



(11) **EP 2 650 291 A1**

(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:  
**16.10.2013 Bulletin 2013/42**

(51) Int Cl.:  
**C07D 403/06** <sup>(2006.01)</sup> **A61K 31/517** <sup>(2006.01)</sup>  
**A61P 31/00** <sup>(2006.01)</sup>

(21) Application number: **13167750.2**

(22) Date of filing: **06.08.2007**

(84) Designated Contracting States:  
**AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IS IT LI LT LU LV MC MT NL PL PT RO SE SI SK TR**  
Designated Extension States:  
**AL BA HR MK RS**

(30) Priority: **07.08.2006 US 835863 P**  
**07.08.2006 EP 06380225**  
**26.06.2007 US 929408 P**  
**26.06.2007 EP 07380186**

(62) Document number(s) of the earlier application(s) in accordance with Art. 76 EPC:  
**07836538.4 / 2 049 504**

(71) Applicant: **Palau Pharma, S.A.**  
**08184 Palau-solità i Plegamans (ES)**

(72) Inventors:  
• **Paredes, Antonio Canovas**  
**08551 Tona (ES)**  
• **Orpi, Javier Bartroli**  
**08034 Barcelona (ES)**  
• **Grau, Elies Molins**  
**08193 Bellaterra (ES)**  
• **Serra, Anna Roig**  
**08193 Bellaterra (ES)**  
• **Meyer, Kevin**  
**Lebanon, IN Indiana 46052 (US)**  
• **Lorimer, Keith**  
**Craigavon, BT63 5QD (GB)**

(74) Representative: **Patentanwälte**  
**Ruff, Wilhelm, Beier, Dauster & Partner**  
**Kronenstrasse 30**  
**70174 Stuttgart (DE)**

(54) **Crystalline Antifungal Compounds**

(57) Describe is a crystalline form of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quina-zolin-4(3H)-one that is Form III and that is suitable for use in treating and/or preventing a microbial or fungal infection, and/or Chagas

disease, in a mammal. Further described are pharmaceutical compositions comprising an effective amount of the crystalline form and a pharmaceutically acceptable carrier

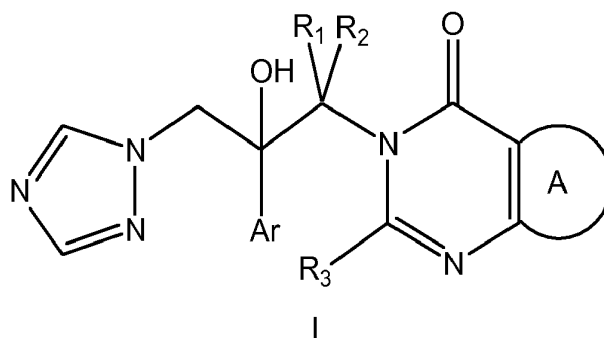
**EP 2 650 291 A1**

**Description****FIELD OF THE INVENTION**

**[0001]** The present subject matter relates to novel crystalline forms of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl]- quinazolin- 4 (3H)- one, pharmaceutical compositions containing these crystalline forms, methods of using these crystalline forms for treating and/or preventing various microbial and/or fungal infections or disorders, and processes for obtaining these crystalline forms. In particular, the present subject matter relates to the specific crystalline Forms I, II, III, IV V, and VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**BACKGROUND OF THE INVENTION**

**[0002]** U.S. Patent No. 5, 807, 854 discloses various novel antifungal compounds of the formula I:

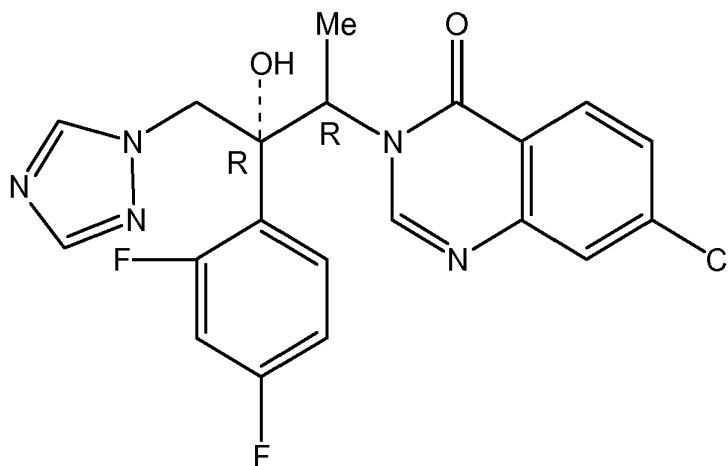


methods of making these compounds, pharmaceutical compositions containing these compounds, and their use in the treatment or prevention of fungal infections in animals. One of the specifically exemplified compounds falling within this genus is albaconazole, which also has the chemical name (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, this patent does not disclose, refer to, or even suggest the benefits to obtaining specific crystalline forms of the compounds of formula I.

**[0003]** To prepare pharmaceutical compositions containing (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4- (3H)- one for administration to mammals in accordance with exacting health registration requirements of the U.S. and international health registration authorities, e.g. the U.S. Food and Drug Administration's Good Manufacturing Practice ("GMP") requirements, there is a need to produce (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4- (3H)- one in as pure a form as possible, especially a form having constant and consistent physical properties.

**SUMMARY OF THE INVENTION**

**[0004]** Accordingly, the present subject matter provides various crystalline forms of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- tria- zol- 1- yl) propyl] quinazolin- 4 (3H)- one represented by formula II:



II

**[0005]** This compound was previously known to exist in an amorphous form. It has now been recognized that the compound can exist as one of six crystalline forms that are presently identified. Accordingly, a crystalline form of this compound is contemplated herein, such as a pure crystalline form substantially devoid of the compound's amorphous form and any residual solvent. In this regard, substantially pure crystalline forms of each of Forms I, II, III, IV, V, and VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one are contemplated herein.

**[0006]** In this regard, a preferred embodiment of the present subject matter relates to a crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising the reaction product of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and an organic solvent. In a further preferred embodiment, the present subject matter relates to a crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising the reaction product of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and an organic solvent in the presence of water.

**[0007]** Another preferred embodiment of the present subject matter relates to a substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one selected from the group consisting of Form I, Form II, Form III, Form IV, Form V, and Form VI..

**[0008]** Still another preferred embodiment of the present subject matter relates to a pharmaceutical composition comprising an anti- microbially or anti- fungally effective amount of a substantially pure crystalline Form I, II, III, IV, V, or VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof and a pharmaceutically acceptable carrier. The crystalline Form III, Form IV, or Form VI is particularly preferred in this regard.

**[0009]** Still yet another preferred embodiment of the present subject matter relates to a method of treating and/or preventing microbial and/or fungal infections in a mammal comprising administering to a mammal in need thereof an effective amount of a substantially pure crystalline Form I, II, III, IV, V, or VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof. The crystalline Form III, Form IV, or Form VI is particularly preferred in this regard.

**[0010]** Still yet another preferred embodiment of the present subject matter relates to a method of treating and/or preventing Chagas Disease in a mammal comprising administering to a mammal in need thereof an effective amount of a substantially pure crystalline Form I, II, III, IV, V, or VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof. The crystalline Form III, Form IV, or Form VI is particularly preferred in this regard.

**[0011]** In still another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form I or II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Crystallizing a crystalline Form I or II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from an amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one using supercritical CO<sub>2</sub> crystallization conditions.

**[0012]** Yet another preferred embodiment of the present subject matter relates to a process for preparing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding an amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to a solvent selected from the group consisting of ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate to form a solution or suspension; and
- Crystallizing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension.

**[0013]** Moreover, another preferred embodiment of the present subject matter relates to a process for preparing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Dissolving (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in a solvent selected from the group consisting of ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate to form a solution;
- Crystallizing crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Drying said crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0014]** In a particularly preferred embodiment in this regard, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one dissolved in the solvent is the amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for these purposes.

**[0015]** Still another preferred embodiment of the present subject matter relates to a process for preparing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to ethanol to form a solution or suspension;
- Crystallizing crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Drying said crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0016]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Storing an amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one for 3 months at 40°C and 75% HR;
- Transforming some of said amorphous (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to a crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one during storage; and
- Obtaining said crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0017]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Dissolving (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in ethanol to form a solution;
- Adding the solution to water to form a suspension;

- Stirring the suspension for more than 30 minutes;
- Obtaining crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Separating said crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0018]** In a particularly preferred embodiment in this regard, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one dissolved in the solvent is the amorphous form or Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for these purposes.

**[0019]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Dissolving (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in ethyl acetate to form a solution;
- Adding hexane to the solution;
- Optionally adding diethyl ether to the solution;
- Crystallizing crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Separating said crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0020]** In a particularly preferred embodiment in this regard, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one dissolved in the solvent is the amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for these purposes.

**[0021]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Forming a solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in an aqueous ethanol solvent;
- Crystallizing Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0022]** In one embodiment, the present subject matter relates to a process of making a substantially pure crystalline monohydrate form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in certain organic solvents and water. This monohydrate form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one can also be known as Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In one embodiment in this regard, the present subject matter relates to a process of making a substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in certain organic solvents and water.

**[0023]** In one particular embodiment of the present subject matter, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one, comprises the reaction product of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and an organic solvent and water. In a preferred embodiment, the organic solvent is a polar solvent. In a further preferred embodiment, the polar solvent is acetone or an alcohol. In a preferred embodiment in this regard, the alcohol is selected from the group consisting of ethanol, methanol, isopropanol, n- propanol, and acetone. In an especially preferred embodiment, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one, comprises the reaction product of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and ethanol and water. In another preferred embodiment, the

substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4- (3H)- one comprises the monohydrate of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0024]** In another embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Forming a solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in an aqueous solvent and an organic solvent selected from the group consisting of ethanol, methanol, isopropanol, n- propanol, and acetone;
- Crystallizing Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0025]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to aqueous ethanol to form a solution or suspension;
- Crystallizing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0026]** In an alternative preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to ethanol to form a mixture;
- Adding said mixture to water to form a solution or suspension;
- Crystallizing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In a particularly preferred embodiment, the mixture of albaconazole and ethanol is a solution of albaconazole in ethanol.

**[0027]** In a particularly preferred embodiment in this regard, the (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one used to form the mixture, or the solution or suspension, is the amorphous form, Form III, Form IV, Form V, or combinations thereof of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for these purposes.

## BRIEF DESCRIPTION OF THE FIGURES

### [0028]

Fig. 1 presents a characteristic X- ray powder diffraction (XRPD) pattern of the crystalline Form I of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 2 presents a characteristic infrared (IR) spectrum of the crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 3 presents a characteristic differential scanning calorimetry (DSC) thermogram of the crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin-

4 (3H)- one.

Fig. 4 presents a characteristic XRPD pattern of the crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 5 presents a characteristic IR spectrum of the crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 6 presents a characteristic DSC thermogram of the crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 7 presents a characteristic XRPD pattern of the crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 8 presents a characteristic IR spectrum of the crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 9 presents a characteristic DSC thermogram of the crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 10 presents a characteristic XRPD pattern of the crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 11 presents a characteristic IR spectrum of the crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 12 presents a characteristic DSC thermogram of the crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 13 presents a characteristic XRPD pattern of the crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 14 presents a characteristic IR spectrum of the crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 15 presents a characteristic DSC thermogram of the crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 16 presents a characteristic XRPD pattern of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Figs. 17a- c present a characteristic IR spectrum of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. The complete IR Spectrum of Fig. 17a is displayed in higher resolution in Figures 17b, 17c, and 17d to show characteristic peaks.

Figs. 18a- c present three characteristic DSC thermograms of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

Fig. 19 presents an ORTEP drawing of the crystalline Form VI of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. Atoms are represented by 50% probability anisotropic ellipsoids.

Fig. 20 presents a packing diagram of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one viewed along the a crystallographic axis. Hydrogen atoms omitted for clarity.

Fig. 21 presents a packing diagram of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one viewed along the b crystallographic

axis. Hydrogen atoms omitted for clarity.

Fig. 22 presents a packing diagram of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one viewed along the c crystallographic axis. Hydrogen atoms are omitted for clarity.

Fig. 23 presents a packing diagram of the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one viewed along the a crystallographic axis with hydrogen bonding represented with dashed lines. Hydrogen bonding between adjacent molecules of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and water molecules generates a three- dimensional network. The water molecules reside in channels running parallel to the crystallographic a axis.

## DETAILED DESCRIPTION OF THE INVENTION

**[0029]** As used herein, the terms "administering", "administration", and like terms refer to any method which, in sound medical or cosmetic practice, delivers the composition to a subject in such a manner as to provide a therapeutic effect.

**[0030]** The phrase "aqueous solvent" as used herein refers to a solvent such as water or containing water. Other dissolved components may be present in small amounts, such as, for example, salts, buffers, and other components understood by one of ordinary skill in the art to be optionally present in an aqueous solution.

**[0031]** The phrase "crystalline form" as used herein refers to crystals of the same molecule having different physical properties as a result of the order of the molecules in the crystal lattice. Accordingly, as used herein the phrase "crystalline form" is intended to refer to different crystalline forms, polymorphs, pseudopolymorphs, and solvate forms of a single molecular entity. Different crystalline forms of a single compound may have different chemical, physical, mechanical, electrical, thermodynamic, and/or biological properties from each other. The differences in physical properties exhibited by crystalline forms affect pharmaceutical parameters such as storage stability, compressibility, density (important in formulation and product manufacturing), dissolution rates (an important factor in determining bio- availability), solubility, melting point, chemical stability, physical stability, powder flowability, compaction, and particle morphology. Each separate crystalline form of an individual compound will exhibit consistent chemical, physical, mechanical, electrical, thermodynamic, and biological properties.

**[0032]** The phrase "derivative" as used herein refers to any hydrate, solvate, salt, racemate, isomer, enantiomer, prodrug, metabolite, ester, or other analog or derivative of a particular chemical compound or molecule. The term "derivative" may also mean a modification to the disclosed compounds including, but not limited to, hydrolysis, reduction, or oxidation products, of the disclosed compounds. Hydrolysis, reduction, and oxidation reactions are known in the art.

**[0033]** Differences in stability can result from changes in chemical reactivity (e.g. differential oxidation, such that a dosage form discolors more rapidly when comprised of one crystalline form than when comprised of another crystalline form) or mechanical changes (e.g. tablets crumble on storage as a kinetically favored crystalline form converts to thermodynamically more stable crystalline form) or both (e.g. tablets of one crystalline form are more susceptible to breakdown at high humidity). As a result of solubility/dissolution differences, some crystalline form transitions affect potency and/or toxicity. In addition, the physical properties of the crystal may be important in processing; for example, one crystalline form might be more likely to form solvates or might be difficult to filter and wash free of impurities (i.e. particle shape and size distribution might be different between one crystalline form relative to the other).

**[0034]** As used herein, the phrases an "effective amount" or a "therapeutically effective amount" of an active agent or ingredient, or pharmaceutically active agent or ingredient, which are synonymous herein, refer to an amount of the pharmaceutically active agent sufficient enough to have a therapeutic effect upon administration. A therapeutically effective amount of the pharmaceutically active agent may, will, or is expected to cause a relief of symptoms. Effective amounts of the pharmaceutically active agent will vary with the particular condition or conditions being treated, the severity of the condition, the duration of the treatment, the specific components of the composition being used, and like factors.

**[0035]** As used herein, the phrase "pharmaceutically acceptable salts" refers to salts of certain ingredient(s) which possess the same activity as the unmodified compound(s) and which are neither biologically nor otherwise undesirable. A salt can be formed with, for example, organic or inorganic acids. Non-limiting examples of suitable acids include acetic acid, acetylsalicylic acid, adipic acid, alginic acid, ascorbic acid, aspartic acid, benzoic acid, benzenesulfonic acid, bisulfic acid, boric acid, butyric acid, camphoric acid, camphorsulfonic acid, carbonic acid, citric acid, cyclopentanepropionic acid, digluconic acid, dodecylsulfic acid, ethanesulfonic acid, formic acid, fumaric acid, glyceric acid, glycerophosphoric acid, glycine, glucoheptanoic acid, gluconic acid, glutamic acid, glutaric acid, glycolic acid, hemisulfic acid, heptanoic acid, hexanoic acid, hippuric acid, hydrobromic acid, hydrochloric acid, hydroiodic acid, hydroxyethanesulfonic acid, lactic acid, maleic acid, malic acid, malonic acid, mandelic acid, methanesulfonic acid, mucic acid, naphthylanesulfonic



acid, naphthyllic acid, nicotinic acid, nitrous acid, oxalic acid, pelargonic, phosphoric acid, propionic acid, saccharin, salicylic acid, sorbic acid, succinic acid, sulfuric acid, tartaric acid, thiocyanic acid, thioglycolic acid, thiosulfuric acid, tosylic acid, undecylenic acid, and naturally and synthetically derived amino acids.

**[0036]** If organic bases are used, poorly volatile bases are preferably employed, for example low molecular weight alkanolamines such as ethanolamine, diethanolamine, N-ethylethanolamine, N-methyldiethanolamine, triethanolamine, diethylaminoethanol, 2-amino-2-methyl-n-propanol, dimethylaminopropanol, 2-amino-2-methylpropanediol, and triisopropanolamine. Ethanolamine is particularly preferred in this regard. Further poorly volatile bases which may be mentioned are, for example, ethylenediamine, hexamethylenediamine, morpholine, piperidine, piperazine, cyclohexylamine, tributylamine, dodecylamine, N,N-dimethyldodecylamine, stearylamine, oleylamine, benzylamine, dibenzylamine, N-ethylbenzylamine, dimethylstearylamine, N-methylmorpholine, N-methylpiperazine, 4-methylcyclohexylamine, and N-hydroxyethylmorpholine.

**[0037]** Salts of quaternary ammonium hydroxides such as trimethylbenzylammonium hydroxide, tetramethylammonium hydroxide, or tetraethylammonium hydroxide can also be used, as can guanidine and its derivatives, in particular its alkylation products. However, it is also possible to employ as salt-forming agents, for example, low molecular weight alkylamines such as methylamine, ethylamine, or triethylamine. Suitable salts for the components to be employed according to the present subject matter are also those with inorganic cations, for example alkali metal salts, in particular sodium, potassium, or ammonium salts, alkaline earth metal salts such as, in particular, the magnesium or calcium salts, as well as salts with bi- or tetravalent cations, for example the zinc, aluminum, or zirconium salts. Also contemplated are salts with organic bases, such as dicyclohexylamine salts; methyl-D-glucamine; and salts with amino acids, such as arginine, lysine, and so forth. Also, the basic nitrogen-containing groups can be quaternized with such agents as lower alkyl halides, such as methyl, ethyl, propyl, and butyl chlorides, bromides, and iodides; dialkyl sulfates, such as dimethyl, diethyl, dibutyl, and diamyl sulfates; long chain halides, such as decyl, lauryl, myristyl, and stearyl chlorides, bromides, and iodides; asthma halides, such as benzyl and phenethyl bromides; and others. Water or oil-soluble or dispersible products are thereby obtained.

**[0038]** The phrase "reaction product" as used herein refers to any crystalline form obtained from the processes described herein, including but not limited to anhydrides, hydrates, polymorphs, solvates, N-oxides, and/or salts of these crystalline forms.

**[0039]** The phrase "substantially pure" as used herein refers to an individual crystalline form which is substantially devoid of all other crystalline forms, as well as degradation products of the crystalline form, the amorphous form, and any residual solvent, and is at least 85% pure on a % weight basis, unless otherwise specified. Preferably, the crystalline form has at least 90% purity on a % weight basis. More preferably, the crystalline form has at least 93% purity on a % weight basis. Yet more preferably, the crystalline form has at least 95% purity on a % weight basis. Yet even more preferably, the crystalline form has at least 97% purity on a % weight basis.

**[0040]** The term "treating" as used herein refers to the process of producing an effect on biological activity, function, health, or condition of an organism in which such activity is maintained, enhanced, diminished, or applied in a manner consistent with the general health and well-being of the organism.

**[0041]** Other terms as used herein are meant to be defined by their well-known meanings in the art.

#### **Crystalline Forms of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one**

**[0042]** Six different crystalline forms of the compound (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one have presently been identified: crystalline Forms I, II, III, IV, V, and VI. Of these different crystalline forms, the most preferred will provide an antifungal, antimicrobial drug having the lowest impurity content, most consistent product quality, most consistent physical characteristics including color, rate of dissolution, and ease of handling, and greatest long term stability in comparison to the other crystalline forms or the amorphous form.

**[0043]** Accordingly, the present subject matter relates to methods of identifying, obtaining, and purifying the various crystalline forms of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one. These crystalline forms, namely Forms I-VI, were identified as six distinct crystal forms. Certain physical characteristics of these crystalline forms generated during crystallization studies, are as follows:

- Form I: typically obtained by crystallizing the amorphous form using supercritical CO<sub>2</sub> crystallization conditions. This form displays a characteristic XRPD pattern, a characteristic IR spectrum, and a characteristic DSC profile.
- Form II: typically obtained by crystallizing the amorphous form using supercritical CO<sub>2</sub> crystallization conditions. This form displays a characteristic XRPD pattern, a characteristic IR spectrum, and a characteristic DSC profile.

- Form III: typically obtained under standard crystallization conditions using a variety of solvents such as ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate. This form displays a characteristic XRPD pattern, a characteristic IR spectrum, and a DSC profile with a strong endothermic peak onset at about 99°C. No degradation products of form III were detected after 6 months of storage at 30°C/65% RH and 25°C/60% RH.
- Form IV: typically obtained by first dissolving any form in ethanol, then suspending this solution in water and stirring for a certain amount of time. This form may also be obtained directly from the amorphous form or Forms III or VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one after storage for an extended period of time, for example after storage for 3 months at 40°C and 75% RH, as some but not all of the original form converts to Form IV. This form displays a characteristic XRPD pattern, a characteristic IR spectrum, and a DSC profile with a strong endothermic peak onset at about 121°C.
- Form V: typically obtained under standard crystallization conditions, typically by first dissolving the amorphous form in ethyl acetate, then adding hexane to this solution. Diethyl ether can then be optionally added. This form displays a characteristic XRPD pattern, a characteristic IR spectrum, and a DSC profile with a strong endothermic peak onset at about 108°C. Form V can best be characterized as a crystalline phase containing between about 2 and about 7% by weight of ethyl acetate and about 0.5 to about 2.5% by weight of hexane.
- Form VI: typically obtained by slurrying (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in aqueous ethanol or by slowly cooling a saturated aqueous ethanol solution that was seeded. This form may be obtained when the slurry comprises any other Form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one, such as, for example, the amorphous form, Form III, Form IV, Form V, or combinations thereof. Complete conversion of a Form III slurry at 45°C to Form VI was noted after 3 hours in ethanol- water (1: 9) . High water activities are preferred for preparing Form VI to ensure that Forms III and/or V are not generated. This form displays a characteristic XRPD pattern, a characteristic IR spectrum, and a DSC profile with a strong endothermic peak ranging from about 102°C to 108°C. Form VI can best be characterized as a non- hygroscopic monohydrate.

#### Purity

**[0044]** The present subject matter contemplates substantially pure and/or isolated crystalline Forms I, II, III, IV, V, and/or VI of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In this regard, the present subject matter contemplates each of these crystalline forms which is substantially devoid of the amorphous form and of any residual solvent on a weight % basis, except as otherwise noted herein. In a preferred embodiment, the present subject matter particularly contemplates each of these crystalline forms free of any residual solvent, except as otherwise noted herein. Notwithstanding the above, crystalline forms contemplated herein may be in a hydrated form and thus may contain a certain amount of water. In preferred embodiments in this regard, crystalline forms may be formed as hydrates containing about 10% or less of water. In an alternative preferred embodiment, the present subject matter contemplates each of these crystalline forms further substantially devoid of the other crystalline forms.

**[0045]** In a preferred embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has less than about 10% by weight of a different crystalline form or an amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one as determined on a % weight basis.

**[0046]** In another preferred embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has at least 90% purity as defined by X- ray powder diffraction.

**[0047]** In another embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has less than about 10% by weight of any residual solvent.

**[0048]** In yet another embodiment, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is storage stable for at least 6 months at about 25°C and 60% HR.

**[0049]** For example, the present subject matter contemplates crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one which has less than about 10% by weight of any residual solvent and the amorphous form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one as determined on a weight % basis; crystalline Form II of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1-

20

35

50

55

forms of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one as determined on a weight % basis.

**[0053]** In another preferred embodiment, the present subject matter contemplates each of the crystalline forms I, II, III, IV, V, and VI of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one which have less than about 10% by weight of any residual solvent. In a further preferred embodiment in this regard, the present subject matter contemplates each of the crystalline forms I, II, III, IV, and V of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one which have less than 7%, more preferably less than 5% by weight, of any residual solvent. In another further preferred embodiment in this regard, the present subject matter contemplates the crystalline form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one which has less than 7%, more preferably less than 5% by weight of any residual solvent. In a most preferred embodiment, the crystalline form free of residual solvent is crystalline Form VI. Notwithstanding the above, Form VI contemplated herein is a monohydrated form and thus contains a certain amount of water. In preferred embodiments in this regard, Form VI is a monohydrate containing about 4% water.

**[0054]** In one embodiment, the substantially pure crystalline form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one, more preferably at least 3, most preferably at least 5, 2- theta position selected from the group consisting of those at about 2.72, 3.74, 4.08, 4.11, 4.15, 4.17, 5.31, 5.73, 5.83, 6.22, 6.28, 6.35, 7.5, 7.77, 7.96, 7.98, 8.15, 8.22, 8.33, 8.35, 8.37, 8.80, 9.01, 9.39, 9.61, 10.1, 11.16, 11.25, 11.29, 11.35, 11.4, 11.47, 11.61, 11.66, 11.7, 12.1, 12.41, 12.44, 12.49, 12.57, 12.6, 13.06, 13.09, 13.15, 13.21, 13.29, 13.3, 13.34, 13.62, 13.64, 14.11, 14.33, 14.34, 14.41, 14.42, 14.43, 14.5, 14.52, 14.68, 14.89, 14.93, 14.98, 15.0, 15.09, 15.43, 15.57, 15.7, 15.74, 15.93, 15.95, 16.0, 16.35, 16.6, 16.68, 16.74, 16.77, 16.90, 16.98, 17.0, 17.21, 17.27, 17.3, 17.4, 17.49, 17.56, 17.57, 17.63, 17.71, 17.91, 18.25, 18.66, 18.74, 18.79, 18.8, 18.82, 18.86, 18.9, 19.2, 19.30, 19.32, 19.37, 19.7, 20.36, 20.43, 20.85, 20.88, 21.08, 21.1, 21.47, 21.78, 21.79, 21.88, 22.12, 22.27, 22.3, 22.31, 22.49, 22.62, 22.82, 22.88, 23.20, 23.58, 23.64, 23.82, 23.84, 23.86, 23.9, 24.2, 24.26, 24.63, 24.78, 24.8, 25.02, 25.11, 25.2, 25.3, 25.32, 25.7, 25.95, 26, 26.03, 26.2, 26.65, 26.7, 26.74, 26.77, 26.83, 26.86, 27.04, 27.12, 27.21, 27.25, 27.35, 27.44, 27.6, 28.00, 28.43, 28.5, 28.57, 28.6, 28.74, 28.9, 28.96, 28.98, 29.05, 29.11, 29.16, 29.3, 29.38, 29.41, 29.7, 29.81, 29.97, 30.0, 30.07, 30.13, 30.14, 30.5, 30.73, 30.75, 30.8, 30.91, 30.98, 31.3, 31.35, 31.35, 31.58, 31.78, 32.05, 32.36, 32.41, 33.3, 33.48, 33.61, 33.63, 33.7, 33.9, 34.3, 34.35, 34.62, 34.94, 35.0, 35.5, 36.5, 36.7, 37.4, 39.5, 45.28, 46.1, 48.87, and 55.02 +/- 0.2.

**[0055]** In a preferred embodiment, the substantially pure crystalline form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern containing at least one, more preferably at least 3, most preferably at least 5, 2- theta position selected from the group consisting of those at about 4.08, 4.11, 4.15, 4.17, 5.73, 5.83, 6.22, 6.28, 6.35, 7.5, 7.77, 7.96, 7.98, 8.15, 8.22, 8.33, 8.35, 8.37, 8.8, 9.39, 9.61, 10.1, 11.16, 11.25, 11.29, 11.35, 11.4, 11.47, 11.66, 11.7, 12.41, 12.44, 12.49, 12.57, 12.6, 13.09, 13.15, 13.29, 13.62, 13.64, 14.34, 14.41, 14.42, 14.5, 14.89, 14.98, 15.57, 15.95, 16.0, 16.74, 16.77, 16.9, 17.49, 17.56, 17.57, 17.63, 18.66, 18.74, 18.79, 18.86, 18.9, 20.85, 21.08, 21.2, 21.2, 21.2, 23.82, 23.84, 23.86, 24.78, 24.8, 25.3, 25.11, 25.2, 25.32, 25.7, 26.65, 26.83, 27.04, 27.12, 27.35, 27.44, and 30.13 +/- 0.2. In other preferred embodiments, the substantially pure crystal form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has two or more of the XRPD peaks identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified above.

