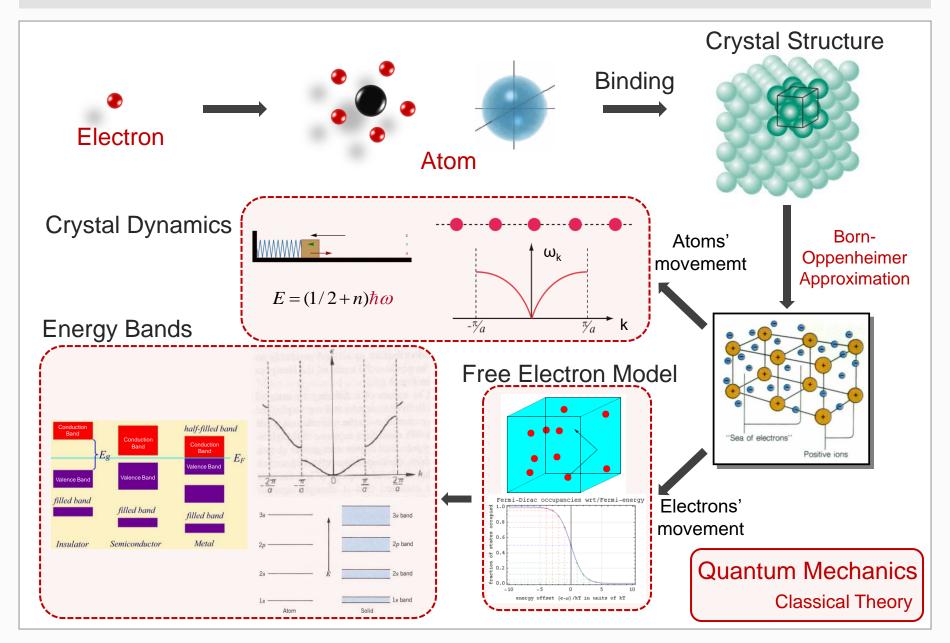
Chapter 3

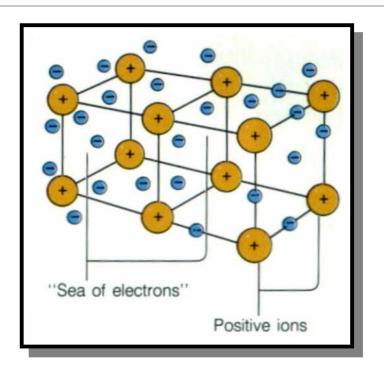
Free electrons in solids

Today's lecture

Profile



Introduction



Crystal dynamics

Free electron theory

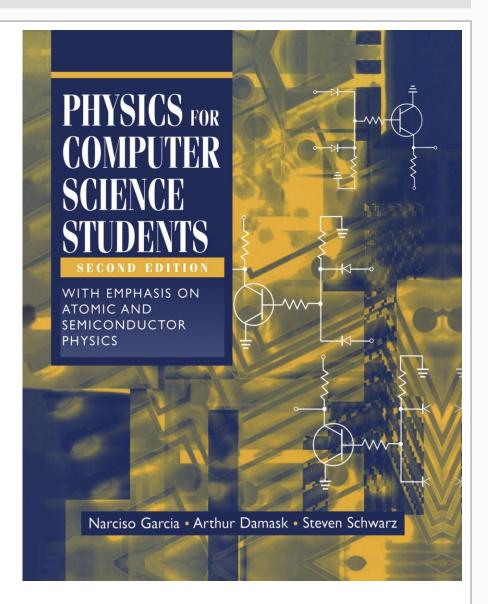
Energy band theory

theory

Introduction

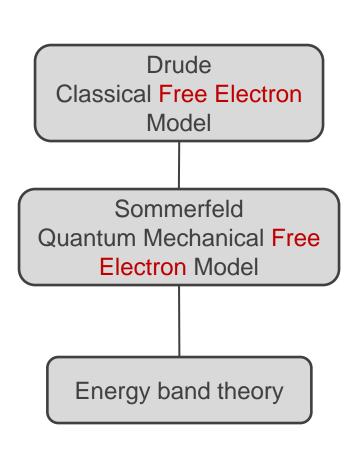
Acknowledgement:

