



# TA201A

## Manufacturing Processes

**Week-11**

**01 Nov, 2022**

**2022-2023 Semester-I**

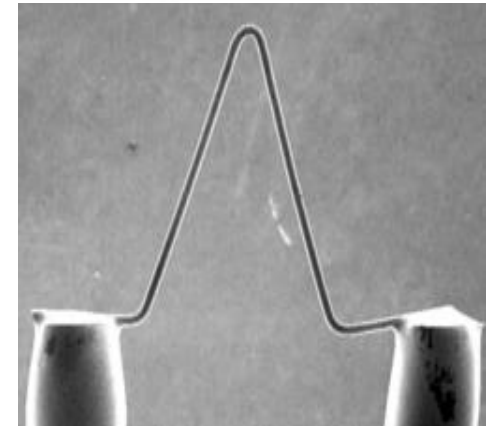
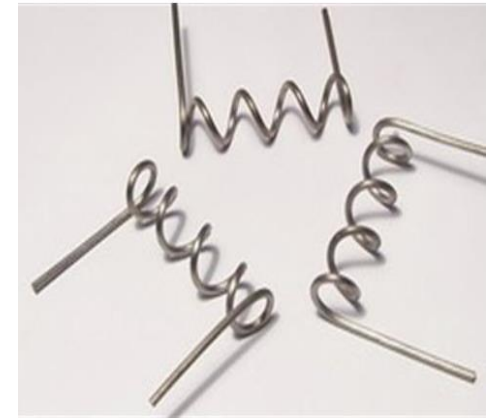
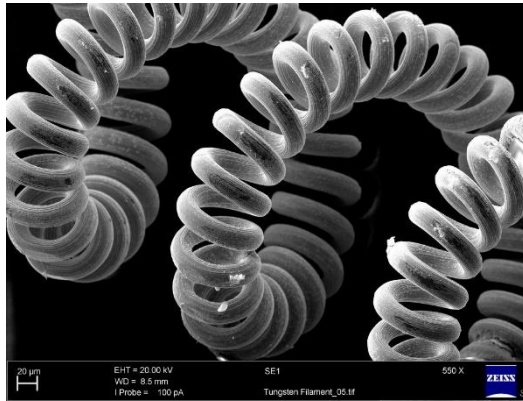
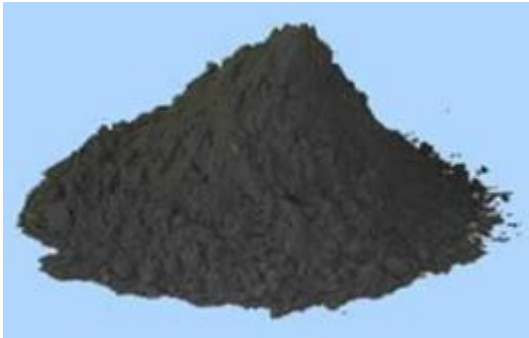
**Lecture 11**



# Powder Metallurgy

## Tungsten Filament

W:  $T_m = 3422^\circ\text{C}$



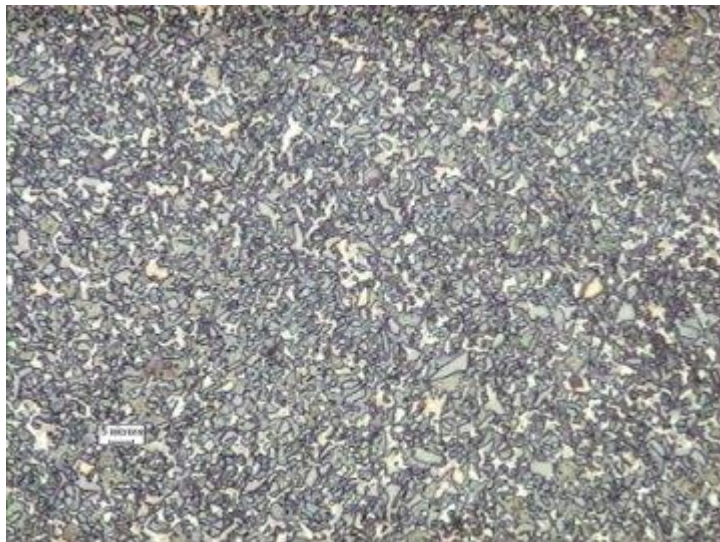
## Powder metallurgy



# Powder Metallurgy

## Cermet cutting tools (Ceramic-metal composite)

Microstructure:  
ceramic particles in metal matrix



Cermet-tipped saw blade for long life





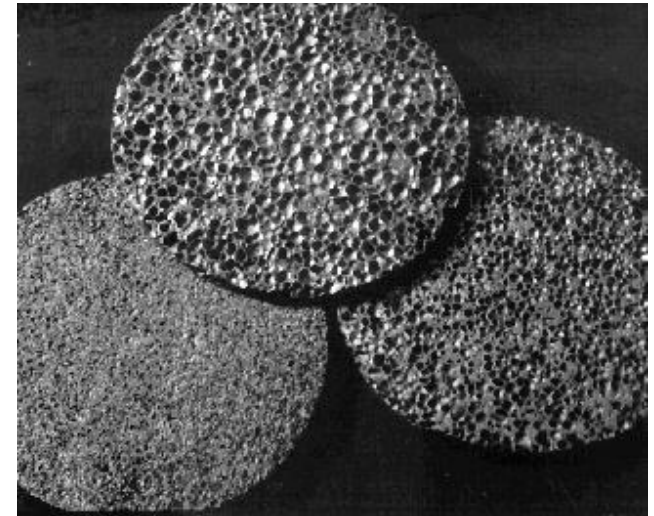
# Powder Metallurgy

## Porous Metals

### Oil-impregnated Porous Bronze Bearings



## Metal filters



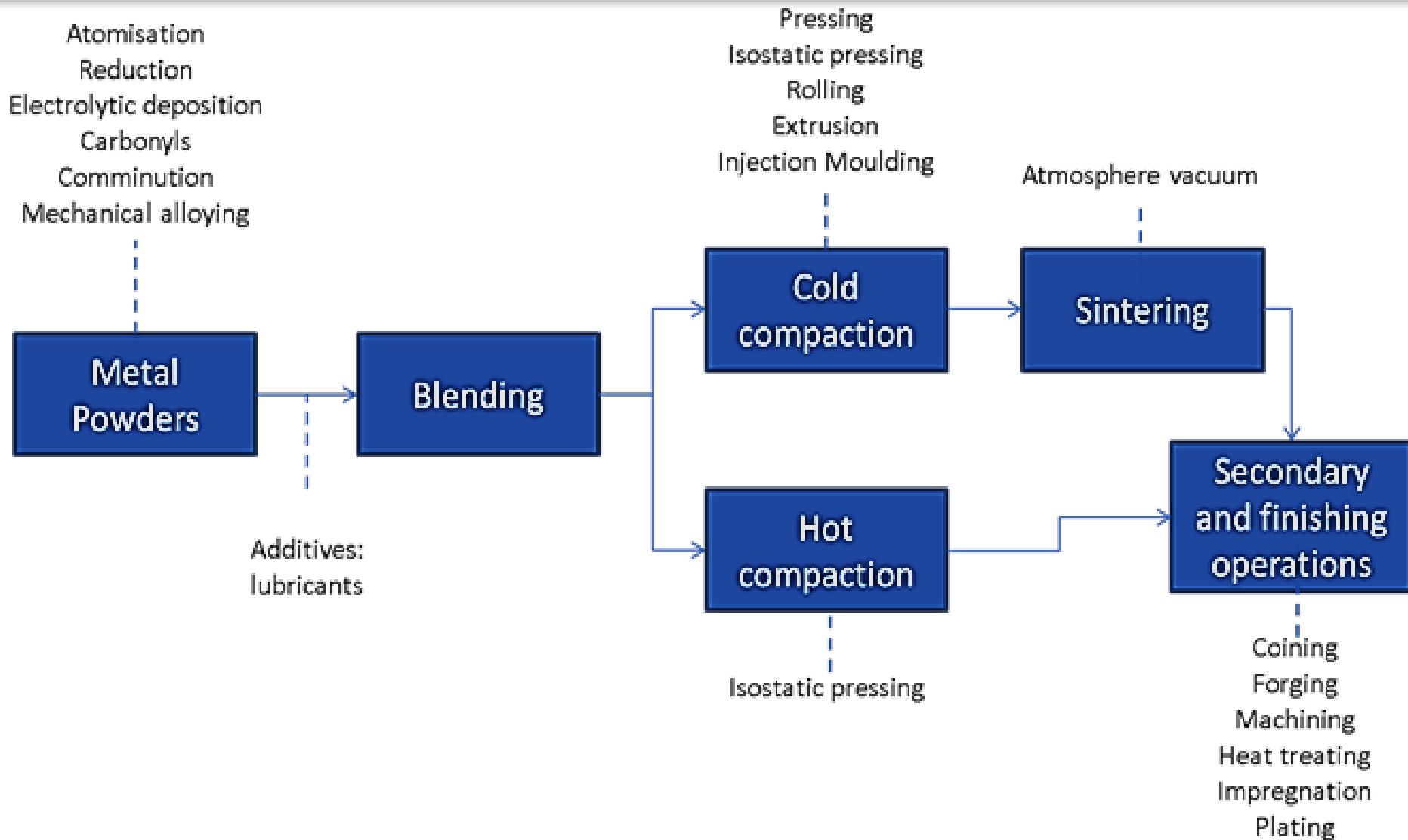


# Importance of P/M

- Versatile and used in numerous industries
- Eliminates or minimizes machining
- Minimizes scrap
- Maintains close dimensional tolerances
- Permits a wide variety of alloy systems
- Facilitates manufacture of complex shapes which would be impractical with other processes\*
- Provides excellent part to part repeatability
- Cost Effective
- Energy and environmentally efficient



