

Roll No.

Total No. of Questions: 28]

[Total No. of Printed Pages: 7

XIIARJKUT23

9122-Y

CHEMISTRY

Time: 3 Hours]

[Maximum Marks: 70

SECTION-A

I each

- 1. Define Unit Cell.
- 2. Colloidal solution has phases.
 - 3. Phenols are:
 - (A) Basic
 - Acidic
 - (C) Neutral
 - (D) None of these

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4. Primary amines give carbylamine reaction.

(True/False)

- 5. What is IUPAC name of $[Cu(NH_3)_4SO_4]$?
- 6. Sunshine Vitamin is:
 - (A) Vitamin A
 - (B) Vitamin D
 - (C) Vitamin E
 - (D) Vitamin K
- 7. Ketones have:

- (B) —CHO group
- (C) -COOH group
- (D) —OH group

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SECTION-B

2 each

- State Faraday's Ist law of Electrolysis.
- 9. Define order of Reaction.
- 10. Define d-Block elements.

Or

Why d-Block elements form complexes?

Define Ligand. Name its types.

Or

What is Linkage Isomerism? Give an example.

- 12. Define Haloalkanes.
- 13. Give two uses of Chloroform.

SECTION-C

3 each

- 14. Explain Frankel defect.
- 15. Draw a labelled diagram of Galvanic cell.

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- Calculate Half-life period for 1st order reaction.
 - 17. Explain Tyndal effect.
- 18. Explain Zone refining.
- 19. Give the electronic configuration of group 15 elements.
 - 20. Give three oxidising properties of K₂Cr₂O₇.
- Discuss the nutrition of phenol with conc. HNO_3 in presence of conc. H_2SO_4 .
- 22. How Benzene diazonium chloride is prepared?

Or

What are Amines? Give their classification.

23. Define Carbohydrates and give their classification.

0r

What are Enzymes? Give their importance.

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What are Polymers? How are they classified on the basis of structure?

Or

Give the preparation and two uses of polythene.

25. Define Analgesics, Antiseptics and Antibiotics.

Or

Why detergents are preferred over soaps?

SECTION-D

5 each

26. Define elevation in B.P. How can you calculate the molecular mass of a non-volatile solute with it?

Or

Define Osmotic Pressure. How is it determined by Berkeley-Hartley method ?

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Discuss the oxidation states of group 16 elements.

Or

Discuss the manufacture of NH₃ by Haber's process.

- 28. (a) Show how each of the following compounds can be converted into benzoic acid? https://www.jkboseonline.com
 - (i) Ethyl benzene
 - (ii) Acetophenone
 - (iii) Bromobenzene
 - (b) Account for the following:
 - (i) Electrophilic substitution in benzoic acid takes place at meta position
 - (ii) Chloroacetic acid is stronger than acetic acid

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Or

Explain the following:

- (a) Aldol condensation
- (b) Cannizzaro's reaction

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