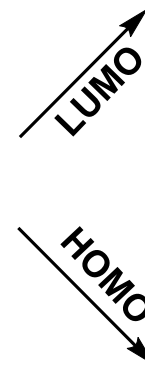
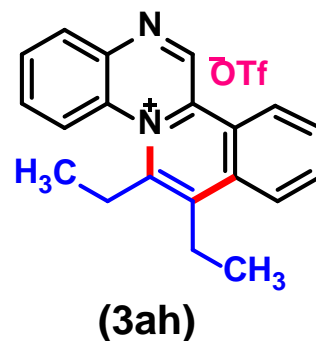
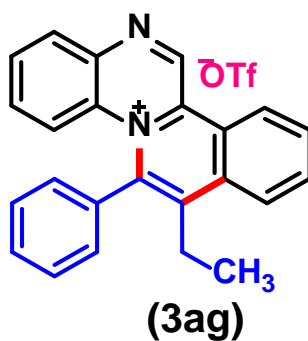
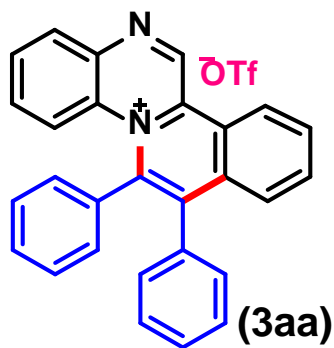


$$\Delta E_{\text{(LUMO-HOMO)}} = 3.30 > 3.21 > 3.09 \text{ eV}$$

$$\lambda_{\text{max,em}} = 535 < 542 < 571 \text{ nm}$$

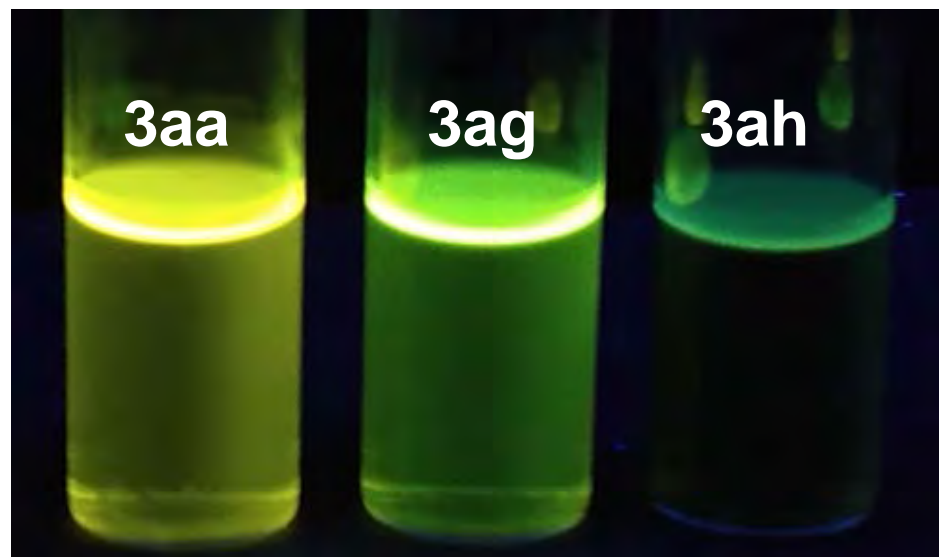
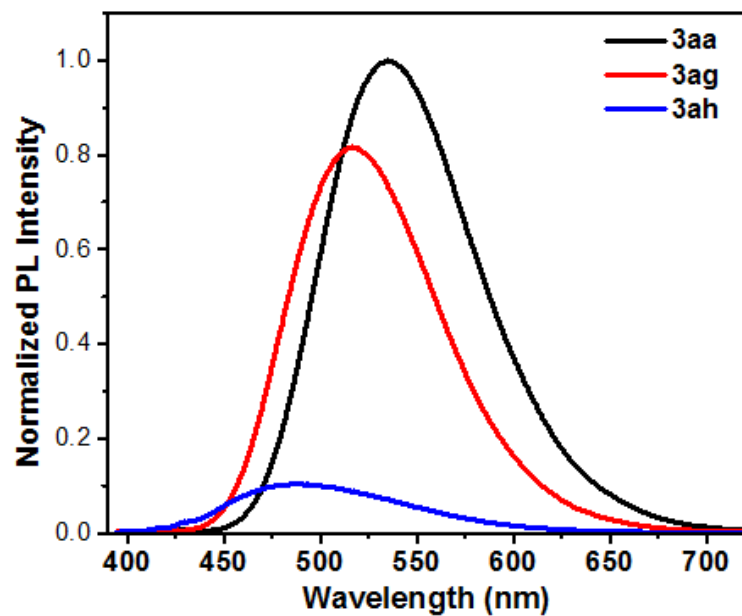


# Replacement of Phenyl with Alkyl Group

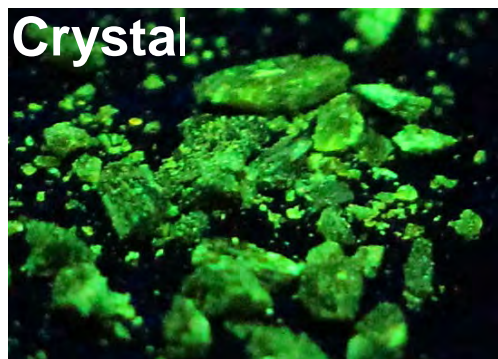
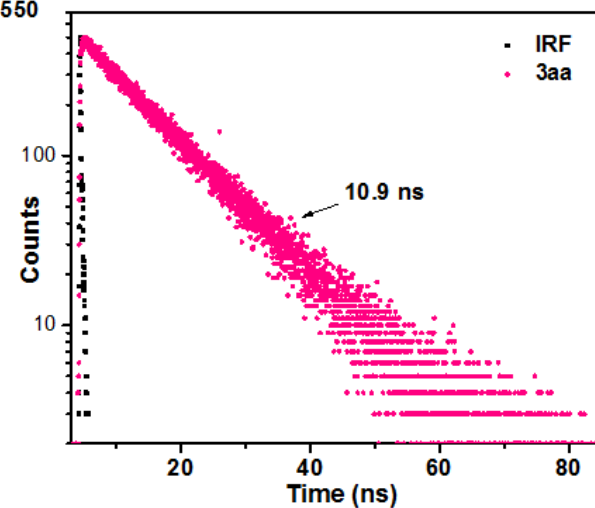
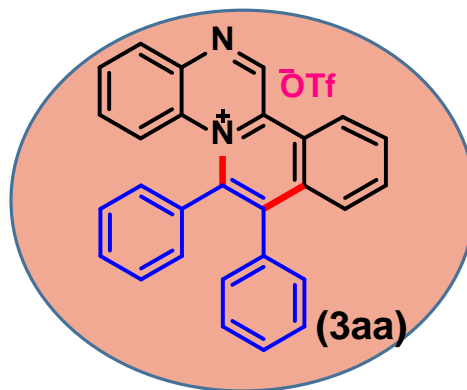
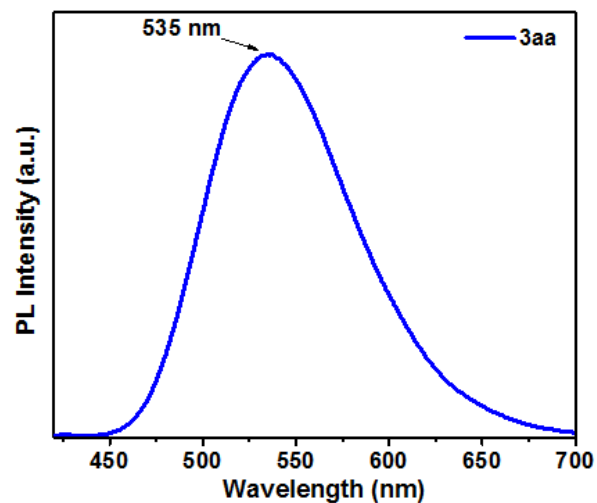
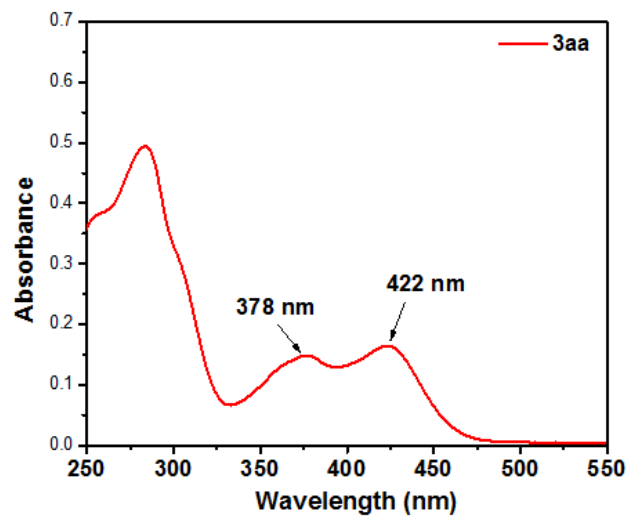


$$\Delta E_{(\text{LUMO-HOMO})} = 3.30 < 3.43 < 3.48 \text{ eV}$$

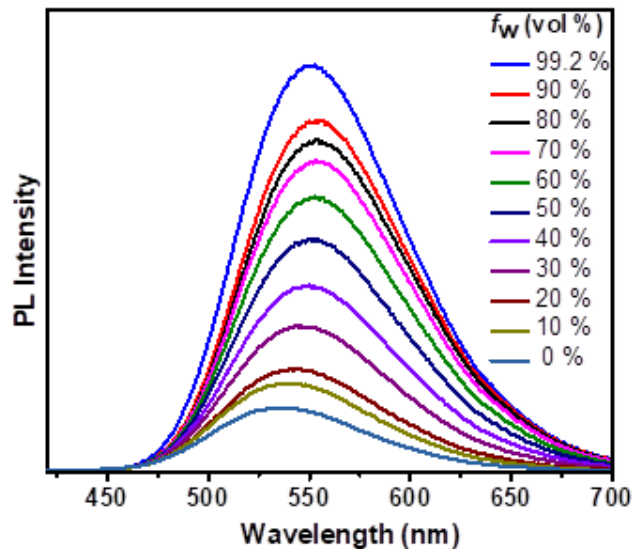
$$\lambda_{\text{max,em}} = 535 > 515 > 488 \text{ nm}$$



