

Figure 1: Overview of the system simulated using the presented MATLAB code

Physbased Diffusion Embrittlement: a MATLAB code to accurately predict hydrogen embrittlement within the phase-field framework

Tim Hageman*, Emilio Martínez-Pañeda

Department of Civil and Environmental Engineering, Imperial College London, London SW7 2AZ, UK

Abstract

Documentation that accompanies the *MATLAB* code Physbased Diffusion Embrittlement, available from here. This documentation explains the usage of the implemented finite element framework, and highlight the main files. Special attention is paid to the parts of the code that implement the physics-based diffusion model for electrolytes within cracks, and how the crack opening height is obtained.

If using this module for research or industrial purposes, please cite: T. Hageman & E. Martínez-Pañeda, A phase field-based framework for electro-chemo-mechanical fracture: crack-contained electrolytes, chemical reactions and stabilisation. Computer Methods in Applied Mechanics and Engineering XX (2023) YY [1].

update citeMe with correct reference once main paper is published

Keywords: MATLAB, electrochemistry, phasefield, hydrogen embrittlement, finite element method

Contents

1 Introduction		
	1.1 Basic usage	2
	Summary of included files	3
	2.1 main.m	3

Email address: t.hageman@imperial.ac.uk (Tim Hageman)

Preprint submitted to July 11, 2023

^{*}Corresponding author

	2.2	Models	3
		2.2.1 BaseModel	4
		2.2.2 Constrainer	4
		2.2.3 LinearElastic	4
		2.2.4 HydrogenDiffusionUnDamaged	5
		2.2.5 Electrolyte	
		2.2.6 ElectrolyteInterface	
		2.2.7 PhasefieldDamage	7
		2.2.8 HydrogenDiffusion	g
		2.2.9 PhasefieldElectrolyte	C
	2.3	Mesh	1
	2.4	Shapes	1
	2.5	Physics	1
	2.6	Dofspace	2
	2.7	Solver	3
3	\mathbf{Spe}	ifics: Fracture opening height	4
4	Sam	ple results	6
	4.1	Static cracks	6
	4.2	Propagating cracks	6

1. Introduction

Development of cracks due to interactions with the environment is a common failure mechanism, with causes of this environment-induced failure including stress-corrosion-cracking and hydrogen embrittlement. Accurately predicting these phenomena does not only require capturing the metal and neighbouring electrolyte, but also requires the effects of newly created and propagating cracks to be captured. One framework enabling crack propagation in arbitrary directions and geometries is the phase field method, representing cracks in a smeared manner. While the mechanical side of phase field fracture models is well-developed, models to represent fluids and species contained within the cracks are usually of a more empirical nature (for instance, requiring fitting parameters that are problem-dependent). Here, we present a MATLAB implementation of a physics-based model for ionic diffusion and reactions within cracks compatible with the phase field framework. This physics-based model is described in T. Hageman & E. Martínez-Pañeda, A phase field-based framework for electro-chemo-mechanical fracture: crack-contained electrolytes, chemical reactions and stabilisation. Computer Methods in Applied Mechanics and Engineering XX (2023) YY [1], where its ability to accurately predict hydrogen uptake and expected failure times is shown. In the remainder of this documentation, the physical models and mathematical objects included within this code will be detailed. Special attention is given to the manner in which crack opening heights are reconstructed, as this requires procedures that are non-standard in finite element methods.

Note: as this code is developed along other codes, parts of this code (and the documentation) correspond to the codes accompanying [2] (hydrogen absorption stabilised with lumped integration) and [3] (corrosion under charge conservation conditions).

1.1. Basic usage

For simulating the model as provided, running the function "main_ Static.m" performs all required actions for simulating a static fracture: It automatically generates the geometry and mesh, initialises all simulation components, and prints outputs to the screen and saves them to a folder within results. Simple changes, e.g. editing parameters, can be done within main.m without requiring altering other files. Similarly, "main.m" performs the simulations for a dynamic fracture. Files are also provided for the post-processing: "Animations.m" to generate time-series for the output files, and "compare.m" to plot multiple simulations at a set time.

2. Summary of included files

The code is set up in a object-oriented manner, defining matlab classes for each sub-component and providing their accompanying methods. As a result, a clear distinction is made between different components, and each can be used and altered with limited/no impact on other components. Here, the different classes are described. The commenting style employed within the code is compatible with the matlab help function, as such information about all usable methods within a class can be accessed by including the relevant folders, and typing, for instance, "help Solver" to print all variables contained within and all function available from the solver.

2.1. main.m

74

78

79

178

182 183 This is the main file, from which all classes are constructed and the actual simulation is performed. Within it, all properties used within other classes are defined as inputs, for instance for the momentum balance and phase-field evolution equations within the solid domain:

```
main.m
%Momentum balance and phase-field evolution
physics_in{1}.type = "PhaseFieldDamage";
physics_in{1}.Egroup = "Internal";
physics_in{1}.young = 200e9;
                                 %Youngs Modulus [Pa]
physics_in{1}.poisson = 0.3;
                                 %Poisson ratio [-]
physics_in\{1\}.kmin = 1e-10;
                                 %residual stiffness factor [-]
physics_in{1}.l=1;
                                 %Phase-field length scale [m]
physics_in{1}.Gc=2.0e3;
                                 %Fracture release energy [J/m^2]
physics_in{1}.GDegrade = 0.9;
                                 %Maximum hydrogen degradation factor
physics_in{1}.NL = 1e6;
                                 %Concentration of interstitial lattice sites [mol/m
    ^31
physics_in{1}.gb = 30e3;
                                %Grain boundary binding energy [J/mol]
```

where "physics_in" is the array of options (in this case, physical models) passed to the physics object at construction.

The actual time-dependent simulations are also performed within this file:

```
main.m
for tstep = startstep:n_max
    disp("Step: "+string(tstep));
    disp("Time: "+string(physics.time));
    physics.dt = dt*1.05^(min(100,tstep-1));
    disp("dTime: "+string(physics.dt));

%solve for current time increment
    solver.Solve();
```

Notably, while this performs the time-stepping scheme and controls the time increment size and termination of the simulations, it does not by itself solve anything, instead calling the "solver.Solve()" function which performs a Newton-Raphson procedure using the parameters used to initialize the class, and once the current timestep is converged returns to the main code.

2.2. Models

The files included within the Models folder form the main implementation of all the physical phenomena involved. They implement the assembly of tangential matrices and force vectors, when requested by the solving procedures, and store model-specific parameters.

