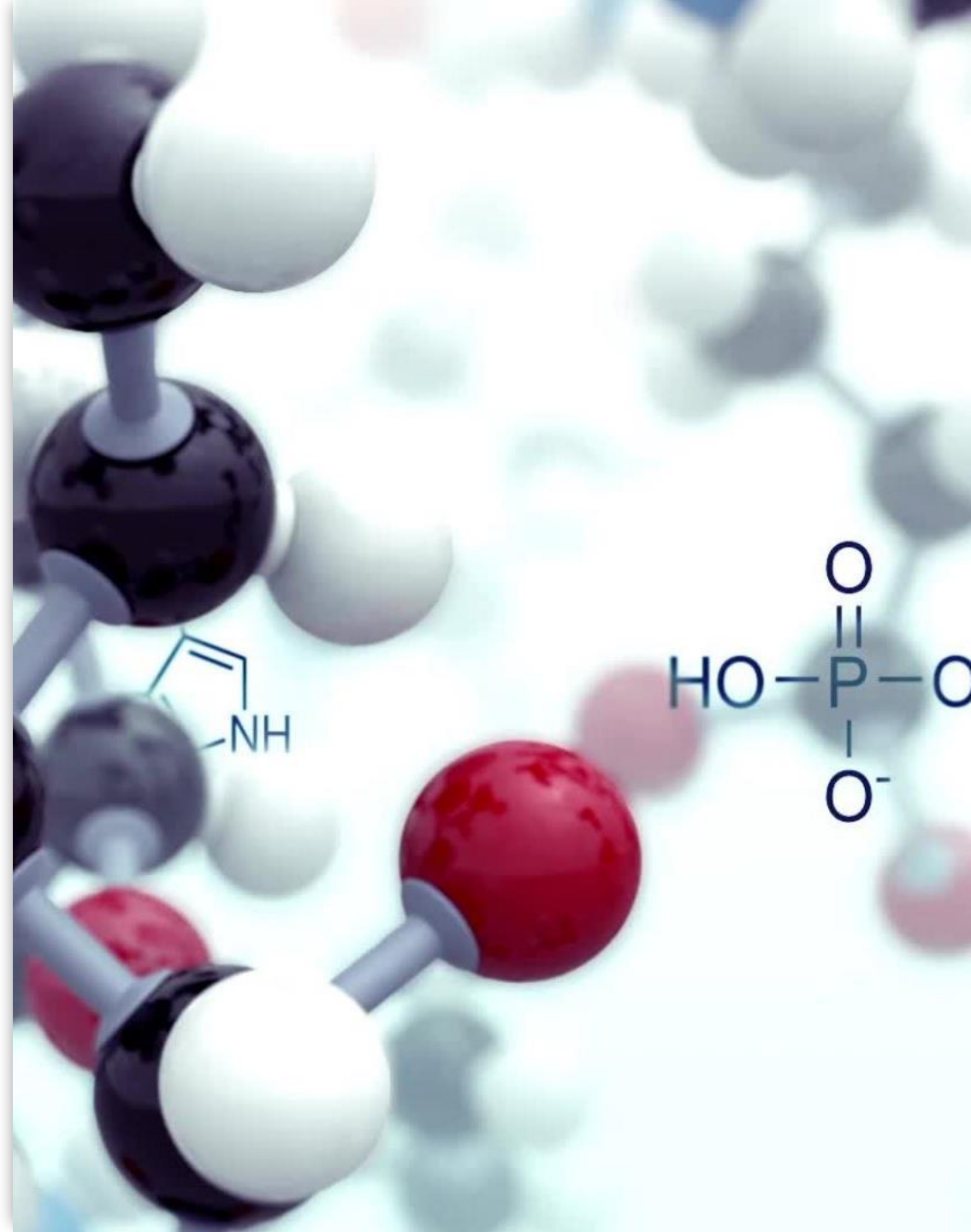


PFA Removal

By: Henri Chainet

What are PFAs?

PFAs or forever chemicals (per-polyfluoroalkyl) are synthetic chemicals mainly used to make non-stick surfaces. They are made up of carbon and fluorine bonds, one of the strongest natural chemical bonds, making it really hard to remove them from the environment?



So how do we remove them?

There are a few ways to remove PFAs from the environment, most notably:

- Granulated activated carbon (GAC), traps the PFAs in the pores of activated carbon.
- Ion exchange resins (IX resins), the specialized resin sites bind to the PFAs via ionic interactions.
- Reverse Osmosis (RO), pressurizes water to push it through a membrane that is too large for PFAs to go through.
- The Guillotine (DMSO+ lye), takes advantage of the fact that the PFAs have a reactive head made of oxygen, creating a reaction that splits the carbon atoms off the PFA, creating fluoride (often added to water to prevent tooth decay), carbon dioxide and formic acid (which is non harmful)

What is my plan?

My plan is to use some of the previously mentioned methods to separate the water and PFAs, then destroy the PFAs through the Guillotine process.