《Physics for computer science students: with emphasis on atomic and semiconductor physics》



- 3.1 Free electron model
 - 3.1.1 Drude Model
 - Classical Free Electron Model
 - 3.1.2 Sommerfeld Model
 - Quantum Mechanical Free Electron Model
- 3.2 Heat capacity of free electron gas
- 3.3 Transport properties of conductive electrons

Introduction



Independent electron approximation
Free electron approximation
Collision assumption
Relaxation time approximation

Independent electron approximation Free electron approximation

No collision

Quantum statistics: Fermi-Dirac Distribution

Free Electron Model

+

Periodical potential field

- 3.1 Free electron model
 - 3.1.1 Drude Model
 - Classical Free Electron Model
 - 3.1.2 Sommerfeld Model
 - Quantum Mechanical Free Electron Model
- 3.2 Heat capacity of free electron gas
- 3.3 Transport properties of conductive electrons
- 3.4 Electron emission and contacting voltage

Drude Model Failure of Drude Model

Ohm's Law

$$U = IR$$
 $J = \sigma \varepsilon$

J: Current density

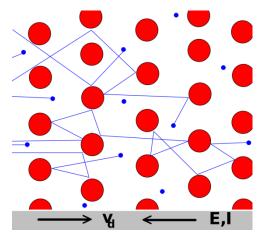
ε: Electric field intensity

σ: Conductivity

$$\sigma = \frac{1}{\rho}$$
 ρ : Resistivity



Paul Karl Ludwig Drude



Drude Model electrons

The Drude model of electrical conduction was proposed in 1900 by Paul Drude to explain the transport properties of electrons in materials (especially metals). The model, which is an application of kinetic theory, assumes that the microscopic behavior of electrons in a solid may be treated classically and looks much like a pinball machine, with a sea of constantly jittering electrons bouncing and re-bouncing off heavier, relatively immobile positive ions.

Drude Model Failure of Drude Model

Assumption of the electron gas:

(1) Independent electron approximation

No electrostatic interaction and collision among free electrons

(2) Free electron approximation

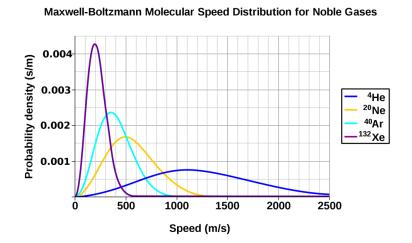
No electrostatic interaction between free electrons and ions

(3) Collision assumption

Velocity of electrons after collision with ions only concerns with temperature, but not the velocity before collision

(4) Relaxation time approximation

Relaxation time τ is independent with the position and velocity of electrons



Free electron gas

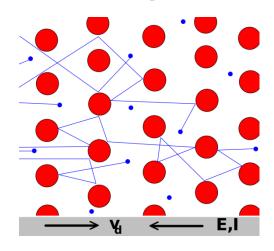
Drude Model - Failure of Drude Model

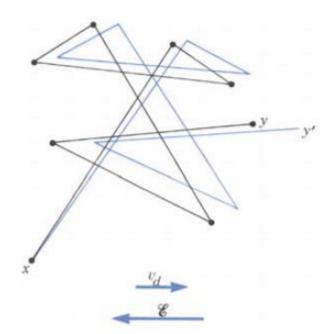
Without electric field, free e- moves randomly

At room temperature,
$$v_{RMS} = (3k_BT/m)^{1/2} = 1.2 \times 10^5 \,\mathrm{m} \cdot \mathrm{sec}^{-1} (\sim 120 \,\mathrm{km} \cdot \mathrm{sec}^{-1})$$

