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Heat Treatment of AA6061 Aluminium Alloy

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Abstract:

This study examines how different aging treatments influence the hardness of AA6061 aluminum alloy. Tensile specimens were solution heat-treated at 530°C for 20 minutes and then aged under various conditions—T6 temper at 160°C, T4 temper at room temperature, and S3 with limited measurements—with hardness measured via the HRF-30T test. The results demonstrate that artificial aging (T6) rapidly increases hardness, peaking within 24 hours, while natural aging (T4) yields a slower, more gradual enhancement over several weeks.

Introduction:

This experiment investigates the effects of various heat treatments on the hardness of AA6061 aluminum alloy, a material renowned for its balance of strength, corrosion resistance, and formability. By subjecting specimens to a solution treatment followed by different aging conditions—namely artificial aging at 160°C (T6 temper), natural aging at room temperature (T4 temper), and limited aging for the S3 sample—the study aims to elucidate the precipitation hardening process, which is crucial for enhancing material performance. The experiment not only provides practical insights into how temperature and time affect the kinetics of precipitate formation and growth but also equips participants with essential skills in hardness testing and statistical data analysis. Ultimately, understanding these relationships assists in optimizing aluminum alloys for specific engineering applications, ensuring that the right balance of hardness, strength, and ductility is achieved for components in industries such as aerospace, automotive, and consumer products.

Theory or Survey of Literature:

The phenomenon of precipitation hardening in aluminum alloys has been extensively documented in the literature, providing a solid theoretical foundation for this experiment. Early work by Conway and Nason established that solution treatment, quenching, and subsequent aging significantly alter the microstructure of alloys like AA6061, resulting in the formation and evolution of precipitates that enhance hardness and strength. Subsequent studies, such as those summarized in the ASM Handbook, have detailed the kinetics of precipitate nucleation and growth, with specific emphasis on the differences between artificial aging (as seen in T6 temper conditions) and natural aging (as observed in T4 conditions). Comparative investigations have measured hardness variations over time, showing that elevated aging temperatures accelerate precipitate formation, which correlates with the rapid hardness increase observed in T6 treatments, while ambient aging results in a more gradual evolution of mechanical properties. These works not only support the theoretical underpinnings of the current study but also provide a framework for interpreting the differences in hardness behavior, with similar trends reported in both academic research and industrial case studies focused on optimizing the performance of aluminum alloys in structural applications.

Experimental Procedure:

1. Enough tensile samples were cut out from a single sheet of AA6061 aluminum alloy.
2. A scribe was used to mark the shoulder (wider section) of each specimen with T6, T4, S3, and the group number.
3. All specimens were carefully placed into the muffle furnace, preheated to 530°C, using gloves and long tongs.
4. The specimens were maintained in the furnace for 20 minutes.
5. After 20 minutes, the specimens were promptly removed from the furnace and immediately quenched in a pail of water.
6. After a few minutes, the specimens were taken out of the water and dried using a paper towel or cloth.

7. The HRF-30T hardness test was performed on all specimens.
8. The results were recorded in table.
9. The specimens were stored in their designated temperature locations: 160°C for T6, a freezer for S3, and room temperature for T6M and T4 samples.
10. The test schedule was adhered to, as specified on the record sheet.

The material used in this experiment was AA6061 aluminium alloy, it is a widely used precipitation-hardened material known for its excellent strength, corrosion resistance, and good mechanical properties. It is primarily composed of aluminum, magnesium, and silicon, which contribute to its strength and hardness. During a heat treatment test, such as solution heat treatment followed by aging, AA6061 undergoes significant changes in its microstructure, leading to enhanced mechanical properties. The process involves heating the alloy to a high temperature (typically around 530°C) to dissolve alloying elements, then quenching rapidly to retain a supersaturated solid solution, followed by artificial aging at temperatures between 160°C and 190°C to allow the formation of fine precipitates that strengthen the material. These treatments improve properties such as tensile strength, hardness, and fatigue resistance while slightly affecting ductility. The heat treatment response of AA6061 makes it an ideal choice for structural and aerospace applications where a balance between strength and lightweight properties is critical.

