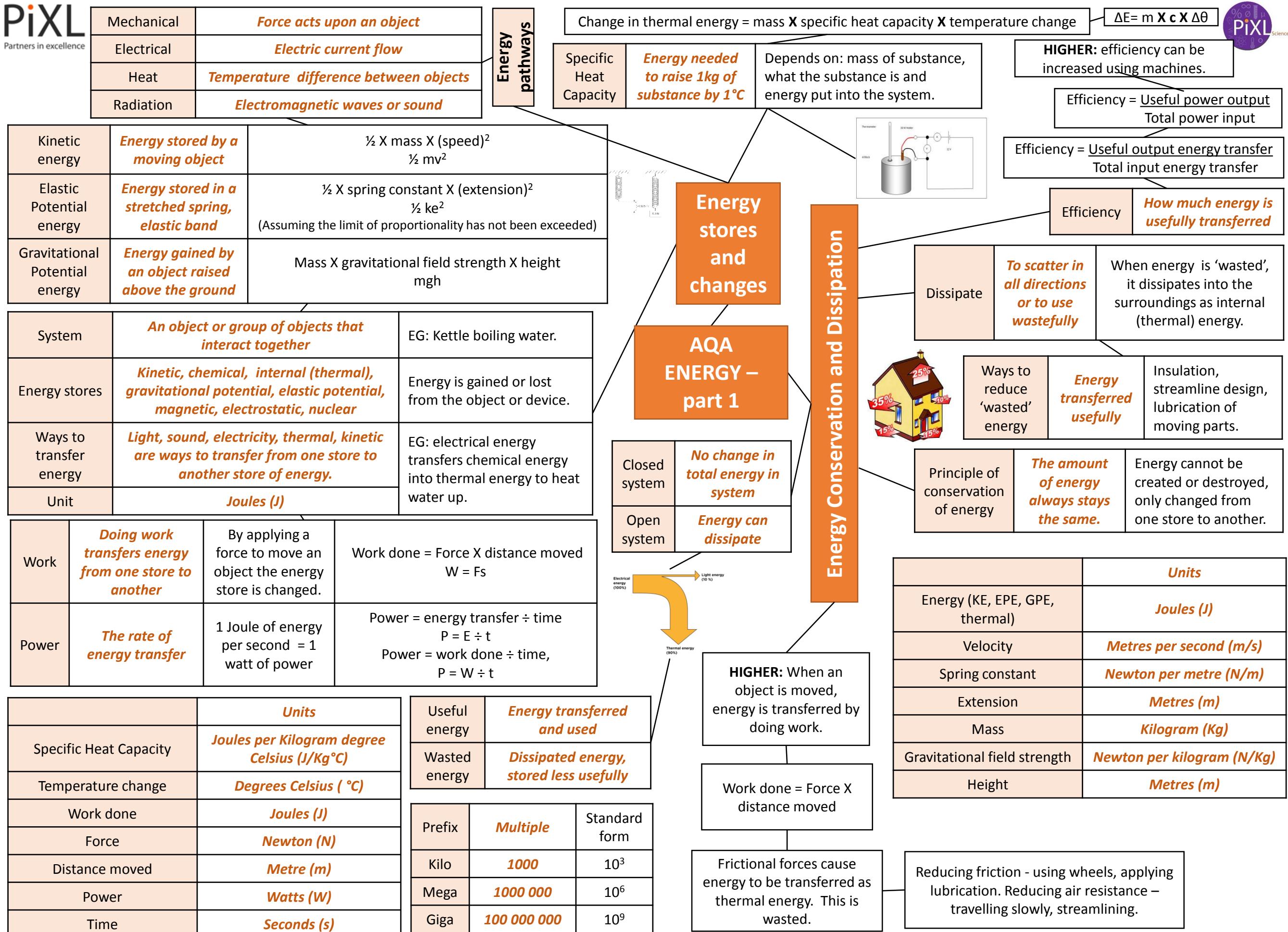
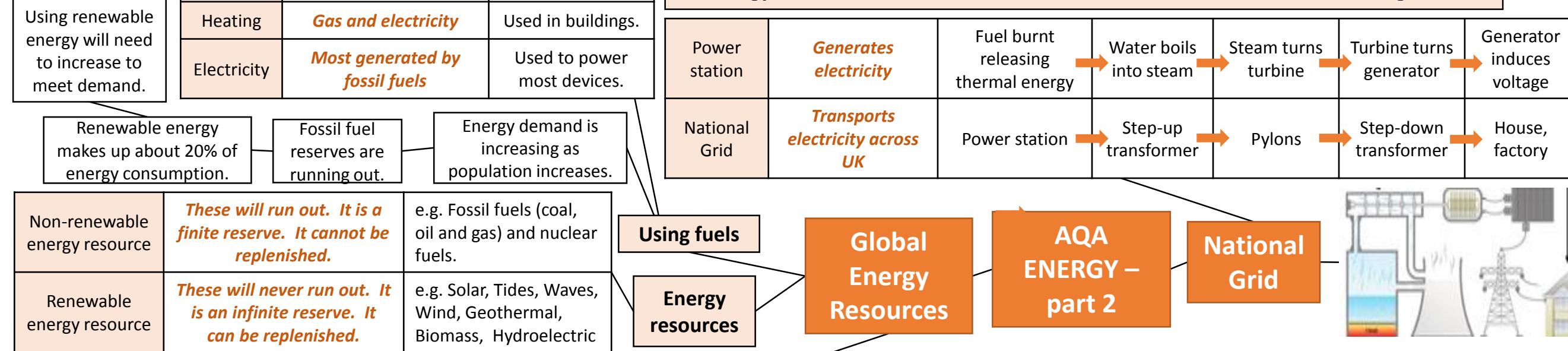