2.2.1. BaseModel

This is an empty model, inherited by all other models to provide consistency within the available functions. While empty within here, the potential functions that can be defined within other models include assembling the system matrix and force vector:

Models/@BaseModel/BaseModel.m

```
function getKf(obj, physics, stp)
```

and committing history dependent or path dependent variables:

```
function Commit(obj, physics, commit_type)
```

where the keyword "commit_type" indicates the type of history or path dependence to commit at the current point. It also provides a function for performing once per staggered solution step actions:

```
function OncePerStep(obj, physics, stp)
```

2.2.2. Constrainer

30

90

91

93

This model is used to apply fixed boundary constraints to a degree of freedom at a set location. Within the main file, the inputs required are:

main.m

```
%displacement constrain at bottom boundary
physics_in{3}.type = "Constrainer";
physics_in{3}.Ngroup = "Bottom";
physics_in{3}.dofs = {"dy"};
physics_in{3}.conVal = [0];
```

and multiple definitions of this model are allowed, allowing for constraints to be applied to several element groups. These constraints are integrated within the tangential matrix and force vector through allocation matrices C_{con} and C_{uncon} , reordering the system into a constrained and unconstrained part. This allows the constrained system to be solved as:

$$C_{uncon}^{T}KC_{uncon}\mathbf{y} = -\left(C_{uncon}^{T}\mathbf{f} + C_{uncon}^{T}KC_{con}\mathbf{c}\right)$$
(1)

with the values of the boundary constraints contained in the vector \mathbf{c} . After solving, the state vector is then incremented through:

$$\mathbf{x}^{new} = \mathbf{x}^{old} + C_{uncon}\mathbf{y} + C_{con}\mathbf{c} \tag{2}$$

2.2.3. LinearElastic

The linear-elastic model implements the momentum balance for the metal domain:

$$\nabla \cdot \boldsymbol{\sigma} = \mathbf{0} \tag{3}$$

where the stresses σ are based on the displacement $\mathbf{u} = [\text{"dx" "dy"}]$. The properties used to initialize this model given as input by:

main.m

```
physics_in{1}.type = "LinearElastic";
physics_in{1}.Egroup = "Metal";
physics_in{1}.young = 200e9;
physics_in{1}.poisson = 0.3;
```

Notably, since the tangential matrix for linear-elasticity is constant, it is assembled once and saved locally within the model, after which during the global matrix assembly process, it is copied over to the global matrix:

Models/@LinearElastic/LinearElastic.m

with the force vector also being updated based on this locally saved stiffness matrix.

2.2.4. HydrogenDiffusionUnDamaged

This model implements the hydrogen mass conservation, through the diffusion equation [4, 5, 6, 7, 8]:

$$\left(1 + \frac{N_{\mathrm{T}}/N_{\mathrm{L}}\exp\left(\Delta g_{\mathrm{b}}/RT\right)}{\left(C_{\mathrm{L}}/N_{\mathrm{L}} + \exp\left(\Delta g_{\mathrm{b}}/RT\right)\right)^{2}}\right)\dot{C}_{\mathrm{L}} - \boldsymbol{\nabla}\cdot\left(\frac{D_{\mathrm{L}}}{1 - C_{\mathrm{L}}/N_{\mathrm{L}}}\boldsymbol{\nabla}C_{\mathrm{L}}\right) + \boldsymbol{\nabla}\cdot\left(\frac{D_{\mathrm{L}}C_{\mathrm{L}}\overline{V}_{\mathrm{H}}}{RT}\boldsymbol{\nabla}\sigma_{\mathrm{H}}\right) = 0$$
(4)

with the interstitial lattice hydrogen concentration C_L indicated within the code by "CL". Input properties for this model constitute:

mainDiscreteFrac.m

```
physics_in{2}.type = "HydrogenDiffusionUnDamaged";
physics_in{2}.Egroup = "Metal";
physics_in{2}.DL = 1e-9;
physics_in{2}.NL = 1e6;
physics_in{2}.gb = 30e3;
physics_in{2}.NT = 1e2;
```

This model presumes the first model provides the Young's modulus and Poisson ratio. The hydrostatic stress gradients required in this model are obtained as $\nabla \sigma_h = B_u^* \mathbf{u}$, with the B^* matrix defined as:

$$\boldsymbol{B}_{u}^{*} = E/(3(1-2\nu)) \begin{bmatrix} \frac{\partial^{2} N_{u1}}{\partial x^{2}} & \frac{\partial^{2} N_{u2}}{\partial x^{2}} & \dots & \frac{\partial^{2} N_{u1}}{\partial x \partial y} & \frac{\partial^{2} N_{u2}}{\partial x \partial y} & \dots \\ \frac{\partial^{2} N_{u1}}{\partial x \partial y} & \frac{\partial^{2} N_{u2}}{\partial x \partial y} & \dots & \frac{\partial^{2} N_{u1}}{\partial y^{2}} & \frac{\partial^{2} N_{u2}}{\partial y^{2}} & \dots \end{bmatrix}$$
 (5)

2.2.5. Electrolyte

The electrolyte model implements the Nernst-Planck mass balance for non-crack contained electrolytes [?]:

$$\dot{C}_{\pi} + \nabla \cdot (-D_{\pi} \nabla C_{\pi}) + \frac{z_{\pi} F}{RT} \nabla \cdot (-D_{\pi} C_{\pi} \nabla \varphi) + R_{\pi} = 0$$
(6)

for the ionic species and their name within the model file: H⁺ ("H"), OH⁻ ("OH"), Na⁺ ("Na"), Cl⁻ ("Cl", using lower case l; upper case L provides the lattice hydrogen concentration), Fe²⁺ ("Fe"), and FeOH⁺ ("FeOH"). Additionally, it implements the electro-neutrality condition [9, 10]:

$$\sum z_{\pi}C_{\pi} = 0 \tag{7}$$

and bulk reactions:

$$H_2O = \frac{k_w}{k_w'} H^+ + OH^-$$
(8)

$$Fe^{2+} + H_2O \xrightarrow{k_{fe}} FeOH^+ + H^+$$

$$(9)$$

$$FeOH^{+} + H_{2}O \xrightarrow{k_{feoh}} Fe(OH)_{2} + H^{+}$$

$$\tag{10}$$

with reaction rates:

$$R_{H^{+}w} = R_{OH^{-}} = k_{w}C_{H_{2}O} - k'_{w}C_{H^{+}}C_{OH^{-}} = k_{eq}(K_{w} - C_{H^{+}}C_{OH^{-}})$$
(11)

$$R_{\text{Fe}^{2+}} = -k_{fe}C_{\text{Fe}^{2+}} + k'_{fe}C_{\text{FeOH}^{+}}C_{\text{H}^{+}}$$
(12)

$$R_{\text{FeOH}^{+}} = k_{fe}C_{Fe^{2+}} - C_{\text{FeOH}^{+}}(k_{feoh} + k'_{fe}C_{\text{H}^{+}})$$
(13)

$$R_{H^{+},fe} = k_{fe}C_{Fe^{2+}} - C_{FeOH^{+}}(k'_{fe}C_{H^{+}} - k_{feoh})$$
(14)

For this model, the input properties required are:

mainDiscreteFrac.m

This model employs a lumped integration scheme when the vector "Lumped" contains true. Details for the implementation of this lumped scheme are given in [2].

