



# Week 8, Diffusion and Self Assembly

1/ Assembly instructions: Distributed pattern storage and self assembly

2/ Biology is 'small' - thermal energy matters! Behold - diffusion!

3/ Diffusion and self-assembly are paired at the hip

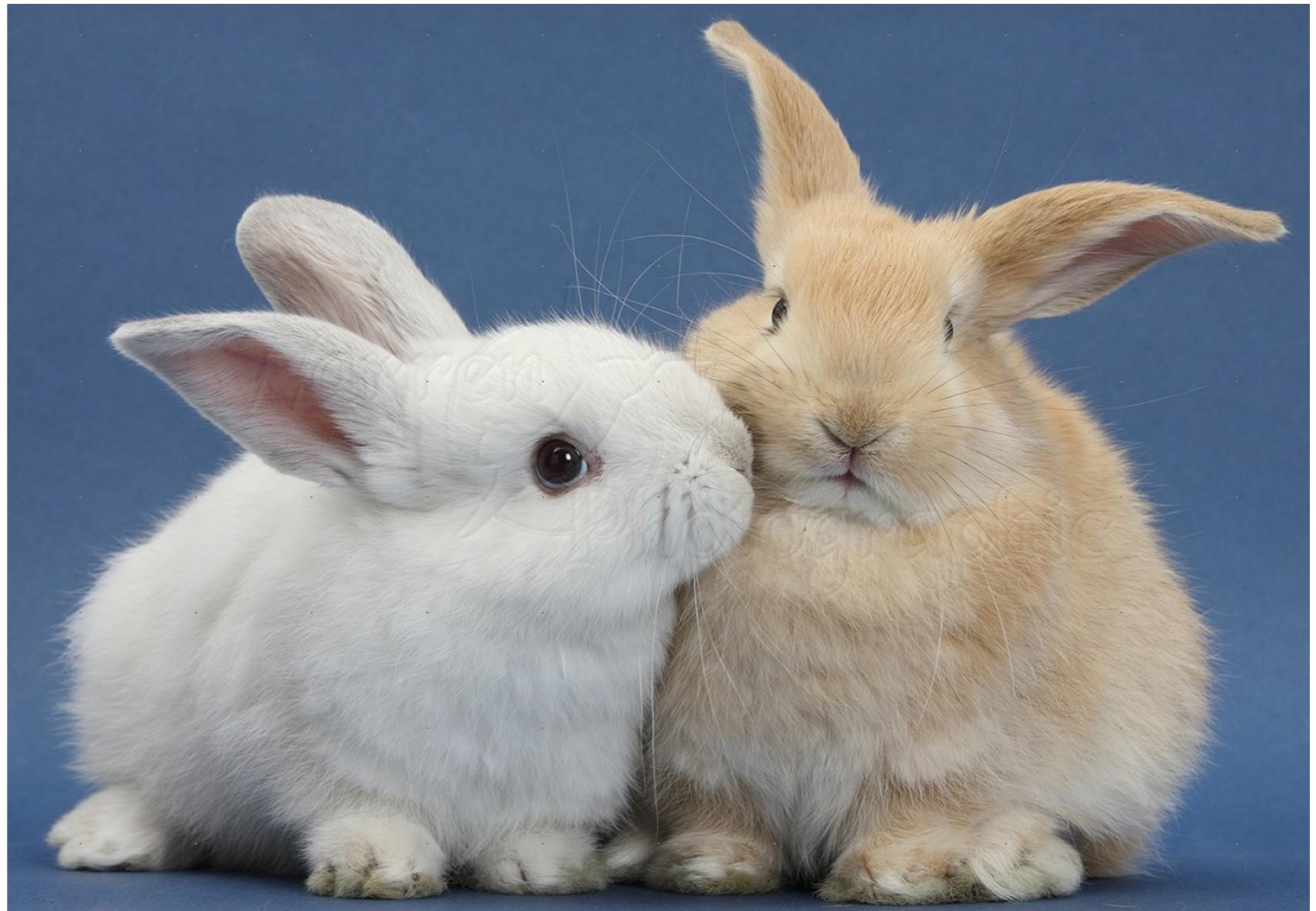
**4/ Diffusion only**

**5/ Diffusion + Reaction**

**6/ Diffusion + Reaction + Logic**

# The fundamental mystery of living matter

Something from soup (Patterns, Animals, Life...)

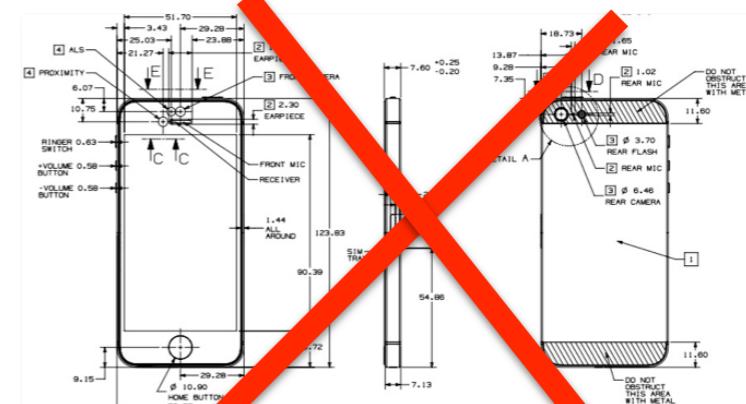


Forget macroscopic construction sites, orchestra conductors reading a score, etc.



# Explicit and Concentrated Specification

```
void Initialize_Zebra( void )
{
    zebra_length          = 2.80; /*meters*/
    zebra_stripe_period = 0.18; /*meters*/
    zebra_stripe_colorFormat(_T("Black"));
}
```



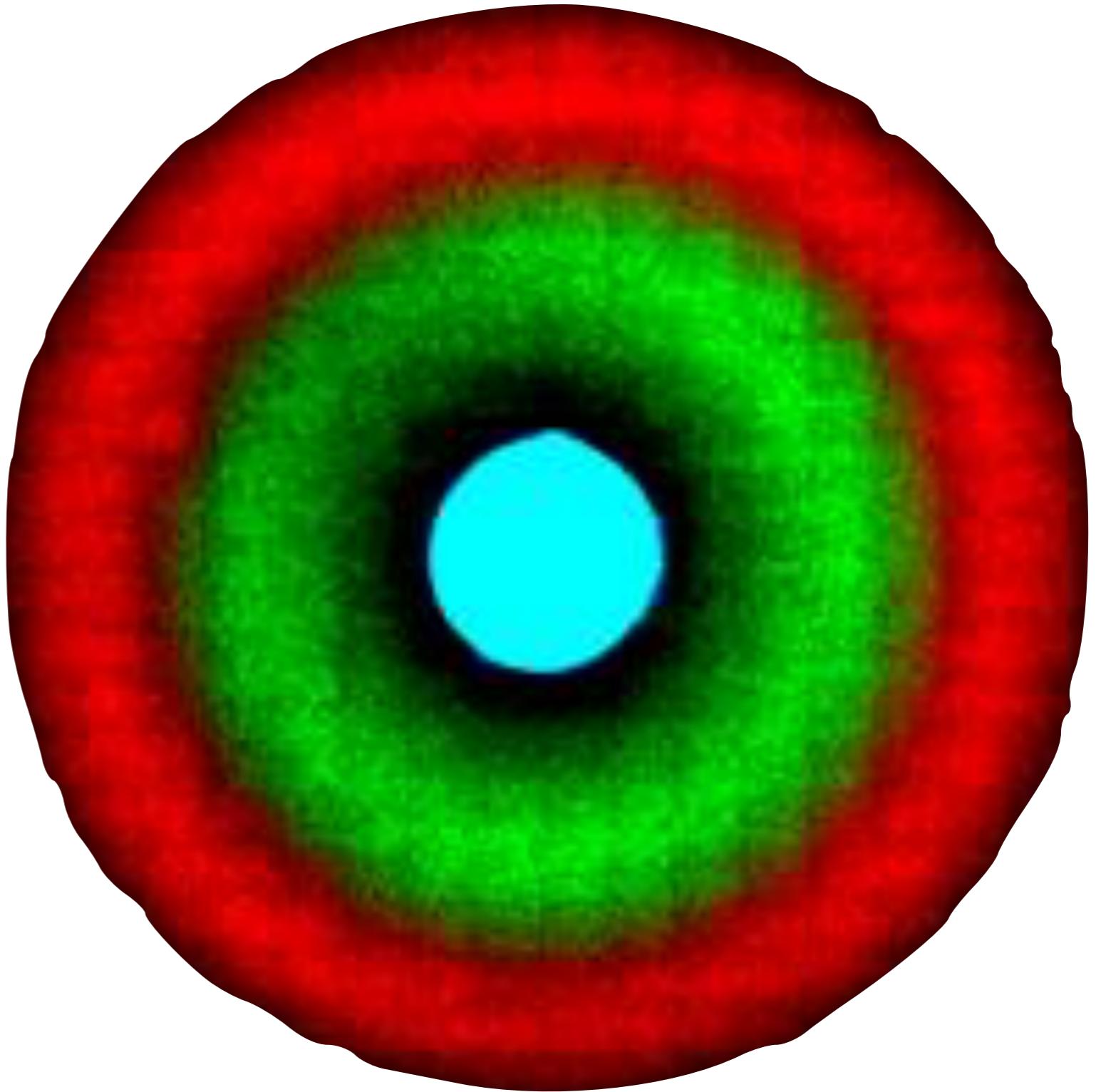
## Implicit and Distributed

Hard to hack

Key question:

For a given pattern, where and how is it stored?