**[0056]** In one embodiment, the substantially pure crystalline form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1723.8, 1677.0, 1676.0, 1673.3, 1671.0, 1607, 1601.0, 1600.0, 1599.0, 1557.5, 1555, 1503.3, 1501.2, 1499.4, 1498.8, 1498.3, 1468, 1462.7, 1462.6, 1462.3, 1462.2, 1404.1, 1403.9, 1403.0, 1402.4, 1400, 1361, 1319.8, 1319.4, 1318.4, 1318.2, 1318.0, 1316, 1280, 1274.4, 1273.9, 1272.5, 1272.4, 1254.9, 1254.6, 1254.1, 1253.7, 1253.0, 1218, 1210.2, 1170.2, 1170.0, 1169.9, 1165, 1139.1, 1139.0, 1138.7, 1138.0, 1137.7, 1102.7, 1102.1, 1102, 1101.8, 1101.6, 1062.2, 1061.8, 1061.4, 1060.7, 1060.2, 1016.4, 1014, 976, 967.4, 967.2, 967.0, 966.7, 938, 933.5, 932.8, 932.7, 932.6, 902.8, 902.7, 902.4, 902.0, 857.4, 857.2, 855.5, 845.0, 801.5, 801.4, 801.3, 801.3, 801.2, 785.9, 785.8, 785.1, 783.6, 782.9, 760, 698, 694.0, 693.9, 693.8, 693.5, 677.9, 677.7, 677.2, 665.4, 665.1, 664.9, 664.1, 663.6, 631.8, 631.7, 630.7, 630.2, 630.1, 533.4, 532.7, and 411.6 cm<sup>-1</sup>.

**[0057]** In another embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is characterized as having an infrared spectra having at least two of the characteristic infrared spectral peak positions identified above. In yet another embodiment, the substantially pure crystalline form is characterized as having an infrared spectra having at least three of the characteristic infrared spectral peak positions identified above. In a further embodiment, the substantially pure crystalline

form is characterized as having an infrared spectra having at least four of the characteristic infrared spectral peak positions identified above.

**[0058]** In a preferred embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern selected from the group consisting of (A) 2- theta positions at about 4.11, 8.22, 9.39, 11.29, 12.41, 13.62, 16.74, 17.57, 18.79, 23.82, and 25.2 +/- 0.2; (B) 2- theta positions at about 6.35, 7.98, 8.37, 11.4, 11.7, 12.6, 13.15, 14.42, 14.98, 16.77, 17.56, 18.86, 21.08, 23.84, 25.32, 26.83, and 27.35 +/- 0.2; (C) 2- theta positions at about 4.08, 5.73, 6.22, 7.77, 8.15, 8.80, 11.25, 11.47, 12.44, 13.09, 15.57, 17.63, 18.66, 20.85, 26.65, and 27.12 +/- 0.2; (D) 2- theta positions at about 4.15, 7.5, 8.33, 9.61, 11.16, 12.49, 13.29, 13.64, 14.41, 16.90, 18.74, 24.78, and 25.11 +/- 0.2; (E) 2- theta positions at about 4.17, 5.83, 6.28, 7.96, 8.35, 11.35, 11.66, 12.57, 14.34, 14.89, 15.95, 16.74, 17.49, 18.9, 21.1, 23.86, 25.3, 27.04, 27.44, and 30.13 +/- 0.2; and (F) 2- theta positions at about 21.1, 24.8, and 25.7 +/- 0.2.

**[0059]** In one embodiment, the substantially pure crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.11, 8.22, 9.39, 11.29, 12.41, 13.21, 13.62, 14.43, 14.93, 15.7, 16.74, 17.3, 17.57, 18.79, 20.88, 21.88, 22.62, 23.64, 23.82, 25.2, 26.77, 27.21, 28.57, 29.16, 29.97, 30.75, 31.35, 45.28, 48.87, and 55.02 +/- 0.2.

**[0060]** In a preferred embodiment, the substantially pure crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.11, 8.22, 9.39, 11.29, 12.41, 13.62, 16.74, 17.57, 18.79, 23.82, and 25.2 +/- 0.2. In other preferred embodiments, the substantially pure crystal Form I has two or more of the XRPD peaks identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified above.

**[0061]** In one embodiment, the substantially pure crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1673.3, 1600.0, 1557.5, 1501.2, 1462.7, 1403.9, 1319.4, 1273.9, 1254.6, 1139.0, 1101.8, 1061.8, 967.2, 902.7, 801.3, 783.6, 664.1, and 630.1 cm<sup>-1</sup>.

**[0062]** In another embodiment, the substantially pure crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 2.72, 5.31, 6.35, 7.98, 8.37, 9.01, 11.4, 11.7, 12.6, 13.15, 14.42, 14.98, 15.93, 16.77, 17.56, 17.91, 18.86, 19.37, 21.08, 21.78, 22.31, 22.82, 23.84, 25.32, 26, 26.83, 27.35, 28.5, 28.96, 29.38, 30.14, 31.58, 32.41, 33.63, 34.94, and 46.1 +/- 0.2.

**[0063]** In a preferred embodiment, the substantially pure crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 6.35, 7.98, 8.37, 11.4, 11.7, 12.6, 13.15, 14.42, 14.98, 16.77, 17.56, 18.86, 21.08, 23.84, 25.32, 26.83, and 27.35 +/- 0.2. In other preferred embodiments, the substantially pure crystal Form II has two or more of the XRPD peaks identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified above.

**[0064]** In one embodiment, the substantially pure crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1677.0, 1600.0, 1557.5, 1498.8, 1462.3, 1318.2, 1272.4, 1253.0, 1170.2, 1137.7, 1102.0, 1060.7, 967.0, 932.6, 902.0, 857.2, 801.3, 785.1, 693.5, 664.9, and 630.7 cm<sup>-1</sup>.

**[0065]** In another embodiment, the substantially pure crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.08, 5.73, 6.22, 7.77, 8.15, 8.80, 11.25, 11.47, 12.44, 13.09, 14.33, 14.68, 14.89, 15.57, 16.35, 16.68, 17.27, 17.63, 18.66, 19.32, 20.85, 22.12, 22.49, 23.58, 24.63, 25.02, 26.65, 27.12, 28.74, 29.11, 29.81, 31.35, and 33.48 +/- 0.2.

**[0066]** In a preferred embodiment, the substantially pure crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.08, 5.73, 6.22, 7.77, 8.15, 8.80, 11.25, 11.47, 12.44, 13.09, 15.57, 17.63, 18.66, 20.85, 26.65, and 27.12 +/- 0.2. In other preferred embodiments, the substantially pure crystal Form III has two or more of the XRPD peaks

identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified above.

**[0067]** In one embodiment, the substantially pure crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1677.0, 1600.0, 1557.5, 1498.3, 1462.6, 1403.0, 1318.4, 1272.5, 1254.1, 1170.0, 1138.7, 1101.6, 1060.2, 1016.4, 966.7, 932.7, 902.4, 855.5, 801.5, 785.8, 694.0, 677.9, 665.4, 631.7, 532.7, and 411.6 cm<sup>-1</sup>.

**[0068]** In another embodiment, the substantially pure crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 3.74, 4.15, 7.5, 8.33, 9.61, 11.16, 11.61, 12.49, 13.29, 13.64, 14.41, 15.43, 15.74, 16.90, 17.71, 18.25, 18.74, 19.30, 20.43, 21.78, 23.20, 24.26, 24.78, 25.11, 26.03, 26.86, 27.25, 28.00, 29.05, 30.07, 30.91, and 32.05 +/- 0.2.

**[0069]** In a preferred embodiment, the substantially pure crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.15, 7.5, 8.33, 9.61, 11.16, 12.49, 13.29, 13.64, 14.41, 16.90, 18.74, 24.78, and 25.11 +/- 0.2. In other preferred embodiments, the substantially pure crystal Form IV has two or more of the XRPD peaks identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified above.

**[0070]** In one embodiment, the substantially pure crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1671.0, 1601.0, 1557.5, 1503.3, 1462.7, 1404.1, 1319.8, 1274.4, 1254.9, 1210.2, 1139.1, 1102.1, 1062.2, 967.4, 933.5, 902.8, 845.0, 801.4, 782.9, 693.8, 677.7, 663.6, and 630.2 cm<sup>-1</sup>.

**[0071]** In another embodiment, the substantially pure crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.17, 5.83, 6.28, 7.96, 8.35, 11.35, 11.66, 12.57, 13.06, 13.34, 14.11, 14.34, 14.52, 14.89, 15.09, 15.95, 16.74, 16.98, 17.21, 17.49, 17.91, 18.82, 18.9, 20.36, 21.1, 21.47, 21.79, 22.27, 22.88, 23.86, 25.3, 25.95, 26.2, 26.74, 27.04, 27.44, 28.43, 28.98, 29.41, 30.13, 30.73, 30.98, 31.78, 32.36, 33.61, 33.9, 34.35, and 34.62 +/- 0.2.

**[0072]** In a preferred embodiment, the substantially pure crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl)- propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 4.17, 5.83, 6.28, 7.96, 8.35, 11.35, 11.66, 12.57, 14.34, 14.89, 15.95, 16.74, 17.49, 18.9, 21.1, 23.86, 25.3, 27.04, 27.44, and 30.13 +/- 0.2. In other preferred embodiments, the substantially pure crystal Form V has two or more of the XRPD peaks identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified above.

**[0073]** In one embodiment, the substantially pure crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1723.8, 1676.0, 1599.0, 1557.5, 1499.4, 1462.2, 1402.4, 1318.0, 1272.4, 1253.7, 1169.9, 1138.0, 1102.7, 1061.4, 967.2, 932.8, 902.0, 857.4, 801.2, 785.9, 693.9, 677.2, 665.1, 631.8, and 533.4 cm<sup>-1</sup>.

**[0074]** In another embodiment, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 10.1, 12.1, 13.3, 14.5, 15.0, 16.0, 16.6, 17.0, 17.4, 18.8, 19.2, 19.7, 21.1, 22.3, 23.9, 24.2, 24.8, 25.7, 26.7, 27.6, 28.6, 28.9, 29.3, 29.7, 30.0, 30.5, 30.8, 31.3, 33.3, 33.7, 34.3, 35.0, 35.5, 36.5, 36.7, 37.4, and 39.5 +/- 0.2. In other preferred embodiments, the substantially pure crystal Form VI has two or more of the XRPD peaks identified above. In further embodiments, the crystal form has three or more of the XRPD peaks identified above.

**[0075]** In a preferred embodiment, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern that may contain at least one 2- theta position selected from the group consisting of those at about 10.1, 14.5, 16.0, 21.1, 24.8, and 25.7 +/- 0.2. In a further preferred embodiment, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has a characteristic X- ray powder diffraction (XRPD) pattern selected from the group consisting of (F1) 2- theta positions at about 10.1, 14.5, 16.0, 21.1, 24.8, and 25.7 +/- 0.2; (F2) 2- theta positions at about 14.5, 16.0, 21.1, 24.8, and 25.7 +/- 0.2; (F3) 2- theta positions at about 21.1, 24.8, and 25.7 +/- 0.2; and (F4) a 2- theta position at about 10.1 +/- 0.2.

**[0076]** In one embodiment, the substantially pure crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has an infrared spectra pattern that may contain at least one spectral line position selected from the group consisting of those at 1607, 1555, 1468, 1400, 1361, 1316, 1280, 1218, 1165, 1102, 1014, 976, 938, 760, and 698  $\text{cm}^{-1}$ .

**[0077]** In other preferred embodiments, the crystal form has two or more of the XRPD peaks identified in any one of the embodiments disclosed herein. In further embodiments, the crystal form has three or more of the XRPD peaks identified in any one of the embodiments disclosed herein. In yet further embodiments, the crystal form has four or more of the XRPD peaks identified in any one of the embodiments disclosed herein.

**[0078]** In another embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is characterized by at least one of the following properties: an X- ray powder diffraction pattern substantially similar to that presented in the figure selected from the group consisting of Fig. 1, Fig. 4, Fig. 7, Fig. 10, Fig. 13, and Fig. 16; a differential scanning calorimetry thermogram substantially similar to that presented in the figure selected from the group consisting of Fig. 3, Fig. 6, Fig. 9, Fig. 12, Fig. 15, Fig. 18a, Fig. 18b, and Fig. 18c; or an infrared spectra substantially similar to that presented in the figure selected from the group consisting of Fig. 17a, 17b, 17c, and 17d.

**[0079]** In a preferred embodiment, the substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is characterized by at least one of the following properties: a differential scanning calorimetry thermogram exhibiting a large endotherm at about 102°C, and having an onset temperature at about 87°C; or an infrared spectra having characteristic infrared spectral peak positions at 1607, 1555, 1468, 1400, 1361, 1316, 1280, 1218, 1165, 1102, 1014, 976, 938, 760, and 698  $\text{cm}^{-1}$ . In one preferred embodiment, the substantially pure crystalline form is characterized as having an infrared spectra having at least two of the characteristic infrared spectral peak positions at 1607, 1555, 1468, 1400, 1361, 1316, 1280, 1218, 1165, 1102, 1014, 976, 938, 760, and 698  $\text{cm}^{-1}$ . In another preferred embodiment, the substantially pure crystalline form is characterized as having an infrared spectra having at least three of the characteristic infrared spectral peak positions. In yet another preferred embodiment, the substantially pure crystalline form is characterized as having an infrared spectra having at least four of the characteristic infrared spectral peak positions.

**[0080]** In a particularly preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has characteristic X- ray powder diffraction (XRPD) 2- theta positions at about 21.1, 24.8, and 25.7  $\pm$  0.2. In another preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has characteristic X- ray powder diffraction (XRPD) 2- theta positions at about 14.5, 16.0, 21.1, 24.8, and 25.7  $\pm$  0.2. In a further preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has characteristic X- ray powder diffraction (XRPD) 2- theta positions at about 10.1, 14.5, 16.0, 21.1, 24.8, and 25.7  $\pm$  0.2. In a further preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has characteristic X- ray powder diffraction (XRPD) 2- theta position at about 10.1  $\pm$  0.2.

**[0081]** In a particularly preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is characterized by at least one of the following properties: an X- ray powder diffraction pattern substantially similar to that presented in Fig. 16; a differential scanning calorimetry thermogram substantially similar to that presented in Fig. 18a, Fig. 18b, or Fig. 18c; or an infrared spectra substantially similar to that presented in Fig. 17a, 17b, or 17c.

**[0082]** In a preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is characterized by a differential scanning calorimetry thermogram exhibiting a large endotherm at a temperature ranging from about 102°C to 108°C due to the melt. In a particularly preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is characterized by a differential scanning calorimetry thermogram exhibiting a large endotherm at about 102°C due to the melt, and wherein the onset temperature is at about 87°C.

**[0083]** In a particularly preferred embodiment, the present subject matter contemplates the substantially pure crystalline Form VI, wherein the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one has characteristic Fourier transform infrared spectral peak positions at 1607, 1555, 1468, 1400, 1361, 1316, 1280, 1218, 1165, 1102, 1014, 976, 938, 760, and 698  $\text{cm}^{-1}$ .

**[0084]** Additionally, the present subject matter contemplates substantially pure crystalline forms of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one substantially devoid of degradation products of the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In this regard, the present subject matter further contemplates a substantially pure crystalline Form I, II, III, IV, V, or VI of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one having not less than 97% purity with respect to other forms of the compound or any degradation products.

#### Physical Characteristics

**[0085]** The six crystalline forms of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one designated as Form I, II, III, IV, V, and VI can readily be distinguished from one another by examination of their respective characteristic x- ray diffraction (XRPD) patterns (see Figs 1, 4, 7, 10, 13, and 16, respectively), characteristic IR spectra (see Figs. 2, 5, 8, 11, 14, and 17a- c, respectively), or differential scanning calorimetry (DSC) thermograms (see Fig. 3, 6, 9, 12, 15, and 18a- c, respectively) . Form VI can also be characterized by its physical structure and atomic bond orientation (see Fig. 19) and by packing diagrams of Form VI along the a, b, and c, crystallographic axes (see Fig. 20, 21, 22, and 23) .

#### Form I

**[0086]** Crystalline Form I of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1 H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was in part identified by its characteristic XRPD pattern. The XRPD pattern for Form I was measured on a D5000 Siemens X- ray diffractometer, having a graphite secondary monochromator and a scintillation detector. The anode was of copper (wavelength  $\text{CuK}\alpha$ : 1.541838 Å, V= 50 kV, I=20 mA) and the ambient temperature was kept at 21°C.

**[0087]** A characteristic XRPD pattern distinctive to Form I was observed, as represented by Fig. 1. The characteristic 2-theta positions and corresponding intensities observed for this specific pattern are summarized below in Table 1.

**TABLE 1**

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form I</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Intensities (counts)</b>
4.11	115
8.22	128
9.39	66
11.29	161
12.41	102
13.21	95
13.62	160
14.43	206
14.93	231
15.7	104
16.74	215
17.3	134
17.57	178
18.79	296
20.88	151
21.88	125
22.62	98
23.64	119



(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form I</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Intensities (counts)</b>
23.82	177
25.2	234
26.77	206
27.21	245
28.57	116
29.16	113
29.97	110
30.75	81
31.35	93
45.28	56
48.87	42
55.02	36

**[0088]** The most relevant 2-theta positions and corresponding intensities observed for this specific XRPD pattern for Form I are summarized below in Table 2.

**TABLE 2**

<b>Most Relevant XRPD 2-Theta Positions and Intensities for Form I</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Intensities (counts)</b>
4.11	115
8.22	128
9.39	66
11.29	161
12.41	102
13.62	160
16.74	215
17.57	178
18.79	296
23.82	177
25.2	234

**[0089]** A set of XRPD peaks that uniquely characterizes Form I has 2-theta positions at about 4.11, 8.22, 9.39, 11.29, 12.41, 13.62, 16.74, 17.57, 18.79, 23.82, and 25.2 +/- 0.2.

**[0090]** The complete infrared (IR) spectrum of the crystalline Form I is shown in Fig. 2 and is characterized as summarized below in Table 3. All IR spectrum analyses obtained and discussed herein were generated on a Bomem BM-100 spectrophotometer, unless otherwise specified.

TABLE 3

<i>Infrared Spectrum for Form I</i>	
Frequency (cm <sup>-1</sup> ) (+/- 0.1)	
1673.3	
1600.0	
1557.5	
1501.2	
1462.7	
1403.9	
1319.4	
1273.9	
1254.6	
1139.0	
1101.8	
1061.8	
967.2	
902.7	
801.3	
783.6	
664.1	
630.1	

[0091] Form I was further observed to have a characteristic endothermic peak using DSC analysis, as represented by Fig. 3. All DSC analyses obtained herein were measured on a Mettler-Toledo DSC-20 instrument, unless otherwise specified. The samples (each of about 1 mg), were enclosed in aluminium pans having small holes accurately weighted, heated over a temperature range of 40-180°C, at a scan rate of 10°/min, under a nitrogen purge. The apparatus was calibrated with (99.9% pure) indium.

#### Form II

[0092] Crystalline Form II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was likewise in part identified by its characteristic XRPD pattern. The XRPD pattern for Form II was measured on a D5000 Siemens X- ray diffractometer, having a graphite secondary monochromator and a scintillation detector. The anode was of copper (wavelength CuK $\alpha$ : 1.541838 Å, V= 50 kV, I=20 mA) and the ambient temperature was kept at 21°C.

[0093] A characteristic XRPD pattern distinctive to Form II was observed, as represented by Fig. 4. The characteristic 2-theta positions and corresponding intensities observed for this specific pattern are summarized below in Table 4.

TABLE 4

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form II</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Intensities (counts)</b>
2.72	176
5.31	62
6.35	126
7.98	124

# EP 2 650 291 A1

(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form II</b>		
	<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Intensities (counts)</b>
5	8.37	177
	9.01	56
	11.4	112
10	11.7	188
	12.6	208
	13.15	160
	14.42	402
15	14.98	568
	15.93	128
	16.77	511
20	17.56	348
	17.91	181
	18.86	589
	19.37	108
25	21.08	272
	21.78	147
	22.31	161
30	22.82	115
	23.84	300
	25.32	294
	26	125
35	26.83	379
	27.35	432
	28.5	161
40	28.96	122
	29.38	147
	30.14	120
	31.58	131
45	32.41	109
	33.63	105
	34.94	77
50	46.1	85

**[0094]** The most relevant 2-theta positions and corresponding intensities observed for this specific XRPD pattern for Form II are summarized below in Table 5.

55

**TABLE 5 Most Relevant XRPD 2-Theta Positions and Intensities for Form II**

Characteristic XRPD 2-theta positions (+/-0.2)	Characteristic XRPD Intensities (counts)
6.35	126
7.98	124
8.37	177
11.4	112
11.7	188
12.6	208
13.15	160
14.42	402
14.98	568
16.77	511
17.56	348
18.86	589
21.08	272
23.84	300
25.32	294
26.83	379
27.35	432

**[0095]** A set of XRPD peaks that uniquely characterizes Form II has 2-theta positions at about 6.35, 7.98, 8.37, 11.4, 11.7, 12.6, 13.15, 14.42, 14.98, 16.77, 17.56, 18.86, 21.08, 23.84, 25.32, 26.83, and 27.35 +/- 0.2.

**[0096]** The complete infrared spectrum of the crystalline Form II is shown in Fig. 5 and is characterized as summarized below in Table 6.

**TABLE 6**

<i>Infrared Spectrum for Form II</i>	
Frequency (cm <sup>-1</sup> ) (+/- 0.1)	
1677.0	
1600.0	
1557.5	
1498.8	
1462.3	
1318.2	
1272.4	
1253.0	
1170.2	
1137.7	
1102.0	
1060.7	
967.0	
932.6	

(continued)

<b>Infrared Spectrum for Form II</b>	
<b>Frequency (cm<sup>-1</sup>) (+/- 0.1)</b>	
902.0	
857.2	
801.3	
785.1	
693.5	
664.9	
630.7	

**[0097]** Form II was further observed to have a characteristic endothermic peak using DSC analysis, as represented by Fig. 6.

### Form III

**[0098]** Crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was in part identified by its characteristic XRPD pattern. The XRPD pattern for Form III was measured at room temperature using a Philips X'Pert diffractometer equipped with a  $\theta/2\theta$  goniometer, a Cu tube working at 50 kV and 40 mA (CuK $\alpha$  radiation,  $\lambda = 1.5419$  Å), a divergence slit =  $1/4^\circ$ , Soller slits = 0.04 rad, an anti- scatter slit =  $1/4^\circ$ , a receiving slit = 0.10 mm, and a secondary curved graphite monochromator. Data were collected in the range 2-  $35^\circ$  of 2theta using a step- scan technique with a step size =  $0.02^\circ$  and a time per step = 20 s.

**[0099]** A characteristic XRPD pattern distinctive to Form III was observed, as represented by Fig. 7. The characteristic 2-theta positions and corresponding intensities observed for this specific pattern are summarized below in Table 7.

**TABLE 7**

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form III</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
4.08	51
5.73	34
6.22	13
7.77	36
8.15	60
8.80	30
11.25	61
11.47	29
12.44	48
13.09	17
14.33	46
14.68	98
14.89	60
15.57	68
16.35	43
16.68	58
17.27	29

# EP 2 650 291 A1

(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form III</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
17.63	40
18.66	100
19.32	12
20.85	71
22.12	25
22.49	18
23.58	29
24.63	20
25.02	24
26.65	48
27.12	77
28.74	15
29.11	17
29.81	23
31.35	10
33.48	8

**[0100]** The most relevant 2-theta positions and corresponding intensities observed for this specific XRPD pattern for Form III are summarized below in Table 8.

**TABLE 8**

<b>Most Relevant XRPD 2-Theta Positions and Intensities for Form III</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
4.08	51
5.73	34
6.22	13
7.77	36
8.15	60
8.80	30
11.25	61
11.47	29
12.44	48
13.09	17
15.57	68
17.63	40
18.66	100
20.85	71
26.65	48

## EP 2 650 291 A1

(continued)

<b>Most Relevant XRPD 2-Theta Positions and Intensities for Form III</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
27.12	77

**[0101]** A set of XRPD peaks that uniquely characterizes Form III has 2-theta positions at about 4.08, 5.73, 6.22, 7.77, 8.15, 8.80, 11.25, 11.47, 12.44, 13.09, 15.57, 17.63, 18.66, 20.85, 26.65, and 27.12 +/- 0.2.

**[0102]** The complete infrared spectrum of the crystalline Form III is shown in Fig. 8 and is characterized as summarized below in Table 9.

**TABLE 9**

<b>Infrared Spectrum for Form III</b>	
<b>Frequency (cm<sup>-1</sup>) (+/-0.1)</b>	
1677.0	
1600.0	
1557.5	
1498.3	
1462.6	
1403.0	
1318.4	
1272.5	
1254.1	
1170.0	
1138.7	
1101.6	
1060.2	
1016.4	
966.7	
932.7	
902.4	
855.5	
801.5	
785.8	
694.0	
677.9	
665.4	
631.7	
532.7	
411.6	

**[0103]** Form III was further observed to have a characteristic endothermic peak onset observed at about 99 +/- 5 degrees C using DSC analysis, as represented by Fig. 9.

Form IV

**[0104]** Crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was likewise in part identified by its characteristic XRPD pattern. The XRPD pattern for Form IV was measured at room temperature using a Philips X'Pert diffractometer equipped with a  $\theta/2\theta$  goniometer, a Cu tube working at 50 kV and 40 mA ( $\text{CuK}\alpha$  radiation,  $\lambda = 1.5419 \text{ \AA}$ ), a divergence slit =  $1/4^\circ$ , Soller slits =  $0.04 \text{ rad}$ , an anti- scatter slit =  $1/4^\circ$ , a receiving slit =  $0.10 \text{ mm}$ , and a secondary curved graphite monochromator. Data were collected in the range  $2-35^\circ$  of  $2\theta$  using a step- scan technique with a step size =  $0.02^\circ$  and a time per step =  $20 \text{ s}$ .

**[0105]** A characteristic XRPD pattern distinctive to Form IV was observed, as represented by Fig. 10. The characteristic  $2\theta$ -theta positions and corresponding intensities observed for this specific pattern are summarized below in Table 10.

**TABLE 10**

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form IV</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
3.74	14
4.15	39
7.5	8
8.33	64
9.61	30
11.16	60
11.61	7
12.49	34
13.29	84
13.64	96
14.41	51
15.43	22
15.74	22
16.90	42
17.71	12
18.25	20
18.74	51
19.30	21
20.43	11
21.78	10
23.20	15
24.26	24
24.78	57
25.11	100
26.03	8
26.86	16
27.25	15
28.00	10
29.05	7



(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form IV</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
30.07	8
30.91	5
32.05	5

**[0106]** The most relevant 2-theta positions and corresponding intensities observed for this specific XRPD pattern for Form IV are summarized below in Table 11.

**TABLE 11**

<b>Most Relevant XRPD 2-Theta Positions and Intensities for Form IV</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
4.15	39
7.5	8
8.33	64
9.61	30
11.16	60
12.49	34
13.29	84
13.64	96
14.41	51
16.90	42
18.74	51
24.78	57
25.11	100

**[0107]** A set of XRPD peaks that uniquely characterizes Form IV has 2-theta positions at about 4.15, 7.5, 8.33, 9.61, 11.16, 12.49, 13.29, 13.64, 14.41, 16.90, 18.74, 24.78, and 25.11 +/- 0.2.

**[0108]** The complete infrared spectrum of the crystalline Form IV is shown in Fig. 11 and is characterized as summarized below in Table 12.

**TABLE 12**

<b>Infrared Spectrum for Form IV</b>	
<b>Frequency (cm<sup>-1</sup>) (+/-0.1)</b>	
1671.0	
1601.0	
1557.5	
1503.3	
1462.7	
1404.1	
1319.8	
1274.4	

(continued)

<b>Infrared Spectrum for Form IV</b>	
<b>Frequency (cm<sup>-1</sup>) (+/-0.1)</b>	
1254.9	
1210.2	
1139.1	
1102.1	
1062.2	
967.4	
933.5	
902.8	
845.0	
801.4	
782.9	
693.8	
677.7	
663.6	
630.2	

**[0109]** Form IV was further observed to have a characteristic endothermic peak onset observed at about 121 +/- 5 degrees C using DSC analysis, as represented by Fig. 12.

#### Form V

**[0110]** Crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was likewise in part identified by its characteristic XRPD pattern. The XRPD pattern for Form V was measured at room temperature using a Philips X'Pert diffractometer equipped with a  $\theta/\theta$  goniometer, a Cu tube working at 50 kV and 40 mA (CuK $\alpha$  radiation,  $\lambda = 1.5419$  Å), a divergence slit =  $1/4^\circ$ , Soller slits = 0.04 rad, an anti- scatter slit =  $1/4^\circ$ , a receiving slit = 0.10 mm, and a secondary curved graphite monochromator. Data were collected in the range 2- 35° of 2theta using a step- scan technique with a step size = 0.02° and a time per step = 20 s.

**[0111]** A characteristic XRPD pattern distinctive to Form V was observed, as represented by Fig. 13. The characteristic 2-theta positions and corresponding intensities observed for this specific pattern are summarized below in Table 13.

