



TA201A

Manufacturing Processes

Week-4

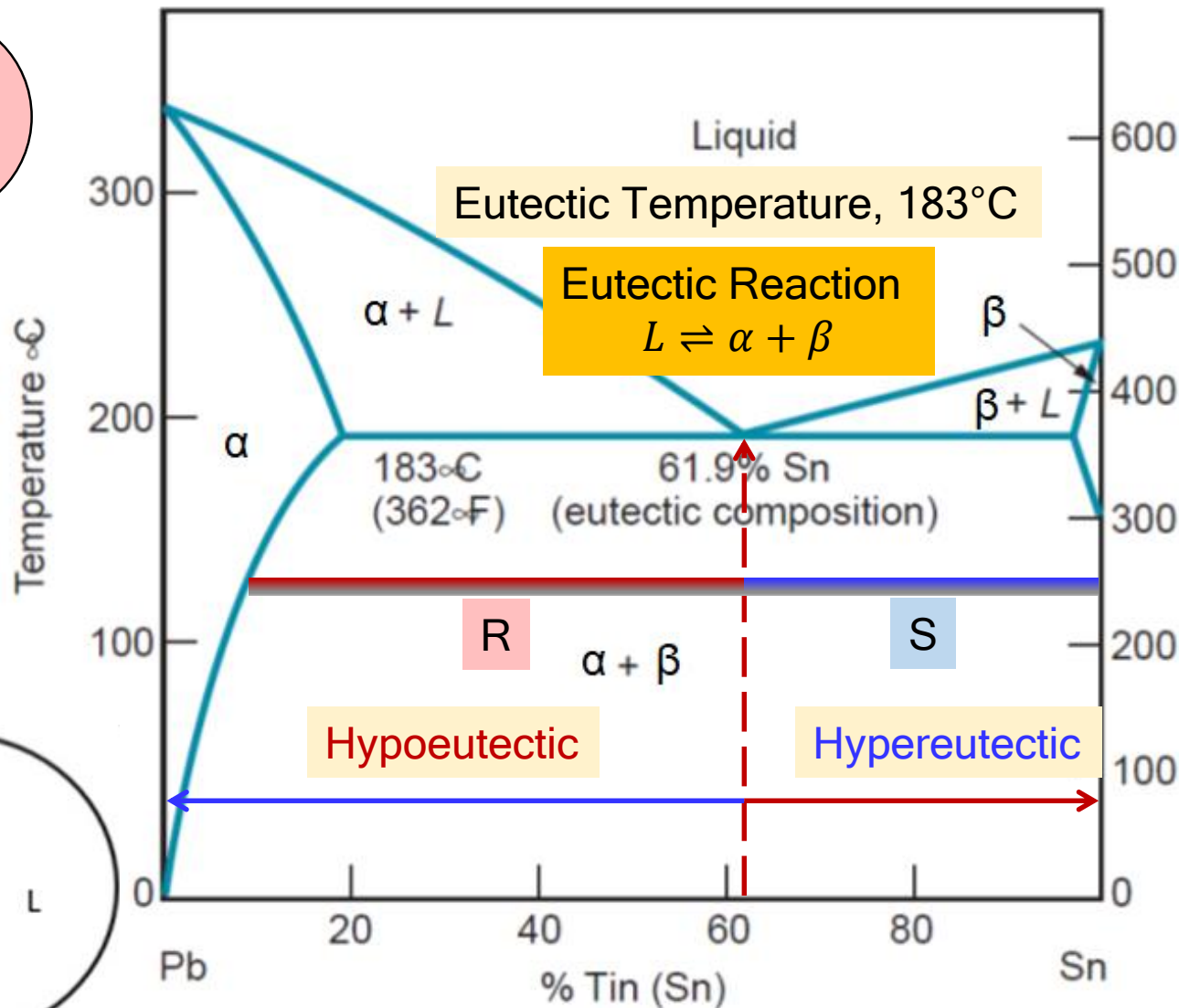
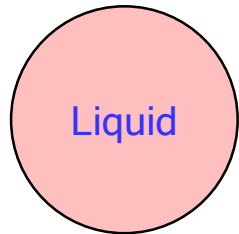
30 Aug, 2022