2.2.6. ElectrolyteInterface

The electrolyteInterface model implements the metal-electrolyte coupling through the surface reactions for explicitly represented electrolytes [11, 8]:

Volmer (acid):
$$H^{+} + M + e^{-\frac{k_{Va}}{k'_{Va}}} MH_{ads}$$
 (15)

Heyrovsky (acid):
$$H^+ + e^- + MH_{ads} \frac{k_{Ha}}{k'_{Ha}} M + H_2$$
 (16)

Volmer (base):
$$H_2O + M + e^{-\frac{k_{Vb}}{k'_{Vb}}} MH_{ads} + OH^{-}$$
(17)

Heyrovsky (base):
$$H_2O + e^- + MH_{ads} \frac{k_{Hb}}{k'_{Hb}} M + H_2 + OH^-$$
 (18)

Tafel:
$$2MH_{ads} \xrightarrow{k_T} 2M + H_2$$
 (19)

Absorption:
$$MH_{ads} \xrightarrow{k_A} MH_{abs}$$
 (20)

Corrosion:
$$\operatorname{Fe}^{2+} + 2e^{-\frac{k_c}{k_c'}}\operatorname{Fe}$$
 (21)

with reaction rates:

Forward Backward

Volmer(acid):
$$\nu_{Va} = k_{Va} C_{H^+} (1 - \theta_{ads}) e^{-\alpha_{Va} \frac{\eta F}{RT}}$$
 $\nu'_{Va} = k'_{Va} \theta_{ads} e^{(1 - \alpha_{Va}) \frac{\eta F}{RT}}$ (22)

$$\text{Heyrovsky(acid)}: \quad \nu_{Ha} = k_{Ha} C_{\text{H}^{+}} \theta_{ads} e^{-\alpha_{Ha} \frac{\eta F}{RT}} \qquad \qquad \nu'_{Ha} = k'_{Ha} (1 - \theta_{ads}) p_{\text{H}_{2}} e^{(1 - \alpha_{Ha}) \frac{\eta F}{RT}}$$
 (23)

$$Volmer(base): \quad \nu_{Vb} = k_{Vb}(1 - \theta_{ads})e^{-\alpha_{Vb}\frac{\eta F}{RT}} \qquad \qquad \nu'_{Vb} = k'_{Vb}C_{OH} - \theta_{ads}e^{(1 - \alpha_{Vb})\frac{\eta F}{RT}}$$

$$(24)$$

Heyrovsky(base):
$$\nu_{Hb} = k_{Hb}\theta_{ads}e^{-\alpha_{Hb}\frac{\eta F}{RT}}$$
 $\nu'_{Hb} = k'_{Hb}(1 - \theta_{ads})p_{\rm H_2}C_{\rm OH} - e^{(1-\alpha_{Hb})\frac{\eta F}{RT}}$ (25)

Tafel:
$$\nu_T = k_T |\theta_{ads}| \theta_{ads} \qquad \qquad \nu_T' = k_T' (1 - \theta_{ads}) p_{\rm H_2} \qquad (26)$$

Absorption:
$$\nu_A = k_A (N_L - C_L) \theta_{ads}$$
 $\nu'_A = k'_A C_L (1 - \theta_{ads})$ (27)

Corrosion:
$$\nu_c = k_c C_{\text{Fe}^2} + e^{-\alpha_c \frac{\eta F}{RT}} \qquad \qquad \nu_c' = k_c' e^{(1-\alpha_c) \frac{\eta F}{RT}}$$
 (28)

These reaction rates are implemented in a separate function from the matrix assembly:

Models/@ElectrolyteInterface/ElectrolyteInterface.m

```
function [react, dreact, products] = reactions(obj, CH, COH, CFE, theta, phil, CLat
    )
```

which takes the local hydrogen, hydroxide, and iron concentrations, the surface coverage, electrolyte potential, and interstitial lattice hydrogen concentration. It functions for both the integration-point variables as

well as for the nodal values. In addition to the reaction rates, the electrolyte interface model also resolves the surface mass balance:

$$N_{ads}\dot{\theta}_{ads} - (\nu_{Va} - \nu'_{Va}) + \nu_{Ha} + 2\nu_T + (\nu_A - \nu'_A) - (\nu_{Vb} - \nu'_{Vb}) + \nu_{Hb} = 0$$
(29)

For this model, the input variables to define are given as:

mainDiscreteFrac.m

```
physics_in{8}.type = "ElectrolyteInterface";
                physics_in{8}.Egroup = "Interface";
                physics_in{8}.NAds = 1e-3;
                physics_in{8}.k = k;
                physics_in{8}.NL = 1e6;
                physics_in{8}.Em = Em;
                physics_in{8}.Lumped = [1 1 1 1 1 1];
108
 9
            k = [1e-4,
                        1e-10,
                                0.3,
                1e-10,
                        Ο,
                                Ο,
                1e-6.
                        Ο,
                1e1,
                        7e5,
                        1e-13,
                1e-8,
14
                1e-10,
                        1e-14,
                                0.3,
                3e-5/(2*96485.3329), 3e-5/(2*96485.3329), 0.5, -0.4];
```

with the vector "Lumped" allowing for individual interface reactions to be either integrated using a standard Gauss integration scheme (0) or a lumped integration scheme (1). the reaction constants matrix k is defined as:

$$k = \begin{bmatrix} k_{Va} & k'_{Va} & \alpha_{Va} & E_{eq,Va} \\ k_{Ha} & k'_{Ha} & \alpha_{Ha} & E_{eq,Ha} \\ k_{T} & k'_{T} & - & - \\ k_{A} & k'_{A} & - & - \\ k_{Vb} & k'_{Vb} & \alpha_{Vb} & E_{eq,Vb} \\ k_{Hb} & k'_{Hb} & \alpha_{Hb} & E_{eq,Hb} \\ k_{c} & k'_{c} & \alpha_{c} & E_{eq,c} \end{bmatrix}$$

$$(30)$$

with the empty entries not used within the model.