# Photo Physical Properties

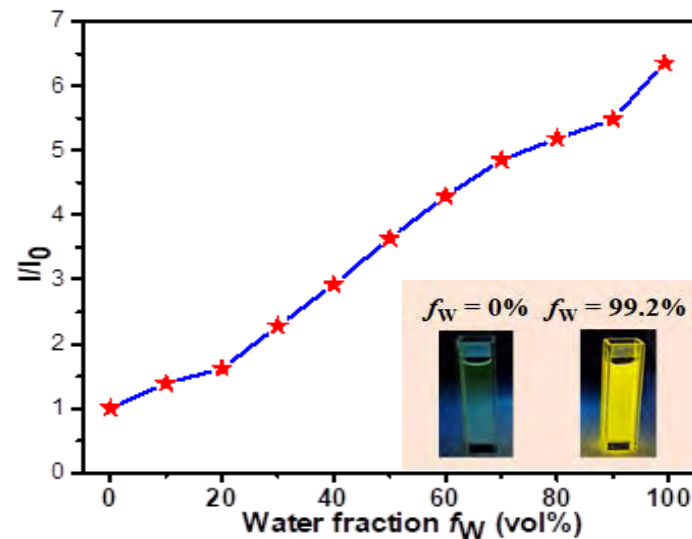
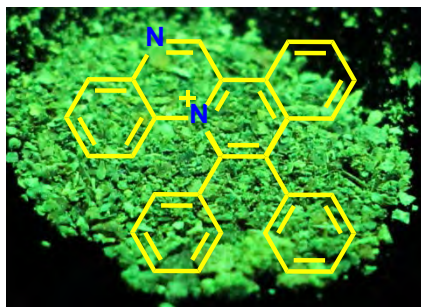




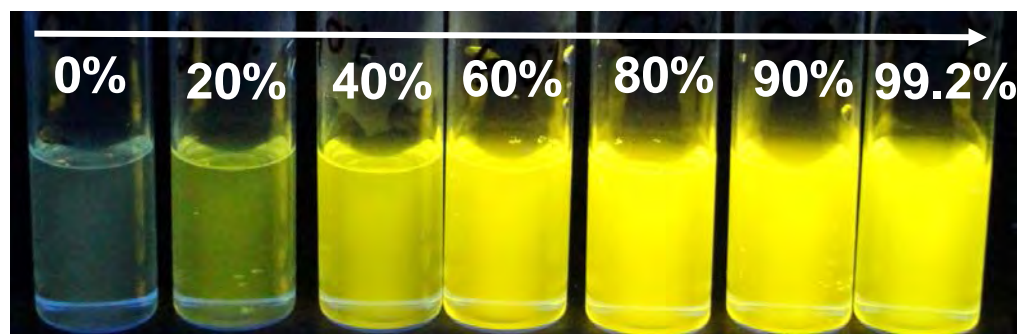


PL titration with varying  $f_W$  in DMF

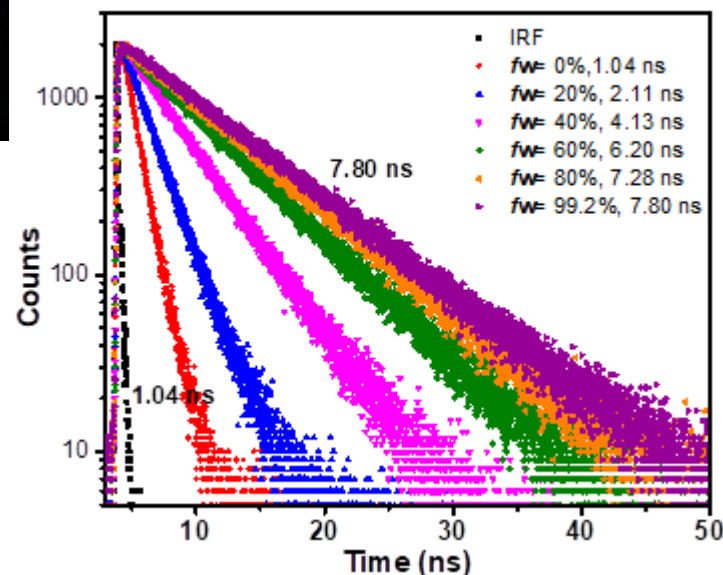
## AIEE Investigation



Relative intensity versus  $f_W$  in DMF

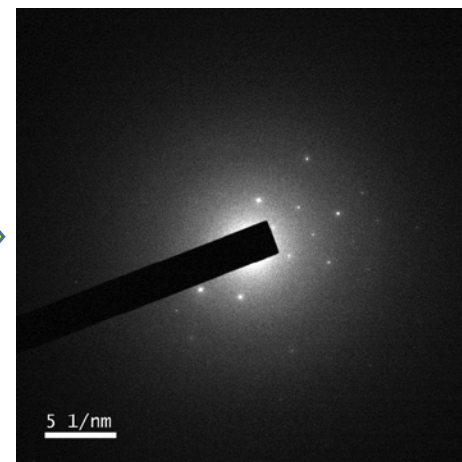
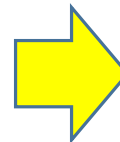
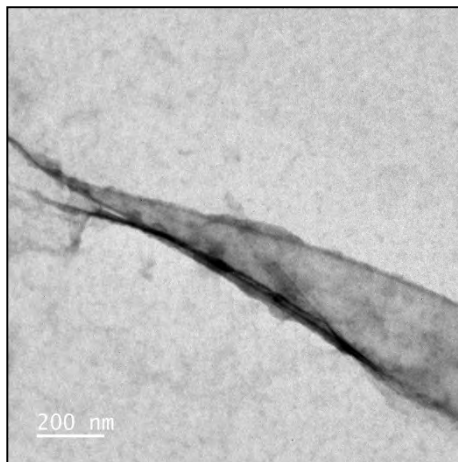
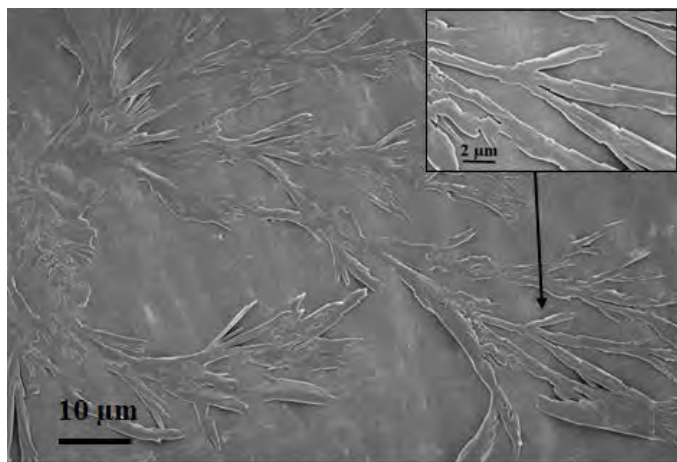


Fluorescence photographs in DMF/water at 365 nm



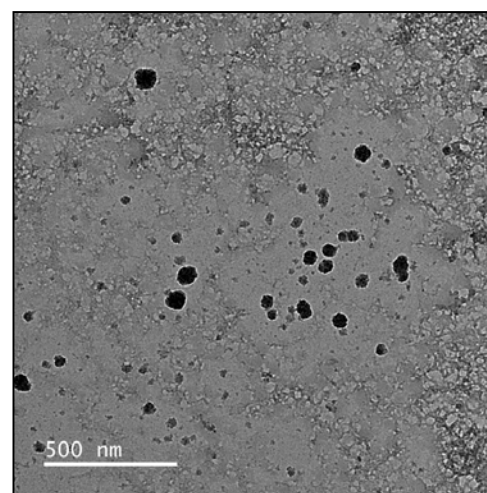
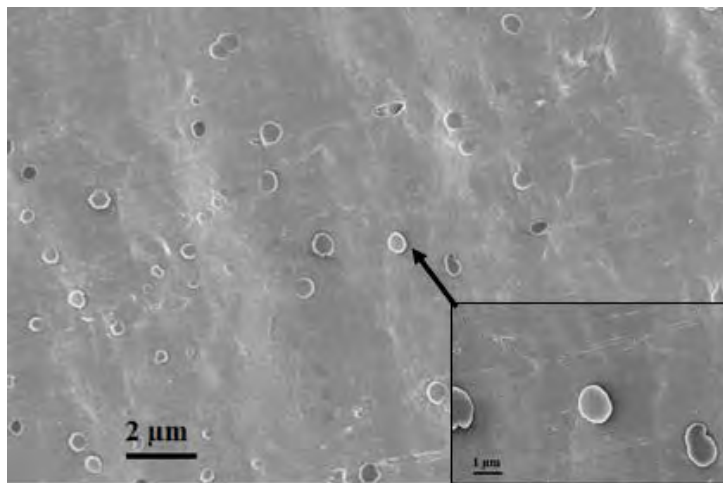
TRPL spectra with varying  $f_W$  in DMF

