# Improving Efficiency and Reducing Cost in Steam Power Plant Air Preheaters

Mei Chung, Mark Lewis, Vijay Ramanujan, Chris Stanczak

### Overview: schematic of a pulverized coal-fired boiler

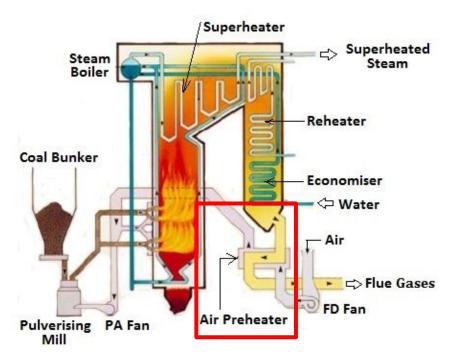


Figure 1. Schematic of a pulverized coal-fired boiler [2]

#### **Boiler:**

 Transfers heat produced by the combustion process of a fuel to the working fluid [1]

#### Air-preheater:

- Use heat from the hot flue gas to raise the temperature of the air before entering the boiler [3]
- This is important because:
  - T<sub>air</sub> increases → efficiency of the boiler also increases
- Credited with saving 25% fuel, provides up to 20% of the heat transfer in the boiler [4]

## Overview: schematic of a rotary air-preheater

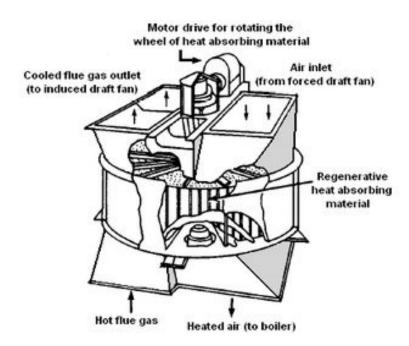


Figure 2. Schematic of rotary air preheater [4]

#### Ljungström Air Preheater Components [3]

- Rotating basket divided into multiple segments
- Heat conductive materials
  - Absorb heat from the hot flue gas, and transfer to the cold air
  - Ceramic sheets
- Seals (single or multiple)



Figure 3. Heating elements of an air preheater [5]

### Previous work done to improve the air preheater

#### On the effects of rotational speed on preheater performance:

- Increasing the rotational speed  $\rightarrow$  increased hot area of the outlet section  $\rightarrow$  better heat transfer  $\rightarrow$  T<sub>air,out</sub> increased and T<sub>gas,out</sub> decreased  $\rightarrow$  increased efficiency [6]
  - Efficiency is only increased up to a certain limit
  - Trade off between better heat transfer and max/min temperatures achieved
- Velocity of the hot and cold air have more significant effect on efficiency than the rotational speed [7]

#### On the effects of materials [6]:

• Effect of plate material change on the efficiency limited

#### On the use of multiple heat sources:

- Integration of combustion air preheating system with a bypass flue (BPF) configuration [8]
  - Waste energy of the flue gas after the APH and BPF exploited to warm the primary and secondary air → preliminarily heated (by wasted flue gas) primary air and secondary air enters the APH and absorb energy from the flue gas

### Previous work done to improve the air preheater

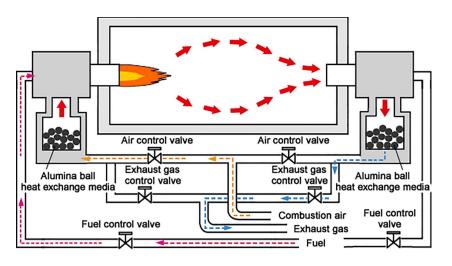


Figure 4. Regenerative burner system schematic [9]

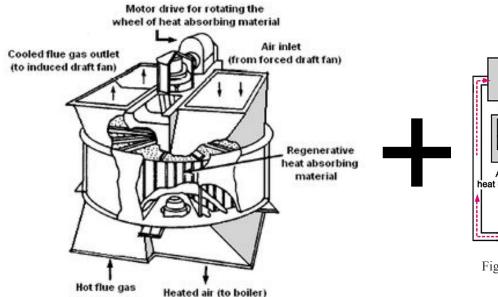
#### On Heat Recovery Burner Systems [9]:

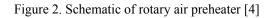
- New design for the air preheater
- Two alternating burners
- One burner is igniting the coal, opposite burner ejects exhaust gases through a medium that collects and stores heat