**TABLE 13**

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form V</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
4.17	5
5.83	12
6.28	51
7.96	21
8.35	23
11.35	13
11.66	36
12.57	59
13.06	2

EP 2 650 291 A1

(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form V</b>		
	<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
5	13.34	1
	14.11	3
	14.34	7
10	14.52	4
	14.89	12
	15.09	7
15	15.95	5
	16.74	14
	16.98	2
20	17.21	3
	17.49	12
	17.91	11
25	18.82	25
	18.9	27
	20.36	1
30	21.1	37
	21.47	4
	21.79	4
35	22.27	4
	22.88	4
	23.86	28
40	25.3	100
	25.95	5
	26.2	3
45	26.74	7
	27.04	16
	27.44	25
50	28.43	2
	28.98	2
	29.41	6
55	30.13	13
	30.73	1
	30.98	1
	31.78	7
	32.36	1
	33.61	2
	33.9	1

(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form V</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
34.35	1
34.62	2

**[0112]** The most relevant 2-theta positions and corresponding intensities observed for this specific XRPD pattern for Form V are summarized below in Table 14.

**TABLE 14**

<b>Most Relevant XRPD 2-Theta Positions and Intensities for Form V</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
4.17	5
5.83	12
6.28	51
7.96	21
8.35	23
11.35	13
11.66	36
12.57	59
14.34	7
14.89	12
15.95	5
16.74	14
17.49	12
18.9	27
21.1	37
23.86	28
25.3	100
27.04	16
27.44	25
30.13	13

**[0113]** A set of XRPD peaks that uniquely characterizes Form V has 2-theta positions at about 4.17, 5.83, 6.28, 7.96, 8.35, 11.35, 11.66, 12.57, 14.34, 14.89, 15.95, 16.74, 17.49, 18.9, 21.1, 23.86, 25.3, 27.04, 27.44, and 30.13 +/- 0.2.

**[0114]** The complete infrared spectrum of the crystalline Form V is shown in Fig. 14 and is characterized as summarized below in Table 15.

**TABLE 15**

<b>Infrared Spectrum for Form V</b>	
<b>Frequency (cm<sup>-1</sup>) (+/-0.1)</b>	
1723.8	
1676.0	

## EP 2 650 291 A1

(continued)

<b>Infrared Spectrum for Form V</b>	
<b>Frequency (cm<sup>-1</sup>) (+/-0.1)</b>	
5	1599.0
	1557.5
	1499.4
10	1462.2
	1402.4
	1318.0
	1272.4
15	1253.7
	1169.9
	1138.0
20	1102.7
	1061.4
	967.2
25	932.8
	902.0
	857.4
	801.2
30	785.9
	693.9
	677.2
35	665.1
	631.8
	533.4

**[0115]** Form V was further observed to have a characteristic endothermic peak onset observed at about 108 +/- 5 degrees C using DSC analysis, as represented by Fig. 15.

### Form VI

**[0116]** Crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was likewise in part identified by its characteristic XRPD pattern. Optimal XRPD methods to analyze Form VI can typically either entail spinning a capillary (Inel) using t method (1) shown below or spinning a flat sample (Shimadzu) using a method (2) shown below. Figure 16 is obtained using the following method (1) .

(1) The XRPD pattern for Form VI was measured using an Inel XRG-3000 diffractometer equipped with a CPS (Curved Position Sensitive) detector with a 2θ range of 120°. Real time data were collected using Cu-Kα radiation. The tube voltage and amperage were set to 40 kV and 30 mA, respectively. The monochromator slit was set at 5 mm by 160 μm. The pattern is displayed from 2.5-40 °2θ. Samples were prepared for analysis by packing them into thin-walled glass capillaries. Each capillary was mounted onto a goniometer head that is motorized to permit spinning of the capillary during data acquisition. The samples were analyzed for 300 sec. Instrument calibration was performed using a silicon reference standard.

## EP 2 650 291 A1

(2) Alternatively, X-ray powder diffraction (XRPD) analyses were performed using a Shimadzu XRD-6000 X-ray powder diffractometer using Cu K $\alpha$  radiation. The instrument was equipped with a long fine focus X-ray tube. The tube voltage and amperage were set to 40 kV and 40 mA, respectively. The divergence and scattering slits were set at 1° and the receiving slit was set at 0.15 mm. Diffracted radiation was detected by a NaI scintillation detector. A  $\theta$ -2 $\theta$  continuous scan at 1°/min (0.4 sec/0.02° step) from 2.5 to 40° 2 $\theta$  was used. The sample was spun at a rate of 25 rpm. A silicon standard was analyzed to check the instrument alignment. Data were collected and analyzed using XRD-6100/7000 v. 5.0. Samples were prepared for analysis by placing them in an aluminum holder with silicon well.

**[0117]** A characteristic XRPD pattern distinctive to Form VI was observed using the above method (1), as represented by Fig. 16. The characteristic 2-theta positions and corresponding intensities observed for this specific pattern are summarized below in Table 16.

**TABLE 16**

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form VI</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
10.1	20
12.1	26
13.3	31
14.5	60
15.0	13
16.0	100
16.6	32
17.0	14
17.4	29
18.8	12
19.2	12
19.7	15
21.1	63
22.3	45
23.9	12
24.2	12
24.8	24
25.7	39
26.7	23
27.6	39
28.6	35
28.9	13
29.3	16
29.7	10
30.0	12
30.5	15
30.8	13
31.3	19

(continued)

<b>Characteristic XRPD 2-Theta Positions and Intensities for Form VI</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
33.3	9
33.7	8
34.3	8
35.0	11
35.5	8
36.5	9
36.7	8
37.4	10
39.5	8

**[0118]** The most relevant 2-theta positions and corresponding intensities observed for this specific XRPD pattern for Form VI are summarized below in Table 17.

**TABLE 17**

<b>Most Relevant XRPD 2-Theta Positions and Intensities for Form VI</b>	
<b>Characteristic XRPD 2-theta positions (+/-0.2)</b>	<b>Characteristic XRPD Relative Intensity (%)</b>
10.1	20
14.5	60
16.0	100
16.6	32
17.0	14
17.4	29
19.7	15
21.1	63
22.3	45
24.8	24
25.7	39
26.7	23
27.6	39
28.6	35
28.9	13
29.3	16
30.5	15
30.8	13
31.3	19

**[0119]** A first set of XRPD peaks that uniquely characterizes Form VI has 2-theta positions at about 21.1, 24.8, and 25.7 +/- 0.2. A second set of XRPD peaks that uniquely characterizes Form VI has 2-theta positions at about 14.5, 16.0, 21.1, 24.8, and 25.7 +/- 0.2. A third set of XRPD peaks that uniquely characterizes Form VI has 2-theta positions at about 10.1, 14.5, 16.0, 21.1, 24.8, and 25.7 +/- 0.2. A fourth set of XRPD peaks that uniquely characterizes Form VI

has a 2-theta position at about 10.1 +/- 0.2.

**[0120]** Infrared spectra were acquired on a Magna-IR 860® Fourier transform infrared (FT-IR) spectrophotometer (Thermo Nicolet) equipped with an Ever-Glo mid/far IR source, an extended range potassium bromide (KBr) beamsplitter, and a deuterated triglycine sulfate (DTGS) detector. A diffuse reflectance accessory (the Collector™, Thermo Spectra-Tech) was used for sampling. Each spectrum represents 256 co-added scans collected at a spectral resolution of 2 cm<sup>-1</sup>. Sample preparation consisted of placing the sample into a 13-mm diameter cup and leveling the material with a frosted glass slide. A background data set was acquired with an alignment mirror in place. A Log 1/R (R = reflectance) spectrum was acquired by taking a ratio of these two data sets against each other. Wavelength calibration was performed using polystyrene.

**[0121]** The infrared spectrum of the crystalline Form VI is shown in Figs. 17a-c and is characterized as summarized below in Table 18.

**TABLE 18**

<i>Infrared Spectrum for Form VI</i>	
<b>Frequency (cm<sup>-1</sup>) (+/-0.1)</b>	
1607	
1555	
1468	
1400	
1361	
1316	
1280	
1218	
1165	
1102	
1014	
976	
938	
760	
698	

**[0122]** Form VI was further observed to have a characteristic endothermic peak onset observed at a temperature ranging from about 102 to about 108 +/- 5 degrees C using DSC analysis, as represented by Figs. 18a-c.

**[0123]** Hot stage Microscopic Analysis was also performed on Form VI as follows. A sample of Form VI was sandwiched between glass coverslips and placed on a hot stage. The sample was heated at a controlled rate and visually observed using a microscope with crossed-polarized light. Visual changes with the sample were recorded. The melting temperature of Form VI was 100-103°C and no further changes were noted at 154°C.

**[0124]** Hot stage microscopy (HSM) was performed using a Linkam hot stage (model FTIR 600) mounted on a Leica DMLP microscope. Samples were observed using crossed polarized light. Images were captured using a SPOT Insight™ color digital camera with SPOT Software v. 4.5.9. The hot stage was calibrated using USP melting point standards.

#### **Structure Determination of Form VI**

**[0125]** Crystal Data and Data Collection Parameters Used for Determining the Structure of Form VI are provided (Table 19), calculated positional parameters and their estimated standard deviations (Table 20), calculated anisotropic temperature factor coefficients (Tables 21), calculated bond distances (Table 22), calculated bond angles (Table 23), calculated torsion angles (Tables 24), and calculated hydrogen bond distances and angles (Table 25).



## Data Collection

[0126] A colorless needle of  $C_{20}H_{16}ClF_2N_5O_2 \cdot H_2O$  was isolated from 1: 1 ethanol- water slurry and the structure determined by single crystal X- ray diffraction. The needle having approximate dimensions of 0.45 x 0.13 x 0.13 mm was mounted on a glass fiber in a random orientation. Preliminary examination and data collection were performed using

Mo  $K\alpha$  radiation ( $\lambda = 0.71073\text{\AA}$ ) on a Nonius KappaCCD equipped with a graphite crystal, incident beam monochromator. [0127] Cell constants for data collection were obtained from least-squares refinement, using the setting angles of 6937 reflections in the range  $3 < \theta < 25^\circ$ . The refined mosaicity from DENZO/SCALEPACK (Z. Otwinowski and W. Minor, Methods Enzymol., 276, 307, 1997, the contents of which are hereby incorporated by reference in their entirety) was  $1.24^\circ$  indicating poor crystal quality. The space group was determined by the program XPREP (XPREP in SHELXTL version 6.12, Bruker AXS Inc., Madison, Wisconsin, USA, 2002, the contents of the program, and user manuals of the program, are hereby incorporated by reference in their entirety.) From the systematic presences of  $h00\ h=2n$ ,  $0k0\ k=2n$ ,  $00l\ l=2n$ , and from subsequent least-squares refinement, the space group was determined to be  $P2_12_12_1$  (No. 19).

[0128] The data were collected at a temperature of 150K. Data were collected to a maximum  $2\theta$  of  $50.1^\circ$ .

## Data Reduction

[0129] A total of 6937 reflections were collected, of which 3516 were unique. Frames were integrated with DENZO-SMN (Z. Otwinowski and W. Minor, Methods Enzymol., 276, 307, 1997).

[0130] Lorentz and polarization corrections were applied to the data. The linear absorption coefficient is  $2.3\text{ cm}^{-1}$  for Mo  $K\alpha$  radiation. An empirical absorption correction using SCALEPACK (Z. Otwinowski and W. Minor, Methods Enzymol., 276, 307, 1997) was applied. Transmission coefficients ranged from 0.934 to 0.970. Intensities of equivalent reflections were averaged. The agreement factor for the averaging was 14.2% based on intensity.

## Structure Solution and Refinement

[0131] The structure was solved by direct methods using SIR2004 (M. C. Burla, R. Caliendo, M. Camalli, B. Carrozzini, G. L. Cascarano, L. De Caro, C. Giacovazzo, G. Polidori, and R. Spagna., J. Appl. Cryst., 38, 381, 2005, the contents of which are hereby incorporated by reference in their entirety). The remaining atoms were located in succeeding difference Fourier syntheses. Hydrogen atoms were included in the refinement but restrained to ride on the atom to which they are bonded. The structure was refined in full-matrix least-squares by minimizing the function:

$$\sum w \left( |F_o|^2 - |F_c|^2 \right)^2$$

[0132] The weight  $w$  is defined as  $1 / [\sigma^2(F_o^2) + 1.932_2 P]$  where  $P = (F_o^2 + 2F_c^2) / 3$ .

[0133] Scattering factors were taken from the "International Tables for Crystallography" ("International Tables for Crystallography", Vol. C, Kluwer Academic Publishers, Utrecht, The Netherlands, 1992, Tables 4.2.6.8 and 6.1.1.4., the contents of which are hereby incorporated by reference in their entirety). Of the 3516 reflections used in the refinements only 2654 reflections with  $F_o^2 > 2\sigma(F_o^2)$  were used in calculating  $R1$ . The final cycle of refinement included 293 variable parameters and converged (largest parameter shift was  $< 0.01$  times its estimated standard deviation) with unweighted and weighted agreement factors of:

$$R = \sum |F_o - F_c| / \sum F_o = 0.069$$

$$R_w = \sqrt{\left( \sum w (F_o^2 - F_c^2)^2 / \sum w (F_o^2)^2 \right)} = 0.102$$

[0134] The standard deviation of an observation of unit weight was 1.09. The highest peak in the final difference Fourier had a height of  $0.26\text{ e}/\text{\AA}^3$ . The minimum negative peak had a height of  $-0.30\text{ e}/\text{\AA}^3$ . The factor for the determination of the absolute structure (H. D. Flack, Acta Cryst., A39, 876, 1983, the contents of which are hereby incorporated by reference in their entirety) refined to 0.02.

[0135] Refinement was performed on a LINUX PC using SHELX-97 (G. M. Sheldrick, SHELXL97. A Program for

Crystal Structure Refinement. Univ. of Gottingen, Germany, 1997, the contents of the program, and user manuals of the program, are hereby incorporated by reference in their entirety). Crystallographic drawings were done using programs ORTEP (C. K. Johnson, ORTEP II, Report ORNL-5138, Oak Ridge National Laboratory, Tennessee, USA, 1976 the contents of the report, the program, and user manuals of the program, are hereby incorporated by reference in their entirety) and Mercury (Mercury 1.4.1, Cambridge Crystallographic Diffraction Center, Cambridge, 2005, the contents of the program, and user manuals of the program, are hereby incorporated by reference in their entirety).

**[0136]** The orthorhombic cell parameters and calculated volume were:  $a = 12.0968$  (15) Å,  $b = 12.6245$  (16) Å,  $c = 13.3520$  (19) Å,  $\alpha = \beta = \gamma = 90^\circ$ ,  $V = 2039.1$  (5) Å<sup>3</sup>. For  $Z = 4$  and formula weight = 449.85 g/mol, the calculated density is 1.47 g/cm<sup>3</sup>. The quality of the structure obtained was reasonable, as indicated by the  $R$ -value of 6.9%. Usually  $R$ -values in the range of 2 to 6% are quoted for the most reliably determined structures (J. Glusker, K. Trueblood, Crystal Structure Analysis: A Primer, 2nd ed.; Oxford University press: New York, 1985; p. 87, the contents of the entire book are hereby incorporated by reference in their entirety).

### Structure Drawings and Packing Diagrams

**[0137]** An ORTEP drawing of Form VI is shown in Figure 19. The asymmetric unit shown in Figure 19 contains one molecule of Form VI and one water molecule.

**[0138]** Packing diagrams viewed along the  $a$ ,  $b$ , and  $c$  crystallographic axes are shown in Figure 20, Figure 21, and Figure 22, respectively. Hydrogen bonding between adjacent Form VI and water molecules generates a three-dimensional network. The water molecules reside in channels running parallel to the crystallographic  $a$  axis (Figure 23). Packing diagrams were prepared using Mercury modeling software. Hydrogen bonds are represented as dashed lines.

TABLE 19

Crystal Data and Data Collection Parameters for Form VI

formula	$C_{20}H_{18}ClF_2N_5O_3$
formula weight	449.85
space group	$P2_12_12_1$ (No. 19)
$a$ , Å	12.0968(15)
$b$ , Å	12.6245(16)
$c$ , Å	13.3520(19)
$V$ , Å <sup>3</sup>	2039.1(5)
$Z$	4
$d_{calc}$ , g cm <sup>-3</sup>	1.465
crystal dimensions, mm	0.45x0.13x0.13
temperature, K	150.
radiation (wavelength, Å)	Mo K $\alpha$ (0.71073)
monochromator	graphite
linear abs coef, mm <sup>-1</sup>	0.234
absorption correction applied	empirical <sup>a</sup>
transmission factors: min, max	0.93, 0.97
diffractometer	Nonius KappaCCD
$h, k, l$ range	-14 to 13 -13 to 15 -15 to 14
$2\theta$ range, deg	6.10-50.06
mosaicity, deg	1.24
programs used	SHELXTL
$F_{000}$	928.0
weighting	$1/[\sigma^2(F_o^2)+1.9322P]$ where $P=(F_o^2+2F_c^2)/3$
data collected	6937
unique data	3516
$R_{int}$	0.142
data used in refinement	3516
cutoff used in $R$ -factor calculations	$F_o^2 > 2.0\sigma(F_o^2)$
data with $I > 2.0\sigma(I)$	2654

(continued)

## Crystal Data and Data Collection Parameters for Form VI

number of variables	293
largest shift/esd in final cycle	0.00
$R(F_o)$	0.069
$R_w(F_o^2)$	0.102
goodness of fit	1.090
absolute structure determination	Flack parameter <sup>b</sup> (0.0(1))

<sup>a</sup>Otwinowski Z. & Minor, W. *Methods Enzymol.* **1997**, 276, 307.<sup>b</sup>Flack, H. D. *Acta Cryst., Sect. A* **1983**, A39, 876.**[0139]** The contents of each of the above documents are hereby incorporated by reference in their entirety.**TABLE 20**

Positional Parameters and Their Estimated Standard Deviation for Form VI.

Atom	x	y	z	U(Å <sup>2</sup> )
Cl(6)	0.38011(12)	0.06405(10)	0.20395(11)	0.0426(4)
F(122)	0.1484(2)	-0.7003(2)	-0.0092(2)	0.0421(9)
F(124)	0.2838(3)	-1.04231(19)	-0.0623(2)	0.0380(9)
O(2)	0.1710(3)	-0.4220(2)	0.1181(2)	0.0254(9)
O(1W)	0.5178(3)	-0.6048(3)	-0.2666(2)	0.0177(9)
O(121)	0.4677(2)	-0.5746(2)	-0.0753(2)	0.0142(8)
N(1)	0.3503(3)	-0.4423(3)	0.0691(3)	0.0164(10)
N(9)	0.4826(3)	-0.3028(3)	0.0776(3)	0.0257(12)
N(14)	0.2910(3)	-0.4375(3)	-0.1488(2)	0.0154(10)
N(15)	0.2114(3)	-0.3697(3)	-0.1160(3)	0.0187(10)
N(17)	0.3499(3)	-0.2784(3)	-0.1897(3)	0.0247(12)
C(2)	0.2617(4)	-0.3816(4)	0.1032(3)	0.0183(14)
C(3)	0.2891(4)	-0.2712(3)	0.1223(3)	0.0200(12)
C(4)	0.2081(4)	-0.2000(4)	0.1565(3)	0.0297(15)
C(5)	0.2348(4)	-0.0974(4)	0.1799(3)	0.0310(15)
C(6)	0.3431(4)	-0.0648(4)	0.1691(3)	0.0267(14)
C(7)	0.4250(4)	-0.1310(4)	0.1345(3)	0.0267(15)
C(8)	0.3984(4)	-0.2358(3)	0.1100(3)	0.0207(12)
C(10)	0.4543(4)	-0.3990(4)	0.0599(3)	0.0240(15)
C(11)	0.3343(4)	-0.5585(3)	0.0555(3)	0.0177(12)
C(12)	0.3568(3)	-0.5985(3)	-0.0518(3)	0.0137(10)
C(13)	0.2799(3)	-0.5508(3)	-0.1335(3)	0.0147(12)
C(16)	0.2501(4)	-0.2765(3)	-0.1432(3)	0.0207(14)
C(18)	0.3719(4)	-0.3820(3)	-0.1928(3)	0.0220(14)
C(111)	0.4023(4)	-0.6178(4)	0.1350(3)	0.0287(15)
C(121)	0.3371(3)	-0.7191(3)	-0.0559(3)	0.0147(10)
C(122)	0.2358(4)	-0.7645(4)	-0.0344(4)	0.0230(14)
C(123)	0.2148(4)	-0.8713(4)	-0.0346(3)	0.0280(15)
C(124)	0.3019(4)	-0.9365(4)	-0.0591(3)	0.0243(15)
C(125)	0.4059(4)	-0.8976(4)	-0.0816(3)	0.0240(15)
C(126)	0.4207(4)	-0.7891(3)	-0.0799(3)	0.0167(12)
H(121)	0.472(5)	-0.597(5)	-0.140(5)	0.09(2)*
H(1W1)	0.450(6)	-0.606(5)	-0.305(5)	0.08(2)*

(continued)

Positional Parameters and Their Estimated Standard Deviation for Form VI.

Atom	$\bar{x}$	$\bar{y}$	$\bar{z}$	$U(\text{\AA}^2)$
H(1W2)	0.554(6)	-0.671(6)	-0.281(5)	0.10(2)*

Starred atoms were refined isotropically

$$U_{eq} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_j^* a_i a_j$$

TABLE 21

Anisotropic Temperature Factor Coefficients - U's for Form VI.

Name	$U(1,1)$	$U(2,2)$	$U(3,3)$	$U(1,2)$	$U(1,3)$	$U(2,3)$
Cl(6)	0.0667(10)	0.0160(6)	0.0452(8)	0.0063(7)	-0.0101(7)	-0.0137(6)
F(122)	0.0162(15)	0.0290(17)	0.081(2)	0.0003(14)	0.0181(15)	0.0111(15)
F(124)	0.065(2)	0.0081(14)	0.0408(17)	-0.0112(14)	-0.0032(16)	0.0013(12)
O(2)	0.0259(19)	0.0216(18)	0.0286(19)	0.0062(16)	0.0115(15)	0.0013(15)
O(1W)	0.0132(17)	0.0184(18)	0.0216(19)	0.0062(15)	0.0012(14)	-0.0012(14)
O(121)	0.0086(15)	0.0142(16)	0.0197(18)	-0.0045(14)	0.0019(13)	-0.0024(14)
N(1)	0.0129(19)	0.0162(19)	0.020(2)	0.0030(19)	0.0018(16)	-0.0032(17)
N(9)	0.022(2)	0.015(2)	0.040(3)	0.0064(19)	-0.0055(19)	-0.0124(18)
N(14)	0.0161(19)	0.0140(19)	0.016(2)	0.0026(19)	0.0006(16)	0.0007(17)
N(15)	0.019(2)	0.016(2)	0.021(2)	0.0089(18)	-0.0013(17)	-0.0018(16)
N(17)	0.022(2)	0.013(2)	0.039(3)	-0.0065(19)	-0.001(2)	0.0042(18)
C(2)	0.013(3)	0.028(3)	0.014(2)	0.007(2)	0.0034(19)	-0.001(2)
C(3)	0.028(3)	0.017(2)	0.015(2)	0.008(2)	-0.005(2)	-0.0020(19)
C(4)	0.038(3)	0.025(3)	0.026(3)	0.010(3)	0.012(2)	0.002(2)
C(5)	0.047(3)	0.018(3)	0.028(3)	0.017(3)	0.009(3)	-0.001(2)
C(6)	0.040(3)	0.016(2)	0.024(3)	0.008(3)	-0.004(2)	-0.009(2)
C(7)	0.033(3)	0.021(3)	0.026(3)	0.000(2)	-0.011(2)	-0.007(2)
C(8)	0.026(3)	0.018(2)	0.018(2)	0.005(2)	-0.005(2)	-0.006(2)
C(10)	0.020(3)	0.023(3)	0.029(3)	0.014(2)	-0.001(2)	-0.005(2)
C(11)	0.020(2)	0.012(2)	0.021(3)	0.001(2)	0.002(2)	0.000(2)
C(12)	0.010(2)	0.012(2)	0.019(2)	0.0033(19)	0.0014(19)	0.0011(18)
C(13)	0.016(2)	0.005(2)	0.023(3)	0.000(2)	0.001(2)	0.0005(19)
C(16)	0.030(3)	0.005(2)	0.027(3)	0.003(2)	-0.005(2)	-0.001(2)
C(18)	0.015(2)	0.024(3)	0.027(3)	-0.002(2)	0.002(2)	0.007(2)
C(111)	0.042(3)	0.027(3)	0.017(3)	0.003(3)	0.002(2)	0.002(2)
C(121)	0.013(2)	0.015(2)	0.016(2)	-0.003(2)	-0.0034(19)	-0.0021(19)
C(122)	0.011(2)	0.025(3)	0.033(3)	0.001(2)	0.006(2)	0.004(2)
C(123)	0.027(3)	0.022(3)	0.035(3)	-0.017(3)	-0.002(2)	0.010(2)
C(124)	0.038(3)	0.013(3)	0.022(3)	-0.006(3)	-0.009(2)	0.000(2)
C(125)	0.036(3)	0.018(3)	0.018(3)	0.008(2)	-0.003(2)	-0.003(2)
C(126)	0.015(2)	0.021(3)	0.014(2)	0.004(2)	-0.0002(19)	-0.0050(19)

[0140] The form of the anisotropic temperature factor is:

$\exp [-2\pi h^2 a^{*2} U(1,1) + k^2 b^{*2} U(2,2) + l^2 c^{*2} U(3,3) + 2hka^* b^* U(1,2) + 2hla^* c^* U(1,3) + 2klb^* c^* U(2,3)]$  where  $a^*$ ,  $b^*$ , and  $c^*$  are reciprocal lattice constants.

# EP 2 650 291 A1

**TABLE 22**

Table of Bond Distances in Angstrom for Form VI.

	<u>Atom 1</u>	<u>Atom 2</u>	<u>Distance</u>	<u>Atom 1</u>	<u>Atom 2</u>	<u>Distance</u>
5	Cl6	C6	1.750(5)	C6	C7	1.376(6)
	F122	C122	1.374(5)	C7	C8	1.400(6)
	F124	C124	1.354(5)	C7	H7	0.950
	O2	C2	1.226(5)	C10	H10	0.950
	O1W	H1W1	0.96(7)	C11	C111	1.538(6)
10	O1W	H1W2	0.96(7)	C11	C12	1.544(6)
	O121	C12	1.410(5)	C11	H11	1.000
	O121	H121	0.91(6)	C12	C121	1.541(6)
	N1	C10	1.378(5)	C12	C13	1.554(6)
15	N1	C2	1.393(5)	C13	H13A	0.990
	N1	C11	1.490(5)	C13	H13B	0.990
	N9	C10	1.283(5)	C16	H16	0.950
	N9	C8	1.393(5)	C18	H18	0.950
	N14	C18	1.340(5)	C111	H11A	0.980
20	N14	N15	1.361(5)	C111	H11B	0.980
	N14	C13	1.451(5)	C111	H11C	0.980
	N15	C16	1.317(5)	C121	C126	1.381(6)
	N17	C18	1.335(5)	C121	C122	1.383(6)
25	N17	C16	1.357(6)	C122	C123	1.371(6)
	C2	C3	1.456(6)	C123	C124	1.376(7)
	C3	C8	1.406(7)	C123	H123	0.950
	C3	C4	1.406(6)	C124	C125	1.383(7)
	C4	C5	1.371(6)	C125	C126	1.382(6)
30	C4	H4	0.950	C125	H125	0.950
	C5	C6	1.381(7)	C126	H126	0.950
	C5	H5	0.950			

**[0141]** Numbers in parentheses are estimated standard deviations in the least significant digits.

**TABLE 23**

Table of Bond Angles in Degrees for Form VI.

	<u>Atom 1</u>	<u>Atom 2</u>	<u>Atom 3</u>	<u>Angle</u>	<u>Atom 1</u>	<u>Atom 2</u>	<u>Atom 3</u>	<u>Angle</u>
40	H1W1	O1W	H1W2	105(5)	O121	C12	C11	107.7(3)
	C12	O121	H121	102(4)	C121	C12	C11	109.2(3)
	C10	N1	C2	120.8(3)	O121	C12	C13	109.3(3)
	C10	N1	C11	119.9(4)	C121	C12	C13	105.4(3)
	C2	N1	C11	118.8(3)	C11	C12	C13	114.7(3)
45	C10	N9	C8	115.9(4)	N14	C13	C12	115.2(3)
	C18	N14	N15	109.2(3)	N14	C13	H13A	108.50
	C18	N14	C13	130.2(4)	C12	C13	H13A	108.50
	N15	N14	C13	120.7(3)	N14	C13	H13B	108.50
50	C16	N15	N14	102.8(3)	C12	C13	H13B	108.50
	C18	N17	C16	102.1(4)	H13A	C13	H13B	107.50
	O2	C2	N1	120.8(4)	N15	C16	N17	115.2(4)
	O2	C2	C3	124.9(4)	N15	C16	H16	122.40
	N1	C2	C3	114.2(4)	N17	C16	H16	122.40
55	C8	C3	C4	119.4(4)	N17	C18	N14	110.7(4)
	C8	C3	C2	119.9(4)	N17	C18	H18	124.70
	C4	C3	C2	120.7(4)	N14	C18	H18	124.70

# EP 2 650 291 A1

(continued)

Table of Bond Angles in Degrees for Form VI.

