

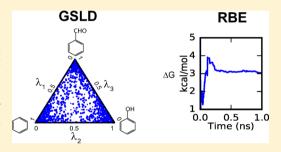
Gibbs Sampler-Based λ -Dynamics and Rao-Blackwell Estimator for **Alchemical Free Energy Calculation**

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Supporting Information

ABSTRACT: λ-dynamics is a generalized ensemble method for alchemical free energy calculations. In traditional λ -dynamics, the alchemical switch variable λ is treated as a continuous variable ranging from 0 to 1 and an empirical estimator is utilized to approximate the free energy. In the present article, we describe an alternative formulation of λ -dynamics that utilizes the Gibbs sampler framework, which we call Gibbs sampler-based λ dynamics (GSLD). GSLD, like traditional λ -dynamics, can be readily extended to calculate free energy differences between multiple ligands in one simulation. We also introduce a new free energy estimator, the Rao-Blackwell estimator (RBE), for use in conjunction with GSLD. Compared



with the current empirical estimator, the advantage of RBE is that RBE is an unbiased estimator and its variance is usually smaller than the current empirical estimator. We also show that the multistate Bennett acceptance ratio equation or the unbinned weighted histogram analysis method equation can be derived using the RBE. We illustrate the use and performance of this new free energy computational framework by application to a simple harmonic system as well as relevant calculations of small molecule relative free energies of solvation and binding to a protein receptor. Our findings demonstrate consistent and improved performance compared with conventional alchemical free energy methods.

1. INTRODUCTION

Free energy calculation is fundamental for understanding many important biophysical processes, such as protein conformational changes, protein-protein interactions, and proteinligand binding processes. 1,2 Calculating protein-ligand binding free energy has important applications in drug discovery, especially in the lead compound generation and optimization stages.3-5 These stages require calculating only protein-ligand relative binding free energy, which has been shown to be easier than calculating protein-ligand absolute binding free energy.^{3,4}

One widely used methodology for calculating protein-ligand relative binding free energy is the alchemical free energy approach. $^{3-5}$ This approach utilizes the thermodynamic cycle shown in Figure 1.2 This thermodynamic cycle specifies that

Figure 1. Thermodynamic cycle used for calculating a relative binding free energy between ligands L₀ and L₁ with receptor R.

 $\Delta \Delta G_{L_0 \to L_1}^{binding} = \Delta G_{L_1}^{binding} - \Delta G_{L_0}^{binding} = \Delta G_{L_0 \to L_1}^{bound} - \Delta G_{L_0 \to L_1}^{unbound}. \ \text{To}$ calculate the relative binding free energy between ligands L₀ and L_1 with receptor R, i.e., $\Delta\Delta G_{L_0\to L_1}^{binding}$, the alchemical free energy method calculates $\Delta G_{L_0\to L_1}^{unbound}$ and $\Delta G_{L_0\to L_1}^{bound}$ by employing alchemical transformations morphing ligand \boldsymbol{L}_0 into ligand \boldsymbol{L}_1 in both unbound and bound environments, respectively.

Several alchemical free energy calculation methods have been developed over the last several decades, such as free energy perturbation,^{6,7} thermodynamic integration,^{2,8} enveloping distribution sampling^{9,10} and λ -dynamics.^{11–17} λ -dynamics is a generalized ensemble method in which the alchemical transformation variable λ is a continuous variable ranging from 0 to 1, with $\lambda = 0$, $0 < \lambda < 1$, and $\lambda = 1$ corresponding to the ligand being in the L₀ state, intermediate hybrid states, and L₁ state, respectively. The potential energy corresponding to λ is

$$V(\lambda, \{x_i\}_{i=0}^{1}, X) = (1 - \lambda)V_0(x_0, X) + \lambda(V_1(x_1, X) + G_1^{b}) + V_{ann}(X)$$
(1)

where X, x_0 , and x_1 are atomic coordinates associated with the environment, ligand L_0 , and ligand L_1 , respectively. $V_i(x_i, X)$ is the potential energy between ligand Li and the environment, and $V_{\text{env}}(X)$ is the potential energy of the environment. G_1^b is a biasing potential to ensure that the two physical states, corresponding to $\lambda = 0$ and $\lambda = 1$, are both sampled in the

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simulation. The biasing potential G_1^b is determined iteratively by running multiple short simulations. ^{14,15,18,19} The dynamics of the system $(\lambda, \{x_i\}_{i=0}^1, X)$ is generated from the extended Hamiltonian

$$H(\lambda, \{x_i\}_{i=0}^l, X) = T_{x,X} + T_{\lambda} + V(\lambda, \{x_i\}_{i=0}^l, X)$$
(2)

where $T_{x,X}$ and T_{λ} correspond to the kinetic energy associated with coordinates $(\{x\}_{i=0}^1, X)$ and λ , respectively. The free energy difference between ligands L_0 and L_1 , with the biasing potential G_1^b , is

$$\Delta G = -\beta^{-1} \ln \frac{P(\lambda = 1)}{P(\lambda = 0)}$$
(3)

where β is the inverse temperature; $P(\lambda = 0)$ and $P(\lambda = 1)$ are probability denisties of λ at points 0 and 1, respectively. In practice, this free energy difference ΔG is estimated using the following empirical estimator based on the trajectory of λ

$$\Delta \hat{G} = -\beta^{-1} \ln \frac{P(\lambda > \lambda_{\text{cutoff}})}{P(\lambda < 1 - \lambda_{\text{cutoff}})}$$
(4)

where λ_{cutoff} (0 < λ_{cutoff} < 1) is a cutoff value that is chosen to be close to 1.14

Although the empirical estimator is straightforward to evaluate based on the λ trajectory, it is not necessarily optimal. One issue is that the empirical estimator is systematically biased as it uses $P(\lambda < 1 - \lambda_{\text{cutoff}})$ and $P(\lambda > \lambda_{\text{cutoff}})$ to approximate $P(\lambda = 0)$ and $P(\lambda = 1)$, respectively. Additionally, the bias depends on the cutoff value λ_{cutoff} , which is chosen empirically and is difficult to quantify as it may vary among different systems.

In the current work, we present a novel form of λ -dynamics called Gibbs sampler-based λ -dynamics (GSLD) with the Rao-Blackwell estimator (RBE). The Gibbs sampler framework for calculating free energy differences between two ligands was first suggested by Chodera and Shirts. ²⁰ In their work, λ was treated as a discrete variable and the multistate Bennett acceptance ratio (MBAR) method²¹ was used to estimate the free energy change. In this study, we show that GSLD and RBE can treat λ as either a discrete variable or a continuous variable when calculating free energy differences between two ligands. When λ is treated as a continuous variable, GSLD and RBE can be generalized to simultaneously calculate free energies of multiple ligands in one simulation, as in the generalization of λ dynamics.14 We explore these new methods through applications to three model systems in this article. This article is organized as follows. In section 2, we describe GSLD and its generalization to multiple ligands. Then we introduce the RBE and show that the MBAR/unbinned weighted histogram analysis method (UWHAM) equations²¹⁻²³ can be derived from the RBE. In section 3, we give detailed setup information for the setup and simulation of the three systems with which we tested the methods. Our results for these three systems are presented in section 4. We conclude with a discussion of how the GSLD and RBE can be used for other applications.