2.2.7. PhasefieldDamage

This model implements the momentum conservation for a damaged linear-elastic material:

$$0 = \nabla \cdot \left(d(\phi) \frac{\partial \psi_0}{\partial \varepsilon} \right) \tag{31}$$

and the phase-field evolution equation [12, 13, 14, 15]:

$$\frac{\phi}{\ell} - \ell \nabla^2 \phi = -\frac{\partial d(\phi)}{\partial \phi} \mathcal{H} \quad \text{where} \quad \mathcal{H} = \frac{\psi_0}{G_c(C_L)}, \quad \dot{\mathcal{H}} >= 0$$
 (32)

using the linear-elastic energy function for an undamaged material:

$$\psi_0 = \varepsilon : \mathcal{D} : \varepsilon \tag{33}$$

and the damage function:

$$d(\phi) = k_0 + (1 - k_0)(1 - \phi)^2 \tag{34}$$

Within this model, the displacement and phase-field damage variable can not be updated at the same time, instead requiring Eqs. (31) and (32) to be set to different staggered solver steps. The history field required within this model is calculated each time the tangent matrix is assembled:

```
275
276
277
278
279
280
281
282
283
284
285
286
287
```

98

106

118 119

120

126

128

130

```
if (H>=Energy_el)
    f_el = f_el + d_dam_fun*w(ip)*N(ip,:)'*H;

    K_el = K_el + d_dam_dphi*w(ip)*H*N(ip,:)'*N(ip,:);

    HNew(n_el, ip) = obj.HistOld(n_el, ip);
else
    f_el = f_el + d_dam_fun*w(ip)*N(ip,:)'*Energy_el;
```

K_el = K_el + w(ip)*d_dam_dphi*Energy_el*N(ip,:)'*N(ip,:);

and changes are committed to the new time increments once the phsics object calls its "Commit" function. For the initialization of the phase-field parameter, an estimated history field is set via:

HNew(n_el, ip) = Energy_el;

Models/@PhaseFieldDamage/PhaseFieldDamage.m

$$\phi^{\text{init}} = \exp\left(-\left| dx \right| / \ell\right) \qquad \mathcal{H}^{\text{init}} = \frac{1 / \ell \ \mathbf{N}_{\phi} \mathbf{\Phi}^{\text{init}} + \ell \left(\mathbf{\nabla} \mathbf{N}_{\phi} \mathbf{\Phi}^{\text{init}}\right)^{T} \left(\mathbf{\nabla} \mathbf{N}_{\phi} \mathbf{\Phi}^{\text{init}}\right)}{k_{0} - 2(1 - k_{0}) \left(1 - \mathbf{N}_{\phi} \mathbf{\Phi}^{\text{init}}\right)}$$
(35)

the first time the "OncePerStep" function is called:

Models/@PhaseFieldDamage/PhaseFieldDamage.m

```
function OncePerStep(obj, physics, stp)
    if (stp == obj.phi_step && obj.doInit)
        Hdom = max(obj.mesh.Nodes(:,2));
        Lfrac = 5e-3;
        %% set Values
        nodecons = [];
        allNodes = obj.mesh.GetAllNodesForGroup(obj.myGroupIndex);
        PhiDofs = obj.dofSpace.getDofIndices(obj.dofTypeIndices(3), allNodes);
        initvals = 0*allNodes;
        for i=1:length(allNodes)
            xy = [obj.mesh.Nodes(allNodes(i),1), obj.mesh.Nodes(allNodes(i),2)];
            if (xy(1) < Lfrac)</pre>
                dst = Hdom/2-xy(2);
            else
                dst=sqrt((xy(2)-Hdom/2)^2+(xy(1)-Lfrac)^2);
            pf = exp(-abs(dst)/(obj.1));
            initvals(i) = pf;
        end
        physics.StateVec{obj.phi_step}(PhiDofs) = initvals;
        physics.StateVec_Old{obj.phi_step}(PhiDofs) = initvals;
        %% set history field
        for n_el=1:size(obj.mesh.Elementgroups{obj.myGroupIndex}.Elems, 1)
            Elem_Nodes = obj.mesh.getNodes(obj.myGroupIndex, n_el);
            [N, G, w] = obj.mesh.getVals(obj.myGroupIndex, n_el);
            G2 = obj.mesh.getG2(obj.myGroupIndex, n_el);
            dofsPhi = obj.dofSpace.getDofIndices(obj.dofTypeIndices(3), Elem_Nodes)
            PHI = physics.StateVec{obj.phi_step}(dofsPhi);
            for ip=1:length(w)
```

```
NPhi = N(ip,:)*PHI;
GPhi = squeeze(G(ip,:,:))'*PHI;

d_dam_fun = -2*(1-obj.kmin)*(1-NPhi);
H = abs((NPhi/(obj.1)+obj.1*(GPhi'*GPhi))/(d_dam_fun+obj.kmin));

obj.Hist(n_el, ip) = H;
obj.HistOld(n_el, ip) = H;

end

end

obj.doInit = false;
```

As can be seen, this initial field is set based on the fracture length "Lfrac", and inserted in the centre of the domain.

Input parameters that need to be set for this model are:

```
%Momentum balance and phase-field evolution
physics_in{1}.type = "PhaseFieldDamage";
physics_in{1}.Egroup = "Internal";
physics_in{1}.young = 200e9;
                                %Youngs Modulus [Pa]
physics_in{1}.poisson = 0.3;
                                %Poisson ratio [-]
physics_in{1}.kmin = 1e-10;
                                %residual stiffness factor [-]
physics_in{1}.1=1;
                                %Phase-field length scale [m]
physics_in{1}.Gc=2.0e3;
                                %Fracture release energy [J/m^2]
physics_in{1}.GDegrade = 0.9;
                                 %Maximum hydrogen degradation factor
physics_in{1}.NL = 1e6;
                                %Concentration of interstitial lattice sites [mol/m
    ^31
```

