

# Title: Solubility

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8/31/21

Purpose: To determine whether a variety of organic compounds are soluble or insoluble in various solvents

## Table of Chemical Data

Name	MW	MP	BP	d	safety
Bi phenyl	154.21	71	255	0.992	environmental Hazard, irritant
Hexane	86.18	-95	69	0.654	Flammable, environmental Hazard, Health Hazard, Irritant
Methanol	32.04	-98	64.7	0.79	Flammable, Toxic, Health Hazard
1-octanol	130.23	-16	195	0.824	Irritant
ethanol	46.07	-90	77	0.8	Flammable, Toxic, Health Hazard, Irritant
diethyl ether	74.12	-116	34.6	0.706	Health Hazard, Flammable, Irritant
Methylene Chloride	84.93	-97	39	1.33	Health Hazard, Irritant
Dichloromethane	84.93	-97	39	1.33	Health Hazard, Irritant
Sodium Chloride (M)	58.44	0	100	1.04	
hydrochloric acid	36.46	-74	81.5	1.2	Corrosive, Irritant



Predictions: As mentioned in the introduction the solubility of these compounds will likely relate with polarity. Therefore compounds like hexane and biphenyl won't be miscible with  $H_2O$ , while the alcohols and certain organic molecules will.

### Procedure:

A: 4mg biphenyl in 2 test tubes  
add 1ml of water to 1 and 1ml  
of hexane to other

repeat with Succinic acid

biphenyl

Succinic Acid

B: add 1ml of  $H_2O$  and Hexane  
to separate test tubes. add methanol  
of 1-octanol and shake

Methanol

1-octanol

C: Make predictions for each  
pair then add 1ml of both  
and observe

$H_2O$  + ethanol

$H_2O$  + diethyl ether

$H_2O$  + Methylene chloride

$H_2O$  + hexane

hexane + Methylene chloride



D. ~~the~~ 30 mg of benzoic acid in ?  
add water, NaOH, DCM  
then add HCl to NaOH