# Morphological Investigation



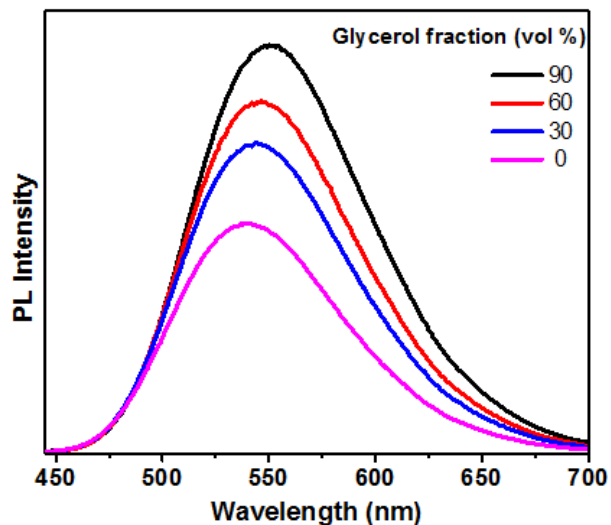
**FESEM and TEM images of crystalline aggregates at  $f_w$  99.2% (80  $\mu$ M)**

**SAED pattern in TEM**

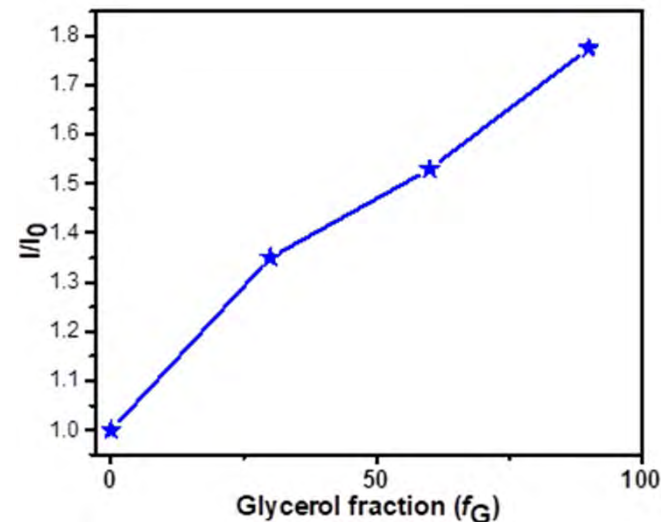


**FESEM and TEM images of distorted spherical aggregates at  $f_w$  99.6% (40  $\mu$ M)**

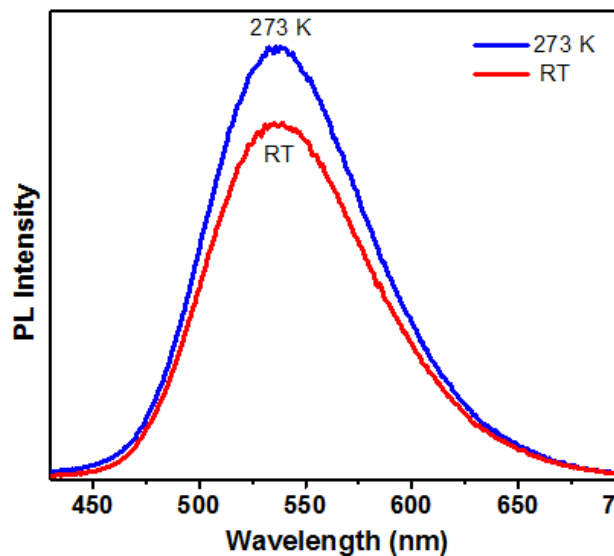
# Visco- and Thermochromism



PL titration with varying  $f_G$  in MeOH



Relative intensity versus  $f_W$  in DMF



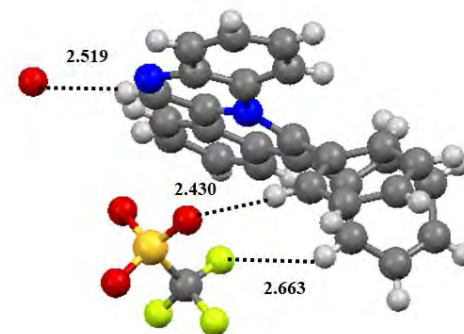
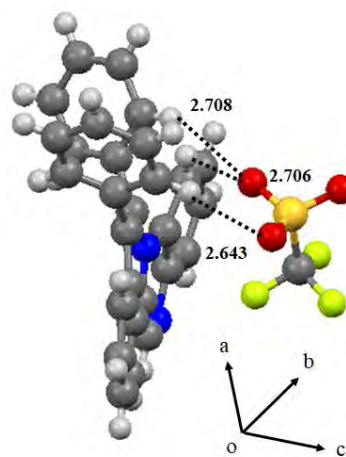
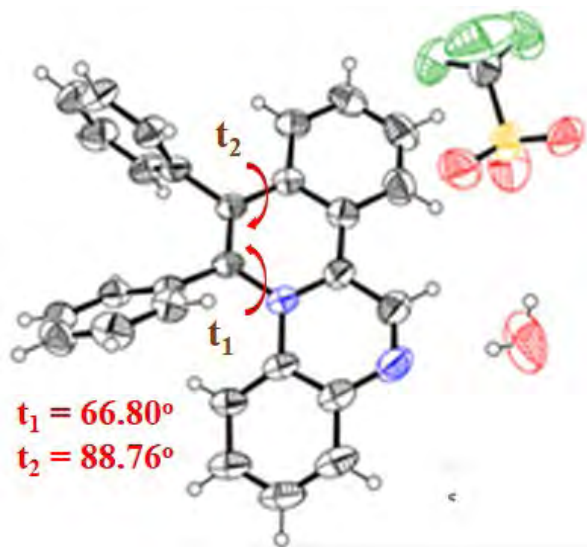
PL titration with varying temperature in DMF solvent

- **Increased in viscosity and lowering of temperature** cause restriction in intramolecular motion that offers enhanced emission intensity in aggregated state



