

Assignment 3

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Q1.1) a) spherical water droplet



$$\text{at } t \quad r = r_s$$

~~Concave~~

C_As

$$@ r = r_s$$

$$\text{as } r \rightarrow \infty, C_A = C_{A\infty}$$

$$\omega = N_A \cdot 4\pi r^2 = \text{const}$$

$$J_A = -D_{AB} \cdot \frac{dc_A}{dr}$$

Since only the droplet is diffusing,
 $N_B = 0$

$$N_A = N_A \cdot \frac{c_A}{C_{A\infty}} + J_A$$

$$N_A = \frac{C_{A\infty} c}{1 - c_A} \cdot J_A$$

$$\therefore N_A = -\frac{c}{1 - c_A} \cdot D_{AB} \cdot \frac{dc_A}{dr}$$

$$\int_{r_s}^{\infty} \frac{c \omega dr}{4\pi r^2} = - \int \frac{c \cdot D_{AB} dc_A}{1 - c_A}$$

$$\therefore \frac{\omega}{4\pi} \cdot \frac{1}{(1 - c_A)} \left(\frac{1}{r_s} \right) = \frac{c \cdot D_{AB} \ln \left(\frac{c_{A\infty}}{c_A} \right)}{1 - c_A}$$

$$\therefore \frac{\omega}{4\pi r_s} = D_{AB} \cdot c \cdot \ln \left(\frac{c_{A\infty}}{1 - c_A} \right)$$

(2)

$$\therefore \boxed{\omega = 4\pi r_s \cdot D_{AB} \cdot c \ln \left(\frac{c - c_{\infty}}{c - c_{ns}} \right)}$$

Now,

$$\omega = - \frac{d}{dt} \left(\frac{4}{3} \pi r_s^3 \cdot \frac{P_{av}}{M_{av}} \right)$$

$$\therefore \cancel{4\pi} D_{AB} \cdot c \cdot \ln \left(\frac{c - c_{\infty}}{c - c_{ns}} \right) \cdot r_s \\ = - \frac{4}{3} \pi \left(\frac{P}{M} \right)_{av} \cdot \frac{d(r_s^3)}{dt}$$

$$\therefore \int_0^t \cancel{4\pi} D_{AB} \cdot c \ln \left(\frac{c - c_{\infty}}{c - c_{ns}} \right) \cdot \left(\frac{P}{M} \right)_{av} dt \\ = - \int_0^{r_s} \frac{r_s^2 dr_s}{\cancel{4\pi}}$$

$$\therefore \boxed{2 D_{AB} \cdot \left(\frac{P}{M} \right)_{av} \ln \left(\frac{c - c_{\infty}}{c - c_{ns}} \right) t = r_s^2 - r_{s0}^2}$$

(b) Sublimation of cylindrical naphthalen rod.

$$\omega = N_A \cdot 2\pi r \cdot L = \text{const.}$$

$$N_B = 0 \Rightarrow N_A = N_A \cdot \frac{P_n}{P} + J_A$$

$$\therefore N_A = \frac{P}{P - P_A} \cdot J_A$$

$$= \frac{P \cdot D_{AB}}{P - P_A} \cdot \frac{dR_A}{dr}$$

$$\therefore N_A = \frac{-P}{P - P_A} \cdot \frac{D_{AB}}{RT} \cdot \frac{dP_A}{d\gamma}$$

~~$$N_A = \frac{RT}{D_{AB} \cdot P}$$~~

$$\therefore \frac{\omega}{2\pi RL} = - \frac{D_{AB} \cdot P}{RT} \int_{P_A}^P \frac{dP_A}{P - P_A}$$

ρ_{as}

$$\Rightarrow \frac{\omega}{2\pi L} \cdot \ln \left(\frac{\gamma_c + \delta}{\gamma_c} \right) = - \frac{D_{AB} \cdot P}{RT} \ln \left(\frac{P}{P - P_A} \right)$$

$$\omega = - \frac{d}{dt} \left(4\pi r_s^2 \left(\frac{f}{M} \right)_{av} \right)$$

$$\therefore \omega dt = \left(\frac{f}{M} \right)_{av} \cdot 4\pi \cdot 2r_s \cdot dr_s$$

$$\therefore \int_0^t \frac{D_{AB} \cdot P}{RT} \cdot \frac{2\pi L}{\ln \left(\frac{\gamma_c + \delta}{\gamma_c} \right)} \cdot dt = \int_0^{\gamma_s} \left(\frac{f}{M} \right)_{av} \cdot 8\pi r_s dr_s$$

$$\frac{D_{AB} \cdot P}{RT} \cdot \frac{2\pi L}{\ln \left(\frac{\gamma_c + \delta}{\gamma_c} \right)} \cdot t = \left(\frac{f}{M} \right)_{av}^2 \left(\gamma_{s0}^2 - \gamma_s^2 \right)$$