2022-2023 Semester-I

Lecture 4

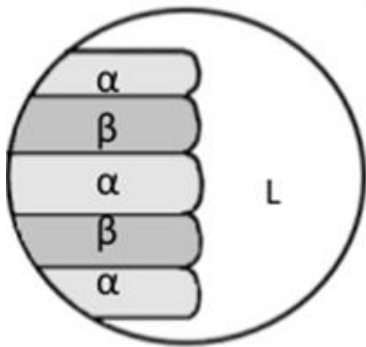


Eutectic Phase Diagram: Pb-Sn system



$$W_{\alpha} = \frac{S}{R + S}$$

$$W_{\beta} = \frac{R}{R + S}$$

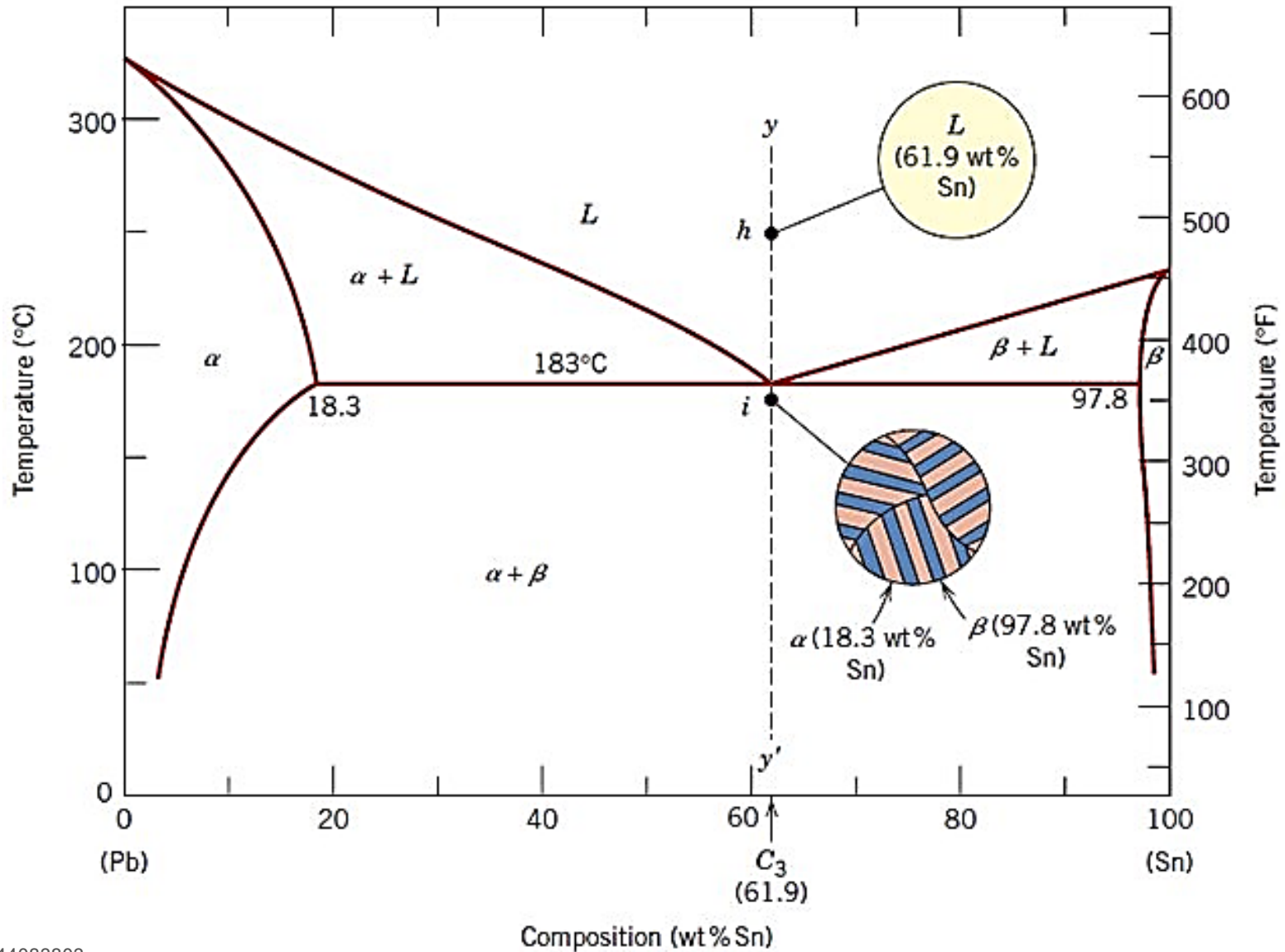


Eutectic $\alpha + \beta$

•ISBN: 978-1544083803

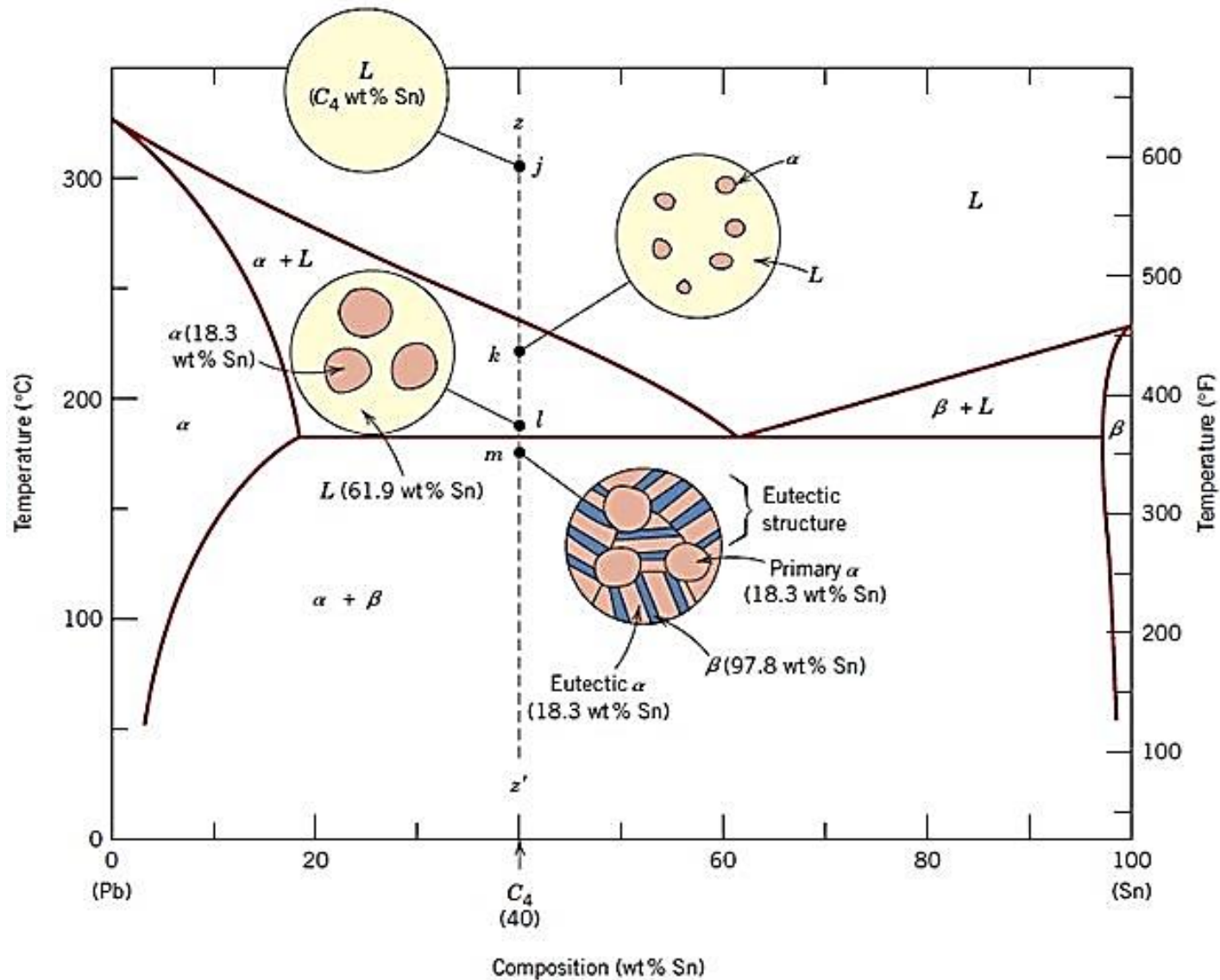


Solidification at Eutectic Composition



