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(54) **DISACCHARIDES USED IN FIRE FIGHTING  
FOAMS**

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**ABSTRACT**

An aqueous fire-fighting foam concentrate includes a disaccharide sugar; a surfactant component selected from an anionic surfactant and a zwitterionic surfactant; a water-miscible organic solvent; at least about 30 wt % water; and optionally includes a polysaccharide thickener and/or a nonionic surfactant. The fire-fighting foam concentrate does not include a monosaccharide sugar and does not include a eutectic solvent.

## DISACCHARIDES USED IN FIRE FIGHTING FOAMS

### CROSS REFERENCE TO RELATED APPLICATIONS

**[0001]** This application claims the benefit of priority to U.S. Provisional Patent Application No. 63/484,829 filed Feb. 14, 2023, which is hereby incorporated by reference, in its entirety for any and all purposes.

### BACKGROUND

**[0002]** Firefighting foams are often able to fight Class A and Class B fires. Class A fires are those involving combustible material such as paper, wood, etc. and can be fought by quenching and cooling with large quantities of water or solutions containing water. Class B fires are those involving flammable liquid fuels, gasoline, and other hydrocarbons and are difficult to extinguish. Most flammable liquids exhibit high vapor pressure along with low fire and flash points. This typically results in a wide flammability range. In this type of fire, the use of water as the sole firefighting agent is generally ineffective because the only means of fighting fire with water is through cooling.

**[0003]** Conventional foam-forming firefighting compositions commonly include fluorinated surfactants. There is a strong desire in the marketplace to replace these fluorinated firefighting products with non-fluorinated products. There is therefore a continuing need to produce non-fluorinated firefighting compositions, also known as synthetic fluorine-free foams or SFFF that can be deployed to fight Class A and Class B fires.

### SUMMARY

**[0004]** The present application is directed to aqueous concentrates, which can be diluted with an aqueous diluent to provide a foam precursor composition, which may be aerated to form a firefighting foam. The present aqueous firefighting concentrates include a sugar component including a disaccharide, and a surfactant component containing one or more of an anionic surfactant, a zwitterionic surfactant, and optionally a nonionic surfactant. The aqueous firefighting concentrates may also include an organic solvent, e.g., a water-miscible organic solvent such as an alkylene glycol, glycerol, a water-soluble polyethylene glycol, and/or a glycol ether. The aqueous firefighting concentrates may further include a polysaccharide thickener. The concentrate does not comprise monosaccharide sugars and does not comprise a eutectic solvent. The concentrate may be substantially free of any fluorinated compounds, e.g., contain no more than 70 parts per trillion (ppt) fluorinated surfactant(s) and, often, is completely free of any fluorinated surfactant or other fluorinated compounds.

**[0005]** In one aspect, an aqueous fire-fighting foam concentrate is provided, which comprises a sugar component comprising a disaccharide sugar; a surfactant component comprising an anionic surfactant, a zwitterionic surfactant, or a mixture of any two or more thereof; a water-miscible organic solvent; and at least about 30 wt % water. The aqueous fire-fighting foam concentrate does not comprise a monosaccharide sugar and does not comprise a eutectic solvent. In some embodiments, the aqueous fire-fighting foam concentrate further comprises a polysaccharide thickener and/or a nonionic surfactant.

**[0006]** In some embodiments, the aqueous fire-fighting foam concentrate comprises about 5 to 25 wt % of the sugar component; about 2 to 20 wt % of the anionic surfactant; about 1 to 10 wt % of the zwitterionic surfactant; about 1 to 20 wt % of the water-miscible organic solvent; and at least about 30 wt % water.

**[0007]** In some embodiments, the disaccharide sugar comprises sucrose, maltose, lactose, lactulose, trehalose, cellobiose, chitobiose, or a combination of any two or more thereof.

**[0008]** In some embodiments, the anionic surfactant comprises an alkyl sulfate salt, an alkyl sulfonate salt, an alkyl ether sulfate surfactant, an alkyl ether sulfonate surfactant, or a mixture of any two or more thereof. In one embodiment, the anionic surfactant comprises C<sub>8</sub>-C<sub>14</sub> alkyl sulfate, C<sub>8</sub>-C<sub>14</sub> alkyl sulfonate, C<sub>10</sub>-C<sub>14</sub> alkyl ether sulfate, or a mixture of any two or more thereof. In another embodiment, the anionic surfactant comprises a decyl sulfate salt, a lauryl sulfate salt, or a mixture of any two or more thereof.

**[0009]** In some embodiments, the zwitterionic surfactant comprises an alkylamidoalkyl hydroxysultaine, an alkylamidoalkyl betaine, an alkyl sulfobetaine, an alkyl betaine, or a mixture of any two or more thereof. In an embodiment, the zwitterionic surfactant comprises an alkylamidopropyl hydroxysultaine, cocamidopropyl hydroxysultaine, laurylamidopropyl hydroxysultaine, or myristylamidopropyl hydroxysultaine.

**[0010]** In some embodiments, the water-miscible organic solvent comprises an alkylene glycol, a glycerol, a water-soluble polyethylene glycol, a glycol ether, or a mixture of any two or more thereof. In an embodiment, the organic solvent comprises propylene glycol and diethylene glycol monobutyl ether.

**[0011]** In some embodiments, the surfactant component further comprises a nonionic surfactant. In some embodiments, the nonionic surfactant comprises an alkyl polyglycoside, an aliphatic alcohol-based nonionic surfactant, or a mixture of any two or more thereof. In some embodiments, the nonionic surfactant comprises an alkyl polyglucoside, an aliphatic alcohol, an aliphatic alcohol ethoxylate, or a mixture of any two or more thereof. In an embodiment, the aqueous fire-fighting foam concentrate further comprises about 0.2 to 5 wt % of a nonionic surfactant.

**[0012]** In some embodiments, the aqueous fire-fighting foam concentrate further comprises a polysaccharide thickener. In some embodiments, the polysaccharide thickener comprises agar, sodium alginate, carrageenan, gum arabic, gum guaicum, neem gum, pistacia lentiscus, gum chatti, caranna, galactomannan, gum tragacanth, karaya gum, guar gum, welan gum, rhamnsam gum, locust bean gum, beta-glucan, cellulose, methylcellulose, chicle gum, kino gum, dammar gum, glucomannan, succinoglycan, mastic gum, spruce gum, tara gum, gellan gum, xanthan gum, acacia gum, cassia gum, diutan gum, fenugreek gum, ghatti gum, hydroxyethylcellulose, hydroxypropylmethylcellulose, karaya gum, konjac gum, pectin, propylene glycol alginate, or a mixture of any two or more thereof. In some embodiments, any one of the aqueous fire-fighting foam concentrates described herein further comprises about 0.1 to 5 wt % of a polysaccharide thickener.

**[0013]** In some embodiments, the aqueous fire-fighting foam concentrate further comprising a reducing agent. In

some embodiments, the reducing agent comprises sodium sulfite, sodium metabisulfite, sodium bisulfite, or a mixture of any two or more thereof.

**[0014]** In some embodiments, the aqueous fire-fighting foam concentrate contains no more than 0.01 wt % fluorinated surfactants.

**[0015]** In an aspect, provided is a method of forming a firefighting foam, the method comprising: mixing any one of the firefighting foam concentrates described herein with an aqueous diluent to form a foam precursor solution; and aerating the foam precursor solution to form the firefighting foam. In some embodiments, the aqueous diluent is selected from the group consisting of fresh water, brackish water, sea water, and a combination of any two or more thereof. In some embodiments, the aqueous diluent comprises water from a municipal water source.

**[0016]** In another aspect, a firefighting foam is provided, the firefighting foam comprising any one of the firefighting foam concentrates described herein and an aqueous diluent. In an aspect, a method of fighting a fire is provided, the method comprising applying the firefighting foam to the fire.

#### DETAILED DESCRIPTION

**[0017]** Various embodiments are described hereinafter. It should be noted that the specific embodiments are not intended as an exhaustive description or as a limitation to the broader aspects discussed herein. One aspect described in conjunction with a particular embodiment is not necessarily limited to that embodiment and can be practiced with any other embodiment(s).