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$$x_c^2 - x_c^0 =$$

DAB · P.L

2 RT

$\left(\frac{P}{RT}\right) \ln\left(\frac{H_2}{H_1}\right)$

an

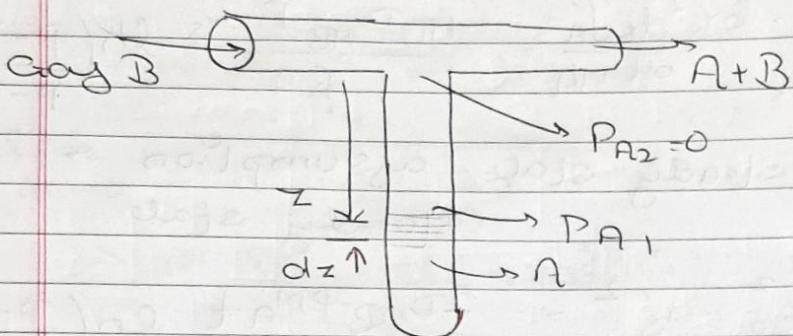
Q5

Assignment 3

Q01.02 Stefan's Tube

A → volatile liquid

B → gas which is not soluble in A.



D of A thru non-D B

$$N_A = (N_A + N_B)$$

$$N_A = D_{AB} \cdot P (P_{A1} - P_{A2})$$

$$\frac{RTz \cdot P_{BM}}{}$$

$$\frac{a \cdot dz \cdot P_A}{M_A} = a \cdot \frac{N_A \cdot dt}{} = \frac{a \cdot D_{AB} \cdot P (P_{A1} - P_{A2})}{RTz \cdot P_{BM}} dt$$

number of
moles of A that
diffuse out

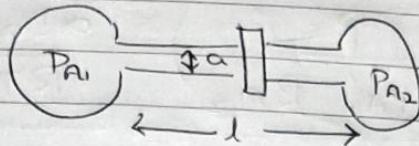
limits $z \rightarrow z_0 \rightarrow z'$
 $t \rightarrow 0 \rightarrow t'$

$$D_{AB} = \frac{RT P_{BM} \ln (z'^2 - z_0^2)}{27 M_A (P_{A1} - P_{A2}) t'}$$

$$P = P_{A_1}' + P_{B_1}' = P_{A_2}' + P_{B_2}'$$

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Q. 13



$$N_A = (N_A + N_B) \cdot \frac{P_A}{P_B} - D_{AB} \frac{dC_A}{dx}$$

$$N_A = -N_B$$

$$\Rightarrow N_A = -D_{AB} \frac{dC_A}{dx}$$

$$\Rightarrow N_A \cdot \int_0^l dC_A = -\frac{D_{AB}}{RT} \frac{P_{A_2}}{P_{A_1}} \int_0^l dP$$

$$\Rightarrow N_A = -\frac{D_{AB}}{RT \cdot l} \cdot (P_{A_2} - P_{A_1})$$

$$N_A = +\frac{D_{AB} (P_{A_1} - P_{A_2})}{RT \cdot l}$$

$$-\nabla_1 \frac{dC_{A_1}}{dt} = a \cdot N_A \quad \text{Inner cross-sectional area}$$

$$\Rightarrow -\nabla_1 \cdot \frac{dP_{A_1}}{RT \cdot dt} = a \cdot N_A$$

$$\text{Hence } \frac{\nabla_2}{RT} \cdot \frac{dP_{A_2}}{dt} = a \cdot N_A$$

$$\therefore \frac{d}{dt} (P_{A_1} - P_{A_2}) = a \cdot N_A \cdot RT \left(\frac{1}{-\nabla_1} - \frac{1}{\nabla_2} \right)$$

$$\frac{d}{dt} (P_{A_1} - P_{A_2}) = -a \cdot N_A \cdot RT \left(\frac{1}{\nabla_1} + \frac{1}{\nabla_2} \right)$$