2. METHODS

2.1. GSLD. As a generalized ensemble method, GSLD samples from the joint distribution of λ and the atomic coordinates of the system using the Gibbs sampler. In this section, we first briefly introduce the Gibbs sampler. We then use the Gibbs sampler to formulate pairwise GSLD. We conclude by showing how the GSLD can be generalized to work for multiple ligands.

- 2.1.1. Gibbs Sampler. The Gibbs sampler, which is widely used in both statistics and machine learning, is a Markov chain Monte Carlo (MCMC) method for sampling from multivariate distributions. ^{24,25} To sample (X, Y) from the joint distribution $(X, Y) \sim P(X, Y)$, the Gibbs sampler generates a Markov chain of states $\{(X_t, Y_t), t = 0, 1, 2, ..., N\}$ using the following procedure:
 - Step 0: initialize the starting state (X_0, Y_0) .
 - Step t: sample from the conditional distributions
 - **Updating** X: given the state (X_{t-1}, Y_{t-1}) from step t-1, sample X_t from the conditional distribution of $X_t \sim P(X_t|Y_{t-1})$.
 - **Updating** Y: given X_t from the above update step, sample Y_t from the conditional distribution of $Y_t \sim P(Y_t|X_t)$. The resulting sample (X_t, Y_t) is the state for step t.

Because the above procedure satisfies the detailed balance condition with respect to the joint distribution $(X, Y) \sim P(X, Y)$, the sampled states $\{(X_t, Y_t), t = 0, 1, 2, ..., N\}$ converge to the joint distribution. The update steps require sampling from both conditional distributions: $X_t \sim P(X_t|Y_{t-1})$ and $Y_t \sim P(Y_t|X_t)$. If direct sampling from the conditional distributions is possible, then independent samples can be directly drawn using numerical pseudorandom number generators. Otherwise, samples can be drawn using other Monte Carlo methods or Hamiltonian dynamics, as long as the method satisfies the detailed balance condition with respect to the corresponding conditional distributions. This property of the Gibbs sampler makes it quite flexible for choosing appropriate sampling methods based on the conditional distributions.

2.1.2. Pairwise GSLD. Pairwise GSLD calculates the free energy difference between two ligands: ligands L_0 and L_1 . In pairwise GSLD, λ can be treated as either a continuous variable or a discrete variable.

2.1.2.1. Continuous λ . When λ is treated as a continuous variable, pairwise GSLD samples from the joint distribution of $(\lambda, \{x_i\}_{i=0}^1 X)$

$$P(\lambda, x_0, x_1, X) = \frac{\exp(-\beta[(1 - \lambda)V_0(x_0, X) + \lambda(V_1(x_1, X) + G_1^b) + V_{env}(X)])}{Z}$$
(5)

where Z is the partition function of the generalized ensemble and G_1^b is a biasing potential. G_1^b is determined automatically in the current simulations using a Wang–Landau-like algorithm, which is described in Appendix A. The Gibbs sampler for sampling from the above joint distribution is as follows:

- Step 0: initialize the starting state $(\lambda^0, \{x_i^0\}_{i=0}^1, X^0)$.
- **Step** *t*: sample from the conditional distributions:
 - **Updating** $(\{x_i\}_{i=0}^1, X)$: given the state $(\lambda^{t-1}, \{x_i^{t-1}\}_{i=0}^1, X^{t-1})$ from step t-1, sample $(\{x_i^t\}_{i=0}^1, X^t)$ from the conditional distribution $P(\{x_i^t\}_{i=0}^1, X^t|\lambda^{t-1}) \propto \exp(-\beta \left[(1-\lambda^{t-1}) \ V_0(x_0^t, X^t) + \lambda^{t-1} (V_1(x_1^t, X^t) + G_1^b) + V_{\text{env}}(X^t)\right])$, which is the canonical ensemble distribution at the inverse temperature β . A sample can be drawn from this distribution using molecular dynamics simulation.
 - **Updating** λ : given the atomic coordinates $(\{x_i^t\}_{i=0}^t, X^t)$ sampled from the above update step, sample λ^t directly from the conditional distribution $P(\lambda^t \mid \{x_i^t\}_{i=0}^t, X^t)$ using numerical pseudorandom number

generator. The conditional distribution $P(\lambda^t | \{x_i^t\}_{i=0}^1, X^t)$ is

$$\begin{split} &P(\lambda^{t}|\{x_{i}^{t}\}_{i=0}^{l},X^{t})\\ &= \left(\exp(-\beta[(1-\lambda^{t})V_{0}(x_{0}^{t},X^{t})+\lambda^{t}(V_{1}(x_{1}^{t},X^{t})+G_{1}^{b})\right.\\ &+V_{\text{env}}(X^{t})])\Big)\Big/\bigg(\int_{0}^{1}\exp(-\beta[(1-\lambda^{t})V_{0}(x_{0}^{t},X^{t})+\lambda^{t}(V_{1}(x_{1}^{t},X^{t})+G_{1}^{b})+V_{\text{env}}(X^{t})])\mathrm{d}\lambda\bigg)\\ &= \left(\beta\cdot(\Delta V_{0\to 1}^{t}+G_{1}^{b})+V_{\text{env}}(X^{t})])\mathrm{d}\lambda\bigg)\\ &= \left(\beta\cdot(\Delta V_{0\to 1}^{t}+G_{1}^{b})\exp(-\lambda^{t}\cdot\beta\cdot[\Delta V_{0\to 1}^{t}+G_{1}^{b}])\right)\\ &/\bigg(1-\exp(-\beta\cdot[\Delta V_{0\to 1}^{t}+G_{1}^{b}])\bigg) \quad (0\leq\lambda^{t}\leq1) \end{split}$$

where $\Delta V_{0\rightarrow 1}^t = V_1(x_1^t, X^t) - V_0(x_0^t, X^t)$. This is an exponential distribution of λ^t restricted on the interval of [0, 1]. Therefore, sampling λ^t directly from this distribution can be done using the inverse transformation method

$$\lambda^{t} = -\frac{1}{\beta \cdot (\Delta V_{0 \to 1}^{t} + G_{1}^{b})} \ln \left[1 - \left[1 - e^{-\beta \cdot (\Delta V_{0 \to 1}^{t} + G_{1}^{b})}\right] \cdot u\right]$$
(7)

where u is a random sample from the uniform distribution on [0, 1]. The resulting sample $(\lambda^t, \{x_i^t\}_{i=0}^t, X^t)$ is the state for step t.