# Powder Metallurgy: Flow Chart

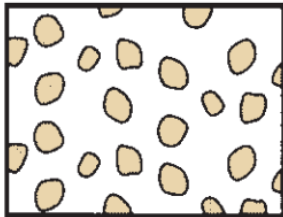






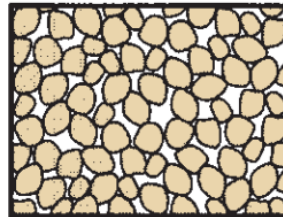
# Conventional powder metallurgy route

Blending

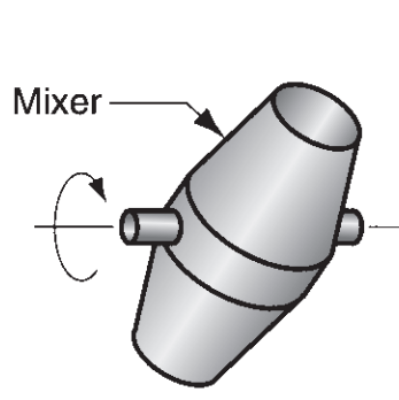
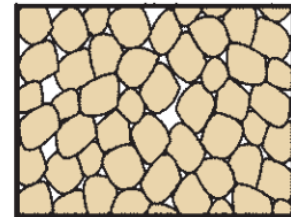


(a)

Compacting

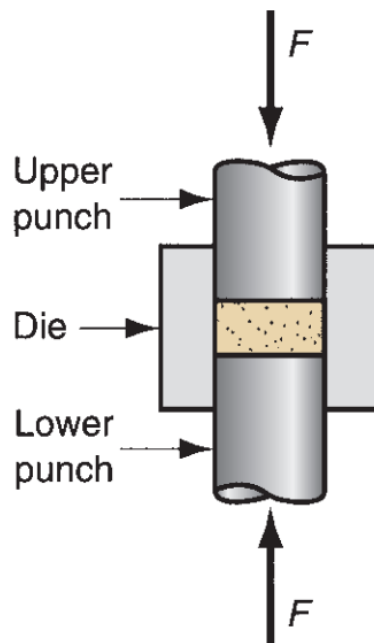


Sintering

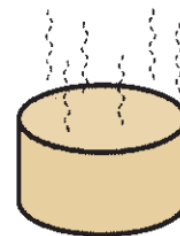


(b)

(1)



(2)



(3)

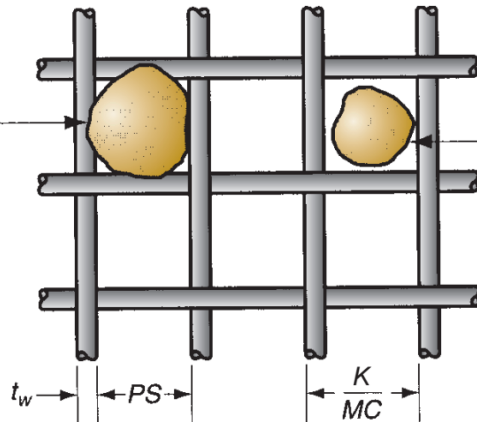


# Powder characterization

## Particle size and distribution



Particle size that would not pass through mesh

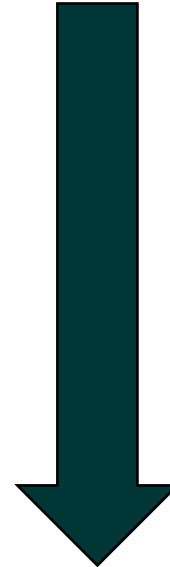


Particle size that would pass through mesh

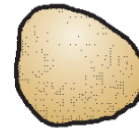
$$PS = \frac{K}{MC} - t_w$$

$PS$  - particle size, mm (in);  
 $MC$  - mesh count, openings per linear inch;  
 $tw$  - wire thickness of screen mesh, mm (in);  
 $K$ : a constant whose value 25.4  
when the size units are mm  
(and  $K = 1.0$  when the units are inches)

## Particle shape and structure



Spherical



Rounded



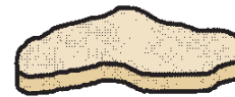
Cylindrical



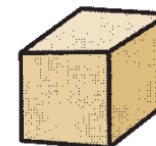
Spongey



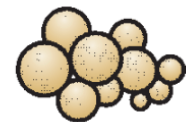
Acicular



Flakey



Cubic



Aggregated

## Particle surface area

$$A = \pi D^2$$

$$\frac{A}{V} = \frac{6}{D}$$

$$V = \frac{\pi D^3}{6}$$

$$\frac{A}{V} = \frac{K_s}{D} \quad \text{or} \quad K_s = \frac{AD}{V}$$

$K_s$  = shape factor





# Powder fabrication techniques

- These can be classified into following main categories

- **Mechanical**

- Milling
- Attritioning and Mechanical alloying

- **Physical**

- Atomization

- **Chemical**

- Decomposition of a solid by a gas
- Thermal decomposition
- Solid-solid reactive synthesis

- **Other**

- Electrolytic techniques
- Microorganism Synthesis



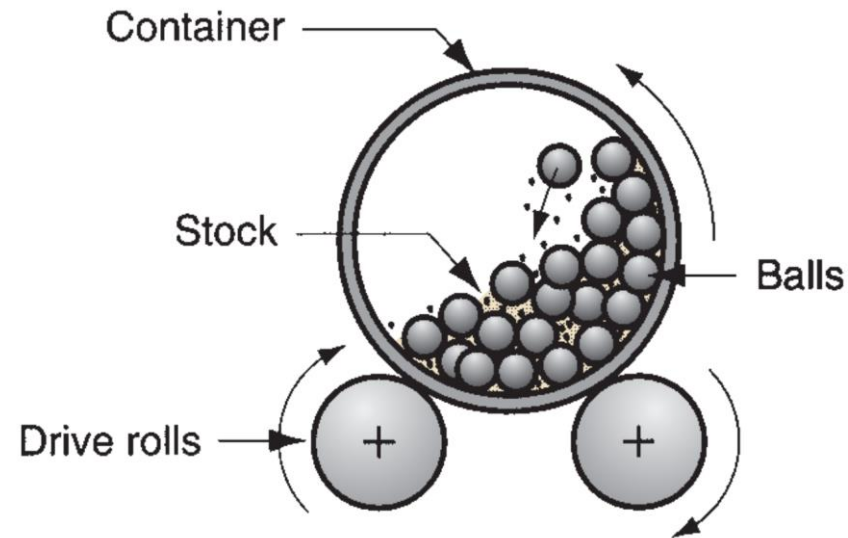
# Powder fabrication: Milling

- Milling implies mechanical impaction using hard balls, rods, or hammers and is a classic approach to fabricating ceramic powders
- Material must be brittle.

[What can you do if a material is ductile like metal?]

- The impact stress required for fracture increases with decreasing particle size

[So How should the size vary with time?]



[What will happen if the speed of the mill is too low or too high?]