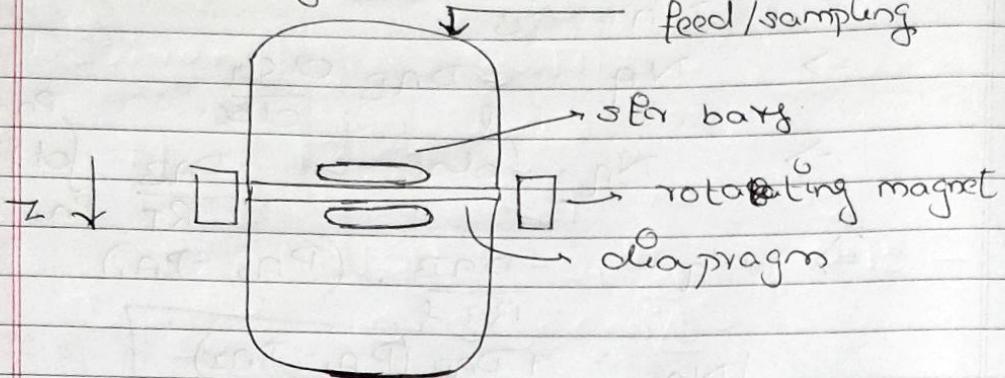
$$\frac{P_{A_1}' - P_{A_2}'}{(P_{A_1} - P_{A_2})} = -\frac{a \cdot D_{AB} (P_{A_1} - P_{A_2})}{RT \cdot l} \left(\frac{1}{\nabla_1} + \frac{1}{\nabla_2} \right)$$

$$\therefore \int \frac{d(P_{A_1}' - P_{A_2}')}{(P_{A_1}' - P_{A_2})} = -\frac{a \cdot D_{AB} (\frac{1}{\nabla_1} + \frac{1}{\nabla_2})}{RT \cdot l} \cdot dt$$

$$\therefore \ln \left(\frac{P}{P_{A_1} - P_{A_2}} \right) = \frac{\alpha D_{AB}}{RTl} \cdot \left(\frac{1}{v_1} + \frac{1}{v_2} \right) \cdot t'$$

→ solve for D_{AB}

Q. 1.4 Diaphragm cell



a = area of cross section

ϵ → porosity of diaphragm

τ → tortuosity factor = $\frac{\text{length of average diffusion path}}{\text{thickness of the diaphragm}}$

If the solution is dilute, we can neglect the bulk flow term.

$$N_A = (N_A + N_B) \tau \alpha_A - D_{AB} \left(\frac{P}{\tau \alpha_A} \right) \frac{dx_A}{dz}$$

put $N_B = 0$

$$\rightarrow N_A = - D_{AB} \left(\frac{P}{\tau \alpha_A} \right) \frac{dx_A}{dz}$$

$$\begin{aligned} \cancel{N_A} &= \cancel{N_A} \cdot \frac{\cancel{C}}{C} - D_{A|B} \frac{\partial \cancel{C}_A}{\partial x} \\ \therefore \cancel{N_A} &= -D_{A|B} \frac{\partial \cancel{C}_A}{\cancel{C}/C} \cdot \frac{\partial C_A}{\partial x} \end{aligned}$$

$$N_A = \frac{(N_A + N_B)Y_A}{c} - D_{A|B} \frac{\partial c_A}{\partial x} \quad \frac{c_A}{c} \approx 0$$

$$\therefore N_A = -D_{A|B} \frac{\partial c_A}{\partial x}$$

$$\therefore N_A = -\frac{D_{A|B}}{t} \cdot (c_{A_1}' - c_{A_2}')$$

conc. at time t

$$-\nu_1 \frac{dc_{A_1}}{dt} = a \cdot N_A \cdot \epsilon$$

$$\nu_2 \frac{dc_{A_2}}{dt} = a \cdot \epsilon \cdot N_A$$

$$\frac{dc_{A_1}' - c_{A_2}'}{dt} = -a \epsilon N_A \left(\frac{1}{\nu_1} + \frac{1}{\nu_2} \right)$$

$$\therefore \int \frac{dc_{A_1}' - c_{A_2}'}{c_{A_1}' - c_{A_2}'} dt = \int \frac{t}{t + a \epsilon D_{A|B} \left(\frac{1}{\nu_1} + \frac{1}{\nu_2} \right)} dt$$

$$c_{A_1 0} - c_{A_2 0}$$

$$\therefore \ln \left(\frac{c_{A_1 t} - c_{A_2 t}}{c_{A_1 0} - c_{A_2 0}} \right) = \frac{a \epsilon \bullet D_{A|B}}{t} \cdot t \left(\frac{1}{\nu_1} + \frac{1}{\nu_2} \right)$$