# PHYSICS KNOWLEDGE ORGANISERS

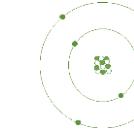




Energy resource	How it works	Uses	Positive	Negative
Fossil Fuels (coal, oil and gas)	<b>Burnt to release thermal energy used to turn water into steam to turn turbines</b>	Generating electricity, heating and transport	Provides most of the UK energy. Large reserves. Cheap to extract. Used in transport, heating and making electricity. Easy to transport.	Non-renewable. Burning coal and oil releases sulfur dioxide. When mixed with rain makes acid rain. Acid rain damages building and kills plants. Burning fossil fuels releases carbon dioxide which contributes to global warming. Serious environmental damage if oil spilt.
Nuclear	<b>Nuclear fission process</b>	Generating electricity	No greenhouse gases produced. Lots of energy produced from small amounts of fuel.	Non-renewable. Dangers of radioactive materials being released into air or water. Nuclear sites need high levels of security. Start up costs and decommission costs very expensive. Toxic waste needs careful storing.
Biofuel	<b>Plant matter burnt to release thermal energy</b>	Transport and generating electricity	Renewable. As plants grow, they remove carbon dioxide. They are 'carbon neutral'.	Large areas of land needed to grow fuel crops. Habitats destroyed and food not grown. Emits carbon dioxide when burnt thus adding to greenhouse gases and global warming.
Tides	<b>Every day tides rise and fall, so generation of electricity can be predicted</b>	Generating electricity	Renewable. Predictable due to consistency of tides. No greenhouse gases produced.	Expensive to set up. A dam like structure is built across an estuary, altering habitats and causing problems for ships and boats.
Waves	<b>Up and down motion turns turbines</b>	Generating electricity	Renewable. No waste products.	Can be unreliable depends on wave output as large waves can stop the pistons working.
Hydroelectric	<b>Falling water spins a turbine</b>	Generating electricity	Renewable. No waste products.	Habitats destroyed when dam is built.
Wind	<b>Movement causes turbine to spin which turns a generator</b>	Generating electricity	Renewable. No waste products.	Unreliable – wind varies. Visual and noise pollution. Dangerous to migrating birds.
Solar	<b>Directly heats objects in solar panels or sunlight captured in photovoltaic cells</b>	Generating electricity and some heating	Renewable. No waste products.	Making and installing solar panels expensive. Unreliable due to light intensity.
Geothermal	<b>Hot rocks under the ground heats water to produce steam to turn turbine</b>	Generating electricity and heating	Renewable. Clean. No greenhouse gases produced.	Limited to a small number of countries. Geothermal power stations can cause earthquake tremors.

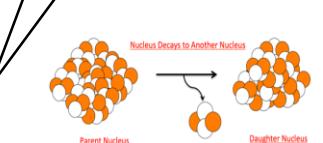


Radius of an atom  
 $1 \times 10^{-10}\text{m}$



**Electrons gained**  
Negative ion

**Electrons lost**  
Positive ion



Atom	<b>Same number of protons and electrons</b>
Ion	<b>Unequal number of electrons to protons</b>
Mass number	<b>Number of protons and neutrons</b>
Atomic number	<b>Number of protons</b>

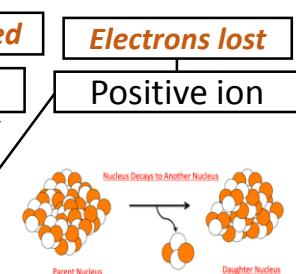
Particle	Charge	Size	Found
Neutron	<b>None</b>	<b>1</b>	<i>In the nucleus</i>
Proton	<b>+</b>	<b>1</b>	
Electron	<b>-</b>	<b>Tiny</b>	<b>Orbits the nucleus</b>

Isotope	${}^6_3\text{Li}$		${}^7_3\text{Li}$	
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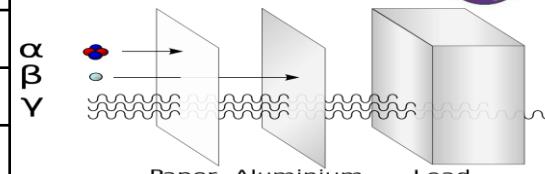
**Different forms of an element with the same number of protons but different number of neutrons**

### Discovery of the nucleus

Democritus	Suggested idea of atoms as small spheres that cannot be cut.
J J Thomson (1897)	Discovered electrons – emitted from surface of hot metal. Showed electrons are negatively charged and that they are much less massive than atoms.
Thomson (1904)	Proposed 'plum pudding' model – atoms are a ball of positive charge with negative electrons embedded in it.
Geiger and Marsden (1909)	Directed beam of alpha particles ( $\text{He}^{2+}$ ) at a thin sheet of gold foil. Found some travelled through, some were deflected, some bounced back.
Rutherford (1911)	Used above evidence to suggest alpha particles deflected due to electrostatic interaction between the very small charged nucleus, nucleus was massive. Proposed mass and positive charge contained in nucleus while electrons found outside the nucleus which cancel the positive charge exactly.
Bohr (1913)	Suggested modern model of atom – electrons in circular orbits around nucleus, electrons can change orbits by emitting or absorbing electromagnetic radiation. His research led to the idea of some particles within the nucleus having positive charge; these were named protons.
Chadwick (1932)	Discovered neutrons in nucleus – enabling other scientists to account for mass of atom.



