

37 Hydrogen and deuterium can be represented by the nuclide symbols ${}^1_1\text{H}$ and ${}^2_1\text{H}$ respectively.

What is a difference between hydrogen and deuterium?

- A** The deuterium atom has twice the number of electrons as the hydrogen atom.
- B** The deuterium nucleus has a charge, but the hydrogen nucleus has no charge.
- C** The deuterium nucleus has less mass than the hydrogen nucleus.
- D** The deuterium nucleus has half the charge per unit mass of the hydrogen nucleus.

38 A radioactive sample decays by emitting β^- particles.

The energy released in the decay process is the same for each nucleus that decays, but the β^- particles emitted have a continuous range of kinetic energies.

Which statement explains why the β^- particles are emitted with a continuous range of kinetic energies?

- A** Some of the energy released is given to the remaining nucleons in the nucleus.
- B** Some of the energy released is taken by an emitted antineutrino.
- C** Some of the energy released is used to create the β^- particle.
- D** Some of the energy released is used to create a new nucleon.

39 Which particle is **not** a fundamental particle?

- A** electron
- B** neutrino
- C** neutron
- D** top quark

40 What is the charge of an anti-top quark?

- A** $-\frac{2}{3}e$
- B** $-\frac{1}{3}e$
- C** $+\frac{1}{3}e$
- D** $+\frac{2}{3}e$