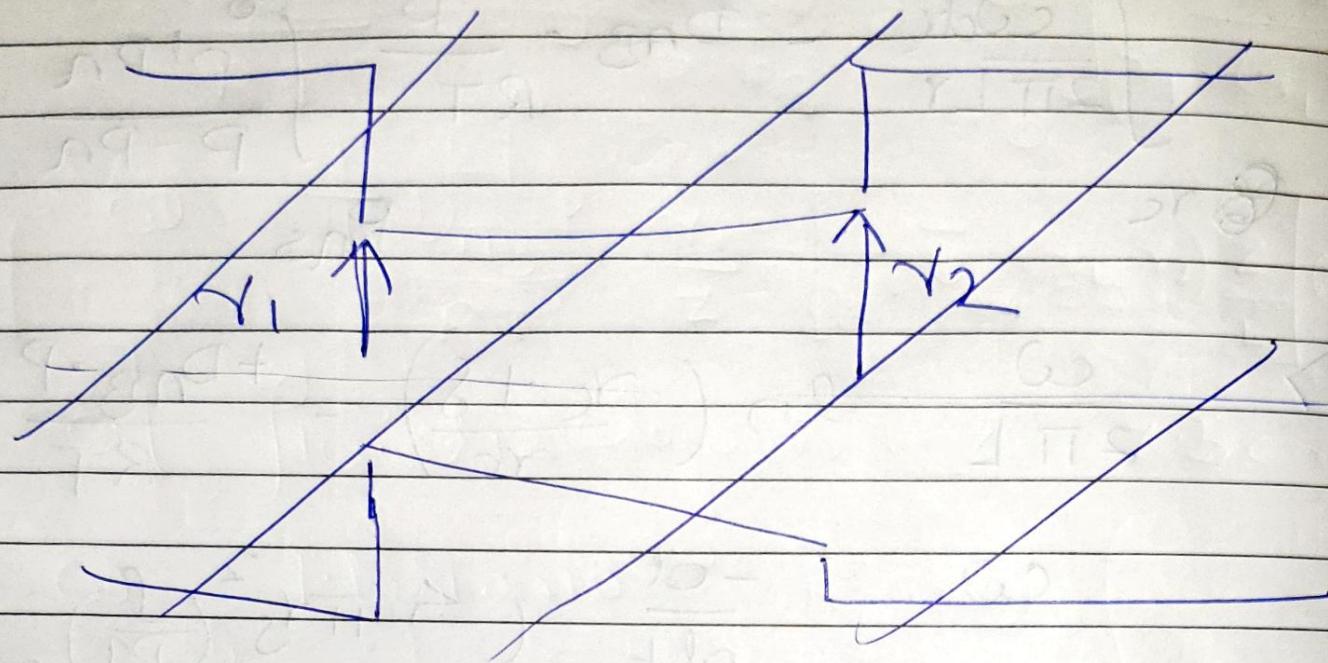
$$= \beta \cdot D_{A|B} \cdot t$$

sove for $D_{A|B}$

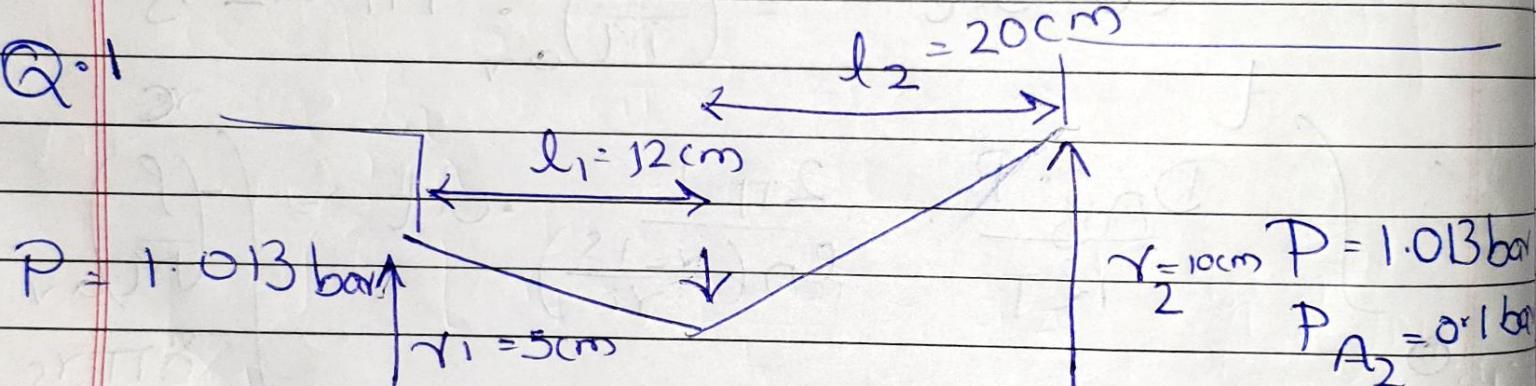
cell constant

Section 2

Q



Q.1



$$P_{A1} = 0.3 \text{ bar}$$

$$2\gamma = 5 \text{ cm}$$

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Using the condition of continuity of rate of transport

$$\omega = \frac{\pi D_{AB}}{RT} \cdot \frac{\gamma_1 \gamma}{l_1} (P_{A1} - P_{An})$$

$$= \frac{\pi \cdot D_{AB}}{RT} \cdot \frac{\gamma_2 \gamma}{l_2} (P_{An} - P_{A2})$$

where

$$P_{An} = \frac{\gamma_1 \cdot l_2 \cdot P_{A1} + \gamma_2 \cdot l_1 \cdot P_{A2}}{\gamma_1 \cdot l_2 + \gamma_2 \cdot l_1}$$

$$\omega = \frac{\pi \cdot D_{AB}}{RT} \cdot \frac{\gamma_1 \gamma_2 \cdot \gamma}{\gamma_1 \cdot l_2 + \gamma_2 \cdot l_1} (P_{A1} - P_{A2})$$

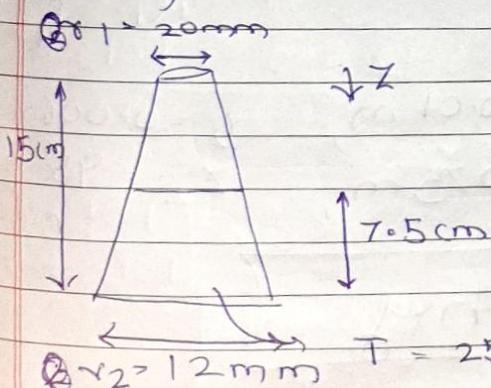
$$\left[\frac{dP_B}{dz} \right]_n = - \left[\frac{dP_n}{dz} \right]_n = \left[\frac{\omega}{\pi r^2} \cdot \frac{RT}{D_{AB}} \right]_n$$