2.1.2.2. Discrete λ . When λ is a discrete variable specified by the set $\{l_1, l_2, ..., l_M\}$, GSLD samples from the joint distribution

$$P(\lambda = l_j, x_0, x_1, X) \propto \exp(-\beta [V_0(x_0, X, 1 - l_j) + V_1(x_1, X, l_j) + G_j^b + V_{env}(X)])$$
(8)

where G_j^b is the biasing potential added to the state corresponding to $\lambda = l_j$. Sampling from this distribution is done in the same way as the case where λ is continuous except that the conditional distribution $P(\lambda^t | \{x_i^t\}_{i=0}^1, X^t)$ becomes a multinomial distribution

$$\begin{split} P(\lambda^{t} &= l_{j} | \{x_{i}^{t}\}_{i=0}^{1}, X^{t}\} \\ &= \frac{\exp(-\beta [V_{0}(x_{0}^{t}, X^{t}, 1 - l_{j}) + V_{I}(x_{1}^{t}, X^{t}, l_{j}) + G_{j}^{b}])}{\sum_{k=1}^{M} \exp(-\beta [V_{0}(x_{0}^{t}, X^{t}, 1 - l_{k}) + V_{I}(x_{1}^{t}, X^{t}, l_{k}) + G_{k}^{b}]])} \end{split}$$

$$(9)$$

from which samples can also be drawn directly using numerical methods. The biasing potentials G_j^b are determined similarly as the case when λ is continuous. We note that eq 9 is similar to the distribution calculated using the infinite swap limit in replica exchange methods. $^{28-31}$

The advantage of using λ as a discrete variable is that the pairwise GSLD still works when the potential energy $V_i(x_i, X, \lambda)$ is λ dependent, such as when a soft-core Lennard-Jones potential is employed to facilitate sampling. When λ is continuous, using λ dependent $V_i(x_i, X, \lambda)$ will make the normalization constant of the conditional distribution $P(\lambda | \{x_i\}_{i=0}^1, X)$ not analytically integrable and prevent direct sampling from the conditional distribution $P(\lambda | \{x_i\}_{i=0}^1, X)$. However, as shown below, the advantage of using λ as a continuous variable is that the GSLD can be generalized for multiple ligands.

2.1.3. Generalizing GSLD for Multiple Ligands. Like λ -dynamics, GSLD can be generalized to calculate the free energies for multiple ligands in one simulation. Assuming there

are n ligands, the fraction of the ith ligand in the hybrid state is represented by λ_p for i=1,2,...,n. The hybrid state is specified by the value of $(\lambda_1,\lambda_2,...,\lambda_n)$ that satisfies the conditions $\sum_{i=1}^n \lambda_i = 1$ and $0 \le \lambda_i \le 1$, i=1,2,...,n. The hybrid state's potential energy is defined as $V(\{\lambda_i\}_{i=1}^n \{x_i\}_{i=1}^n X) = \sum_{i=1}^n \lambda_i (V_i(x_i,X) + G_i^b) + V_{\text{env}}(X)$, where x_i and X are atomic coordinates associated with the ith ligand and environment, respectively; G_i^b is the biasing potential added for the ith ligand and can be determined similarly as in the pairwise GSLD. Sampling from the generalized ensemble distribution: $P(\{\lambda_i\}_{i=1}^n, \{x_i\}_{i=1}^n, X) \propto \exp(-\beta \cdot V(\{\lambda_i\}_{i=1}^n, \{x_i\}_{i=1}^n, X))$ can be done using the following Gibbs sampler procedure:

- **Step 0**: initialize the starting state $(\{\lambda_i^0\}_{i=1}^n, \{x_i^0\}_{i=1}^n, X^0)$.
- Step t: sample from the conditional distributions.
 - **Updating** $(\{x_i\}_{i=1}^n, X)$: given the state $(\{\lambda_i^{t-1}\}_{i=1}^n, X^{t-1}\}$ from step t-1, sample $(\{x_i^t\}_{i=1}^n, X^t)$ from the conditional distribution $P(\{x_i^t\}_{i=1}^n, X^t)$ $\{\lambda_i^{t-1}\}_{i=1}^n$ using molecular dynamics simulation.
 - **Updating** $\{\lambda_i\}_{i=1}^n$: given the sample $(\{x_i^t\}_{i=1}^n, X^t)$ from the above update step, the conditional distribution of $\{\lambda_i^t\}_{i=1}^n$ in the set $S = \{(\lambda_1, ..., \lambda_n)|\sum_{i=1}^n \lambda_i = 1 \text{ and } \lambda_i \geq 0, i = 1, ...n\}$ is given by

$$P(\{\lambda_{i}^{t}\}_{i=1}^{n} | \{x_{i}^{t}\}_{i=1}^{n}, X^{t}) = \frac{\exp(-\beta [\sum_{i=1}^{n} \lambda_{i}^{t} [V_{i}(x_{i}^{t}, X^{t}) + G_{i}^{b}] + V_{\text{env}}(X^{t})])}{Z}$$
(10)

where

$$Z = \int_{S} \exp(-\beta \left[\sum_{i=1}^{n} \lambda_{i}^{t} \left[V_{i}(x_{i}^{t}, X^{t}) + G_{i}^{b}\right] + V_{\text{env}}(X^{t})\right])$$

$$dm_{S}(\lambda)$$

$$= e^{-\beta V_{\text{env}}(X^{t})} \sum_{i=1}^{n} \left(e^{-\beta \left[V_{i}(x_{i}^{t}, X^{t}) + G_{i}^{b}\right]}\right)$$

$$/\left(\beta^{n-1} \prod_{j \neq i} \left(\left[V_{j}(x_{j}^{t}, X^{t}) + G_{j}^{b}\right]\right) - \left[V_{i}(x_{i}^{t}, X^{t}) + G_{i}^{b}\right]\right)$$

$$(11)$$

and $dm_S(\lambda)$ is the infinitesimal volume element of the simplex S. Because $\sum_{i=1}^n \lambda_i^t = 1$, the conditional distribution $P(\{\lambda_i^t\}_{i=1}^n|\{x_i^t\}_{i=1}^n, X^t)$ has only n-1 degrees of freedom. Sampling from this conditional distribution is equivalent to sampling from the n-1-dimensional distribution

$$P(\{\lambda_{i}^{t}\}_{i=1}^{n-1}|\{x_{i}^{t}\}_{i=1}^{n}, X^{t})$$

$$\propto \exp(-\beta[\sum_{i=1}^{n-1} \lambda_{i}^{t}[V_{i}(x_{i}^{t}, X^{t}) + G_{i}^{b} - V_{n}(x_{n}^{t}, X^{t})$$

$$- G_{n}^{b}]))$$
(12)

where $0 \leq \sum_{i=1}^{n-1} \lambda_i^t \leq 1$, and $\lambda_i^t \geq 0$. The environment atom energy term, $V_{\text{env}}(X^t)$, does not appear in eq 12 because it is part of both the numerator and denominator of eq 10 and can be canceled out as a constant when $(\{x_i^t\}_{i=1}^n, X^t)$ is fixed. Sampling from this n-1-dimensional distribution $P(\{\lambda_i^t\}_{i=1}^{n-1}|\{x_i^t\}_{i=1}^n, X^t)$ is done using the rejection method. In the rejection method, each $\{\lambda_i^t\}_{i=1}^{n-1}$ is sampled independently from the distribution: $P(\lambda_i^t) \propto \exp(-\beta \lambda_i^t | V_i(x_i^t, X^t) + G_i^b - V_i^t)$

 $V_n(x_n^t, X^t) - G_n^b]$), where $0 \le \lambda_i^t \le 1$. If the sample $\{\lambda_i^t\}_{i=1}^{n-1}$ satisfies the condition $0 \le \sum_{i=1}^{n-1} \lambda_i^t \le 1$, it is accepted; otherwise, the sample $\{\lambda_i^t\}_{i=1}^{n-1}$ is rejected. This procedure is repeated until a sample $\{\lambda_i^t\}_{i=1}^{n-1}$ is accepted. Set $\lambda_n^t = 1 - \sum_{j=1}^{n-1} \lambda_j^t$ and the resulting sample $\{\{\lambda_i^t\}_{i=1}^n, \{x_i^t\}_{i=1}^n, X^t\}$ is the state for step t.