**[0018]** As utilized herein with respect to numerical ranges, the terms “approximately,” “about,” “substantially,” and similar terms will be understood by persons of ordinary skill in the art and will vary to some extent depending upon the context in which it is used. If there are uses of the terms that are not clear to persons of ordinary skill in the art, given the context in which it is used, the terms will be plus or minus 10% of the disclosed values. When “approximately,” “about,” “substantially,” and similar terms are applied to a structural feature (e.g., to describe its shape, size, orientation, direction, etc.), these terms are meant to cover minor variations in structure that may result from, for example, the manufacturing or assembly process and are intended to have a broad meaning in harmony with the common and accepted usage by those of ordinary skill in the art to which the subject matter of this disclosure pertains. Accordingly, these terms should be interpreted as indicating that insubstantial or inconsequential modifications or alterations of the subject matter described and claimed are considered to be within the scope of the disclosure as recited in the appended claims.

**[0019]** The use of the terms “a” and “an” and “the” and similar referents in the context of describing the elements (especially in the context of the following claims) are to be construed to cover both the singular and the plural, unless otherwise indicated herein or clearly contradicted by context. Recitation of ranges of values herein are merely intended to serve as a shorthand method of referring individually to each separate value falling within the range, unless otherwise indicated herein, and each separate value is incorporated into the specification as if it were individually recited herein. All methods described herein can be performed in any suitable order unless otherwise indicated herein or otherwise clearly contradicted by context. The use of any and all examples, or exemplary language (e.g., “such

as”) provided herein, is intended merely to better illuminate the embodiments and does not pose a limitation on the scope of the claims unless otherwise stated. No language in the specification should be construed as indicating any non-claimed element as essential.

**[0020]** As used herein, the term “eutectic solvent” or “deep eutectic solvent” refers to a mixture of two or more compounds, which demonstrates a melting point that is lower than either of the compounds alone. For example, a eutectic mixture of two compounds A and B would have a melting point that is lower than compound A or B alone and is known as a binary eutectic mixture. Similarly, a eutectic mixture of three compounds A, B, and C would have a melting point that is lower than compound A, B, or C alone and is known as a ternary eutectic mixture. See, for example, Liu, Y.-T. et al., *Synthesis And Characterization of Novel Ternary Deep Eutectic Solvents*. *Chin. Chem. Lett.* 2014, 25, 104-106. The point in a phase diagram, where the chemical composition and temperature correspond to the lowest melting point of a mixture of components is the eutectic point of the mixture. Generally, eutectic solvents having a freezing point depression greater than 150° C. are referred to as “deep eutectic solvents.”

**[0021]** Provided herein are aqueous firefighting foam concentrates comprising a sugar component which includes a disaccharide sugar. Other fire-fighting foam concentrates include only monosaccharide sugars or have found only carbonized (caramelized) sugars to work. However, the present aqueous firefighting foam concentrates prove sugars in raw forms (e.g., sucrose, which is table sugar) can provide performance in firefighting foams.

**[0022]** In one aspect, the aqueous firefighting foam concentrates of the present disclosure include a sugar component, which includes a disaccharide; a surfactant component, which comprises an anionic surfactant and/or a zwitterionic surfactant; a water-miscible organic solvent; and at least about 30 wt % water. The present fire-fighting foam concentrates do not comprise monosaccharide sugars and do not comprise a eutectic solvent. In some embodiments, the firefighting foam concentrate further comprises a polysaccharide thickener and/or a nonionic surfactant. In some embodiments, the firefighting foam concentrate is free of a nonionic surfactant.

**[0023]** Saccharides are generally disaccharide sugars, such as common sugar (sucrose/dextrose) derived from sugar cane or sugar beets. Sucrose is a disaccharide composed from the basic, simple sugar molecules glucose and fructose. Sucrose is readily available in view of its world production from cane and sugar beet on the order of millions of tons per annum. Those skilled in the art will also be aware that other commercially available disaccharides can be utilized in the present foam concentrates. Examples of suitable disaccharides for use in the present foam concentrates include sucrose, maltose, lactose, lactulose, trehalose, cellobiose, and chitobiose. Disaccharides in raw forms (e.g., non-carbonized) can be utilized in the present foam concentrates.

**[0024]** The sugar component may include one or more disaccharide sugars. Suitable examples include a sugar component containing one or more of sucrose, maltose, lactose, lactulose, trehalose, cellobiose, and chitobiose. The aqueous fire-fighting foam concentrate includes about 5 to 25 wt % of the sugar component. In other embodiments, the aqueous fire-fighting foam concentrate includes about 10 to

20 wt % of the sugar component. In some embodiments, the aqueous fire-fighting foam concentrate comprises about 5 wt. %, about 10 wt. %, about 15 wt. %, about 20 wt. %, or about 25 wt. % of the sugar component.

**[0025]** Illustrative anionic surfactants include an alkyl sulfate surfactant, an alkyl sulfonate surfactant, an alkyl ether sulfate surfactant, an alkyl ether sulfonate surfactant, or a mixture of any two or more thereof. In some embodiments, the anionic surfactant may include an alkyl sulfate surfactant and/or an alkyl sulfonate surfactant. In some embodiments, the alkyl sulfate salt surfactant includes a  $C_{8-12}$ -alkyl sulfate salt. Non-limiting examples of the  $C_{8-12}$ -alkyl sulfate salt include a dodecyl sulfate salt, a decyl sulfate salt, an octyl sulfate salt, or a mixture of any two or more thereof. In some embodiments, the alkyl sulfate salt includes an alkyl sulfate sodium salt, such as a sodium decyl sulfate, sodium octyl sulfate, or a mixture of any two or more thereof. In some embodiments, the alkyl sulfate salt includes an alkyl sulfate ammonium salt, such as an ammonium decyl sulfate, ammonium octyl sulfate, ammonium lauryl sulfate, or a mixture of any two or more thereof. In some embodiments, the anionic surfactant includes a  $C_{8-14}$ -alkyl sulfate salt and/or a  $C_{8-14}$ -alkyl sulfonate salt. In some embodiments, the aqueous fire-fighting foam concentrates include an anionic surfactant, which comprises one or more surfactants selected from  $C_{8-12}$ -alkyl sulfate salts and/or a  $C_{8-12}$ -alkyl sulfonate salts. As a non-limiting example, one or more of octyl sulfate salts, decyl sulfate salts, dodecyl sulfate salts and tetradecyl sulfate salts may be used as anionic surfactants in the present aqueous fire-fighting foam concentrates. The anionic surfactant may be a sodium, potassium and/or ammonium salt. In some embodiments, the aqueous fire-fighting foam concentrates may include about 1 to 25 wt % or about 1 to 20 wt % of the anionic surfactant. In some embodiments, the aqueous firefighting foam concentrates include about 2 to 15 wt %, about 2 to 10 wt % and, in some instances, about 3 to 10 wt % of the anionic surfactant.

**[0026]** In some embodiments, the present aqueous fire-fighting foam concentrates include a zwitterionic surfactant. Illustrative zwitterionic surfactants include an alkylamido-alkyl betaine surfactant, an alkyl betaine surfactant, an alkyl sulfobetaine surfactant, an alkylamidoalkylene hydroxysultaine surfactant, such as an alkylamidopropyl hydroxysultaine surfactant, or a mixture of two or more thereof. As a non-limiting example, the aqueous fire-fighting foam concentrates include a zwitterionic surfactant, which comprises one or more of a  $C_{8-18}$ -alkylamidopropyl hydroxysultaine surfactant, a  $C_{8-18}$ -alkylamidopropyl betaine surfactant, a  $C_{8-18}$ -alkyl sulfobetaine surfactant, a  $C_{8-18}$ -alkyl betaine surfactant, or a mixture of two or more thereof. Non-limiting examples of the alkylamidoalkylene hydroxysultaine surfactant include a  $C_{8-18}$ -alkylamidopropyl hydroxysultaine surfactant, such as a cocamidopropyl hydroxysultaine surfactant, which includes a laurylamidopropyl hydroxysultaine and a myristylamidopropyl hydroxysultaine. Non-limiting examples of the alkylamidoalkyl betaine surfactant include a  $C_{8-18}$ -alkylamidoalkyl betaine surfactant, such as a cocamidopropyl betaine, a tallowamidopropyl betaine, a laurylamidopropyl betaine or a myristylamidopropyl betaine. In some embodiments, the zwitterionic surfactant includes a  $C_{8-14}$ -alkylamidopropyl hydroxysultaine, such as a cocamidopropyl hydroxysultaine. In some embodiments, the zwitterionic surfactant includes laurylamidopropyl

hydroxysultaine and/or myristylamidopropyl hydroxysultaine. In some embodiments, the aqueous fire-fighting foam concentrates include about 2 to 15 wt. % of the zwitterionic surfactant. In some embodiments, the aqueous fire-fighting foam concentrates include about 1 to 15 wt. %, about 5 to 15 wt. %, or about 1 to 10 wt. % of the zwitterionic surfactant.