Decay	Range in air	Ionising power	Penetration power
Alpha	Few cm	Very strong	Stopped by paper
Beta	Few m	Medium	Stopped by Aluminium
Gamma	Great distances	Weak	Stopped by thick lead



Radioactive decay	<b>Unstable atoms randomly emit radiation to become stable</b>
Detecting	<b>Use Geiger Muller tube</b>
Unit	<b>Becquerel</b>
Ionisation	<b>All radiation ionises</b>

### Atoms and Isotopes

### Atoms and Nuclear Radiation

## AQA ATOMIC STRUCTURE

## PHYSICS ONLY: Hazards and uses of Radioactive emissions and of background radiation

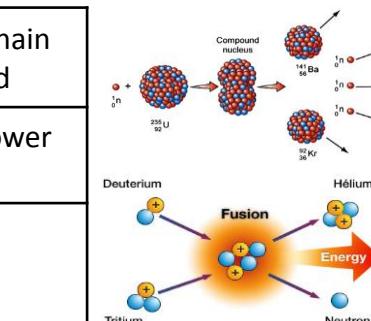
### Nuclear fission and fusion

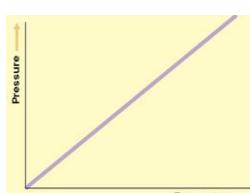
### PHYSICS ONLY: Nuclear energy

Uses	<b>Different isotopes have different half lives</b>	Short half-lives used in high doses, long half lives used in low doses.
Tracers	<b>Used within body</b>	Isotope with short half life injected, allowed to circulate and collect in damaged areas. PET scanner used to detect emitting radiation. Must be beta or gamma as alpha does not penetrate the body.
Radiation therapy	<b>Used to treat illnesses e.g. cancer</b>	Cancer cells killed by gamma rays. High dose used to kill cells. Damage to healthy cells prevented by focussed gamma ray gun.

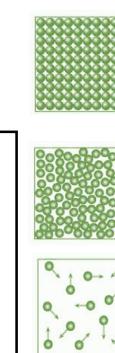
Fuel rods	Made of U-238, 'enriched' with U-235 (3%). Long and thin to allow neutrons to escape, hitting nuclei.
Control rods	Made of Boron. Controls the rate of reaction. Boron absorbs excess neutrons.
Concrete	Neutrons hazardous to humans – thick concrete shield protects workers.

Nuclear fission	<b>One large unstable nucleus splits to make two smaller nuclei</b>	Neutron hits U-235 nucleus, nucleus absorbs neutron, splits emitting two or three neutrons and two smaller nuclei. Process also releases energy.	Process repeats, chain reaction formed
Nuclear fusion	<b>Two small nuclei join to make one larger nucleus</b>	Difficult to do on Earth – huge amounts of pressure and temperature needed.	Used in nuclear power stations
			Occurs in stars





Pressure of a fixed volume of gas increases as temperature increases (temperature increases, speed increases, collisions occur more frequently and with more force so pressure increases).



State	Particle arrangement	Properties
Solid	<i>Packed in a regular structure. Strong forces hold in place so cannot move.</i>	Difficult to change shape.
Liquid	<i>Close together, forces keep contact but can move about.</i>	Can change shape but difficult to compress.
Gas	<i>Separated by large distances. Weak forces so constantly randomly moving.</i>	Can expand to fill a space, easy to compress.

Temperature of gas is linked to the average kinetic energy of the particles.

If kinetic energy increases so does the temperature of gas.

No kinetic energy is lost when gas particles collide with each other or the container.

Gas particles are in a constant state of random motion.

$$P = m \div V$$

Density = mass  $\div$  volume.



Freezing	Liquid turns to a solid. Internal energy decreases.
Melting	Solid turns to a liquid. Internal energy increases.
Boiling / Evaporating	Liquid turns to a gas. Internal energy increases.
Condensation	Gas turns to a liquid. Internal energy decreases.
Sublimation	Solid turns directly into a gas. Internal energy increases.
Conservation of mass	When substances change state, mass is conserved.
Physical change	No new substance is made, process can be reversed.

## Particle model

### AQA PARTICLE MODEL OF MATTER

#### Pressure

**PHYSICS ONLY:** when you do work the temperature increases e.g. pump air quickly into a ball, the air gets hot because as the piston in the pump moves the particles bounce off increasing kinetic energy, which causes a temperature rise.

Reducing the volume of a fixed mass of gas increases the pressure.

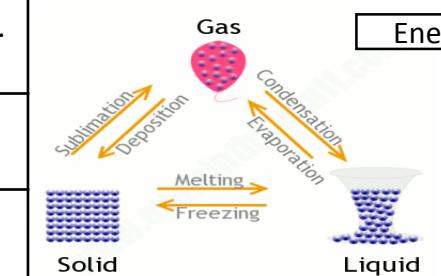
$$PV = \text{constant.}$$

$$P_1 V_1 = P_2 V_2$$

Halving the volume doubles the pressure.

## Change of state

### Internal energy and energy transfers



$$\text{Energy needed} = \text{mass} \times \text{specific latent heat.}$$

$$\Delta E = m \times L$$

	Units
Density	Kilograms per metre cubed ( $\text{kg/m}^3$ )
Mass	Kilograms (kg)
Volume	Metres cubed ( $\text{m}^3$ )
Energy needed	Joules (J)
Specific latent heat	Joule per kilogram ( $\text{J/kg}$ )
Change in thermal energy	Joules (J)
Specific heat capacity	Joule per kilogram degrees Celsius ( $\text{J/kg}^\circ\text{C}$ )
Temperature change	Degrees Celsius ( $^\circ\text{C}$ )
Pressure	Pascals (Pa)

Specific Heat Capacity

**Energy needed to raise 1kg of substance by  $1^\circ\text{C}$**

- Depends on:
- Mass of substance
  - What the substance is
  - Energy put into the system.