2.2. Rao-Blackwell Estimator (RBE). Although the empirical estimator used in λ -dynamics can also be utilized in GSLD to estimate the free energy, it is not an optimal estimator and may contain a system-dependent bias. RBE is introduced here to eliminate these potential issues. RBE is the estimator derived by applying the Rao-Blackwellization transformation to the empirical estimator. Rao-Blackwellization is a statistical method, inspired by the Rao-Blackwell theorem, 33,34 to transform a crude estimator into a better estimator that has smaller mean squared error for estimating the quantity of interest. 35

For pairwise GSLD with continuous λ , the quantity of interest is the free energy $\Delta G = -\beta^{-1} \ln[P(\lambda=1)/P(\lambda=0)]$. To estimate ΔG , the empirical estimator approximates $P(\lambda=1)$ and $P(\lambda=0)$ directly by calculating the fraction of λ values that are close to 1 and 0, respectively, based on the λ trajectory. In contrast, the RBE ignores the λ trajectory and uses only the atomic coordinate trajectory. It is based on the fact that $P(\lambda=1)$ and $P(\lambda=0)$ are equal to the expectation of the conditional probability of λ with respect to the atomic coordinates, i.e.

$$P(\lambda = 1) = \mathbb{E}_{\{\{x_i\}_{i=0}^1, X\}}[P(\lambda = 1|\{x_i\}_{i=0}^1, X)]$$

and

$$P(\lambda = 0) = \mathbb{E}_{\{x_i\}_{i=0}^1, X\}}[P(\lambda = 0|\{x_i\}_{i=0}^1, X)]$$

Therefore, RBE uses the following formula to estimate the free energy ΔG

$$\begin{split} \Delta G_{\text{RBE}} &= -\beta^{-1} \ln \frac{P(\lambda = 1)}{P(\lambda = 0)} \\ &= -\beta^{-1} \ln \frac{\mathbb{E}_{\{\{x_i\}_{i=0}^{l}, X\}}[P(\lambda = 1|\{x_i\}_{i=0}^{l}, X)]}{\mathbb{E}_{\{\{x_i\}_{i=0}^{l}, X\}}[P(\lambda = 0|\{x_i\}_{i=0}^{l}, X)]} \\ &= -\beta^{-1} \ln \frac{1/N \cdot \sum_{t=0}^{N} P(\lambda = 1|\{x_i^t\}_{i=0}^{l}, X^t)}{1/N \cdot \sum_{t=0}^{N} P(\lambda = 0|\{x_i^t\}_{i=0}^{l}, X^t)} \end{split}$$

where

$$P(\lambda = 1 | \{x_i^t\}_{i=0}^l, X^t) = \frac{\beta \cdot (\Delta V_{0 \to 1}^t + G_1^b) \cdot \exp(-\beta \cdot [\Delta V_{0 \to 1}^t + G_1^b])}{1 - \exp(-\beta \cdot [\Delta V_{0 \to 1}^t + G_1^b])}$$

$$P(\lambda = 0 | \{x_i^t\}_{i=0}^l, X^t) = \frac{\beta \cdot (\Delta V_{0 \to 1}^t + G_1^b)}{1 - \exp(-\beta \cdot [\Delta V_{0 \to 1}^t + G_1^b])}$$
(14)

and N is the number of samples.

For the generalized GSLD with multiple ligands, the RBE can be derived similarly. To estimate the free energy of the *i*th ligand given by $G(\lambda_i = 1, \lambda_{j \neq i} = 0) = -\beta^{-1} \ln P(\lambda_i = 1, \lambda_{j \neq i} = 0)$, the RBE uses the following formula

$$\begin{split} G_{\text{RBE}}(\lambda_{i} = 1, \, \lambda_{j \neq i} = 0) \\ &= -\beta^{-1} \ln P(\lambda_{i} = 1, \, \lambda_{j \neq i} = 0) \\ &= -\beta^{-1} \ln \mathbb{E}_{\{\{x_{k}\}_{k=1}^{n}, X\}} [P(\lambda_{i} = 1, \, \lambda_{j \neq i} = 0 | \{\{x_{k}\}_{k=1}^{n}, \, X\})] \\ &= -\beta^{-1} \ln [1/N \cdot \sum_{t=0}^{N} P(\lambda_{i} = 1, \, \lambda_{j \neq i} = 0 | \{\{x_{i}^{t}\}_{k=1}^{n}, \, X\})] \\ &= -\beta^{-1} \ln \left[1/N \cdot \sum_{t=0}^{N} \frac{\exp(-\beta [V_{i}(x_{i}^{t}, \, X^{t}) + G_{i}^{b}])}{Z} \right] \end{split}$$

$$(15)$$

where Z is given in eq 11 in section 2.1.3.

As shown in the above formulas, the RBE estimator $\Delta G_{\rm RBE}$ does not depend on the empirical cutoff value of $\lambda_{\rm cutoff}$. On the basis of the Rao–Blackwell theorem, $\Delta G_{\rm RBE}$ is an unbiased estimator. In addition, if the samples from GSLD are independent, the mean squared error of RBE is guaranteed to be smaller than or equal to that of the empirical estimator. Although the samples from GSLD are usually not truly independent, the advantage of RBE can often be justified empirically. ³⁶

2.3. Derivation of the MBAR/UWHAM Equations Using RBE. Although RBE is originally introduced to estimate free energies based on sampling from GSLD, RBE can also be used when multiple equilibrium states are sampled independently. When RBE is applied to this case, it generates the MBAR/UWHAM equations, ^{21–23} which are widely used in current alchemical free energy methods.