Your task - make  
me this thing with  
living matter



# Thermal energy leads to Diffusion



## Atomistic view:

All particles move due to thermal energy ( $1/2 k_B T$  per DOF)

Movement direction is random

True for everything but  $1/2 k_B T$  is EXTREMELY SMALL so 'invisible' unless you are tiny!

Albert Einstein - worked out Stokes-Einstein equation in his Ph.D thesis for the diffusion coefficient of a "Stokes" particle undergoing [Brownian Motion](#) in a quiescent fluid at uniform temperature

## Phenomenological view:

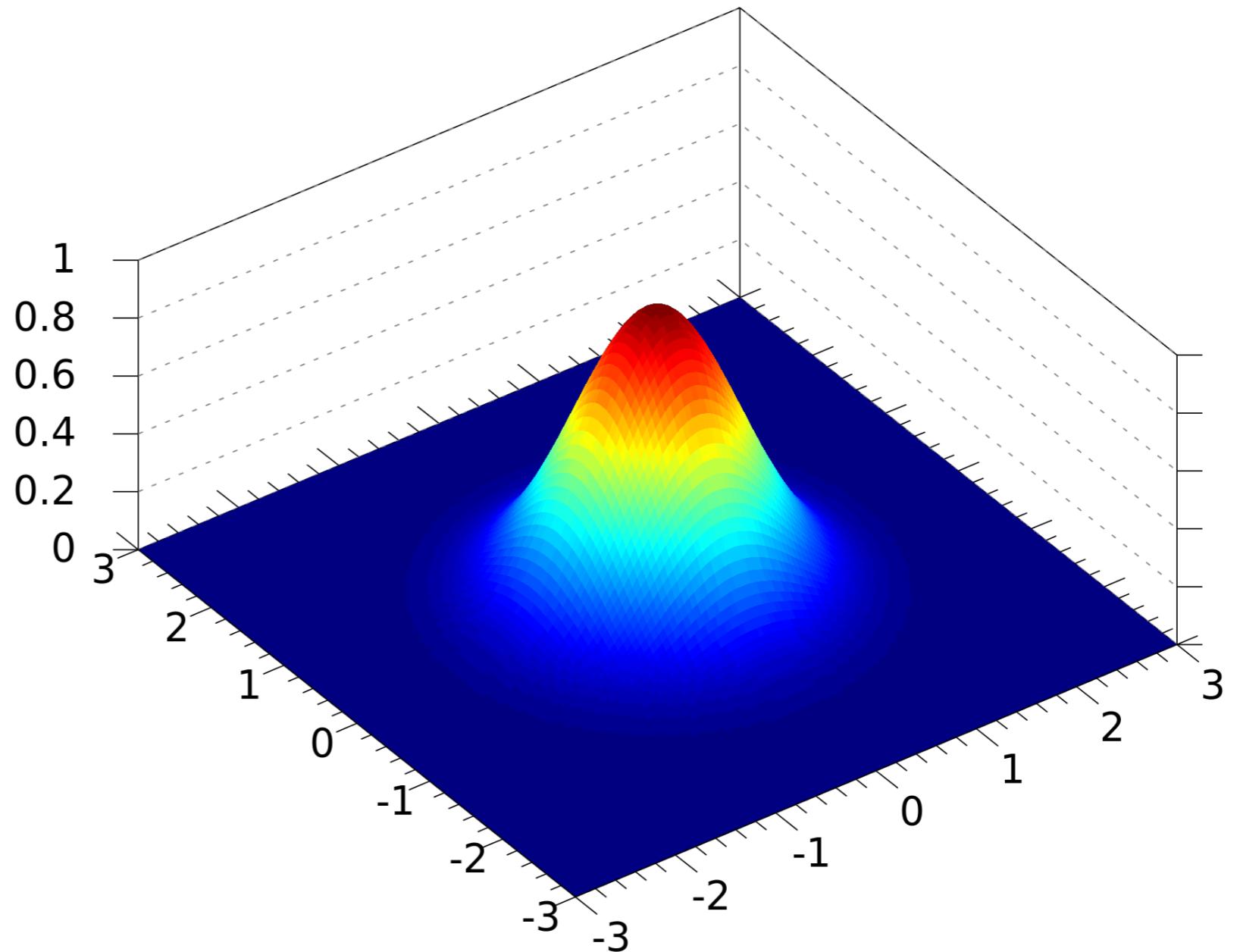
The diffusion [flux](#) is proportional to the negative [gradient](#) of concentrations

"Driving force" due to concentration difference, stuff moves until the concentration difference is zero.