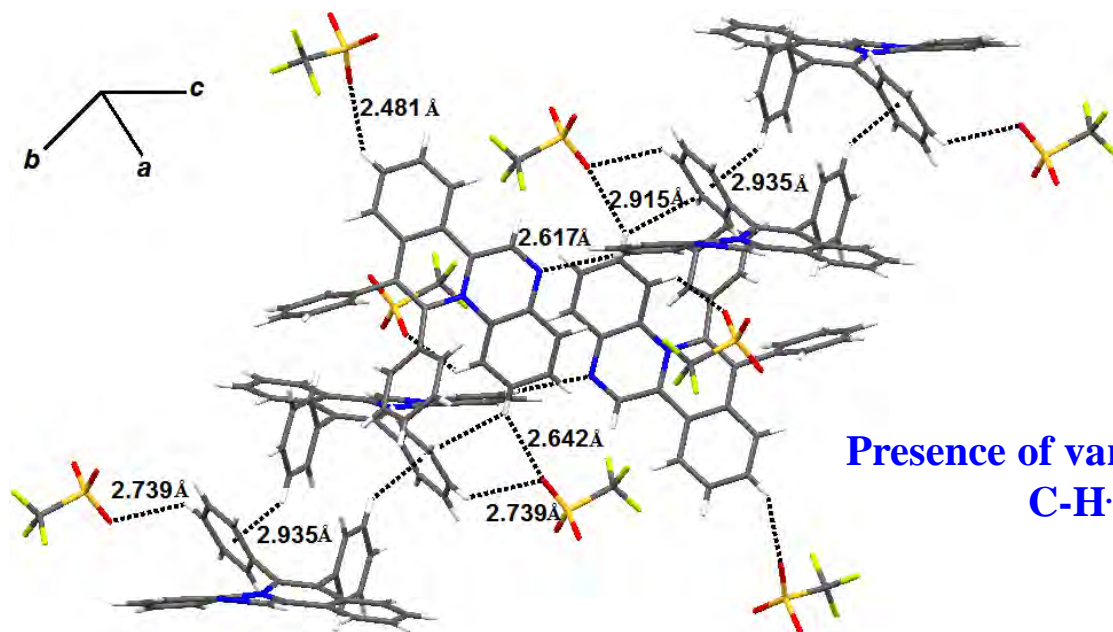


# Crystal Packing

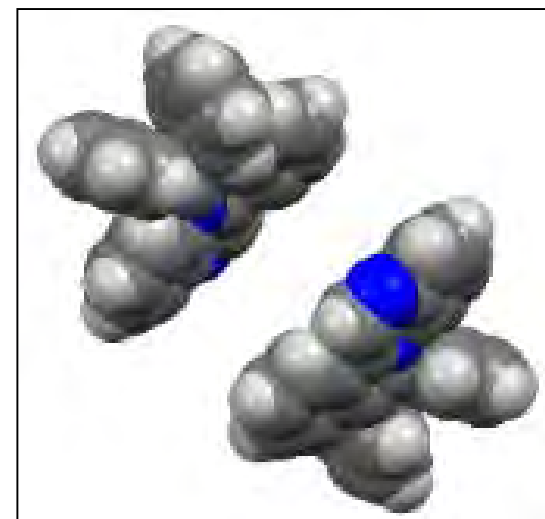
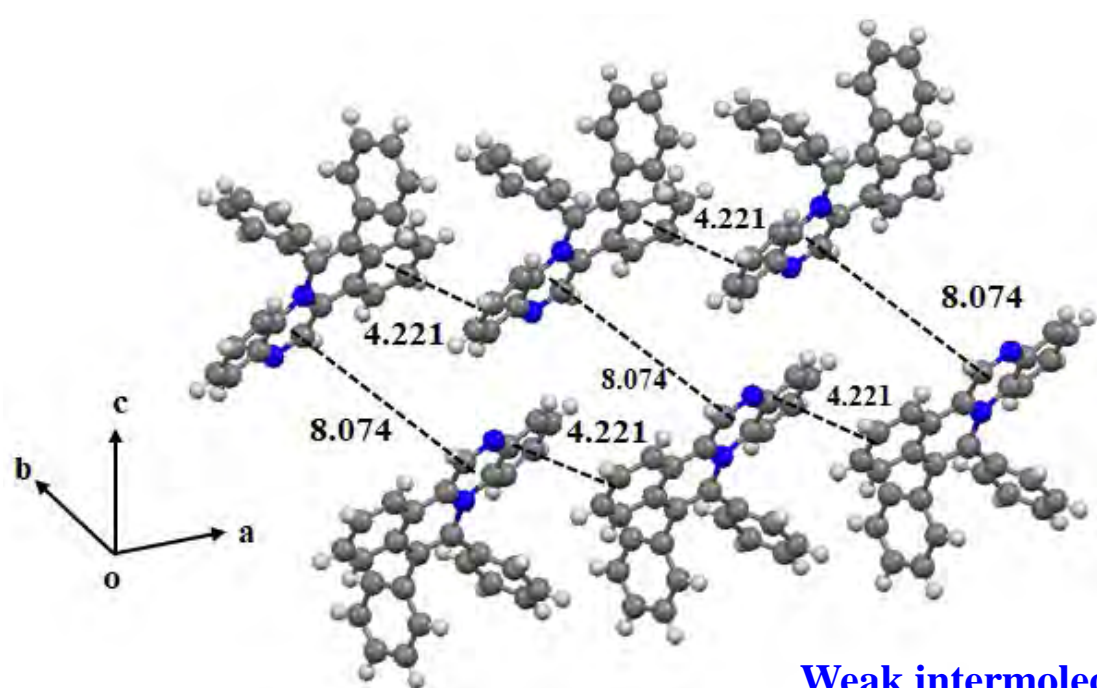


ORTEP view of 3aa with torsional angles

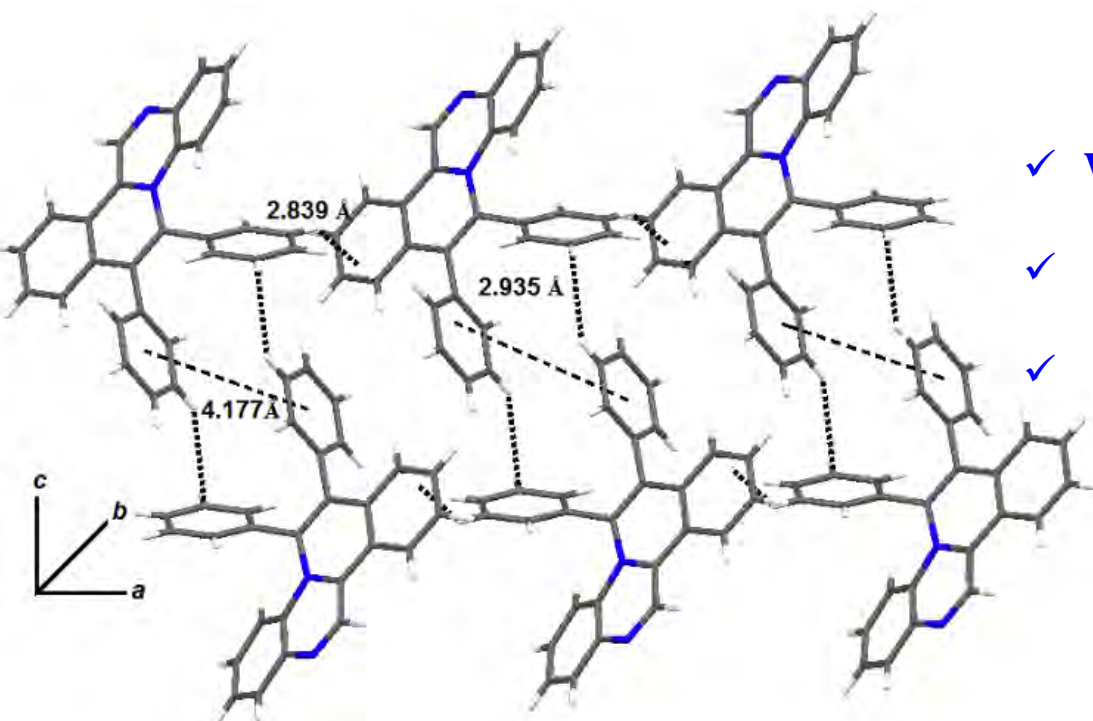
Existence of strong H-bonding interactions



Presence of various non-covalent interactions including C-H $\cdots$ N, C-H $\cdots$  $\pi$ , C-H $\cdots$ O and C-H $\cdots$ F