**[0027]** The water miscible organic solvent may suitably include one or more of a glycol, a glycol ether, glycerin (i.e., glycerol) and a water-soluble polyethylene glycol. Examples of suitable water miscible organic solvents include diethylene glycol n-butyl ether, dipropylene glycol n-propyl ether, hexylene glycol, ethylene glycol, propylene glycol, diethylene glycol, dipropylene glycol, tripropylene glycol, dipropylene glycol monobutyl ether, dipropylene glycol monomethyl ether, diethylene glycol monobutyl ether, ethylene glycol monobutyl ether, tripropylene glycol monomethyl ether, dipropylene glycol monoethyl ether, glycerol, and mixtures of two or more thereof. In some embodiments, the water miscible organic solvent includes a mixture of an alkylene glycol and a glycol ether, such as a glycol butyl ether. In some embodiments, the water miscible organic solvent includes an alkylene glycol ether, such as ethylene glycol monoalkyl ether, propylene glycol monoalkyl ether, dipropylene glycol monoalkyl ether, and/or diethylene glycol monoalkyl ether. In some embodiments, the water miscible organic solvent includes an alkylene glycol, such as ethylene glycol, propylene glycol, dipropylene glycol and/or diethylene glycol. In some embodiments, the water miscible organic solvent may include a mixture of a glycol ether, such as diethylene glycol monobutyl ether, and a glycol, such as ethylene glycol and/or propylene glycol. For example, the water miscible organic solvent can include ethylene glycol and diethylene glycol monobutyl ether. In another suitable example, the water miscible organic solvent includes propylene glycol and diethylene glycol monobutyl ether.

**[0028]** The aqueous fire-fighting foam concentrates may include about 0.5 to about 30 wt. % of the water miscible organic solvent. This may include from about 1 to about 30 wt. %, about 1 to about 15 wt. %, or about 2 to about 25 wt. % of the water miscible organic solvent. In many embodiments, the aqueous firefighting foam concentrate includes a water miscible organic solvent including one or more of an alkylene glycol, glycerol, and a glycol ether. The alkylene glycol typically includes propylene glycol and/or ethylene glycol. The glycol ether typically includes ethylene glycol monoalkyl ether, diethylene glycol monoalkyl ether, dipropylene glycol monoalkyl ether, triethylene glycol monoalkyl ether, and 1-butoxyethoxy-2-propanol. In some embodiments, the water miscible organic solvent may be a mixture of an alkylene glycol and a glycol ether. In some embodiments, the water miscible organic solvent may be a mixture of glycerin and a glycol ether. In some embodiments, the water miscible organic solvent may be a mixture of propylene glycol and a glycol ether. In such embodiments, the water miscible organic solvent includes the alkylene glycol and glycol ether in a weight ratio of about 0.1:1 to about 10:1 or about 0.2:1 to about 5:1. In some embodiments, the water miscible organic solvent may be a mixture of glycerin and diethylene glycol monobutyl ether. In some embodiments, the water miscible organic solvent may include about 1 to about 20 wt. % or about 1 to about 15 wt. % of alkylene glycol, such as ethylene glycol, or glycerin together with about 1 to about 15 wt. % or about 1 to about 10 wt. % of a glycol ether, such as diethylene glycol monobutyl ether.

**[0029]** In some embodiments, the aqueous firefighting foam concentrate further includes a thickener, such as a polysaccharide thickener. The polysaccharide thickener may include a polysaccharide that is soluble in the aqueous firefighting foam concentrate and a second polysaccharide that is less soluble or insoluble in the aqueous firefighting foam concentrate. In some embodiments, the second polysaccharide may be at least partially insoluble (and dispersed) in the aqueous firefighting concentrate but may be soluble in water alone or in solutions where the concentrate has been diluted with a much larger volume of water. In other embodiments, the concentrate may only include one or more polysaccharides that are completely soluble in the concentrate. The foam concentrate may include about 0.1 to 5 wt %, about 0.2 to 3 wt %, about 0.5 to 3 wt % and, more commonly, about 0.5 to 2 wt % of the polysaccharide thickener.

**[0030]** Illustrative polysaccharide thickeners include agar, sodium alginate, carrageenan, gum arabic, gum guaiacum, neem gum, pistacia lentiscus, gum chatti, caranna, galactomannan, gum tragacanth, karaya gum, guar gum, welan gum, rhamnsam gum, locust bean gum, beta-glucan, cellulose, methylcellulose, chicle gum, kino gum, dammar gum, glucomannan, succinoglycan, mastic gum, spruce gum, tara gum, gellan gum, acacia gum, cassia gum, diutan gum, fenugreek gum, ghatti gum, hydroxyethylcellulose, hydroxypropylmethylcellulose, karaya gum, konjac gum, pectin, propylene glycol alginate, and a mixture of two or more thereof.

**[0031]** In some embodiments, the polysaccharide thickener may include one or more of xanthan gum, diutan gum, rhamnsam gum, welan gum, gellan gum, guar gum, konjac gum, tara gum, succinoglycan, and methylcellulose. In some embodiments, it may include a mixture of xanthan gum and one or more of diutan gum, rhamnsam gum, welan gum, gellan gum, guar gum, konjac gum, tara gum, succinoglycan, and methylcellulose. In other embodiments, the foam concentrate may include a mixture of xanthan gum and one or more of diutan gum, rhamnsam gum, welan gum, succinoglycan, and gellan gum as the polysaccharide thickener. In other embodiments, the foam concentrate may include a mixture of xanthan gum and diutan gum and/or rhamnsam gum. In other embodiments, the foam concentrate may include a mixture of xanthan gum and succinoglycan. In other embodiments, the foam concentrate may include a mixture of xanthan gum and welan gum. In other embodiments, the foam concentrate may include welan gum.

**[0032]** Polysaccharide thickeners, which include a combination of xanthan gum and diutan gum, are suitable for use in the present foam concentrates, according to some embodiments. The foam concentrate may include about 0.2 to 3 wt % of xanthan gum and about 0.1 to 2 wt % diutan gum. This may include from about 0.3 to 2 wt %, about 0.5 to 1.5 wt %, or about 0.5 to 1 wt % xanthan gum, and about 0.2 to 1.5 wt % or about 0.2 to 1 wt % diutan gum.

**[0033]** In other embodiments, polysaccharide thickeners that include a combination of xanthan gum and welan gum are suitable for use in the present foam concentrates. For example, the foam concentrate may include about 0.2 to 3 wt % of xanthan gum and about 0.1 to 5 wt % welan gum. This includes about 0.3 to 2 wt %, about 0.5 to 1.5 wt %, or about 0.5 to 1 wt % xanthan gum, and about 0.2 to 4 wt % or about 0.5 to 3 wt % welan gum.

**[0034]** In other embodiments, polysaccharide thickeners that include a combination of xanthan gum and succinoglycan are suitable for use in the present foam concentrates. For example, the foam concentrate may include about 0.2 to 3 wt % of xanthan gum and about 0.1 to 5 wt % of succinoglycan. This includes about 0.3 to 2 wt %, about 0.5 to 1.5 wt %, or about 0.5 to 1 wt % of xanthan gum, and about 0.2 to 4 wt % or about 0.5 to 3 wt % of succinoglycan.

