

# Study of Physics

Toma-Stefan Cezar

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## Abstract

This article was written for the purpose of documenting my process of studying physics.

## 1 Classical mechanics

### 1.1 Introduction

Just like every other "Physics documentation", this document starts with the field of mechanics. In this part we will focus on Classical mechanics. Classical mechanics is the area of physics concerned with the relationships between force, matter and motion among physical objects. This branch of physics has its origins in Ancient Greece, for instance, in the writings of Aristotle and Archimedes. Classical mechanics further breaks down to the sub topics of Dynamics, Kinematics, Continuum, Statistical and Celestial.

### 1.2 General properties and units

These properties and units are important for every branch of classical mechanics. First of all everything around us that we perceive in any kind of way, for instance, a cube out of stone is a physical object (*Ger. Körper*). Every physical object has a fix amount of properties. (The stone cube has a mass, takes up a certain amount of space, etc.)

#### 1.2.1 Volume, mass and density

Volume, mass and density are three important physical quantities.

The Volume is a scalar quantity expressing the amount of three-dimensional spaced enclosed by a closed surface. For example, the space that the stone cube occupies. It can be usually measured using tools (like flow meters, etc.). For expressing the volume we use the symbol  $V$  and the units litre  $l$ , or cubic metre  $m^3$ .

$$1m^3 = 1000l$$

Mass is the quantity of matter in a physical body. Unlike the weight of an object, the mass of a specific object is the same everywhere (The stone cube has a mass of 1kg on the earth, on the mass, etc.). Mass has the symbol  $m$  and the unit kilogram  $kg$ . Mass can be measured using a scale. In an closed system, under the assumption of classical physics, the sum of all masses stays constant (Conservation of mass).

$$m = \sum_{i=1}^n m_i = constant$$

The density of a substance is its mass per unit volume. The symbol most often used for density is  $\rho$  (the lower case Greek letter rho). Mathematically, density is defined as mass divided by volume.

$$\rho = \frac{m}{V}$$

Thus the unit for density is kilogram per cubic metre  $\frac{kg}{m^3}$ . Every substance has a defined constant density.

### 1.2.2 Avogadro constant, mole and dalton

Every object consists of an one or more substances. Every substance consists of particles. These particles may be atoms, molecules or ions. A sample of a substance can be characterized using the number of particles. The number of particles  $N$  expresses the number of particles in the sample of a substance. The amount of substance  $n$  expresses the same thing "*the number of particles in the sample of a substance*" but using the unit mole  $1\text{mol}$ . The Avogadro constant  $N_A$  is the amount of particles in one mole of any substance.

$$N_A = 6.02214076 \times 10^{23} \frac{1}{\text{mol}}$$

One can calculate these values using the following equation.

$$n = \frac{N}{N_A} = \frac{m}{M} = \frac{V}{V_m}$$

$m$  is the mass of the substance sample.  $M$  is the molar mass of the substance. This value is different for every substance.  $V_m$  is the molar volume i.e. the volume that one mole of given substance has and can be calculated with  $V_m = \frac{M}{\rho}$ .

**Example:** How many water molecules are in 10g of water.

$$m = 10\text{g}$$

$$M_{H_2O} = 18\text{g} \times \text{mol}^{-1}$$

$$\frac{N}{N_A} = \frac{m}{M}$$

$$N = N_A \times \frac{m}{M}$$

$$N = \frac{6.022 \times 10^{23} \times 10\text{g} \times \text{mol}}{\text{mol} \times 18} \approx 3.35 \times 10^{23}$$

10 grams of water consist of approximately  $3.35 \times 10^{23}$  molecules.

The unit dalton  $u$  is defined as  $\frac{1}{12}$  of the mass of an unbound neutral atom of carbon-12.

$$1u = \frac{1\text{g}}{N_A} \approx 1,660540 \times 10^{-24}\text{g}$$

### 1.2.3 Cohesion and Adhesion

Cohesion is the action of two like molecules sticking together, being mutually attractive. Adhesion is the tendency of dissimilar particles to stick together.

## 1.3 Kinematics

Kinematics is all about movements and the laws of movements. Kinematics doesn't care about the cause of movement.

### 1.3.1 Frame of reference

The movement of an object is defined as change of position. Change of position can only be measured in relation to an reference object or frame of reference (*Ger. Bezugssystem*), e.g. a coordinate system.

### 1.3.2 Types of movements

There are two types of movements that are important for now. Linear and Nonlinear movements. Linear movements occur with a constant velocity. Nonlinear movements may change velocity over time.

### 1.3.3 Physical quantities for describing movements

The important quantities for kinematics are location, distance, time, velocity and acceleration. The distance  $\vec{s}$  expresses the distance travelled by an object, measured in metre  $m$ . Velocity expresses the rate at which the distance changes. The velocity has the symbol  $\vec{v}$  and the unit metre per second  $\frac{m}{s}$ . Acceleration expresses the rate at which the velocity changes. The acceleration has the symbol  $\vec{a}$  and the unit metre per square second  $\frac{m}{s^2}$ .