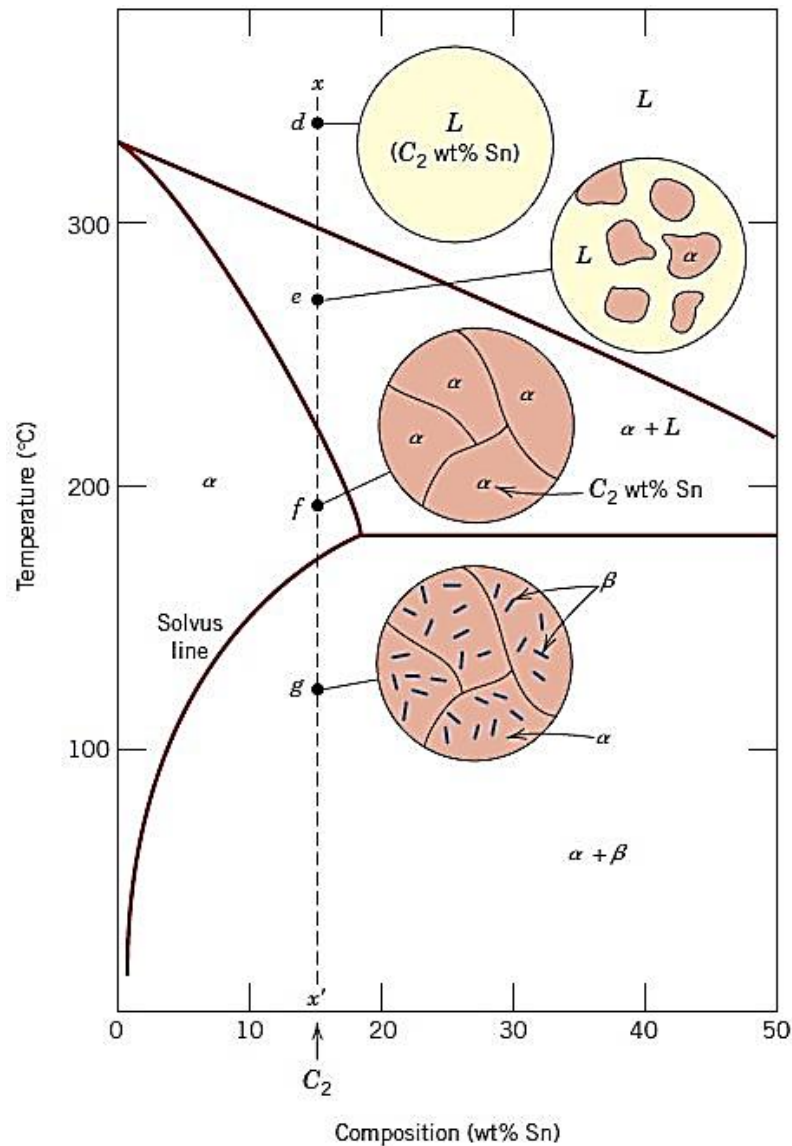


Solidification at Proeutectic Composition





Solidification at Proeutectic Composition with single phase region



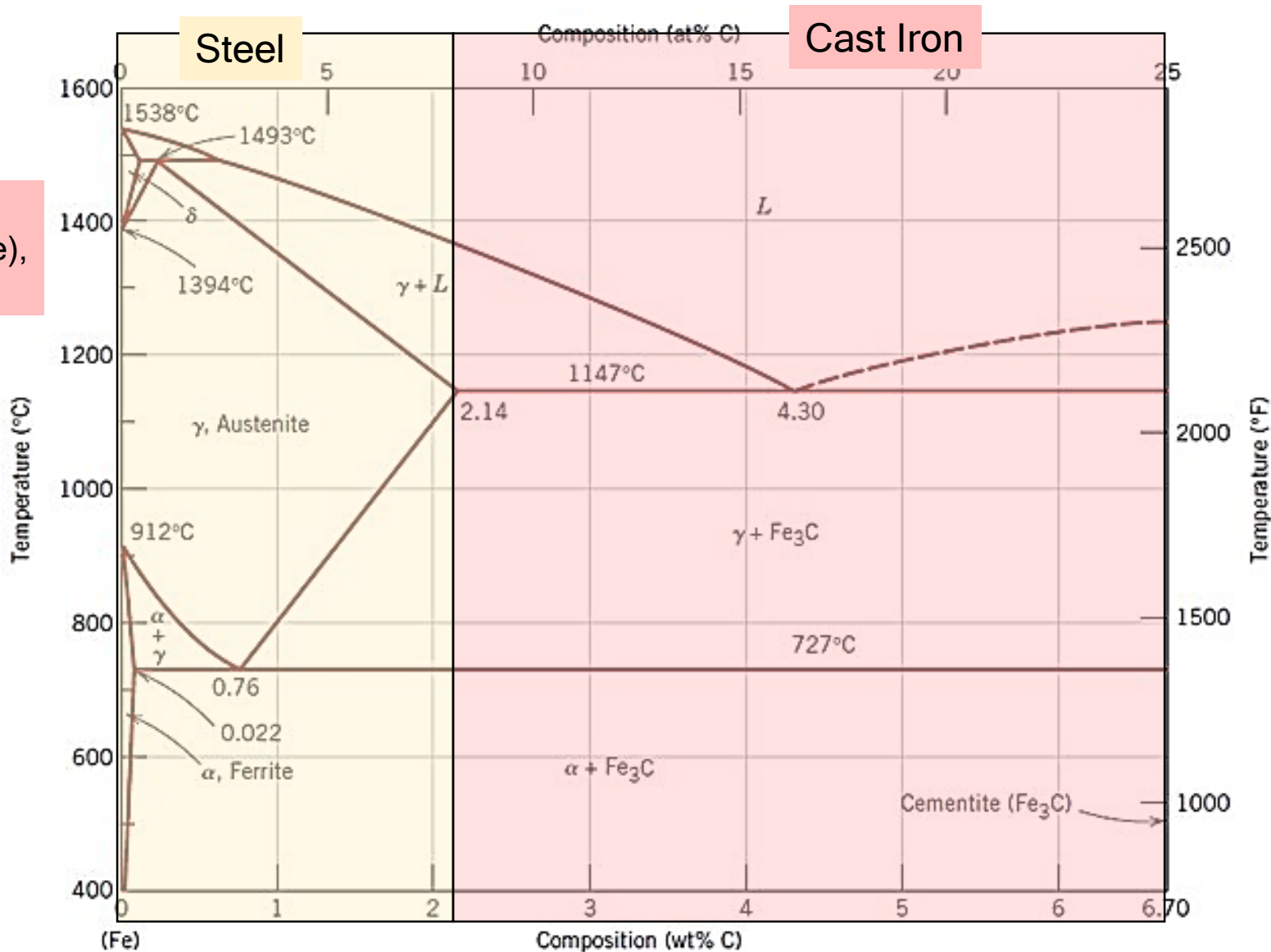


Fe-Fe₃C phase Diagram| Iron-Iron Carbide phase diagram

Allotropes

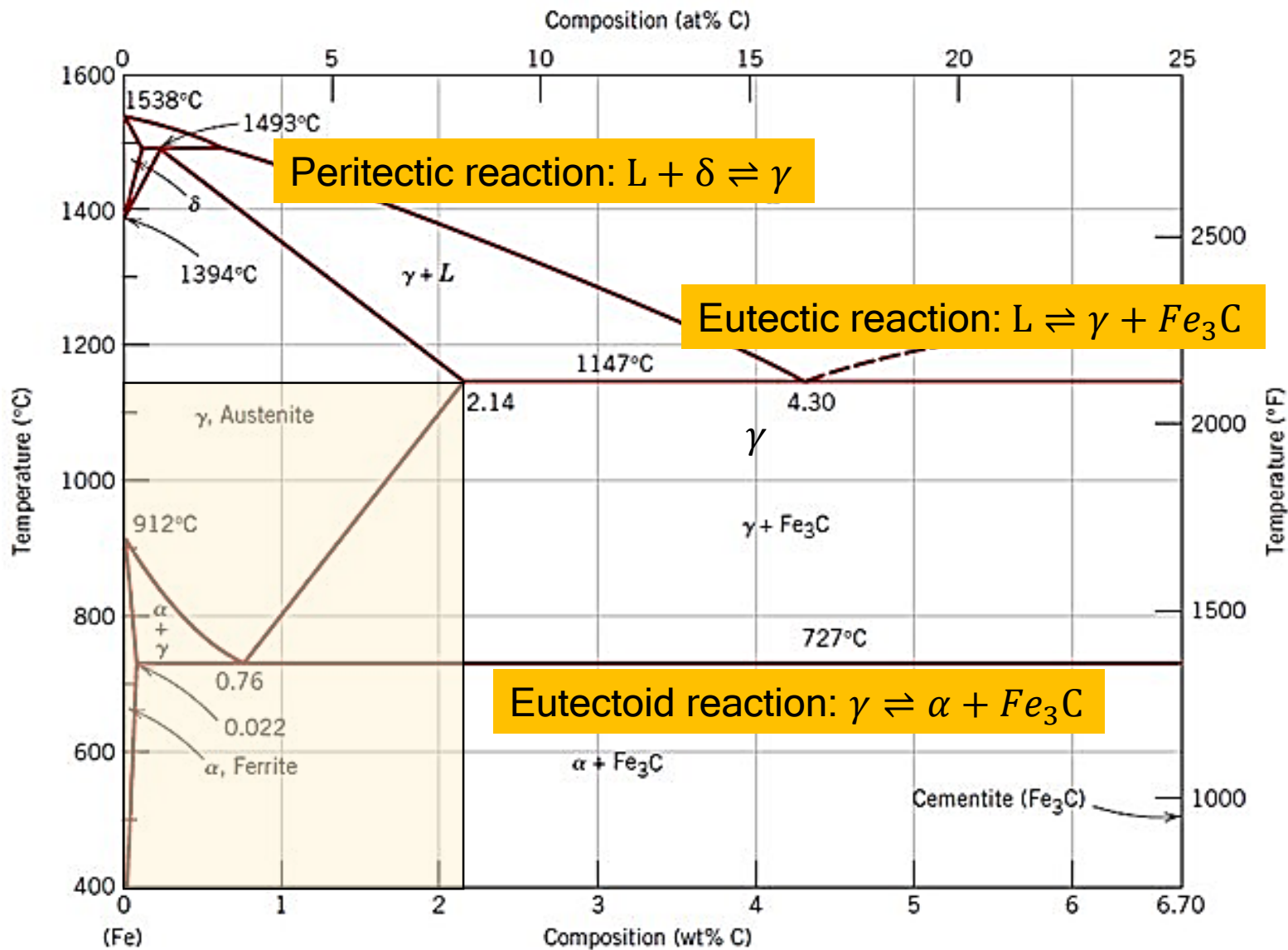
alpha iron (α -Fe):
gamma iron (γ -Fe),
delta iron (δ -Fe)