Change in thermal energy = mass  $\times$  specific heat capacity  $\times$  temperature change.

$$\Delta E = m \times c \times \Delta \theta$$

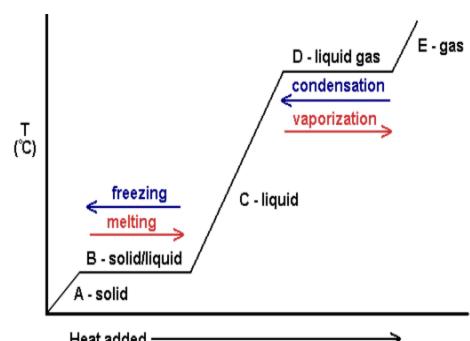
Internal energy

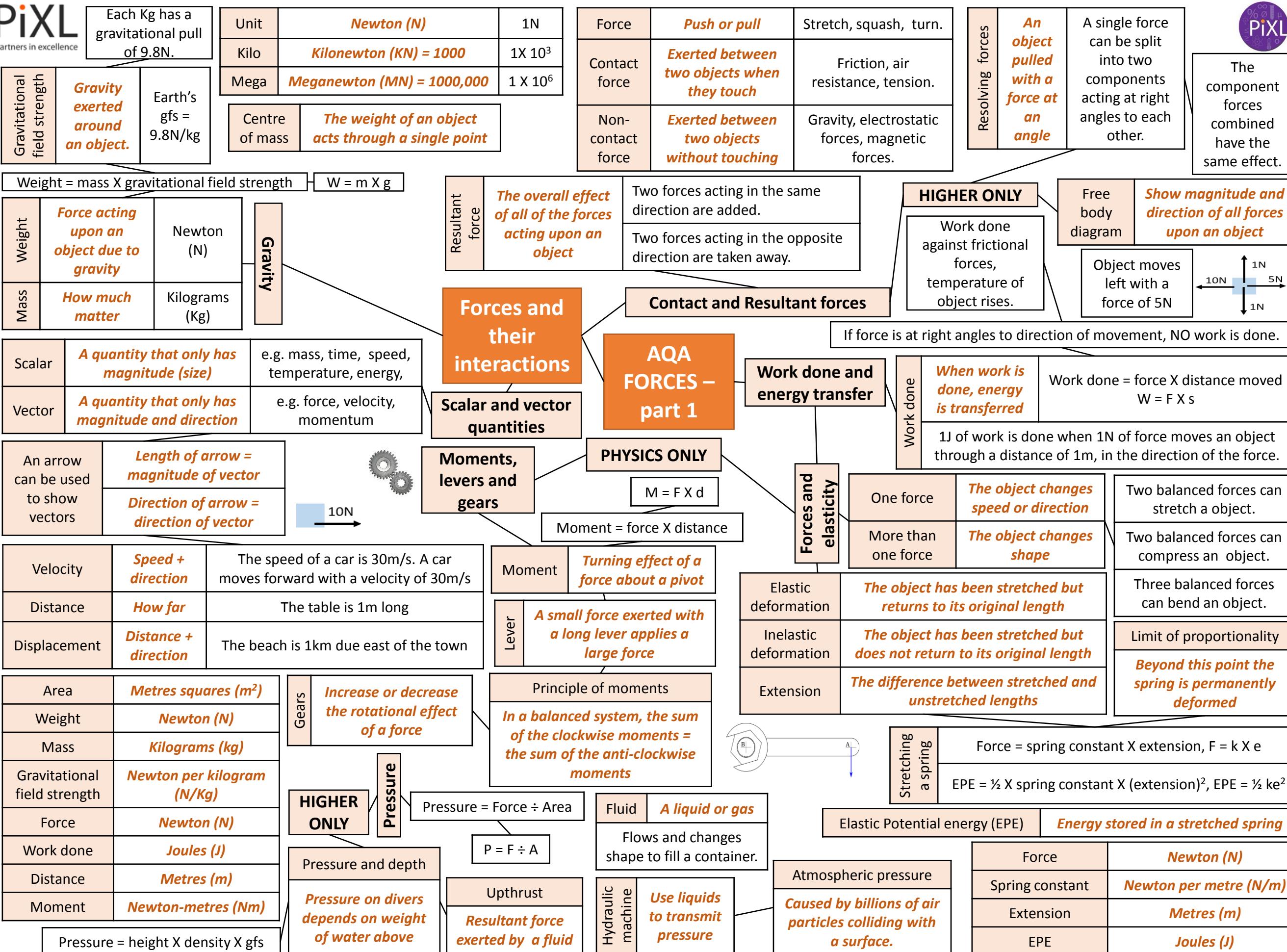
**Energy stored inside a system by particles**