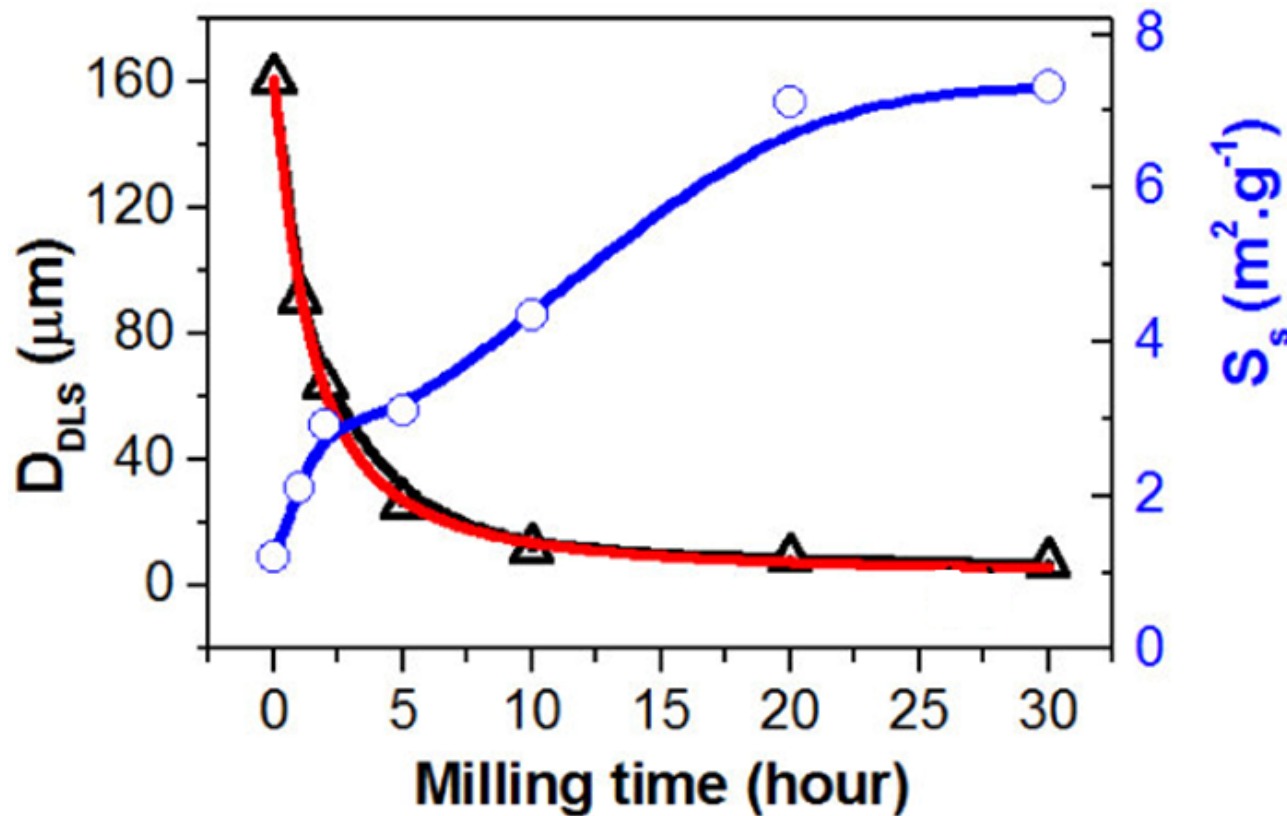
# Powder fabrication: Ball Milling

- Ball milling is **inefficient** because most of the energy goes into **noise and heat**
- For optimal milling
  - The **ball diameter** should be approximately **30 times the diameter of the powder**
  - The **balls** should fill **about half of the jar volume**
  - The **particles** should be about **25% of the jar volume**
- Fluids or protective atmosphere are used to reduce oxidation and aid grinding
- When wet, liquid clings to the media surface.

**How can this influence the particle size that can be obtained?**



# Powder fabrication: Ball Milling



Grain size, specific surface area variation with milling time.

[What will be the change when you add liquid to increase adhesion?]

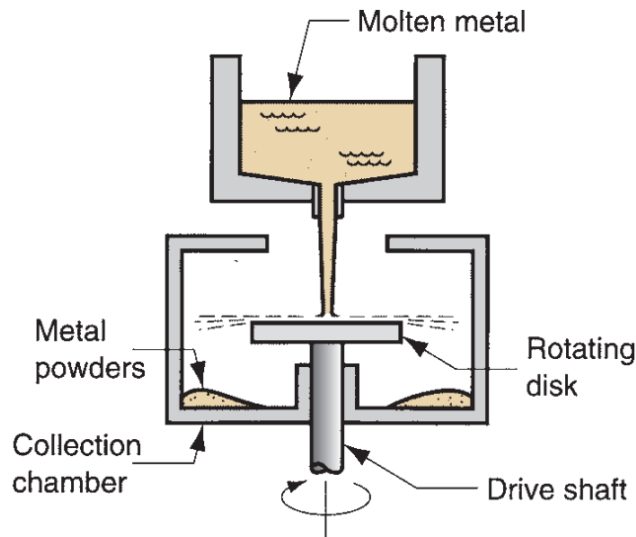
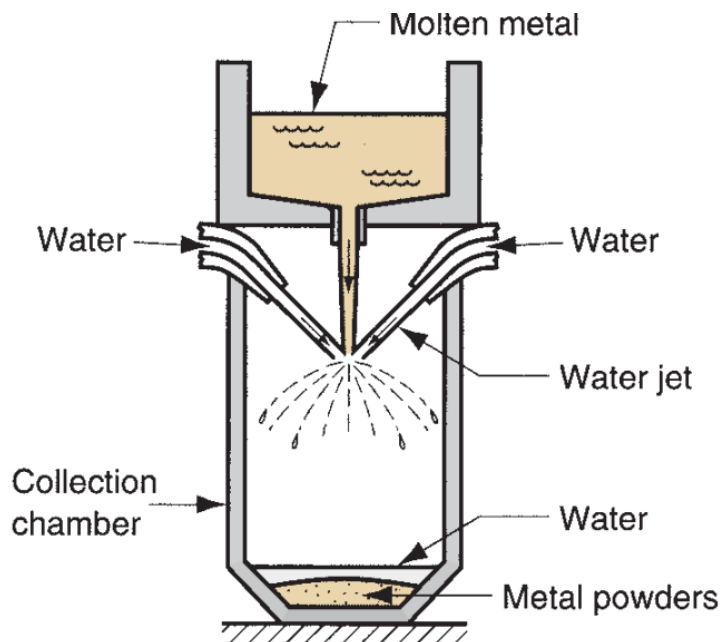
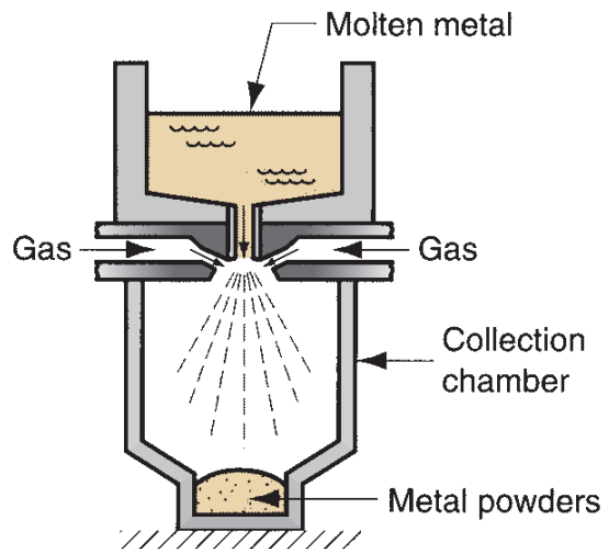
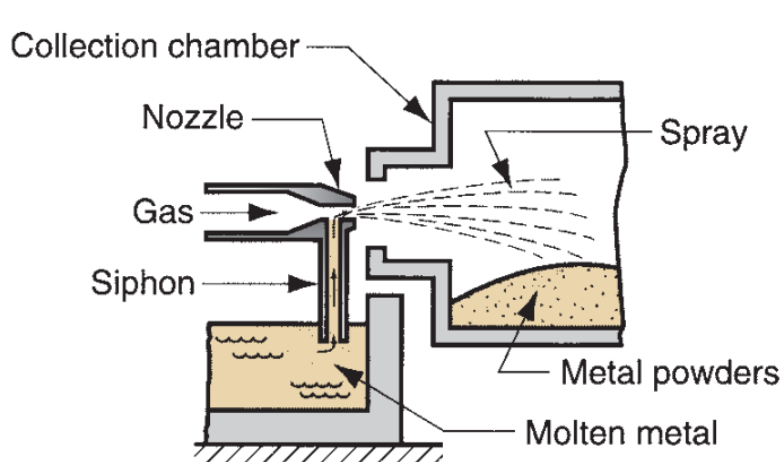


# Powder fabrication: Atomization Techniques

- Relies on disintegration of melt into droplets that freeze into particles
- Commercial atomization units operate at production rates as high as 400 kg/min
- Mostly used for metals, alloys and intermetallics with recent applications in polymers and ceramics
- Two main methods of atomization
  - Gas atomization
  - Liquid atomization



# Powder fabrication: Atomization - Powder production

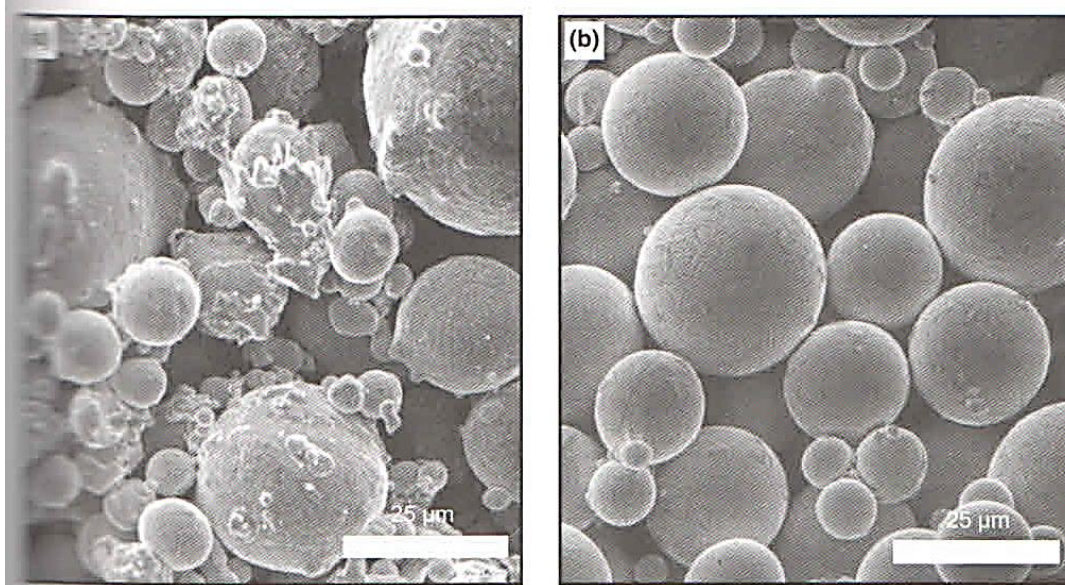






# Gas vs Liquid Atomization

## Gas Atomization

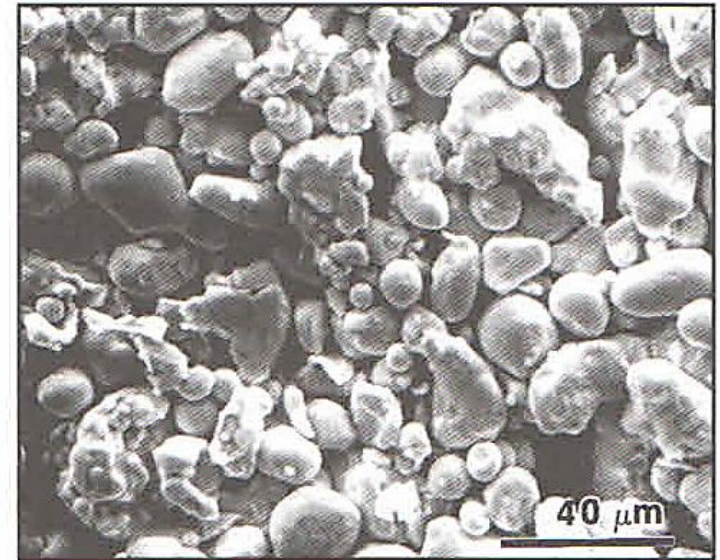


**Figure 3.19.** Scanning electron micrographs of roughly 25 µm inert gas-atomized powders. Although the particles are nearly the same median size, there is a dramatic difference in particle agglomeration and satellite formation associated with control of turbulence and particle reentry into the atomization zone; a) exhibits splats, agglomerates, and satellites and b) is free of satellites.