### Current air-preheaters can still improve effectiveness

- **Air Leakage** pressure differences between hot gas and cool air cause air leakages [10]
  - Leak rates with well designed seals should be below 10%
  - Typical leak rates are about 15-20%, with rates of larger than 30% not being uncommon [11]
- Low heat transfer and corrosion current air preheaters have low heat transfer, which is proportional to the surface area of the heating elements [3]
  - Heat transfer area is reduced due to erosion and corrosion of the heating elements, mainly due to sulfuric acid dew-point corrosion [12]
- Reduced efficiency due to fresh air purging [13]

## Our Proposed Modification





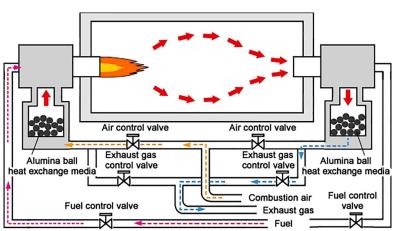
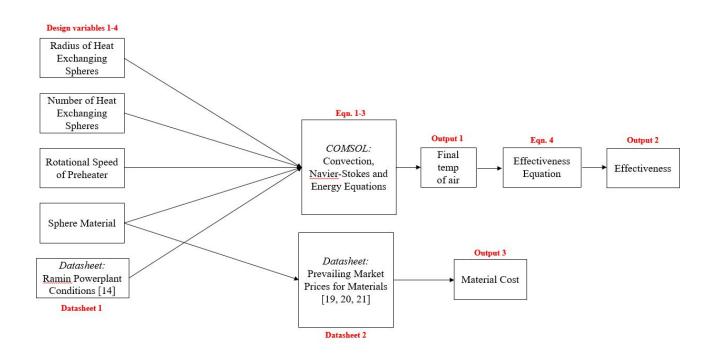


Figure 4. Regenerative burner system schematic [9]

### Our Goals: Increase effectiveness while decreasing material costs

	Goal 1	Goal 2
Problem	Low effectiveness of Ljungström air preheater due to low heat transfer	Material cost of air preheater can be lowered
Goal	Increase preheater effectiveness to 70%	Reduce material costs to \$0.650/kg
State of the Art Value from Previous Work	~61% [14, 15, 16]	\$.900 / kg [17]
Design Variable(s) to be Manipulated in this Project to Attain Target	Radius of heat exchange heating element spheres [9], number of spheres, RPM	Material of spheres

# Design Calculations Flow Diagram



# Main equations: Navier Stokes and Energy Equations [18]

$$abla \cdot ec{V} = 0$$
Divergence of Velocity

(1) Mass Continuity

$$\rho \frac{\partial \vec{V}}{\partial t} + \rho \left( \vec{V} \cdot \nabla \right) \vec{V} = -\nabla p + \mu \nabla^2 \vec{V} + \rho \vec{g}$$
Pressure Viscous Gravity
Forces Force Force

$$\rho C_p \frac{\partial T}{\partial t} + \rho C_p \overrightarrow{V} \cdot \nabla T = \nabla \cdot (k \nabla T) + Q$$
Rate of change of energy per into a point by flow

Net heat flux Internal heat source unit volume

Net heat source

# Main equations: Effectiveness Equation [14]

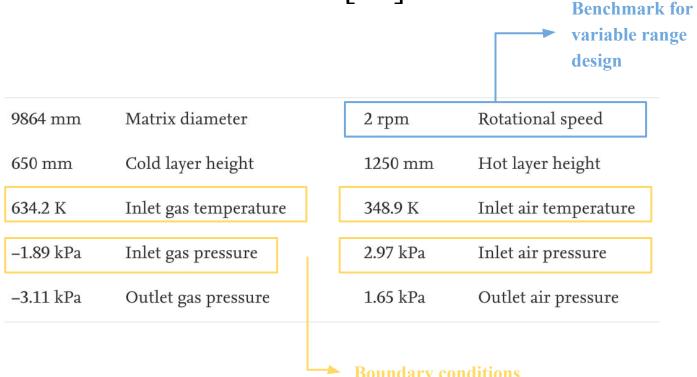
$$\epsilon = \frac{\dot{m}_{air,in}C_{P,air}(T_{air,out} - T_{air,in})}{\dot{m}_{flue,in}C_{P,flue}(T_{flue,in} - T_{air,in})}$$
(4)

- Effectiveness = ratio of heat transfer to ideal possible heat transfer
- Numerator = change in enthalpy of air
- Denominator = Change in enthalpy of incoming flue gas

```
\dot{m}_{air,in} = mass flow rate of air
C_{P,air} = heat capacity of air
T_{air, out} = outlet temperature of air
T_{air, in} = inlet temperature of air
```