# Pattern at specific time T

$$P(r,t) = \frac{e^{\left(\frac{-r^2}{4D \cdot t}\right)}}{4\pi D \cdot t}$$

$$D = \frac{k_B T}{6\pi\eta \cdot R}$$



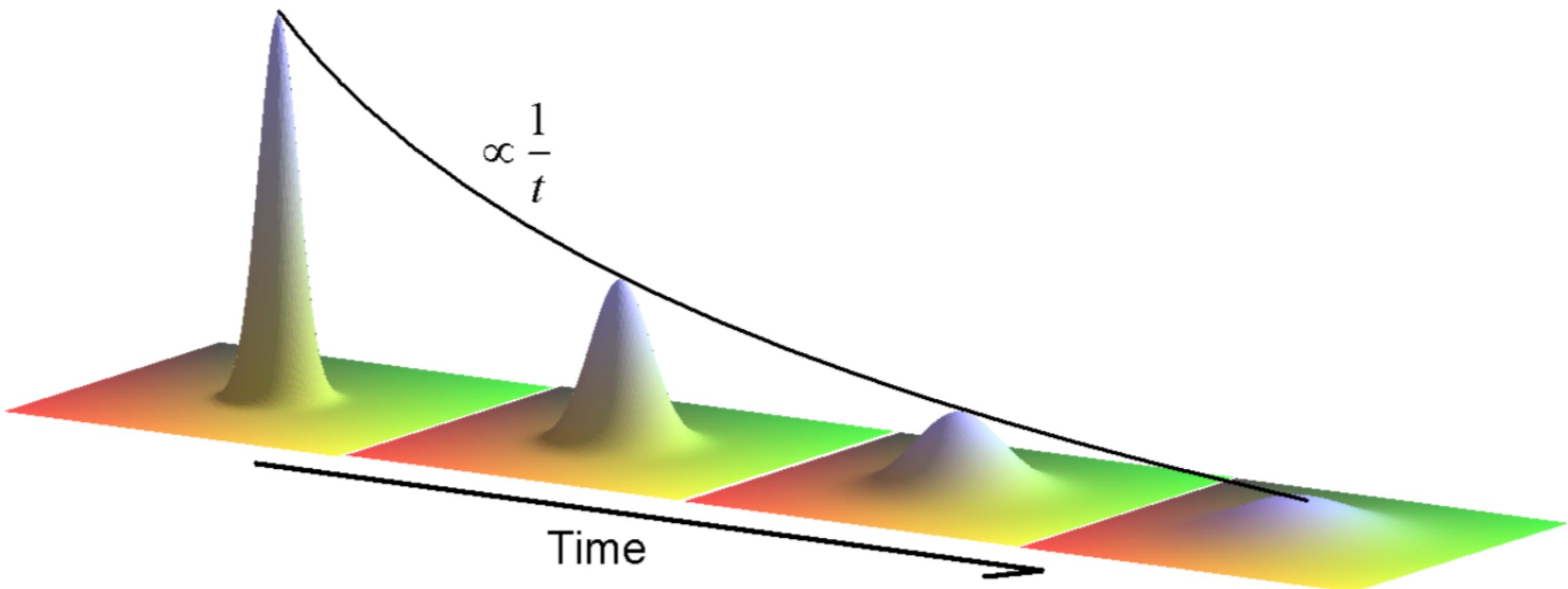
Further reading

<http://www.rpgroup.caltech.edu/courses/aph162/2006/Protocols/diffusion.pdf>

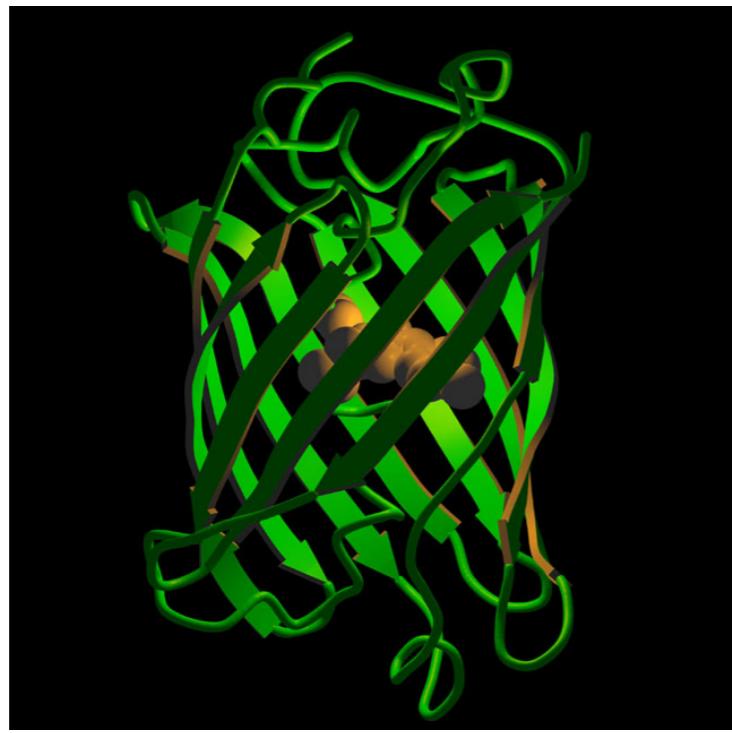
# Pattern at changes over time

$$P(r,t) = \frac{e^{\left(\frac{-r^2}{4D \cdot t}\right)}}{4\pi D \cdot t}$$

$$D = \frac{k_B T}{6\pi\eta \cdot R}$$



# Bigger proteins diffuse more slowly!



**GFP**  
~238 aa

## Sequences

Sequence	Length	Mass (Da)	Tools
<input type="checkbox"/> P42212 [UniParc]. <a href="#">FASTA</a> 238 26,886 <a href="#">Blast</a> <a href="#">go</a>			

Last modified November 1, 1995. Version 1.  
Checksum: EA5A6F21FBFB6E05

10 20 30 40 50 60  
MSKGEELFTG VVPILVELDG DVNGHKFSVS GEGEGDATYG KLTLKFICTT GKLPVWPWTL

70 80 90 100 110 120  
VTTFSYGVQC FSRYPDHMKQ HDFFKSAMPE GYVQERTIFF KDDGNYKTRA EVKFEGDTLV

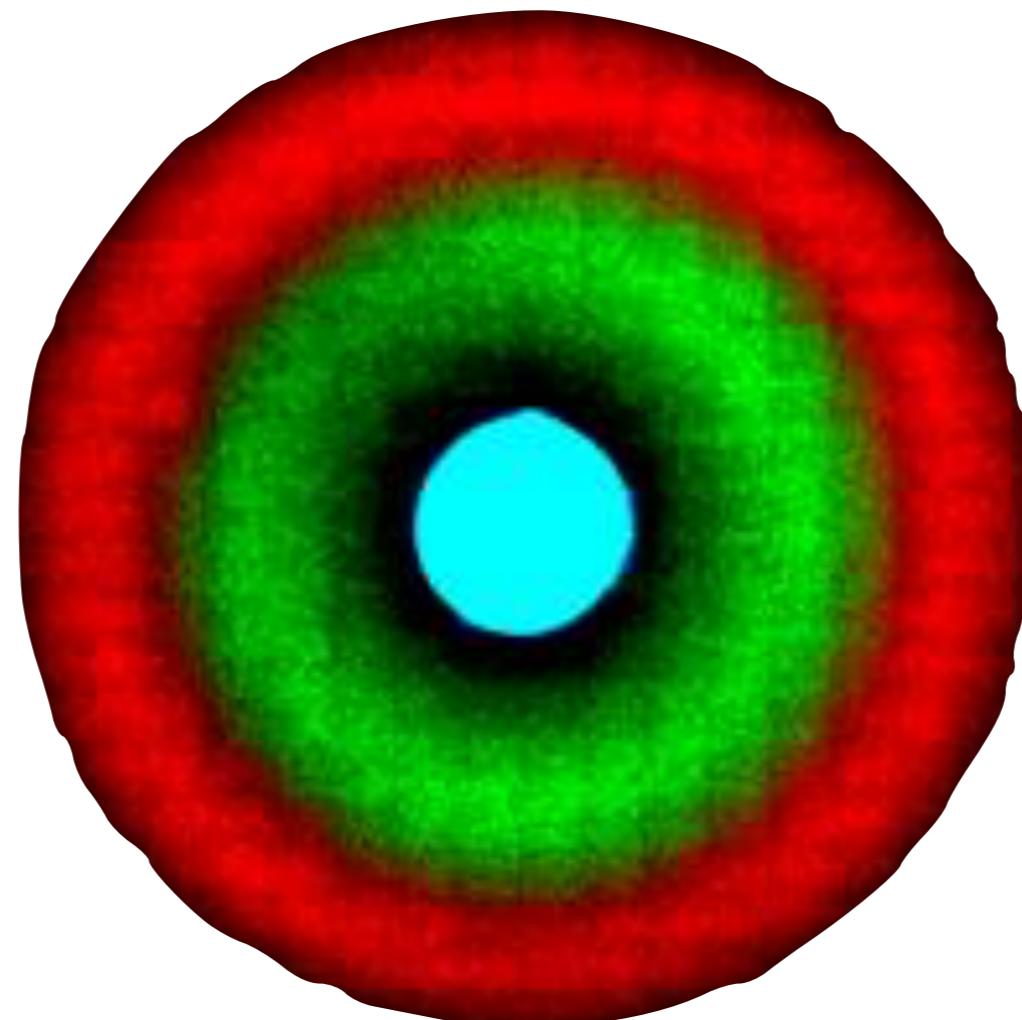
130 140 150 160 170 180  
NRIELKGIDF KEDGNILGHK LEYNYNSHNV YIMADKQKNG IKVNFKIRHN IEDGSVQLAD

190 200 210 220 230  
HYQQNTPIGD GPVLLPDNHY LSTQSALSKD PNEKRDHMVL LEFVTAAGIT HGMDELYK

[« Hide](#)

# Thought Experiment

Imagine you had a genome that encodes a **large green protein** and a **small red protein**, and the cell secretes these proteins. Which pattern will you get?



# Implicit and Distributed

Hard to hack

Key question:

For a given pattern, where and how is it stored?

For a given pattern, where and how is it stored?

**Protein size - YES**

E.g. the green protein is 238 aa, and red protein is 450 aa

**Relative protein sizes - YES - 238/450**

**Protein production rate - YES** - for Green238 and Red450

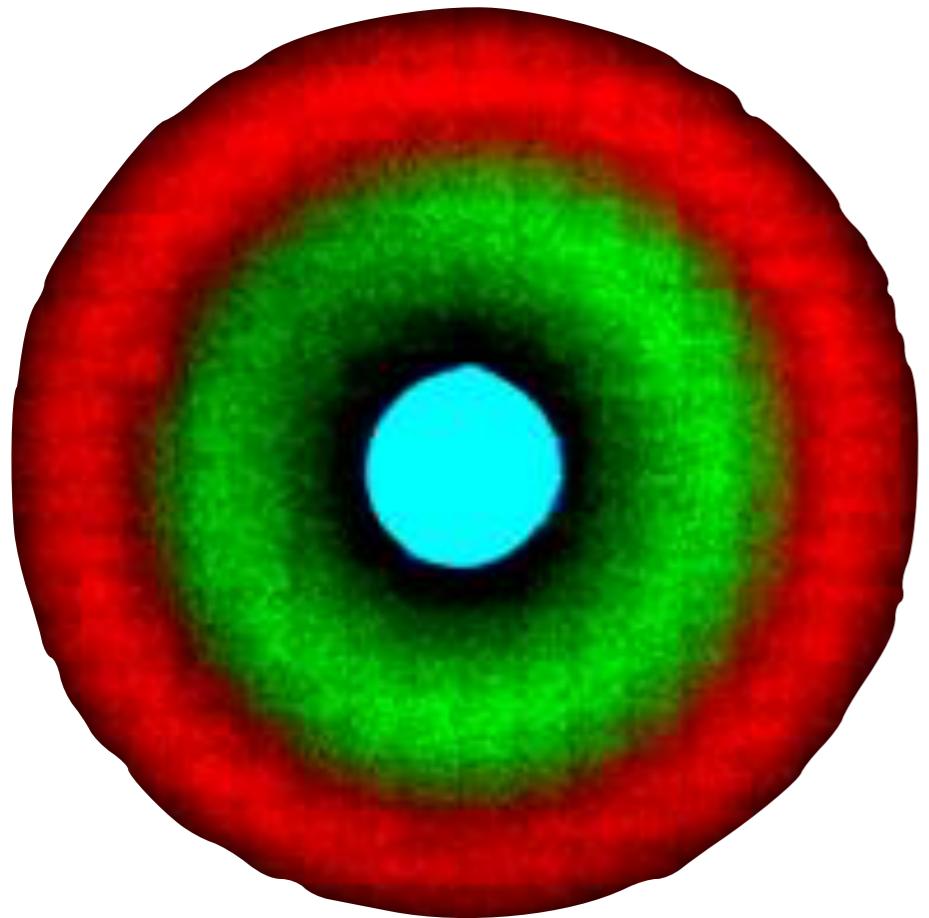
$$D = \frac{k_B T}{6\pi\eta \cdot R}$$

**Temperature - Yes**

**Viscosity of water - Yes**

*Our simple bullseye pattern is stored across (at least) 5 places, three of which are good purchase points aka easy to hack*