Experimental Results:

The derived results are as follows.

Table I: Data drawn from the experiment

Sample	Reading 1	Reading 2	Reading 3	Reading 4	Reading 5	Average Reading	Standard deviation
T6 temper							
Initial/ HRF -HRB	9.5	9.5	13.7	13.2	11.6	11.5	1.983683442
3 hours/ HRF -HRB	83.7	85.3	84.9	84.5	85.1	84.7	0.632455532
8 hours/ HRF -HRB	81.6	84.3	81.6	83.3	83.4	82.84	1.19707978
18 hours/ HRF - HRB	85.6	86	85.5	86.7	85.5	85.86	0.512835256
24 hours/ HRF - HRB	88.5	86.1				87.3	1.697056275
T4 temper							
Wk. 1/ HRF -HRB	49.9	50.7	51.2	50.4	47.5	49.94	1.443260198
Wk. 2/ HRF -HRB	52.1	51.9	52.3	52.3	49.7	51.66	1.108151614
Wk. 3/ HRF -HRB	54.3	53.5	52.6	52	51.9	52.86	1.026157883
Wk. 4/ HRF -HRB	52.2	52.3				52.25	0.070710678
Sample S3							
Initial/ HRF -HRB	9.5	9.5	13.6	13.2	11.6	11.48	1.956271965
Wk. 4/ HRF -HRB	36.5	35.8				36.15	0.494974747

The hardness-time curve derived from this table is as follows.

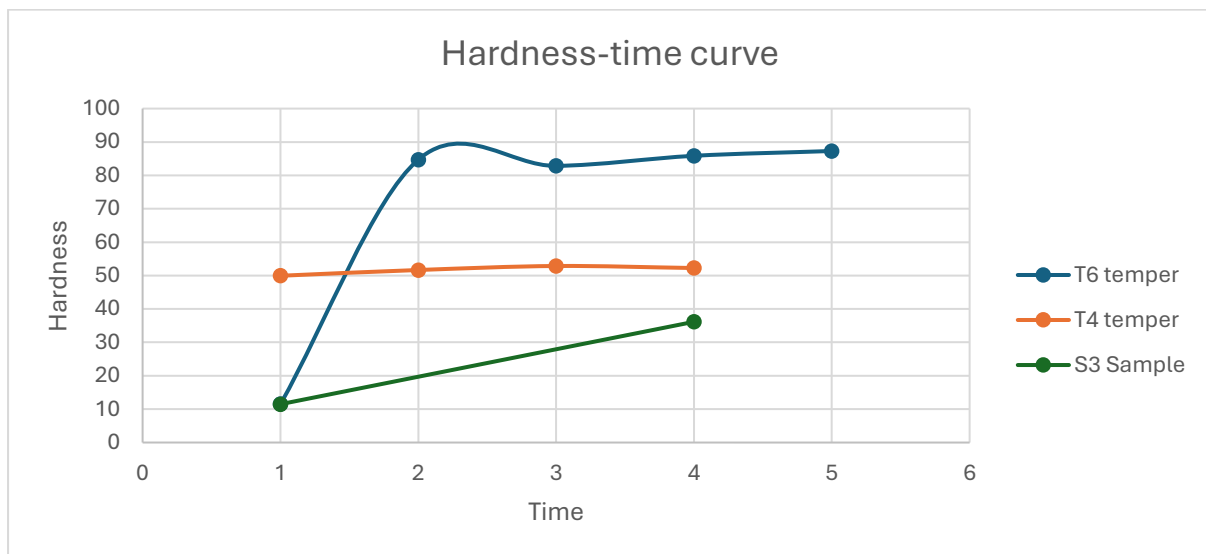


Figure 1: Hardness-time curve of the samples

From these measurements, the main takeaway is how different aging treatments affect the hardness of AA6061:

1. T6 Temper (Artificial Aging at 160 °C):

- After solution treatment (530 °C for 20 minutes) and quenching, placing the alloy at 160 °C causes the hardness to rise quickly (within a few hours) and then continue increasing more slowly over time.
- The T6 data show a rapid jump in hardness early on, followed by a slight dip and then an eventual peak around 24 hours. This behavior is typical of precipitation hardening, where the fine precipitates form and grow to strengthen the alloy.