**[0035]** When present in the aqueous fire-fighting foam concentrate, the non-ionic surfactants include an alkyl polyglycoside, an aliphatic alcohol-based nonionic surfactant, or a mixture thereof. In some embodiments, the non-ionic surfactant comprises an alkyl polyglucoside, an aliphatic alcohol, an aliphatic alcohol ethoxylate, or a mixture of two or more thereof. A suitable alkyl polyglucoside for use as a non-ionic surfactant includes a C<sub>8-16</sub> alkyl polyglucoside. In some embodiments, the alkyl polyglucoside is a C<sub>8</sub>-alkyl polyglucoside. As a non-limiting example, the aqueous firefighting foam concentrate may include an aliphatic alcohol-based nonionic surfactant including an aliphatic alcohol having 8 to 14 carbon atoms and/or an aliphatic alcohol ethoxylate having 10 to 16 carbon atoms in its alcohol portion. The aliphatic alcohol ethoxylate may have an average degree of polymerization (i.e., the average number of ethylene oxide units) of about 0.7-2.0 and often of no more than about 1.5, no more than about 1.2, or no more than about 1.0. Aliphatic alcohols, which include a linear C<sub>8-14</sub>-aliphatic alcohol, such as a C<sub>8-14</sub>-fatty alcohol, may be used as a nonionic surfactant in the present firefighting concentrates. Non-limiting examples of such alcohols include one or more of octyl alcohol, decyl alcohol, lauryl alcohol and myristyl alcohol. The aqueous firefighting foam concentrate may include an aliphatic alcohol ethoxylate having an average of no more than about 2 ethylene oxide units. In some embodiments, the aliphatic alcohol portion of such ethoxylates has about 10 to 16 carbon atoms. Non-limiting examples include decyl alcohol ethoxylates, lauryl alcohol ethoxylates and/or myristyl alcohol ethoxylates. In some embodiments, the alcohol ethoxylates have an average of no more than about 2 ethylene oxide units, no more than about 1.5 ethylene oxide units, no more than about 1.2 ethylene oxide units and, in some instances, no more than about 1 ethylene oxide units. In some embodiments, the aliphatic alcohol ethoxylate comprises an ethoxylate of a linear C<sub>8-14</sub>-aliphatic alcohol having no more than about 1.2 ethylene oxide units.

**[0036]** In embodiments that include the nonionic surfactant, the aqueous fire-fighting foam concentrate includes about 1 to 15 wt %, about 1 to 10 wt %, about 3 to 10 wt %, or about 5 to 10 wt % of the non-ionic surfactant. In some embodiments, the aqueous fire-fighting foam concentrates include about 1 wt %, about 3 wt %, about 5 wt %, about 10 wt %, or about 15 wt % of the non-ionic surfactant.

**[0037]** As discussed above, the aqueous firefighting foam concentrate includes water. In some embodiments, the water is water from a municipal water source (e.g., tap water). In some embodiments, the water is a purified water, such as purified water that meets the standards set forth in the United States Pharmacopeia, which is incorporated by reference herein in relevant part. In some embodiments, the aqueous firefighting foam concentrate includes at least about 30 wt % water, often at least about 40 wt % water and, more commonly, at least about 50 wt % water. In some embodiments, the aqueous firefighting foam concentrate includes

greater than about 60 wt % water. In some embodiments, the aqueous firefighting foam concentrate may be produced using a source of water that has a total concentration of fluorine atoms on a weight percentage basis of no more than about 1 ppm F.

**[0038]** The present aqueous firefighting foam concentrates may be substantially free of any fluorinated compounds. As used herein, the “phrase substantially free of fluorinated compounds” means that the aqueous firefighting foam concentrate includes no more than 0.1 wt % fluorinated compounds. In some embodiments, the aqueous firefighting foam concentrate includes no more than 0.01 wt % and, in some instances, no more than about 0.005 wt % fluorinated compounds. The aqueous firefighting foam concentrate of the present disclosure may be substantially free of fluorine in any form. As used herein, the phrase “substantially free of fluorine” means that the aqueous firefighting foam concentrate has a total concentration of fluorine atoms on a weight percentage basis of no more than about 1 ppm F. In some embodiments, the aqueous firefighting foam concentrate has a total concentration of fluorine atoms of less than 0.01 wt. %. In some embodiments, the aqueous firefighting foam concentrate has a total concentration of fluorine atoms of less than 0.005 wt. %. In some embodiments, the aqueous firefighting foam concentrate has a total concentration of fluorine atoms on a weight percentage basis of no more than about 70 parts per trillion (ppt) F. The aqueous firefighting foam concentrate of the present disclosure may include substantially less than 70 ppt F. In some embodiments, the total concentration of fluorine in the aqueous firefighting foam concentrate is below detection limits.

**[0039]** In any of the above embodiments, the aqueous firefighting foam concentrate includes one or more chelators or sequestering buffers. Illustrative and non-limiting chelators and sequestering buffers include agents that sequester and chelate metal ions, including polyamminopolycarboxylic acids, ethylenediaminetetraacetic acid, citric acid, tartaric acid, nitrilotriacetic acid, hydroxyethylethylenediaminetriacetic acid and salts thereof. Illustrative buffers include Sorensen’s phosphate or Mellvaine’s citrate buffers.

**[0040]** In any of the above embodiments, the aqueous firefighting foam concentrate includes one or more corrosion inhibitors. Illustrative and non-limiting corrosion inhibitors includes ortho-phenylphenol, tolyltriazole, and phosphate ester acids. In some embodiments, the corrosion inhibitor is tolyltriazole.

**[0041]** In any of the above embodiments, the aqueous firefighting foam concentrate may also include a metallic salt, typically a metallic salt, which includes a multi-valent cation. For example, suitable salts may include a cation selected from the group consisting of aluminum, calcium, copper, iron, magnesium, potassium, and calcium cations. The counter anion may suitably be a sulfate, bicarbonate, and/or phosphate anion. In one embodiment, the metallic salt may include magnesium sulfate. In some embodiments, the metallic salt includes potassium bicarbonate.

**[0042]** In any of the above embodiments, the aqueous firefighting foam concentrate may include a reducing agent. Accordingly, an aqueous fire-fighting foam concentrate may include a sugar component including a disaccharide sugar, and a reducing agent. The aqueous fire-fighting foam concentrate may further include a surfactant component including an anionic surfactant and a zwitterionic surfactant, a water-miscible organic solvent, and at least about 30 wt %

water. The aqueous firefighting foam concentrate may further include a polysaccharide thickener and/or an aliphatic alcohol-based nonionic surfactant. The aqueous fire-fighting foam concentrate does not contain monosaccharide sugars. The aqueous firefighting foam concentrate does not comprise a eutectic solvent. When present, the reducing agent may be present in the aqueous firefighting foam concentrate from about 0.01 wt % to about 5 wt %. This may include from about 0.01 wt % to about 3 wt %, from about 0.05 wt % to about 5 wt %, from about 1 wt % to about 5 wt %, or from about 1 wt % to about 3 wt %.

**[0043]** The reducing agent may be selected such that it is more readily oxidized compared to other components of the foam. For example, the reducing agent may be oxidized more readily than the sugar component or polysaccharide components. Illustrative reducing agents include, but are not limited to, sodium sulfite, sodium bisulfite, sodium metabisulfite, or a mixture of any two or more thereof.

**[0044]** In any of the above embodiments, the aqueous firefighting foam concentrate may also include a preservative, such as one or more antimicrobial compounds and/or biocidal compounds. These components are included to prevent the biological decomposition of natural product based polymers that are incorporated as polymeric film formers (e.g., a polysaccharide gum). Examples include Kathon CG/ICP (Rohm & Haas Company), Givgard G-4 40 (Givaudan, Inc.), Dowicil 75, and Dowacide A (Dow Chemical Company).

**[0045]** In any of the above embodiments, the fire-fighting foam composition may include an alkanolamine, such as monoethanolamine, diethanolamine, diisopropanolamine, and/or triethanolamine. In some embodiments, where the fire-fighting foam composition includes an alkanolamine, it may include triethanolamine. In some embodiments, the composition includes about 1 to 10 wt. % of the alkanolamine.