2.2.8. HydrogenDiffusion

Similar to the model presented in Section 2.2.4 for hydrogen diffusion in undamaged materials, this model simulates the diffusion in damaged materials. While the governing equations displayed in Eq. (4) are still being simulated, the hydrostatic stress gradient is now interacts with the phase-field damage as:

$$\nabla \sigma_h = \frac{E}{3(1 - 2\nu)} \left(d(\phi) \mathbf{B}^* \mathbf{u} + \frac{\partial d(\phi)}{\partial \phi} \nabla \phi \overline{\mathbf{B}} \mathbf{u} \right)$$
(36)

%Grain boundary binding energy [J/mol]

where

69

71

76

78

81

82

83 84

85

87

$$\overline{B} = \begin{bmatrix} \frac{\partial N_{u1}}{\partial x} & \frac{\partial N_{u1}}{\partial x} & \cdots & \frac{\partial N_{u1}}{\partial y} & \frac{\partial N_{u1}}{\partial y} & \cdots \end{bmatrix}$$
(37)

Input parameters required for this model are:

physics_in{1}.gb = 30e3;

main.m

2.2.9. PhasefieldElectrolyte

This model simulates the Nernst-Planck and electroneutrelity equations, surface adsorbed hydrogen concentration, and bulk and volume reactions for a crack-contained electrolyte. This system is described by:

$$0 = \beta_{c} \dot{C}_{\pi} - \nabla \cdot \left(\mathbf{R}^{T} \boldsymbol{\beta}_{d} \mathbf{R} D_{\pi} \nabla C_{\pi} \right) - \frac{z_{\pi} F}{RT} \nabla \cdot \left(\mathbf{R}^{T} \boldsymbol{\beta}_{d} \mathbf{R} D_{\pi} C_{\pi} \nabla \varphi \right) + \beta_{c} R_{\pi} + \beta_{s} \nu_{\pi}$$
(38)

$$0 = \beta_{\rm c} \sum z_{\pi} C_{\pi} \tag{39}$$

$$0 = \beta_{\rm s} \left(N_{\rm ads} \dot{\theta}_{\rm ads} - (\nu_{\rm Va} - \nu'_{\rm Va}) + \nu_{\rm Ha} + 2\nu_{\rm T} + (\nu_{\rm A} - \nu'_{\rm A}) - (\nu_{\rm Vb} - \nu'_{\rm Vb}) + \nu_{\rm Hb} \right)$$
(40)

The capacity, diffusion, and surface distributors are accordingly defined as:

Distributed diffusion Physics—based

$$\beta_{c} = \phi^{m} \qquad \beta_{c} = h \left(\frac{1}{2\ell} \phi^{2} + \frac{\ell}{2} |\nabla \phi|^{2} \right)$$

$$(41)$$

$$\boldsymbol{\beta}_{d} = \begin{bmatrix} \phi^{m} D_{\pi,2} / D_{\pi} & 0 \\ 0 & 0 \end{bmatrix} \qquad \boldsymbol{\beta}_{d} = \left(\frac{1}{2\ell} \phi^{2} + \frac{\ell}{2} \left| \boldsymbol{\nabla} \phi \right|^{2} \right) \begin{bmatrix} h & 0 \\ 0 & D_{\infty} \end{bmatrix}$$
(42)

$$\beta_{s} = \frac{1}{\ell} \phi^{2} + \ell \left| \boldsymbol{\nabla} \phi \right|^{2} \qquad \beta_{s} = \frac{1}{\ell} \phi^{2} + \ell \left| \boldsymbol{\nabla} \phi \right|^{2}$$

$$(43)$$

where the distribution method is set within the code to either "WuLorenzis" for the distributed diffusion model described in [15] or "Subgrid" for the physics-based model from this work [1]. The manner in which the crack opening height is obtained is described in detail in Section 3.

Input parameters for this model are:

main.m

```
%Crack-contained electrolyte diffusion, electro-migration, and reactions
108
               physics_in{6}.type = "PhaseFieldElectrolyte";
               physics_in{6}.Egroup = "Internal";
               physics_in{6}.D = [9.3; 5.3; 1.3; 2; 1.4; 1]*1e-9;
                                                                     %Diffusion coefficients for [H
                    OH Na CL Fe FeOH] species respectively [m/s]
               physics_in\{6\}.z = [1; -1; 1; -1; 2; 1];
                                                                     %ionic charges for [H OH Na CL
                   Fe FeOH] [-]
               physics_in\{6\}.pH0 = 5;
                                                                      %Initial and boudnary pH [-]
               physics_in{6}.NaCl = 0.6e3;
                                                                     %Initial and boundary Cl-
                    concentration [mol/m^3]
               physics_in{6}.Lumped = [true; true];
                                                                     %Flag indicationg whtehr to ude
                     lumped integration for the water autoionisation and metal ion reactions
               physics_in{6}.k = [1e6; 1e-1; 1e-3; 1e-3];
                                                                     %Dummy constant for the water
                   auto-ionisation reaction, and reaction rates for Fe, Fe', FeOH reactions
                physics_in{6}.NAds = 1e-3;
                                                                     %Concentration of surface
                    adsorption sites
               physics_in\{6\}.ksurf = [ 1e-4,
                                                1e-10, 0.5,
                                                                 0:
                                                                     %Reaction constants for surface
                     reactions, [k k' alpha E_eq]
                                        1e-10,
                                                                 0;
                                        1e-6.
                                                         Ο,
                                                                 0;
                                                 0.
                                        1e1,
                                                 7e5,
                                                         Ο,
                                        1e-8,
                                                 1e-13,
                                        1e-10.
                                                1e-14,
                                                        0.3,
                                                                 0;
                                        3e-5/(2*96485.3329), 3e-5/(2*96485.3329), 0.5, -0.4];
               physics_in{6}.NL = physics_in{1}.NL;
                                                                     %Concentration of interstitial
                   lattice sites [mol/m^3]
126
               physics_in\{6\}.Em = 0;
                                                                     %Metal electric potential
               physics_in{6}.Lumpedsurf = [1 1 1 1 1 1 1];
                                                                     %Flags to indicate the use of
                    lumped integration for surface reactions
128
               physics_in\{6\}.h0 = 1e-12;
                                                                      %small offset used to prevent
                    ill-conditioned systems
```

```
129 physics_in{6}.Flowtype = model; % Model to use for electrolyte diffusion, either Subgrid WuLorenzis
130 physics_in{6}.l = physics_in{1}.l; %Phase-field length scale
```

where the surface reaction rates k_{surf} follows the format from Eq. (30). This model also implements post-processing functions specifically for the electro-chemical system contained within the crevasse:

```
Models/@PhaseFieldElectrolyte/PhaseFieldElectrolyte.m\\
```

```
function plotHeightData(obj, x_eval, h_est)
```

which plots the crack opening height and the crack-normal vectors,

```
function plotFields(obj, physics)
```

plotting electrolyte concentrations and the electrolyte potential, and

```
function plotFieldspH(obj, physics)
```

plotting the pH of the electrolyte. This last function solely show results where $\phi > 0.1$, filtering out the non-physical state of the electrolyte.