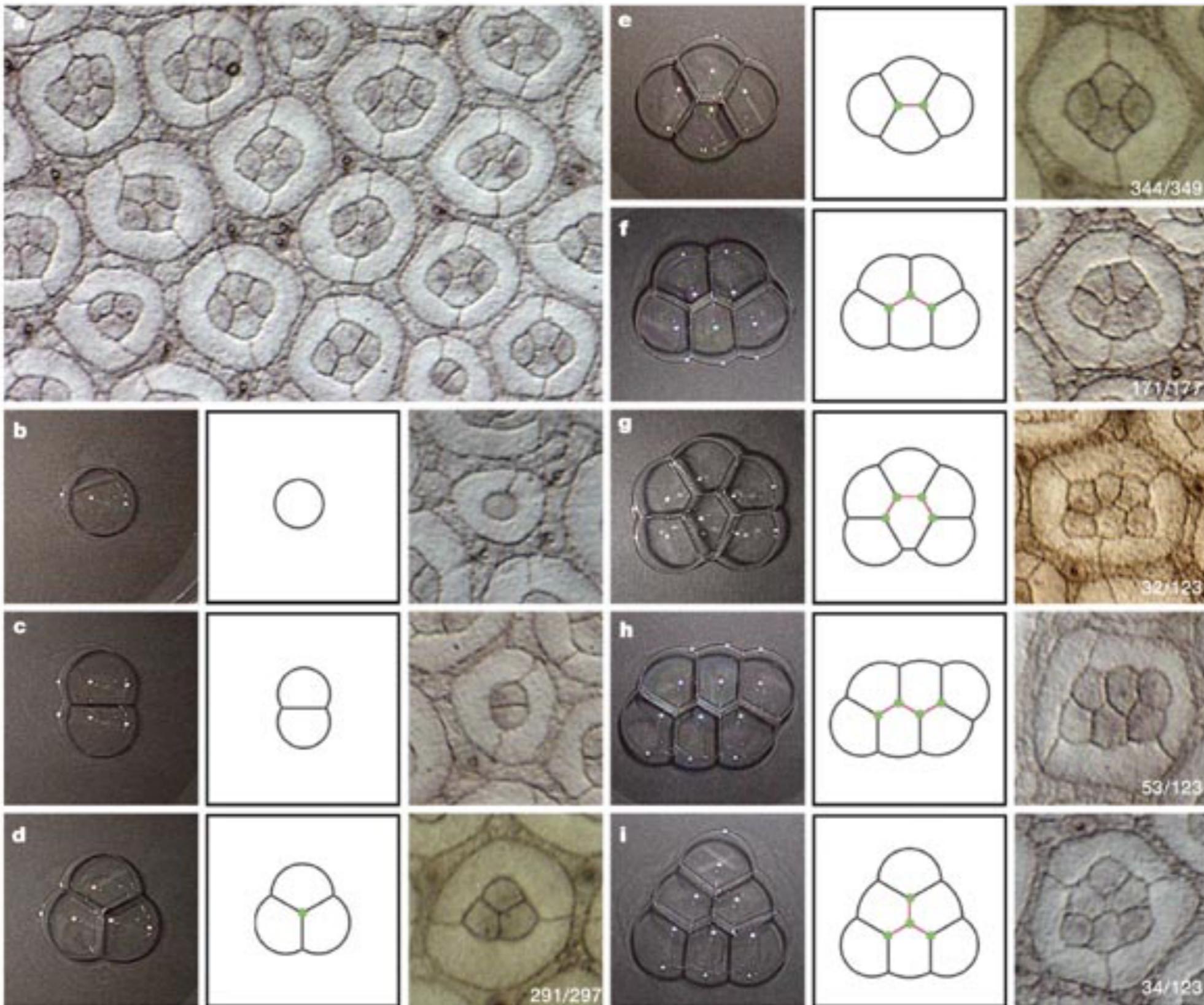
That was conceptually useful, but, there is no way you can tell me that diffusion, surface tension, + genome etc. can give more interesting patterns...



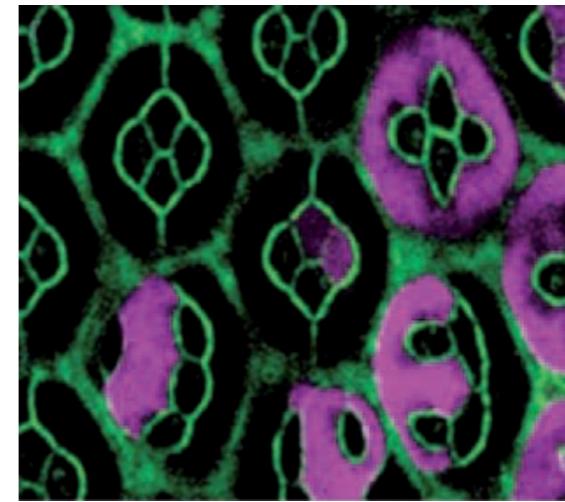
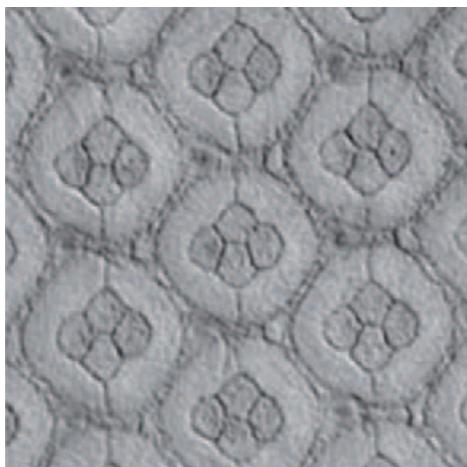
Nevermind...

# Surface mechanics mediate pattern formation in the developing retina

Takashi Hayashi & Richard W. Carthew



# More realistic expression (and a taste of 21st century biology)



Physical modeling of cell geometric order in an epithelial tissue,  
S. Hilgenfeldt, S. Erisken, and RW. Carthew, *PNAS* (2008).

Energy functional ~  
architecture

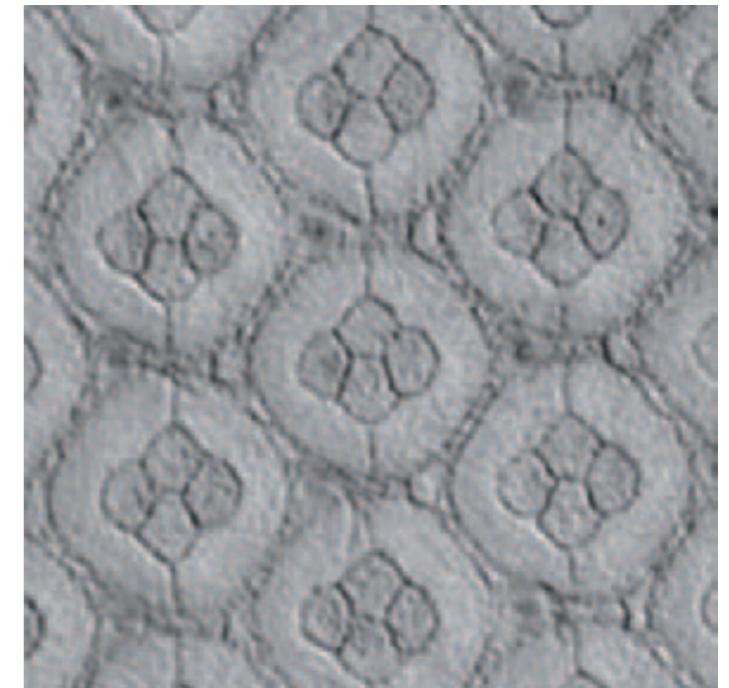
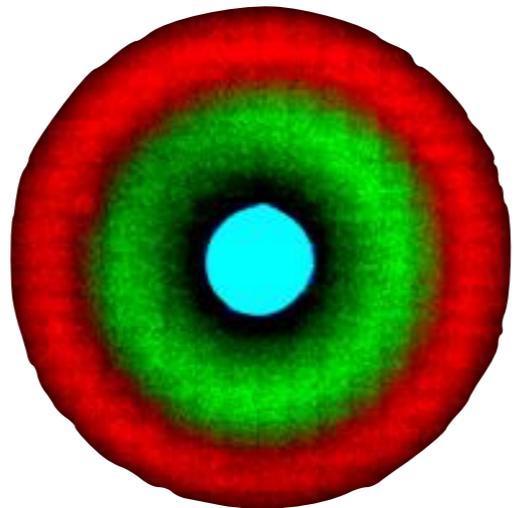
$$\mathcal{E} = \sum_i \frac{1}{2} \Delta_i^2 L_{0i} - \sum_{i,j} L_{ij} \gamma_E \delta_{i,E} \delta_{j,E} - \sum_{i,j} L_{ij} \gamma_N \delta_{i,N} \delta_{j,N}$$