Difference due to turbulent vs not-so-turbulent flow of gas from the nozzle

Turbulence causes particles to reenter the gas expansion zone, leading to the formation of satellite particles

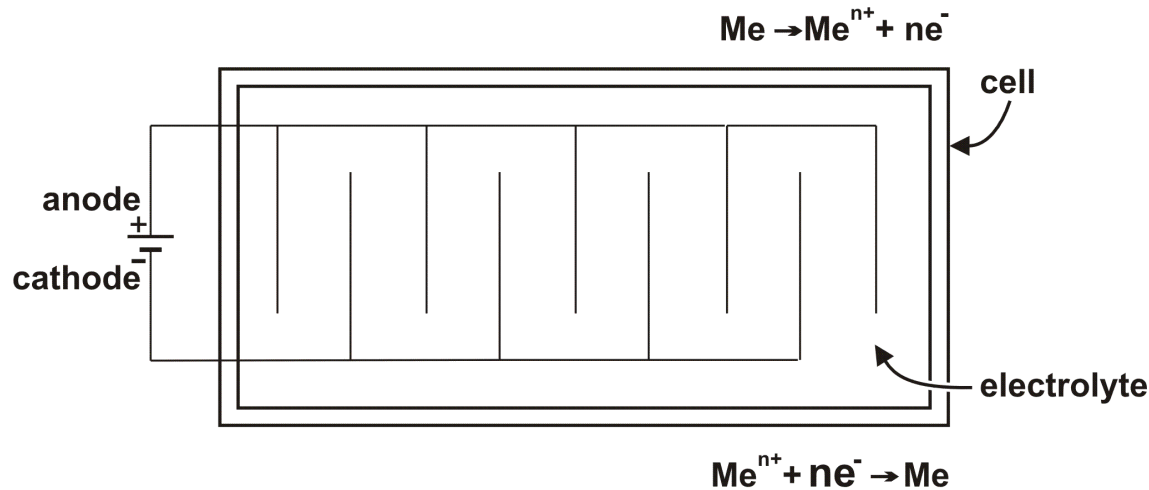
## Water Atomization



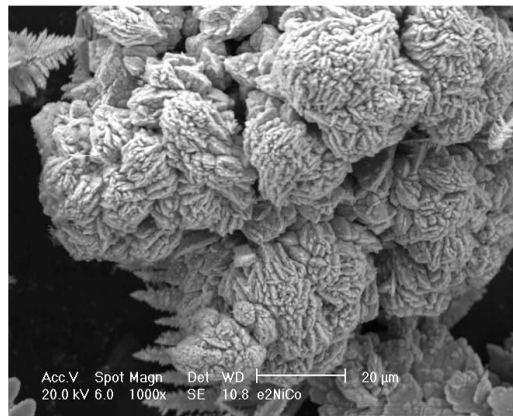
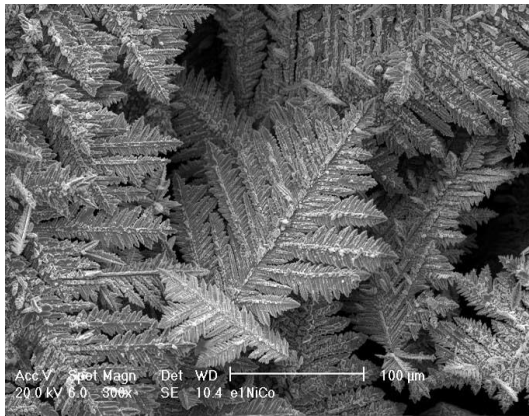
**Figure 3.24.** A scanning electron micrograph of -325 mesh water-atomized stainless steel powder intentionally produced with a rounded shape.



# Powder fabrication: Electrolytic Technique

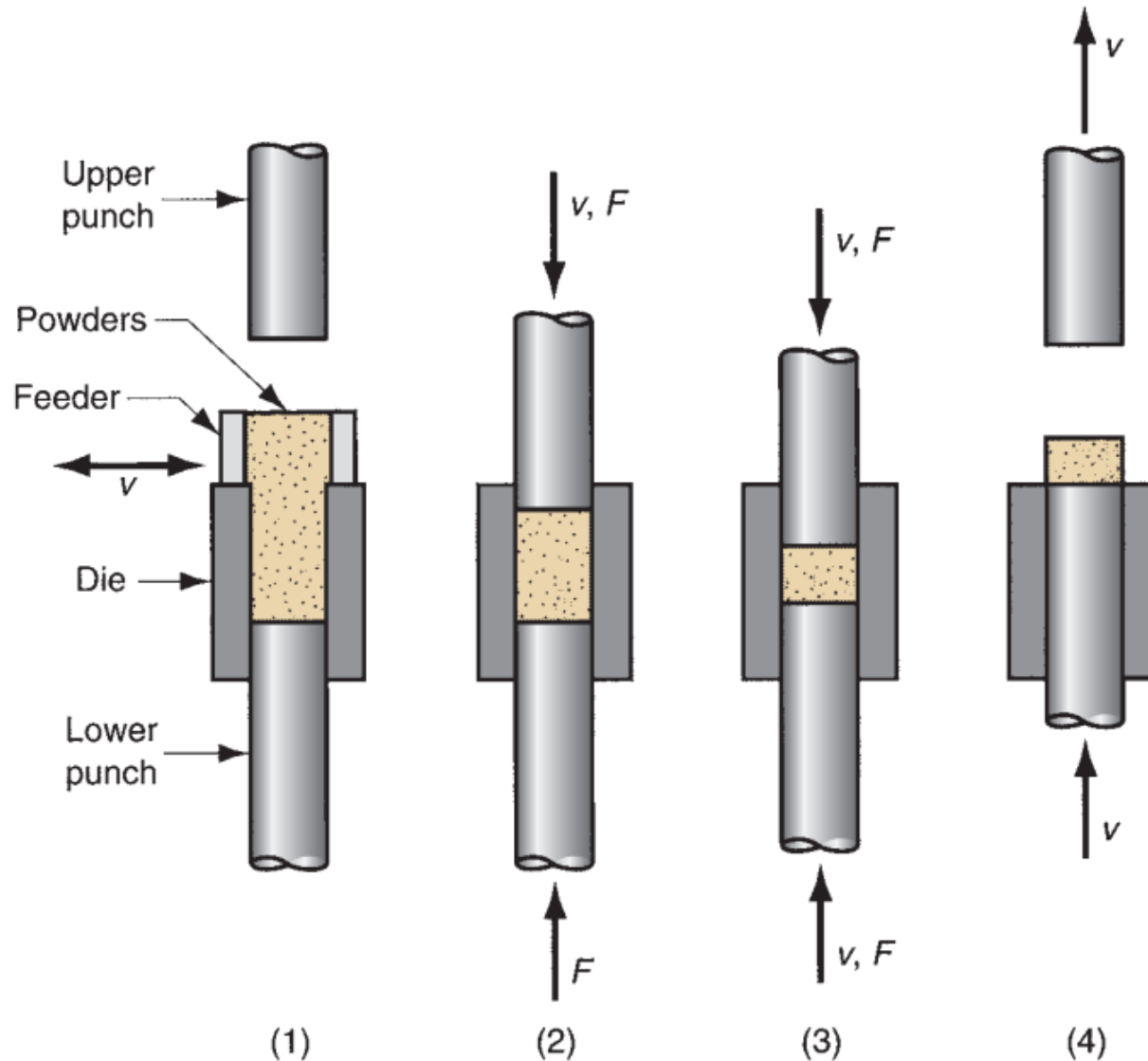


- Elemental powders can be deposited at the cathode of the electrolytic cell
- Raw metal is dissolved at the anode and deposited at the cathode
- After deposition, the cathode deposit is washed, dried, ground, screened, and annealed to form a powder
- Very high purity particulates are obtained
- Most common examples are palladium, chromium, copper, iron, zinc, manganese, and silver