α -Fe: bcc
 γ -Fe: fcc
 δ -Fe: bcc





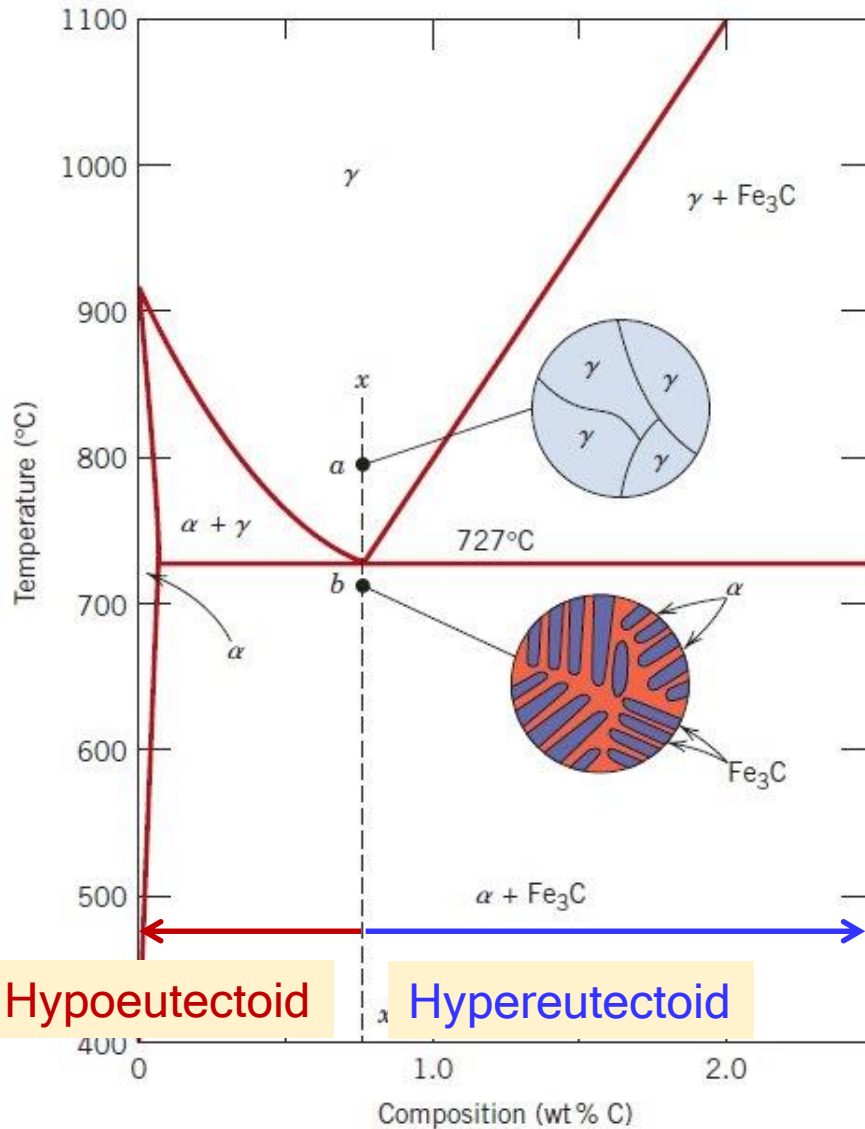
Fe-Fe₃C phase Diagram



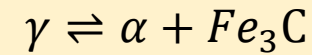
•ISBN: 978-1544083803



Eutectoid Steel

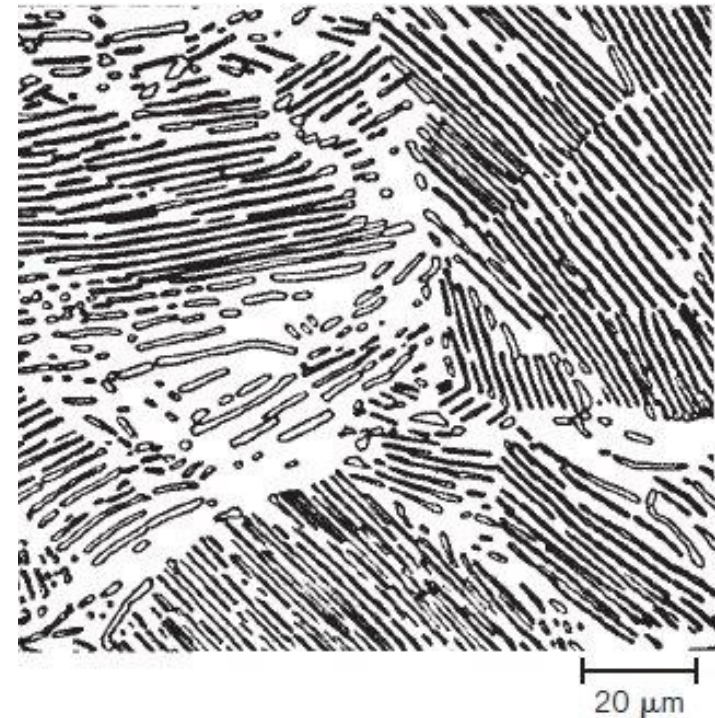


Eutectoid reaction



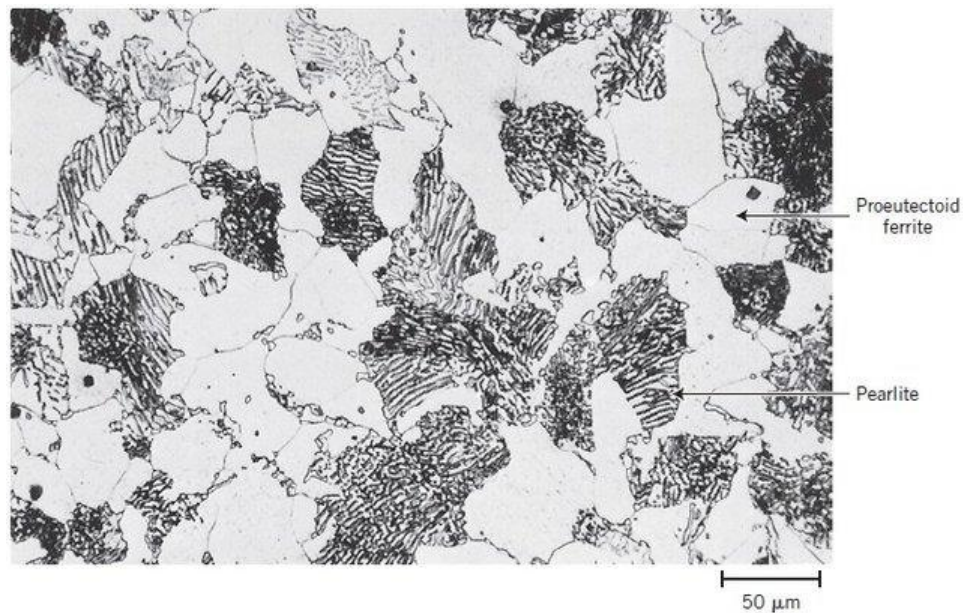
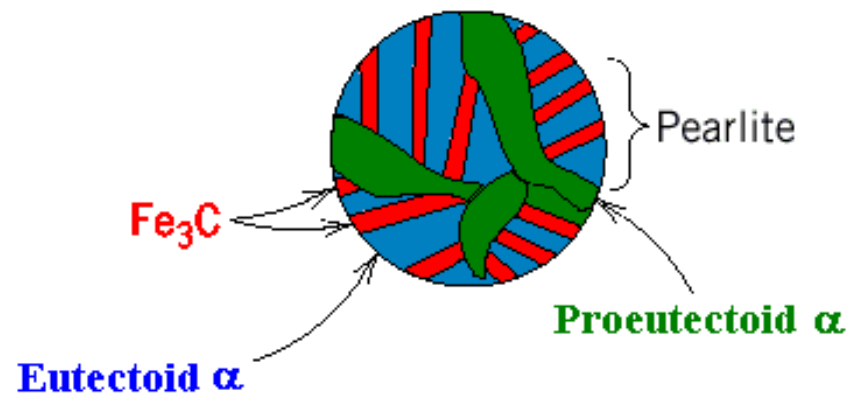
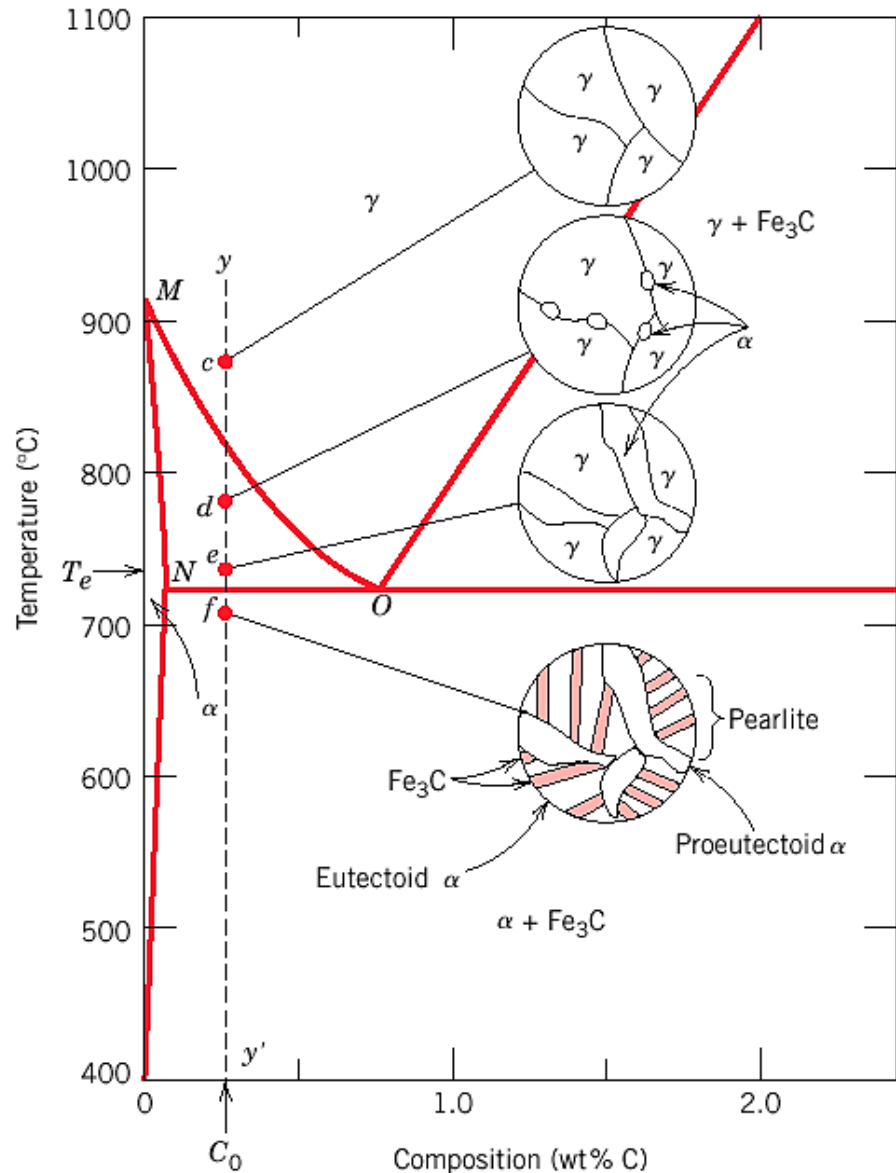
$T = 727^\circ\text{C}$

Composition = 0.76 wt.% C





Hypoeutectoid steel





Microstructure

- The microstructure of crystalline materials is defined by the **type, structure, number, shape and topological** arrangement of phases and/or lattice defects .
- **Elements of microstructure:** Point defects, point-defect clusters, dislocations, stacking faults, grain boundaries, interphase interfaces are important elements of the microstructure of most materials.



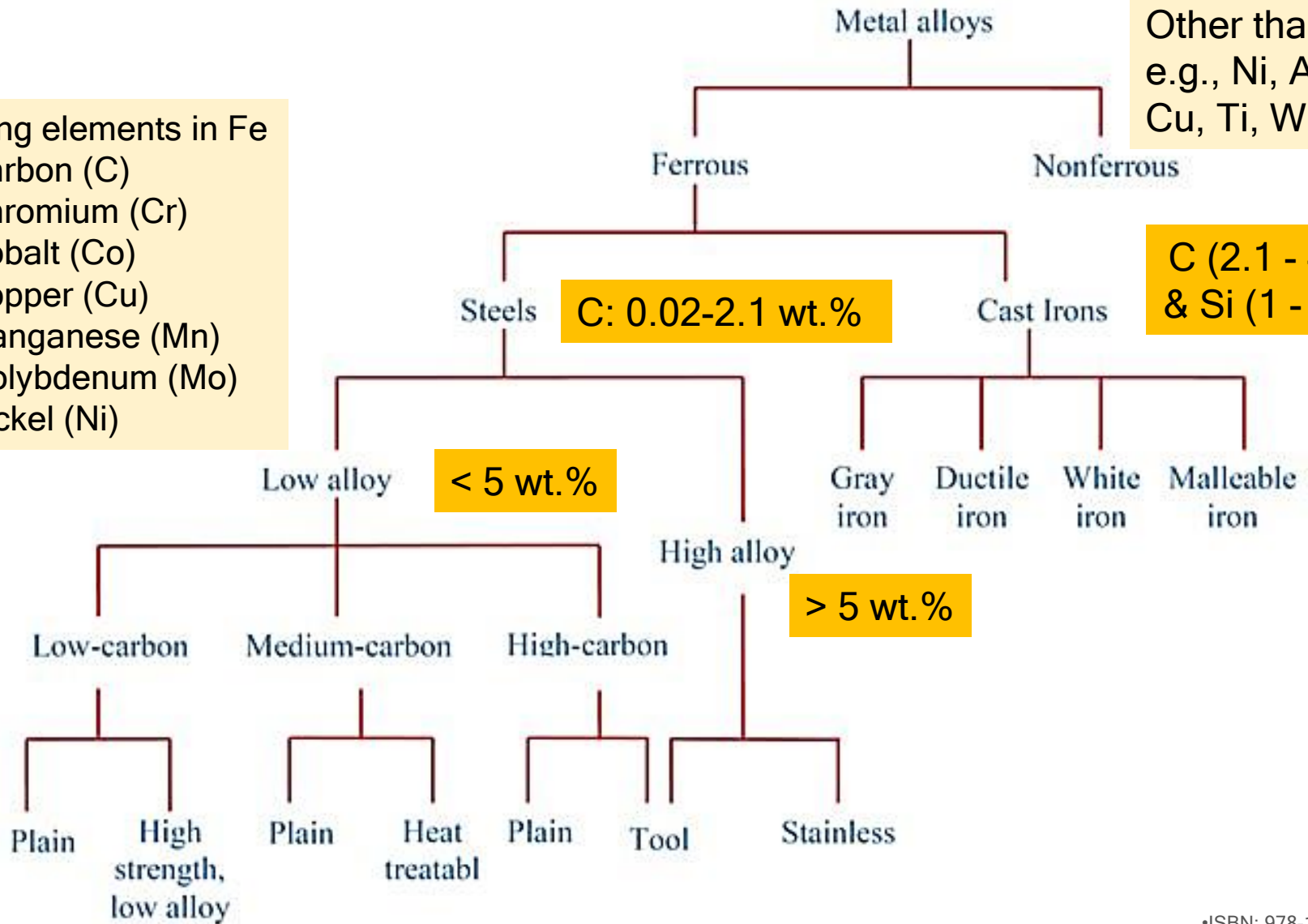
Classification of alloys

Alloying elements in Fe

- Carbon (C)
- Chromium (Cr)
- Cobalt (Co)
- Copper (Cu)
- Manganese (Mn)
- Molybdenum (Mo)
- Nickel (Ni)