	<u>Atom 1</u>	<u>Atom 2</u>	<u>Atom 3</u>	<u>Angle</u>	<u>Atom 1</u>	<u>Atom 2</u>	<u>Atom 3</u>	<u>Angle</u>
5	C5	C4	C3	120.9(5)	C11	C111	H11A	109.50
	C5	C4	H4	119.50	C11	C111	H11B	109.50
	C3	C4	H4	119.50	H11A	C111	H11B	109.50
	C4	C5	C6	118.8(5)	C11	C111	H11C	109.50
	C4	C5	H5	120.60	H11A	C111	H11C	109.50
10	C6	C5	H5	120.60	H11B	C111	H11C	109.50
	C7	C6	C5	122.5(4)	C126	C121	C122	115.6(4)
	C7	C6	Cl6	118.0(4)	C126	C121	C12	121.8(4)
	C5	C6	Cl6	119.4(4)	C122	C121	C12	122.6(4)
15	C6	C7	C8	119.1(4)	C123	C122	F122	116.0(4)
	C6	C7	H7	120.40	C123	C122	C121	124.8(4)
	C8	C7	H7	120.40	F122	C122	C121	119.2(4)
	N9	C8	C7	118.6(4)	C122	C123	C124	116.6(4)
	N9	C8	C3	122.1(4)	C122	C123	H123	121.70
20	C7	C8	C3	119.3(4)	C124	C123	H123	121.70
	N9	C10	N1	127.1(4)	F124	C124	C123	118.3(4)
	N9	C10	H10	116.40	F124	C124	C125	119.3(4)
	N1	C10	H10	116.40	C123	C124	C125	122.4(4)
	N1	C11	C111	109.0(3)	C126	C125	C124	117.8(4)
25	N1	C11	C12	114.4(3)	C126	C125	H125	121.10
	C111	C11	C12	112.8(3)	C124	C125	H125	121.10
	N1	C11	H11	106.70	C121	C126	C125	122.9(4)
	C111	C11	H11	106.70	C121	C126	H126	118.50
	C12	C11	H11	106.70	C125	C126	H126	118.50
30	O121	C12	C121	110.5(3)				

[0142] Numbers in parentheses are estimated standard deviations in the least significant digits.

TABLE 24

Table of Torsional Angles in Degrees for Form VI.

	<u>Atom 1</u>	<u>Atom 2</u>	<u>Atom 3</u>	<u>Atom 4</u>	<u>Angle</u>
35	C(10)	N(1)	C(2)	O(2)	-174.81 ( 0.39)
	C(10)	N(1)	C(2)	C(3)	2.77 ( 0.57)
	C(11)	N(1)	C(2)	O(2)	-2.80 ( 0.60)
	C(11)	N(1)	C(2)	C(3)	174.78 ( 0.35)
	C(2)	N(1)	C(10)	N(9)	-1.77 ( 0.69)
40	C(11)	N(1)	C(10)	N(9)	-173.69 ( 0.42)
	C(2)	N(1)	C(11)	C(12)	121.76 ( 0.42)
	C(2)	N(1)	C(11)	C(111)	-110.99 ( 0.43)
	C(10)	N(1)	C(11)	C(12)	-66.15 ( 0.51)
	C(10)	N(1)	C(11)	C(111)	61.10 ( 0.50)
45	C(10)	N(9)	C(8)	C(3)	-0.18 ( 0.61)
	C(10)	N(9)	C(8)	C(7)	177.40 ( 0.39)
	C(8)	N(9)	C(10)	N(1)	0.30 ( 0.66)
	C(13)	N(14)	N(15)	C(16)	-179.88 ( 0.34)
	C(18)	N(14)	N(15)	C(16)	0.23 ( 0.44)
50	N(15)	N(14)	C(13)	C(12)	-106.76 ( 0.39)
	C(18)	N(14)	C(13)	C(12)	73.10 ( 0.49)
	N(15)	N(14)	C(18)	N(17)	0.70 ( 0.46)

# EP 2 650 291 A1

(continued)

Table of Torsional Angles in Degrees for Form VI.

	<u>Atom 1</u>	<u>Atom 2</u>	<u>Atom 3</u>	<u>Atom 4</u>	<u>Angle</u>
5	C(13)	N(14)	C(18)	N(17)	-179.17 ( 0.36)
	N(14)	N(15)	C(16)	N(17)	-1.11 ( 0.49)
	C(18)	N(17)	C(16)	N(15)	1.51 ( 0.51)
	C(16)	N(17)	C(18)	N(14)	-1.27 ( 0.46)
	O(2)	C(2)	C(3)	C(4)	-2.88 ( 0.65)
10	O(2)	C(2)	C(3)	C(8)	174.82 ( 0.39)
	N(1)	C(2)	C(3)	C(4)	179.65 ( 0.38)
	N(1)	C(2)	C(3)	C(8)	-2.64 ( 0.56)
	C(2)	C(3)	C(4)	C(5)	176.35 ( 0.39)
	C(8)	C(3)	C(4)	C(5)	-1.36 ( 0.61)
15	C(2)	C(3)	C(8)	N(9)	1.44 ( 0.61)
	C(2)	C(3)	C(8)	C(7)	-176.12 ( 0.38)
	C(4)	C(3)	C(8)	N(9)	179.18 ( 0.39)
	C(4)	C(3)	C(8)	C(7)	1.62 ( 0.59)
	C(3)	C(4)	C(5)	C(6)	0.13 ( 0.63)
20	C(4)	C(5)	C(6)	Cl(6)	-177.11 ( 0.32)
	C(4)	C(5)	C(6)	C(7)	0.87 ( 0.64)
	Cl(6)	C(6)	C(7)	C(8)	177.41 ( 0.32)
	C(5)	C(6)	C(7)	C(8)	-0.59 ( 0.64)
	C(6)	C(7)	C(8)	N(9)	-178.31 ( 0.38)
25	C(6)	C(7)	C(8)	C(3)	-0.67 ( 0.60)
	N(1)	C(11)	C(12)	O(121)	60.12 ( 0.44)
	N(1)	C(11)	C(12)	C(13)	-61.82 ( 0.47)
	N(1)	C(11)	C(12)	C(121)	-179.83 ( 0.35)
	C(111)	C(11)	C(12)	O(121)	-65.16 ( 0.42)
30	C(111)	C(11)	C(12)	C(13)	172.90 ( 0.34)
	C(111)	C(11)	C(12)	C(121)	54.88 ( 0.44)
	O(121)	C(12)	C(13)	N(14)	-56.58 ( 0.41)
	C(11)	C(12)	C(13)	N(14)	64.51 ( 0.43)
	C(121)	C(12)	C(13)	N(14)	-175.33 ( 0.31)
35	O(121)	C(12)	C(121)	C(122)	178.91 ( 0.39)
	O(121)	C(12)	C(121)	C(126)	-0.02 ( 0.76)
	C(11)	C(12)	C(121)	C(122)	60.61 ( 0.52)
	C(11)	C(12)	C(121)	C(126)	-118.32 ( 0.42)
	C(13)	C(12)	C(121)	C(122)	-63.16 ( 0.50)
40	C(13)	C(12)	C(121)	C(126)	117.91 ( 0.40)
	C(12)	C(121)	C(122)	F(122)	0.62 ( 0.68)
	C(12)	C(121)	C(122)	C(123)	-178.33 ( 0.43)
	C(126)	C(121)	C(122)	F(122)	179.61 ( 0.39)
	C(126)	C(121)	C(122)	C(123)	0.66 ( 0.71)
45	C(12)	C(121)	C(126)	C(125)	178.31 ( 0.38)
	C(122)	C(121)	C(126)	C(125)	-0.69 ( 0.63)
	F(122)	C(122)	C(123)	C(124)	-179.60 ( 0.38)
	C(121)	C(122)	C(123)	C(124)	-0.62 ( 0.72)
	C(122)	C(123)	C(124)	F(124)	-178.55 ( 0.39)
50	C(122)	C(123)	C(124)	C(125)	0.59 ( 0.63)
	F(124)	C(124)	C(125)	C(126)	178.50 ( 0.36)
	C(123)	C(124)	C(125)	C(126)	-0.64 ( 0.61)
	C(124)	C(125)	C(126)	C(121)	0.70 ( 0.62)
55					

TABLE 25

Table of Hydrogen Bond Distances and Angles for Form VI.

<u>D</u>	<u>H</u>	<u>A</u>	<u>D-H</u>	<u>A-H</u>	<u>D-A</u>	<u>D-H-A</u>
O(121)	H(121)	O(1W)	0.91(7)	1.78(7)	2.653(5)	159(6)
O(1W)	H(1W1)	O(2)	0.97(7)	1.82(7)	2.775(6)	168(6)
O(1W)	H(1W2)	N(17)	0.96(7)	1.83(7)	2.776(7)	167(7)

## Pharmaceutical Compositions

[0143] In another preferred embodiment, the present subject matter relates to pharmaceutical compositions containing an anti- microbially or anti- fungally effective amount of any of the pure crystalline polymorphs I, II, III, IV, V, or VI of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl)- propyl] quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof, and a pharmaceutically acceptable carrier. The crystalline Form III, Form IV, or Form VI is particularly preferred in this regard. Form VI is most preferred in this regard. The present subject matter also contemplates compositions comprising one or more of the crystalline forms described herein. In a particularly preferred embodiment, these pharmaceutical compositions comprise any one of the pure crystalline Forms I, II, III, IV, V, or VI having a purity of at least 85% by weight, or not more than 15% by weight of other forms of the compound and any residual solvents. In a more preferred embodiment, these pharmaceutical compositions comprise any one of the pure crystalline Forms I, II, III, IV, V, or VI having a purity of at least 90% by weight, with a purity of at least 95% by weight being most preferred. In another preferred embodiment, these pharmaceutical compositions comprise any one of the pure crystalline Forms I, II, III, IV, V, or VI having a purity ranging from about 85% to 99.99% purity on a % weight basis. More preferably, the compositions comprise the crystalline form having a purity ranging from about 90% to 99.99% purity on a % weight basis. Further more preferably, the crystalline form has a purity ranging from about 93% to 99.99% purity on a % weight basis. Yet more preferably, the compositions comprise the crystalline form having a purity ranging from about 95% to 99.99% purity on a % weight basis. Yet even more preferably, the compositions comprise the crystalline form having a purity ranging from about 97% to 99.99% purity on a % weight basis. Of these embodiments, compositions comprising Form VI are especially preferred.

[0144] The phrase "pharmaceutically acceptable carrier" as used in this regard refers to any inactive ingredient present in one of the herein described compositions in an amount effective to enhance the stability, effectiveness, or otherwise of said composition. Non-limiting examples of such pharmaceutically acceptable carriers include diluents, excipients, suspending agents, lubricating agents, adjuvants, vehicles, delivery systems, emulsifiers, disintegrants, absorbants, adsorbents, preservatives, surfactants, colorants, flavorants, emollients, buffers, pH modifiers, thickeners, water softening agents, humectants, fragrances, stabilizers, conditioning agents, chelating agents, sweeteners, propellants, anticaking agents, viscosity increasing agents, solubilizers, plasticizers, penetration enhancing agents, glidants, film forming agents, fillers, coating agents, binders, antioxidants, stiffening agents, wetting agents, or any mixture of these components.

[0145] The carriers useful herein may further include one or more compatible solid or liquid filler, diluents, or encapsulating materials which are suitable for human or animal administration.

[0146] Biocompatible carriers, as used herein, are components that do not cause any interactions which substantially reduce the efficacy of the pharmaceutical composition in an ordinary user environment. Possible pharmaceutical carriers must be of sufficiently low toxicity to make them suitable for administration to the subject of treatment.

[0147] Some non-limiting examples of substances which can serve as a carrier herein include sugar, starch, cellulose and its derivatives, powered tragacanth, malt, gelatin, talc, stearic acid, magnesium stearate, calcium sulfate, vegetable oils, polyols, alginic acid, pyrogen-free water, isotonic saline, phosphate buffer solutions, cocoa butter (suppository base), emulsifier as well as other non-toxic pharmaceutically compatible substances used in other pharmaceutical formulations. Wetting agents and lubricants such as sodium lauryl sulfate, as well as coloring agents, flavoring agents, excipients, tableting agents, stabilizers, antioxidants, and preservatives may also be present.

[0148] Any non-toxic, inert, and effective carrier may be used to formulate the compositions contemplated herein. Suitable pharmaceutically acceptable carriers, excipients, and diluents in this regard are well known to those of skill in the art, such as those described in The Merck Index, Thirteenth Edition, Budavari et al., Eds., Merck & Co., Inc., Rahway, N.J. (2001); the CTFA (Cosmetic, Toiletry, and Fragrance Association) International Cosmetic Ingredient Dictionary and Handbook, Tenth Edition (2004); and the "Inactive Ingredient Guide", U.S. Food and Drug Administration (FDA) Center for Drug Evaluation and Research (CDER) Office of Management, <http://www.accessdata.fda.gov/scripts/cder/iig/index.cfm>, the contents of all of which are hereby incorporated by reference in their entirety. Examples of preferred pharmaceutically acceptable excipients, carriers and diluents useful in the present compositions include distilled water, physiological saline, Ringer's solution, dextrose solution, Hank's solution, and DMSO.



[0149] These additional inactive components, as well as effective formulations and administration procedures, are well known in the art and are described in standard textbooks, such as Goodman and Gilman's: The Pharmacological Bases of Therapeutics, 8th Ed., Gilman et al. Eds. Pergamon Press (1990) and Remington's Pharmaceutical Sciences, 18th Ed., Mack Publishing Co., Easton, Pa. (1990), both of which are incorporated by reference herein in their entirety.

[0150] The carrier may comprise, in total, from about 0.1% to about 99.99999% by weight of the pharmaceutical compositions presented herein.

[0151] Preferred compositions herein can be in the form of oral or topical compositions. The oral compositions contemplated herein may take the form of tablets, capsules, soft-gels, hard gels, solutions, suspensions, powders, dispersible granules, cachets, combinations thereof, or any other oral pharmaceutical dosage form as would commonly be known in the art.

[0152] A solid carrier can be one or more substances which may also act as diluents, flavoring agents, solubilizers, lubricants, suspending agents, binders, or tablet disintegration agents; it can also be an encapsulating material. In powders, the carrier can be a finely divided solid which is in admixture with the active compound. In a tablet, the active compound can be mixed with a carrier having the necessary binding properties in suitable proportions and compacted in the size and shape desired. Non-limiting examples of suitable solid carriers include magnesium carbonate, magnesium stearate, talc, sugar, lactose, pectin, dextrin, starch, gelatin, tragacanth, methyl cellulose, sodium carboxymethylcellulose, hydroxypropyl methylcellulose, hydroxyethylcellulose, hydroxypropylcellulose, other cellulose derivatives, a low melting wax, cocoa butter, and the like.

[0153] Similarly, the topical compositions contemplated herein may take the form of a gel, cream, lotion, suspension, emulsion, aerosol, ointment, foam, mousse, shampoo, nail lacquer, nail product, vaginal product, combinations thereof, or any other topical pharmaceutical dosage form commonly known in the art. Other pharmaceutical and cosmetic treatment compositions known to those skilled in the art, including liquids and balms, are additionally contemplated as falling within the scope of the present subject matter. Further, the present subject matter contemplates applying any of these compositions with an applicator. Non-limiting examples of useful applicators in this regard include a pledget, a pad, and combinations thereof. Additionally, the present subject matter further contemplates that any of these topical compositions can be provided in a package of less than 5 g topical composition as a unit of use.

[0154] Emulsions, such as oil-in-water or water-in-oil systems, as well as a base (vehicle or carrier) for the topical formulation can be selected to provide effectiveness of the active ingredient and/or avoid allergic and irritating reactions (e.g., contact dermatitis) caused by ingredients of the base or by the active ingredient.

[0155] In severe cases, occlusive therapy may be useful herein. Covering the treated area with a nonporous occlusive dressing can increase the absorption and effectiveness of the compounds and compositions described herein. Usually, a polyethylene film (plastic household wrap) is applied overnight over cream or ointment, since a cream or ointment is usually less irritating than lotion in occlusive therapy. Plastic tapes may be impregnated with drug and are especially convenient for treating isolated or recalcitrant lesions; children and (less often) adults may experience pituitary and adrenal suppression after prolonged occlusive therapy over large areas.

[0156] Suitable gelling agents which may be useful in the present compositions include but are not limited to aqueous gelling agents, such as neutral, anionic, and cationic polymers, and mixtures thereof. Exemplary polymers which may be useful in the instant compositions include carboxy vinyl polymers, such as carboxypolymethylene. A preferred gelling agent is Carbopol® brand polymer such as is available from Noveon Inc., Cleveland, OH. Carbopol® polymers are high molecular weight, crosslinked, acrylic acid-based polymers. Carbopol® homopolymers are polymers of acrylic acid crosslinked with allyl sucrose or allylpentaerythritol. Carbopol® copolymers are polymers of acrylic acid, modified by long chain (C10-C30) alkyl acrylates, and crosslinked with allyl-pentaerythritol.

[0157] Other suitable gelling agents include cellulosic polymers, such as gum arabic, gum tragacanth, locust bean gum, guar gum, xanthan gum, cellulose gum, methylcellulose, hydroxyethylcellulose, hydroxypropylcellulose, and hydroxypropylmethylcellulose.

### **Methods of Treatment**

[0158] In another preferred embodiment, the present subject matter relates to a method of treating and/or preventing microbial or fungal infections or disorders in a mammal comprising administering to a mammal in need thereof an effective amount of any of the crystalline forms I, II, III, IV, V, or VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl]- quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof. In a particularly preferred embodiment, these methods comprise the use of any one of the pure crystalline Forms I, II, III, IV, V, or VI having a purity of at least 85% by weight, or having not more than 15% by weight of other forms of the compound and any residual solvents, with a purity of at least 90% by weight being most preferred. In an especially preferred embodiment, the present crystalline Forms I, II, III, IV, V, or VI can be used in a method of treating and/or preventing Chagas Disease in a mammal comprising administering to a mammal in need thereof an effective amount of a substantially pure crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H-

1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof. The crystalline Form III, Form IV, or Form VI is particularly preferred in any of these methods in this regard. Form VI is most preferred in this regard. The present subject matter also contemplates use of one or more of the crystalline forms in combination for the methods of treatment described herein.

**[0159]** In another preferred embodiment, the present methods of treatment result in an improvement of the patient's condition, reduction of symptoms, an improvement in the patient's appearance, or combinations thereof. In another preferred embodiment, the present methods of treatment include the use any of the crystalline forms I, II, III, IV, V, or VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one or a pharmaceutically acceptable salt or derivative thereof, for the preparation of a medicament for treating and/or preventing microbial or fungal infections or disorders, or for treating and/or preventing Chagas Disease, in a mammal in need thereof.

**[0160]** In preferred embodiments, the administration of the present compositions is capable of reducing the number of microbes, preferably pathogenic microbes, in or on the mammal to which it is administered. The microbes that can be acted on by the present compositions are selected from the group consisting of fungi, molds, and combinations thereof. The present compositions are further capable of treating infections of the parasite *Trypanosoma cruzi*.

**[0161]** Preferred, non- limiting examples of such fungi are those selected from the group consisting of *P. ovale*, *P. oviculare*, *M. furfur*, *C. neoformans*, *S. prolificans*, *S. shenkii*, *Epidermophyton floccosum*, *Microsporum canis*, *Candida* sp., *Trichophyton* sp., and combinations thereof.

**[0162]** Preferred, non- limiting examples of such *Candida* sp. are those selected from the group consisting of *C. albicans*, *C. krusei*, *C. glabrata*, *C. guilliermondii*, *C. parapsilosis*, *C. tropicalis*, and combinations thereof.

**[0163]** Preferred, non- limiting examples of such *Trichophyton* sp. are those selected from the group consisting of *T. rubrum*, *T. mentagrophytes*, *T. tonsurans*, *T. violaceum*, and combinations thereof.

**[0164]** Preferred, non- limiting examples of such molds are *Aspergillus* sp.

**[0165]** Preferred, non- limiting examples of such *Aspergillus* sp. are those selected from the group consisting of *A. flavus*, *A. fumigatus*, *A. niger*, and combinations thereof.

**[0166]** Several specific skin disorders may also be treated according to the present methods. Exemplary among these skin disorders are seborrheic dermatitis, *Pityrosporum* infections, tinea versicolor, tinea pedis, tinea cruris, tinea corporis, cutaneous candidiasis, onychomycosis, and combinations thereof. Other skin disorders known to those of ordinary skill in the art as effectively treatable by an antimicrobial or antifungal composition are further contemplated as within the scope of the present subject matter.

**[0167]** Further, the present methods provide antimycotic activity against *Pityrosporum* strains, such as *Pityrosporum ovale* and *Pityriasis versicolor*.

**[0168]** In this regard, the present compositions are particularly effective in treating the specific dermatophytes *Tinea* and/or *Candida* fungi.

#### Combination Therapy

**[0169]** In another preferred embodiment, the present preferred compositions may be used in combination with an additional pharmaceutical dosage form to enhance their effectiveness in treating a microbial or fungal infection or disorder. In this regard, the present preferred compositions may be administered as part of a regimen additionally including any other pharmaceutical and/or pharmaceutical dosage form known in the art as effective for the treatment of a microbial and/or fungal infection. Similarly, a pharmaceutically active ingredient other than those specified herein can be added to the present preferred compositions to enhance their effectiveness in treating a microbial and/or fungal infection or disorder. Accordingly, this additional pharmaceutically active ingredient or additional pharmaceutical dosage form can be administered to a patient either directly or indirectly, and concomitantly or sequentially, with the preferred compositions described herein.

**[0170]** In this regard, antimicrobial agents other than the (1 R, 2R)- 7- chloro- 3- [2- (2, 4- di- fluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one discussed above are additionally contemplated as useful for combination therapy discussed herein. Included among these other antimicrobial agents are those selected from the group consisting of imidazoles, allylamines, triazoles, glucan synthase inhibitors, chitin synthase inhibitors, polyenes, griseofulvin, morpholine derivatives, triazines, pyrimidines, any other antimicrobial azole, pharmaceutically acceptable salts or derivatives thereof, and mixtures thereof. Other antimicrobial agents known in the art as effective upon administration to a patient are further contemplated as effective within the present subject matter.

**[0171]** In a preferred embodiment, these other antimicrobial agents are those selected from the group consisting of amorolfine, amphotericin B, bacitracin, benzalkonium chloride, benzethonium chloride, bifonazole, butenafine, butocanazole, chloroxine, cilofungin, chlordanol, chlortetracycline, ciclopirox, clindamycin, clioquinol, clotrimazole, econazole, elubiol, faeriefungin, fezatione, fluconazole, flucytosine, fungimycin, gentamicin, griseofulvin, haloprogin, hexylresorcinol, itraconazole, ketoconazole, methylbenzethonium chloride, miconazole, mupirocin, naftifine, nikkomycin Z, nystatin, l-

ofloxacin, oxiconazole, oxytetracycline, phenol, polymyxin B, pyrido [3, 4- e]- 1, 2, 4- triazine, pyrrolnitrin, quaternary ammonium compounds, salicylic acid, saperconazole, sulconazole, tea tree oil, terbinafine, terconazole, tetracyclines, thiabendazole, ticlatone, tioconazole, tolnaftate, triacetin, triclocarbon, triclosan, undecylenic acid, voriconazole, zinc and sodium pyrithione, a pharmaceutically acceptable salt or derivative thereof, and a mixture thereof. Combinations of any of the foregoing anti- microbial agents or their pharmaceutically acceptable salts or derivatives are contemplated herein.

**[0172]** In one embodiment in this regard, the present preferred compositions and the additional pharmaceutical dosage form can be administered to a patient at the same time. In an alternative embodiment, one of the present preferred compositions and the additional pharmaceutical dosage form can be administered in the morning and the other can be administered in the evening.

**[0173]** In another preferred embodiment, the presently described compounds can be administered to a patient in need thereof in multiple pharmaceutical dosage forms. This combination therapy may maximize the effectiveness of the present composition in treating a microbial or fungal infection or disorder. In one preferred embodiment in this regard, both an oral and a topical composition, each containing Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1-methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one, can be administered, either concomitantly or sequentially, to a patient suffering from a microbial and/or fungal infection or disorder. In the alternative, the oral and topical compositions can contain different amorphous and/or crystalline forms of the (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one compound.

## Methods of Production

**[0174]** Each of the crystalline forms I, II, III, IV, V, and VI of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one can be prepared by a separate process to arrive at a separate crystalline form.

**[0175]** For example, the present subject matter relates to a process for preparing a crystalline Form I or II of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- crystallizing a crystalline Form I or II of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluoro- phenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from an amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one using supercritical CO<sub>2</sub> crystallization conditions.

**[0176]** Similarly, the present subject matter relates to a process for preparing a crystalline Form III of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1 H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising recrystallizing (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from a solution or suspension of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in an organic solvent, particularly an organic solvent selected from the group consisting of ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate. In a most preferred embodiment, the organic solvent is ethanol. In a further preferred embodiment, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one forms a solution in the organic solvent.

**[0177]** Further, the solution or suspension is preferably formed at a temperature of at least 40° C. This solution or suspension can then optionally be cooled to a temperature of about 10° C to about 20° C. Accordingly, the present subject matter relates to a crystalline form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising the reaction product of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and an organic solvent, particularly an organic solvent selected from the group consisting of ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate, with ethanol being most preferred. One of skill in the art would understand that combinations of two or more organic solvents, with or without water, may also be useful for the purpose of producing a crystalline form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In a preferred embodiment, combinations of two or more organic solvents, without water, may also be useful for the purpose of producing a crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in this regard.

**[0178]** In a further preferred embodiment in this regard, the present subject matter relates to a process for preparing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to a solvent selected from the group consisting of ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate to form a solution or suspension; and
- Crystallizing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension.

**[0179]** In a particularly preferred embodiment in this regard, the (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one added to the solvent is the amorphous form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In a further particularly preferred embodiment, the solvent is ethanol and is used to form a solution of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0180]** In still another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Dissolving (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in a solvent selected from the group consisting of ethanol, ethyl acetate, dichloromethane, and a combination of ethanol and ethyl acetate to form a solution;
- Crystallizing crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Drying said crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0181]** In a preferred embodiment, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one added to the solvent is the amorphous form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. Further in this regard, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in the solvent is preferably first filtered and then heated to about 65° C to about 75° C in a vacuum to reduce the solvent content. In a preferred embodiment, the (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in the solvent is heated to about 70° C. And again, the most preferred solvent used in this regard is ethanol.

**[0182]** In an alternative embodiment, this process further comprises the step of refluxing the solution before crystallizing Form III. In a further alternative embodiment, this process comprises the additional step of filtering the solution after it is refluxed.

**[0183]** In a further preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluoro- phenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to an ethanol solvent to form a solution or suspension;
- Crystallizing crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Drying said crystalline Form III of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0184]** Likewise, the present subject matter relates to a process for preparing a crystalline Form IV of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1 H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Storing an amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one for 3 months at about 40°C and 75% HR;
- Transforming some of said amorphous (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to a crystalline Form IV of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one during storage; and
- Obtaining said crystalline Form IV of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0185]** In this regard, this process results in a mixture of both the amorphous form and Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0186]** In another preferred embodiment in this regard, the present subject matter relates to an alternative process for preparing a crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Dissolving (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in ethanol to form a solution;
- Adding the solution to water to form a suspension;
- stirring the suspension for more than 30 minutes;
- Obtaining crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Separating said crystalline Form IV of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluoro- phenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0187]** In a particularly preferred embodiment in this regard, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one dissolved in the solvent is the amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for this purpose.

**[0188]** In this regard, the amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is preferably dissolved in hot ethanol, i.e. ethanol at a temperature of at least 40° C. This solution can then preferably be added to the water at about 5° C to about 10° C. In an alternative embodiment, the solution can be added to the water at room temperature. Further, the suspension is preferably stirred for at least 6 hours. In one preferred embodiment in this regard, the suspension is stirred for at least 18 hours. In another preferred embodiment in this regard, the suspension is stirred for about 4- 5 hours at a temperature of about 28° C .

**[0189]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Dissolving (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in ethyl acetate to form a solution;
- Adding hexane to the solution;
- Optionally adding diethyl ether to the solution;
- Crystallizing Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one; and
- Separating said crystalline Form V of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one.

**[0190]** In a particularly preferred embodiment in this regard, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one dissolved in the solvent is the amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for this purpose.

**[0191]** In this regard, the amorphous form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one is preferably dissolved in hot ethyl acetate, i.e. ethyl acetate at a temperature of at least 40° C. Further, the solution can preferably be cooled after addition of the hexane.

**[0192]** The present subject matter also relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising recrystallizing (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from a solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in an organic solvent and water. In one embodiment, the organic solvent is a polar solvent, in a preferred embodiment, the organic solvent is an alcohol. In a particular embodiment, the organic solvent is selected from the group consisting of ethanol, methanol, isopropanol, n- propanol, and acetone. In a most preferred embodiment, the organic solvent is ethanol. One of skill in the art would understand that combinations of two or more organic solvents with water may also be useful for the purpose of producing the crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1-

yl) propyl] quinazolin- 4 (3H)- one. In another preferred embodiment, ethanol is present with water at an ethanol: water ratio ranging from 1: 1 to 1: 20 by volume. In a further preferred embodiment, ethanol is present with water at an ethanol: water ratio ranging from 1: 4 to 1: 10 by volume. In this regard, the use of ethanol in the present processes for making Form VI can optionally be replaced with any of these specified organic solvents. In a further preferred embodiment, the

(1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one forms a solution in the organic solvent.