# Powder Compaction (Dry Pressing)



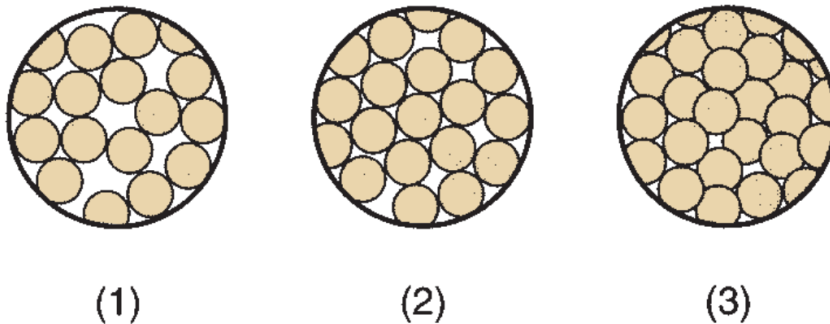
Green compact

Green strength



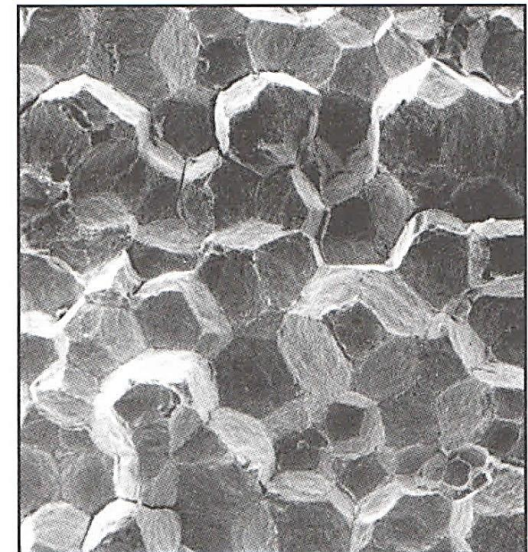
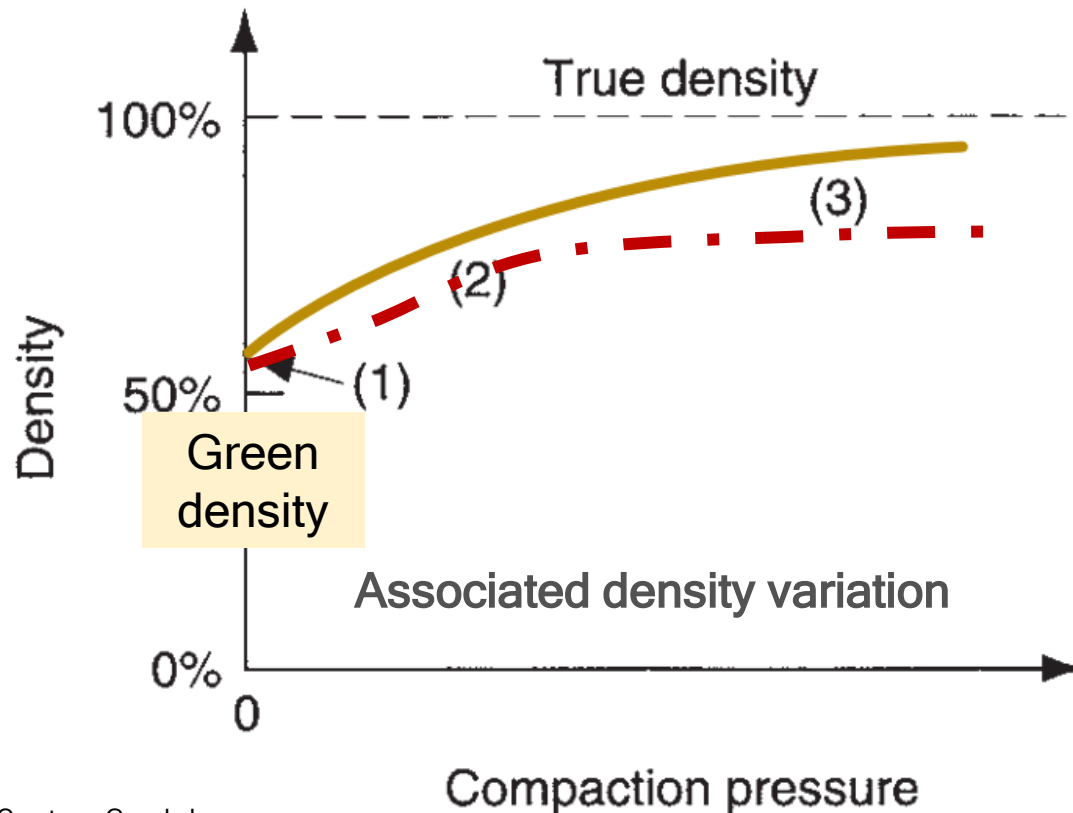


# Compaction: Stages and density of green compact



1. Loose compact after filling
2. Rearrangement
3. Particle deformation and reduction in pores

[How will the plot be different for ceramics?]

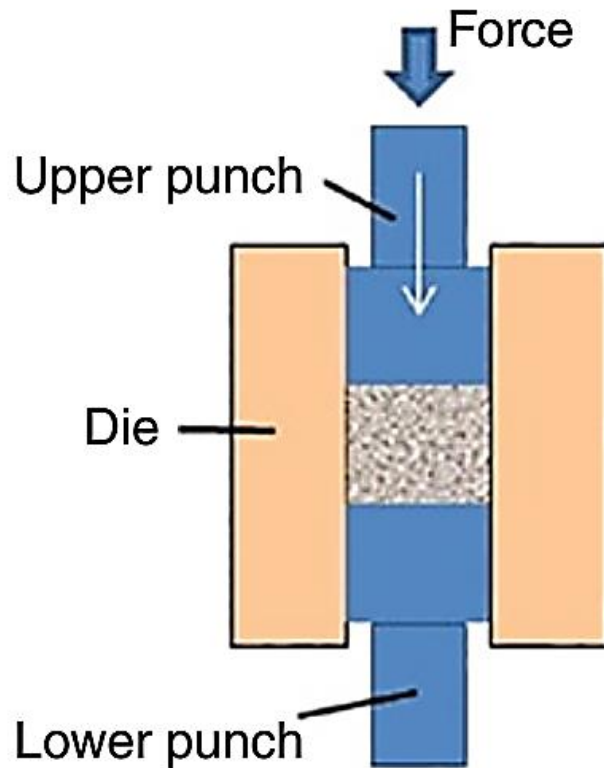


*Figure 7.6. This scanning electron micrograph shows the flattened shape of spherical particles subject to severe compaction. The spheres have been deformed into polygonal shapes with a coordination number approaching 14.*



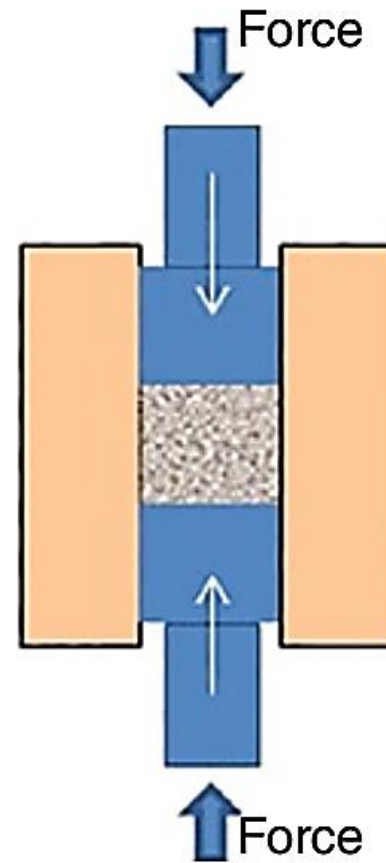
# Compaction: Action

## Single Action Compaction



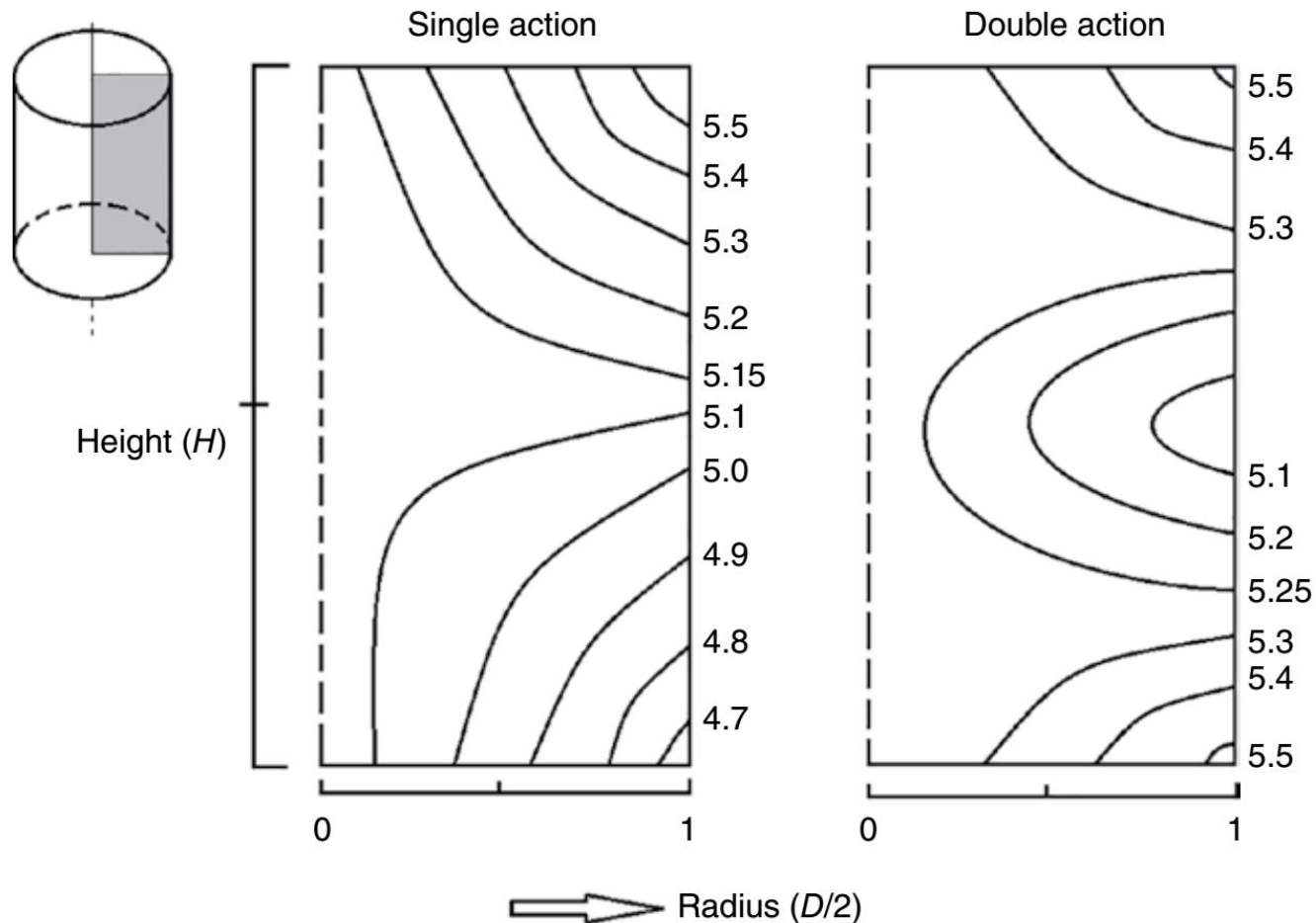
(a)