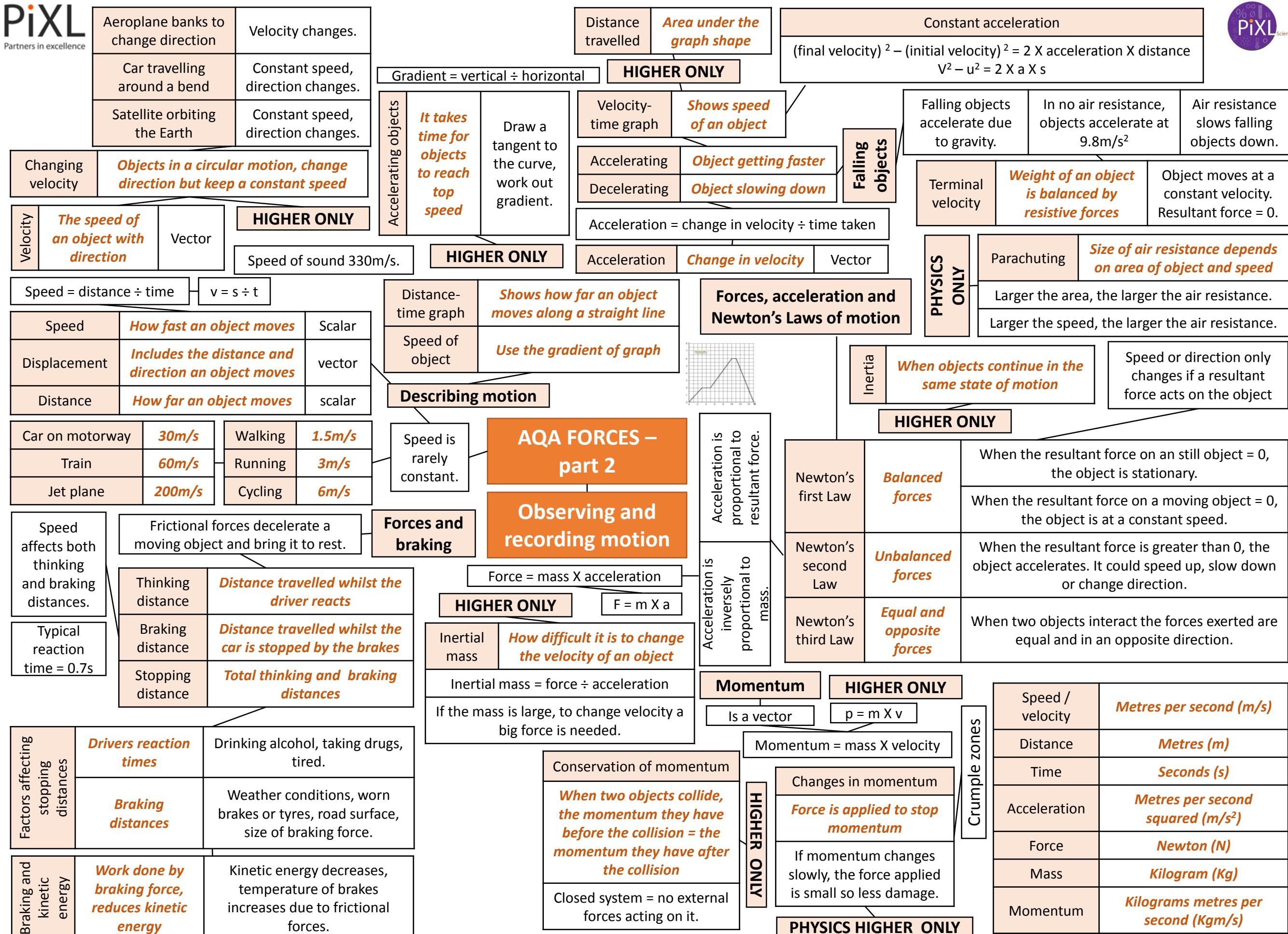
Internal energy is the total kinetic and potential energy of all the particles (atoms and molecules) in a system.