**[0193]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluoro- phenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Forming a solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in an aqueous solvent and an organic solvent selected from the group consisting of ethanol, methanol, isopropanol, n- propanol, and acetone
- Crystallizing Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In a preferred embodiment, the organic solvent is ethanol.

**[0194]** A wide range of ethanol:water ratios may be useful for preparing Form VI solids, including, for example, 1:1, 1:2, 1:3, 1:4, 1:5, 1:6, 1:7, 1:8, 1:9, 1:10, 1:11, 1:12, 1:13, 1:14, 1:15, 1:16, 1:17, 1:18, 1:19 and 1:20, by volume. In a preferred embodiment, the ratio of ethanol to water ranges from 1:1 to 1:20, by volume. In another preferred embodiment, the ratio of ethanol to water ranges from 1:4 to 1:10, by volume. In further embodiments, these ethanol:water ratios would also apply to the other organic solvents discussed herein.

**[0195]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to aqueous ethanol to form a solution or suspension;
- Crystallizing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In a preferred embodiment of the above method, the solution or suspension is formed at a temperature of about 25°C to about 65°C. In a particularly preferred embodiment, the solution or suspension is formed at a temperature of about 45°C. In a preferred embodiment of the above method, the ratio of ethanol to water ranges from 1: 1 to 1: 20, by volume. In another preferred embodiment, the ratio of ethanol to water ranges from 1: 4 to 1: 8, by volume.

**[0196]** In an alternative preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising:

- Adding (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one to ethanol to form a mixture;
- Adding said mixture to water to form a solution or suspension;
- Crystallizing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophe- nyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from said solution or suspension; and
- Separating said crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluoro- phenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. In a particularly preferred embodiment, the mixture of albaconazole and ethanol is a solution of albaconazole in ethanol. In a preferred embodiment of the above method, the ratio of ethanol to water ranges from 1: 1 to 1: 20, by volume. In another preferred embodiment, the ratio of ethanol to water ranges from 1: 4 to 1: 8, by volume.

**[0197]** In a preferred embodiment of the above method, the process whereby albaconazole is first added to ethanol to form a mixture, which mixture is then added to water, this process contemplates the additional step of refluxing the solution or suspension of albaconazole in ethanol, before the mixture is added to water. In an alternative preferred embodiment, this process contemplates the additional step of filtering the solution of albaconazole in ethanol, before the mixture is added to water.

**[0198]** In other preferred embodiments in this regard, the slurry of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be of Form III, Form IV, Form V, amorphous form, or combinations thereof in aqueous ethanol. However, any form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be suitable for this purpose.

**[0199]** In another preferred embodiment, the present subject matter relates to a process for preparing a crystalline Form VI of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one comprising recrystallizing (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one from a solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one in ethanol and water. In a preferred embodiment, the ethanol mixture is a solution. In a further embodiment, the process further comprises refluxing the solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and ethanol before adding the mixture to water. In another further preferred embodiment, the process further comprises filtering the solution or suspension of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one and ethanol before adding the mixture to water.

**[0200]** Form VI may be prepared by slurrying albaconazole in aqueous ethanol or by slowly cooling a saturated aqueous ethanol solution that was seeded. In order to estimate rate of conversion, a slurry experiment of Form III was setup at 45°C in ethanol-water (1:9), seeded with Form VI and aliquots periodically removed. Solids from the first aliquot removed after ~ 3 hours were composed of only Form VI by XRPD analysis. A slow cool method was also attempted by cooling a saturated solution of albaconazole in aqueous ethanol that was seeded. Form VI solids were isolated in a 41 % yield.

**[0201]** It is further noted that the process for making Form VI is similar to the process for making Form IV. Embodiments of both processes may use the same or similar ratios of ethanol: water and similar mixing and stirring methods. In this regard, it is contemplated herein that, in one process of making Form VI, the (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one first proceeds through the crystalline Form IV before continuing on to Form VI when starting with the same aqueous solvent- (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one mixture. In one embodiment, Form IV is an intermediate in the production of Form VI, and the rate of the transformation is a function of the ratio of solvents, temperature, stirring, and time. In general, shorter stirring times and lower stirring temperatures favor formation of Form IV, while longer stir times and higher stirring temperatures favor formation of Form VI. For example, in this regard, Form VI can be obtained in aqueous ethanol at 22°C after about 6 hours of stirring; and at 28°C after about 5 hours of stirring. One can also allow the process to proceed longer, for example, 18- 24 hr, and still obtain Form VI. Selection of solvent may also affect the time it takes for Form IV to convert to Form VI. For example, in some embodiments, Form VI may form in aqueous ethanol or aqueous acetone in 3- 7 hours; or Form VI may form in aqueous methanol or aqueous isopropanol or aqueous n- propanol only after 7- 18 hours. In a preferred embodiment, a minimum stirring time of 5 hr is required to prepare Form VI. The presence of Form IV or Form VI can be monitored by an appropriate technique, such as, for example, microscopy or x- ray diffraction, in order to determine the best time/ conditions for its isolation.

**[0202]** Amorphous material of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one may be produced by a melt- quench technique and exhibited a glass transition at 75°C. Interconversion of the amorphous form in slurry experiments showed that the predominant resulting crystalline forms were Forms III, IV, V, and VI.

## Dosage

**[0203]** Appropriate dosage levels for any of the crystalline forms of any of the active ingredients and/or their specific crystal forms are well known to those of ordinary skill in the art and are selected to maximize the treatment of the previously described microbial and/or fungal conditions. Dosage levels on the order of about 0.001 mg to about 5,000 mg per kilogram body weight of the active ingredient components are known to be useful in the treatment of the diseases, disorders, and conditions contemplated herein. Typically, this effective amount of the active agent will generally comprise from about 0.001 mg to about 100 mg per kilogram of patient body weight per day. Moreover, it will be understood that this dosage of ingredients can be administered in a single or multiple dosage units to provide the desired therapeutic effect.

**[0204]** If desired, other therapeutic agents can be employed in conjunction with those provided in the above-described compositions. The amount of pharmaceutically active ingredients that may be combined with the carrier materials to produce a single dosage form will vary depending upon the host treated, the nature of the disease, disorder, or condition, and the nature of the active ingredients.

**[0205]** The preferred pharmaceutical compositions may be given in a single or multiple doses daily. In a preferred embodiment, the pharmaceutical compositions are given from one to three times daily. Starting with a low dose twice

daily and slowly working up to higher doses if needed is a preferred strategy. The amount of pharmaceutically active ingredients that may be combined with the carrier materials to produce a single dosage form will vary depending upon the host treated, the nature of the disease, disorder, or condition, and the nature of the active ingredients.

**[0206]** It is understood, however, that a specific dose level for any particular patient will vary depending upon a variety of factors, including the activity of the specific pharmaceutically active agent; the age, body weight, general health, sex and diet of the patient; the time of administration; the rate of excretion; possible drug combinations; the severity of the particular condition being treated; and the form of administration. One of ordinary skill in the art would appreciate the variability of such factors and would be able to establish specific dose levels using no more than routine experimentation.

**[0207]** The optimal pharmaceutical formulations will be determined by one skilled in the art depending upon considerations such as the particular pharmaceutically active agent combination and the desired dosage. See, for example, "Remington's Pharmaceutical Sciences", 18th ed. (1990, Mack Publishing Co., Easton, PA 18042), pp. 1435-1712, the disclosure of which is hereby incorporated by reference. Such formulations may influence the physical state, stability, rate of in vivo release, and rate of in vivo clearance of the essential lipids.

## EXAMPLES

**[0208]** The presently presented compounds can be prepared in accordance with the following Examples using commercially available starting materials. The (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one used in the foregoing examples may be obtained by any of the currently available processes that are well known in the art, such as, for example, the processes described in Bartoli et al., J. Med. Chem., Vol. 41, No. 11, pp. 1869- 1882 (1998), the contents of which are hereby incorporated by reference in their entirety.

**[0209]** In solution, no crystalline form exists, and thus the physicochemical solution characteristics, i.e. <sup>1</sup>H NMR spectra, ultraviolet spectra, and specific rotations of the crystalline and amorphous forms of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one are the same.

### EXAMPLE 1

Preparation of crystalline Form I of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0210]** 1.5 g of amorphous (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was dispersed in a column filled with glass beads which act as a disaggregating agent. The column was inserted in an extractor and a flux of supercritical CO<sub>2</sub> (SCCO<sub>2</sub>) was passed (p=250 bar, T=450C) . A filter was installed at the extractor exit to collect the material once dissolved by the SCCO<sub>2</sub>. Only 0.15 g of material remained in the column. This material was crystalline, and was shown to have crystalline Form I.

### EXAMPLE 2

Preparation of crystalline Form II of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0211]** 0.5 g of amorphous (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was dissolved in 200 ml of ethanol. The solution was put in a reactor which was pressurized with CO<sub>2</sub> up to p=100 bar and T=40°C (supercritical conditions) . The reactor was depressurized, and any significant amount of product was collected in the outer filter. Once the ethanol was evacuated, 0.52 g of resulting material was collected. The X- ray diffractogram of this sample showed a crystalline pattern. This material was shown to have crystalline Form II.

### EXAMPLE 3

**[0212]** The procedure of Example 1 was repeated but substituting the glass beads with silica sand. The supercritical conditions were p=250 bar and T=52°C. In this case, sample could be collected in the outer filter, which was of crystalline Form II, as well as inside the reactor, which was of crystalline Form I.



## EXAMPLE 4

Preparation of crystalline Form III of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0213]** Amorphous (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was dissolved in ethanol and recrystallized. The original <sup>1</sup>H NMR spectrum showed about ½ mol of ethanol. The sample was kept in a capped vial at room temperature for 7 years. After 7 years, the samples NMR spectrum showed a total loss of ethanol. The sample was then shown to have crystalline Form III.

## EXAMPLE 5

**[0214]** The procedure of Example 4 was repeated but substituting the ethanol with ethanol and ethyl acetate, ethyl acetate, or dichloromethane. In each case, the recrystallized sample was shown to have crystalline Form III.

## EXAMPLE 6

Preparation of crystalline Form III of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0215]** 6.785 kg of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was dissolved in 17.3 L of ethanol. The solution was cooled at 10- 20°C for about 2 hours. The obtained product was centrifuged and dried in vacuum at 70°C to obtain 5.796 kg of crystalline (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. The product was then shown to have crystalline Form III.

## EXAMPLE 7

Preparation of crystalline Form III of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0216]** (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one was dissolved in an excess of hot ethanol. Charcoal was added and the solution was filtered. The filtered solution was then concentrated in vacuum to a final volume of 2.5- 3 L/kg of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. The concentrated solution was then cooled to 5- 10°C and (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one crystallized. The obtained product was filtered and dried in vacuum at 70°C to obtain crystalline (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one. The product was then shown to have crystalline Form III.

## EXAMPLE 8

Preparation of crystalline Form IV of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0217]** 8.10 g of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were dissolved in 65 mL of hot ethanol. This solution was added to 400 mL of water at about 5 to about 10° C. The suspension obtained was stirred for 18 hours. 7.07 g of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were obtained. The product was then shown to have crystalline Form IV.

## EXAMPLE 9

Preparation of crystalline Form IV of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one.

**[0218]** 18.15 g of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were dissolved in 130 mL of hot ethanol. This solution was added to 800 mL of water at

room temperature. The suspension obtained was stirred for 25 hours at room temperature. The product was filtered and dried in vacuum at 50°C for 18 hours. 16.98 g of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were obtained. The product was then shown to have crystalline Form IV.

#### EXAMPLE 10

Preparation of crystalline Form IV of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one.

**[0219]** 15.0 g of Form III (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1 H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were dissolved in 90 mL of EtOH 96° at 70°C and this solution was filtered and added to 810 mL of water contained in a reactor equipped with a heating jacket connected to a thermostatic bath, reflux cooler, discharge valve at the bottom and mechanical stirrer. The suspension was stirred at 160 rpm at 28°C for 4h 30 min. The product was filtered (wet weight: 73, 34 g) and dried in vacuum at 50°C. 13.59 g of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were obtained. The product was then shown to have crystalline Form IV by using XRPD and DSC.

#### EXAMPLE 11

Preparation of crystalline Form V of (1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one.

**[0220]** 10.0 g of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were dissolved in 30 mL of hot ethyl acetate. 10 mL of Hexane were then added and the solution was slightly cooled. Crystallization was immediately observed. 20 mL of diethyl ether were then added, and the suspension was stirred on an ice bath for one hour. The obtained product was filtered and dried in vacuum at about 40° C for 8 hours. The product was then dried at 60° C in vacuum for 24 hours. The product was then shown to have crystalline Form V.

#### EXAMPLE 12

Preparation of crystalline Form VI of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0221]** 100 mL of ethanol and water in a ratio of 1: 1 by volume were heated to 45° C. When the temperature stabilized, 5.0 g of Form III (1 R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were added. The suspension was stirred, with a magnetic stirrer, for 18 hours at 45° C. The suspension was filtered, at 45° C, with a sintered disc filter funnel, keeping the vacuum for 20 minutes. 3.38 g of wet product was obtained. The product was dried in a vacuum oven at room temperature for 24 hours. The product was then shown to have crystalline Form VI.

#### EXAMPLE 13

Preparation of crystalline Form VI of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0222]** 30.0 g of Form III (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one were dissolved in 210 mL of absolute ethanol at 65° C. This solution, at 70° C, was added to 1270 mL of water at room temperature. The suspension was stirred at room temperature for 8 hours. The product was dried in vacuum at 50° C. The product was then shown to have crystalline Form VI.

#### EXAMPLE 14

Preparation of crystalline Form VI of (1 R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one

**[0223]** (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl]

## EP 2 650 291 A1

quinazolin- 4 (3H)- one (254.4mg) was added to a glass vial, followed by ethanol (2.04mL) and warmed to ~ 70°C to aid dissolution. Solids dissolved and the resulting solution was filtered through 0.2µm nylon filter into a clean vial containing water (12.24mL) . A precipitate formed and approximately half of the volume of the suspension was removed. The remaining suspension was agitated at ~60°C. After ~ 6 hours, an aliquot was removed and solids collected by filtration. The product was then shown to have crystalline Form VI.

### EXAMPLE 15

Preparation of crystalline Form VI of Albaconazole ((1R,2R)-7-chloro-3-[2-(2,4-difluorophenyl)-2-hydroxy-1-methyl-3-(1H-1,2,4-triazol-1-yl)propyl]quinazolin-4(3H)-one)

**[0224]** 150 mL of EtOH-water (10:90) was heated to 45°C. When the temperature stabilized, 20.0 g of Albaconazole Form III was added. The suspension was stirred, with a magnetic stirrer, for 18 hours at 45°C. The suspension was cooled to 35°C in about 20 minutes and then to 5-10°C with an ice bath for one hour. 21.53 g of wet product was obtained. The product was dried in a vacuum oven at 50°C for 24 h. The product was then shown to have crystalline form VI.

### EXAMPLE 16

**[0225]** Sufficient (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one Form III was added to (9: 1) water- ethanol mixture (3mL) such that excess solids remained. The mixture was agitated at ambient temperature overnight and filtered through a 0.2µm nylon filter into a clean vial. Approximately equal amounts of albaconazole Forms III, IV and V were added and the resulting mixture slurried overnight. An aliquot was then removed and solids isolated by filtration. The product was then shown to have crystalline Form VI.

### EXAMPLE 17

**[0226]** 10.0 g of Albaconazole were dissolved in 60 mL of MeOH at about reflux temperature and this solution was filtered and added to 540 mL of water contained in a reactor equipped with a heating jacket connected to a thermostatic bath, reflux cooler, discharge valve at the bottom and mechanical stirrer. The suspension was stirred at 160 rpm at 28°C. Samples were taken to follow the process. After 23 hours the product was filtered and dried in vacuum at 50°C. The product was identified as Form VI by its x-ray diffractogram and DSC.

### EXAMPLE 18

**[0227]** 10.0 g of Albaconazole were dissolved in 60 mL of iso-propanol at about reflux temperature and this solution was filtered and added to 540 mL of water contained in a reactor equipped with a heating jacket connected to a thermostatic bath, reflux cooler, discharge valve at the bottom and mechanical stirrer. The suspension was stirred at 160 rpm at 28°C. Samples were taken follow the process. After 8 hours the product was filtered and dried in vacuum at 50°C. The product was identified as Form VI by its x-ray diffractogram and DSC.

### EXAMPLE 19

**[0228]** 10.0 g of Albaconazole were dissolved in 60 mL of acetone at about reflux temperature and this solution was filtered and added to 540 mL of water contained in a reactor equipped with a heating jacket connected to a thermostatic bath, reflux cooler, discharge valve at the bottom and mechanical stirrer. 10 mL of acetone were used for washing the addition funnel and 90 mL of additional water were added. The suspension was stirred at 160 rpm at 28°C. After 1 hour it was observed that the product was agglomerated and the stirring was increased to 200 rpm and the temperature to 30°C. After 6 hours the product was filtered and dried in vacuum at 50°C. The product was identified as Form VI by its x-ray diffractogram and DSC.

### EXAMPLE 20

**[0229]** 10.0 g of Albaconazole were dissolved in 60 mL of n-propanol at about reflux temperature and this solution was filtered and added to 540 mL of water contained in a reactor equipped with a heating jacket connected to a thermostatic bath, reflux cooler, discharge valve at the bottom and mechanical stirrer. The suspension was stirred at 160 rpm at 28°C. After 1 hour it was observed that the product was agglomerated and the stirring was increased to 200 rpm and the temperature to 35°C. The suspension was stirred for 17 hours 30 min. The product was filtered and dried in vacuum at 50°C. The product was identified as Form VI by its x-ray diffractogram and DSC.

## EXAMPLE 21

[0230] A patient is suffering from onychomycosis. A preferred composition herein is administered to the patient. It would be expected that the patient would improve his/her condition or recover.

## EXAMPLE 22

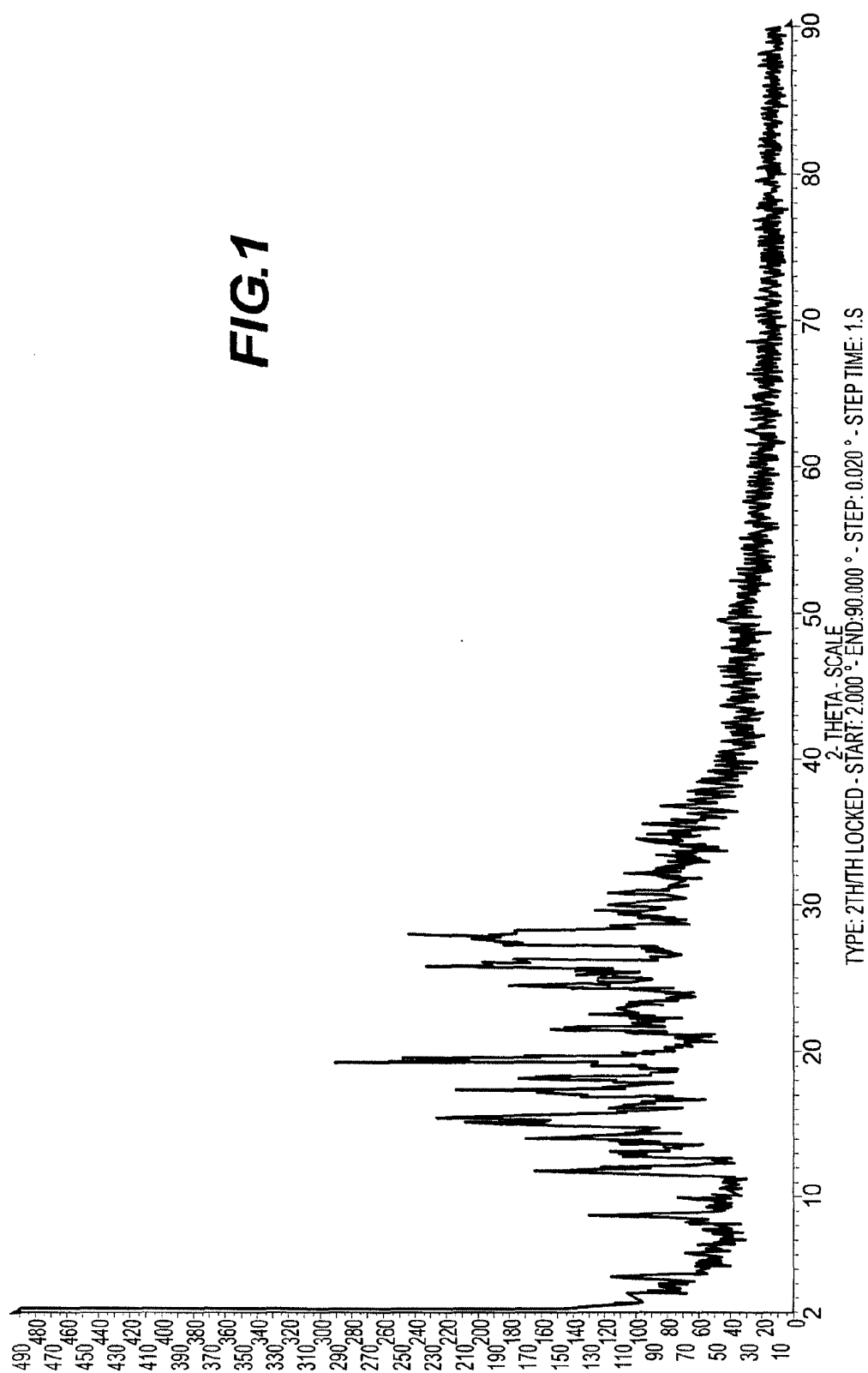
[0231] A patient is suffering from Chagas Disease. A preferred composition herein is administered to the patient. It would be expected that the patient would improve his/her condition or recover.

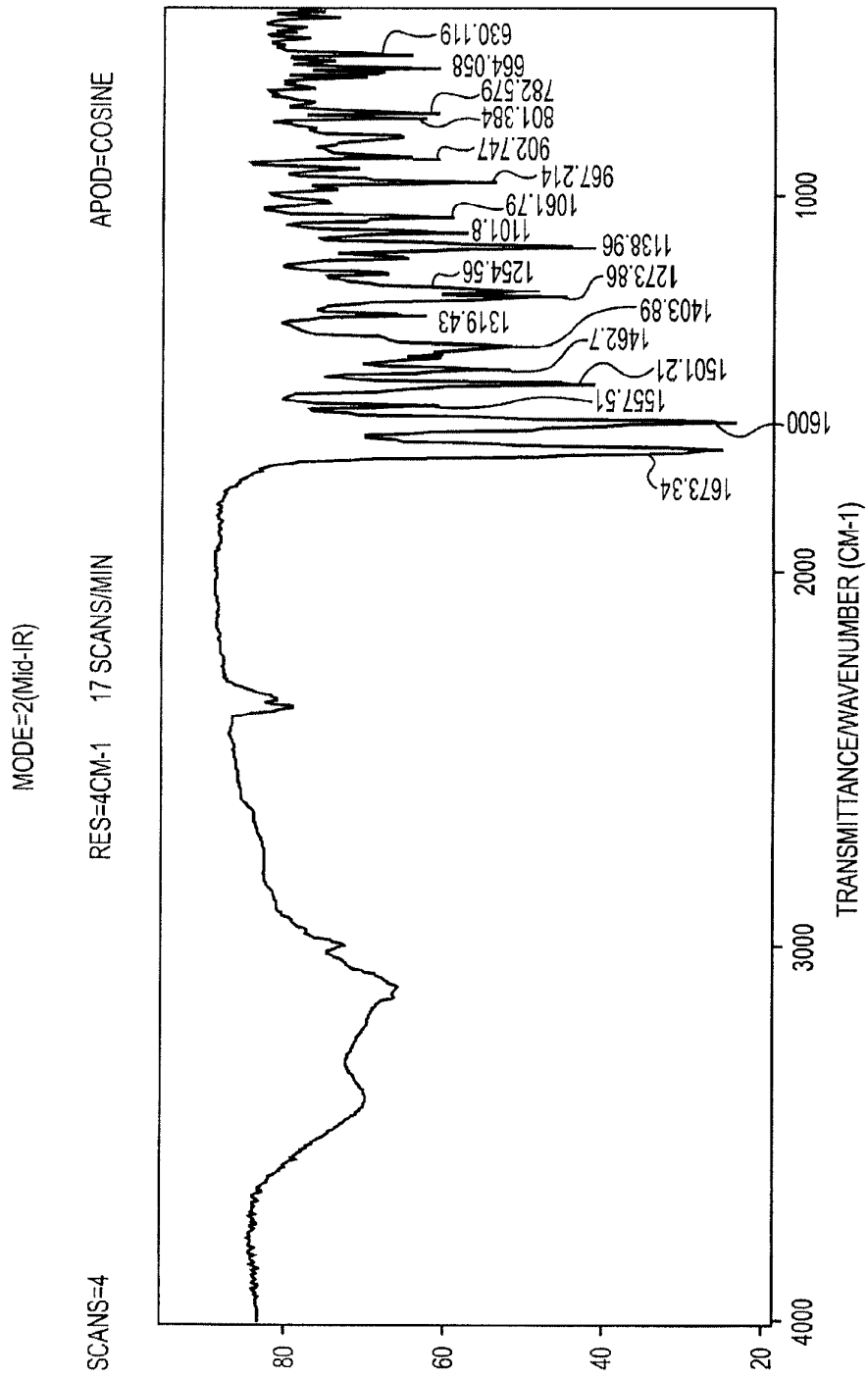
[0232] The present subject matter being thus described, it will be apparent that the same may be modified or varied in many ways. Such modifications and variations are not to be regarded as a departure from the spirit and scope of the present subject matter, and all such modifications and variations are intended to be included within the scope of the following claims.

## Claims

1. A crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one that is Form III.
2. The crystalline form according to claim 1, wherein Form III has a characteristic X-ray powder diffraction (XRPD) pattern comprising 2-theta positions at about 4.08, 5.73, 6.22, 7.77, 8.15, 8.80, 11.25, 11.47, 12.44, 13.09, 15.57, 17.63, 18.66, 20.85, 26.65, and 27.12 +/- 0.2.
3. The crystalline form according to claim 1 or claim 2, wherein Form III has a characteristic X-ray powder diffraction (XRPD) pattern comprising 2-theta positions at about 4.08, 5.73, 6.22, 7.77, 8.15, 8.80, 11.25, 11.47, 12.44, 13.09, 14.33, 14.68, 14.89, 15.57, 16.35, 16.68, 17.27, 17.63, 18.66, 19.32, 20.85, 22.12, 22.49, 23.58, 24.63, 25.02, 26.65, 27.12, 28.74, 29.11, 29.81, 31.35, and 33.48 +/- 0.2;
4. The crystalline form according to one of claims 1 to 3, wherein Form III has an infrared spectra pattern comprising spectral lines at 1677.0, 1600.0, 1557.5, 1498.3, 1462.6, 1403.0, 1318.4, 1272.5, 1254.1, 1170.0, 1138.7, 1101.6, 1060.2, 1016.4, 966.7, 932.7, 902.4, 855.5, 801.5, 785.8, 694.0, 677.9, 665.4, 631.7, 532.7, and 411.6 cm<sup>-1</sup>.
5. The crystalline according to one of claims 1 to 4, wherein Form III has a differential scanning calorimetry thermogram exhibiting an endotherm at 105°C and having an onset temperature at 99°C ± 5°C.
6. The crystalline form of any one of claims 1- 5, which has less than about 10% by weight of a different crystalline form or an amorphous form of (1 R, 2R)- 7- chloro- 3- [2- (2, 4- di- fluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one as determined on a % weight basis.
7. The crystalline form of any one of claims 1-6, wherein the crystalline form has at least 90% purity as defined by X-ray powder diffraction.
8. The crystalline form of any one of claims 1-7, which has less than about 10% by weight of any residual solvent.
9. A pharmaceutical composition comprising an effective amount of the crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1 H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one of any one of claims 1- 8 and a pharmaceutically acceptable carrier.
10. A crystalline form of (1R, 2R)- 7- chloro- 3- [2- (2, 4- difluorophenyl)- 2- hydroxy- 1- methyl- 3- (1H- 1, 2, 4- triazol- 1- yl) propyl] quinazolin- 4 (3H)- one of any one of claims 1- 8, for use in treating and/or preventing a microbial or fungal infection, and/or Chagas disease, in a mammal.