 $\dot{m}_{flue,in}$  = mass flow rate of flue gas  $C_{P,flue}$  = heat capacity of flue gas  $T_{flue,in}$  = inlet temperature of flue gas

### Ramin Power Plant Datasheet [14]



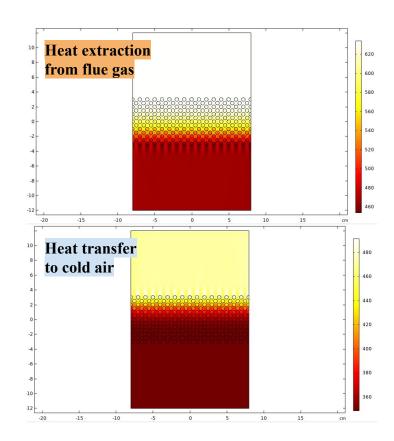
**Boundary conditions** 

### Material Cost Data Sheets

Material	Cost [\$ / ton]
Alumina	1,500 [19]
Steel	1,020 [20]
Aluminum	1,900 [21]

## COMSOL Set Up

- Iterations to simulate rotation
- Spheres of different sizes with area
   taken up by spheres held constant
- 2D heat transfer simulation



# GOAL 1

## Simulation Sweep 1: Variables and Boundary Conditions

#### Variables and Ranges:

- Ball Material: Steel, Alumina, Aluminum
- **Ball diameter:** 0.5 cm, 1 cm, 2 cm
  - Ran simulations with larger diameters, but they were not able to transfer enough heat
- **RPM:** 1, 2, 4

#### **Boundary Conditions [18]:**

- $T_{air, in} = 348.9 \text{ K}$
- $T_{gas, in} = 634.2 \text{ K}$
- $P_{gas, in} = -1.89 \text{ kPa}$
- $P_{air, in} = 2.97 \text{ kPa}$

### Simulation 1: Results

Cools down before end of rotation

Average

End T

N

536.23

389.97

N

459.36

347.85

Trial	1	2	3	4	5	6	7	8
Ball material	alumina	alumina	alumina	alumina	alumina	alumina	alumina	alumina
Ball diameter (cm)	0.5	0.5	0.5	1	1	1	2	2
Duration of half spin (s)	8.5	15	30	8.5	15	30	15	30
Cools down before end of rotation	N	N	Y	Y	N	N	N	N
Average	542.44	545.20	467.49	430.58	541.71	500.65	537.83	516.84
End T	511.70	402.47	347.86	347.58	454.61	370.55	488.48	432.11
Trial	9	10	11	12	13	14		
Ball material	aluminum	aluminum	aluminum	aluminum	aluminum	aluminum		
Ball diameter (cm)	0.5	0.5	1	1	2	2		
Duration of half spin (s)	15	30	15	30	15	30		
Cools down before end of rotation	Y	Y	N	Y	N	N		
Average	466.47	407.95	474.37	416.43	477.64	432.31		
End T	348.62	348.14	356.49	348.75	384.72	350.50		
Trial	15	16	17	18	19	20	21	
Ball material	steel	steel	Steel	steel	steel	steel	steel	
Ball diameter (cm)	0.5	0.5	1	1	1	2	2	
Duration of half spin (s)	15	30	8.5	15	30	15	30	

N

462.11

365.40

N

530.60

434.39

N

485.76

359.36

526.42

470.38

N

502.53

410.72

### Validation of Calculations: Correlation Case Study

- COMSOL heating times for the 0.5 cm diameter sphere were compared to correlations from Heat and Mass Transfer for Spheres subject to Forced Convection [22]
- Values from correlations for 0.5 cm diameter cylinders were also compared for reference
- Derivation in **Appendix IV**

Geometry	Time to for cold air to heat to T = 630 K (s)
2D Sphere (Calculated in COMSOL)	14
Sphere (Correlation)	17.5
Cylinder (Correlation)	23.2

# Simulation Sweep 2: Parameters and Boundary Conditions

#### **Parameters:**

- Material: Alumina
- **Ball diameter:** 0.5 cm
- **RPM**: 2

#### Variable:

• # Rows of Spheres

#### **Boundary Conditions [18]:**

- $T_{air, in} = 348.9 \text{ K}$
- $T_{gas, in} = 634.2 \text{ K}$
- $P_{gas, in} = -1.89 \text{ kPa}$
- $P_{air, in} = 2.97 \text{ kPa}$