$$P_{An} = \frac{5 \times 10^{-2} \times 20 \times 10^{-2} \times 0.3 + 10 \times 10^{-2} \times 12 \times 10^{-2} \times 0.1}{5 \times 10^{-2} \times 20 \times 10^{-2} + 10 \times 10^{-2} \times 12 \times 10^{-2}}$$

$$= 0.19 \text{ bar}$$

$$\therefore \boxed{\omega = 2.26 \times 10^{-6} \frac{\text{mol}}{\text{s}}}$$

Q.2 Diffusion through a tapered region



(the question says diameters are 12, 20 mm, but the have actually taken these to be the radii)

$$T = 25^\circ \text{C} = 298 \text{ K}$$

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$$\omega = \frac{\rho \pi r^2}{M_A} N_A$$

$$N_A = \frac{D_{AB} \cdot P}{RT(P-P_A)} \cdot \frac{dP_n}{dz} \quad (1)$$

$$\gamma = \gamma_1 - \frac{\gamma_1 - \gamma_2}{L} z \quad (2)$$

$$\frac{\pi \cdot D_{AB} \cdot P}{RT} \cdot \int_0^{P_A} \frac{dP_n}{P - P_A} = \omega \int_0^z \frac{dz}{(\gamma_1 - (\gamma_1 - \gamma_2) \frac{z}{L})^2}$$