**FIG.1**





**FIG.2**

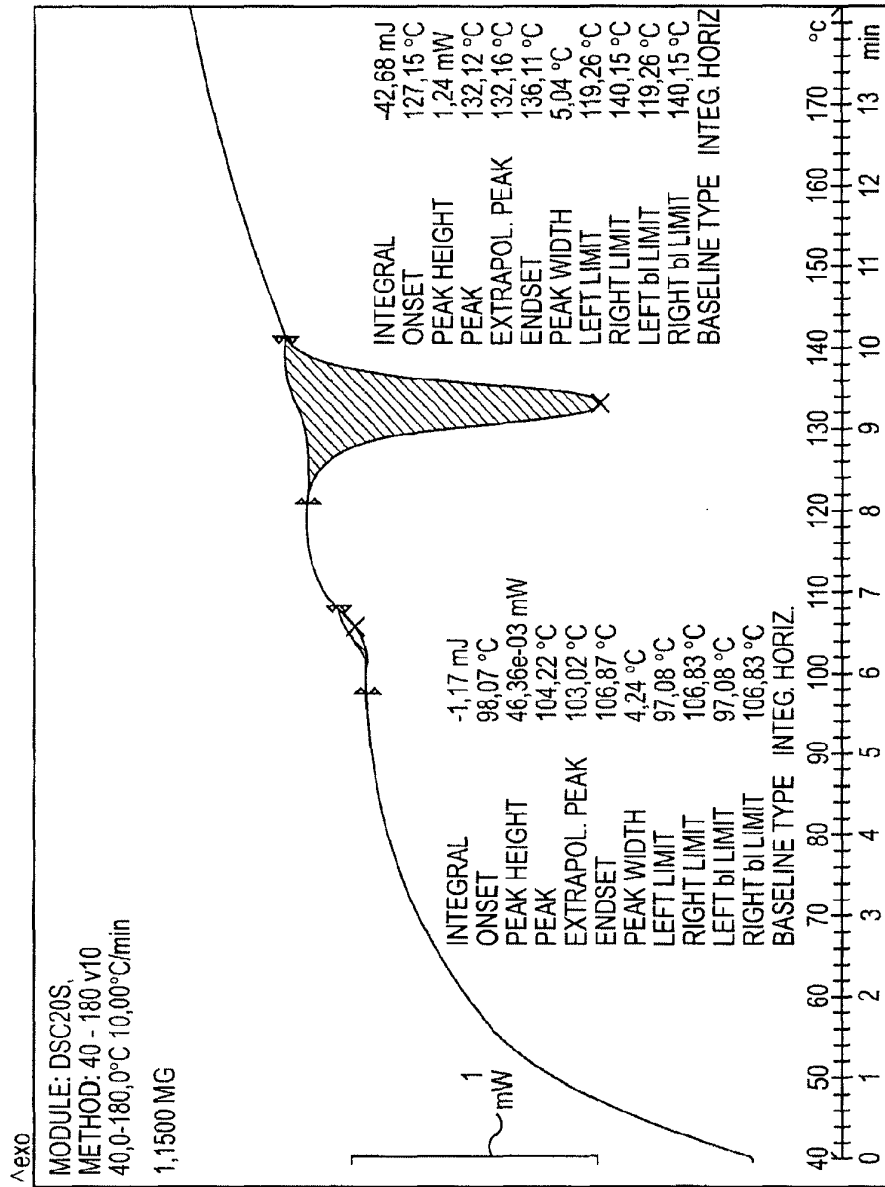
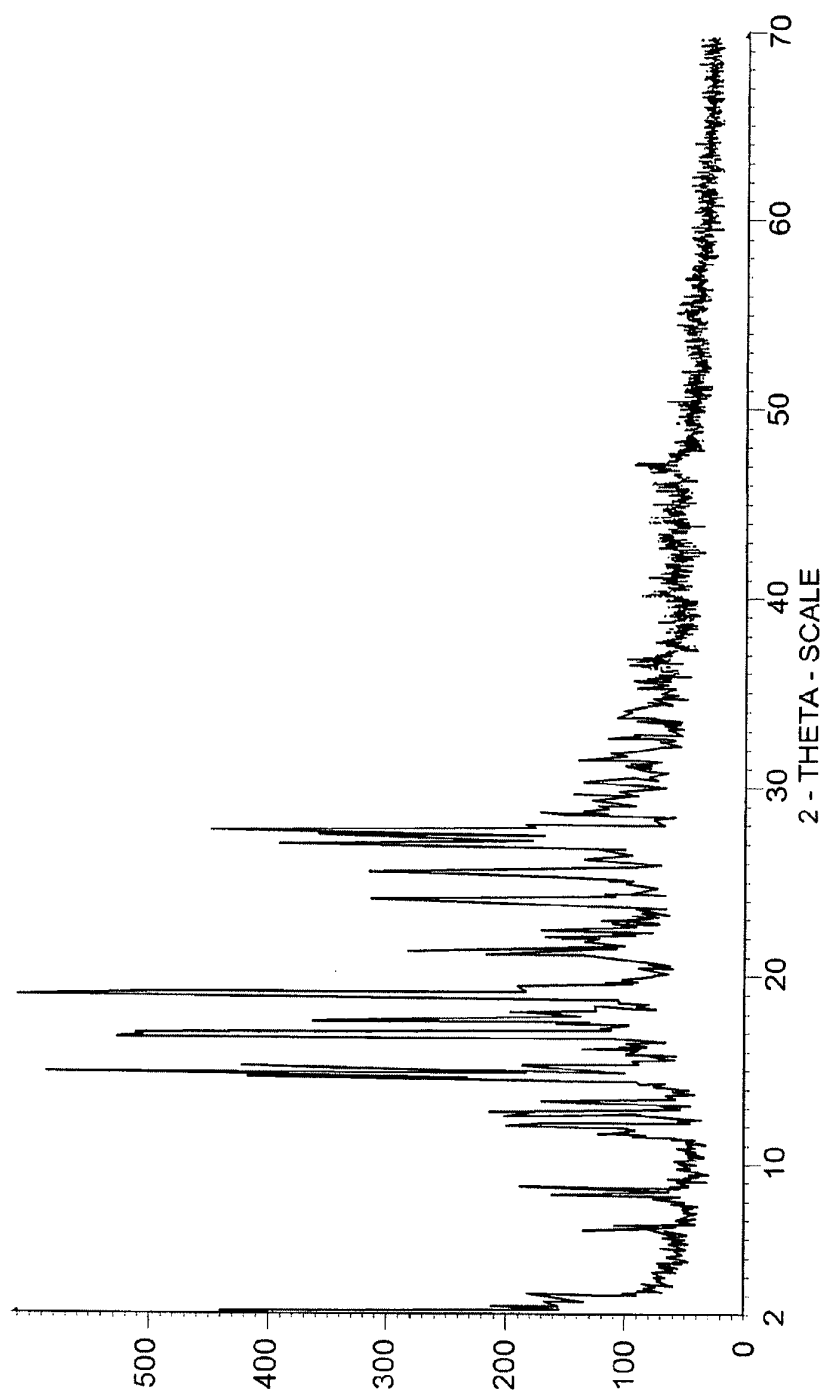


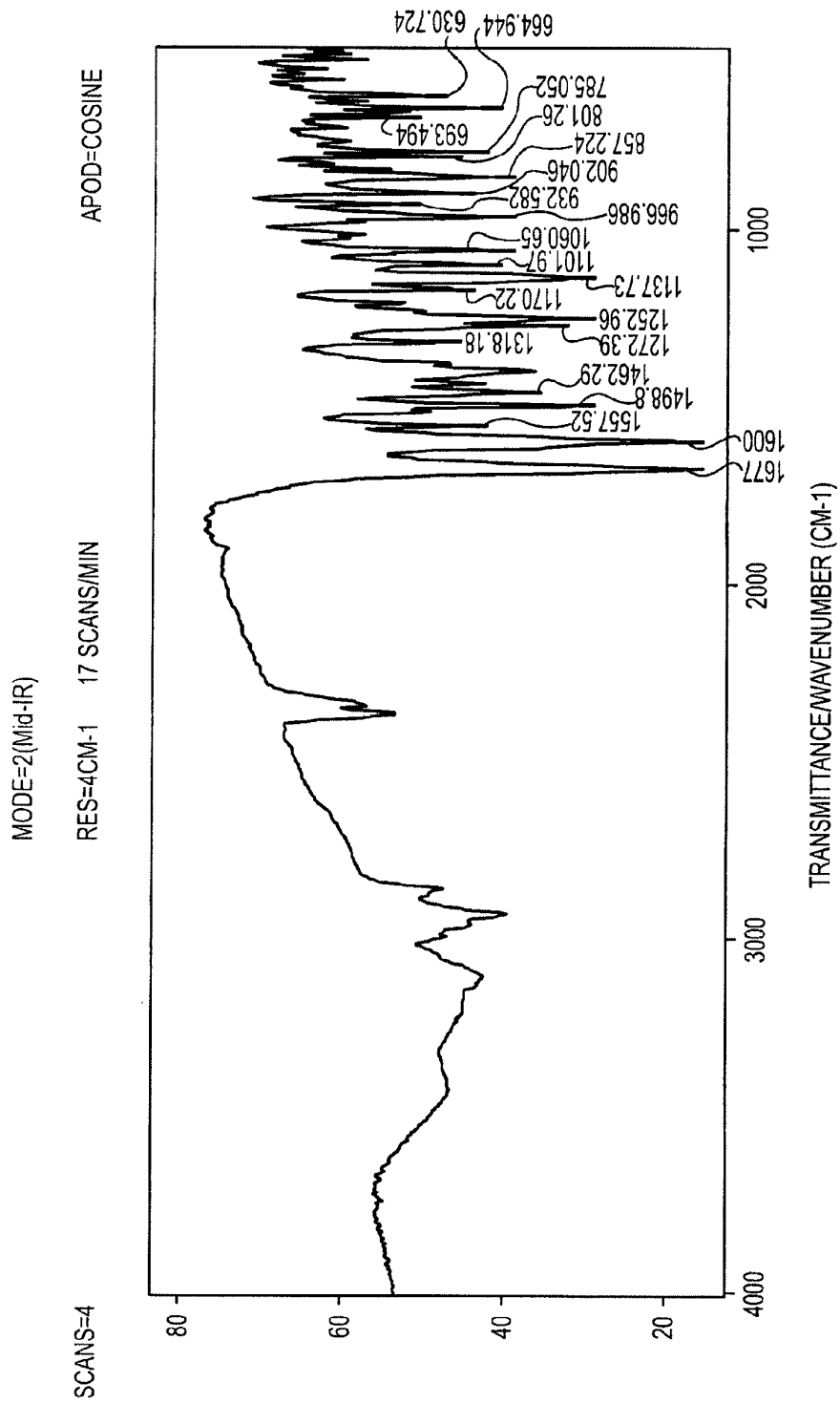
FIG.3



TYPE: 2TH/TH LOCKED - START: 2.00 ° - END: 70.000 ° - STEP: 0.20 ° - STEP TIME: 1.5

**FIG.4**





**FIG.5**

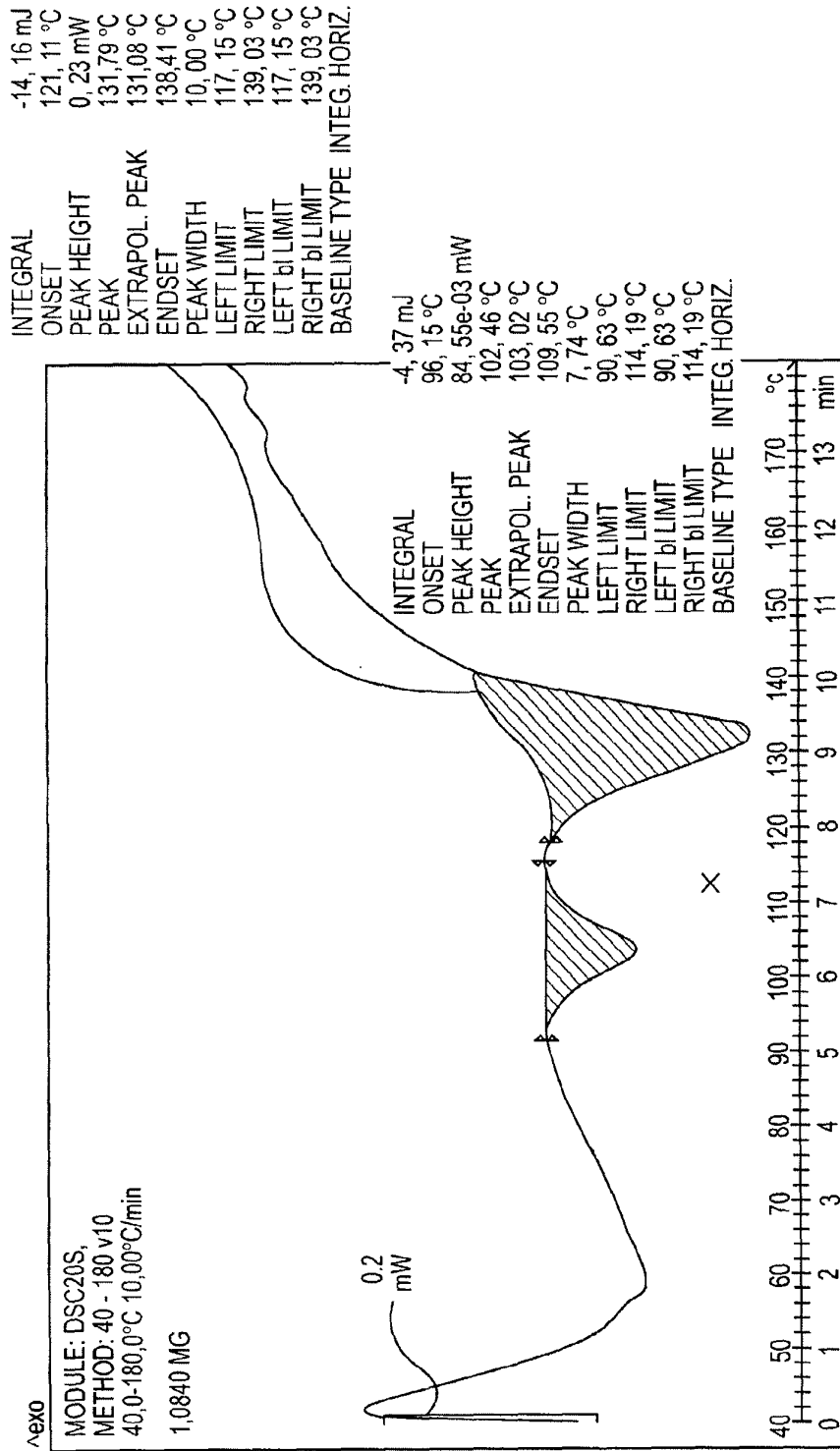
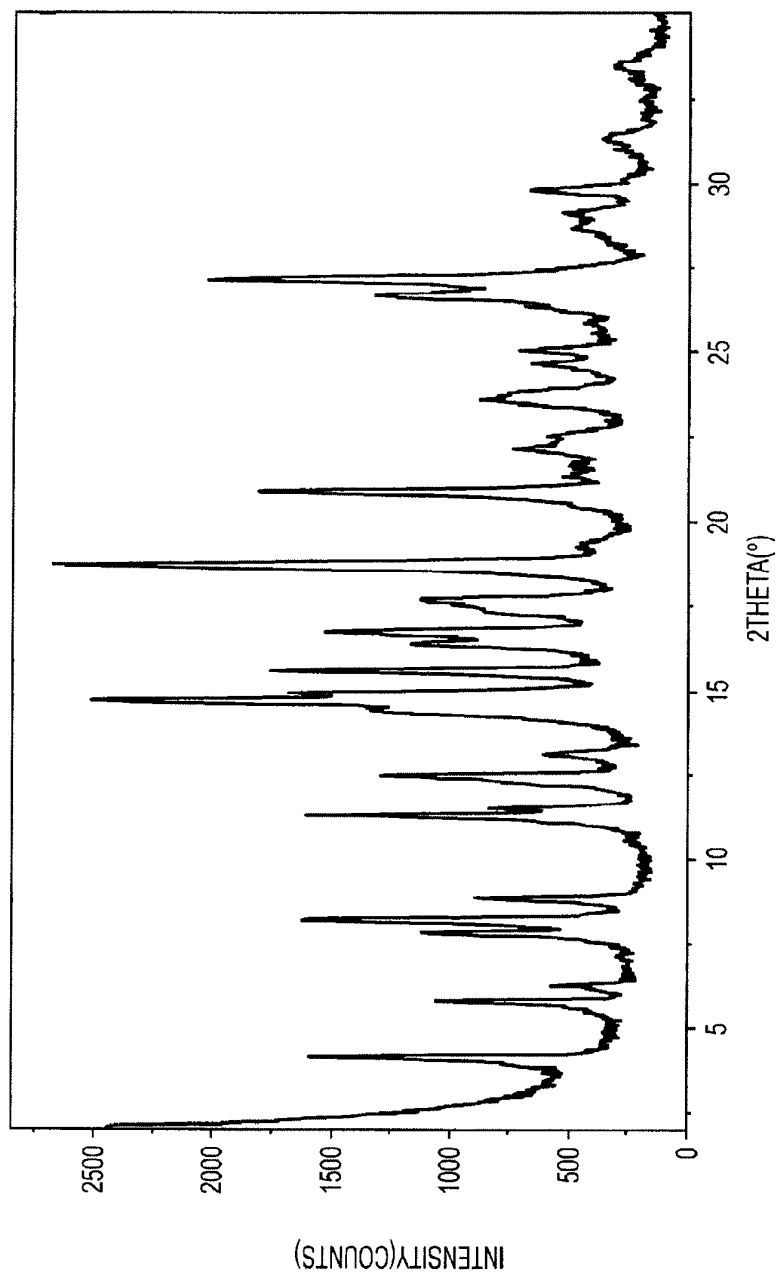


FIG.6



**FIG.7**

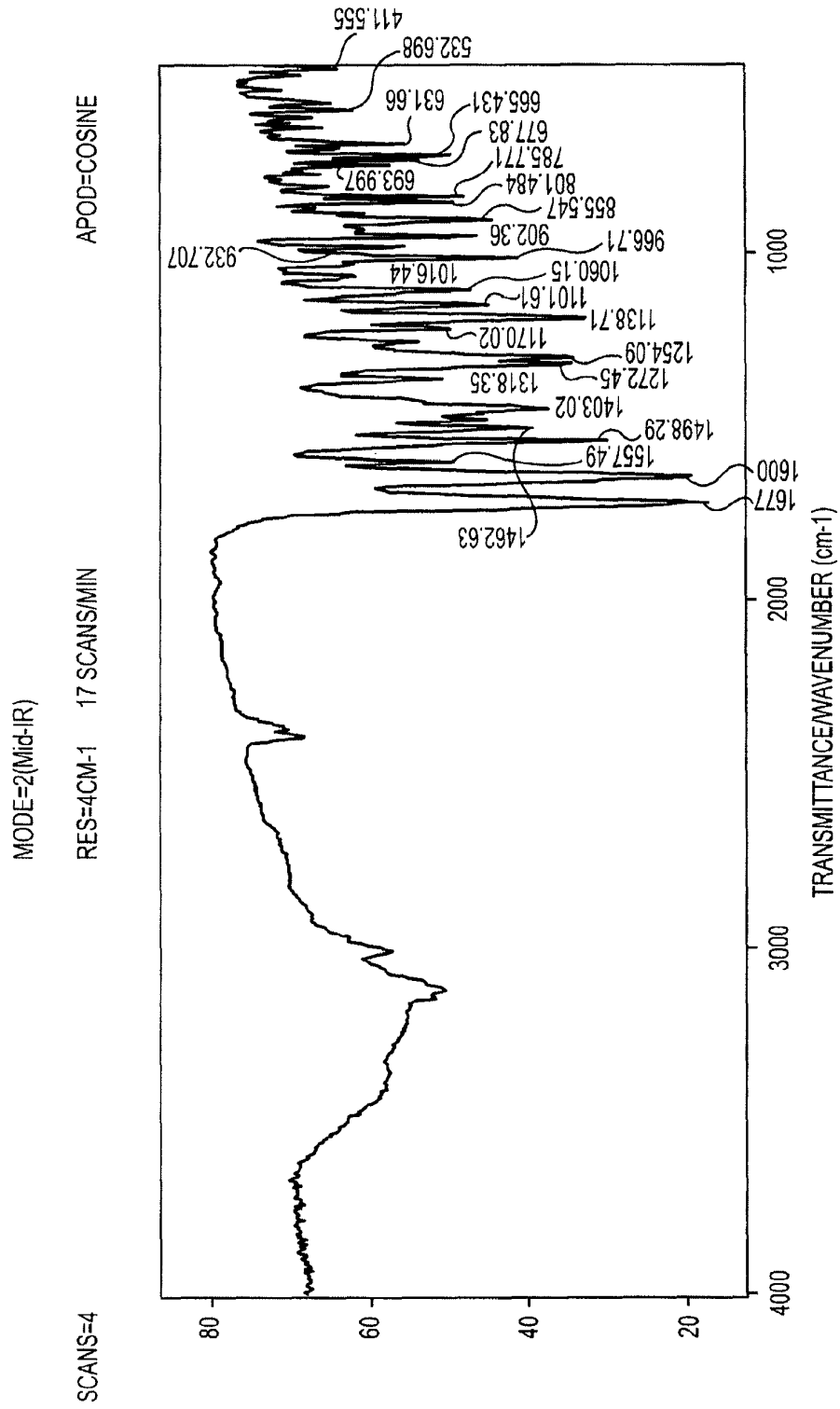
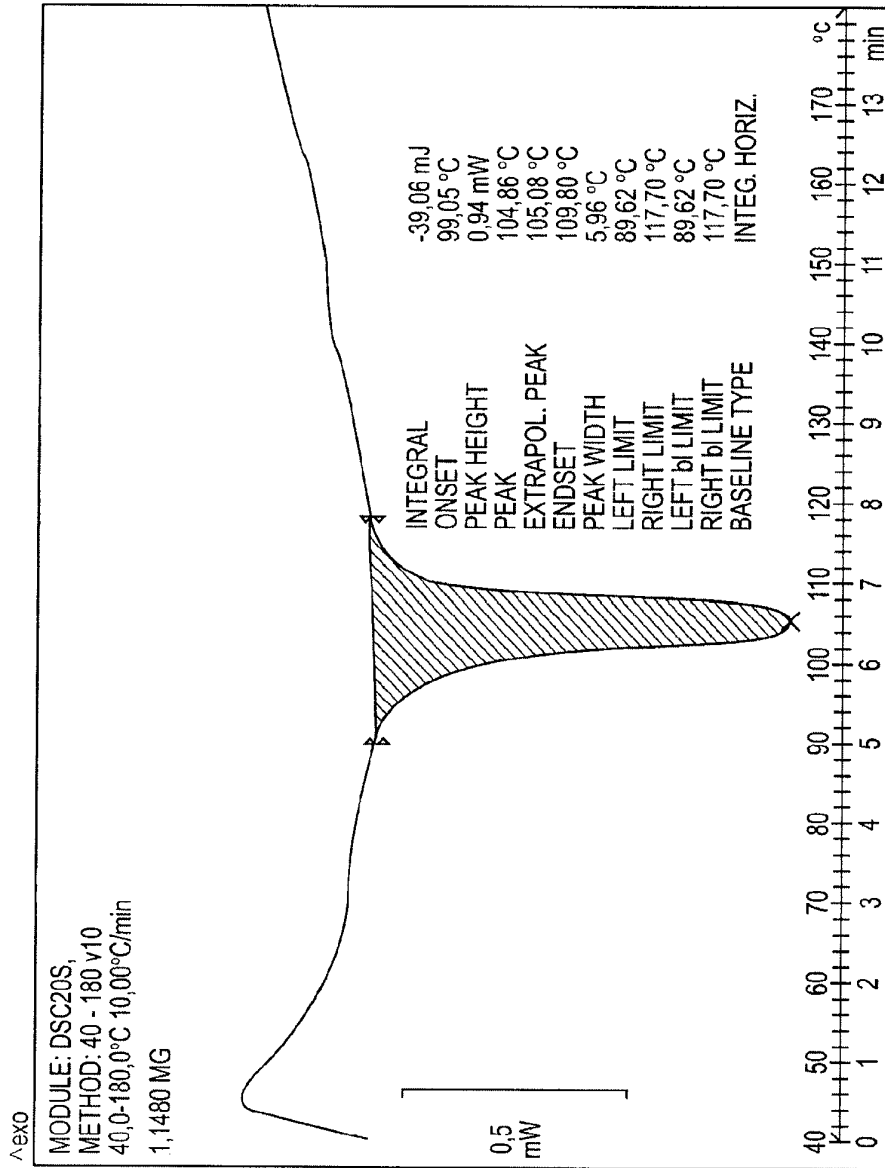
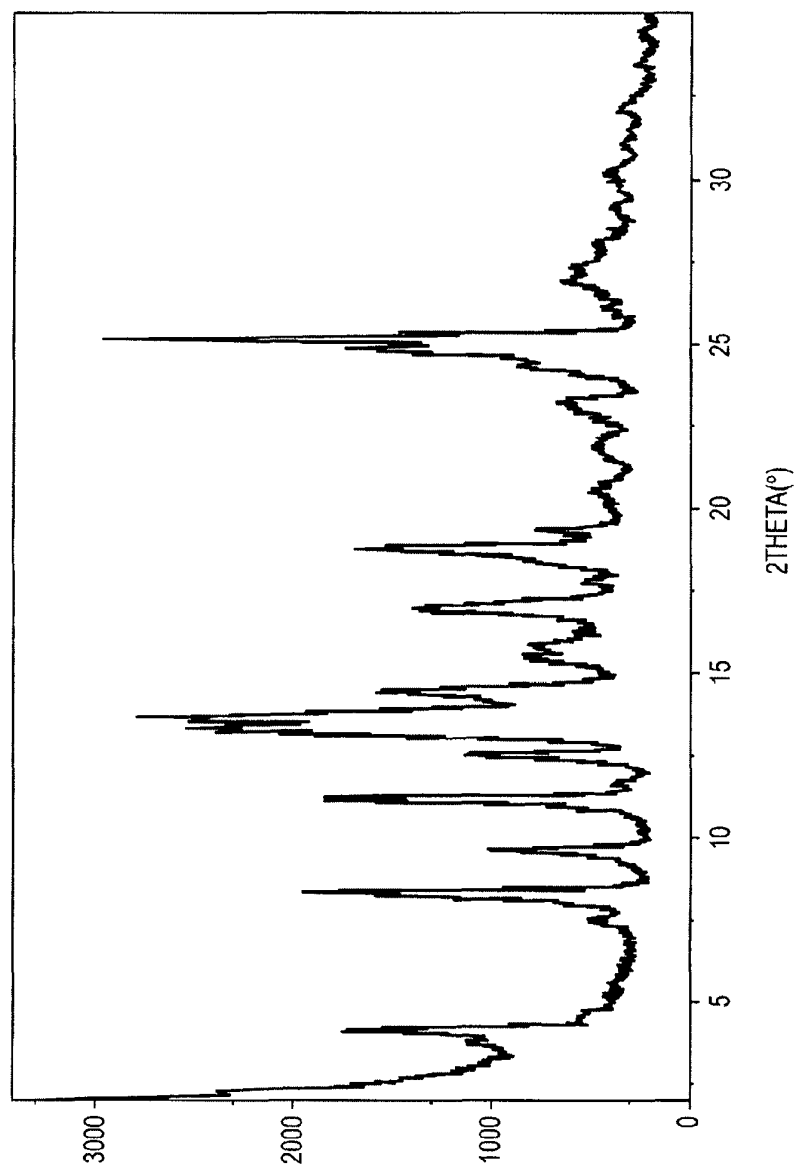