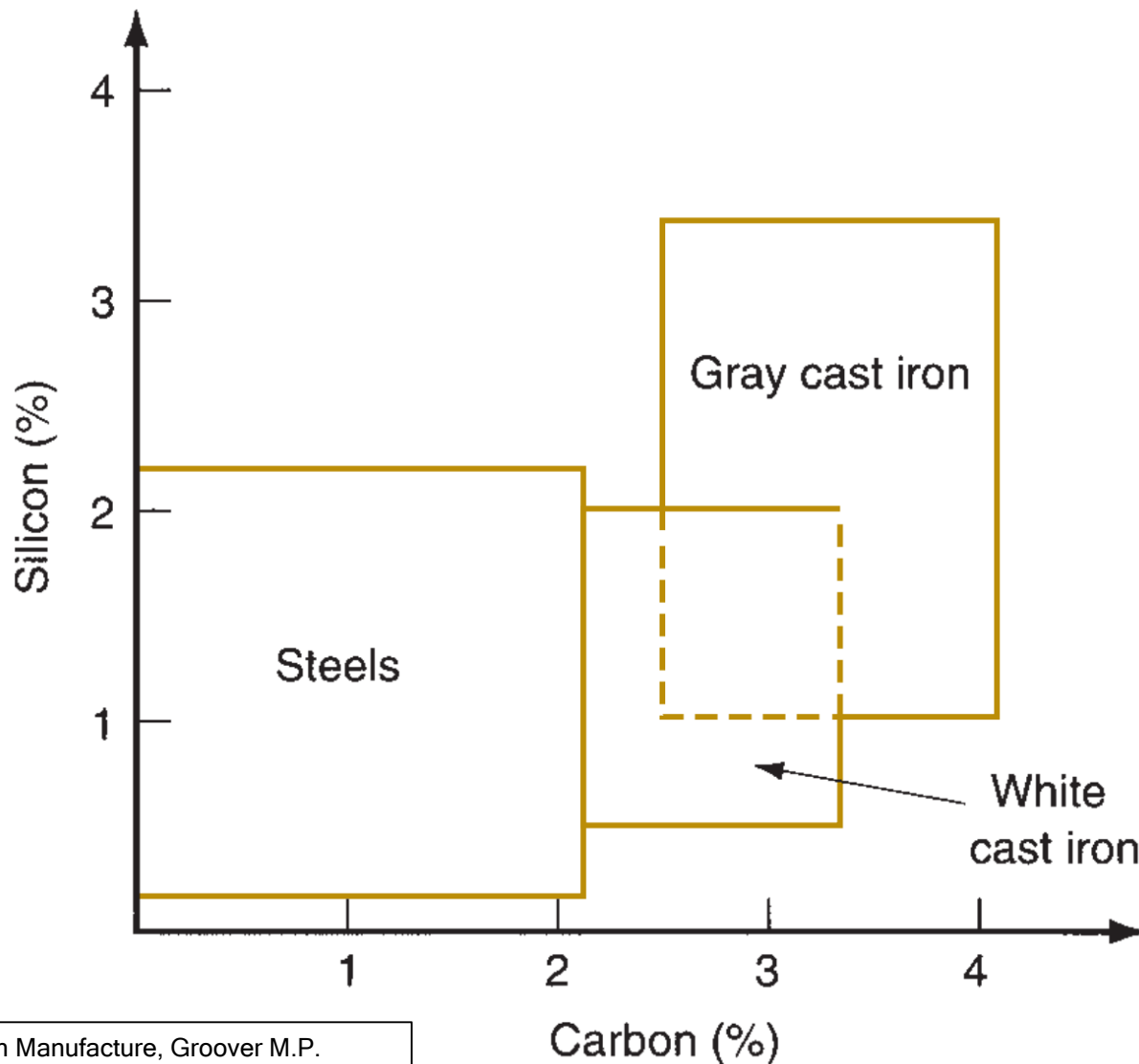
Other than Fe
e.g., Ni, Al,
Cu, Ti, W, Mg

C (2.1 - 4%)
& Si (1 - 3%)