**[0046]** In any of the above embodiments, the aqueous firefighting foam concentrate may further include microfibrillar cellulose. The microfibrillar cellulose included in the aqueous firefighting foam concentrates may include microfibrillar cellulose produced by mechanically disrupting/altering cellulose fibers, e.g., cereal, wood, and/or cotton-based cellulose fibers—commonly referred to as microfibrillated cellulose (MFC). Microfibrillated cellulose can be obtained through a fibrillation process of cellulose fibers. In such a process, the mechanical shearing can strip away the outer layer of the cellulose fibers, exposing the fibril bundles. The macroscopic fibers are typically mechanically sheared until the fibrils are released, resulting in separation of the cellulose fibers into a three dimensional network of microfibrils with a very large surface area. The exposed fibrils are much smaller in diameter compared to the original fibers and can form a network or a web-like structure.

**[0047]** An illustrative and non-limiting example of microfibrillated cellulose is Exilva™ microfibrillated cellulose (available from Borregaard, Sarpsborg, Norway). Exilva™ microfibrillated cellulose is a pre-activated product, available as a 2% suspension or a 10% paste, that is produced from mechanically disrupting cellulose sourced from Norway spruce. Exilva™ microfibrillated cellulose is reported to be an insoluble microfibrillated cellulose consisting of an entanglement of the cellulose fibers, which has the ability to interact both physically through its extreme surface area and chemically through hydrogen bonding. Other commercial

sources of microfibrillar cellulose include Celova® microfibrillated cellulose (available from Weidmann Electrical Technology AG (Rapperswil, Switzerland) and Curran® microfibrillated cellulose (available from CelluComp, Fife, Scotland). Curran® microfibrillated cellulose is produced from extraction of nanocellulose fibers from waste streams of root vegetables, primarily carrots and sugar beet pulp.

**[0048]** Another illustrative and non-limiting example of a source of microfibrillated cellulose for use in the present aqueous firefighting foam concentrates is microfibrillated cellulose-mineral composite commercially available from FiberLean® Technologies (Par Moor Centre, United Kingdom). The FiberLean® MFC-composite is reportedly produced by fibrillating the cellulose fibers in the presence of one of a number of different minerals, such as calcium carbonate, clay (e.g., kaolin or bentonite), alumina, zirconia, graphite, silicate or talc, to obtain a nano-fibrillar cellulose suspension.

**[0049]** When included in the aqueous firefighting foam concentrate, the microfibrillar cellulose may include a fermentation-derived cellulose, such as a microfibrillar cellulose derived from a microbial fermentation process. The microfibrillar cellulose may include cellulose derived from a bacterial fermentation process, e.g., from fermentation of a *Komagataeibacter xylinus* strain or an *Acetobacter xylinum* strain. Fermentation-derived cellulose (FDC) produced by such a method may have an average fiber diameter of about 0.1-0.2 µm. This very small fiber size and diameter means that a given weight of FDC can have up to 200 times more surface area than other common forms of cellulose.

**[0050]** The microfibrillar cellulose employed in the present aqueous firefighting foam concentrates may have an average fiber diameter of no more than about 10 µm, no more than about 1 µm, and in some instances about 50 to 300 nm (0.05-0.3 µm). Quite often, the microfibrillar cellulose is a derived from microbial fermentation. Prior to inclusion in the aqueous firefighting foam concentrates, such microbial fermentation derived cellulose may be activated by combining a powdered microfibrillar cellulose and any optional co-agent with water and then mixing with high shear.

**[0051]** In an embodiment, the aqueous firefighting foam concentrate includes about 5 to 25 wt % of a sugar component, which comprises a disaccharide sugar; about 2 to 20 wt % anionic surfactant, which comprises one or more of an aliphatic sulfate salt, aliphatic sulfonate salt, an aliphatic ether sulfate salt, and aliphatic ether sulfate salt; about 1 to 10 wt % zwitterionic surfactant, which comprises one or more of an alkylamidoalkyl betaine, an alkyl sulfobetaine, an alkylamidoalkyl hydroxysultaine and an alkyl hydroxysultaine; about 1 to 20 wt % organic solvent, which comprises one or more of a glycol, glycol ether, glycerol and/or water-soluble polyethylene glycol (PEG); and at least about 30 wt % water. The aqueous firefighting foam concentrate may further include about 0.2 to 5 wt % of a non-ionic surfactant; and/or about 0.1 to 5 wt % polysaccharide thickener. The concentrate does not comprise monosaccharide sugars. The aqueous firefighting foam concentrate does not comprise a eutectic solvent. The concentrate comprises no more than 0.01 wt % of a fluorinated compound.

**[0052]** In another embodiment, the aqueous firefighting foam concentrate includes at least about 15 wt % of a sugar component, which comprises a disaccharide sugar; about 2-20 wt % of a surfactant mixture comprising one or more

of an alkyl sulfate and/or alkyl sulfonate anionic surfactant and a zwitterionic surfactant; about 2-20 wt % of a water-miscible solvent, which includes one or more of an alkylene glycol, glycerol and a glycol ether; and at least about 50 wt % water. The aqueous firefighting foam concentrate may further include a nonionic surfactant, and/or a polysaccharide thickener. The aqueous firefighting foam concentrate is free of monosaccharide sugars, and the aqueous firefighting foam concentrate does not comprise a eutectic solvent.

**[0053]** In another illustrative embodiment, the aqueous firefighting foam concentrate includes about 5 to 25 wt % disaccharide sugar; about 3 to 15 wt % C<sub>8-14</sub>-alkyl sulfate anionic surfactant; about 2 to 7 wt % alkylamidopropyl hydroxysultaine; about 1 to 5 wt % ethylene glycol and/or propylene glycol; about 1 to 5 wt % diethylene glycol monobutyl ether; and at least about 50 wt % water. The aqueous firefighting foam concentrate may further include about 0.2 to 3 wt % diutan gum and/or about 0.2 to 1.5 wt % xanthan gum. The aqueous firefighting foam concentrate may further include a nonionic surfactant. The aqueous firefighting foam concentrate does not contain monosaccharide sugars and does not comprise a eutectic solvent. Such concentrates commonly contain no more than 0.01 wt % fluorinated surfactants.

**[0054]** In another illustrative embodiment, the aqueous firefighting foam concentrate includes about 5 to 25 wt % disaccharide sugar; about 3 to 15 wt % C<sub>8-14</sub>-alkyl sulfate anionic surfactant; about 2 to 7 wt % alkylamidopropyl hydroxysultaine; about 1 to 5 wt % ethylene glycol and/or propylene glycol; about 1 to 5 wt % diethylene glycol monobutyl ether; and at least about 50 wt % water, and the aqueous firefighting foam concentrate is free of monosaccharide sugars. The aqueous firefighting foam concentrate does not comprise a eutectic solvent. The aqueous firefighting foam concentrate may further include about 0.2 to 4 wt % welan gum and/or about 0.2 to 1.5 wt % xanthan gum. The aqueous firefighting foam concentrate may further include a nonionic surfactant. Such concentrates commonly contain no more than 0.01 wt % fluorinated surfactants.

**[0055]** In another illustrative embodiment, the aqueous firefighting foam concentrate includes about 5 to 25 wt % disaccharide sugar; about 3 to 15 wt % C<sub>8-14</sub>-alkyl sulfate anionic surfactant; about 2 to 7 wt % alkylamidopropyl hydroxysultaine; about 1 to 5 wt % ethylene glycol and/or propylene glycol; about 1 to 5 wt % diethylene glycol monobutyl ether; and at least about 50 wt % water, and the aqueous firefighting foam concentrate does not contain monosaccharide sugars. The aqueous firefighting foam concentrate does not comprise a eutectic solvent. The aqueous firefighting foam concentrate may further include about 0.2 to 4 wt % succinoglycan and/or about 0.2 to 1.5 wt % xanthan gum. The aqueous firefighting foam concentrate may further include a nonionic surfactant. Such concentrates commonly contain no more than 0.01 wt % fluorinated surfactants.