2.3. Mesh

833

894

19

20

This class contains the nodes and elements that describe the geometry, and provides support for evaluating shape functions. Within its implementation, it uses a multi-mesh approach, defining element groups for each entity within the domain (for instance, defining an element group "Interior" for the metal domain composed of surface elements, and defining an element group "Interface" composed of line elements which coincide with the electrolyte-metal interface). The geometry of the problem is defined through procedures within the mesh class

The mesh class also provides a direct interfaces from which to get the element shape functions, second gradients, and surface-normal vectors, providing an element group number and the index of the element itself:

```
@Mesh/mesh.m
[N, G, w] = getVals(obj, group, elem);
G2 = getG2(obj, group, elem);
[n, t] = getNormals(obj, group, elem);
```

which returns a matrix containing the shape functions N within all integration points of the element, gradients of the shape function G, and the integration weights for all integration points w. Additionally, for the construction of the hydrogen diffusion model, the second-order gradients G2 are provided through a separate function.

2.4. Shapes

The classes within this folder provide basic shape functions, and are used by the mesh to provide shape functions and integration weights. The included shape functions are square Lagrangian and triangular Bernstein surface elements (Q9 and T6), quadratic Lagrangian and Bernstein line elements (L3 and L3B), and interface elements (L16, unused).

2.5. Physics

This class provides all the support required for constructing and managing state and force vectors, tangential matrices, and boundary constraints. Most notably, during its initialization it generates an array of all the physical models, from which it then is able to construct the tangential matrix when required:

@Physics/Physics.m

This calls each of the models, and passes a handle to the physics object itself through which the individual models can add their contributions. Notably, the force vector and tangent matrix being assembled by this function are those associated with the staggered solver step "stp", with this step index being passed onto the individual models, which then either add their contributions to the tangent matrix or do not undertake nay action. In a similar manner, actions which should be performed once at the start of each staggered solution step are performed via:

```
@Physics/Physics.m
function OncePerStep(obj, stp)
    %procedures that should be performed once per step
    for m=1:length(obj.models)
        obj.models{m}.OncePerStep(obj, stp);
    end
end
```

The physics class also provides the ability for post-processing the results through the function;

```
@Physics/Physics.m
PlotNodal(obj, dofName, dispscale, plotloc) %exterior defined, plots nodal
   quantities
PlotIP(obj, varName, plotloc) %exterior defined, plots integration
   point quantities
```

This function requires the name of a degree of freedom (for instance "dx" for the horizontal displacements, or "H" for the hydrogen ion concentration), a scale to indicate whether the mesh is plotted in deformed (scale>0) or undeformed (scale=0) configuration, and the name of an element group on which to plot the results. Similarly, the "PlotIP" function plots integration-point specific variables, for instance stresses or history fields.

2.6. Dofspace

67

78

81

82

83

60

17

This class converts the node numbering, degree of freedom type, and solution step to an index for the degree of freedom, corresponding to its location within the unconstrained state vector and tangential matrix. Specific types of degree of freedom are registered on initialization of the dofspace through:

```
@DofSpace/DofSpace.m
```

Algorithm 1 Overview of solution method

```
1: Start of time increment
 2: while not converged do
 3:
        Step 1: Update \phi
        Step 2: Update \mathbf{u}
 4:
        Perform OncePerStep 3: Update h
 5:
        while C_L, C_{\pi}, \phi, \theta are not converged do
 6:
            Step 3: update C_L, C_{\pi}, \varphi, \theta
 7:
            Calculate energy based residual for C_L, C_{\pi}, \varphi, \theta
 8:
        end while
 9:
        Calculate energy based residual for \mathbf{u} and \boldsymbol{\phi}
10:
11: end while
12: Go to next time increment
```

```
%DOFSPACE Construct an instance of this class

obj.NSteps = max(dofs_in.Step);

obj.mesh = mesh;
obj.DofTypes = dofs_in.dofs;
obj.DofSteps = dofs_in.Step;
obj.NDofs = zeros(obj.NSteps,1);
obj.DofNumbering = sparse(length(obj.mesh.Nodes), length(obj.DofTypes));
```

For the numbering scheme saved within "DofNumbering", each solution step retains its own numbering. After initializing the dofspace, degrees of freedom can be added to nodes through:

```
function addDofs(obj, dofIndices, nodeIndex)
  % Adds degrees of freedom for type "dofIndices" to the nodes
  % "nodeIndex"
```

These functions automatically check for duplicates, such that each model can safely add all the degrees of freedom relevant to itself, without taking into account potential interactions with other models. During the finite element assembly, the managed degrees of freedom indices are requestable by providing the degree of freedom type index and the node number:

```
function DofIndices = getDofIndices(obj, dofType, NodeIndices)
% gets the indices for a combination of degree of freedom
% "doftype" and nodes "NodeIndices"
```

which assumes the model already knows which staggered solver step the degree of freedom is associated with. These solver step indices can also be requested through:

```
function [DofTypeIndex, DofStepIndex] = getDofType(obj, dofnames)
  % returns the dof type index for pre-existing degrees of
  % freedom
```

which takes the name of the degree of freedom, and returns its index and the staggered step (and thus state vector) associated with this degree of freedom.

2.7. Solver

18

20

The solver class implements a iteratively staggered Newton-Raphson type nonlinear solver, including the ability to perform linear line-searches to improve the convergence rate and stability. During its creation, it gets linked to the physics object, such that it can automatically request updated tangential matrices. An example of the staggered solution scheme solved through this solver is shown in Algorith 1. Input parameters controlling the solver behaviour are:

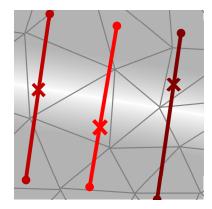


Figure 2: Schematic overview of integrations performed to calculate the crack opening height

main.m

3. Specifics: Fracture opening height

Within the model for the crack-contained electrolyte, Section 2.2.9, the estimated opening height of the crevasse is needed. These opening heights are calculated through an integral normal to the fracture surface [16, 17]:

$$h = \int \mathbf{u} \cdot \nabla \phi \, \mathrm{d}n \tag{44}$$

with the line along which this integral is performed shown in Fig. 2, and the normal vector based on the phase-field as:

$$\boldsymbol{n} = \frac{\boldsymbol{\nabla}\phi}{|\boldsymbol{\nabla}\phi|}\tag{45}$$

These integrals are implemented in:

196

198

201

203

Models/@PhaseFieldElectrolyte/PhaseFieldElectrolyte.m

which fills the "Height" and "normals" matrices with the opening heights and normal vectors for all elements and all integration points. This function is only called once per staggered iteration step, as the phase-field and displacements are assumed to stay constant during the electro-chemical solution step.