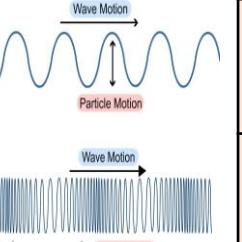
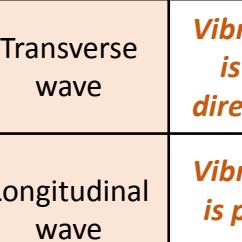
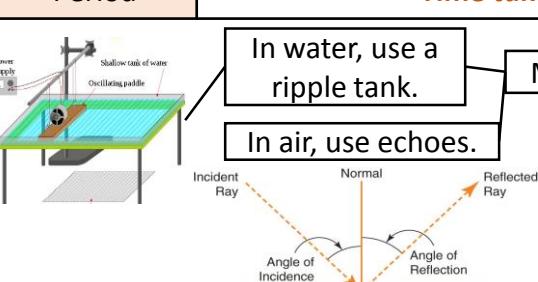
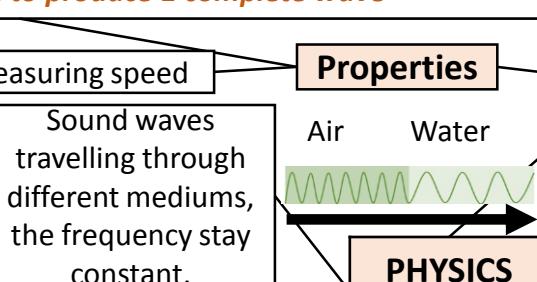
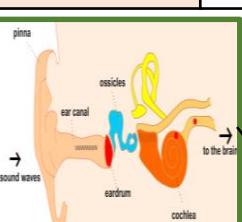
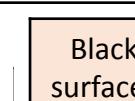
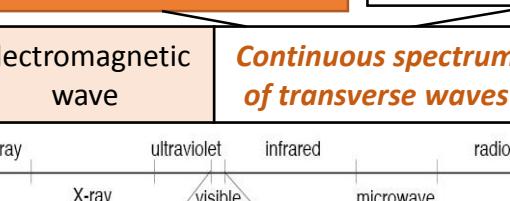
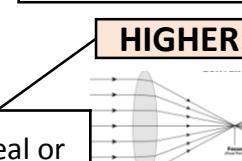
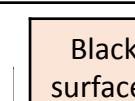
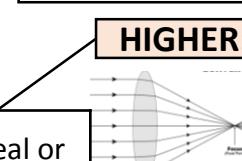
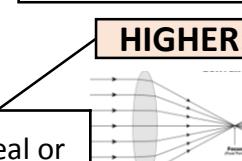
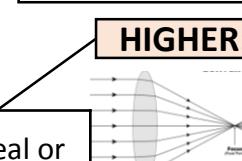
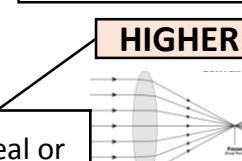
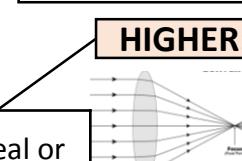
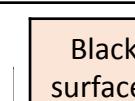
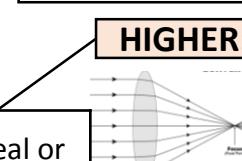
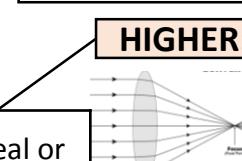
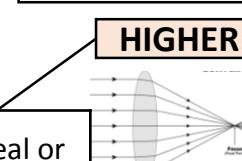
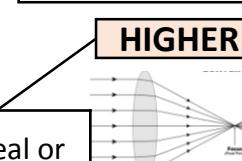
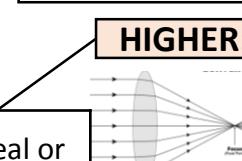
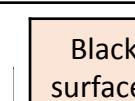
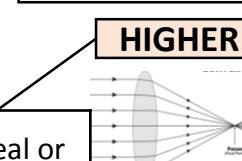
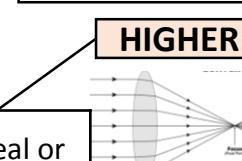
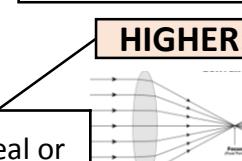
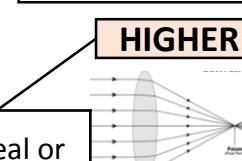
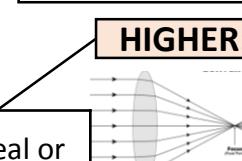
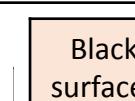
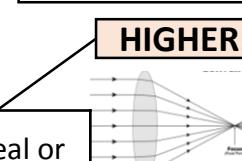
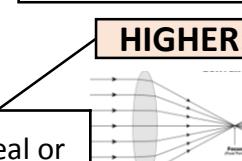
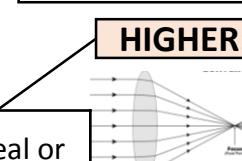
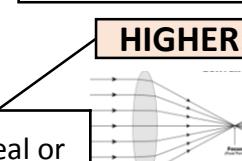
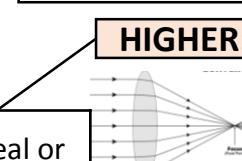
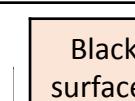
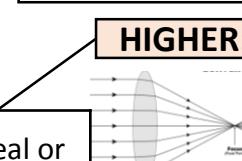
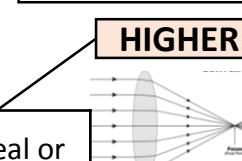
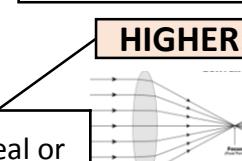
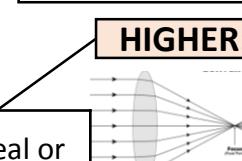
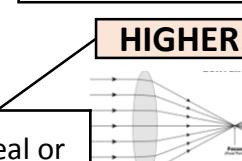
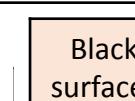
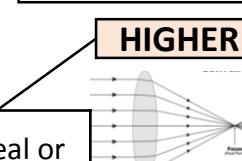
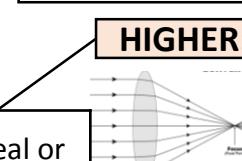
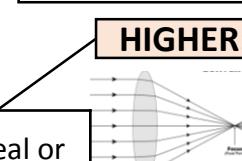
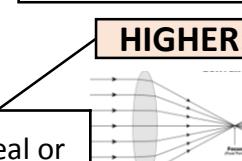
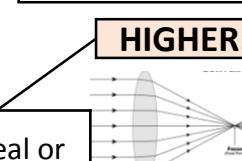
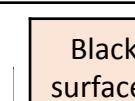
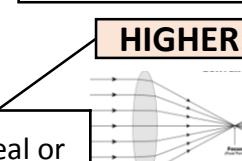
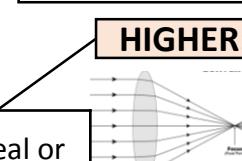
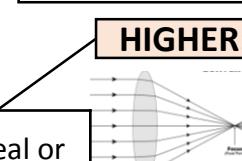
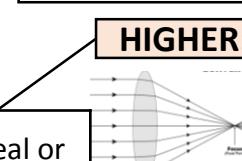
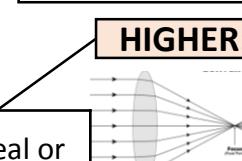
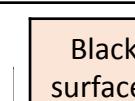
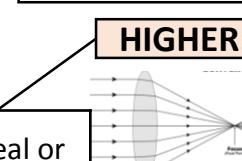
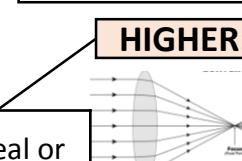
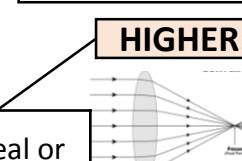
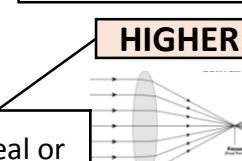
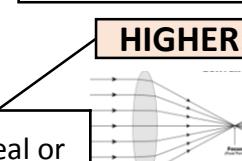
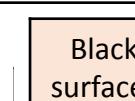
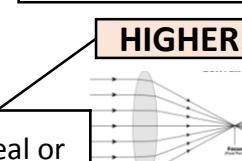
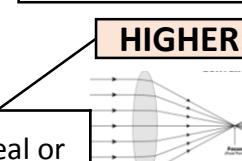
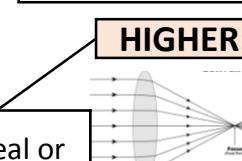
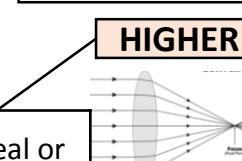
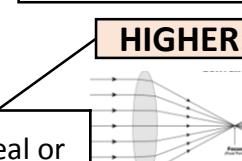
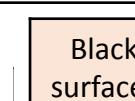
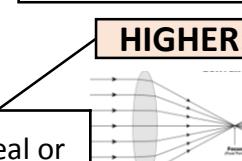
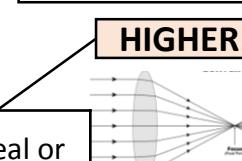
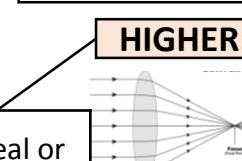
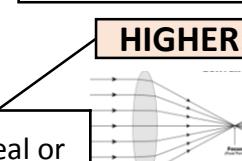
**Heating changes the energy stored within a system**