**Weak intermolecular  $\pi$ - $\pi$  interactions (8.074 – 4.177 Å°)**



✓ Weak  $\pi$ - $\pi$  interactions

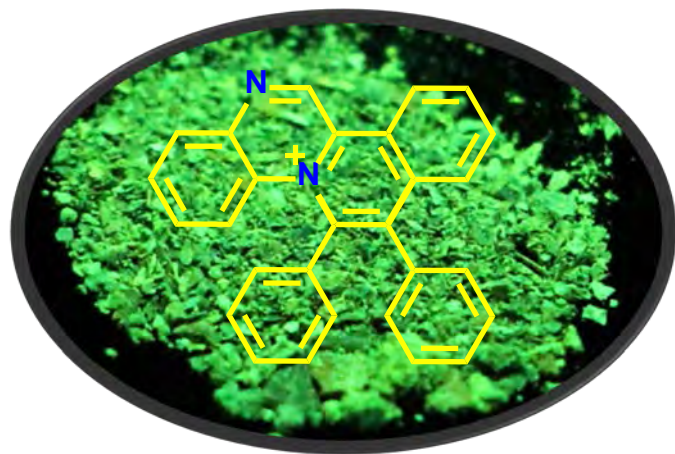
✓ highly twisted conformation

✓ presence of various non-covalent interactions

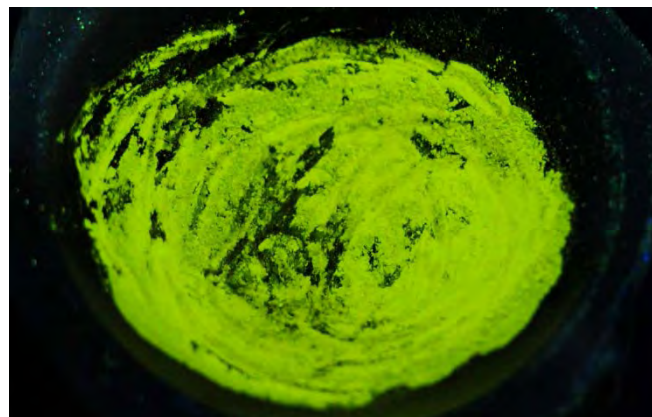


# Mechanochromism

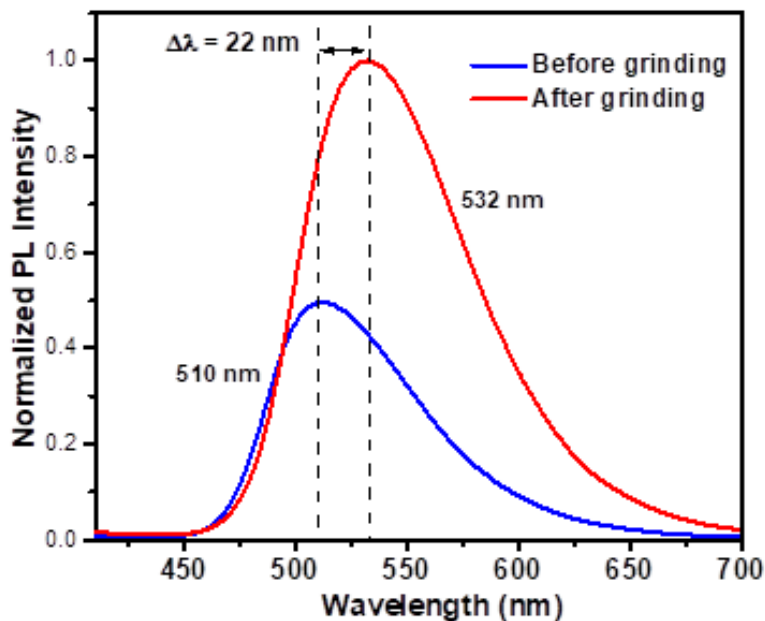
Pristine



Ground



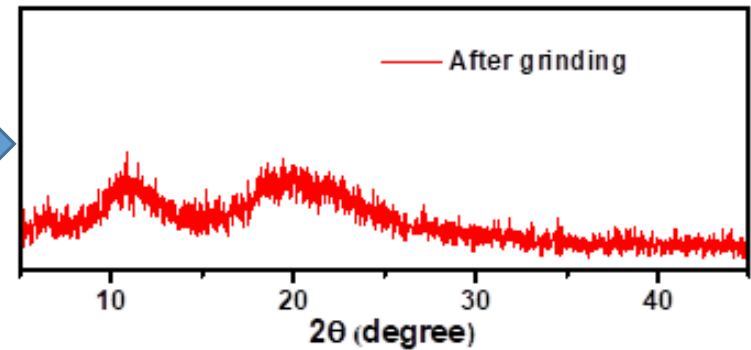
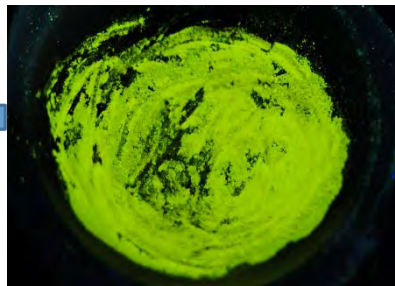
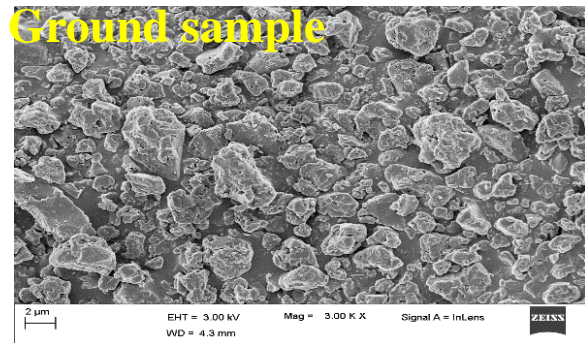
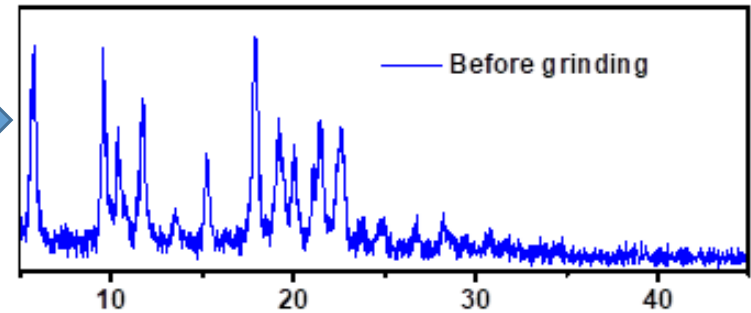
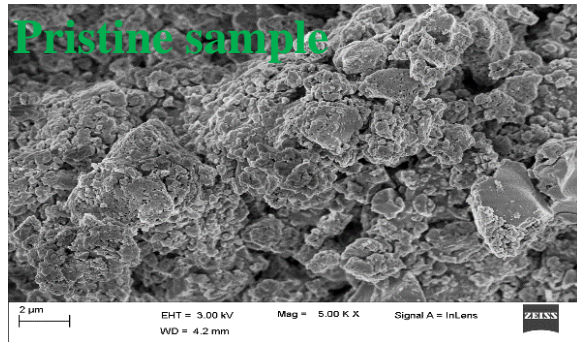
Colour transition from green to yellowish green upon grinding in mortar-pestle (in UV 365 nm)



Is due to weak crystal packing and twisted D-A conformation

During grinding twists the two Ph rings which adopts a coplanar conformation with quinoxaline core leading to the extended conjugation there by shifting the emission toward red