Let us assume there are M equilibrium states with potential energy function of V_i , i=1,2,...,M. Each equilibrium state is sampled independently. The conformations sampled from state i are represented as x_i^k , $k=1,2,...,n_i$, where n_i is the number of conformations from state i. The total number of conformations is $N=\sum_{j=1}^M n_j$. The free energy of state i is represented as G_i^* . We use $\lambda \in \{1,2,...,M\}$ as an index variable to represent the M equilibrium states, with $\lambda=i$ corresponding to state i. To calculate the free energies for all the equilibrium states, all of the conformations $\{x_i^k, i=1,2,...,M, k=1,2,...,n_i\}$ are pooled together and viewed as samples from the generalized ensemble $P(\lambda=i,x) \propto \mathrm{e}^{-\beta[V_i(x)+G_i^b]}$, where G_i^b is the biasing energy added to state i to adjust the relative weight of state i to be proportional to n_i , i.e, G_i^b needs to satisfy the condition

$$G_i = G_i^* + G_i^b = -\beta^{-1} \ln \frac{n_i}{N}$$
 (16)

where G_i is the free energy of state i with the biasing potential of G_i^b and G_i^* is the unbiased free energy of state i. We note that the biasing potentials G_i^b in eq 16 are unknown variables. They are introduced to make eq 16 valid, which is the requirement for applying the RBE. These unknown biasing potentials G_i^b can be calculated after the values of G_i^* are solved. The RBE for this generalized ensemble is

$$G_{i} = -\beta^{-1} \ln P(\lambda = i)$$

$$= -\beta^{-1} \ln \frac{1}{N} \sum_{j=1}^{M} \sum_{k=1}^{n_{j}} P(\lambda = i | x_{j}^{k})$$

$$= -\beta^{-1} \ln \frac{1}{N} \sum_{j=1}^{M} \sum_{k=1}^{n_{j}} \frac{e^{-\beta [V_{i}(x_{j}^{k}) + G_{i}^{b}]}}{\sum_{l=1}^{M} e^{-\beta [V_{l}(x_{j}^{k}) + G_{l}^{b}]}}$$
(17)

Combining eq 16 with eq 17, we have

$$G_i^* = -\beta^{-1} \ln \sum_{j=1}^M \sum_{k=1}^{n_j} \frac{e^{-\beta [V_i(x_j^k)]}}{\sum_{l=1}^M n_l \cdot e^{-\beta [V_l(x_j^k) - G_l^*]}}$$
(18)

which is the same as the MBAR/UWHAM equations. ^{21–23} Previously, the MBAR/UWHAM equations were derived as either a result of the maximum likelihood principle or an unbinned extension of the weighted histogram analysis method (WHAM). ^{21–23} Here, we have shown that the MBAR/UWHAM equations can also be derived using RBE.

3. MODEL SYSTEMS AND COMPUTATIONAL DETAILS

To illustrate how GSLD works and the advantage of RBE over the empirical estimator typically used in λ -dynamics, we applied GSLD and RBE to three test cases: (a) calculation of the free energy difference between two states of a harmonic oscillator system, (b) calculation of the relative hydration free energies of three benzene derivatives, and (c) calculation of the binding free energy difference between benzene and p-xylene bound to the L99A mutant of the protein T4 lysozyme. The simulations in these calculations were run using CHARMM compiled with OpenMM. Each calculation was repeated 10 times. Error bars were calculated as the standard variation of the results from these 10 independent repeats.

3.1. Harmonic System. The harmonic system consists of a one-dimensional particle that switches between two states: states 0 and 1. Each state has a harmonic potential energy. The purpose is to calculate the free energy difference of the particle when it changes from state 0 to state 1, i.e., $\Delta G = G_1 - G_0$. Specifically, state 0 has a potential energy given by $\frac{1}{2}k_0(x-x_0^e)^2$, and state 1 has a potential energy given by $\frac{1}{2}k_1(x-x_1^e)^2$. To prevent the particle from moving too far from the equilibrium position, a restraining potential is added for each state. This restraining potential is not scaled by λ . The resulting hybrid potential energy is

$$\begin{split} V(\lambda, \, x_0, \, x_1) &= \\ & (1 - \lambda) \cdot \frac{1}{2} k_0 (x_0 - x_0^\epsilon)^2 + \lambda \cdot \frac{1}{2} k_1 (x_1 - x_1^\epsilon)^2 + \frac{1}{2} k_{\text{env}} \\ & (|x_0| - x_{\text{env}}^\epsilon)^2 \mathbb{I}\{|x_0| \geq x_{\text{env}}^\epsilon\} + \frac{1}{2} k_{\text{env}} (|x_1| - x_{\text{env}}^\epsilon)^2 \mathbb{I}\{|x_1| \geq x_{\text{env}}^\epsilon\} \end{split}$$

where $\mathbb{I}\{\text{condition}\}$ is equal to 1 if the condition is true; otherwise, it is equal to 0. GSLD is used to sample from the joint distribution of $(\lambda, \{x_i\}_{i=0}^1)$: $P(\lambda, \{x_i\}_{i=0}^1) \propto \exp(-\beta \cdot V(\lambda, \{x_i\}_{i=0}^1))$. Given the value of λ , sampling the coordinates $(\{x_i\}_{i=0}^1)$ is accomplished by running Langevin dynamics for 1 ps with a step size of 1 fs, temperature of 300 K, and friction coefficient of 10 ps⁻¹. The total simulation time is 10 ns. The parameters used for x_0^e , x_1^e , x_{env}^e , and k_{env} are -2.0 Å, 2.0 Å, 4.0 Å, and 2.5 kcal/mol·Å⁻², respectively. Two variations of the model system that correspond to setting different values for k_0 and k_1 are used: a symmetrical system with $k_0 = k_1 = 0.75$ kcal/mol·Å⁻² and an asymmetrical system with $k_0 = 0.75$ kcal/mol·Å⁻² and $k_1 = 0.075$ kcal/mol·Å⁻².

3.2. Relative Hydration Free Energies for Three Benzene Derivatives. Relative hydration free energies for three benzene derivatives, benzene, phenol, and benzaldehyde, were calculated from the differences between alchemical free energy changes computed in vacuum and water. The topology and parameter files for the hybrid ligand were generated using

MATCH⁴¹ and in-house developed scripts based on the CHARMM general force field (CGenFF). 42 The simulation in water was done in a water box consisting of 800 TIP3P⁴³ water molecules with cubic periodic boundary conditions. The water box had a size of 30.0 Å \times 30.0 Å. A nonbonded cutoff of 14 Å was used, and the van der Waals switching function and electrostatic force switching function⁴⁴ were used between 12 and 14 Å. Sampling from the conditional distribution $P(x, X|\lambda)$ was accomplished by running Langevin dynamics at 298.15 K for 0.2 ps. The time step size was 2 fs, and the friction coefficient was 10 ps⁻¹. The length of all bonds involving hydrogen atoms was fixed during the simulation using the SHAKE algorithm. 45 The three relative hydration free energies were first calculated by three independent pairwise GSLDs. Then, they were calculated simultaneously using the generalized GSLD for multiple ligands. For comparison, the three relative hydration free energies were also calculated using the FEP/MBAR method, in which 11 states corresponding to $\lambda =$ 0, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9, 1 were used.