Membrane bending energy  
straight out of physics/eng. textbook

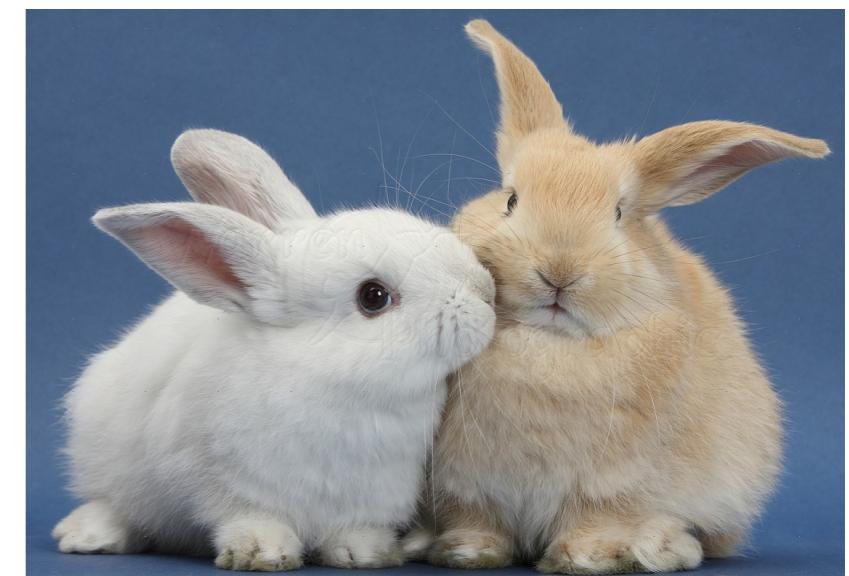
E-cadherin and N-cadherin structural biology  
Expression levels, noise, gene regulation  
(transcription, translation, microRNAs, nuclear export)  
Cancer genomics  
Epigenetics/Chromatin Architecture  
Developmental Biology

# The fundamental mystery of living matter

Something from soup (Patterns, Animals, Life...)



??? ↔

A black double-headed arrow symbol indicating a relationship or comparison between the rabbits and the question marks.

# The fundamental mystery of living matter

## Major limitations of patterns so far

- ★ Not permanent/stable - change with time, and vanish at  $t = \text{Inf}$
- ★ Cover only one length scale
- ★ Are radially symmetric - good for eyes, spheres, and donuts....

## What are we missing?

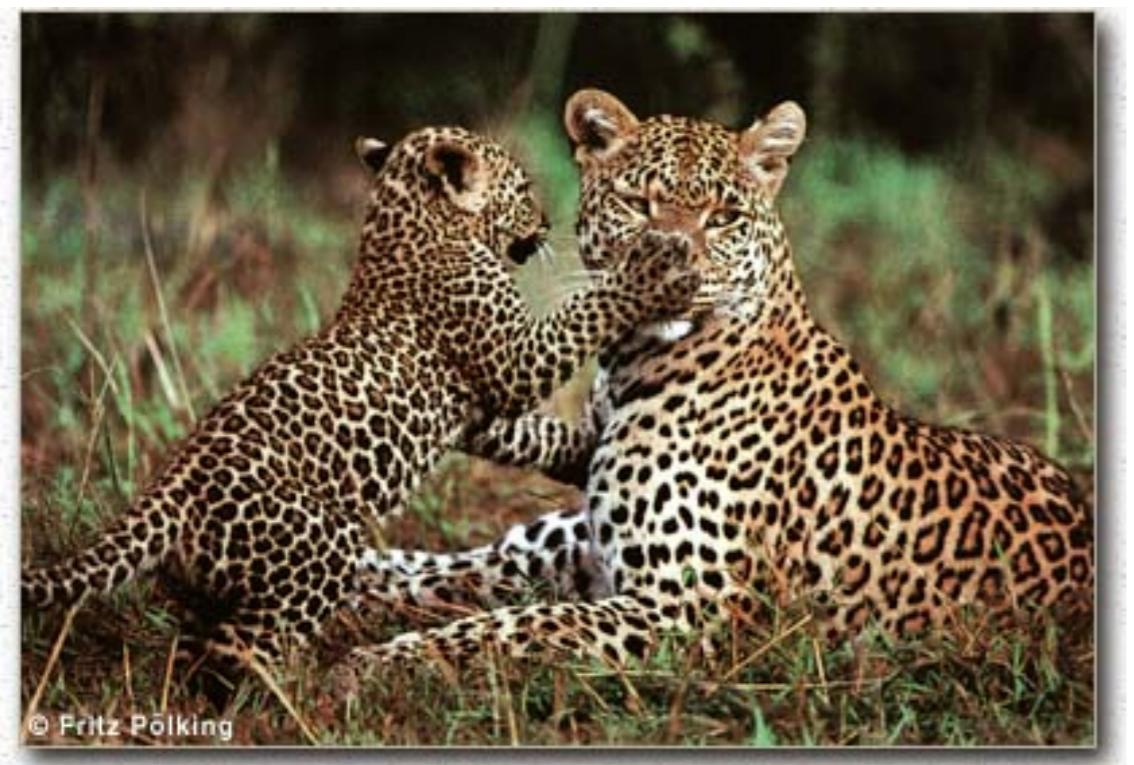
You already answered this last time - *what if the components could react?*

*Stick together, create products, degrade one another,...*

# Stripes, swirls, and dots



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© Fritz Polking



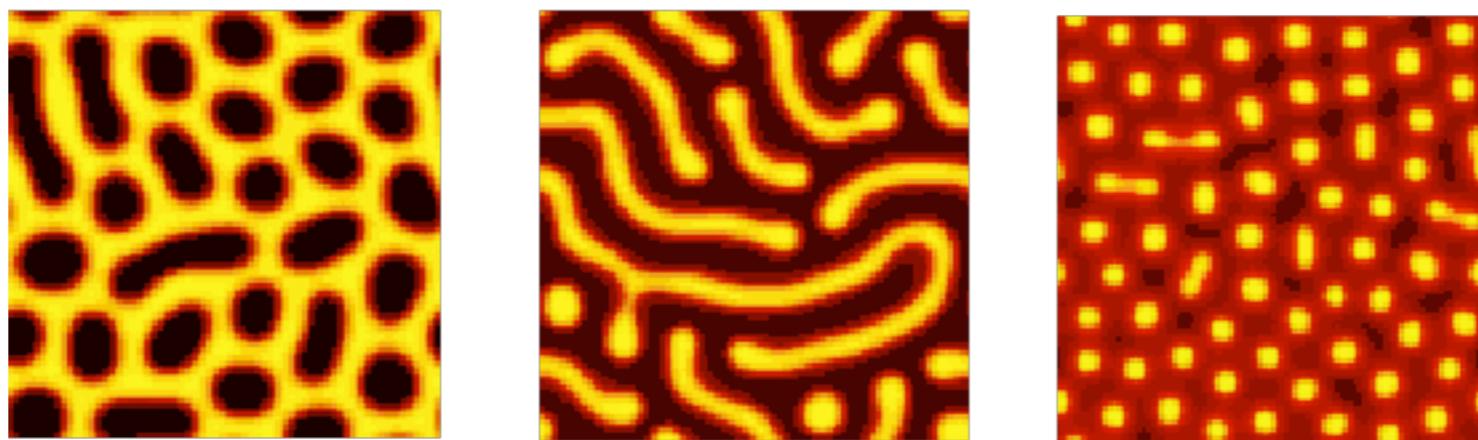
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# Turing patterns



Chemical morphogenesis (Turing, 1952):

- A system of reacting and diffusing chemical species can spontaneously form stationary spatial patterns given a certain set of chemically plausible mechanisms.
- Two reacting chemical species that diffuse at very different rates.
- The system is an intrinsically non-equilibrium; both substances are continuously created (by the cells) at every point in space, and also decay or are removed at specified rates.
- In these reactions the activator makes more activator and inhibitor, and the inhibitor destroys the activator.



