A cylinder contains a mixture of gas X and gas Y. The mean kinetic energy of the molecules of gas X is $E_{\rm x}$ and the mean kinetic energy of the molecules of gas Y is $E_{\rm y}$. The molecules of gas X have twice the mass of the molecules of gas Y.

Which of the following equations is correct?

$$\mathbf{B} \quad E_{\mathbf{X}} = E_{\mathbf{Y}}$$

$$\square$$
 C $E_{\rm x} = 2E_{\rm x}$

$$\mathbf{C}$$
 $E_{\mathrm{X}} =$

$$\square$$
 D $E_{\rm X} = 4E_{\rm Y}$

 \mathbf{C} $E_{\mathrm{X}} = 2E_{\mathrm{Y}}$