$$\omega = \frac{\pi \cdot D_{AB} \cdot P}{RT} \cdot \frac{\gamma_1}{z} \cdot \left[\gamma_1 - (\gamma_1 - \gamma_2) \frac{z}{L} \right] \ln \left(\frac{P}{P - P_A} \right)$$

at fluid depth ($z = 7.5 \text{ cm}$),

$$\omega = 1.58 \times 10^{-10} \frac{\text{kg mol}}{\text{sec}}$$

$$\frac{f_A}{M_A} \cdot \frac{\pi r^2}{M_A} \cdot \frac{dz}{dt} = \omega$$

$$\sum_1^2 \int z \left[\gamma_1 - \frac{(\gamma_1 - \gamma_2)}{L} \cdot z \right] dz = \frac{D_{AB} \cdot P}{RT} \cdot \frac{M_A}{f_A \cdot r_1} \ln \left(\frac{P}{P - P_A} \right)$$

$$\text{Putting } \gamma_1 = 0.01 \text{ m}, \gamma_2 = 0.006 \text{ m}, \\ z_1 = 7.5 \text{ cm} = 0.075 \text{ m}, z_2 = 95 \text{ cm} (= 1.2 \text{ m}) \\ = 0.095 \text{ m}$$

$$\Rightarrow t = 79 \text{ hrs} \quad ($$

Q.3 Diffusion from a sphere

$$4g \rightarrow 1g$$

$$T = 45^\circ C \quad P = 1.013 \text{ bar}$$

$$D_{AB} = 0.92 \times 10^{-6} \frac{\text{m}^2}{\text{s}} \quad f = 1140 \frac{\text{kg}}{\text{m}^3}$$

$$P_{AS} = P_{S\text{lab}} = 0.8654 \text{ mmHg} = 0.00114 \text{ atm}$$

$$\begin{array}{ll} \text{Initial mass} = 4g \Rightarrow \gamma_{S0} = 0.943 \times 10^{-2} \\ \text{final mass} = 1g \Rightarrow \gamma_S' = 0.594 \times 10^{-2} \end{array}$$

$$\gamma_{S0}^2 - \gamma_S'^2 = \frac{2 D_{AB} \cdot P \cdot \ln \frac{t'}{t}}{RT \cdot \ln \frac{P - P_{AS}}{P - P_{S0}}} \quad \stackrel{t=0}{\cancel{t}}$$

Substituting, we get

$$t' = 219.5 \text{ hr}$$

$$\begin{aligned} Q.4 \quad & \frac{\gamma_{C_1}^2}{2} \cdot \ln \left(1 + \frac{S}{\gamma_{C_1}} \right) = \frac{\gamma_{C_2}^2}{2} \ln \left(1 + \frac{S}{\gamma_{C_2}} \right) \\ & + \frac{S}{2} \left[(\gamma_{C_1} - \gamma_{C_2}) S \ln \left(\frac{\gamma_{C_1} + S}{\gamma_{C_2} + S} \right) \right] \\ & = D_{AB} \cdot \frac{P}{RT} \cdot \frac{\gamma \ln \left(\frac{P}{P - P_{AS}} \right)}{\beta_A} \cdot t \end{aligned}$$

$$\text{Initial mass} = 10g, \quad f = 1.14 \text{ g/cm}^3$$

$$\Rightarrow \left[\gamma_{C_1} = 0.3736 \text{ cm} \right], \quad \text{final volume} = \frac{\pi \gamma_{C_1} \cdot L}{2} = 1.386 \text{ cm}^3$$

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$$\therefore | r_{c_2} = 0.2642 \text{ cm} |$$

given that $\delta = 3 \text{ mm}$, $D_{AB} = 0.0692 \text{ cm}^2$
 $P = 1 \text{ atm}$, $P_{AS} = 0.081 \text{ atm}$

substituting, we get \rightarrow

$$| t = 10.35 \text{ h} |$$

Section 3

Q. 3.1] $r_s = 2 \text{ mm}$ $P_{AS} = 0.555 \text{ mm Hg}$
 $T = 318 \text{ K}$ $= 7.33 \times 10^{-5} \text{ Pa}$
 $D_{AB} = 6.92 \times 10^{-6} \frac{\text{m}^2}{\text{sec}}$.

