



Laboratory Gasification Memo Carbon Balance Experiment

Summary

In order to audit the laboratory gasification system and ensure that our gas analysis data is giving us accurate conversion results, a carbon balance was performed. 100.5% of the carbon put into the system was accounted for in the gaseous and solid products, and it was determined that the conversion calculations obtained from the mass spec are accurate.

Experimental Methods

Since the mass balance was performed to obtain information about slag formation and waste disposal at the pilot and commercial scales, run conditions which have historically lead to higher conversions were chosen to mirror the desired conversion at large scales. Biomass flow rate was 2 lbs/hr, and the SiC tube skin temperature was set to 1450 °C. The experiment was run as long as possible to maximize the amount of char that was produced for sampling. Detailed information about the run conditions can be found in Appendix A. Feedstock was taken from barrel 101534, which was produced during Mascoma run MS1231.

Carbon content is typically higher in the filters compared to that in the ash knockout, so the char from each section was collected and weighed separately. Each sample was analyzed by Huffman Laboratories for carbon content and moisture.

The experiment was interrupted and restarted twice because of plugs, so three separate analyses were performed on each steady state section using Sundrop Fuels' analysis software. The total conversion from the software is calculated using Equation 1. These conversions can be compared to the overall conversion calculated by comparing the amount of carbon in the char versus the amount of carbon fed into the system as biomass as shown in Equation 2. This will show both the accuracy of the conversion calculated by the software as well as the ability of the system to reach similar steady states for identical run conditions.

$$X_{tot} = \frac{\dot{n}_{Cout,gas} - \dot{n}_{C_{in,CO_2}}}{\dot{n}_{Cin,biomass}} \tag{1}$$

$$X_{solids} = 1 - \frac{m_{C_{out,solids}}}{m_{C_{in,biomass}}} \tag{2}$$

Also, a carbon balance is performed using Equation 3. This shows the percentage of inlet carbon that is accounted for in the gaseous and solids products.

$$B_C = \frac{m_{C_{in,biomass}} + m_{C_{in,CO_2}}}{m_{C_{out,solids}} + m_{C_{out,gas}}}$$
(3)

Results and Discussion

Comparing Conversions

The conversions were calculated separately for each of the three runs, and are given in Table 1 as X_{tot} . The average total conversion was 0.823, and the standard deviation was 0.00737. This shows that there was good repeatability between the runs.

Conversion was also calculated using Equation 2. Carbon content of the biomass as well as the char are shown in Tables 2 and 3. The conversion found using solids analyses was found to be 0.805. This is only 0.018 lower than the conversion found using the analysis software. Considering that the software only analyzes selected time periods within the mass balance experiment, a difference of 1.8% absolute is certainly within acceptable limits. Periods of start up and instability could have been leading to different conversion rates reflected in the conversion calculated using solids results.



Table 1: Conversion calculations for the mass balance experiment using analysis software (X_{tot}) and carbon analysis data from the char and biomass (X_{solids}) .

Run ID	X_{tot}	X_{tot} Average	X_{solids}
511 512 513	0.820 0.817 0.831	0.823	0.805

Table 2: Solids analyses used to find total carbon fed through the system as biomass.

	Bioma	ass In	
Fed (lb)	H_2O (%wt)	C (%wt)	C Fed (lb)
9.17	3.08	57.8	5.14

Table 3: Carbon analysis data from Huffman Laboratories used to calculate the amount of carbon found in the solid products.

Ash Knockout				
Collected (lb)	H_2O (% wt)	C (% wt)		
0.429	0.83	91.0		
	Filters			
Collected (lb)	H_2O (% wt)	C (% wt)		
0.655	2.92	96.5		

Table 4: Final carbon balance numbers for inlet and outlet streams.

C In (l Biomass	,	C Out Solids	` /	B_C
5.14	1.20	1.00	5.37	1.00

Carbon Balance

The carbon balance was calculated using Equation 3, and is a representation of the fraction of carbon in the inlet that is accounted for in all products by solids and gas analyses. The balance comes out as 1.005, which means that all carbon put into the system was accounted for in the outlet products within 0.5%. This is an extremely good result, and it gives great confidence in the ability of the mass spectrometer to provide us with molar flow rates of each gaseous species based on an argon tracer gas.

Conclusion

The mass balance runs were completed satisfactorily and samples were produced to send out for third party analyses. The carbon balance performed accounted for 100.5% of the carbon fed to the system. This shows that the molar flow rates of each product calculated using mass spec analysis are accurate and can provide accurate data to calculate conversion numbers for experiments. Also, the fact that the total conversions from the analysis software and solids analysis match within 1.8% absolute shows that the analysis software is calculating reliable conversions. We can assume that our conversion numbers going forward in the future will be accurate until another mass balance run is scheduled.



A Experimental Setpoints

Run ID	Temp °C	Pressure psig	Biomass lbs/hr	Steam mL/min		Entrainment SLPM N ₂	Downbed SLPM CO ₂	Argon SLPM
511	1450	50	2	12.08	300	12.08	3.3	2
512	1450	50	2	12.08	300	12.08	3.3	2
513	1450	50	2	12.08	300	12.08	3.3	2

B Additional Results

Run ID	Feed Start	Feed Stop	Steady State Start	Steady State Stop
511	2014-06-09 10:36	2014-06-09 12:24	2014-06-09 11:03	2014-06-09 12:23
512	2014-06-09 12:53	2014-06-09 14:29	2014-06-09 13:04	2014-06-09 14:28
513	2014-06-09 14:58	2014-06-09 16:27	2014-06-09 15:08	2014-06-09 16:26

Run Id	Space Time Seconds	X Good	X Total	CH ₄ Yield	$\begin{array}{c} {\rm Tar\ Loading} \\ {\rm mg/Sm^3} \end{array}$
511	3.66	0.794	0.812	0.00791	30.0
512	3.66	0.792	0.817	0.00723	24.9
513	3.65	0.804	0.831	0.00827	32.8

C Symbol Definitions

Symbol	Definition
B_C	Total carbon balance
$m_{C_{in,i}}$	Total mass of carbon into the system as species i
$m_{C_{out,gas}}$	Total mass of carbon in gas exiting the system
$m_{C_{out,solids}}$	Total mass of carbon in char collected after the experiment was completed
$\dot{n}_{C_{in,i}}$	Molar flow rate of carbon entering the system as species i
$\dot{n}_{C_{out,gas}}$	Molar flow rate of carbon in all gaseous species exiting the system
$\dot{n}_{C_{out,i}}$	Molar flow rate of carbon in species i exiting the system
X_{solids}	Total conversion calculated using carbon contents from biomass and char
X_{tot}	Total conversion calculated using analysis software