**[0056]** In another embodiment, the concentrate includes: about 5 to 25 wt % disaccharide sugar; about 5-15 wt % C<sub>8-14</sub>-alkyl sulfate anionic surfactant; about 2-10 wt % C<sub>8-18</sub>-alkylamidopropyl hydroxysultaine surfactant; propylene glycol and diethylene glycol monobutyl ether; and at least about 50 wt % water; wherein the concentrate contains no more than 0.01 wt % fluorinated surfactants. The aqueous firefighting foam concentrate may further include about 1-4 wt % polysaccharide thickener; and/or about 0.5-5 wt % of

an alkyl polyglycoside, an aliphatic alcohol-based nonionic surfactant, or a mixture of any two or more thereof. In some embodiments, the polysaccharide thickener comprises xanthan gum and diutan gum. In some embodiments, the polysaccharide thickener comprises xanthan gum and welan gum. In some embodiments, the polysaccharide thickener comprises xanthan gum and succinoglycan. The aqueous firefighting foam concentrate does not include monosaccharide sugars and does not include a eutectic solvent.

**[0057]** Any of the illustrative foam concentrates described above may also include one or more of a chelator, a buffer, a corrosion inhibitor and a preservative. Such concentrates may have a pH of about 7 to 9.

**[0058]** Tables 1 and 2 below provide an illustration of suitable aqueous firefighting foam concentrate formulations designed to be combined with a diluent, aerated, and administered to fight a fire as a firefighting foam.

TABLE 1

Ingredient	Amount (wt %)
Biocide	0-1.0
Corrosion Inhibitor	0-0.1
Sugar Component	5-25
Zwitterionic Surfactant	1-10
Aliphatic alcohol-based	0-5
Nonionic Surfactant	
Anionic Surfactant	2-20
Organic Solvent	1-20
Polysaccharide Gum	0-5
Water	30-85

TABLE 2

Ingredient	Amount (wt %)
Biocide	0-1.0
Corrosion Inhibitor	0-0.1
Sugar Component	5-25
Zwitterionic Surfactant	1-5
Aliphatic alcohol-based	0-5
Nonionic Surfactant	
Anionic Surfactant	2-10
Organic Solvent	2-10
Polysaccharide Gum	0-3
Water	30-85

**[0059]** The present invention, thus generally described, will be understood more readily by reference to the following examples, which are provided by way of illustration and are not intended to be limiting of the present invention.

#### EXAMPLES

**[0060]** Example 1. Tables 3-10 include a number of illustrative compositions for the aqueous firefighting foam concentrate. The amounts shown in these tables represent the weight percentage of the particular component based on the total weight of the composition. The formulations include a) a surfactant mixture including an anionic surfactant and a zwitterionic surfactant, b) a water-miscible organic solvent comprising one or more of an alkylene glycol, glycerol and a glycol ether, and c) at least about 30 wt % water. The formulations may further include: d) a polysaccharide thickener and/or e) a nonionic surfactant. The formulations are free of monosaccharide sugars and do not comprise a eutectic solvent.

TABLE 3

Ingredient	Amount (wt %)
Biocide	0-0.2
Corrosion inhibitor	0-0.1
Water	30-80
Zwitterionic surfactant	1-15
C <sub>8-14</sub> -Fatty alcohol	0-5
C <sub>8-14</sub> -Alkyl sulfate surfactant	2-15
Polysaccharide gum	0-5
Alkylene glycol	1-15
Alkyl carbitol	1-15
Total Organic Solvent	1-20
Disaccharide sugar(s)	5-25

TABLE 4

Ingredient	Amount (wt %)
Biocide	0-0.2
Corrosion Inhibitor	0-0.1
Water	50-80
Cocamidoalkyl hydroxysultaine	1-6
C <sub>8-12</sub> -Fatty Alcohol	0-3
C <sub>8-12</sub> -Alkylsulfate	3-10
Diutan Gum	0-3
Xanthan Gum	0-2
Ethylene Glycol and/or	1-10
Propylene Glycol	
Diethylene glycol monobutyl ether	1-10
Disaccharide sugar(s)	5-25

TABLE 5

Ingredient	Amount (wt %)
Biocide	0-0.2
Corrosion Inhibitor	0-0.1
Water	50-80
Cocamidoalkyl hydroxysultaine	1-7
C <sub>8-12</sub> -Fatty Alcohol	0-3
C <sub>8-12</sub> -Alkylsulfate	3-10
Welan Gum	0-4
Xanthan Gum	0-2
Ethylene Glycol and/or	1-10
Propylene Glycol	
Diethylene glycol monobutyl ether	1-10
Disaccharide sugar(s)	5-25

TABLE 6

Ingredient	Amount (wt %)
Biocide	0-0.2
Corrosion Inhibitor	0-0.1
Water	50-80
Cocamidoalkyl hydroxysultaine	1-7
C <sub>8-12</sub> -Fatty Alcohol	0-3
C <sub>8-12</sub> -Alkylsulfate	3-10
Succinoglycan	0-4
Xanthan Gum	0-2
Ethylene Glycol and/or	1-10
Propylene Glycol	
Diethylene glycol monobutyl ether	1-10
Disaccharide sugar(s)	5-25

**[0061]** The aqueous firefighting foam formulations shown in Tables 1-6 typically have a pH of about 7 to 9. According to various embodiments, the formulations shown in Tables 1-6 may have a pH of about 7.5 to 8.5. If necessary, a



pH-adjusting agent may be added to the concentrate to achieve the desired pH range.

**[0062]** The illustrative aqueous firefighting foam formulations shown in Tables 1-6 may have a viscosity of about 1,000 to 5,000 cps and, in some embodiments, may have a viscosity of about 1,000 to 2,500 cps (as determined at room temperature (75° F./24° C.) with a #4 spindle at 30 rpm).

**[0063]** Example 2. Table 7 shows an illustrative composition for an aqueous firefighting foam that includes a reducing agent. The amounts shown in these tables represent the weight percentage of the particular component based on the total weight of the composition. The formulations include a) a surfactant mixture including an anionic surfactant and a zwitterionic surfactant, b) a water-miscible organic solvent comprising one or more of an alkylene glycol, glycerol and a glycol ether, and c) at least about 30 wt % water. The formulations may further include: d) a polysaccharide thickener, e) a nonionic surfactant, and/or f) a reducing agent.

TABLE 7

Ingredient	Amount (wt %)
Biocide	0-0.2
Corrosion inhibitor	0-0.1
Water	30-80
Zwitterionic surfactant	1-15
C <sub>8-14</sub> -Fatty alcohol	0-5
C <sub>8-14</sub> -Alkyl sulfate surfactant	2-15
Polysaccharide gum	0-5
Alkylene glycol	1-15
Alkyl carbitol	1-15
Total Organic Solvent	1-20
Disaccharide sugar(s)	5-25
Reducing agent	0-5

**[0064]** Example 3. Tables 9-12 includes illustrative compositions for an aqueous firefighting foam concentrate comprising a sugar component that is a disaccharide. The amounts shown in these tables represent the weight percentage of the particular component based on the total weight of the composition.

TABLE 9

Ingredient	Amount (wt %)
Deionized Water	43.40
Metallic Salt	2.00
Biocide	0.05
Sucrose	15.00
Zwitterionic surfactant	8.00
Anionic surfactant	20.00
Water miscible organic solvent	6.50

TABLE 9-continued

Ingredient	Amount (wt %)
Corrosion inhibitor	0.05
Polysaccharide thickener	5.00

TABLE 10

Ingredient	Amount (wt %)
DI Water	60.35
Metal Salt	1.50
Biocide	0.05
Sucrose	15.00
Zwitterionic surfactant	6.00
Anionic surfactant	4.00
Anionic surfactant	6.00
Water miscible organic solvent	5.00
Polysaccharide thickener	2.10

TABLE 11

Ingredient	Amount (wt %)
Biocide	0.08
Corrosion inhibitor	0.06
Metal Salt	1.50
Zwitterionic surfactant	8.00
Anionic surfactant	16.00
Water miscible organic solvent	1.50
Sucrose	15.00
Water	57.86

TABLE 12

Ingredient	Amount (wt %)
Biocide	0.05
Corrosion inhibitor	0.06
Metal salt	1.25
Zwitterionic surfactant	7.00
Anionic surfactant	14.00
Water miscible organic solvent	5.00
Polysaccharide thickener	1.00
Sucrose	15.00
Water	56.64

**[0065]** Example 4. Table 13 below includes illustrative compositions for an aqueous firefighting foam concentrate comprising a sugar component that is a disaccharide. The unaged and aged formulations were tested for their ability to extinguish different types of fuel at full-, half-, and double-strength. The results are shown below in Table 14.