$$P = 1 \text{ atm.}$$

We have derived the following relation for evaporation of naphthalene ball \rightarrow

$$\omega = 4\pi r_s \cdot D_{AB} \cdot \frac{P}{RT} \ln \left(\frac{P - P_{AS}}{P - P_{AS}} \right)$$

$$= 4\pi \times 2 \times 10^{-3} \times 6.92 \times 10^{-6} \times 1.013 \times 10^5$$

$$8.314 \times$$

$$\times \ln \left(\frac{1.013 \times 10^5 - 0}{1.013 \times 10^5 - 7.33 \times 10^{-5}} \right)$$

$$\omega = 4.86 \times 10^{-12} \frac{\text{kg mol}}{\text{sec}}$$

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Q 3.3]

 $H_2O \leftarrow$
 $T = 283K \quad z \uparrow$

$c_{A2} = 0.02 \frac{mol}{m^3}$

4mm

$c_{A01} = 0.1 \frac{mol}{m^3}$

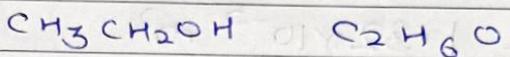
↑

Case 1 : water film.

$D_{AB} = 13.26 \times 10^{-5} \times (\nu_B)^{-1.14} \times (\nu_A)^{-0.589}$

↓

 $\begin{matrix} \text{ethanol} \\ \text{H}_2O \end{matrix}$
 $= 1.4 cP$



$\nu_A = 2\nu_C + 6\nu_H + \nu_O$
 $= 2 \times 14.8 + 6 \times 3.7 + 7.4$

$\therefore \nu_A = 59.2 \frac{cm^3}{gmol}$

$P_{AB} = 8.166 \times 10^{-6} \frac{cm}{s}$

Assume dilute soln $N_A + N_B = 0$

$\Rightarrow N_A = \frac{D_{AB}}{L} (c_{A1} - c_{A2})$

$= \frac{8.166 \times 10^{-6} \times 10^{-4}}{4 \times 10^{-3}} \times (0.1 - 0.02)$

$N_A = 1.63 \times 10^{-8} \frac{mol}{m^2 s}$

take $c_A = c_1 z + c_2$

boundary condition

$c_A(z=0) = c_{A1}$
 $c_A(z=\ell) = c_{A2}$

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$$\left. \begin{aligned} C_A &= \frac{C_{A2} - C_{A1}}{l} \cdot z + C_{A1} \\ C_A &= -20z + 0.1 \end{aligned} \right\}$$

Diffusion in air

$$N_A = -D_{AB} \cdot \frac{c}{l} \ln \left(\frac{c - C_{A2}}{c - C_{A1}} \right)$$

$$D_{AB} = 1.32 \times 10^{-5} \frac{m^2}{s}$$

assuming ideal gas $c = \frac{P}{RT} = \frac{105}{8.314 \times 283}$

(assuming): $c = 43.05 \frac{\text{mol}}{\text{m}^3}$

$$C_{A1} = 0.1 \Rightarrow P_{A1} = 0.1 RT$$

$$C_{A2} = 0.02 \Rightarrow P_{A2} = 0.02 RT$$

$$N_A = - \frac{1.32 \times 10^{-5} \times 43.05}{4 \times 10^{-3}} \times \ln \left(\frac{43.05 - 0.1}{43.05 - 0.02} \right)$$

$$N_A = 2.64 \times 10^{-4} \frac{m^2}{s}$$

Concentration profile would
of course be same)

>>> N_A water

much faster diffusion.

$$Q 3.4] P = 10^5 \frac{N}{m^2}$$

$$P_{A0} = 13000 \frac{N}{m^2} \quad P_{Al} = 6500 \frac{N}{m^2}$$

$$l = 2 \text{ mm}$$

$$D_{AB} = 1.87 \times 10^{-5} \frac{m^2}{s}$$

$$N_A = \frac{D_{AB} \cdot P}{RTl} \cdot \ln\left(\frac{P - P_{Al}}{P - P_{A0}}\right)$$

$$N_A = 5.93 \times 10^{-5} \frac{\text{kg mol}}{m^2 s}$$

$$\omega_A = N_A \cdot A \rightarrow = 1 \text{ m}^2$$

$$\omega_A = 5.93 \times 10^{-5} \frac{\text{kg mol}}{m^2 s}$$