# Morphological Transformation

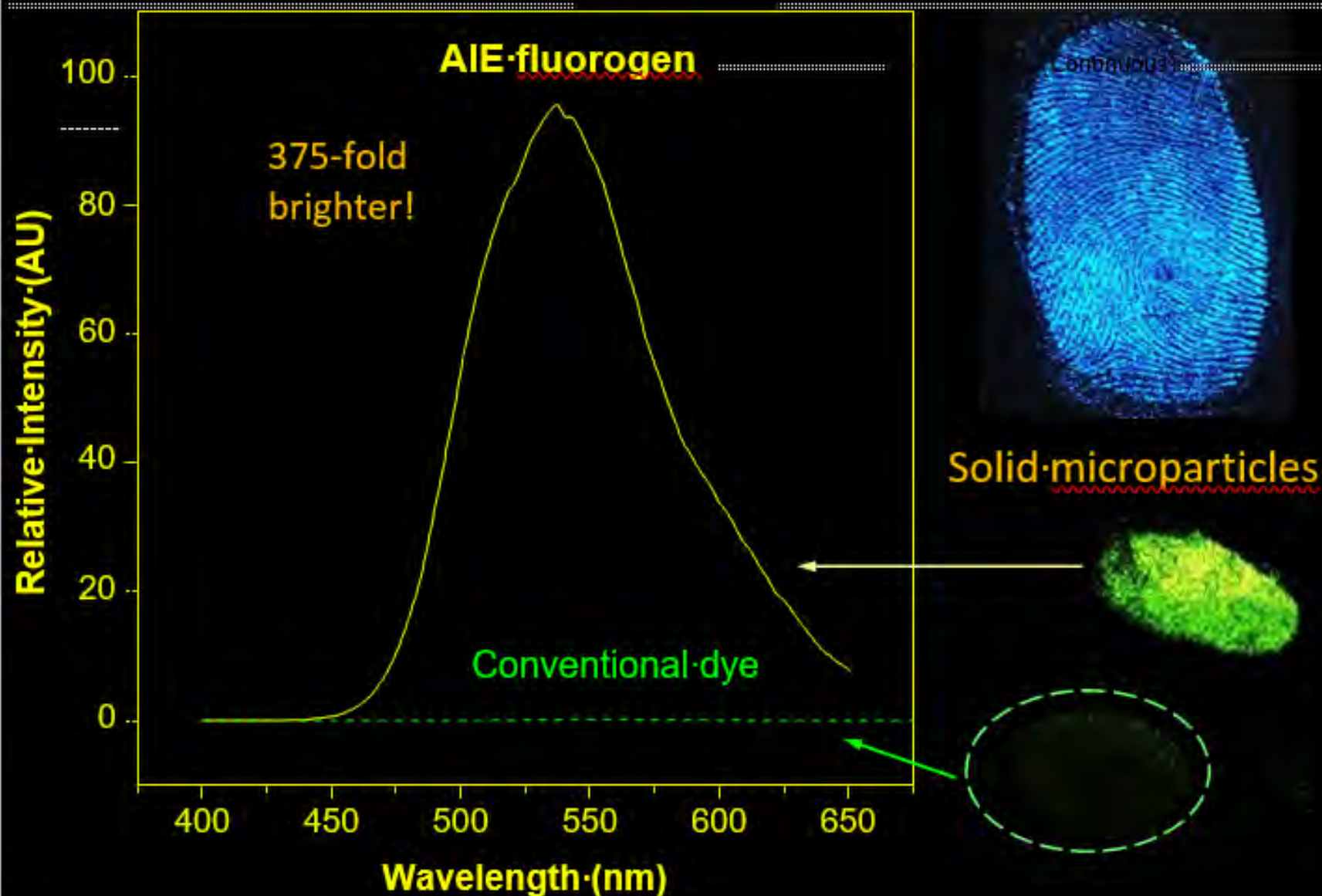


FESEM images of Pristine and Ground sample

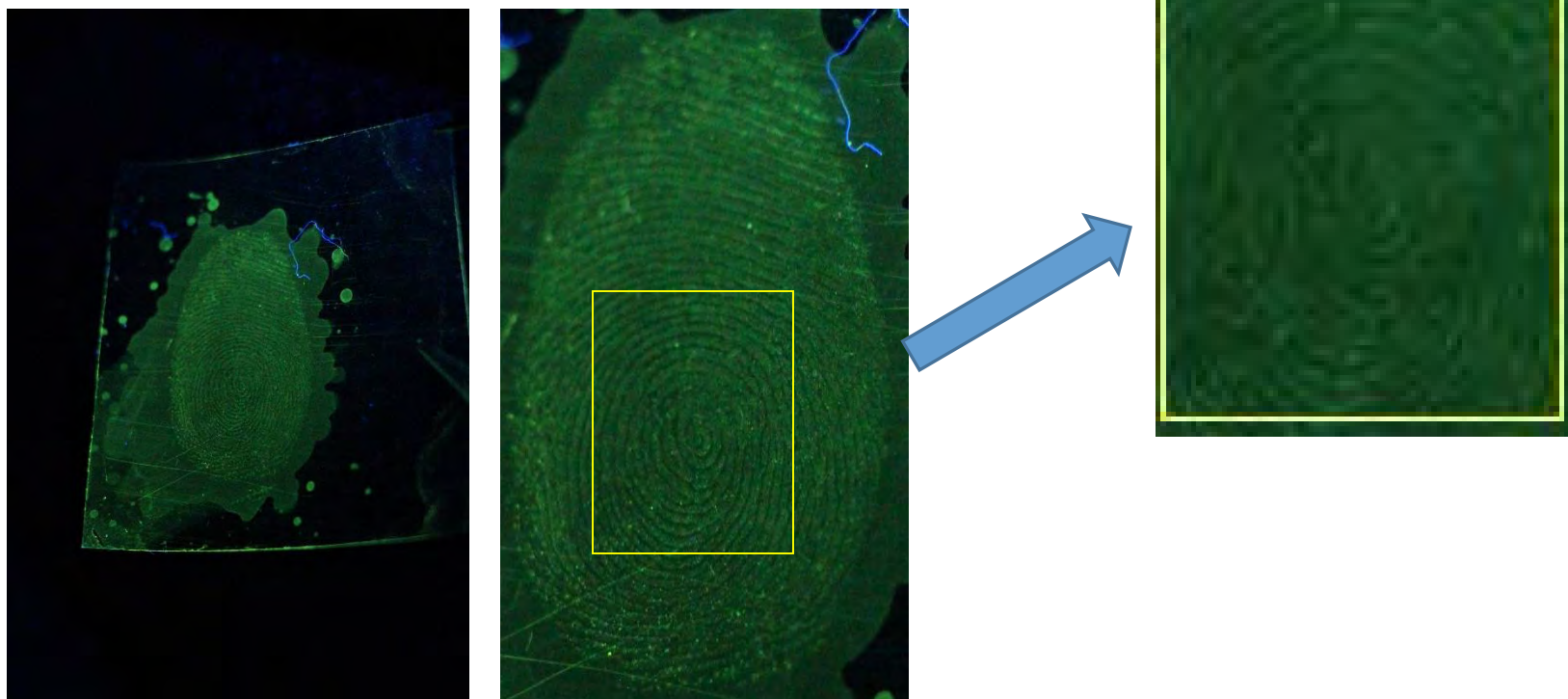
PXRD patterns of Pristine and Ground sample

**PXRD pattern confirms the transformation from crystalline to amorphous state during grinding**

# Forensic-Science-(Criminal-Investigation)



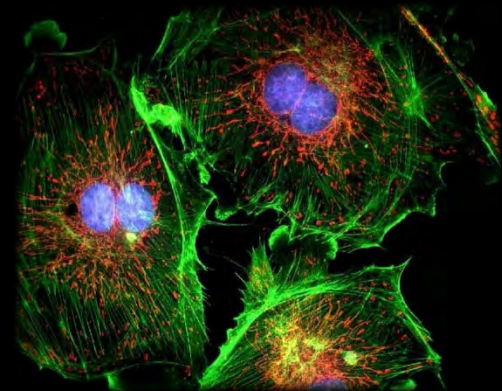
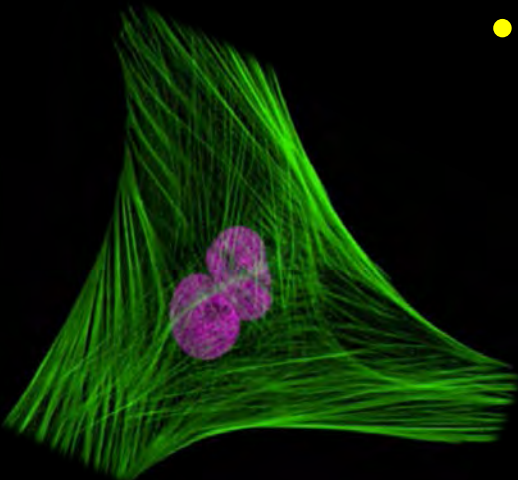
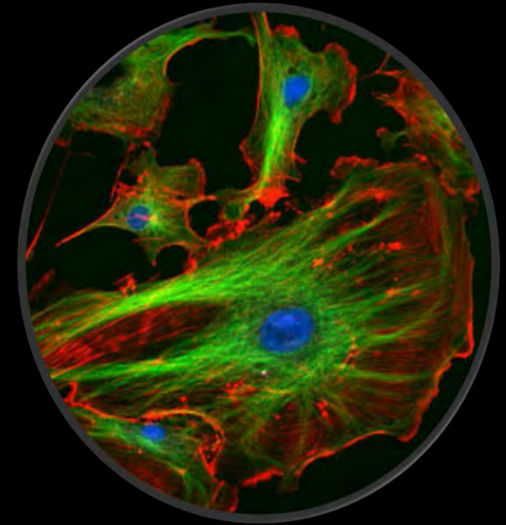
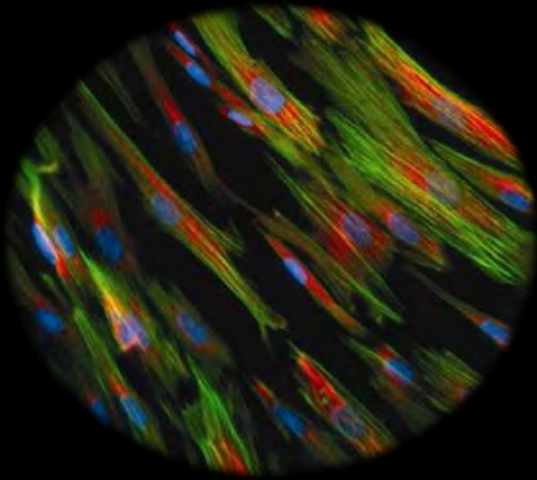




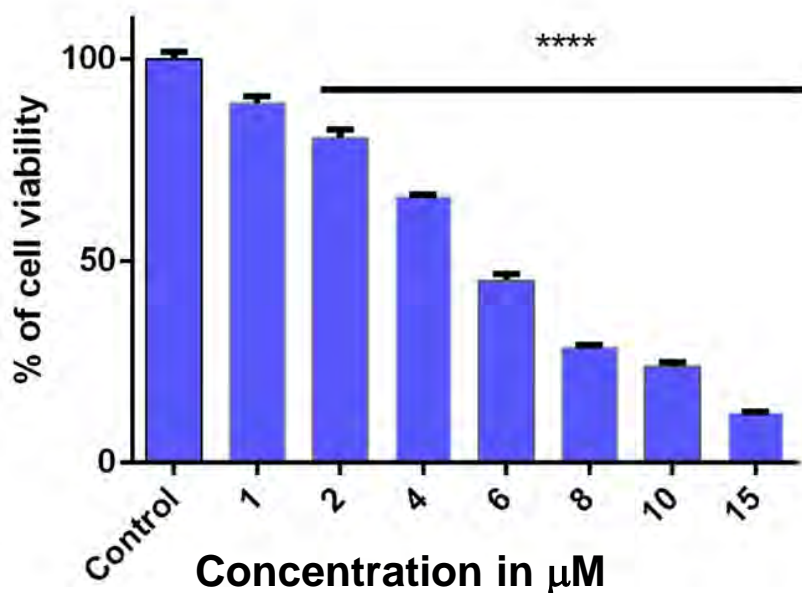
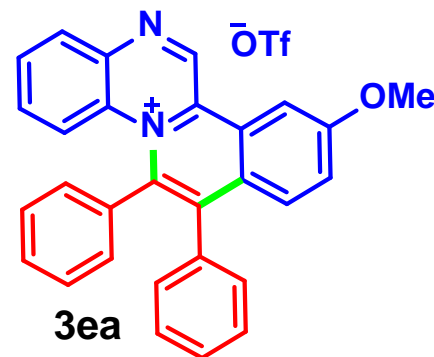
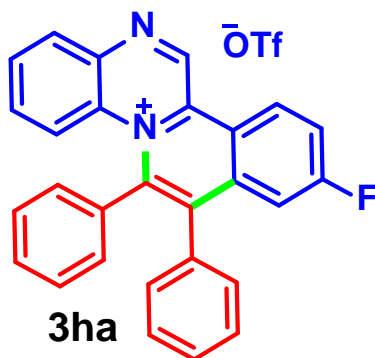
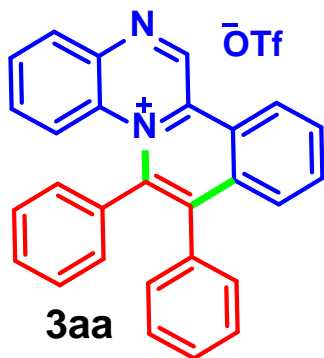
**Fluorescence images of Fingerprints on adhesive tape stained with 3aa in 70:30 H<sub>2</sub>O-MeOH solvent. Concentration = 1 mM**