3.3. Relative Binding Free Energy between Benzene and p-Xylene with T4 Lysozyme. The L99A mutant of T4 lysozyme has been a model protein system for testing free energy calculation methods.^{46–48} It has experimental binding free energy data for a series of benzene derivatives including benzene and *p*-xylene. ^{37,38} The relative binding free energy between benzene and p-xylene was calculated using the difference between the alchemical free energy change in water and in the protein environment. The alchemical free energy change in water was calculated using pairwise GSLD with continuous λ . Calculating the alchemical free energy in the protein environment is challenging, even though the binding site of T4 lysozyme is a relatively simple nonpolar pocket and the alchemical change from benzene to p-xylene is small. This challenge arises from the fact that T4 lysozyme has a conformational change for the side-chain dihedral angle χ (N-CA-CB-CG1) of residue Val111, which accompanies the alchemical transformation from benzene to p-xylene. 46 When T4 lysozyme binds with benzene (PDB ID: 181L), the dihedral angle stays in the trans conformation ($\chi \approx -180^{\circ}$). When it binds with p-xylene (PDB ID: 187L), the dihedral angle changes into the gauche conformation ($\chi \approx -60^{\circ}$). Failing to sample these two relevant conformations in a free energy calculation would cause a quasi-nonergodicity problem, i.e., the calculated free energy will depend on which conformation is used as the starting conformation. 46,48 To address the problem, several methods have been developed. These methods include enhanced sampling methods, such as the two-dimensional replica exchange method (REM),⁴⁷ the free energy perturbation/replica exchange with solute tempering (FEP/REST) method, 48 and the potential of mean force (PMF) method, which was first introduced by Tobias and Brooks for addressing a similar problem in 1989⁴⁹ and rediscovered as the "confineand-release" method by Mobley et al. in 2007.46 Here, we combined the PMF method with GSLD to calculate the alchemical free energy changes between benzene and p-xylene in the protein environment.

To make our computational protocol clear, we reformulated the PMF method using conditional probability as shown in Appendix B. The free energy change $\Delta G(\chi^*)$ was calculated using pairwise GSLD with a harmonic restraint potential on χ to keep it near χ^* during the pairwise GSLD simulation. The force constant of the harmonic restraint potential was 1195.3 kcal/mol·radius⁻². In our calculations, we chose χ^* to be -180°

and -60° , although the final calculated result ΔG did not depend on the choice of χ^* . In the pairwise GSLD, λ was chosen to be a discrete variable specified by the set $\{l_1, l_2, ..., l_n\}$ l_{16} }. $\lambda = l_1$ corresponds to the physical state that the ligand is benzene, and $\lambda = l_{16}$ corresponds to the physical state that the ligand is *p*-xylene. When λ was changed from l_1 to l_{16} , the ligand was alchemically transformed from benzene into p-xylene. During the alchemical transformation, the partial charges on benzene atoms were turned off first. Then, the benzene atoms were transformed into p-xylene atoms before the partial charges on p-xylene atoms were turned on. A soft-core Lennard-Jones potential was used during the transformation.⁵⁰ The formula used for both electrostatic potential and the soft-core Lennard-Jones potential is shown in Table S1. The potential energy scaling factors used for each state $\lambda = l_i$ are also shown in Table S1. The free energy $-\beta^{-1} \ln P(\chi^* | \lambda = l_1)$ and the free energy $-\beta^{-1} \ln P(\chi^* | \lambda = l_{16})$ were computed by calculating the PMF with respect to χ when T4 lysozyme binds with benzene (λ = l_1) and with p-xylene ($\lambda = l_{16}$), respectively. The simulations were run inside a TIP3P water box with a size of 79.0 $\text{Å} \times 56.4$ $\rm \mathring{A} \times 55.4 \ \mathring{A}$, and rectangular periodic boundary conditions were used. The water box had 7112 water molecules in total. The CHARMM36 force field⁵¹ was used for T4 lysozyme, and CGenFF⁴² was used for the ligands. The nonbonded interaction options were the same as that used in the relative hydration free energy calculations.

4. RESULTS

4.1. Harmonic System. As shown in Figure 2A, GSLD is able to sample the continuous λ well for both symmetrical and

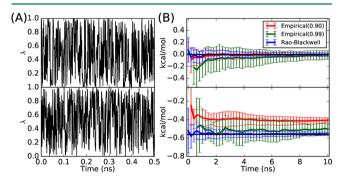


Figure 2. (A) λ trajectories for the symmetrical harmonic system (top) and the asymmetrical harmonic system (bottom). (B) Free energy estimations for the symmetrical (top) and asymmetrical (bottom) systems using the empirical estimators with cutoffs of 0.9 and 0.99 and the RBE. The horizontal black line is the calculated free energy change using numerical integration.

asymmetrical systems. Figure 2B shows the estimated free energy changes ΔG using the RBE and two empirical estimators with cutoff values of 0.9 and 0.99. For the symmetrical system, the true value for the free energy changes is equal to 0 kcal/mol because of the symmetry. The RBE and the empirical estimator with a cutoff of 0.9 converge to 0 kcal/mol within 2 ns, whereas the empirical estimator with a cutoff of 0.99 needs 10 ns of simulation to converge to 0 kcal/mol. Moreover, the RBE has the smallest variance among the three estimators. For the asymmetrical system, the empirical estimator with a cutoff of 0.9 converges to -0.41 ± 0.03 kcal/mol and the empirical estimator with a cutoff of 0.99 converges to -0.50 ± 0.06 kcal/mol, whereas the result from

numerical integration is -0.56 kcal/mol. This shows that the results of empirical estimators can be biased and that the bias depends on the value of the cutoff. Increasing the cutoff value decreases the estimation bias, but it increases the estimation variance because a higher cutoff decreases the number of valid samples used by the empirical estimator. In contrast, the result of RBE converges to -0.56 ± 0.02 kcal/mol, which is closest to the true value and also has the smallest variance. The detailed numerical results can be found in Table S2. Overall, the results suggest that, for this harmonic system, GSLD is able to extensively sample the alchemical states and RBE is better than the empirical estimator in terms of both bias and variance.

4.2. Relative Hydration Free Energies for Three Benzene Derivatives. Results of pairwise GSLD simulations in vacuum and water are shown in Figures S1 and 3,

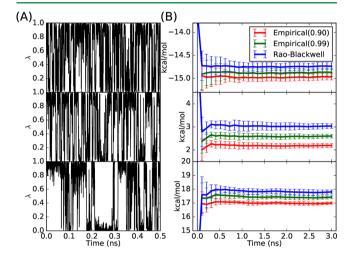


Figure 3. (A) λ trajectories from simulations in water using GSLD for alchemical changes benzene to phenol (top), benzene to benzaldehyde (middle), and phenol to benzaldehyde (bottom). (B) Estimated alchemical free energy changes in water using empirical estimators with different cutoff values and the RBE for alchemical changes benzene to phenol (top), benzene to benzaldehyde (middle), and phenol to benzaldehyde (bottom).

respectively. The pairwise GSLD is able to sample the alchemical states very well for simulations both in vacuum and in water. For the simulation in vacuum, the RBE outperforms empirical estimators in terms of both bias and variance, as in the harmonic system. For the simulation in water, the RBE has a similar variance to that of the empirical estimators because samples from the simulation in water are more correlated than those from the simulations in vacuum. Nevertheless, the RBE is still better than the empirical estimators in terms of the bias. As shown in Figure 3B, the empirical estimator depends on the cutoff. As the cutoff increases from 0.9 to 0.99, the empirical estimator results move toward to the RBE results. As an example, for the alchemical change from benzene to benzaldehyde, when the cutoff increases from 0.9 to 0.99, the empirical estimator result changes from 2.20 \pm 0.08 to 2.60 \pm 0.08 kcal/mol. The RBE result is 3.04 ± 0.09 kcal/mol, which is indistinguishable from the FEP/MBAR result of 3.01 \pm 0.02 kcal/mol. The detailed numerical values from pairwise GSLD and FEP/MBAR can be found in Tables S3 and S4.