TABLE 13

Ingredient (amount in wt %)	Formulation A	Formulation B	Formulation C	Formulation D	Formulation E	Formulation F
Corrosion inhibitor	0.10	0.05	0.05	0.05	0.05	0.05
Biocide	0.25	—	0.05	0.25	0.25	0.25
Metallic salt	1.50	—	—	—	—	0.25
Non-ionic surfactant	—	6.00	6.00	—	—	6.00
Sucrose (table sugar)	20.00	—	—	20.00	20.00	15.00

TABLE 13-continued

Ingredient (amount in wt %)	Formulation A	Formulation B	Formulation C	Formulation D	Formulation E	Formulation F
Sugar	—	—	15.00	—	—	—
alcohol	—	—	—	—	—	—
Zwitterionic surfactant	11.00	6.00	6.00	6.00	6.00	6.00
Anionic surfactant	22.00	18.00	18.00	18.00	18.25	12.00
Glycerol	—	16.00	16.00	—	—	16.00
Alkylene glycol	—	—	—	—	0.50	10.00
Glycol ether	10.00	10.00	8.00	1.00	1.25	1.25
Alkanol- amine	—	—	—	0.50	—	—
Water	35.15	43.95	30.90	54.20	53.70	33.20

TABLE 14

	Formulation A	Formulation B	Formulation C	Formulation D	Formulation E	Formulation F
Jet Fuel A Fire, 48 ft <sup>2</sup> -Unaged, full-strength aqueous fire-fighting concentrate						
90% control (min:sec)	0:38	0:44	0:37	0:38	—	—
Extinguish (min:sec)	0:46	0:58	0:42	0:46	—	—
Burnback (min:sec)	9:55	8:07	10:30	8:21	—	—
Expansion/ 25% drain time (Exp/DT (min:sec))	11.5/3:35	11.1/3:21	11.4/3:34	10.7/3:35	—	—
Jet Fuel A Fire, 50 ft <sup>2</sup> -Unaged, full-strength aqueous fire-fighting concentrate						
90% control (min:sec)	—	—	0:20	0:18	0:24	0:22
Extinguish (min:sec)	—	—	0:35	0:38	0:46	0:42
Burnback (min:sec)	—	—	25% @ 6:35	25% @ 6:47	25% @ 6:57	25% @ 6:35
Exp/DT	—	—	9.09/2:57	7.70/10:36	6.92/9:30	8.47/4:16
Gasoline Fire, 28 ft <sup>2</sup> -Unaged, full-strength aqueous fire-fighting concentrate						
90% control (min:sec)	—	—	0:31	0:30	0:13	0:12
Extinguish (min:sec)	—	—	0:59	0:57	0:21	0:23
Burnback (min:sec)	—	—	25% @ 4:13	25% @ 5:30	25% @ 7:18	25% @ 8:24
Exp/DT	—	—	6.94/4:40	7.69/9:20	9.01/ 8:03	11.11/4:18
Dry Chemical Compatibility-Unaged, full-strength aqueous fire-fighting concentrate						
90% control (min:sec)	—	—	0:13	—	0:35	0:36
Extinguish (min:sec)	—	—	0:28	—	0:50	0:57
Burnback (min:sec)	—	—	25% @ 4:30	—	25% @ 4:27	25% @ 4:23
Exp/DT	—	—	8.26/4:50	—	9.39/9:00	10.00/4:05
Gasoline Fire, 28 ft <sup>2</sup> -Aged, full strength aqueous firefighting concentrate						
90% control (min:sec)	—	—	0:32	0:27	0:17	0:16
Extinguish (min:sec)	—	—	0:54	0:46	0:25	0:24
Burnback (min:sec)	—	—	25% @ 4:44	25% @ 5:15	25% @ 4:40	25% @ 5:10
Exp/DT	—	—	10.53/3:50	8:16	7.01/9:15	7.94/5:10

TABLE 14-continued

	Formulation A	Formulation B	Formulation C	Formulation D	Formulation E	Formulation F
Jet A Fuel Fire, 28 ft <sup>2</sup> - Unaged, Half Strength aqueous firefighting concentrate						
90% control (min:sec)	—	—	0:14	0:22	0:32	0:33
Extinguish (min:sec)	—	—	0:23	0:26	0:52	0:58
Burnback (min:sec)	—	—	25% @ 7:47	25% @ 6:30	25% @ 5:06	25% @ 4:47
Exp/DT	—	—	7.81/4:55	7.87/10:37	8.34/8:56	9.52/4:13
Jet A Fuel Fire, 28 ft <sup>2</sup> -Unaged, Double Strength aqueous firefighting concentrate						
90% control (min:sec)	—	—	0:12	0:16	0:19	0:22
Extinguish (min:sec)	—	—	0:15	0:23	0:26	0:29
Burnback (min:sec)	—	—	25% @ 7:40	25% @ 8:15	25% @ 6:24	25% @ 8:55
Exp/DT	—	—	12.05/3:40	8.33/4:08	7.05/9:40	7.35/6:45
Jet A Fuel Fire, 28 ft <sup>2</sup> - Unaged, full-strength aqueous firefighting concentrate						
90% control (min:sec)	—	—	—	0:19	0:16	0:16
Extinguish (min:sec)	—	—	—	0:23	0:21	0:25
Burnback (min:sec)	—	—	—	25% @ 8:13	25% @ 8:50	25% @ 8:48
Exp/DT	—	—	—	9.35/8:05	7.22/9:14	8.93/4:41
Jet A Fuel Fire, 28 ft <sup>2</sup> -Aged, full-strength aqueous firefighting concentrate						
90% control (min:sec)	—	—	—	—	0:21	0:22
Extinguish (min:sec)	—	—	—	—	0:27	0:26
Burnback (min:sec)	—	—	—	—	25% @ 6:39	25% @ 7:16
Exp/DT	—	—	—	—	7.27/11:18	7.94/5:45
Jet A Fuel Fire, 28 ft <sup>2</sup> -Aged, half-strength aqueous firefighting concentrate						
90% control (min:sec)	—	—	—	0:14	0:10	0:18
Extinguish (min:sec)	—	—	—	0:27	0:19	0:22
Burnback (min:sec)	—	—	—	25% @ 7:28	25% @ 7:53	25% @ 9:33
Exp/DT	—	—	—	9.01/9:17	8.12/4:27	13.51/4:15

**[0066]** Example 5. Method of Producing a Firefighting Foam. The firefighting foam concentrates described herein may be mixed with a diluent to form firefighting foam precursor solution, i.e., a use strength composition. The firefighting foam precursor solution may be aerated (e.g., using a nozzle) to produce a firefighting foam including the firefighting foam concentrate and the diluent. Illustrative diluents may include water, such as fresh water, brackish water, sea water, and combinations thereof. In some embodiments, the aqueous firefighting foam concentrate described above may be 1 vol. %, 3 vol. %, or 5 vol. % concentrate solutions, meaning that the aqueous firefighting foam concentrates are mixed with 99 vol. %, 97 vol. %, or 95 vol. % diluent, respectively, to form the firefighting foam precursor solution.

**[0067]** In some instances, it has been found that the order of addition of ingredients with appropriate agitation may impact the actual firefighting performance as seen in the UL and EN fire tests. It may be suitable to begin by mixing the sugar component with a substantial amount of water and subsequently preparing a solution or slurry of the polysaccharide thickener in the aqueous sugar solution prior to blending in the remaining components of the foam concen-

trate. It was found that first preparing an aqueous sugar solution by combining and mixing the sugars (e.g., sucrose) with water may facilitate later dissolution and/or dispersal of the biogums/biopolymers (e.g., xanthan gum), if present in the aqueous firefighting foam concentrate. This can allow the gums to properly hydrate without encapsulating (clumping) upon the addition of the surfactant(s), other optional compounds and remaining amounts of water. Surfactants and other optional additives can then be added and the resulting mixture may finally be diluted further with water to decrease the viscosity of the preparation, if desired.