### Simulation 2: Results

# Rows	1	2	3	4	5	7	9
Average T_air,out	383.23	398.62	421.61	447.68	470.92	514.18	545.20
End T_air, out	348.90	348.88	349.00	349.99	353.50	370.71	402.47
Average effectiveness	12.03%	17.43%	25.49%	34.62%	42.77%	57.93%	68.80%
Cost	\$1,706.69	\$3,413.37	\$5,120.06	\$6,826.75	\$8,533.43	\$11,946.80	\$15,360.18
# Rows	10	11	13	15	17	19	
# Rows	10	11	13	15	17	19	
Average T_air,out	555.30	564.72	577.17	585.77	588.24	596.33	
End T_air, out	421.85	441.57	473.61	497.68	511.13	532.96	
Average effectiveness	72.35%	75.65%	80.01%	83.03%	83.89%	86.73%	$\neg$
Cost	\$17,066.86	\$18,773.55	\$22,186.92	\$25,600.29	\$29,013.67	\$32,427.04	

- Effectiveness increases with the number of spheres (at a trade off of costs)
- More than 7 **rows** of spheres achieve current <u>state of the art</u> effectiveness
- More than **10 rows** exceed goal effectiveness

# GOAL 2

### Costs Comparison

- Current air preheater heating elements made of Corten Steel [23]
- Cost of Corten Steel = \$900 / ton [17]
- Cost of Alumina = \$1,500 / ton [19]
- Cost of Steel = \$1,020 / ton [20]
- Cost of Aluminum = \$1,900 / ton [21]
- Total cost of air preheater: ~\$4.3 M [24]
  - \$30K negligible, cost impact small

### Our Goals: Increase effectiveness while decreasing material costs

	Goal 1	Goal 2
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# -Thank You-

# **APPENDIX**

## I. List of contributions of each group member to project

#### • Mei Chung

Editor, overview of system design, COMSOL simulations, plot of how goal parameters vary with values of design variables

#### • Christopher Stanczak

 COMSOL simulations, schematic of system and explanation of parts, main equations used in calculations and flow diagram

#### • Mark Lewis

• COMSOL simulations, quantifiable goals for the project, discussion of how we know our calculations are correct

#### Vijay Ramanujan

 Background on system including previous work, remaining unsolved problems, and how project is addressing these challenges

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### III. Main equations: Navier Stokes and Energy Equations [18]

$$abla \cdot ec{V} = 0$$
Divergence of Velocity

(1) Mass Continuity

$$\rho \frac{\partial \vec{V}}{\partial t} + \rho \left( \vec{V} \cdot \nabla \right) \vec{V} = -\nabla p + \mu \nabla^2 \vec{V} + \rho \vec{g}$$
Pressure Viscous Gravity
Forces Forces Force

Internal Forces Forces Force

$$\rho C_p \frac{\partial T}{\partial t} + \rho C_p \overrightarrow{V} \cdot \nabla T = \nabla \cdot (k \nabla T) + Q$$
Rate of change of energy per into a point by flow
Net heat flux Internal heat source unit volume

(3) Total Energy heat source

# III. Main equations: Effectiveness Equation [14]

$$\epsilon = \frac{\dot{m}_{air,in}C_{P,air}(T_{air,out} - T_{air,in})}{\dot{m}_{flue,in}C_{P,flue}(T_{flue,in} - T_{air,in})} \tag{4}$$

- Effectiveness = ratio of heat transfer to ideal possible heat transfer
- Numerator = change in enthalpy of air
- Denominator = Change in enthalpy of incoming flue gas

```
\dot{m}_{air,in} = mass flow rate of air
C_{P,air} = heat capacity of air
T_{air, out} = outlet temperature of air
T_{air, in} = inlet temperature of air
```

 $\dot{m}_{flue,in}$  = mass flow rate of flue gas  $C_{P,flue}$  = heat capacity of flue gas  $T_{flue,in}$  = inlet temperature of flue gas

# III. Ramin Power Plant Datasheet [14]

			Benchmark for variable range design
9864 mm	Matrix diameter	2 rpm	Rotational speed
650 mm	Cold layer height	1250 mm	Hot layer height
634.2 K	Inlet gas temperature	348.9 K	Inlet air temperature
–1.89 kPa	Inlet gas pressure	2.97 kPa	Inlet air pressure
–3.11 kPa	Outlet gas pressure	1.65 kPa	Outlet air pressure
		→ Boundary con	ditions