The simulation results in vacuum and water from generalized GSLD for multiple ligands are shown in Figures S2 and 4,

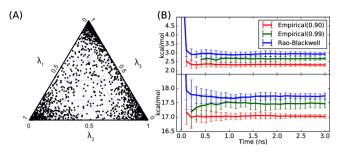


Figure 4. (A) Ternary plot of $(\lambda_1, \lambda_2, \lambda_3)$ sampled using GSLD for multiple ligands in water. (B) Estimated free energy changes in water for alchemical changes, benzene to benzaldehyde (top) and phenol to benzaldehyde (bottom), using an empirical estimator with different cutoff values and RBE.

respectively. The ternary plots⁵² of $(\lambda_1, \lambda_2, \lambda_3)$ trajectories show that the generalized GSLD is able to explore the hybrid ligand configuration space of $(\lambda_1, \lambda_2, \lambda_3)$: the unit simplex $\{(\lambda_1, \lambda_2, \lambda_3)\}$: $|\lambda_3| \sum_{i=1}^3 \lambda_i = 1, \ 0 \le \lambda_i \le 1 \text{ for } i = 1, 2, 3$, in both vacuum and water. In vacuum, the configuration space $(\lambda_1, \lambda_2, \lambda_3)$ is sampled rather uniformly, whereas in water, the configuration space is sampled mostly close to the physical states, i.e., the corners of the ternary plot in Figure 4. This difference is because the biasing potential energy used in this study is a linear biasing potential $\lambda_i G_i^b$. With the linear biasing potential, the biased free energy landscape over the configuration space $(\lambda_1, \lambda_2, \lambda_3)$ in vacuum is almost flat. In water, the corresponding biased free energy landscape is not flat due to the polarization energy of the solvent interacting with reactant and product states, and the biased free energies of the physical states is lower than the intermediate nonphysical states, which explains why the sampled $(\lambda_1, \lambda_2, \lambda_3)$ are mostly around the physical states. On the basis of the trajectory from the generalized GSLD simulation, the calculated free energy using RBE and empirical estimators are shown in Figures S2B and 4B. These results suggests again that, compared with the empirical estimators, the RBE is a better estimator as it has no bias and a smaller variance. The detailed numerical results from the generalized GSLD for multiple ligands is shown in Table S5.

The calculated relative hydration free energies for the three benzene derivatives using pairwise GSLD, generalized GSLD for multiple ligands, and FEP/MBAR methods are combined in Table 1. The results from all three methods agree well with each other. The total simulation time in water for calculating all three relative hydration free energies is 9 ns for pairwise GSLD, 3 ns for generalized GSLD for multiple ligands, and 33 ns for FEP/MBAR methods, which suggests the efficacy of the generalized GSLD for multiple ligands.

4.3. Relative Binding Free Energy of Benzene and p-Xylene with T4 Lysozyme. The λ trajectories from the simulation with T4 lysozyme using pairwise GSLD and the free energy estimations using RBE are shown in Figure 5. For both

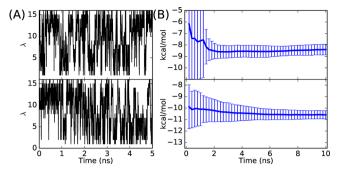


Figure 5. (A) λ trajectories for simulations with T4 lysozyme using pairwise GSLD for $\chi^* = -180^\circ$ (top) and $\chi^* = -60^\circ$ (bottom). (B) Free energy estimation using RBE for $\chi^* = -180^\circ$ (top) and $\chi^* = -60^\circ$ (bottom).

cases where χ is restricted to the trans conformation (χ^* = -180°) and to the gauche ($\chi^* = -60^{\circ}$) conformation, pairwise GSLD is able to sample the alchemical switching variable λ well and the RBE estimations converge in 10 ns of simulation. When χ is restricted to the trans conformation, the estimated free energy converges to -8.40 ± 0.46 kcal/mol. When γ is restricted in the gauche conformation, the estimated free energy converges to -10.60 ± 0.36 kcal/mol. These two free energy estimations are different by 2.20 kcal/mol because the dihedral angle χ is restricted to different conformations. On the basis of the PMF method, in order to get the free energy corresponding to the case where χ is not restricted, the restricting free energies $(-\beta^{-1} \ln P(\chi^* | \lambda = l_1))$ and $-\beta^{-1} \ln P(\chi^* | \lambda = l_1)$ $P(\chi^*|\lambda=l_{16})$) need to be considered and used to correct the free energy $\Delta G(\chi^*)$ using eq 21 in Appendix B. These corrections are shown in Table 2. After the corrections, the estimated free energy ΔG is -9.27 ± 0.50 kcal/mol when $\gamma^* =$ -180° and -9.01 ± 0.40 kcal/mol when $\chi^* = -60^{\circ}$. Therefore, after the corrections, the estimated free energy differences (ΔG) agree very well within statistical uncertainty. On the basis of the these corrected values, the relative binding free energies $(\Delta \Delta G)$ are 0.27 \pm 0.56 and 0.43 \pm 0.46 kcal/mol when χ^* = -180° and $\chi^* = -60^{\circ}$, respectively. These results are close to the relative binding free energy from experiment, which is 0.52 $\pm 0.22 \text{ kcal/mol.}^{37,38}$

5. DISCUSSION AND CONCLUSIONS

Although GSLD and RBE are applied only for calculating relative hydration free energy and relative binding free energy in this study, they could also be used for other purposes. One of the applications would be for calculating the pK_a value of protein amino acids by combining with the constant pH molecular dynamics methods (CPHMD), ^{18,53,54} as several CPHMD methods are based on λ -dynamics. Furthermore, the GSLD framework presented here is not limited to alchemical free energy calculations. The λ variable could be replaced by the pH values, which would correspond to pH generalized

Table 1. Comparison of Relative Hydration Free Energies ($\Delta\Delta G$ in kcal/mol) for The Three Benzene Derivatives^a

		pairwise GSLD	GSLD for multiple ligands	FEP/MBAR
substituents change	$\Delta\Delta G_{ m exp}$	$\Delta\Delta G$ (9 ns)	$\Delta\Delta G$ (3 ns)	$\Delta\Delta G$ (33 ns)
benzene → phenol	-5.77	-4.46 ± 0.08	-4.53 ± 0.15	-4.46 ± 0.03
benzene → benzaldehyde	-3.18	-3.11 ± 0.11	-3.22 ± 0.11	-3.13 ± 0.03
phenol \rightarrow benzaldehyde	2.59	1.39 ± 0.17	1.31 ± 0.10	1.34 ± 0.14

^aThe total simulation time in water for each method is shown in parentheses.