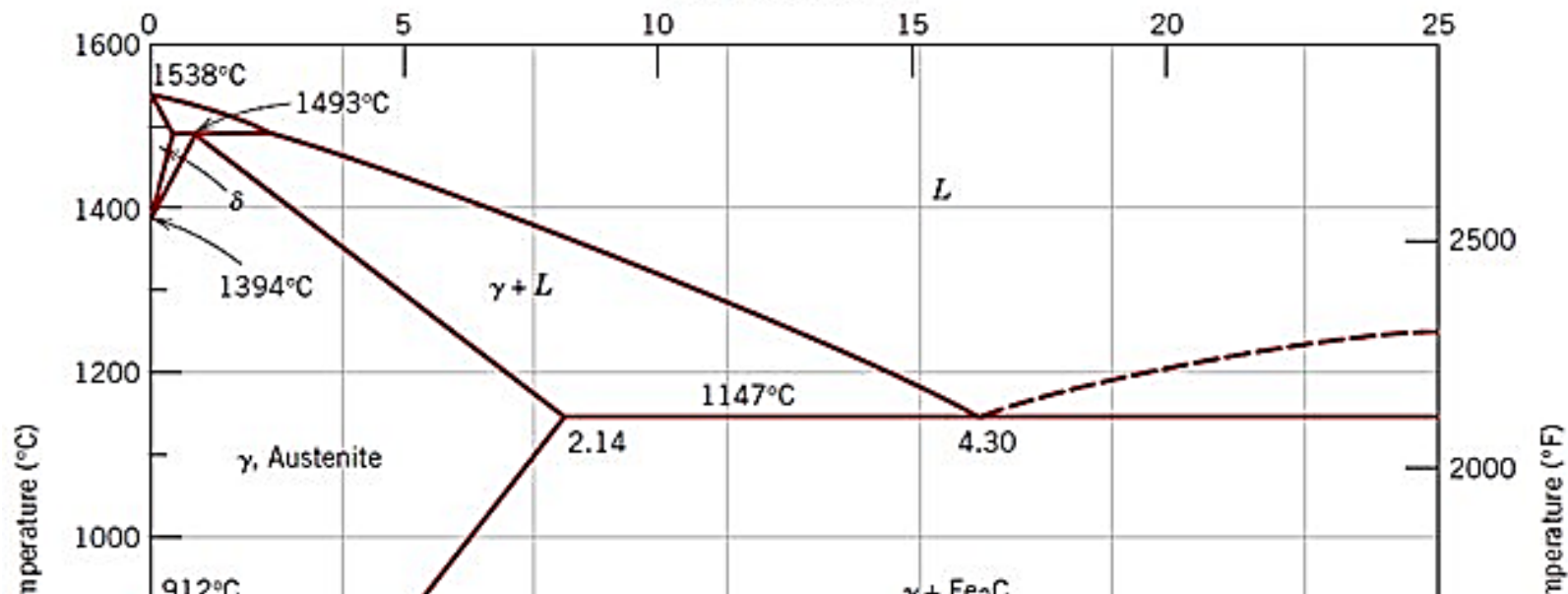


Role of C & Si in Fe

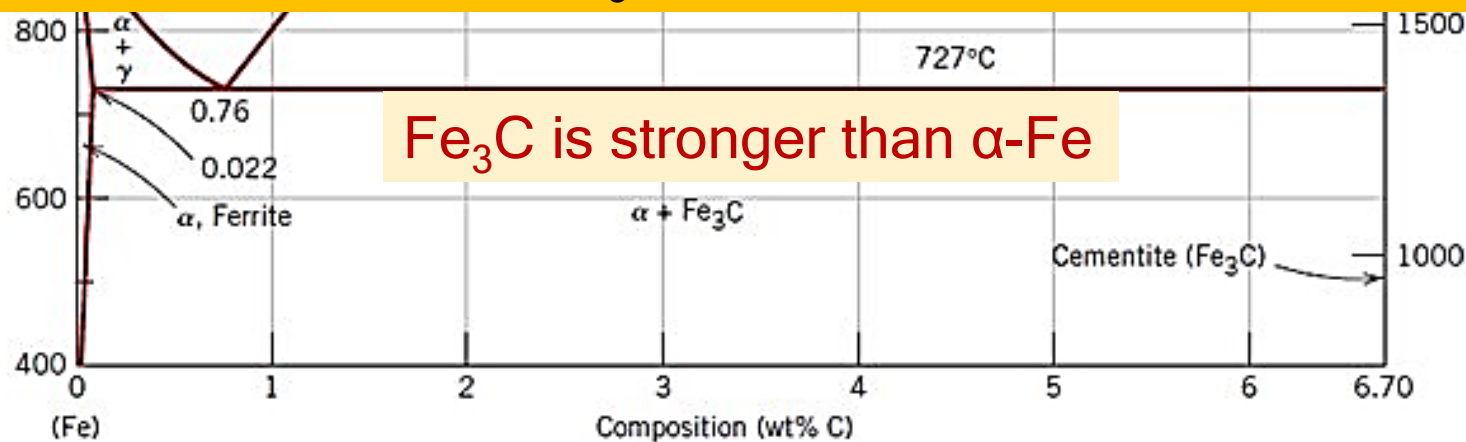




Fe-Fe₃C phase diagram: Revisited



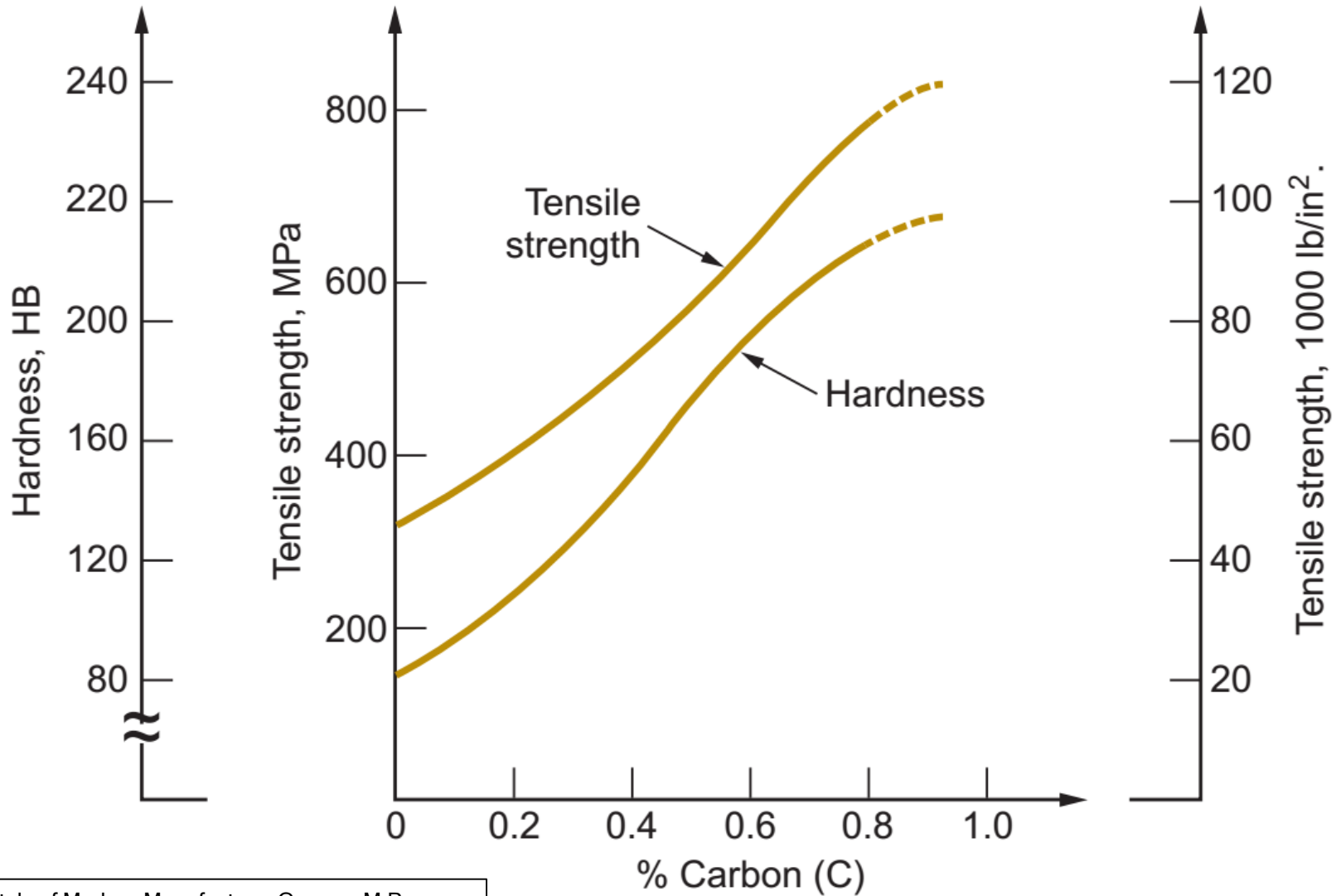
What is the proportion of Fe₃C with increase in C wt.% in steel??