Erik M. Rauch and Mark M. Millonas, J. Theoretical Biology

# THE CHEMICAL BASIS OF MORPHOGENESIS

BY A. M. TURING, F.R.S. *University of Manchester*

(Received 9 November 1951—Revised 15 March 1952)

It is suggested that a system of chemical substances, called morphogens, reacting together and diffusing through a tissue, is adequate to account for the main phenomena of morphogenesis. Such a system, although it may originally be quite homogeneous, may later develop a pattern or structure due to an instability of the homogeneous equilibrium, which is triggered off by random disturbances. Such reaction-diffusion systems are considered in some detail in the case of an isolated ring of cells, a mathematically convenient, though biologically unusual system. The investigation is chiefly concerned with the onset of instability. It is found that there are six essentially different forms which this may take. In the most interesting form stationary waves appear on the ring. It is suggested that this might account, for instance, for the tentacle patterns on *Hydra* and for whorled leaves. A system of reactions and diffusion on a sphere is also considered. Such a system appears to account for gastrulation. Another reaction system in two dimensions gives rise to patterns reminiscent of dappling. It is also suggested that stationary waves in two dimensions could account for the phenomena of phyllotaxis.

# Reaction-diffusion equations

$$U_t = D_U \Delta U + f(U, V),$$

$$V_t = D_V \Delta V + g(U, V).$$

“The Chemical Basis of Morphogenesis”, Alan Turing  
(1952 *Phil. Trans. Roy. Soc.*)

**Reaction-Diffusion Equation (1)**

$$U_t = D_U \Delta U + f(U, V), \\ V_t = D_V \Delta V + g(U, V).$$

**Reaction Equation (2)**

$$U_t = f(U, V), \\ V_t = g(U, V).$$

A constant solution  $u(t,x)=u_0, v(t,x)=v_0$  can be a stable solution of (2), but an unstable solution of (1). *Thus the instability is induced by diffusion.*

Or more simply - if you combine reaction with diffusion, you can generate instabilities - you do not need an external agent to provide lighting bolt.

# Turing patterns

Unfortunately, the math in this paper is not so easy - you will feel more comfortable with this as senior than today. BUT - at least this way, you know *why* it's good to take math classes.

This will not be on an exam and not be in a problem set.

Two morphogens are considered. They will be called  $X$  and  $Y$ , and the same letters will be used for their concentrations. This will not lead to any real confusion. The diffusion constant for  $X$  will be assumed to be  $5 \times 10^{-8}$  cm $^2$ /s and that for  $Y$  to be  $2.5 \times 10^{-8}$  cm $^2$ /s. With cells of diameter 0.01 cm

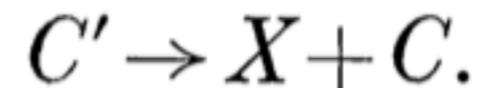
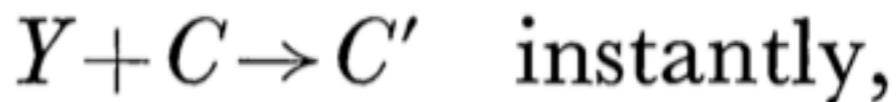
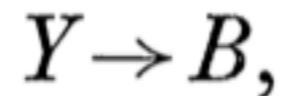
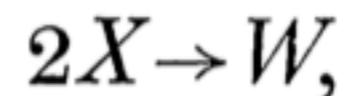
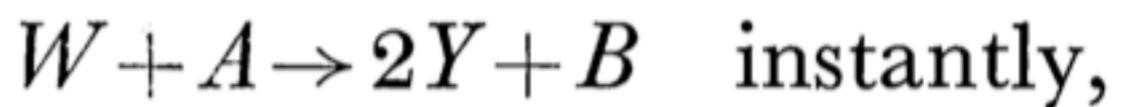
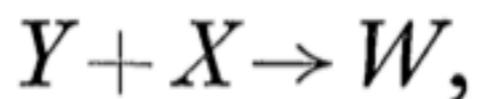
this means that  $X$  flows between neighbouring cells at the rate  $5 \times 10^{-4}$  of the difference of  $X$ -content of the two cells per second.

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The reactions postulated are



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- |                        |   |
|------------------------|---|
| $Y + X \rightarrow W$  | at the rate $\frac{25}{16} YX$ ,                |
| $2X \rightarrow W$     | at the rate $\frac{7}{64} X^2$ ,                |
| $A \rightarrow X$      | at the rate $\frac{1}{16} \times 10^{-3} A$ ,   |
| $C' \rightarrow X + C$ | at the rate $\frac{55}{32} \times 10^{+3} C'$ , |
| $Y \rightarrow B$      | at the rate $\frac{1}{16} Y$ .                  |

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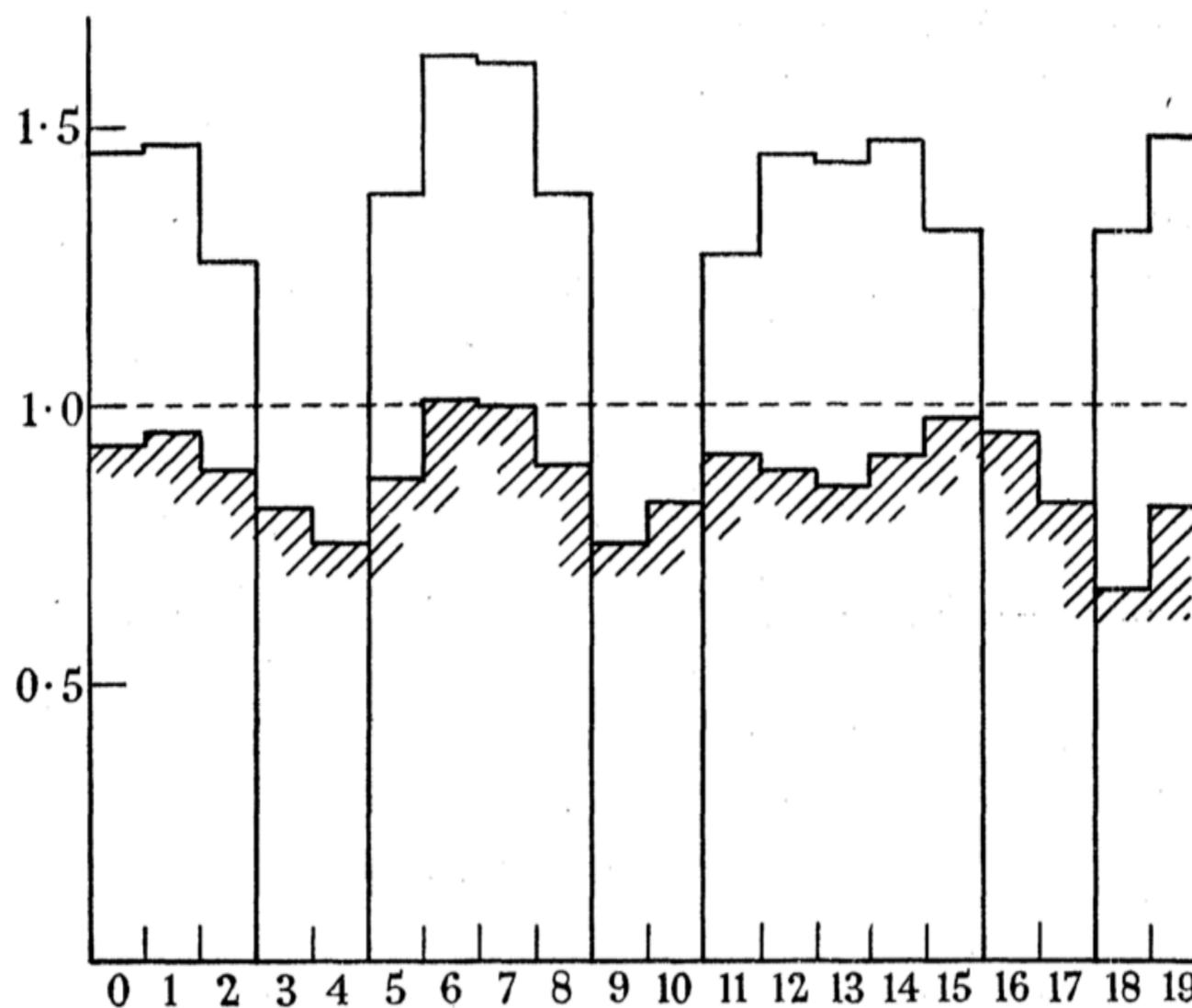


FIGURE 3. Concentrations of  $Y$  in the development of the first specimen (taken from table 1).  
----- original homogeneous equilibrium; // incipient pattern; — final equilibrium.

