

CH3104: Organic Chemistry II (Physical Organic Chemistry)

Syllabus:

- Structure and Bonding in Organic Compound and their Consequence in Reactivity. (12 lectures)
- Kinetics, thermodynamics and selectivity. (8 lectures)
- Solvent and solution properties: Empirical scale of solvent effects. (3 lectures)
- Acid - Base catalysis. (3 lectures)
- Investigation of organic reaction mechanism. (5 lectures)
- Photochemistry and Pericyclic Reaction: (9 Lectures)

Books:

Ian Fleming *Molecular Orbitals and Organic Chemical Reactions*

F. A. Carey and R. J. Sundberg *Advance Organic Chemistry, Part A: Structure and Mechanisms*

Clayden, Greeves, Warren *Organic Chemistry*

Anslyn, Dougherty, *Modern Physical Organic Chemistry*

Grading:* Endsem (complete syllabus)

Midsem

Quiz

Assignments

Molecular Interactions: Steric + Electronics

Elimination vs Substitution reaction:



Molecular Interactions: Steric + Electronics

In modern form,^[3] the Klopman-Salem equation is commonly given as

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$$\Delta E = \text{Sterics} + \text{Electronics}$$

$$= - \sum_{ab} (q_a + q_b) \beta_{ab} S_{ab} + \sum_{k < l} \frac{Q_k Q_l}{\epsilon R_{kl}} + \sum_r^{\text{occ.}} \sum_s^{\text{unocc.}} - \sum_s^{\text{occ.}} \sum_r^{\text{unocc.}} \frac{2(\sum_{ab} c_{ra} c_{sb} \beta_{ab})^2}{E_r - E_s}$$

Molecular Orbitals – Frontier Molecular Orbitals (FMOs)



Orbital interaction = $\sum \sum \frac{\sum (C_{Nu} C_{El} \beta)^2}{(E_{El} - E_{Nu})}$

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Arrow pushing mechanism

- What does the curved arrows represents?

Illustrates the movement of electron pairs from bonds or lone pairs to form new bonds.

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Frontier Molecular Orbitals (FMOs)

Interaction energy = Orbital overlap b/w
HOMO and LUMO



Review FMO Theory

Salem Klopman Equation

$$\Delta E = \underbrace{- \sum_{ab} (q_a + q_b) \beta_{ab} S_{ab}} + \underbrace{\sum_{k < l} \frac{Q_k Q_l}{dR_{kl}}} + \underbrace{\sum_r^{\text{occ.}} \sum_s^{\text{unocc.}} - \sum_s^{\text{occ.}} \sum_r^{\text{unocc.}} \frac{2(\sum_{ab} c_{ra} c_{sb} \beta_{ab})^2}{E_r - E_s}}$$