# Fluorescent Bioprobes

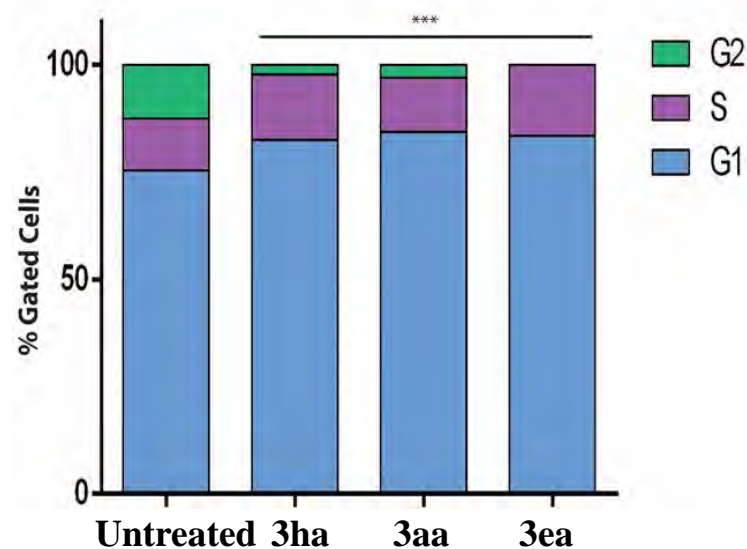
- **Fast response**
- **Sensitive**
- **Low cytotoxicity**
- **In-situ visualization**
- **Simple operation**
- **Excellent penetration**



## Are they bio-compatible?

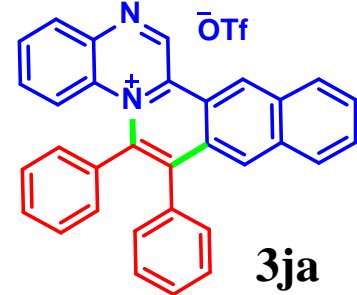
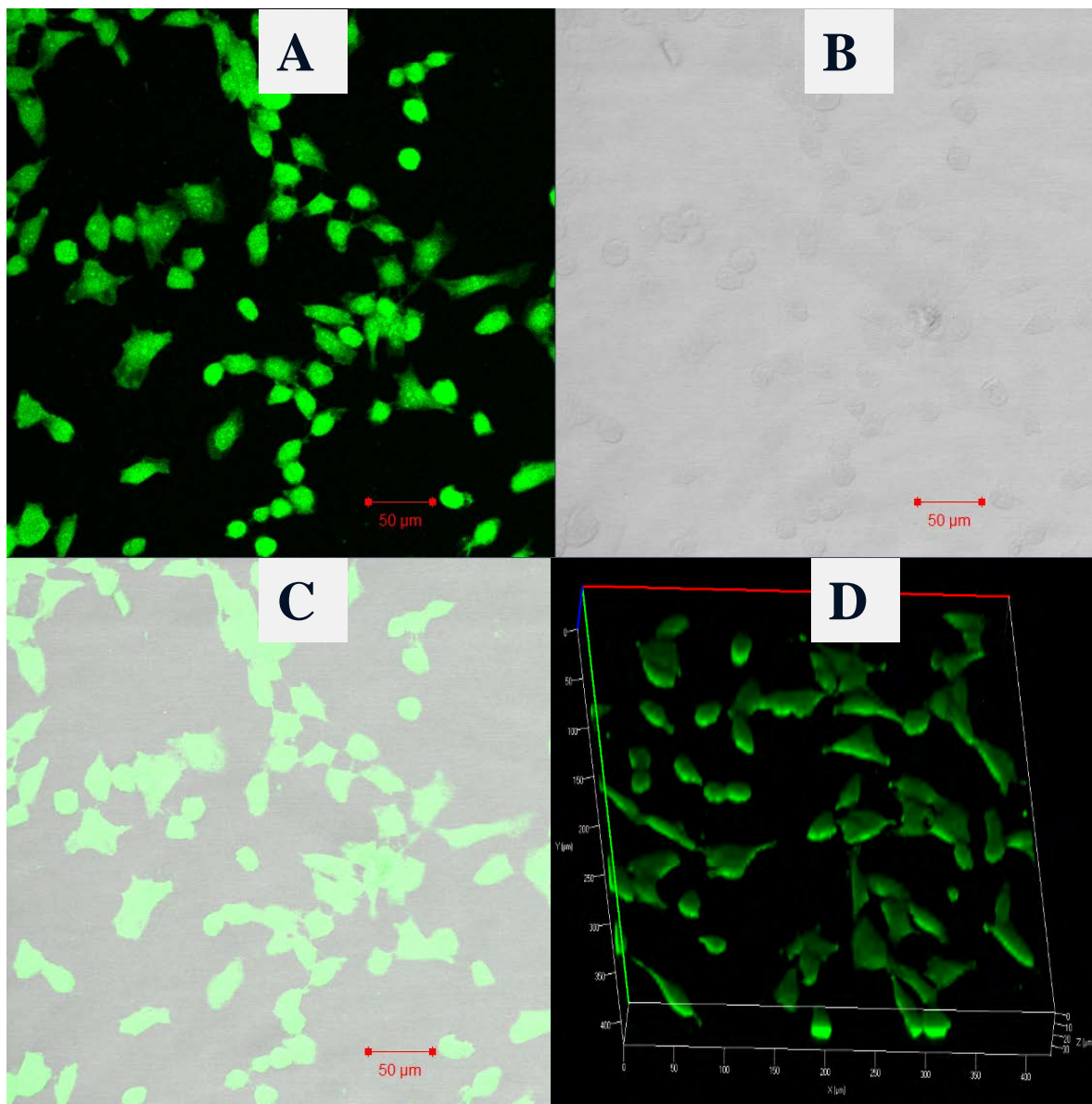


Effect of Compound 3aa in terms of reduction in percentage of viable cells as demonstrated by alamerBlue assay.

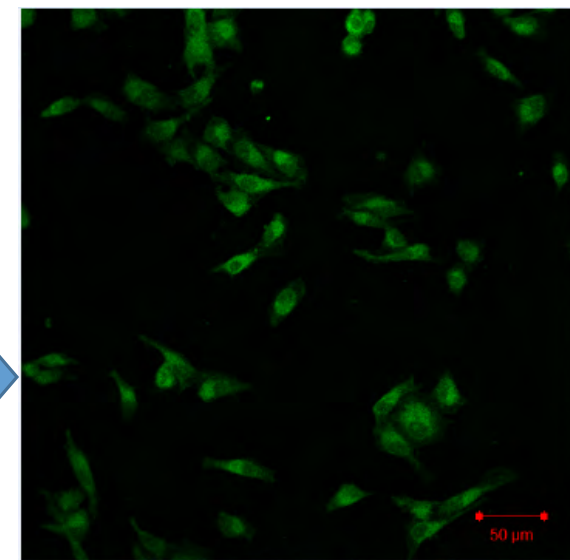
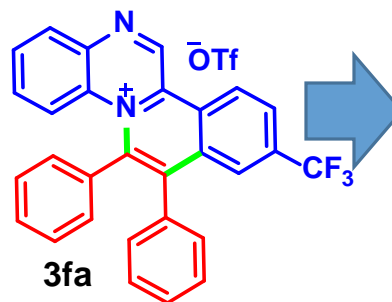
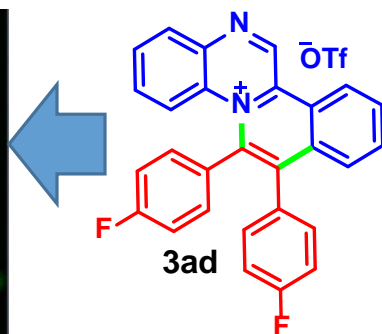
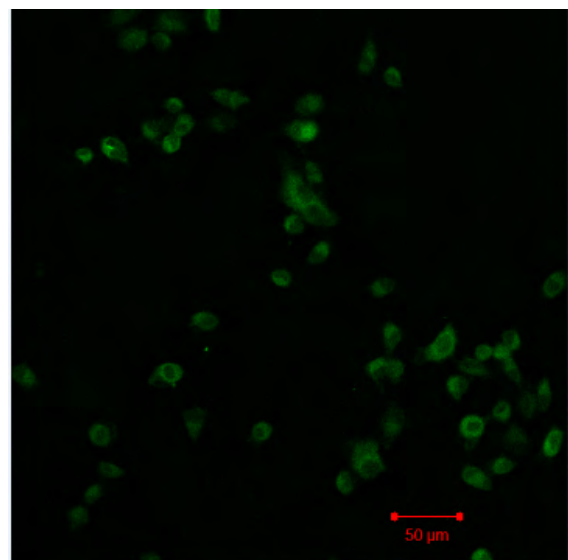
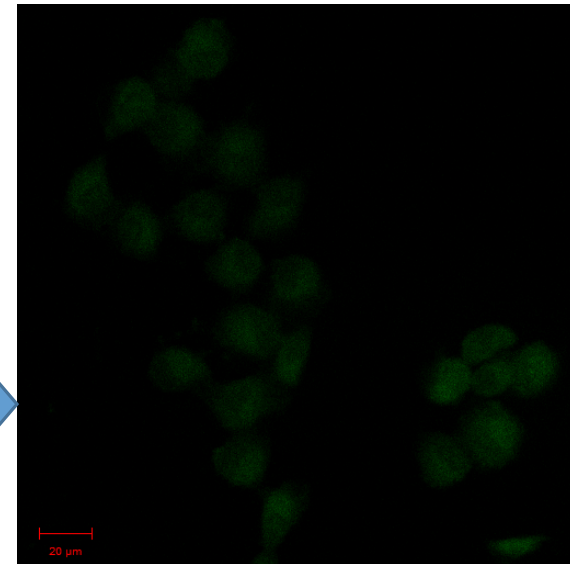
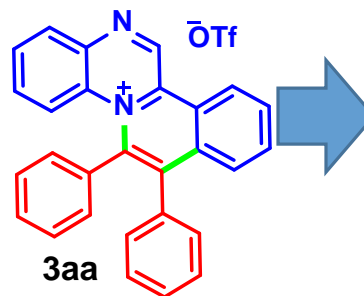
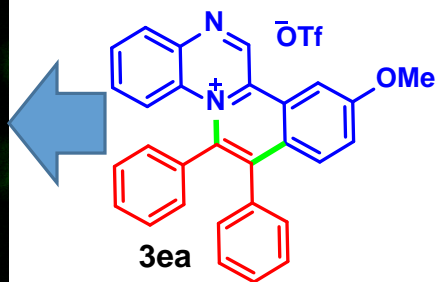
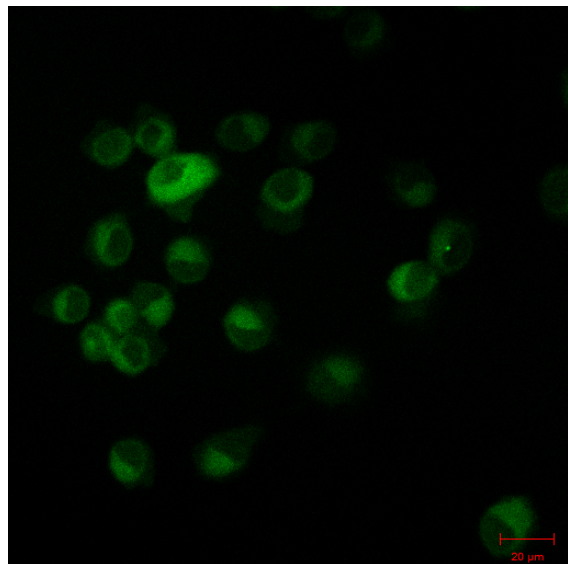