Heating causes a change in state. As particles separate, potential energy stored increases. Heating increases the temperature of a system. Particles move faster so kinetic energy of particles increases.







Wave speed	Wave speed = frequency X wavelength	$V = f \times \lambda$		Transverse wave	<b>Vibration causing the wave is at right angles to the direction of energy transfer</b>	Energy is carried outwards by the wave.	Water and light waves, S waves.
Wave period	Wave period = $1 \div$ frequency	$T = 1 \div f$		Longitudinal wave	<b>Vibration causing the wave is parallel to the direction of energy transfer</b>	Energy is carried along the wave.	Sound waves, P waves.
Wavelength	<b>Distance from one point on a wave to the same point of the next wave</b>						
Amplitude	<b>The maximum disturbance from its rest position</b>						
Frequency	<b>Number of waves per second</b>						
Period	<b>Time taken to produce 1 complete wave</b>						
 <p>Measuring speed</p> <p>Air Water</p> <p>Air Water</p> <p>Properties</p> <p>Sound waves travelling through different mediums, the frequency stay constant.</p>				<b>Transverse and Longitudinal waves</b>			
 <p>Angle of incidence = angle of reflection (i) = (r)</p>				<b>Waves in air, fluids and solids</b>			
Reflection	Wave bounces off the surface.			<b>AQA Waves</b>			
Refraction	Waves changes direction at boundary.			<b>Black body radiation</b>			
Transmitted	Passes through the object.			<b>Earth and Global warming</b>			
Absorbed	Passes into but not out of, transfers energy and heats up the object.			Ultraviolet, visible light, infra-red radiation penetrate atmosphere and heat up Earth's surface. Longer wavelengths are radiated back, trapped by atmosphere.			
 <p><b>PHYSICS HIGHER ONLY</b></p> <p>Hearing Frequencies between 20 – 20,000 Hz</p> <p>Longitudinal waves cause ear drum to vibrate, amplified by three ossicles which creates pressure in the cochlea.</p>				<b>PHYSICS ONLY</b>			
P wave	S wave	Seismograph		<b>Electromagnetic waves</b>			
Longitudinal	Transverse	<b>Shows P and S waves arriving at different times.</b>		 <p>Electromagnetic wave</p> <p>Continuous spectrum of transverse waves</p>			
Fast	Slow			<b>PHYSICS ONLY</b>			
Travel through solids and liquids	Travels through solids	By using the times the waves arrive at the monitoring centres, the epicentre of earthquake can be found. ( $v = x \div t$ ).		 <p>Magnification = image size <math>\div</math> object size</p> <p><b>HIGHER: Lenses</b></p> <p><b>HIGHER: Properties</b></p> <p>Real or virtual images.</p>			
 <p><b>Seismic waves</b></p>				<p>Black surfaces Good emitters, good absorbers</p> <p>White surfaces Poor emitters, poor absorbers</p> <p>Shiny surfaces Good reflectors</p> <p>EM waves refract</p>			
P wave	S wave	Seismograph		 <p>Magnification = image size <math>\div</math> object size</p> <p><b>HIGHER: Lenses</b></p> <p><b>HIGHER: Properties</b></p> <p>Real or virtual images.</p>			
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**AQA MAGNETISM AND ELECTROMAGNETISM**