## Double Action Compaction



(b)

# Compaction: Die-wall friction

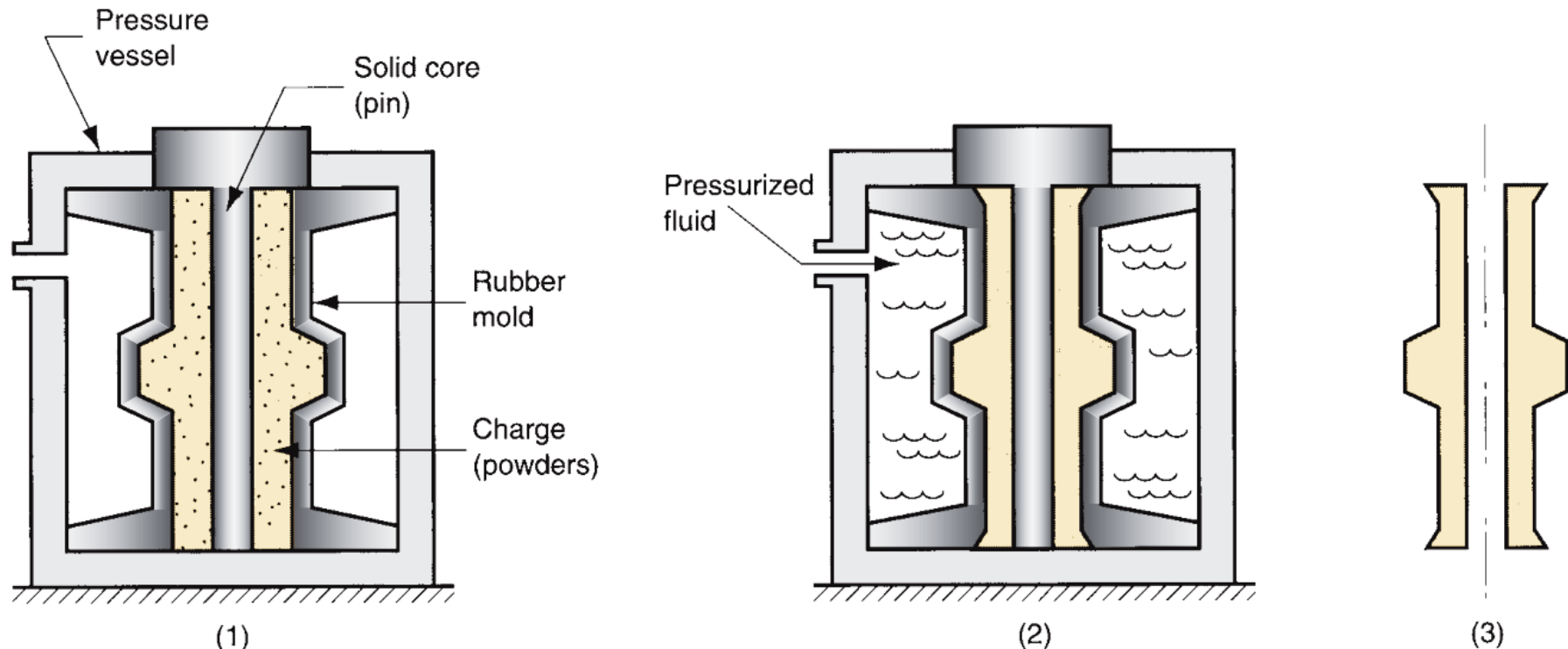


Constant density lines in cylinders of compacted copper powder.





# Compaction: Cold Isostatic Pressing

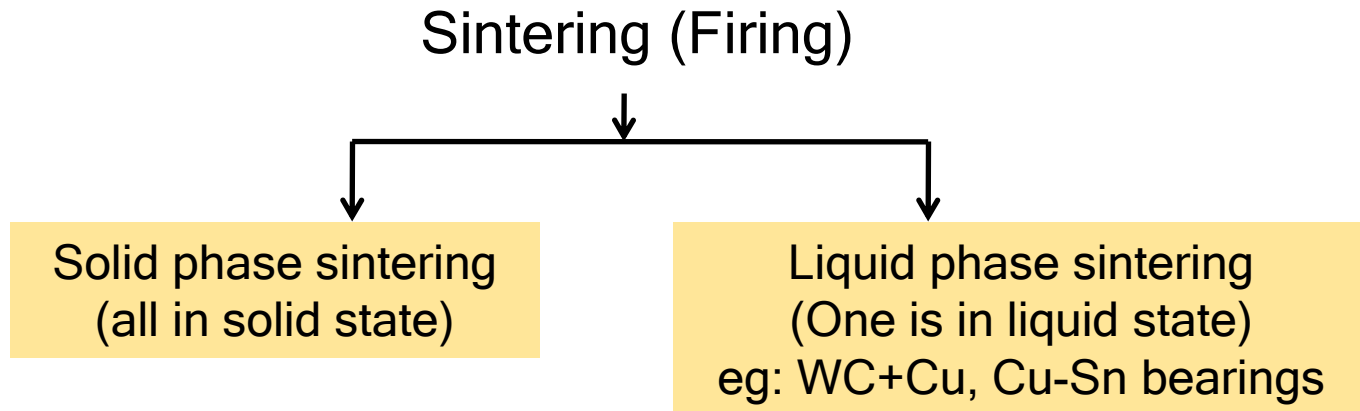


Cold isostatic pressing:

- (1) powders are placed in the flexible mold;
- (2) hydrostatic pressure is applied against the mold to compact the powders; and
- (3) pressure is reduced, and the part is removed.



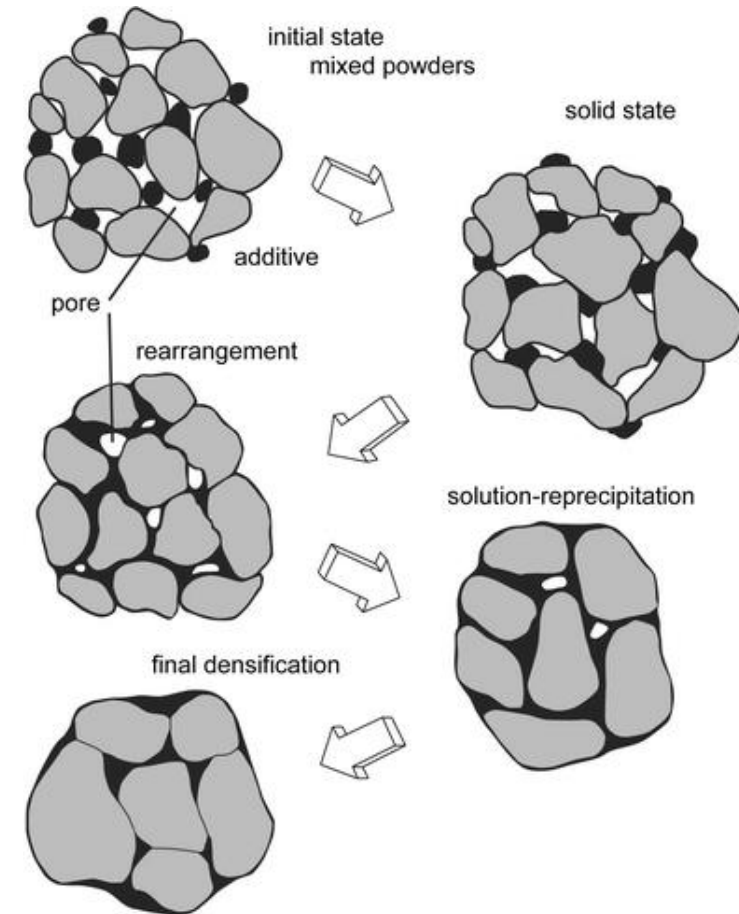
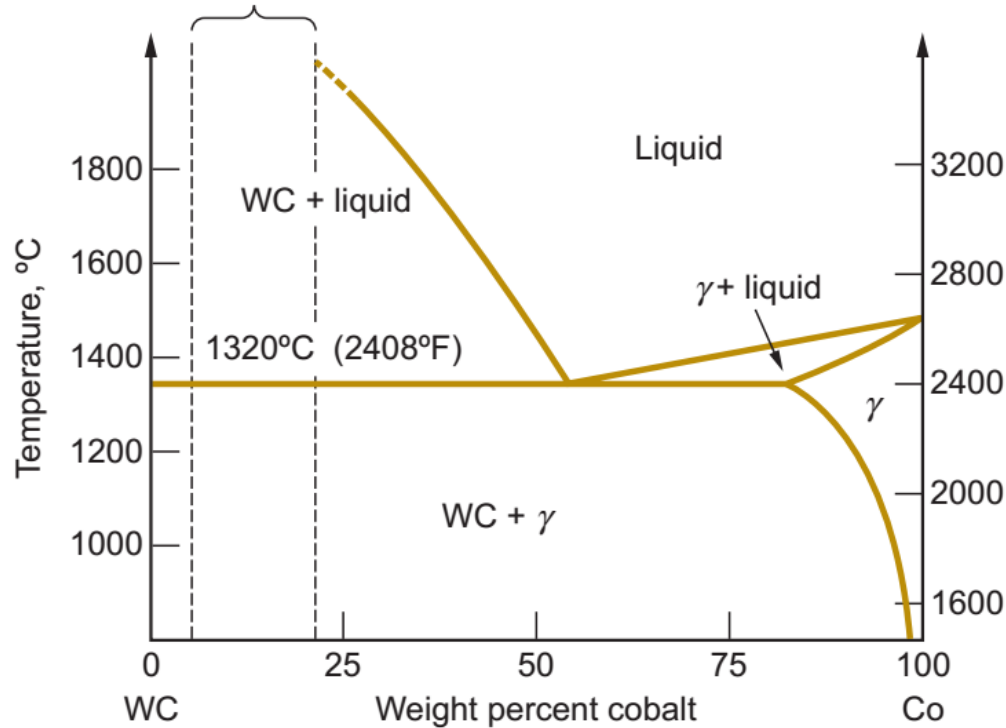
# Sintering