No net current

With electric field, things are different

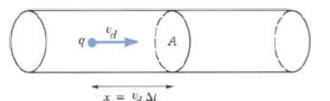




v_d: drift velocity (漂移速度), v_d<<v_{RMS} (a net movement)

Drude Model - Failure of Drude Model

Current density: the electric quantity per unit area of cross section per second



$$J = \frac{n(-e)(\mathbf{v_d}\Delta t) \cdot A}{\Delta t \cdot A} = -ne\mathbf{v_d} \quad \text{vs} \quad J = \sigma \varepsilon$$

v_d must be proportional to electric field intensity

τ: average time between collision

$$\mathbf{v_d} = \mathbf{v_0} + \mathbf{a}t = \frac{F}{m}t = -\frac{e\varepsilon}{m}\tau$$
 $(F = ma = e\varepsilon)$ (suppose $\mathbf{v_0} = 0$)

$$J = -ne\mathbf{v_d} = -ne \cdot -\frac{e\varepsilon}{m}\tau = \sigma\varepsilon \qquad \sigma = \frac{ne^2}{m}\tau$$

Prove: τ independent of ϵ

$$\tau = mean free path/v$$

mean free path: average distance traveled between collisions

$$v = v_{RMS} + v_d \approx v_{RMS} \quad (v_d \ll v_{RMS})$$

$$\tau = mean \ free \ path/v_{RMS}$$
 τ independent of ε

- Drude Model - Failure of Drude Model

(1) Specific Heat

Equipartition theorem of energy: in thermal equilibrium, energy is shared equally among all of its various forms. For example, the average kinetic energy per degree of freedom in the translational motion of a molecule should equal that of its rotational motions.

$$\overline{\varepsilon} = \frac{1}{2}(t + r + 2s)k_BT$$
 t, r, s: free degree of translation, rotation and vibration

For solids,

t=0, r=0, s=3

$$\overline{\varepsilon} = \frac{1}{2}(t + r + 2s)k_B T = 3k_B T$$

Total energy of 1mol solid state matter

$$E=N\overline{\varepsilon}=3Nk_BT$$

$$C_v^{mol} = \frac{\partial E}{\partial T} = 3Nk_B = 3R$$

Dulong-Petit Law

For monoatomic gas (electron gas) t=3, r=s=0

$$\overline{\varepsilon} = \frac{1}{2}(t+r+2s)k_B T = \frac{3}{2}k_B T$$

Total energy of 1mol monoatomic gas (electron gas)

$$E = N\overline{\varepsilon} = \frac{3}{2}Nk_BT$$

$$C_v^{mol} = \frac{\partial E}{\partial T} = \frac{3}{2}Nk_B = \frac{3}{2}R$$

- Drude Model - Failure of Drude Model

(1) Specific Heat

From Drude Model: $C_v^{mol}(from e^-) = 3/2 R$

Experimentally: $C_v^{mol}(from e^-) = 10^{-4} RT = 0.03R$ (If T=300K)

Classical FEM: all free electrons contribute to heat capacity.

Disaster! Heat capacity is much less than predicted. Free e- gives such a small contribution to C_{ν} (free e- have poor ability to absorb heat)