First, the displacements and gradients within all integration points are determined:

Models/@PhaseFieldElectrolyte/PhaseFieldElectrolyte.m

```
%% get integration-point data
                    xy = zeros(length(obj.mesh.Elementgroups{obj.myGroupIndex}.Elems), obj.mesh.
206
                        ipcount1D^2, 2);
                    u = zeros(length(obj.mesh.Elementgroups{obj.myGroupIndex}.Elems), obj.mesh.
                        ipcount1D^2, 2);
                    dphi = zeros(length(obj.mesh.Elementgroups{obj.myGroupIndex}.Elems), obj.mesh.
                        ipcount1D^2, 2);
                    Svec = physics.StateVec;
                    xmax = 0; ymax = 0;
                    for el=1:length(obj.mesh.Elementgroups{obj.myGroupIndex}.Elems)
213
                        [N, G, ~] = obj.mesh.getVals(obj.myGroupIndex, el);
214
                        coords = obj.mesh.getIPCoords(obj.myGroupIndex, el);
216
                        Elem_Nodes = obj.mesh.getNodes(obj.myGroupIndex, el);
217
                        dofsX = obj.dofSpace.getDofIndices(obj.dofTypeIndices(1), Elem_Nodes);
218
                        dofsY = obj.dofSpace.getDofIndices(obj.dofTypeIndices(2), Elem_Nodes);
219
                        dofsPhi = obj.dofSpace.getDofIndices(obj.dofTypeIndices(3), Elem_Nodes);
                        X = Svec{obj.dx_step}(dofsX);
                        Y = Svec{obj.dx_step}(dofsY);
                        PHI= Svec{obj.phi_step}(dofsPhi);
224
                        for ip=1:obj.mesh.ipcount1D^2
                            xy(el,ip,1:2) = coords(:,ip);
                            u(el, ip, 1:2) = [N(ip, :)*X N(ip, :)*Y];
228
                            dphi(el,ip,:) = squeeze(G(ip,:,:))'*PHI;
                        end
                        xmax = max(max(coords(1,:)),xmax);
232
                        ymax = max(max(coords(2,:)),ymax);
233
```

and these opening heights are used to construct interpolation functions:

238 239 240

243

246

247

254

Models/@PhaseFieldElectrolyte/PhaseFieldElectrolyte.m

```
%% construct interpolation functions
x = xy(:,:,1); y = xy(:,:,2);
ux = u(:,:,1); uy = u(:,:,2);
dphix = squeeze(dphi(:,:,1)); dphiy = squeeze(dphi(:,:,2));

F_ux = scatteredInterpolant(x(:),y(:),ux(:));
warning('off','last')
F_uy = scatteredInterpolant(x(:),y(:),uy(:));
F_fx = scatteredInterpolant(x(:),y(:),dphix(:));
F_fy = scatteredInterpolant(x(:),y(:),dphiy(:));
```

These interpolation functions are then used to construct an interpolation function for the crack opening height based on Eq. (44), for instance when the crack path is solely horizontal:

Models/@PhaseFieldElectrolyte/PhaseFieldElectrolyte.m

and finally the crack opening heights and normal vectors within integration points are produced as:

```
269
270
271
272
273
274
275
276
277
278
279
280
281
282
283
284
```

287

Models/@PhaseFieldElectrolyte/PhaseFieldElectrolyte.m

```
for n_el=1:length(obj.ActiveElems)
  el = obj.ActiveElems(n_el);
  coords = obj.mesh.getIPCoords(obj.myGroupIndex, el);

for ip=1:obj.mesh.ipcount1D^2
      coords_ip = coords(:,ip);
    if obj.propDir == "Hor"
        obj.Heights(n_el,ip) = Fh(coords_ip(1));
  else
        obj.Heights(n_el,ip) = Fh(coords_ip(2));
  end
  obj.Normals(n_el,ip,1) = F_fx(coords_ip(1),coords_ip(2));
  obj.Normals(n_el,ip,2) = F_fy(coords_ip(1),coords_ip(2));
  obj.Normals(n_el,ip,:) = obj.Normals(n_el,ip,:)/(sqrt(obj.Normals(n_el,ip,1)^2+obj.Normals(n_el,ip,2)^2));

if obj.Heights(n_el,ip)<obj.h0
        obj.Heights(n_el,ip) = obj.h0;
  end</pre>
```

It should be noted that this does not produce opening heights which exactly match to the expected opening based on the displacement and phase-field. However, as long as enough intermediate points are used (e.g. the 1000 points declared on line 248), these results will be virtually indistinguishable from the opening heights that could be calculated by evaluating the phase-field gradient and displacements exactly. The implementation presented here furthermore does not take into account branching cracks, instead assuming the phase-field tends towards zero away from the crack interface.

4. Sample results

4.1. Static cracks

For verification cases using a static crack, a single simulation can be run via "main_Static(model, l, u)", where model is either "Subgrid" or "WuLorenzis" for the physics-based or distributed diffusion models, l is the phase-field length scale, and u is the displacement (for instance, running "main_Static("Subgrid", 5e-4, 1e-5)"). Once running, a folder is created in "./Results_NoProp/" where output data is saved, and results are plotted after each converged time increment. Results after the first time increment are shown in Figs. 3 and 4.

In addition to running a single case, the file "DoSweep.m" performs a parametric sweep over the two models, and several external displacements and phase field length scales. It also runs simulations for a discrete representation of the crack, and creates a figure showing the comparison between the models. please note, as this function runs quite a few simulations, it takes a while to run fully.

4.2. Propagating cracks

The function "main.m" performs a simulation of a propagating crack in a square plate, using the parameters set in this main file. Please note that this function takes a while to run. Results after the first time increment are shown in Figs. 5 and 6.