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# Turing patterns

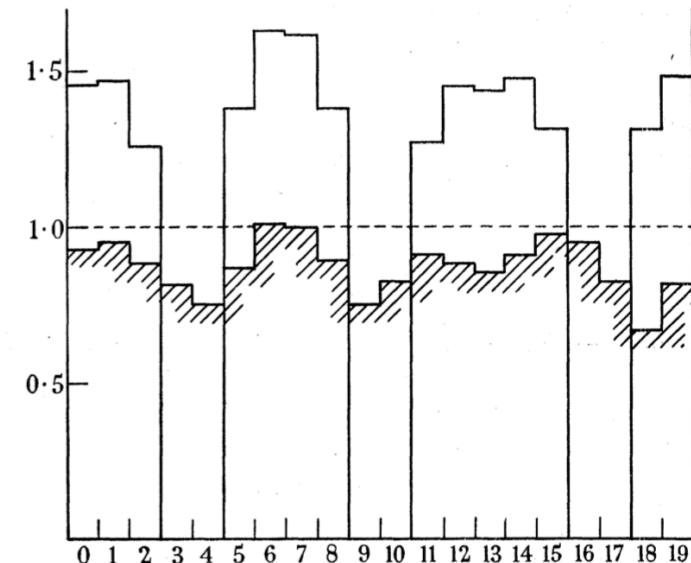


FIGURE 3. Concentrations of  $Y$  in the development of the first specimen (taken from table 1).  
----- original homogeneous equilibrium; ////////////// incipient pattern; —— final equilibrium.

- 1/ You do not need bolt of electricity to "activate" Frankenstein
- 2/ If a system has both reaction and diffusion, you can get instabilities, triggered by random small disturbances. Then, 2D and 3D patterns form. No lightning needed. The system "wakes up" all by itself.
- 3/ Can generate static 2D and 3D patterns, but also dynamic 2D and 3D movies.

# Waves in the Belousov-Zhabotinsky reaction



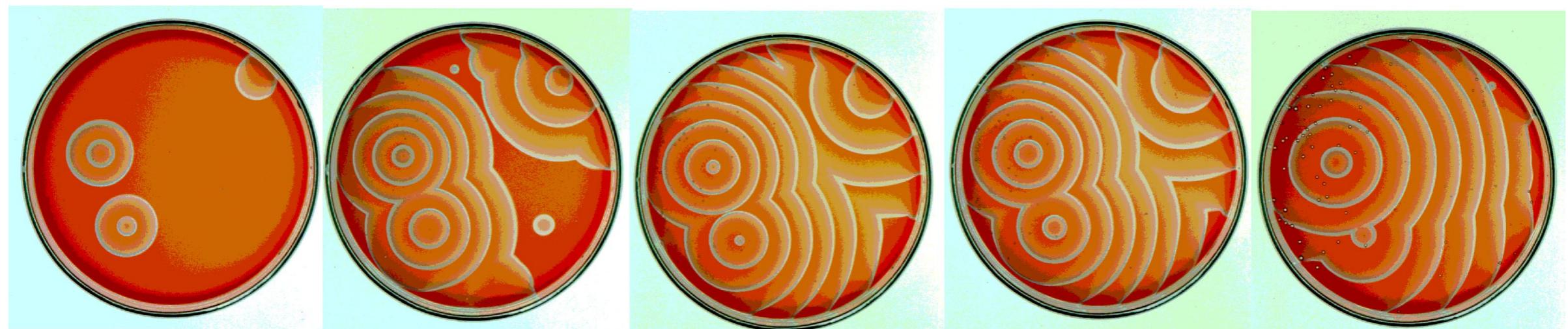
Boris P. Belousov

(Soviet Union, 1951, left)



Anatol M. Zhabotinsky

(Soviet Union, 1961, right)



Chemical reactions can be oscillatory (periodic)!



# Implicit and Distributed

Hard to hack

Key question:

For a given pattern, where and how is it stored?

For a given pattern, where and how is it stored?