- Permanent and Induced Magnetism**
  - Magnets**

Magnetic	<b>Materials attracted by magnets</b>	Uses non-contact force to attract magnetic materials.
North seeking pole	<b>End of magnet pointing north</b>	Compass needle is a bar magnet and points north.
South seeking pole	<b>End of magnet pointing south</b>	Like poles (N – N) repel, unlike poles (N – S) attract.
Magnetic field	<b>Region of force around magnet</b>	Strong field, force big. Weak field, force small. Field is strongest at the poles.
Permanent	<b>A magnet that produces its own magnetic field</b>	Will repel or attract other magnets and magnetic materials.
Induced	<b>A temporary magnet</b>	Becomes magnet when placed in a magnetic field.
- Magnetic field around a wire**
  - Right hand rule**
  - Fleming's left-hand rule**
  - Generators**
  - Electric motor**
  - Relay**
  - Loud speakers**
  - Microphones**
  - Split-ring commutator**
  - Electromagnet**
  - Solenoid**
- Induced potential, transformers and National Grid**
  - National Grid**
  - Transformer**
    - Step-up transformers**
    - Step-down transformers**
  - PHYSICS HIGHER only**
- Electromagnetism**
  - Relay**
  - Electromagnet**
  - Solenoid**
  - Generators**
  - Electric motor**
  - Microphones**
  - Loud speakers**
  - Split-ring commutator**
  - Relay**
  - Thumb**
  - First finger**
  - Second finger**
  - Fleming's left-hand rule**
  - Right hand rule**
  - HIGHER only**
  - Induced potential**
  - Generator effect**
  - Uses of the generator effect**
  - Force**
  - Magnetic flux density**
  - Current**
  - Length**
  - Power**
  - p.d.**
  - Newton (N)**
  - Tesla (T)**
  - Amperes (A)**
  - Metres (m)**
  - Watts (W)**
  - Voltage (V)**



Milky Way  
our galaxy.

Planet	A large body orbiting the Sun
Moon	A natural satellite orbiting a planet
Dwarf planet	A body large enough to have its own gravity which caused a spherical shape
Solar system	Any object orbiting the Sun due to gravity
Galaxy	Collection of billions of stars
Universe	Collection of galaxies



Comets, asteroids, satellites.  
Other objects.

## Solar system

### The life cycle of a star.

Nebula	A cloud of cold hydrogen gas and dust	Cloud collapses due to gravity, particles move very fast colliding with each other, kinetic energy transfers into internal energy and the temperature increases.
Protostar	The large ball of gas contracts to form a star	High temperature causes Hydrogen nuclei to collide and nuclear fusion begins. A star is 'born'.
Main sequence	Stable period of star	Gravity tries to collapse the star but enormous pressure of fusion energy expands and balances the inward force.

## AQA SPACE PHYSICS PHYSICS ONLY

### Red shift

#### Understanding models.

##### Red-shift

The observed increase in wavelength of light from most distance galaxies. Light moves towards the red end of the spectrum.

##### Hubble

He studied light from distant galaxies; found as frequency decreases, wavelength increases.



Light from star in our galaxy.

Light from star in nearby galaxy.

Light from star in distant galaxy.

##### The Big Bang

Universe began 13.8 billion years ago

All matter and space expanded violently from a single point.

Red-shift provides evidence for expansion.

##### Aristotle

Earth at the centre, other heavenly bodies move around the Earth.

##### Copernicus

Sun at the centre, other heavenly bodies move around the Sun.

##### Galileo

Made a telescope, looked at Jupiter, found four moons rotating around planet.

#### Effect of gravity.

Gravity causes moons to orbit planets, planets to orbit the Sun, stars to orbit galaxy centres.

Force of gravity changes the moon's direction not its speed.

Gravity pulls objects towards the ground.

#### Speed of Orbit.

To calculate speed of Orbit: distance object moves in 1 orbit, Distance =  $2\pi r$ , then average speed = distance ÷ time.

Planets close to the Sun, gravity pull is strong. Planets move quickly.

Planets further away from the Sun, gravity pull is weaker. So speed of planet is slower.

When ambulances go past the sound changes from a high pitch to a low pitch.

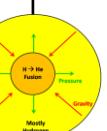
Frequency of sound wave decreases, wavelength increases.

Galaxies are moving away from us in all directions.

Light from distant galaxies is red-shifted, so galaxy is moving away from us.

Galaxies further away have bigger red-shift so are moving faster away.

### Stars the same size as our Sun.



Red giant	A large star that fuses Helium into heavier elements	Hydrogen runs out, star becomes unstable, pressure inside drops causing star to collapse. Atoms now closer together results in atoms fusing and temperature increases. This increase in temperature causes the core to swell.
White dwarf	Star collapses	Nuclear fuel runs out, fusion stops, dense very hot core.
Black dwarf	Cold dark star	White dwarf cools down.

### Stars larger than our Sun.

Red super giant	Star swells greatly	Nuclear fuel begins to run out and star swells (more matter = bigger size).
Supernova	Gigantic explosion due to run away fusion reactions	Rapid collapse, heats to very high temperatures causing run away nuclear reactions, star explodes, flinging remnants out into space. Large gravitational forces collapse the core into a tiny space. Remains of supernova form heavier elements (Iron and above)
Neutron star	Very dense star	Made out of neutrons.

OR if collapse is into a really tiny space.

Black hole

No light escapes

Gravitational forces so strong everything is pulled in.