Table 2. Alchemical Free Energy Changes (kcal/mol) between Benzene and p-Xylene Binding with T4 Lysozyme Calculated Using Pairwise GSLD with Corrections from PMFs^a

χ^*	$\Delta G(\chi^*)$	$-\beta^{-1} \ln P(\chi^* \lambda = 0)$	$-\beta^{-1} \ln P(\chi^* \lambda = 1)$	ΔG	$\Delta\Delta G$
trans $(\chi^* = -180^\circ)$	-8.40 ± 0.46	-0.47 ± 0.01	0.4 ± 0.03	-9.27 ± 0.50	0.27 ± 0.56
gauche ($\chi^* = -60^\circ$)	-10.60 ± 0.36	1.14 ± 0.03	-0.45 ± 0.01	-9.01 ± 0.40	0.43 ± 0.46

^aThe alchemical free energy change ΔG in water is -9.44 ± 0.06 kcal/mol and the experimental relative binding free energy $\Delta \Delta G$ is 0.52 ± 0.22 kcal/mol.

ensemble simulations. In these cases, we can also derive the corresponding RBE similarly.

In this study, we have presented the formalism for GSLD and RBE for alchemical free energy calculations. These methods were successfully demonstrated for three test cases of increasing complexity. The GSLD, a generalized ensemble sampling method, works for the case where λ is a discrete variable and for the case where λ is considered to be continuous. When λ is continuous, GSLD can be generalized to calculate free energies for multiple ligands simultaneously in one simulation. RBE not only eliminates the bias problem of the empirical estimator used in the original λ -dynamics but also has smaller estimation variance than the empirical estimator. Moreover, we have also shown that RBE can be used to derive the MBAR/UWHAM equations, which provides new understanding for the MBAR/UWHAM method. $^{21-23}$

APPENDIX

A. Wang—Landau-Like Algorithm To Automatically Determine the Biasing Potential $G_1^{\rm b}$ Used in Pairwise GSLD When λ Is Continuous

The purpose of the biasing potential $G_1^{\rm b}$ used in the pairwise GSLD when λ is continuous is to make the biased free energy landscape over the λ space flat, i.e., to make the simulation spend about equal time at all λ values between 0 and 1. In current study, a linear biasing potential λ $G_1^{\rm b}$ is utilized, because with the linear biasing potential the biased free energy landscape over λ space is quite flat, i.e., the energy barrier between the two physical states $\lambda = 0$ and 1 is small enough that the λ is well sampled across the interval [0, 1]. If the linear biasing potential energy cannot make the biased free energy landscape over the λ space flat enough, a quadratic form of biasing potential can be utilized as in Hayes et al.'s flattening method.¹⁹ The biasing potential $G_1^{\rm b}$ is determined automatically using the following Wang—Landau-like algorithm

- Set the initial biasing potential $G_1^b = 0$ kcal/mol, the decay parameter α such that $0 < \alpha < 1$ ($\alpha = 0.998$ in this study), the biasing potential increment Δ in each step ($\Delta = 2.0$ kcal/mol in this study), and the number of steps R (R = 3000 in this study). Initialize the starting state (λ^0 , $\{x_i^0\}_{i=0}^1$, X^0).
- For t=1 to R: Sample $(\{x_i^t\}_{i=0}^1, X^t)$ from the conditional distribution $P(\{x_i^t\}_{i=0}^1, X^t|\lambda^{t-1})$ by running molecular dynamics simulations and then sample λ^t from the conditional distribution $P(\lambda^t|\{x_i^t\}_{i=0}^1, X^t)$. Set $G_1^b(t) = G_1^b(t-1) + (\lambda^t 0.5) \times \Delta(t)$ and $\Delta(t) = \alpha \times \Delta(t-1)$.
- The final value of G_1^b from the above step is fixed and used as the biasing potential in following simulations.

B. Reformulation of the PMF Method Using Conditional Probability

The PMF method requires prior knowledge of which slow degree of freedom is affecting the free energy calculation. In the context of T4 lysozyme, the slow degree of freedom is the sidechain dihedral angle N–CA–CB–CG1 (χ) of residue Val111. The joint distribution of (χ , λ), $P(\chi, \lambda)$, is of most interest, as it encapsulates all the relevant information required to calculate the free energy $\Delta G = -\beta^{-1} \ln(P(\lambda=l_{16})/P(\lambda=l_{1}))$. On the basis of the chain rule of conditional probability, we have the following equations

$$P(\chi = \chi^*, \lambda = l_{16}) = P(\chi = \chi^* | \lambda = l_{16}) P(\lambda = l_{16})$$

$$= P(\lambda = l_{16} | \chi = \chi^*) P(\chi = \chi^*)$$

$$P(\chi = \chi^*, \lambda = l_1) = P(\chi = \chi^* | \lambda = l_1) P(\lambda = l_1)$$

$$= P(\lambda = l_1 | \chi = \chi^*) P(\chi = \chi^*)$$
(19)

Combining the above two equation gives us

$$\frac{P(\lambda = l_{16})}{P(\lambda = l_1)} = \frac{P(\lambda = l_{16}|\chi = \chi^*)}{P(\lambda = l_1|\chi = \chi^*)} \cdot \frac{P(\chi = \chi^*|\lambda = l_1)}{P(\chi = \chi^*|\lambda = l_{16})}$$
(20)

Therefore, we can calculate the free energy ΔG as

$$\Delta G = -\beta^{-1} \ln \frac{P(\lambda = l_{16})}{P(\lambda = l_{1})}$$

$$= -\beta^{-1} \ln \frac{P(\lambda = l_{16}|\chi = \chi^{*})}{P(\lambda = l_{1}|\chi = \chi^{*})} - \beta^{-1} \ln \frac{P(\chi = \chi^{*}|\lambda = l_{1})}{P(\chi = \chi^{*}|\lambda = l_{16})}$$

$$= \Delta G(\chi = \chi^{*}) + [-\beta^{-1} \ln P(\chi = \chi^{*}|\lambda = l_{1})]$$

$$- [-\beta^{-1} \ln P(\chi = \chi^{*}|\lambda = l_{16})]$$
(21)

where $\Delta G(\chi=\chi^*)$ is alchemical free energy change when χ is fixed at the value χ^* ; $-\beta^{-1} \ln P(\chi=\chi^*|\lambda=l_1)$ is the free energy required to restrict the dihedral angle χ at the value χ^* when T4 lysozyme binds with benzene, i.e., $\lambda=l_1$; $-\beta^{-1} \ln P(\chi=\chi^*|\lambda=l_{16})$ is the corresponding free energy required when T4 lysozyme binds with p-xylene, i.e., $\lambda=l_{16}$. The above equation holds regardless of the value of χ^* .

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jctc.7b00204.

Scaling factors, free energy differences, free energy estimations, biasing potential energies, and relative potential of mean force (PDF)

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Notes

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