21 At room temperature, the density of liquid mercury is five times greater than the density of solid aluminium.

What is the reason for this?

- A Aluminium atoms are spaced widely apart.
- **B** Aluminium atoms move more freely than mercury atoms.
- **C** Atoms in a liquid take up less space than atoms in a solid.
- **D** Mercury atoms have greater mass than aluminium atoms.
- **22** When white sugar granules are heated, they melt. When the melt is cooled quickly, a brittle solid form of toffee is produced.

How does the structure of the sugar change?

- A amorphous to polymeric
- B crystalline to amorphous
- C crystalline to polymeric
- D polymeric to amorphous
- 23 Which property of a metal wire depends on its Young modulus?
 - **A** ductility
 - **B** elastic limit
 - C spring constant
 - **D** ultimate tensile stress

Space for working