**Protein size - YES**

**Number of distinct types of proteins - YES**

**Reaction Network - YES**

**Reaction Rates - YES**

**Initial concentrations - YES**

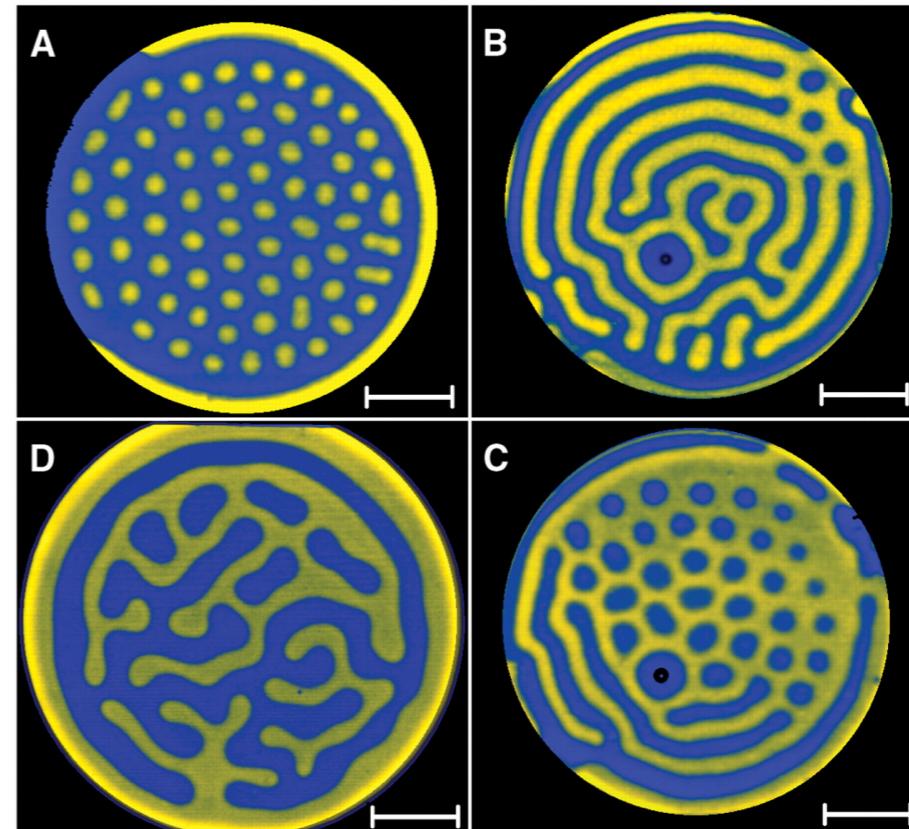
**Synthesis Rates - YES**

**Shape of enclosing dish - YES**

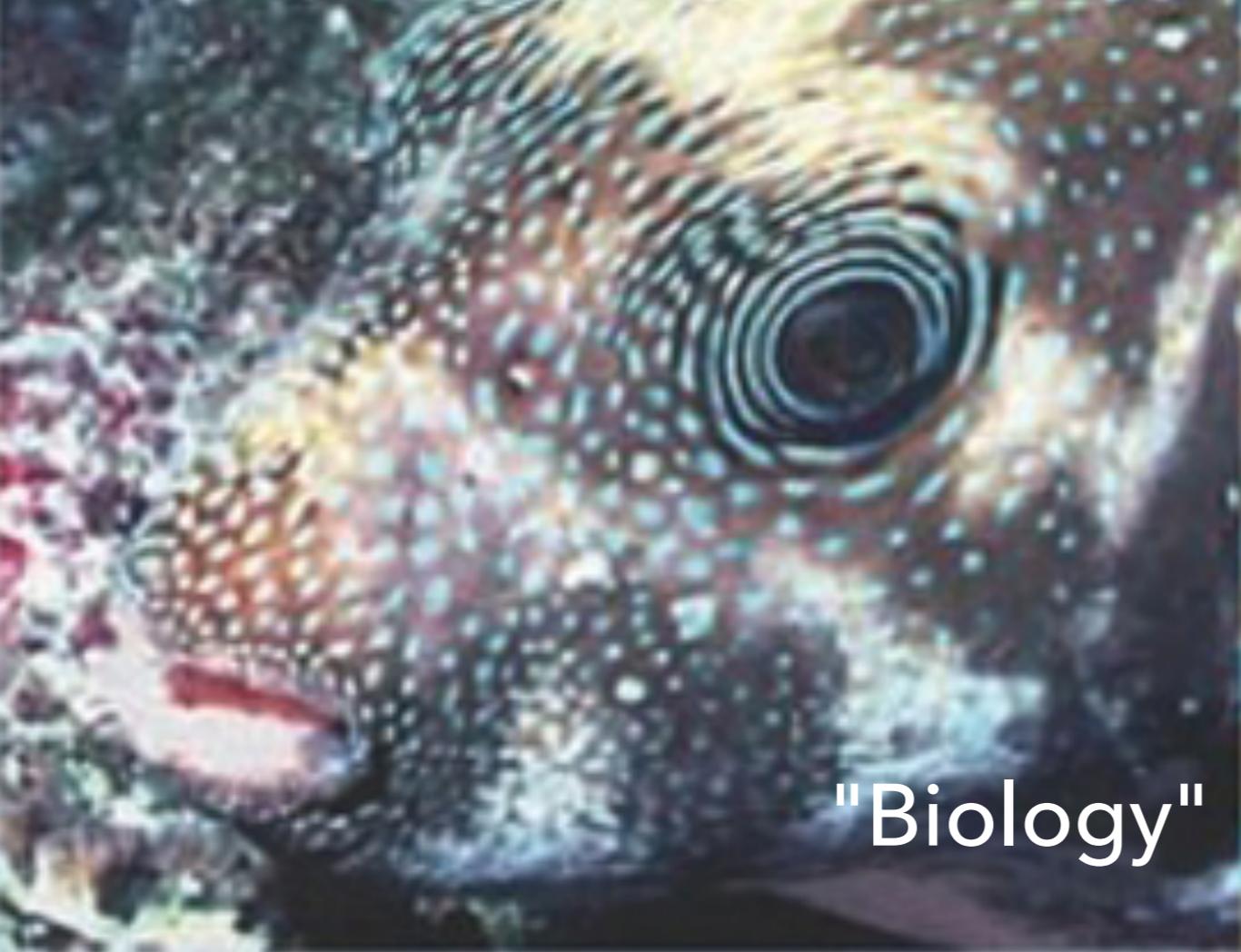
$$D = \frac{k_B T}{6\pi\eta \cdot R}$$

**Temperature - Yes**

**Viscosity of water - Yes**

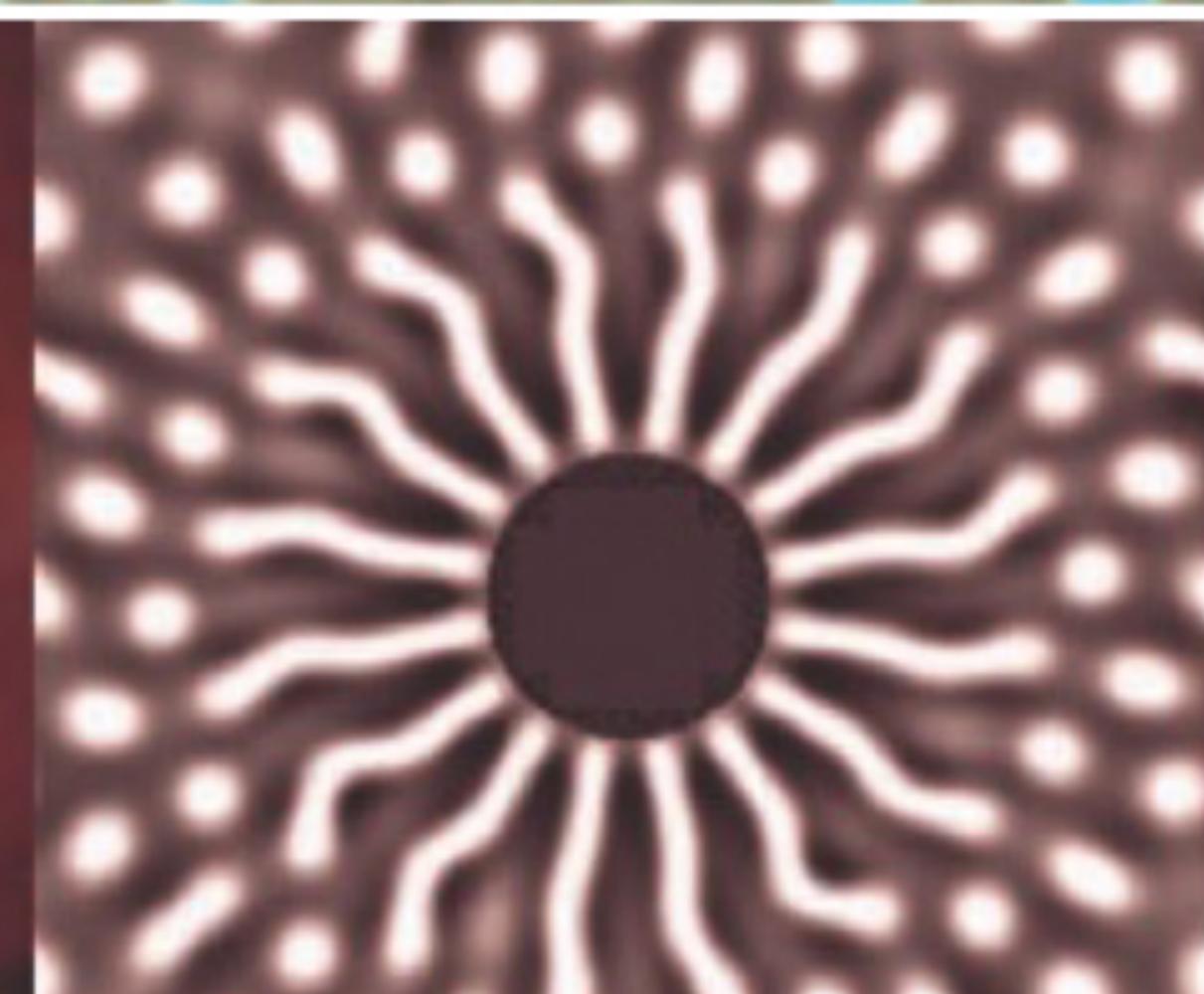
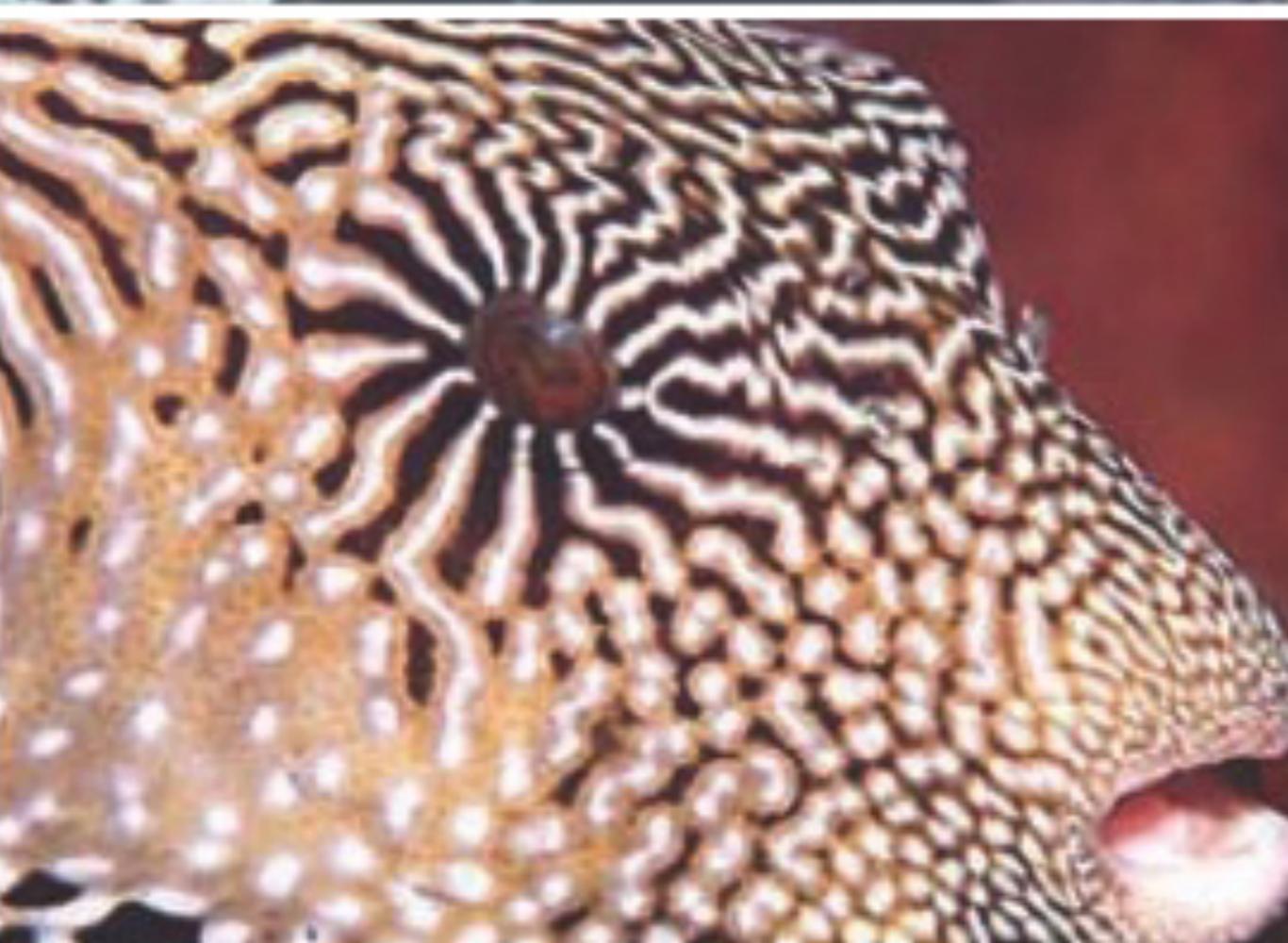


A typically biological 2D, in this framework, is stored across (at least) 9 **places**, two of which are good purchase points aka easy to hack, 5 are ok, and two are poor



"Biology"

"Programmable Reaction diffusion system"



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