- It is an heat treatment process for attaining strength and density in the green compact
- Sintering - Green compacts are heated in a furnace to a temperature below melting point ( $0.7-0.9T_m$ )
- Improves the strength of the material
- Proper furnace control is important for optimum properties
- Part shrinkage occurs during sintering due to pore size reduction

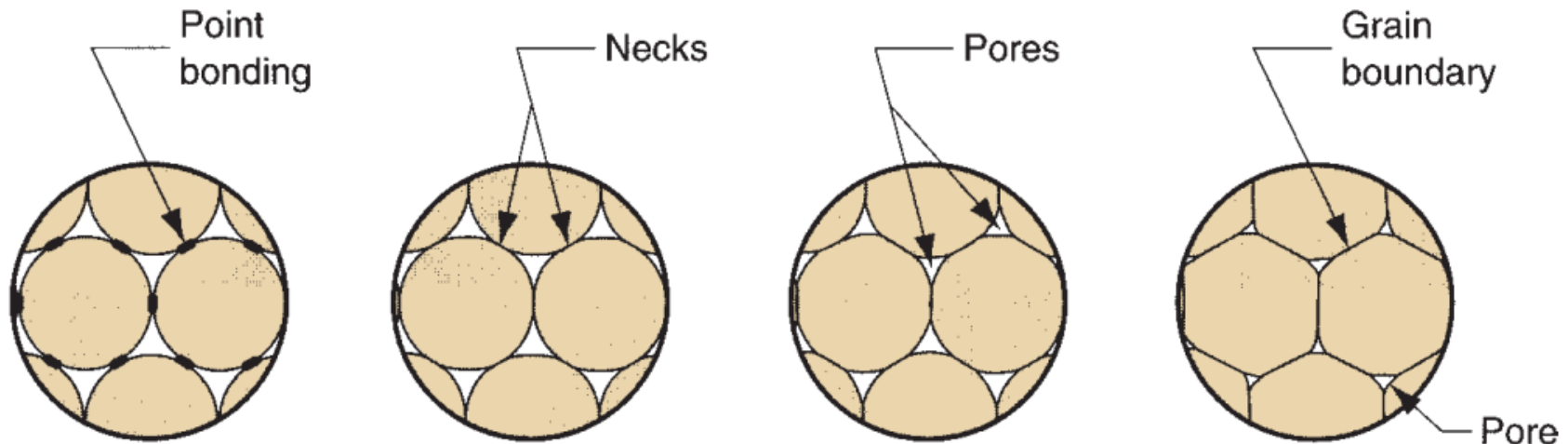
# Liquid Phase sintering

Typical composition range of cemented carbide products





# Solid State Sintering: Stages



(1)

Particle bonding is initiated at contact points

(2)

Contact points grow into necks

(3)

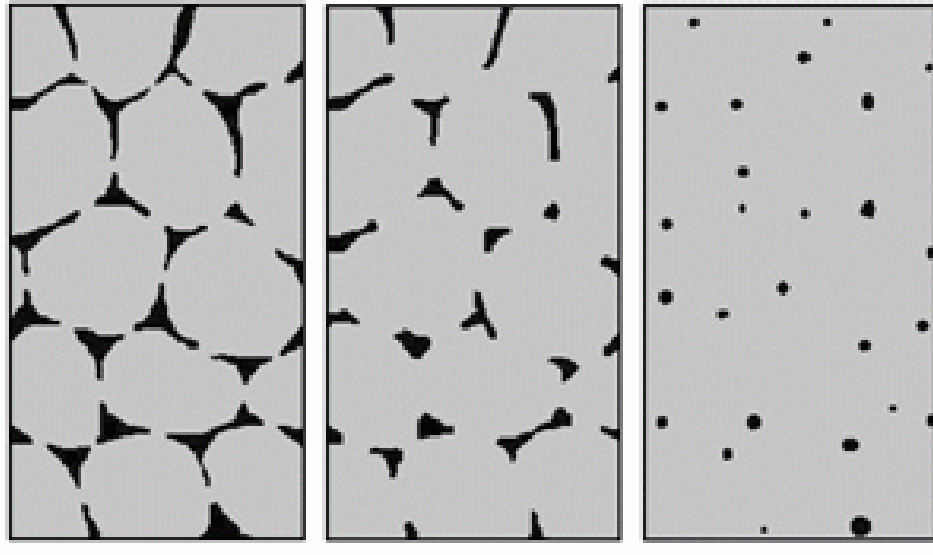
Reduction in pore size

(4)

Grain boundaries develop between particle



# Solid State Sintering: Stages



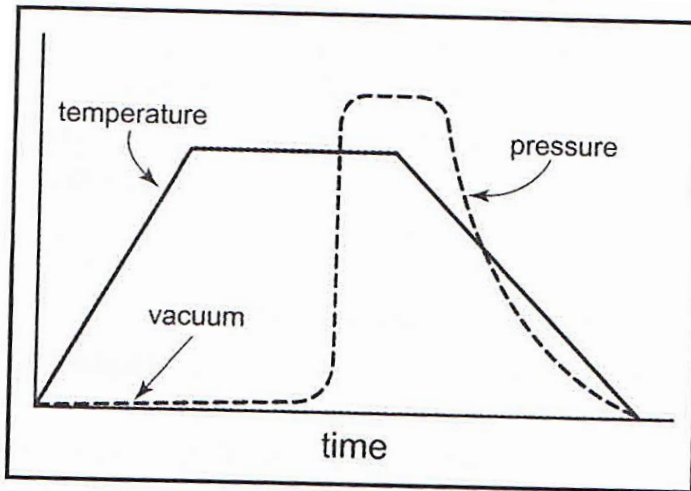
1. **Initial stage:**  
Formation of inter-particle neck
2. **Intermediate stage:**  
Transition occurs from open porosity to closed porosity. Typically, when the overall porosity in the compact is less than 8%, the pores are predominantly closed type
3. **Final stage:**  
Reduction/ elimination of closed pores.

Note that despite sintering at such high temperature, there is still some residual porosity

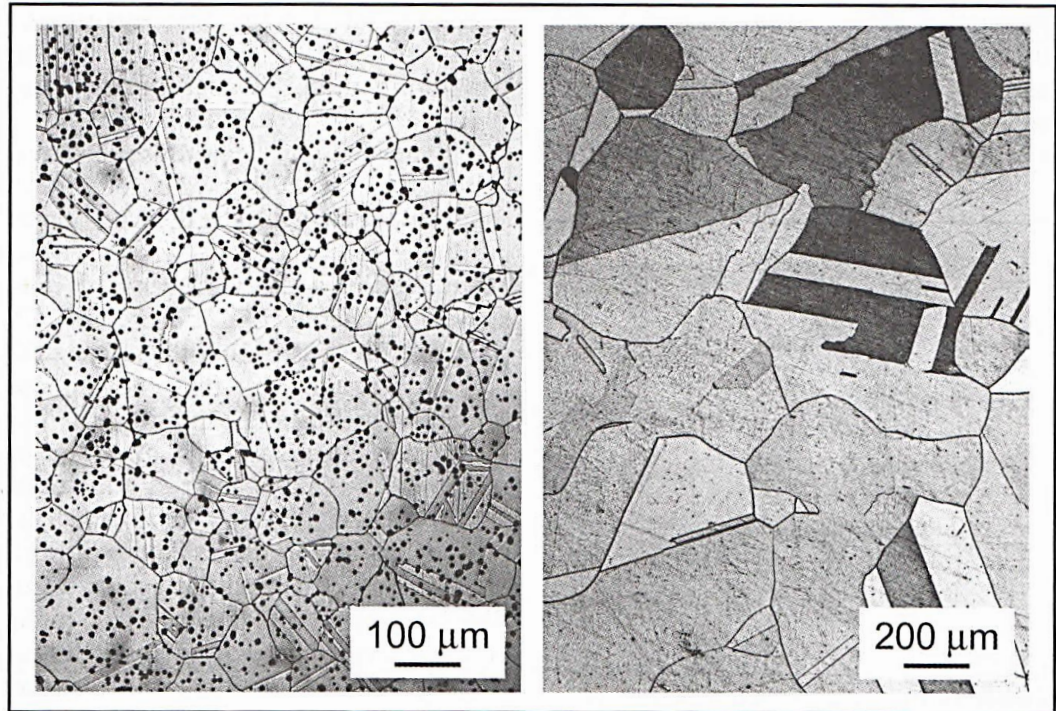




# Pressure-assisted sintering



**Figure 10.22.** Pressure-assisted sintering uses an initial vacuum sintering cycle to reach the closed pore condition, such that subsequent pressurization pushes the evacuated pores closed.