Flow cytometry based analysis of cell cycle of HeLa cells treated with compound 3ha, 3aa and 3ea which shows reduction of G2 phase



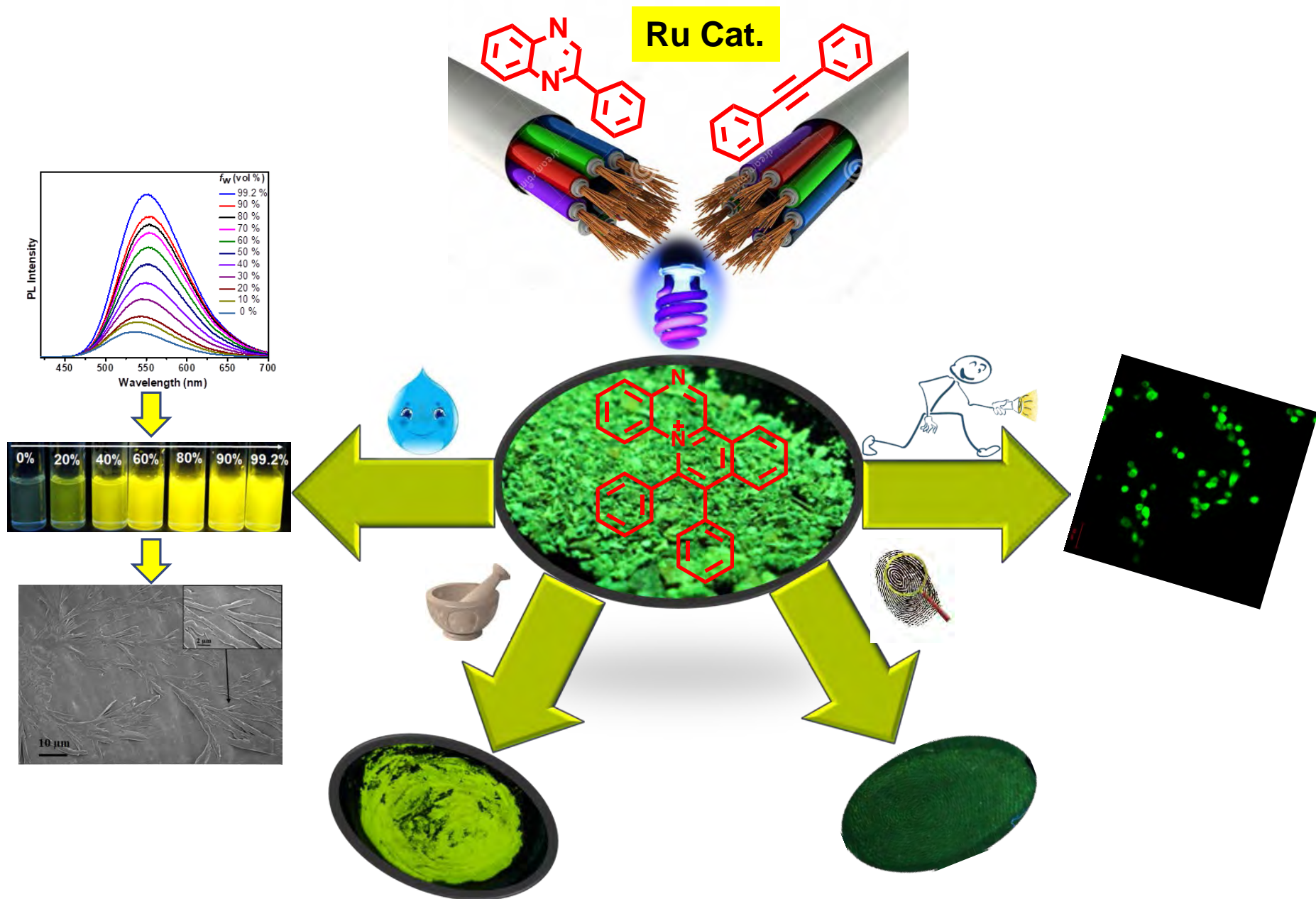


**CLSM images of HeLa cell stained with 3ja, (A) Fluorescence; (B) Bright-field; (C) Merged and (D) 3D images. Cells were incubated with 200  $\mu$ M of 3ja for 10 mins. Excitation: 458 nm Emission range: 480–605 nm.**

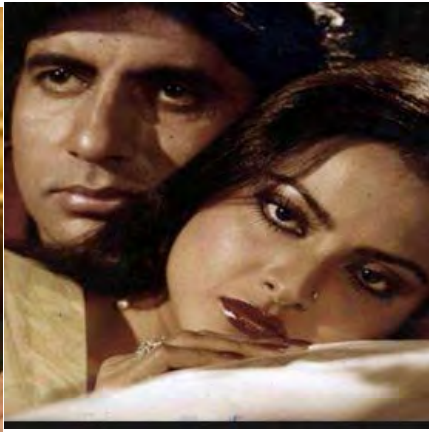


CLSM images of HeLa cell stained with 200  $\mu$ M of **3ea**, **3aa**, **3ad**, **3fa** for 10 mins.

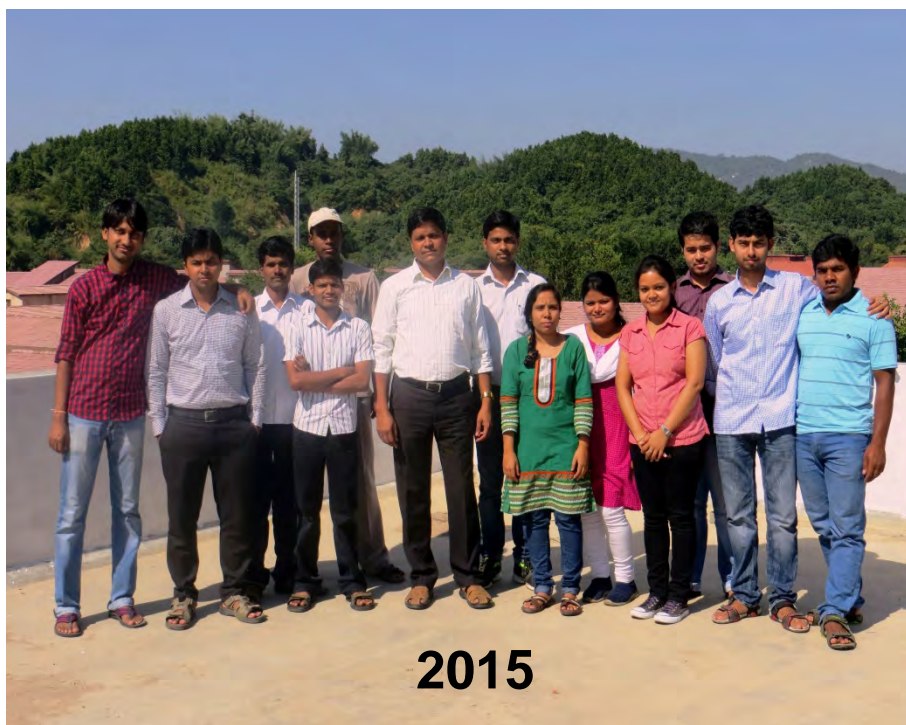
# Multifunctional AIEEgen













# Acknowledgements...

**Financial support from DST/CSIR**

**DST FIST for XRD facility**

**Centre for Instrumentation IIT Guwahati**

**Department of Chemistry IIT Guwahati**



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