

Question Number	Answer	Mark
17ai	Use of $P = VI$ (1) $P = 0.11$ (W) (1) (MP1 - Allow methods where R is calculated and then either $P = I^2R$ or $P = V^2/R$ is used to calculate P) <u>Example of calculation</u> $P = VI = 12.0 \text{ V} \times 9.2 \times 10^{-3} \text{ A} = 0.11 \text{ W}$	2
17aii	Use of $E = Pt$ to calculate energy of LED (1) Use of $\nu = f\lambda$ and $E = hf$ to calculate photon energy (1) Divides total energy in one minute by energy of a photon (1) Number of photons in one minute = 2.1×10^{19} (1) (candidates who do not convert minutes into seconds can score a maximum of 2 marks – MP2 and MP3) (“show that” value leads to 1.9×10^{19}) (allow full e.c.f. from (i)) <u>Example of calculation</u> $E = Pt = 0.11 \text{ W} \times 60 \text{ s} = 6.6 \text{ J}$ $f = \frac{\nu}{\lambda} = \frac{3.00 \times 10^8 \text{ ms}^{-1}}{627 \times 10^{-9} \text{ m}} = 4.78 \times 10^{14} \text{ Hz}$ $E = hf = 6.63 \times 10^{-34} \text{ Js} \times 4.78 \times 10^{14} \text{ Hz} = 3.17 \times 10^{-19} \text{ J}$ Number of photons in one minute = $\frac{6.6 \text{ J}}{3.17 \times 10^{-19} \text{ J}} = 2.1 \times 10^{19}$	4
17b	(Lower wavelength leads to) greater (photon) energy (1) Therefore fewer photons (in one minute) (1) (MP1 – accept hc/λ increases or hf increases) (MP2 dependent on awarding of MP1)	2
17c	Use of $A = 4\pi r^2$ (1) Use of $I = P/A$ (1) $I = 2200 \text{ (W m}^{-2}\text{, which is greater than } 1100 \text{ W m}^{-2}\text{) so student is correct}$ (1) (MP2 – allow if use $A = 0.69/1100$) <u>Example of calculation</u> $A = 4\pi (0.005)^2 = 3.14 \times 10^{-4} \text{ m}^2$ $I = \frac{0.69 \text{ W}}{3.14 \times 10^{-4} \text{ m}^2} = 2196 \text{ W m}^{-2}$	3
Total for question 17		11