**[0068]** Firefighting foams that were prepared not following this order of component addition may result in polysaccharide bio gums that are encapsulated, but not fully hydrated, which can result in the production of foams that are not satisfactory for fire testing. Thus, in some embodiments, the initial formation of an aqueous sugar solution is important in process order and can be used to dissolve/disperse polysaccharide thickener(s) into the foam concentrate before addition of any other ingredients, such as surfactant(s) and/or other additives.

**[0069]** Example 5, Method of Fighting a Fire. The firefighting foam compositions described herein may be used to

fight a fire and/or to suppress flammable vapors by mixing the firefighting foam compositions with a diluent, aerating the resulting firefighting foam precursor solution to form a firefighting foam, and administering the firefighting foam to a fire or applying the firefighting foam to the surface of a volatile flammable liquid (e.g., gasoline or other flammable hydrocarbon or a flammable polar solvent).

[0070] While certain embodiments have been illustrated and described, it should be understood that changes and modifications can be made therein in accordance with ordinary skill in the art without departing from the technology in its broader aspects.

[0071] The embodiments illustratively described herein may suitably be practiced in the absence of any element or elements, limitation or limitations, not specifically disclosed herein. Thus, for example, the terms “comprising,” “including,” “containing,” shall be read expansively and without limitation. Additionally, the terms and expressions employed herein have been used as terms of description and not of limitation, and there is no intention in the use of such terms and expression of excluding any equivalents of the features shown and described or portions thereof, but it is recognized that various modifications are possible within the scope of the claimed technology. Additionally, the phrase “consisting essentially of” will be understood to include those elements specifically recited and those additional elements that do not materially affect the basic and novel characteristics of the claimed technology. The phrase “consisting of” excludes any element not specified.

[0072] Additionally, where features or aspects of the disclosure are described in terms of Markush groups, those skilled in the art will realize that the disclosure is also thereby described in terms of any individual member or subgroup of members of the Markush group.

[0073] As will be understood by one skilled in the art, for any and all purposes, particularly in terms of providing a written description, all ranges disclosed herein also encompass any and all possible subranges and combinations of subranges thereof. Any listed range can be easily recognized as sufficiently describing and enabling the same range being broken down into at least equal halves, thirds, quarters, fifths, tenths, etc. As a non-limiting example, each range discussed herein can be readily broken down into a lower third, middle third and upper third, etc. As will also be understood by one skilled in the art all language such as “up to,” “at least,” “greater than,” “less than,” and the like, include the number recited and refer to ranges which can be subsequently broken down into subranges as discussed above. Finally, as will be understood by one skilled in the art, a range includes each individual member.

[0074] All publications, patent applications, issued patents, and other documents referred to in this specification are herein incorporated by reference as if each individual publication, patent application, issued patent, or other document was specifically and individually indicated to be incorporated by reference in its entirety. Definitions that are contained in text incorporated by reference are excluded to the extent that they contradict definitions in this disclosure.

[0075] Other embodiments are set forth in the following claims.

What is claimed is:

1. An aqueous fire-fighting foam concentrate comprising:
  - a sugar component comprising a disaccharide sugar;
  - a surfactant component comprising an anionic surfactant, a zwitterionic surfactant, or a mixture of any two or more thereof;

a water-miscible organic solvent; and

at least about 30 wt % water,

wherein the aqueous fire-fighting foam concentrate does not comprise a monosaccharide sugar, and wherein the aqueous fire-fighting foam concentrate does not comprise a eutectic solvent.

2. The aqueous fire-fighting foam concentrate of claim 1 comprising:

about 5 to 25 wt % of the sugar component;

about 2 to 20 wt % of the anionic surfactant;

about 1 to 10 wt % of the zwitterionic surfactant;

about 1 to 20 wt % of the water-miscible organic solvent; and

at least about 30 wt % water.

3. The aqueous fire-fighting foam concentrate of claim 1, wherein the disaccharide sugar comprises sucrose, maltose, lactose, lactulose, trehalose, cellobiose, chitobiose, or a combination of any two or more thereof.

4. The aqueous fire-fighting foam concentrate of claim 1, wherein the anionic surfactant comprises an alkyl sulfate salt, an alkyl sulfonate salt, an alkyl ether sulfate surfactant, an alkyl ether sulfonate surfactant, or a mixture of any two or more thereof.

5. The aqueous fire-fighting foam concentrate of claim 1, wherein the anionic surfactant comprises C<sub>8</sub>-C<sub>14</sub> alkyl sulfate, C<sub>8</sub>-C<sub>14</sub> alkyl sulfonate, C<sub>10</sub>-C<sub>14</sub> alkyl ether sulfate, or a mixture of any two or more thereof.

6. The aqueous fire-fighting foam concentrate of claim 1, wherein the zwitterionic surfactant comprises an alkylamidoalkyl hydroxysultaine, an alkylamidoalkyl betaine, an alkyl sulfobetaine, an alkyl betaine, or a mixture of any two or more thereof.

7. The aqueous fire-fighting foam concentrate of claim 1, wherein the water-miscible organic solvent comprises an alkylene glycol, a glycerol, a water-soluble polyethylene glycol, a glycol ether, or a mixture of any two or more thereof.

8. The aqueous fire-fighting foam concentrate of claim 1, wherein the water-miscible organic solvent comprises propylene glycol and diethylene glycol monobutyl ether.

9. The aqueous fire-fighting foam concentrate of claim 1, wherein the surfactant component further comprises a non-ionic surfactant.

10. The aqueous fire-fighting foam concentrate of claim 9, wherein the nonionic surfactant comprises an alkyl polyglycoside, an aliphatic alcohol-based nonionic surfactant, or a mixture of any two or more thereof.

11. The aqueous fire-fighting foam concentrate of claim 1 further comprising about 0.2 to 5 wt % of a nonionic surfactant.

12. The aqueous fire-fighting foam concentrate of claim 1 further comprising a polysaccharide thickener.

13. The aqueous fire-fighting foam concentrate of claim 12, wherein the polysaccharide thickener comprises agar, sodium alginate, carrageenan, gum arabic, gum guaiacum, neem gum, pistacia lentiscus, gum chatti, caranna, galactomannan, gum tragacanth, karaya gum, guar gum, welan gum, rhamnsam gum, locust bean gum, beta-glucan, cellulose, methylcellulose, chicle gum, kino gum, dammar gum, glucomannan, succinoglycan, mastic gum, spruce gum, tara gum, gellan gum, xanthan gum, acacia gum, cassia gum, diutan gum, fenugreek gum, ghatti gum, hydroxyethylcellulose, hydroxypropylmethylcellulose, karaya gum, konjac gum, pectin, propylene glycol alginate, or a mixture of any two or more thereof.

14. The aqueous fire-fighting foam concentrate of claim 1 further comprising about 0.1 to 5 wt % of a polysaccharide thickener.

**15.** The aqueous fire-fighting foam concentrate of claim **1**, wherein the concentrate contains no more than 0.01 wt % fluorinated surfactants.

**16.** A method of forming a firefighting foam, the method comprising:

mixing a firefighting foam concentrate with an aqueous diluent to form a foam precursor solution; and aerating the foam precursor solution to form the firefighting foam,

wherein:

the firefighting foam concentrate comprises:

a sugar component comprising a disaccharide sugar;

a surfactant component comprising an anionic surfactant, a zwitterionic surfactant, or a mixture of any two or more thereof;

a water-miscible organic solvent; and

at least about 30 wt. % water,

wherein the aqueous fire-fighting foam concentrate does not comprise a monosaccharide sugar, and

wherein the aqueous fire-fighting foam concentrate does not comprise a eutectic solvent.

**17.** The method of claim **16**, wherein the aqueous diluent is selected from the group consisting of fresh water, brackish water, sea water, and a combination of any two or more thereof.

**18.** The method of claim **16**, wherein the aqueous diluent comprises water from a municipal water source.

**19.** A firefighting foam comprising the firefighting foam composition of claim **1** and an aqueous diluent.

**20.** A method of fighting a fire comprising applying the firefighting foam of claim **19** to the fire.

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