References

- [1] T. Hageman, E. Martínez-Pañeda, Stabilising Effects of Lumped Integration Schemes for the Simulation of Metal-Electrolyte Reactions, Journal of The Electrochemical Society 170 (2) (2023) 021511.
- [2] T. Hageman, E. Martínez-Pañeda, Stabilising Effects of Lumped Integration Schemes for the Simulation of Metal-Electrolyte Reactions, Journal of The Electrochemical Society 170 (2) (2023) 021511.
- [3] T. Hageman, C. Andrade, E. Martínez-Pañeda, Corrosion rates under charge-conservation conditions, Electrochimica Acta 461 (2023) 142624.
- [4] R. A. Oriani, P. H. Josephic, Equilibrium aspects of hydrogen-induced cracking of steels 22 (9) (1974) 1065–1074.

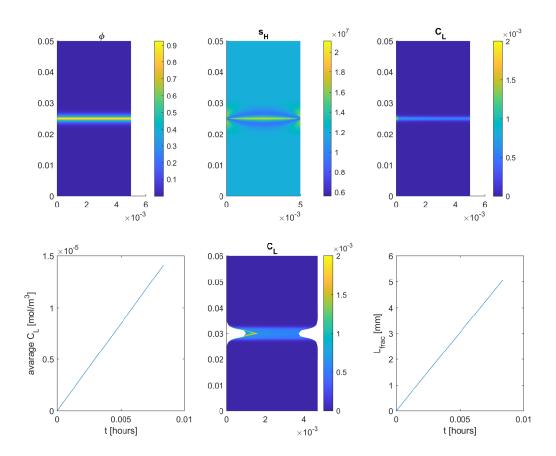


Figure 3: Results produced by running "main_Static("Subgrid", 5e-4, 1e-5)". The first row shows the phase field, hydrostatic stress, and interstitial lattice hydrogen concentration. The second row shows the avarage hydrogen uptake over time, interstitial lattice concentration in the deformed configuration, and fracture length over time.

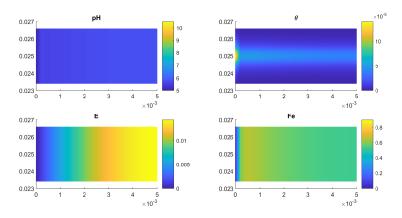


Figure 4: Results produced by running "main_Static("Subgrid", 5e-4, 1e-5)", showing the electrolyte pH, surface adsorbed hydrogen concentration, electrolyte potential and Fe^{2+} concentration.

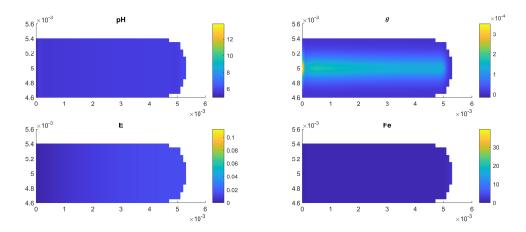


Figure 5: Results produced by running "main(1)". The first row shows the phase field, hydrostatic stress, and interstitial lattice hydrogen concentration. The second row shows the avarage hydrogen uptake over time, interstitial lattice concentration in the deformed configuration, and fracture length over time.

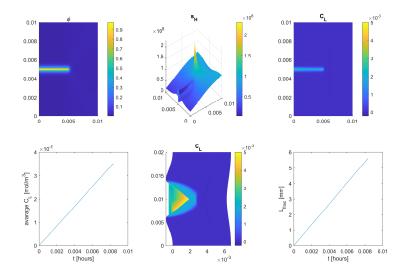


Figure 6: Results produced by running "main(1)", showing the electrolyte pH, surface adsorbed hydrogen concentration, electrolyte potential and Fe^{2+} concentration.

- [5] E. Martínez-Pañeda, A. Díaz, L. Wright, A. Turnbull, Generalised boundary conditions for hydrogen transport at crack tips, Corrosion Science 173 (2020) 108698.
- [6] P. K. Kristensen, C. F. Niordson, E. Martínez-Pañeda, A phase field model for elastic-gradient-plastic solids undergoing hydrogen embrittlement, Journal of the Mechanics and Physics of Solids 143 (2020) 104093.
- [7] A. Golahmar, P. K. Kristensen, C. F. Niordson, E. Martínez-Pañeda, A phase field model for hydrogen-assisted fatigue, International Journal of Fatigue 154 (July 2021) (2022) 106521.
- [8] T. Hageman, E. Martínez-Pañeda, An electro-chemo-mechanical framework for predicting hydrogen uptake in metals due to aqueous electrolytes, Corrosion Science 208 (2022) 110681.
- [9] S. Sarkar, W. Aquino, Electroneutrality and ionic interactions in the modeling of mass transport in dilute electrochemical systems, Electrochimica Acta 56 (24) (2011) 8969–8978.
- [10] S. W. Feldberg, On the dilemma of the use of the electroneutrality constraint in electrochemical calculations, Electrochemistry Communications 2 (7) (2000) 453–456.
- [11] Q. Liu, A. D. Atrens, Z. Shi, K. Verbeken, A. Atrens, Determination of the hydrogen fugacity during electrolytic charging of steel, Corrosion Science 87 (2014) 239–258.
- [12] C. Miehe, M. Hofacker, F. Welschinger, A phase field model for rate-independent crack propagation: Robust algorithmic

- implementation based on operator splits, Computer Methods in Applied Mechanics and Engineering 199 (45-48) (2010) 2765-2778
- [13] C. Miehe, F. Welschinger, M. Hofacker, Thermodynamically consistent phase-field models of fracture: Variational principles and multi-field FE implementations, International Journal for Numerical Methods in Engineering 83 (10) (2010) 1273–1311.
- [14] M. J. Borden, C. V. Verhoosel, M. A. Scott, T. J. Hughes, C. M. Landis, A phase-field description of dynamic brittle fracture, Computer Methods in Applied Mechanics and Engineering 217-220 (2012) 77-95.
- [15] T. Wu, L. De Lorenzis, A phase-field approach to fracture coupled with diffusion, Computer Methods in Applied Mechanics and Engineering 312 (2016) 196–223.
- [16] K. Yoshioka, D. Naumov, O. Kolditz, On crack opening computation in variational phase-field models for fracture, Computer Methods in Applied Mechanics and Engineering 369 (2020) 113210.
- [17] C. Chukwudozie, B. Bourdin, K. Yoshioka, A variational phase-field model for hydraulic fracturing in porous media, Computer Methods in Applied Mechanics and Engineering 347 (2019) 957–982.