(2) Temperature dependence of σ

From Drude Model: $\sigma = \frac{ne^2}{m}\tau$ τ is temperature dependent

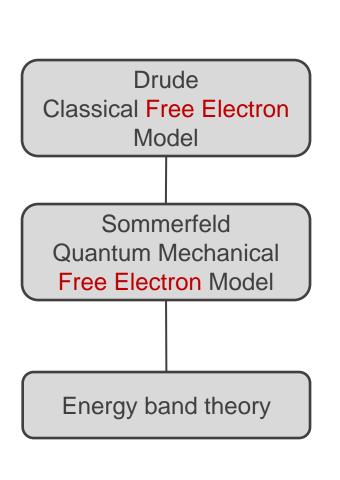
Experimentally: $\sigma \propto T^{-1}$

Drude Model Failure of Drude Model

Summary of Drude Theory

- Based on kinetic theory of gases.
- Successes
 - Ohm's Law.
 - Wiedemann-Franz ratio comes out close to right for most materials
 - Many other transport properties predicted correctly
- Failures
 - Specific Heat of electrons
 - Temperature dependence of σ
 - Wiedemann-Franz ratio
 - The Seebeck/Peltier coefficient come out wrong by a factor of 100.
- Despite the shortcomings of Drude theory, it nonetheless was the only theory of metallic conductivity for a quarter of a century (until the Sommerfeld theory improved it), and it remains quite useful today.

- 3.1 Free electron model
 - 3.1.1 Drude Model
 - Classical Free Electron Model
 - 3.1.2 Sommerfeld Model
 - Quantum Mechanical Free Electron Model
- 3.2 Heat capacity of free electron gas
- 3.3 Transport properties of conductive electrons
- 3.4 Electron emission and contacting voltage



Independent electron approximation
Free electron approximation
Collision assumption
Relaxation time approximation