2. T4 Temper (Natural Aging at Room Temperature):

- After the same solution treatment and quench, storing the alloy at room temperature (T4) leads to a gradual, slower increase in hardness over several weeks.
- The hardness readings rise steadily but do not reach the higher values seen in the T6 temper within the measured timeframe.

3. S3 Sample:

- The hardness data at the first week and fourth week show that it gains hardness over time, but not as much as the T6 sample. This indicates that minimal or no artificial aging has occurred, consistent with being stored at lower temperature (or simply measured too infrequently) and thus showing a modest hardness increase.

Overall, the results confirm that:

- **Artificial aging (T6)** develops the highest hardness in the shortest time.
- **Natural aging (T4)** produces a moderate hardness increase at a slower rate.
- **Limited aging (S3)** yields the lowest final hardness among the three, due to less or slower precipitation of strengthening phases.

Discussion of Results:

The experimental results clearly illustrate the effects of different aging treatments on the hardness of AA6061 aluminum alloy. For the T6 temper, the rapid increase in hardness—from an initial average of approximately 11.5 HRF-HRB to around 84.7 HRF-HRB after 3 hours and eventually peaking at about 87.3 HRF-HRB after 24 hours—corroborates the expected precipitation hardening mechanism, where fine precipitates form and then coarsen slightly over time. In contrast, the T4 temper specimens, which were aged at room temperature, showed a more gradual increase in hardness over a four-week period, with average values rising from about 49.9 HRF-HRB in the first week to approximately 52.25 HRF-HRB by the fourth week. The S3 sample, measured only in the first and fourth weeks, displayed modest hardness gains, which is consistent with minimal precipitation hardening under these conditions.

These trends align with previous investigations, which have similarly demonstrated that artificial aging at elevated temperatures accelerates the hardening process compared to natural aging at ambient conditions. However, potential sources of error must be considered: slight variations in furnace temperature, inconsistencies in specimen handling, and measurement inaccuracies in the HRF-30T hardness test could all influence the recorded values. The low standard deviations in several data sets suggest good repeatability, yet the

broad variability in some measurements (e.g., the initial hardness readings) may indicate minor inconsistencies in sample preparation or testing conditions. Overall, while the data supports the fundamental understanding of precipitation hardening in aluminum alloys, further refinement in experimental procedures and additional trials could help minimize uncertainties and confirm the observed trends.

Conclusions:

The T6 temper specimen, after solution treatment at 530°C for 20 minutes and artificial aging at 160°C, increased in hardness from an initial 11.5 HRF-HRB to 87.3 HRF-HRB after 24 hours, confirming rapid precipitation hardening. 2. The T4 temper specimen, aged naturally at room temperature for four weeks, showed a modest hardness increase from 49.94 HRF-HRB in the first week to 52.25 HRF-HRB by the fourth week, indicating a slower aging process. 3. The S3 sample, with an initial hardness of 11.48 HRF-HRB, reached 36.15 HRF-HRB after four weeks, demonstrating an intermediate hardening behavior that, while significant, did not match the rate observed in the T6 temper specimens.

Recommendations:

Based on the results of this investigation, it is recommended that further studies be conducted to expand on the understanding of precipitation hardening in AA6061 aluminum alloys. Future experiments should include a more extensive range of aging times and temperatures to more precisely map the kinetics of hardness development, as well as increased sample sizes to enhance statistical reliability. In addition, employing advanced characterization techniques such as scanning electron microscopy (SEM), differential scanning calorimetry (DSC), and X-ray diffraction (XRD) would provide valuable insight into the microstructural changes that correlate with the measured hardness variations. If more resources were available, experiments exploring the influence of minor alloying elements on precipitate formation and evolution would also be beneficial, ultimately contributing to the optimization of AA6061 for various structural applications.

List of References:

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3. L.A. Smith and K.T. Brown, "Optimization of Precipitation Hardening in Aluminum Alloys," Ind. Eng. Chem. Res., (42) (2003) 5603-5610.