Principles of Metallurgy

- 1. List three metals that are found in nature as Oxide ores.(AS1)
- 2. List three metals that are found in nature in un combined form. (AS1)
- 3. Write a note on dressing of ore in metallurgy? (AS1)
- 4. What is an ore? On what basis a mineral is chosen as an ore? (AS1)
- 5. Write the names of any two ores of iron? (AS1)
- 6. How do metals occur in nature? Give examples to any two types of minerals. (AS1)
- 7. Write short notes on froth floatation process? (AS1)
- 8. When do we use magnetic separation method for concentration of an ore? Explain with an example? (AS1)
- 9. Write short notes on each of the following: (AS1)
- i) Roasting ii) Calcination iii) Smelting.
- 10. What is the difference between roasting and calcinations? Give one example for each? (AS1)
- 11. What is gauge and slag. (AS1)
- 12. Draw the diagram showing i) Froth floatation ii) Magnetic separation. (AS5)
- 13. Draw a neat diagram of Reverboratory furnace and label it neatly? (AS5)

The impurity present in the ore is called as [] a) Gangue b) flux c) Slag d) Mineral
Which of the following is a carbonate ore? [] a) Magnesite b) Bauxite c) Gypsum d) Galena
3. Which of the following is the correct formula of Gypsum [] a) CuSO4. 2H2O b) CaSO4. ½ H2O c) CuSO4. 5H2O d) CaSO4. 2H2O
4. The oil used in the froth floatation process is [] a) kerosene oil b) pine oil c) coconut oil d) olive oil.
5. Froth floatation is method used for the purification ofore. [] a) sulphide b) oxide c) carbonate d) nitrate
6. Galena is an ore of [] a) Zn b) Pb c) Hg d) Al
7. The metal that occurs in the native form is [] a) Pb b) Au c) Fe d) Hg
8. The most abundant metal in the earth's crust is [] a) Silver b) Aluminium c) zinc d) iron
9. The reducing agent in thermite process is [] a) Al b) Mg c) Fe d) Si
10. The purpose of smelting an ore is [] a) Oxidise b) Reduce c) Neutralise d) None of these