Fe₃C is stronger than α -Fe

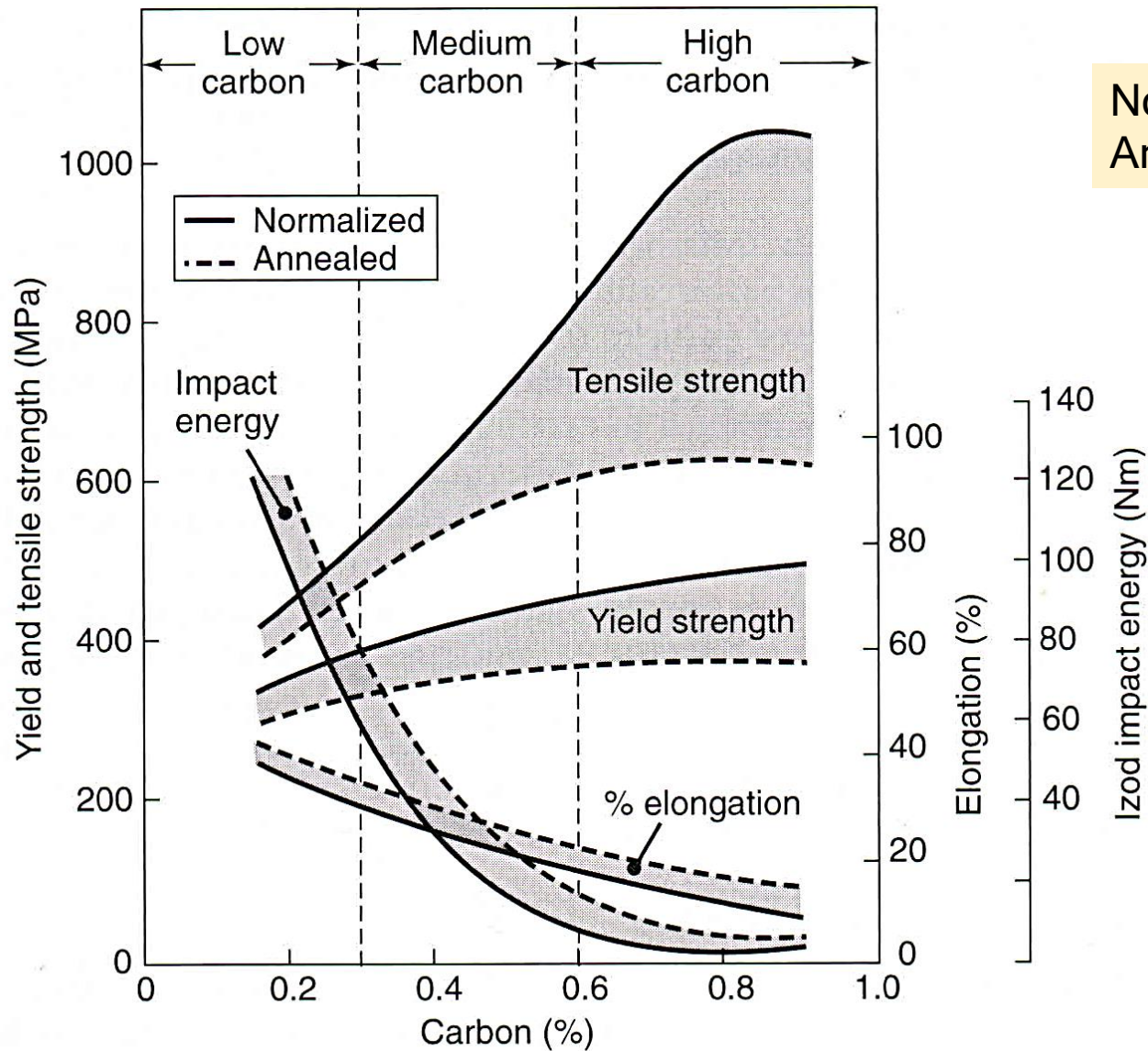


Properties: Role of C in steel





Properties: Role of C in steel





Strength: Low C Vs High C

Low C



Steel wire
Cross section area: 1 cm^2



3 - 4 max

High C

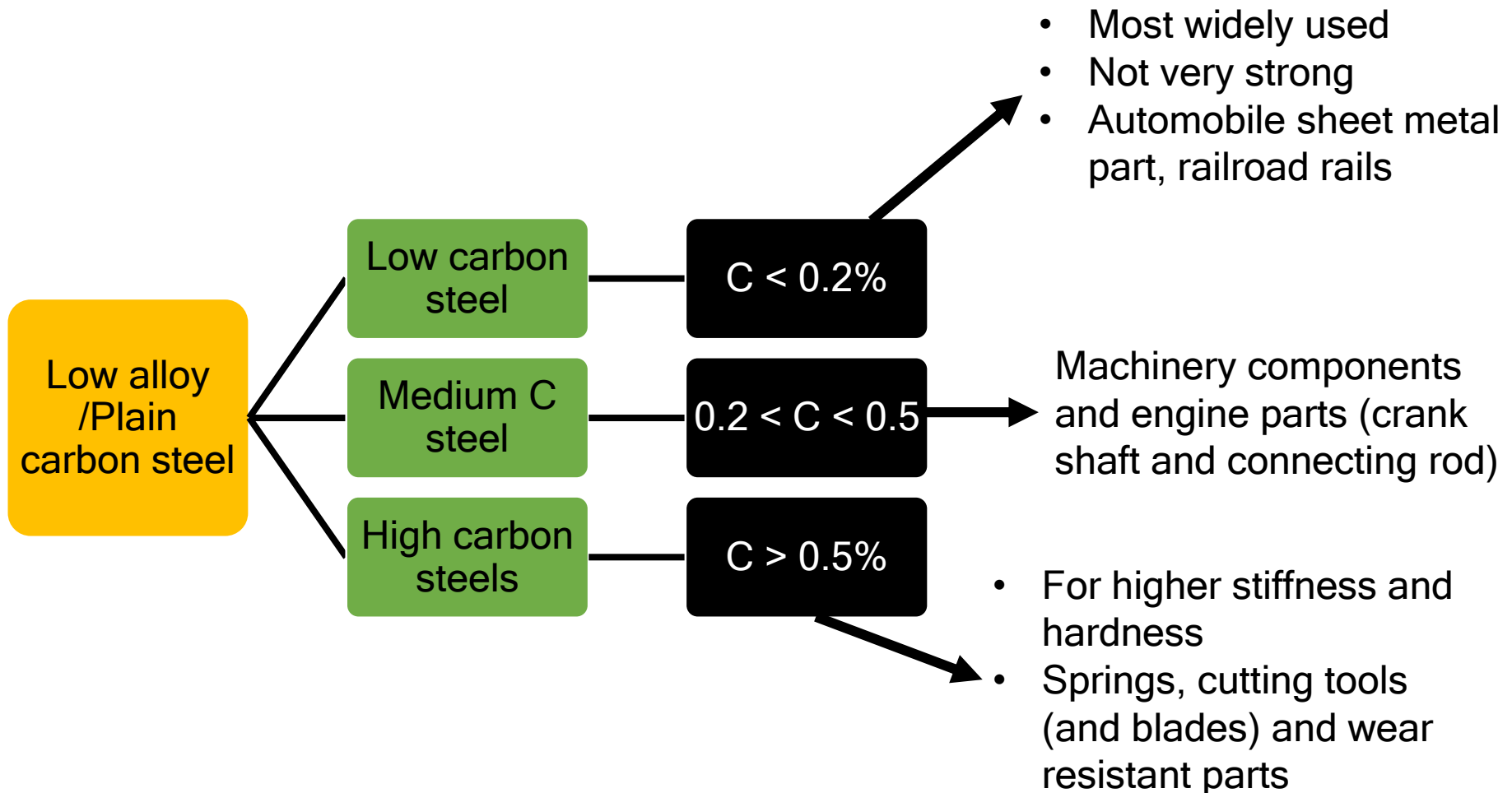


10 - 12



Low Alloy or Plain Carbon Steel

- Plain carbon steel
 - Principal alloying element C





Alloy steels

Low alloy steel

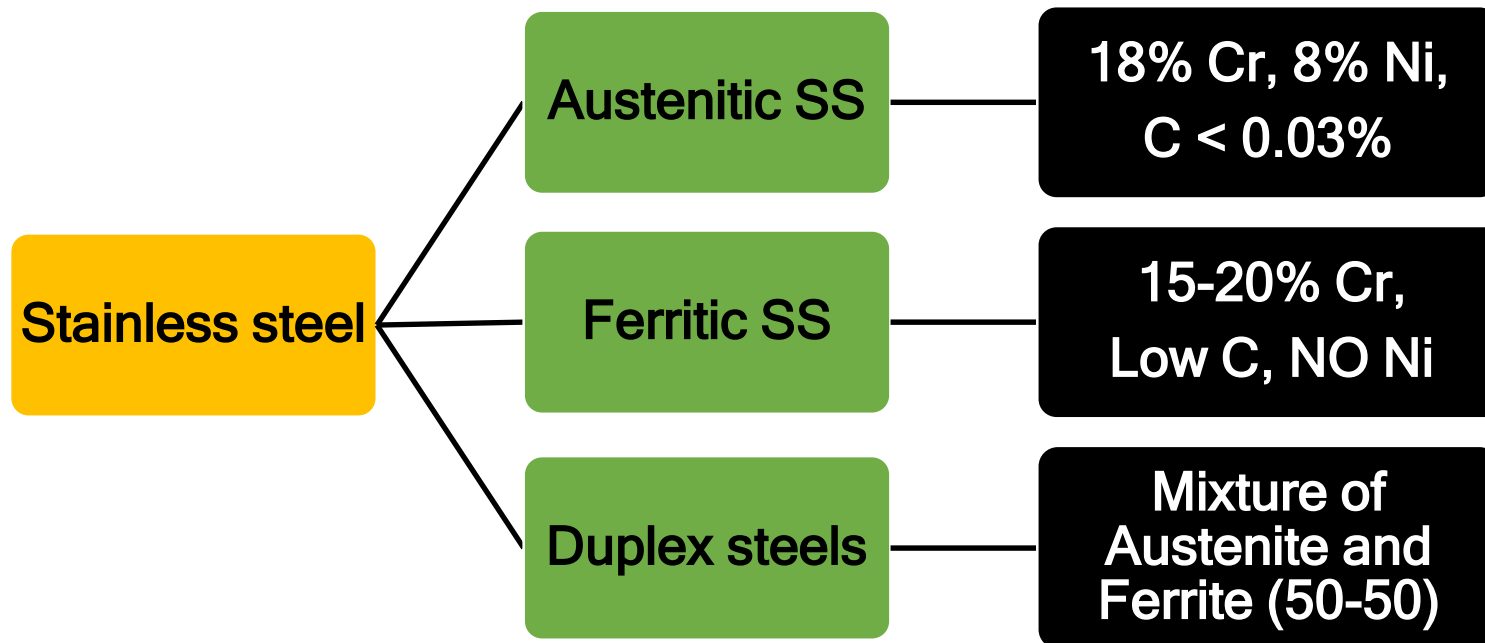
- Iron carbon alloys that contain additional alloying elements in amounts totaling less than 5% (wt.%)
- Better mechanical properties than plain C steel
 - Strength
 - Hardness
 - Wear resistance
 - Toughness
- Heat treatment is often required
- Alloying elements usually added in combination



Stainless steel

Stainless steels: Highly alloyed for high corrosion resistance

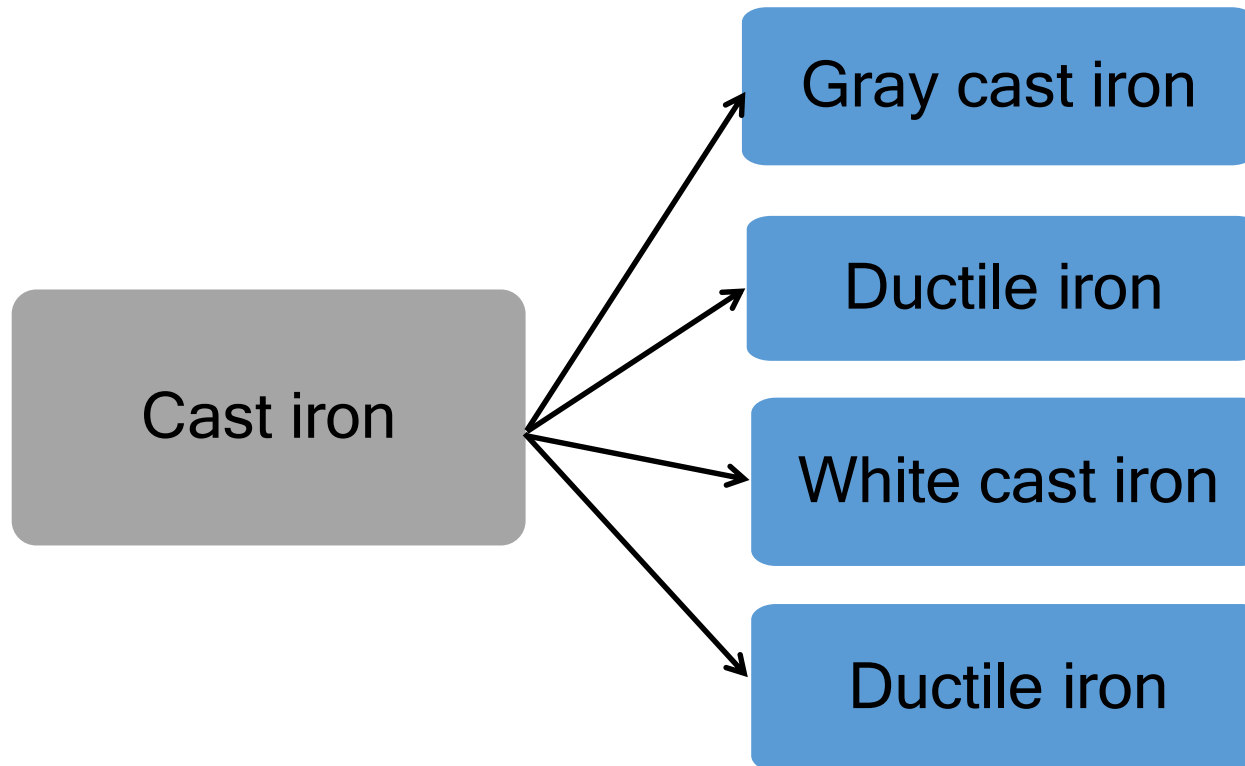
- High strength and ductility
- Cr > 15%





Cast iron

- Cast iron: alloy containing C (2.1 - ~ 4%) and Si (1 - 3%)





Microstructures & Applications of cast iron



Gray Cast Iron



Ductile Cast Iron



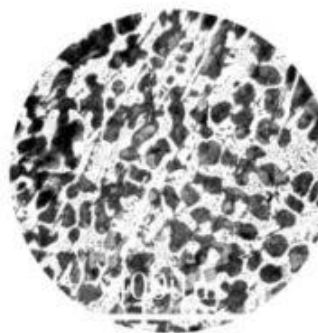
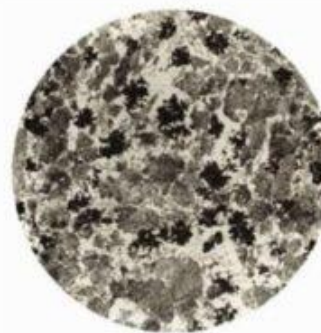
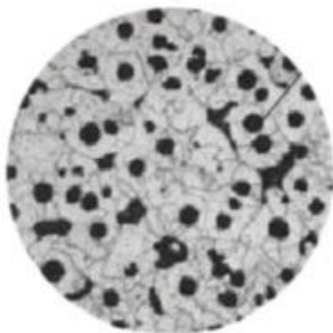
Malleable Cast Iron