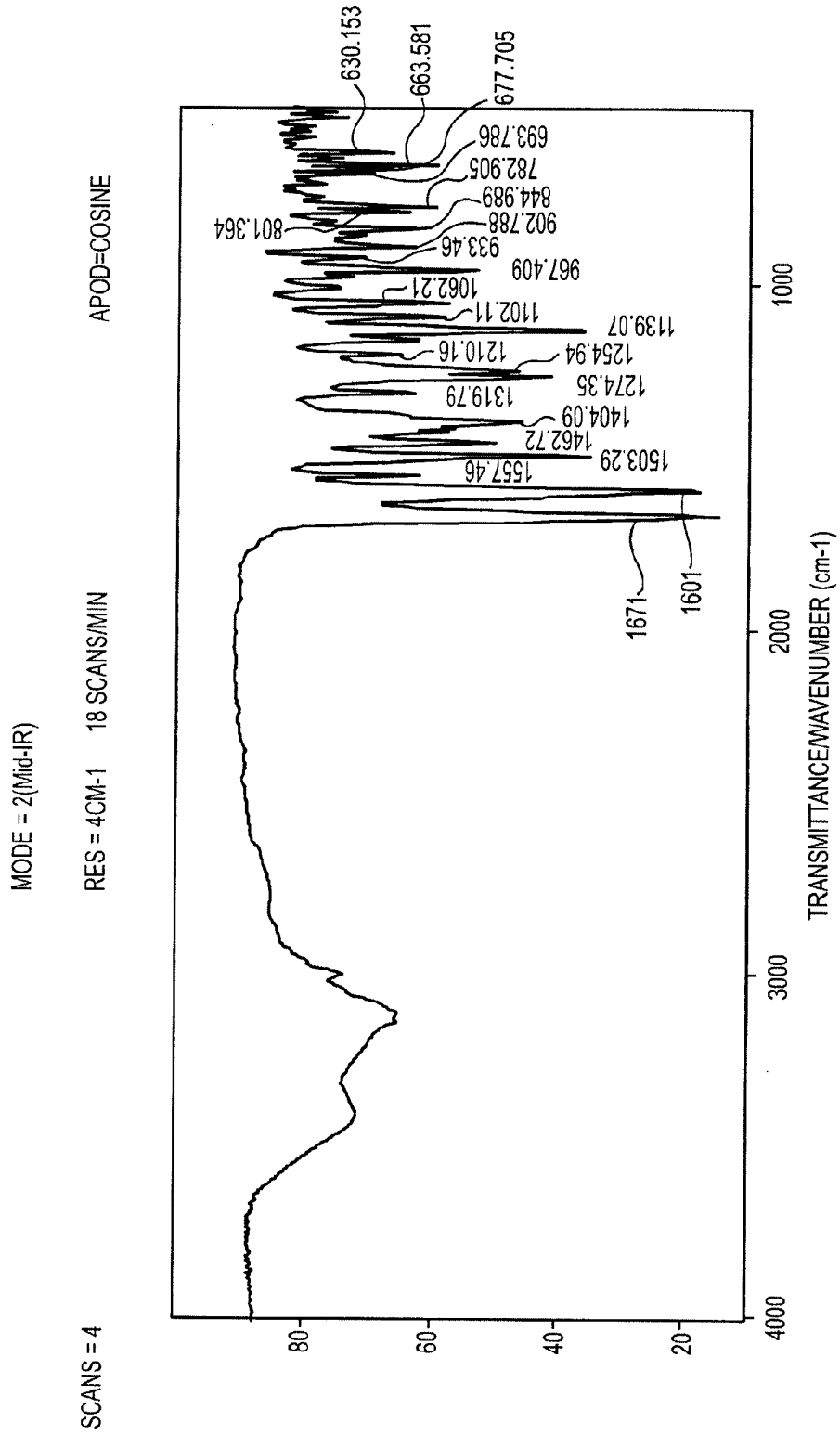
FIG.8



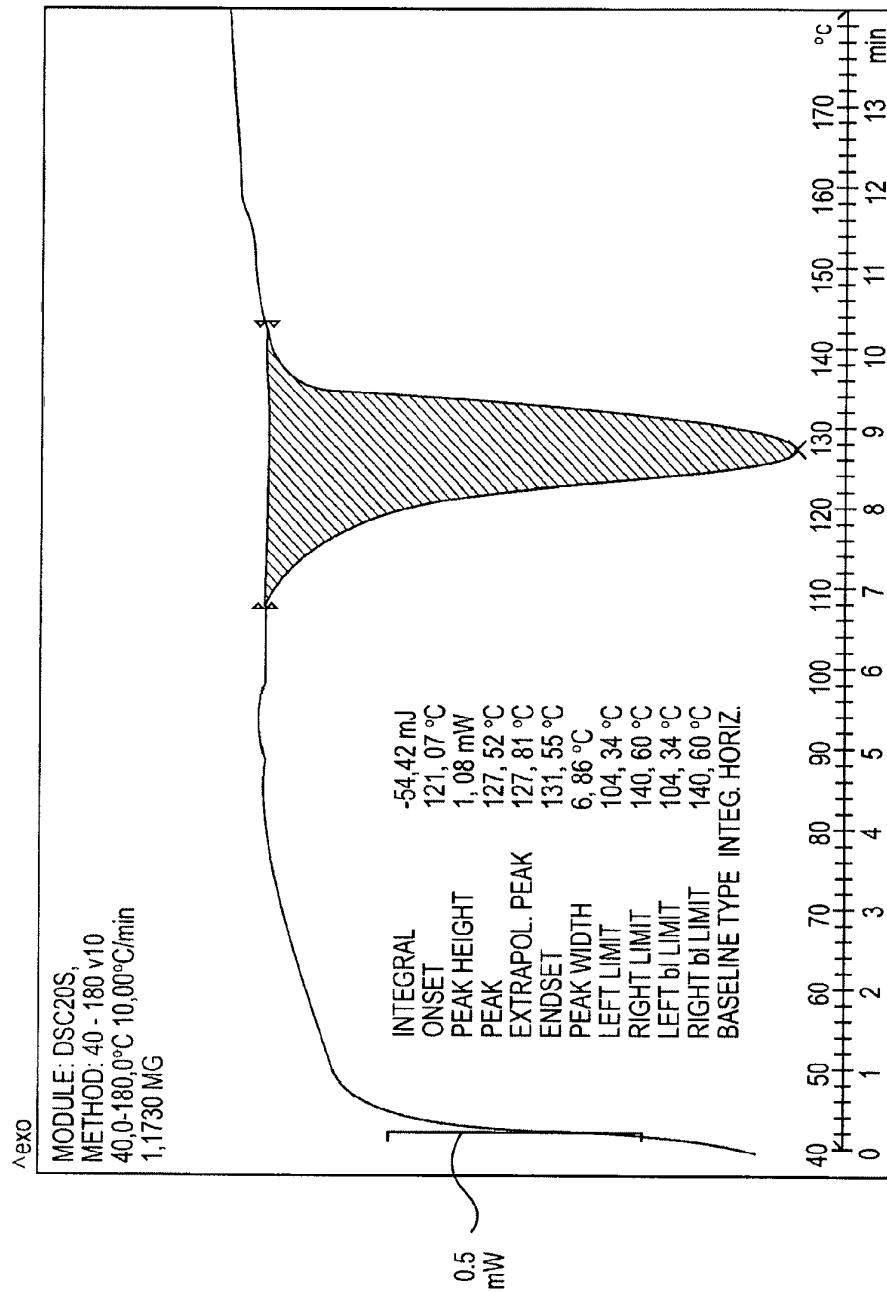
**FIG.9**



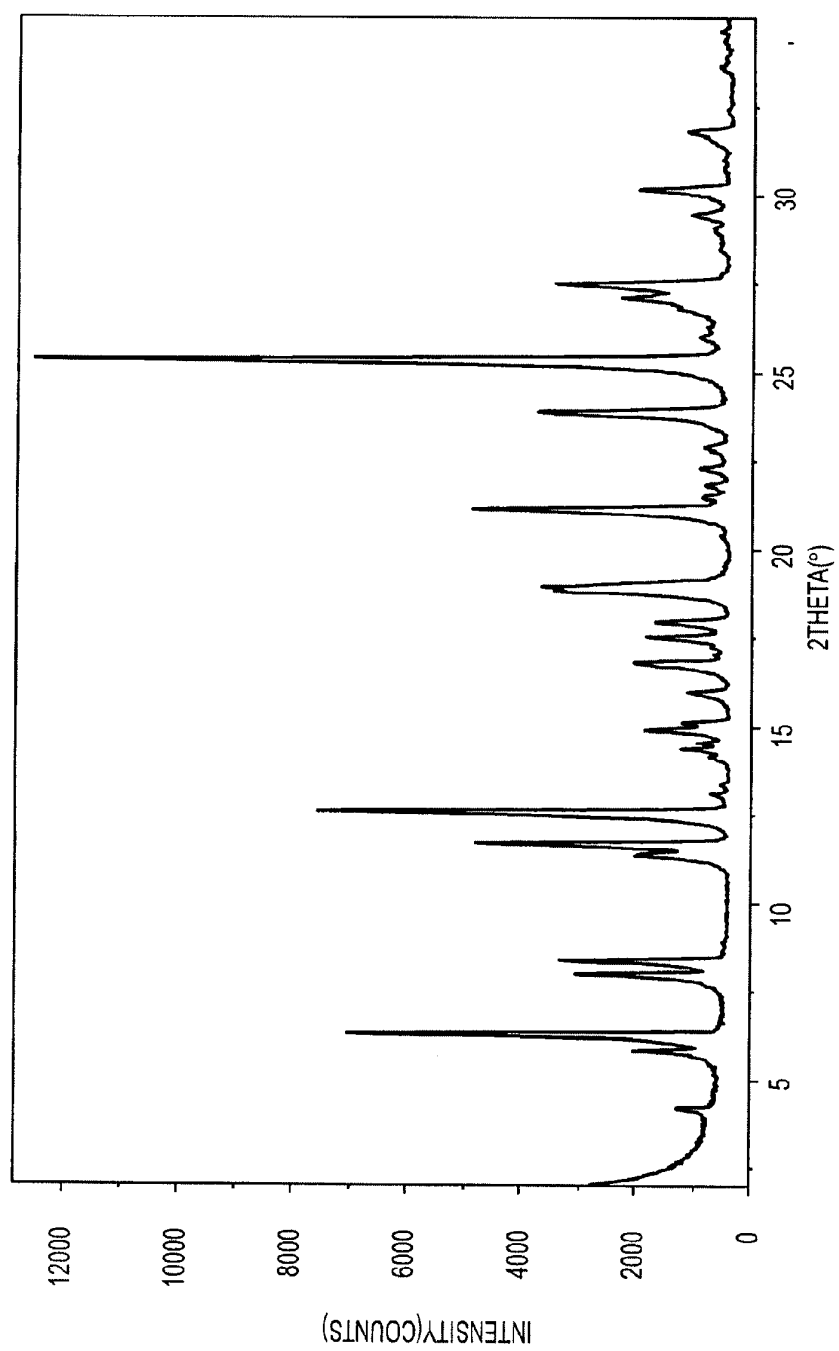
**FIG.10**

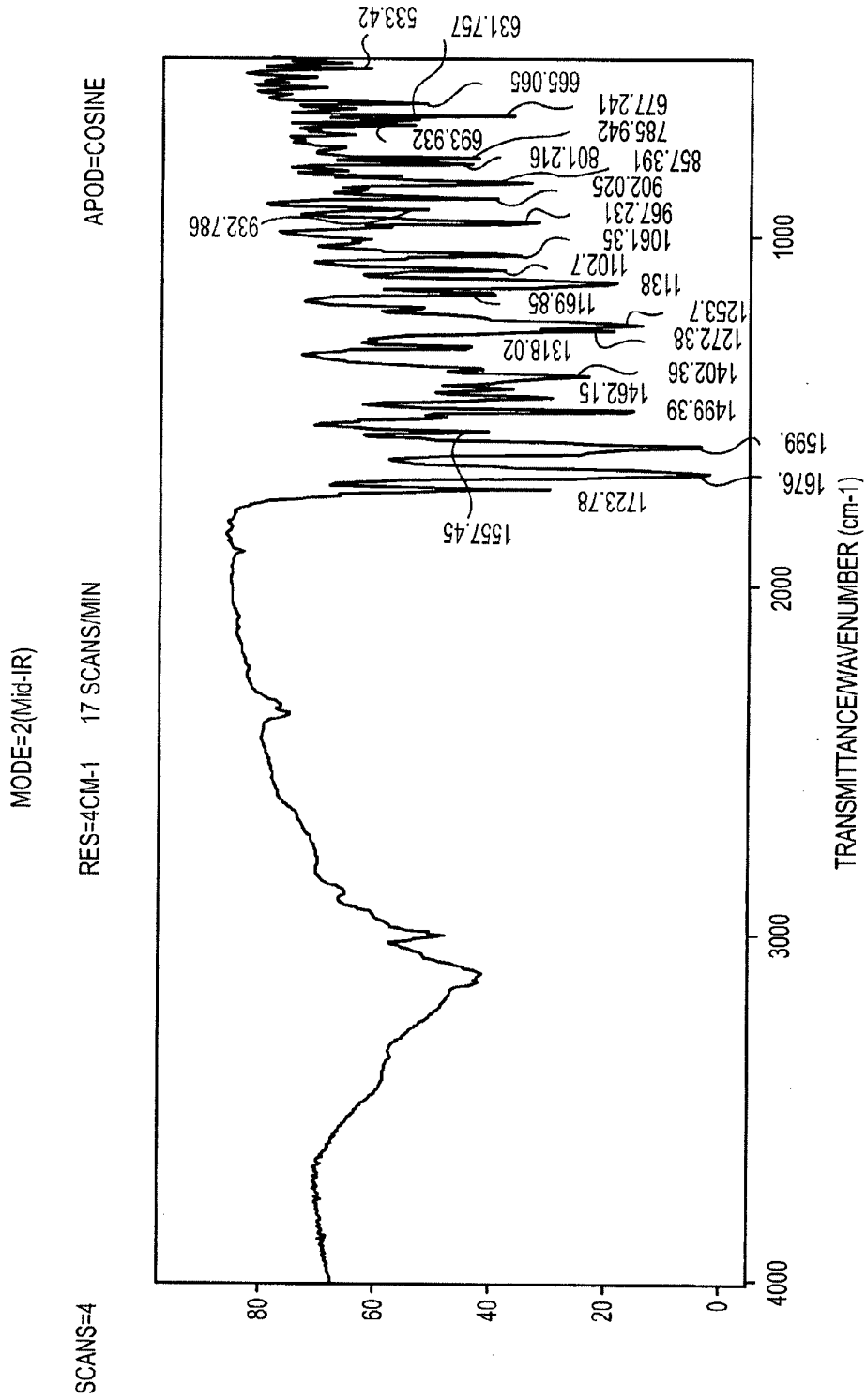


**FIG.11**

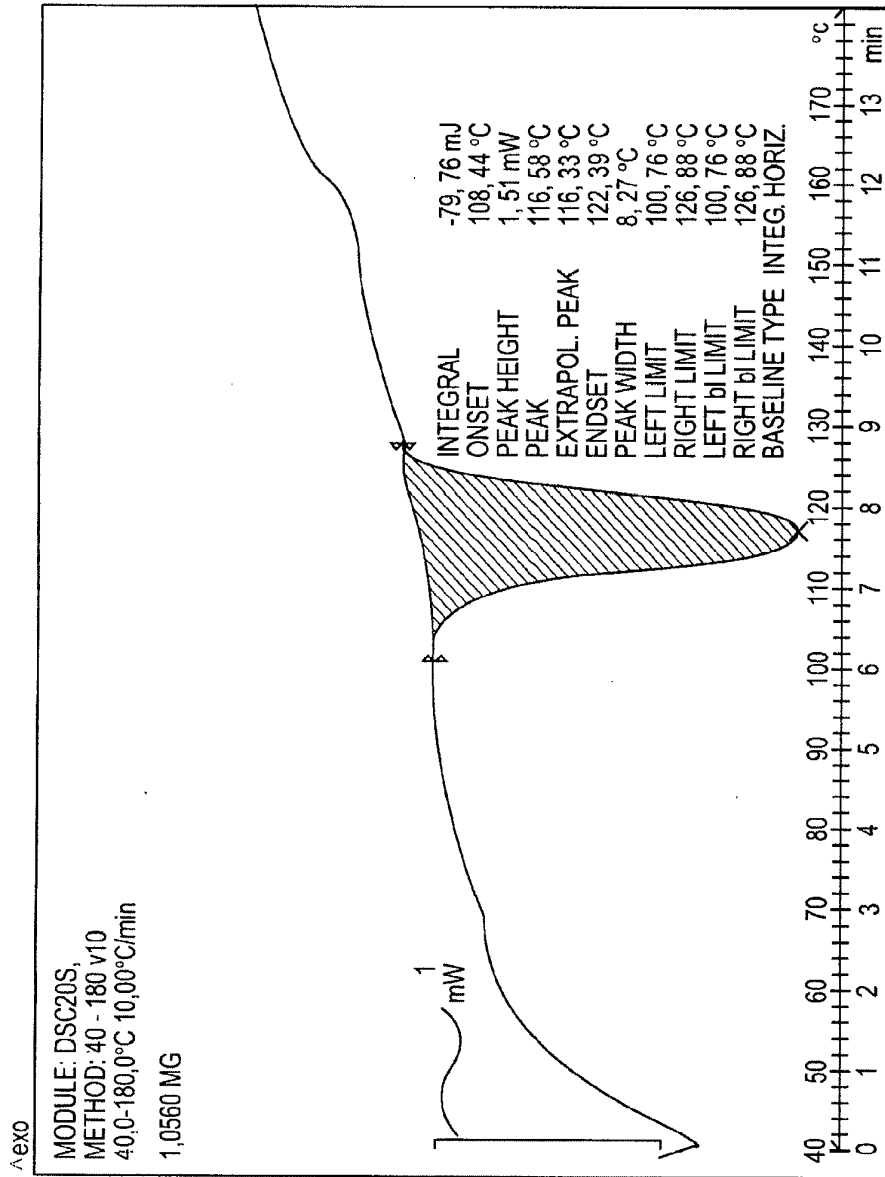
**FIG.12**



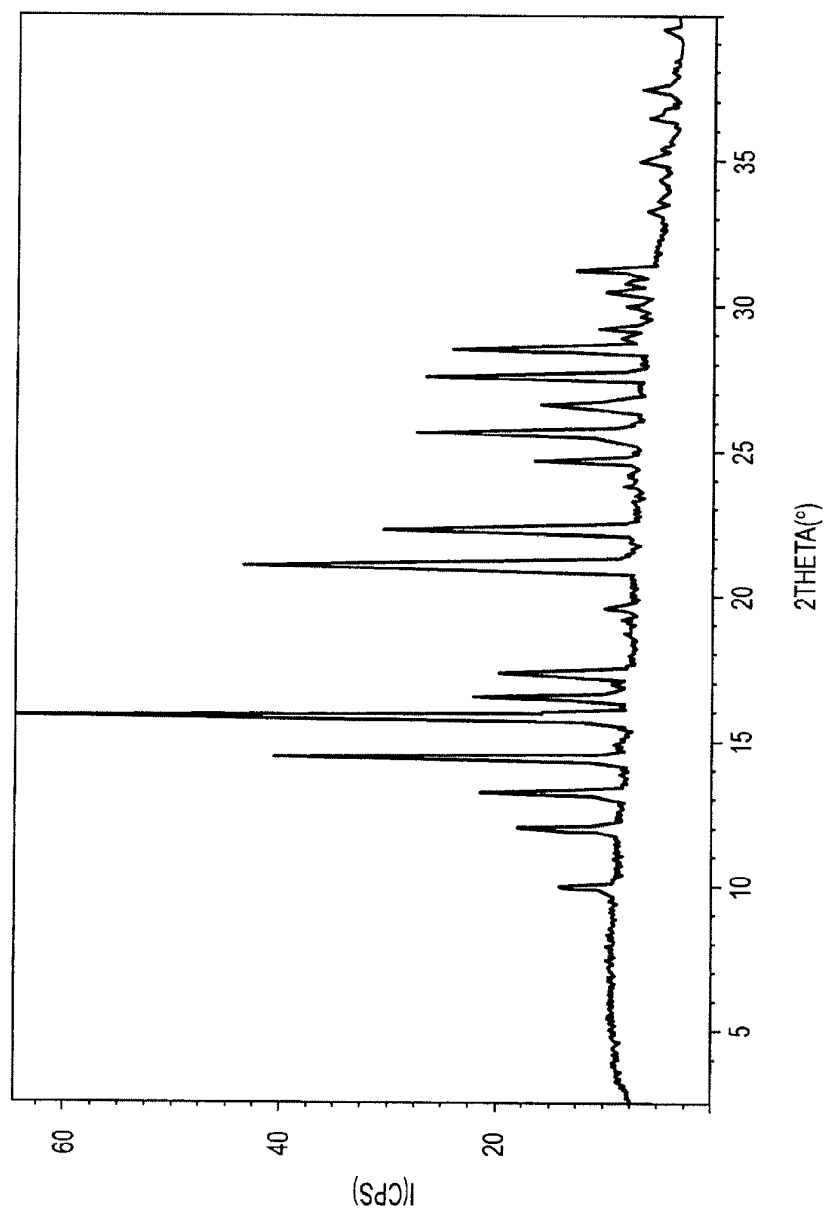
**FIG.13**



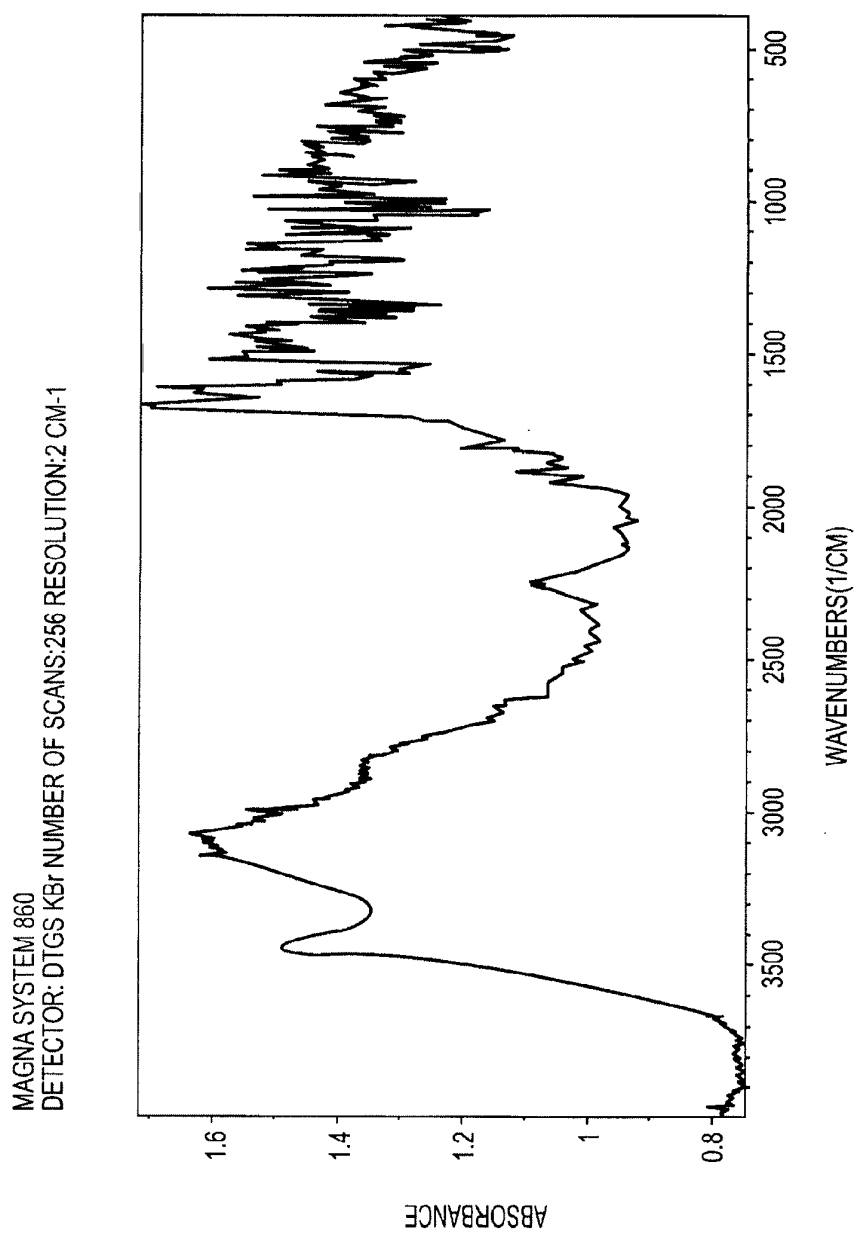
**FIG.14**

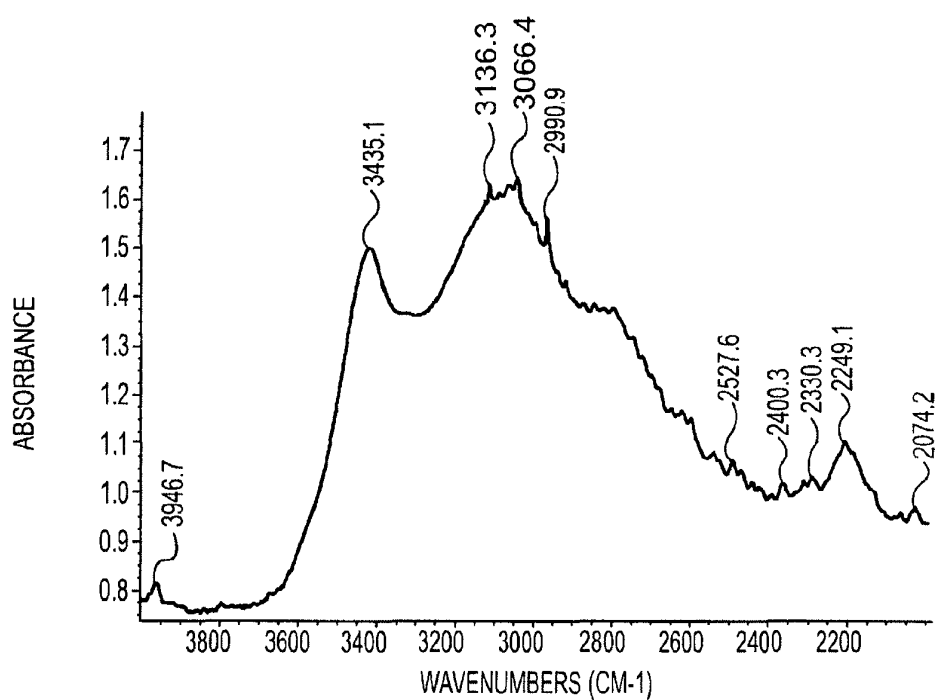
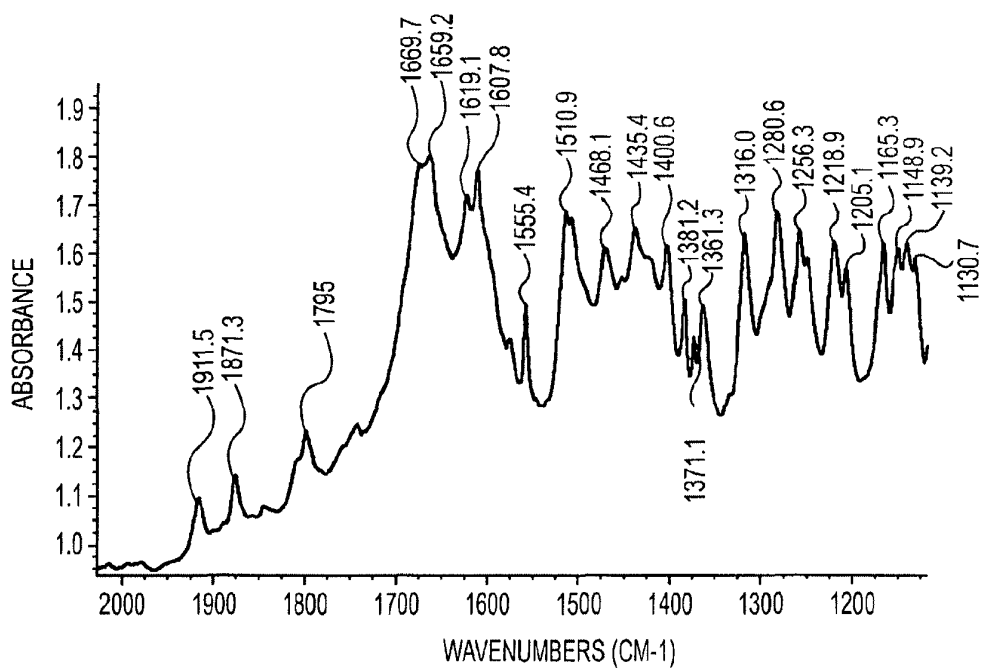