### III. Material Cost Data Sheets

Material	Cost [\$ / ton]
Alumina	1,500 [19]
Steel	1,020 [20]
Aluminum	1,900 [21]

### IV. Correlation Calculations for 0.5 cm diameter Circular Geometries [22]

#### Thermal and Mechanical Properties of Alumina (from COMSOL)

Density (kg/m <sup>3</sup> )	3900
Heat Capacity (J/K)	900
Thermal Conductivity (W/m-K)	27

#### Thermal and Mechanical Properties of Air (from COMSOL)

Thermal Conductivity (mW/m-K)	48
Dynamic Viscosity (T = 634.2 K) (Pa-s)	4.6 x 10 <sup>-5</sup>
Dynamic Viscosity (T = 348.9 K) (Pa-s)	3.93 x 10 <sup>-5</sup>
Prandtl Number (T = 634.2 K) (Pa-s)	0.735

IV. Correlation Calculations for 0.5 cm diameter Circular Geometries [22]

1 T

$$Re_D = \frac{\rho vd}{\mu} = 3519$$

#### **Sphere Correlation**

Average Nusselt Number	Restrictions	$\Rightarrow Nu = \frac{hL}{L} = 36.64$
$\overline{Nu_D} = 2 + \left(0.4 \operatorname{Re}_D^{1/2} + 0.06 \operatorname{Re}_D^{2/3}\right) \operatorname{Pr}^{0.4} \left(\frac{\mu}{\mu_s}\right)^{1/4}$	$0.71 \le \text{Pr} \le 380$ $3.5 \le \text{Re}_D \le 7.6 \times 10^4$ $1.0 \le (\mu/\mu_s) \le 3.2$	$\Rightarrow h = 703.5 \frac{W}{m^2 \cdot K}$

#### **Cylinder Correlation**

$$\overline{Nu_D} = 0.683 \text{Re}_D^{0.466} \text{Pr}^{1/3} \quad \text{Pr} \ge 0.7 \quad \Rightarrow Nu = 27.7 \quad \Rightarrow h = 531 \frac{W}{m^2 \cdot K}$$

### IV. Correlation Calculations for 0.5 cm diameter Circular Geometries [22]

#### **Sphere**

$$Bi = rac{hL}{k_{sphere}} = 0.0651 < 0.1$$
 Since Biot number is less than 0.1, lumped capacitance method can be used

#### **Cylinder**

$$Bi = rac{hL}{k_{culinder}} = 0.0492 < 0.1$$
 Since Biot number is less than 0.1, lumped capacitance method can be used

$$T_{\infty} + (T_i - T_{\infty})exp(-\frac{hAt}{\rho V c_p}) = T - T_{\infty} \longrightarrow$$

$$T_i = 348.9K$$

$$T_{\infty} = T_{fluegas} = 634.2K$$

Geometry	Time to for cold air to heat to T = 630 K (s)
Sphere	17.5
Cylinder	23.2

### V. Nomenclature

m <sub>air,in</sub>	Mass flow rate of air (kg/s)
$C_{P,air}$	Heat capacity of air (J/K)
T <sub>air, out</sub>	Outlet temperature of air (K)
T <sub>air, in</sub>	Inlet temperature of air (K)
ṁ <sub>flue,in</sub>	Mass flow rate of flue gas (kg/s)
C <sub>P,flue</sub>	Heat capacity of flue gas (J/K)
T <sub>flue, in</sub>	Inlet temperature of flue gas (K)
ε	Air preheater effectiveness
V	Velocity (m/s)

ρ	Density of fluid (2D) (kg / m <sup>2</sup> )
t	Time (s)
p	Pressure of fluid (Pa)
μ	Dynamic viscosity (Pa-s)
g	Gravitational acceleration (m / s <sup>2</sup> )
T	Temperature (K)
k	Thermal conductivity (W / m-K)
Q	Heat generated (J/s-m <sup>2</sup> )

### VI. Nomenclature

$Re_D$	Reynolds Number
d	Diameter (m)
Nu	Nusselt Number
h	Convection coefficient (W/m²-K)
Pr	Prandtl Number
Bi	Biot Number
T <sub>i</sub>	Initial temperature (K)
T <sub>∞</sub>	Surrounding fluid temperature (K)
V	Volume (m <sup>3</sup> )

### VII. COMSOL Datasheets

• See attached Excel file