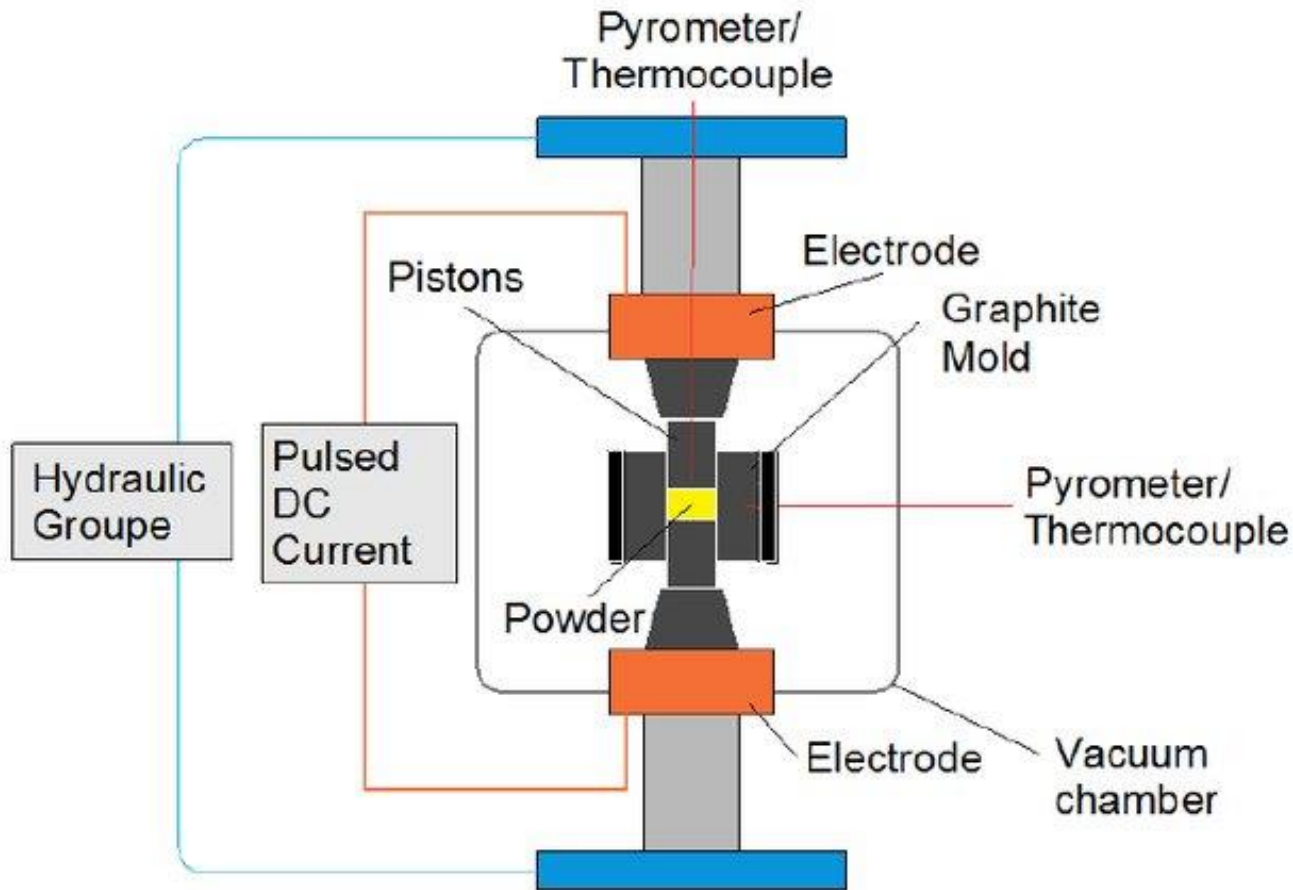


**Figure 10.23.** A nickel-iron powder compact sintered at 1410°C (2570°F) for 1 h to a closed pore condition at 95% density and then further densified by containerless hot isostatic pressing at 103 MPa (15 ksi) and 1200°C (2200°F) for 30 min with elimination of the pores and growth of the grains (micrographs courtesy of A. Bose and G. Camus).





# Spark Plasma Sintering



Field and Pressure Assisted Sintering Technique

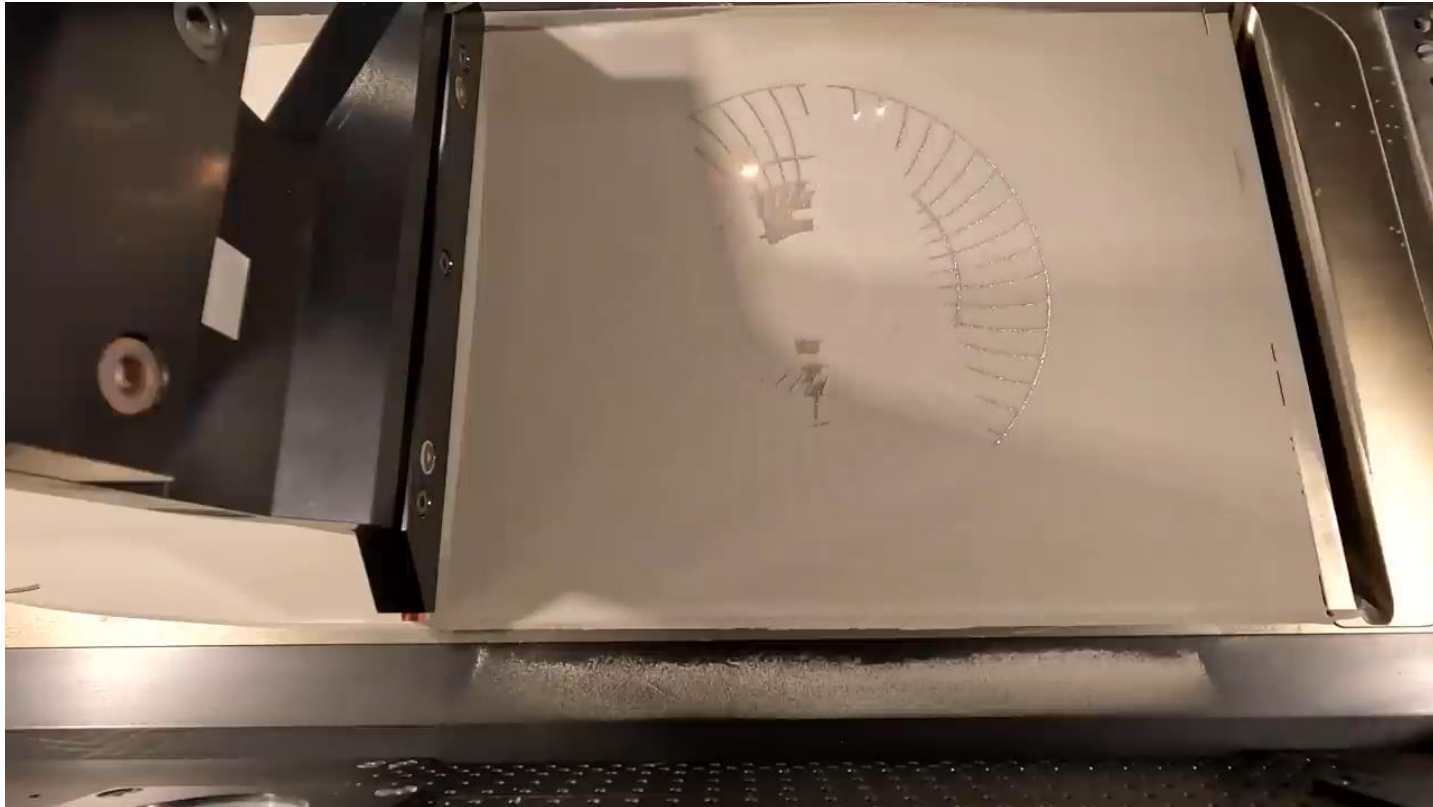


# Powder Metallurgy

- Design principles to consider
  - Shape of the compact must be simple and uniform
  - Bulk production must be met
  - Provision must be made for the ejection of the part
  - Wide tolerances should be used whenever possible
- Limitations
  - High cost of equipment
  - Tooling cost for short production runs
  - Limitations on part size and shape
  - Mechanical properties of the part
    - Strength
    - Ductility



# Freeform Fabrication: 3-D printing



<https://www.youtube.com/watch?v=xM36RpcgcQc>