White Cast Iron



Alloy Cast Iron



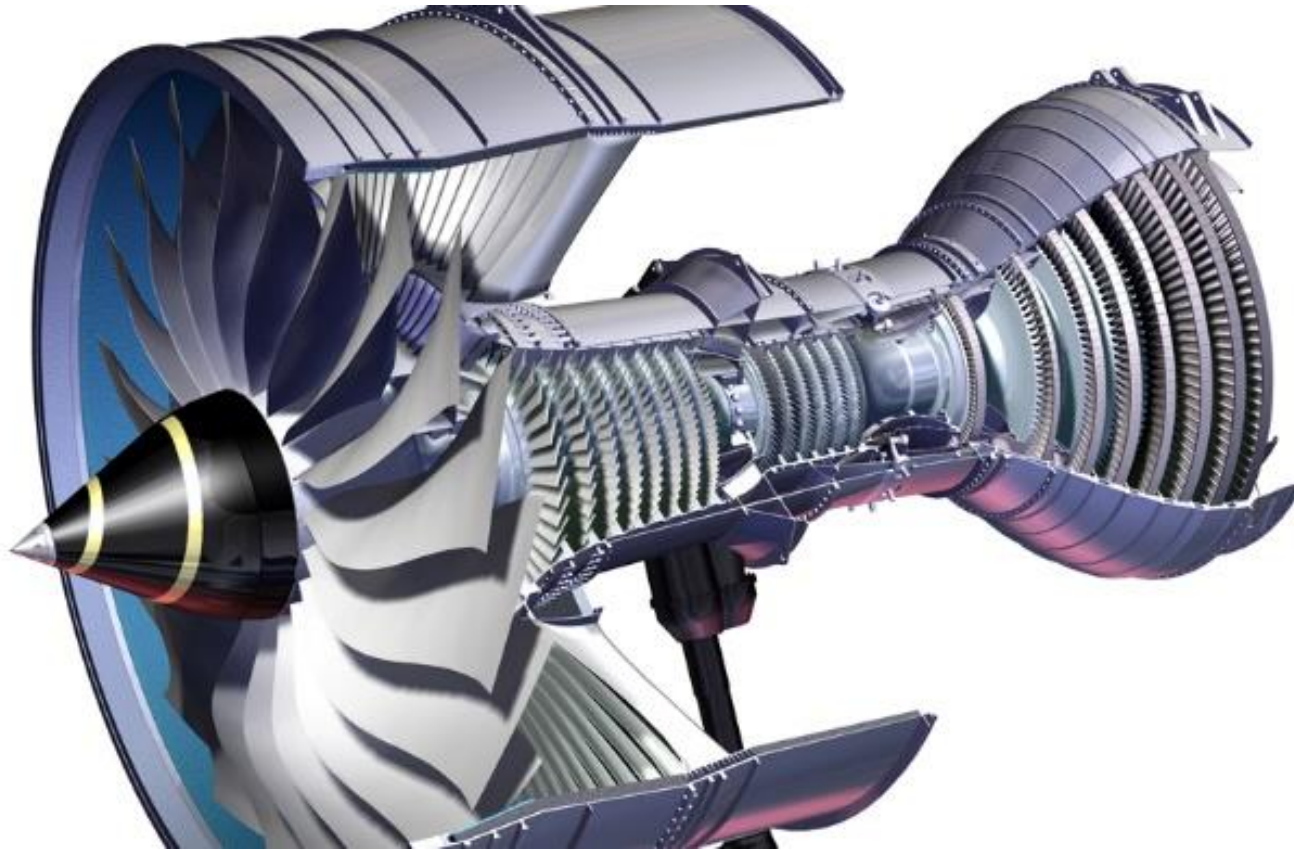


Cast Irons

- Gray Cast Iron (2.5-4% C, 1-3% Si)
 - Graphite flakes form throughout
 - Good vibration damping (engine and other machinery)
 - Good internal lubrication (machineable)
 - Low ductility
 - Application: automotive engine blocks
- Ductile iron (2.5-4% C, 1-3% Si)
 - Chemically treated before pouring so spheroids → ductile
 - Application: Machinery component requiring high strength and good wear resistance
- White Cast Iron (Lesser C and Si than gray CI)
 - Cooling of molten metal is much faster so C remains in the form of Fe_3C and does not precipitate out in form of flakes
 - Hard and brittle
 - Excellent wear resistance
 - Good strength
 - Application: Railway brake shoes



Materials: Strategic & Critical applications

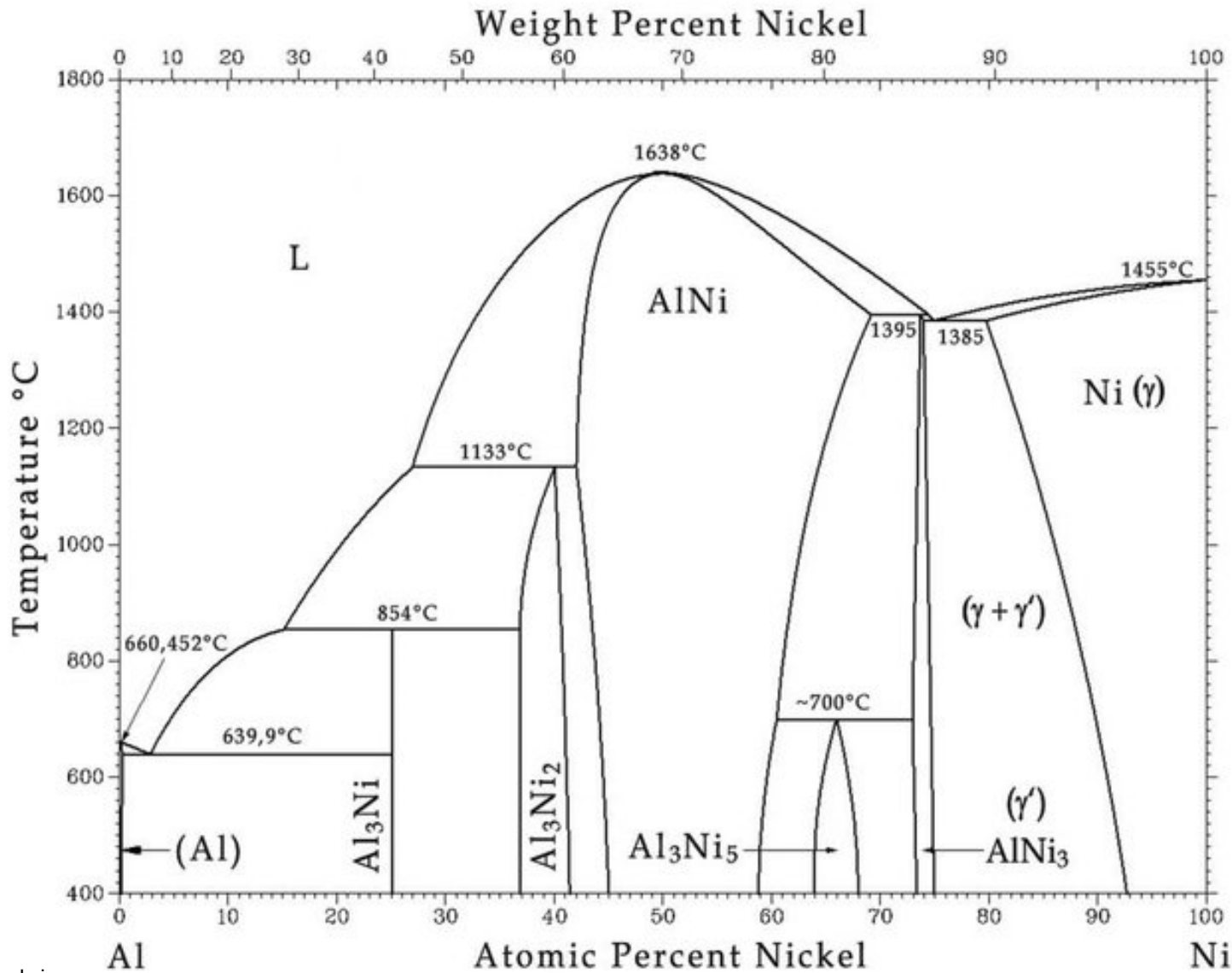


Ni based alloys

<https://www.cam.ac.uk/research/news/super-superalloys-hotter-stronger-for-even-longer>



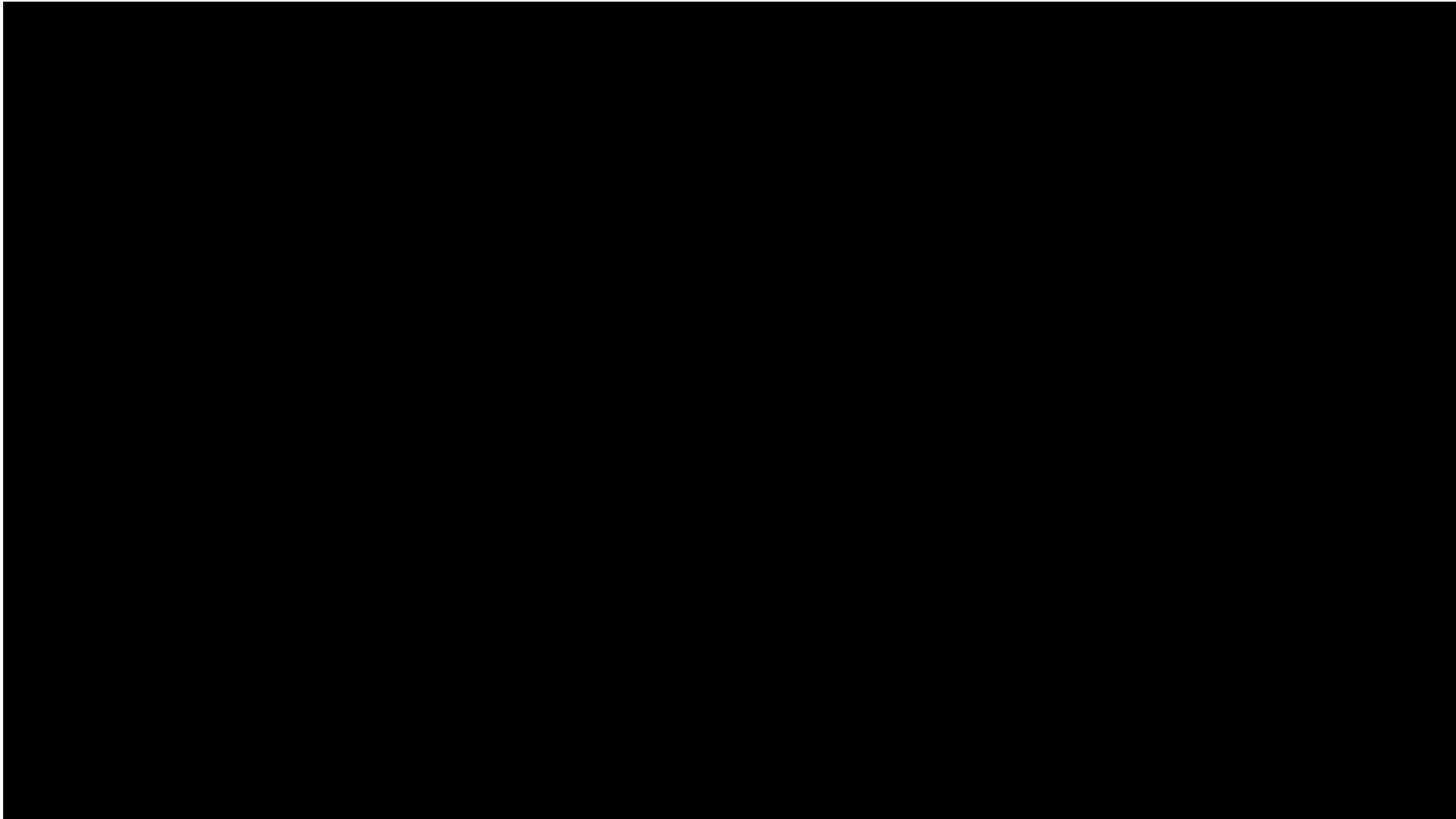
Superalloys: γ - γ' microstructure



Courtesy: Google images



Superalloys: γ - γ' microstructure



<https://www.youtube.com/watch?v=wYHch5QIWTQ>