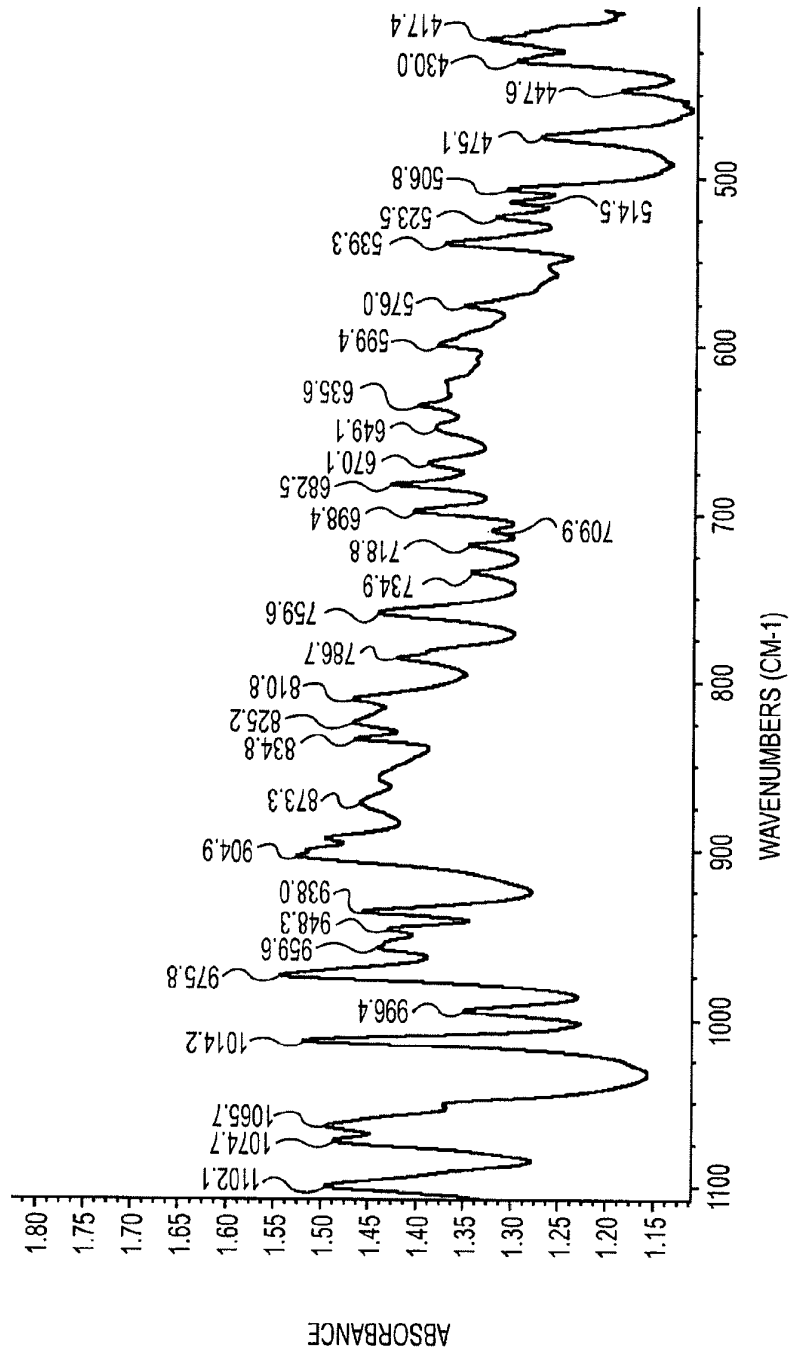
**FIG.15**



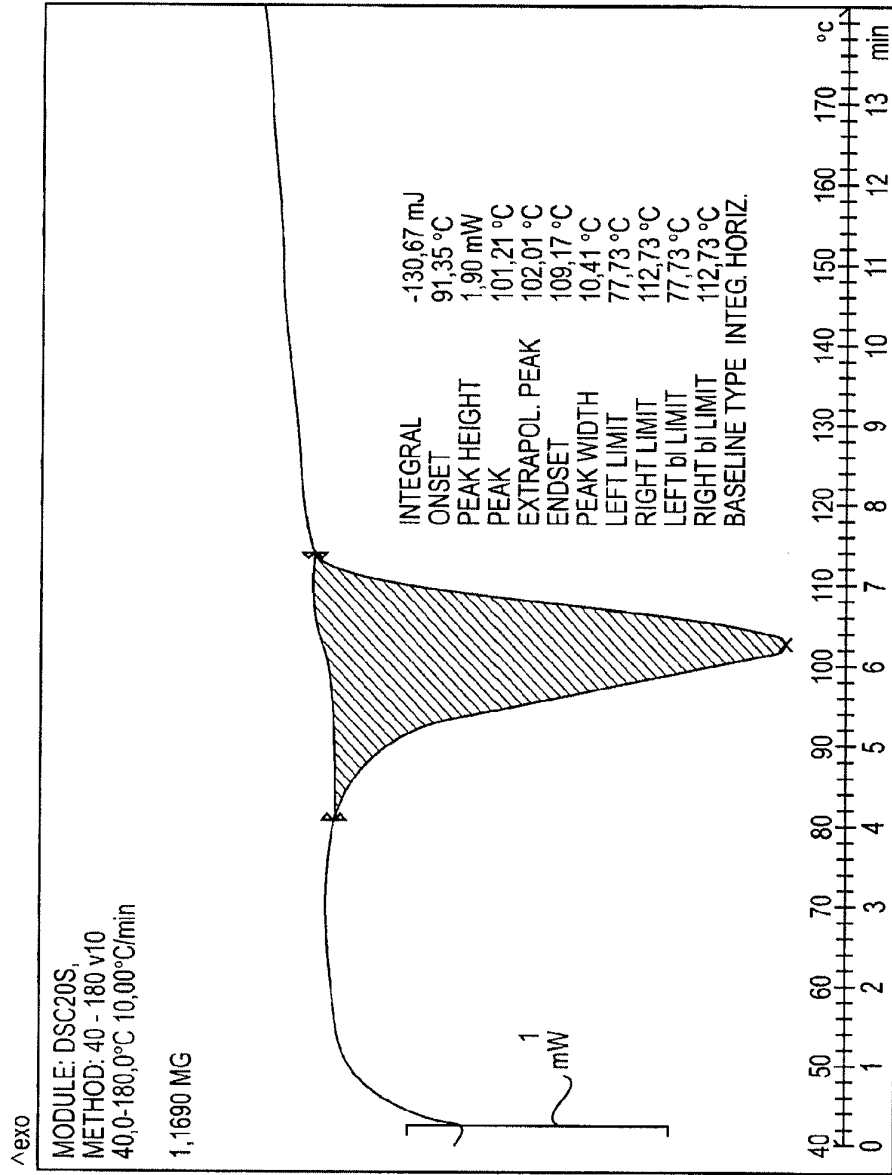
**FIG.16**

**FIG.17a**

**FIG.17b****FIG.17c**

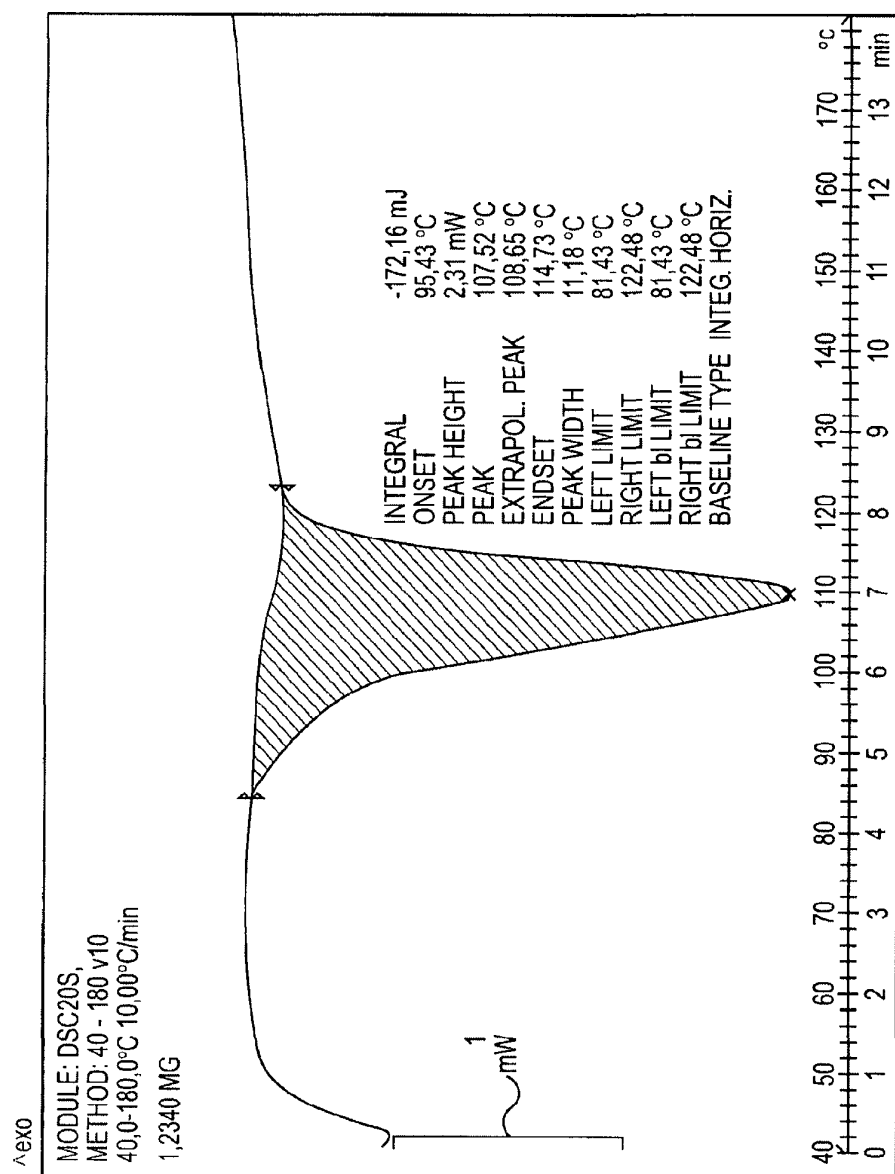


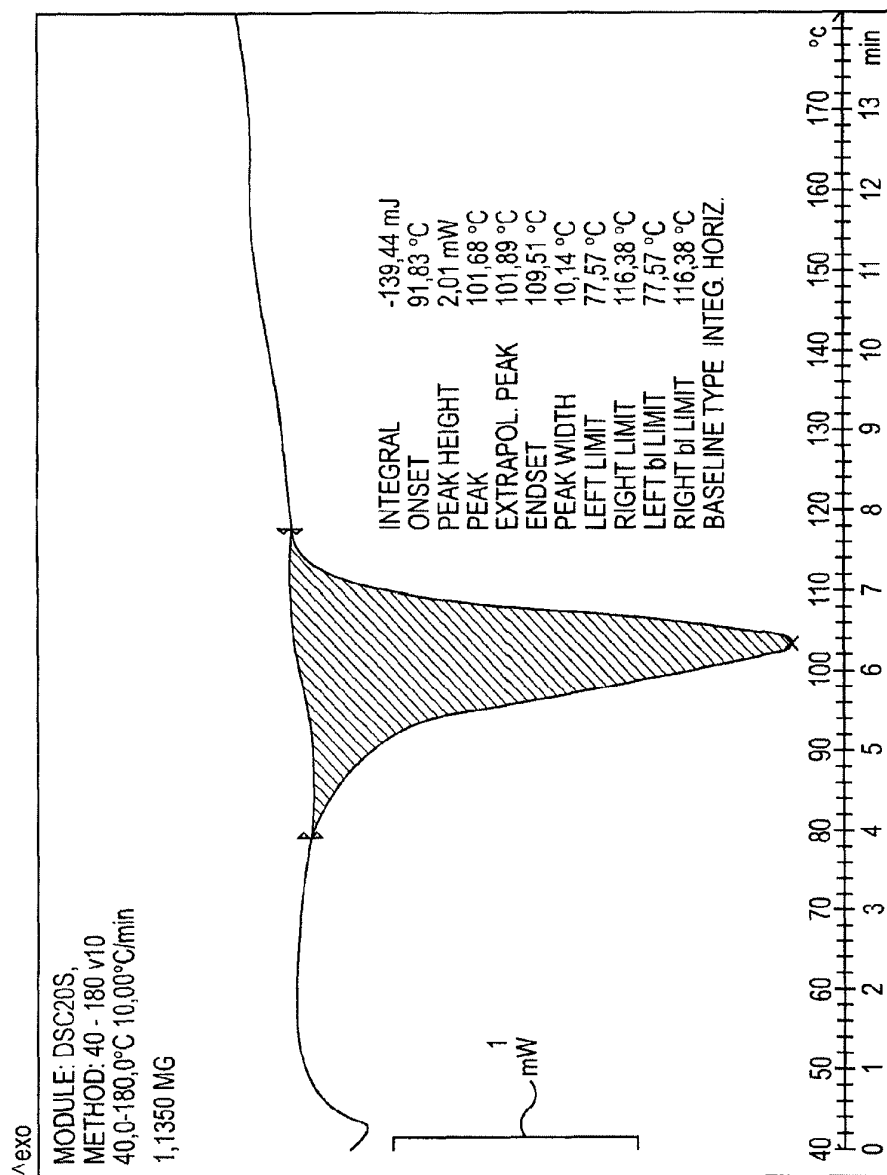
**FIG. 17d**

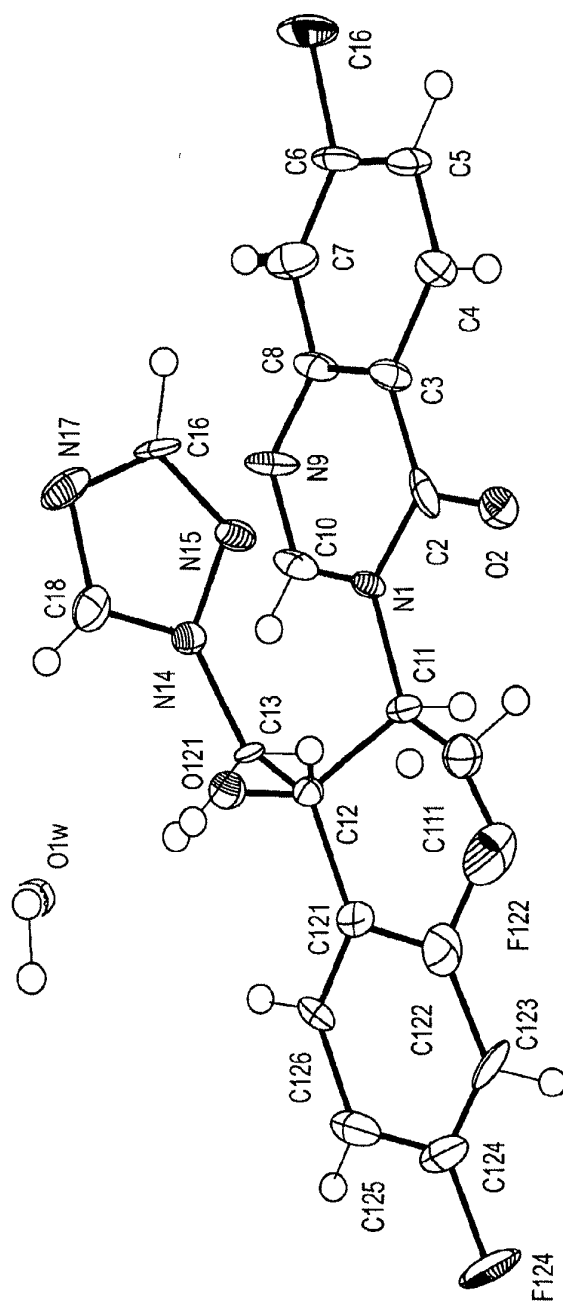


**FIG. 18a**

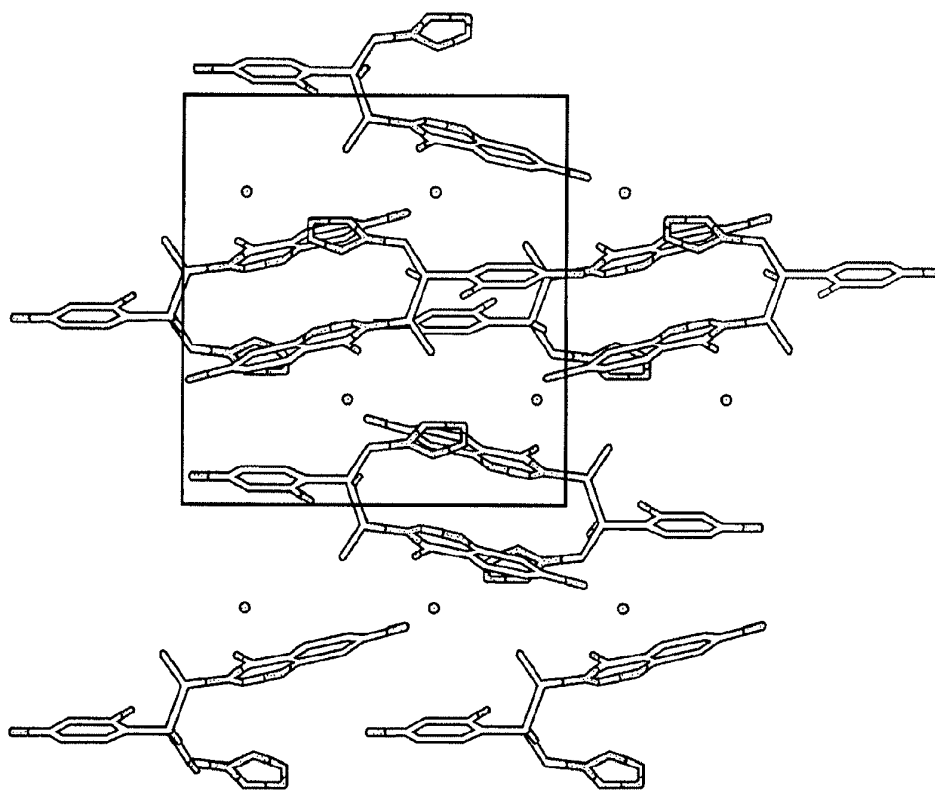


**FIG.18b**

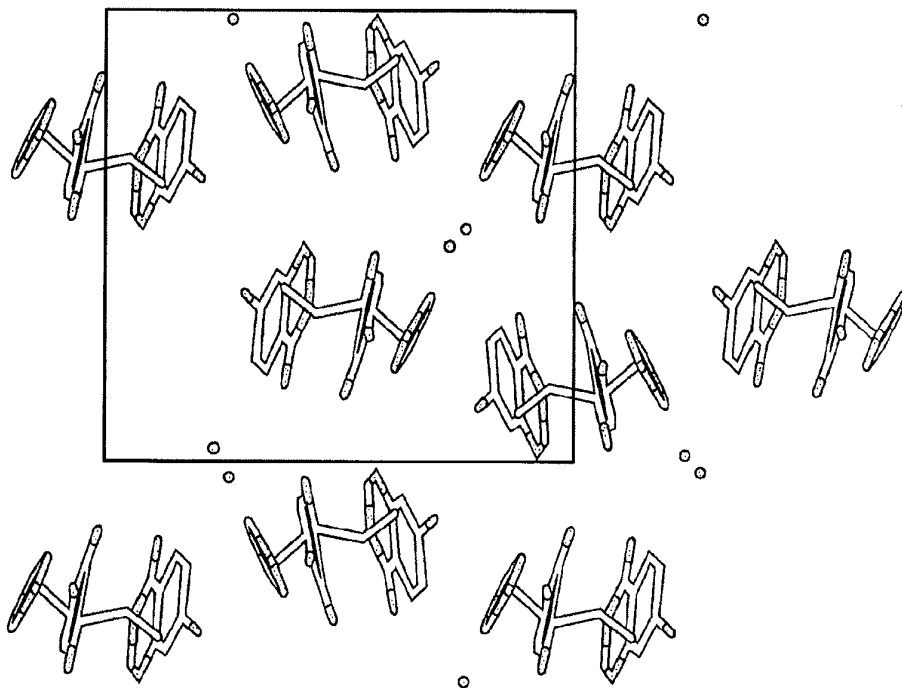
**FIG.18c**



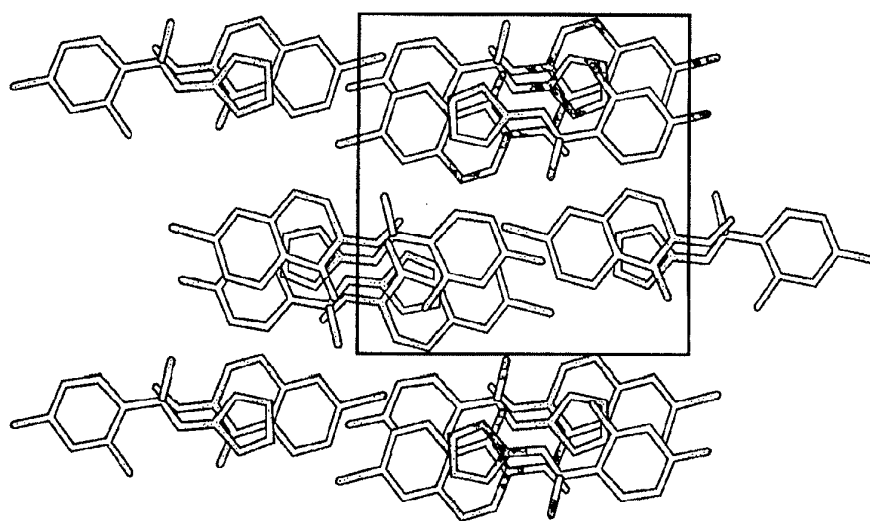
**FIG.19**



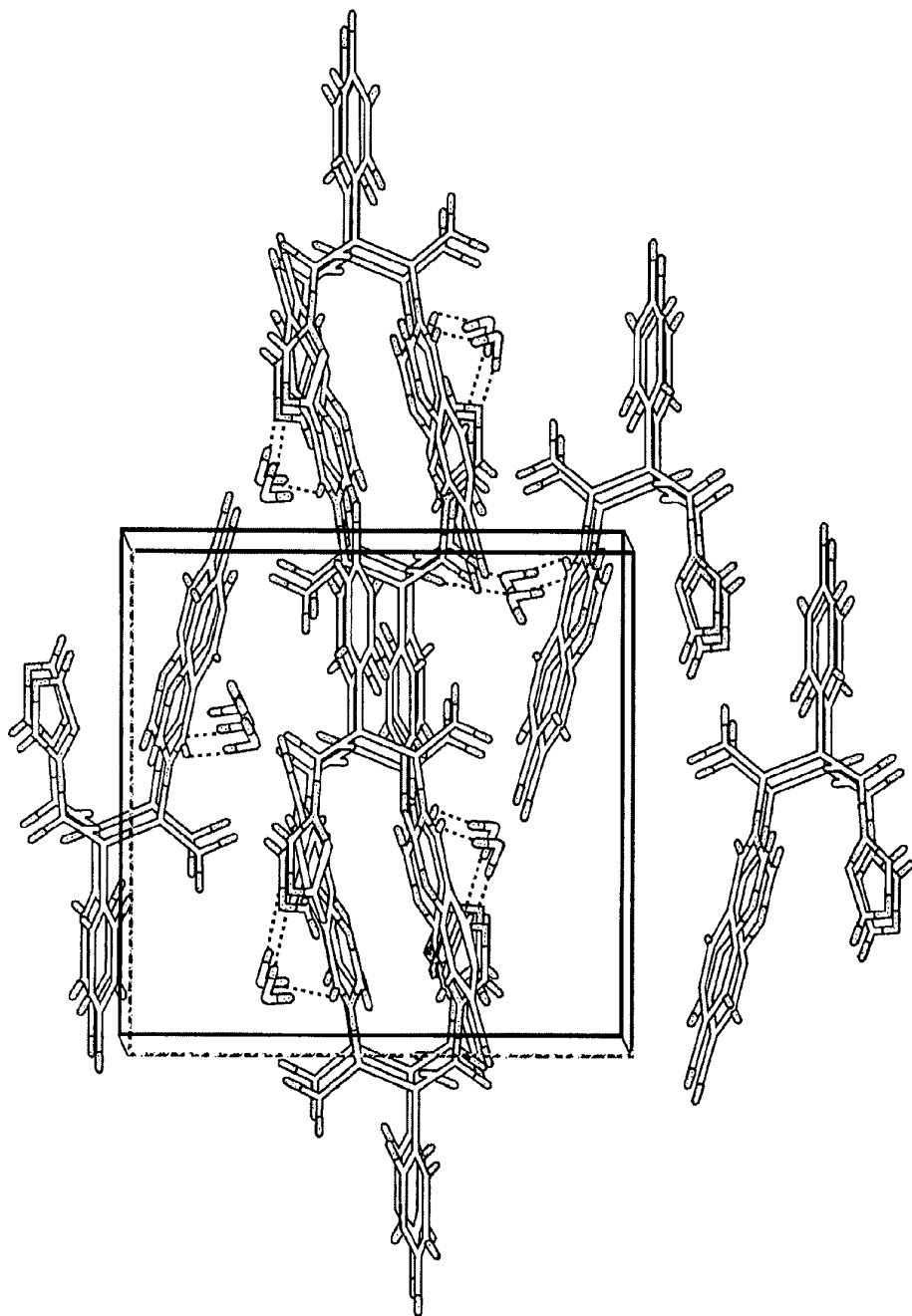
**FIG.20**



**FIG.21**



**FIG.22**



**FIG. 23**



## EUROPEAN SEARCH REPORT

Application Number  
EP 13 16 7750

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (IPC)
X	EP 1 282 084 A (URIACH & CIA SA J [ES]) 5 February 2003 (2003-02-05) * abstract; claims 1-12; example 3 *	1-10	INV. C07D403/06 A61K31/517 A61P31/00
X,D	WO 97/05130 A1 (BARTROLI JAVIER [ES] ET AL) 13 February 1997 (1997-02-13) * abstract; claims 1-18; example 1 *	1-10	
X	BARTROLI J ET AL: "NEW AZOLE ANTIFUNGALS. 3. SYNTHESIS AND ANTIFUNGAL ACTIVITY OF 3-SUBSTITUTED-4(3H)-QUINAZOLINONES", JOURNAL OF MEDICINAL CHEMISTRY, AMERICAN CHEMICAL SOCIETY, WASHINGTON, US, vol. 41, 29 April 1998 (1998-04-29), pages 1869-1882, XP001106675, ISSN: 0022-2623 * abstract; figure 2; tables 1,2,4-9; compound 20 * * page 1876; compound 20 *	1-10	
X	SORBERA, L. A. ET AL: "Albaconazole : antifungal", DRUGS OF THE FUTURE, 28(6), 529-537 CODEN: DRFUD4; ISSN: 0377-8282, 2003, XP002461045, * page 529 *	1-10	TECHNICAL FIELDS SEARCHED (IPC)  C07D A61K A61P
A	CAIRA M R: "CRYSTALLINE POLYMORPHISM OF ORGANIC COMPOUNDS", TOPICS IN CURRENT CHEMISTRY, SPRINGER, BERLIN, DE, vol. 198, 1998, pages 163-208, XP001156954, ISSN: 0340-1022 * the whole document *	1-10	
The present search report has been drawn up for all claims			
Place of search Munich		Date of completion of the search 5 September 2013	Examiner Von Daacke, Axel
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

1  
EPO FORM 1503 03.82 (P04C01)



**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 13 16 7750

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.  
The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

05-09-2013

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 1282084	A	05-02-2003	AR 027607 A1	02-04-2003
			AT 250256 T	15-10-2003
			AU 3744701 A	17-09-2001
			BR 0108990 A	30-12-2003
			CA 2402390 A1	13-09-2001
			DE 60100810 D1	23-10-2003
			DE 60100810 T2	01-07-2004
			DK 1282084 T3	02-02-2004
			EP 1282084 A2	05-02-2003
			ES 2159488 A1	01-10-2001
			ES 2207605 T3	01-06-2004
			HK 1052244 A1	19-12-2003
			JP 4772248 B2	14-09-2011
			JP 2004500391 A	08-01-2004
			MX PA02008556 A	24-02-2003
			NO 20024256 A	06-09-2002
			PT 1282084 E	27-02-2004
			US 2003064986 A1	03-04-2003
			WO 0166519 A2	13-09-2001
WO 9705130	A1	13-02-1997	AT 198593 T	15-01-2001
			AU 6870396 A	26-02-1997
			BR 9606547 A	14-10-1997
			CA 2201477 A1	13-02-1997
			DE 69611512 D1	15-02-2001
			DE 69611512 T2	31-05-2001
			DK 0783501 T3	29-01-2001
			EP 0783501 A1	16-07-1997
			ES 2153975 T3	16-03-2001
			GR 3035621 T3	29-06-2001
			JP 4049207 B2	20-02-2008
			JP H10508317 A	18-08-1998
			NO 971470 A	01-04-1997
			PT 783501 E	30-04-2001
			US 5807854 A	15-09-1998
			WO 9705130 A1	13-02-1997

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

## REFERENCES CITED IN THE DESCRIPTION

*This list of references cited by the applicant is for the reader's convenience only. It does not form part of the European patent document. Even though great care has been taken in compiling the references, errors or omissions cannot be excluded and the EPO disclaims all liability in this regard.*

## Patent documents cited in the description

- US 5807854 A [0002]

## Non-patent literature cited in the description

- **Z. OTWINOWSKI ; W. MINOR.** *Methods Enzymol.*, 1997, vol. 276, 307 [0127] [0129]
- **Z. OTWINOWSKI ; W. MINOR.** *Methods Enzymol.*, 1997, vol. 276, 307 [0130]
- **M. C. BURLA ; R. CALIANDRO ; M. CAMALLI ; B. CARROZZINI ; G. L. CASCARANO ; L. DE CARO ; C. GIACOVAZZO ; G. POLIDORI ; R. SPAGNA.** *J. Appl. Cryst.*, 2005, vol. 38, 381 [0131]
- International Tables for Crystallography. Kluwer Academic Publishers, 1992, vol. C [0133]
- **H. D. FLACK.** *Acta Cryst.*, 1983, vol. A39, 876 [0134]
- **G. M. SHELDRICK.** SHELXL97. *A Program for Crystal Structure Refinement*, 1997 [0135]
- ORTEPII. **C. K. JOHNSON.** Report ORNL-5138. Oak Ridge National Laboratory, 1976 [0135]
- Mercury 1.4.1. Cambridge Crystallographic Diffraction Center, 2005 [0135]
- **J. GLUSKER ; K. TRUEBLOOD.** *Crystal Structure Analysis: A Primer.* Oxford University press, 1985, 87 [0136]
- The Merck Index. Merck & Co., Inc, 2001 [0148]
- International Cosmetic Ingredient Dictionary and Handbook. CTFA, 2004 [0148]
- Inactive Ingredient Guide. U.S. Food and Drug Administration (FDA) Center for Drug Evaluation and Research [0148]
- Goodman and Gillman's: The Pharmacological Bases of Therapeutics. Pergamon Press, 1990 [0149]
- Remington's Pharmaceutical Sciences. Mack Publishing Co, 1990 [0149]
- Remington's Pharmaceutical Sciences. Mack Publishing Co, 1990, 1435-1712 [0207]
- **BARTROLI et al.** *J. Med. Chem.*, 1998, vol. 41 (11), 1869-1882 [0208]