Independent electron approximation Free electron approximation

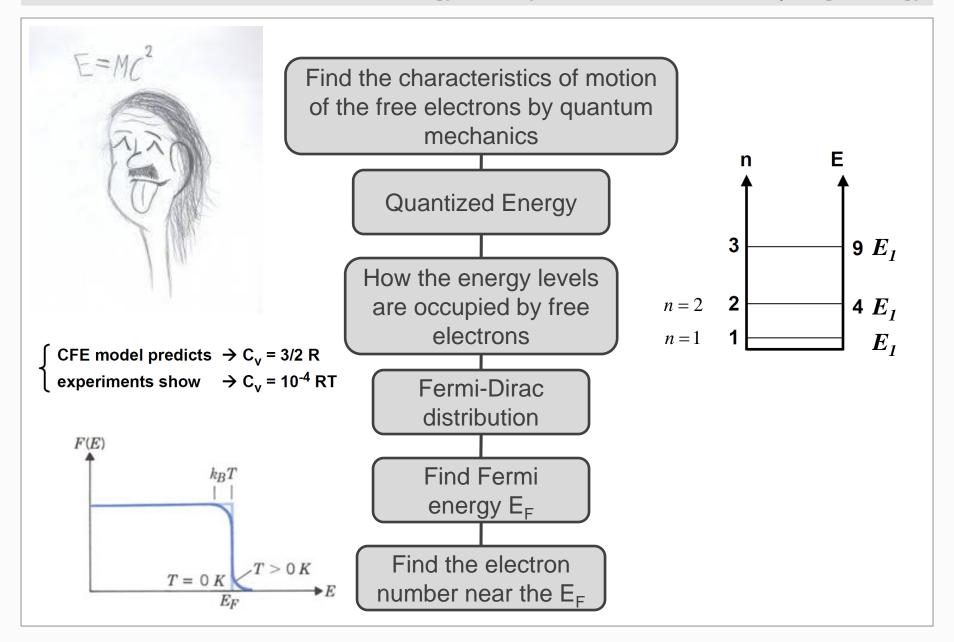
No collision

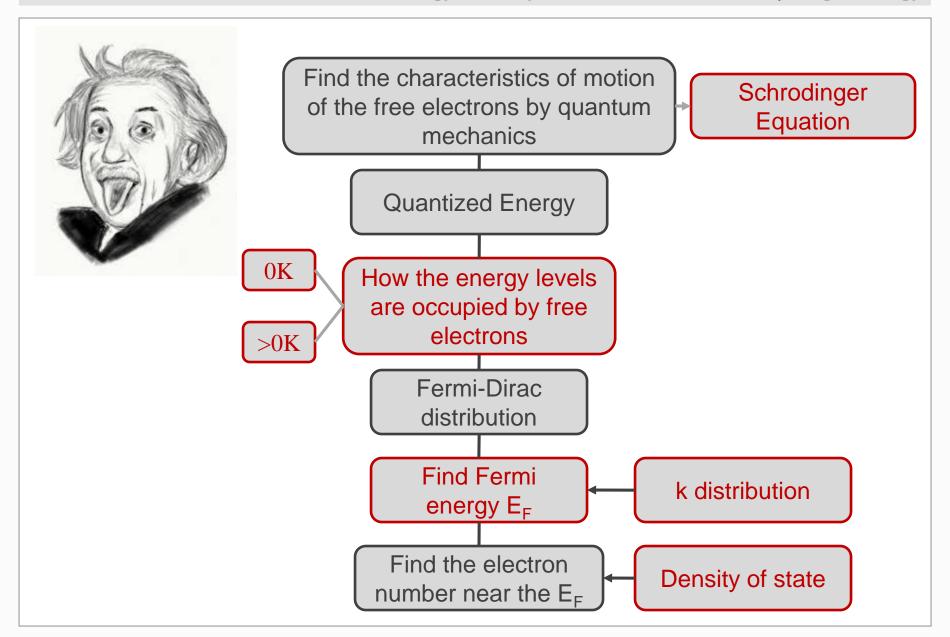
Quantum statistics: Fermi-Dirac Distribution

Free Electron Model

+

Periodical potential field

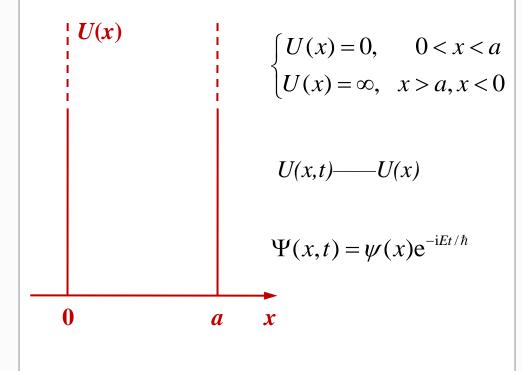




- Potential Well - F-D Distribution - Fermi Energy - Density of State - e⁻ number near E_F - e⁻ gas Energy

Infinite Potential Well

$$\left(-\frac{\hbar^2}{2m}\frac{\mathrm{d}^2}{\mathrm{d}x^2} + U(x)\right)\psi(x) = E\psi(x)$$



$$-\frac{\hbar^2}{2m} \frac{d^2 \psi}{dx^2} = E\psi$$

$$\frac{d^2 \psi}{dx^2} + \frac{2mE}{\hbar^2} \psi = 0$$

$$k = \frac{\sqrt{2mE}}{\hbar}$$

$$\frac{d^2 \psi}{dx^2} + k^2 \psi = 0$$

$$\psi = A \sin kx + B \cos kx$$

$$E = \frac{(\hbar k)^2}{2m}$$

$$\psi(x) = 0, \quad x > a, x < 0$$

$$\psi(0) = B \cos 0 = B = 0$$

$$\psi(a) = A \sin ka + B \cos ka = 0$$

$$B = 0$$

$$A \sin ka = 0$$

• Potential Well • F-D Distribution • Fermi Energy • Density of State • e number near E_F • e gas Energy

$$\frac{d^{2}\psi}{dx^{2}} + \frac{2mE}{\hbar^{2}}\psi = 0$$

$$k = \frac{\sqrt{2mE}}{\hbar}$$

$$\psi = A\sin kx + B\cos kx$$

$$A\sin ka = 0$$

$$\sin ka = 0$$

$$ka = n\pi \quad k = n\frac{\pi}{a}$$

$$n = 1, 2, 3, ...$$

$$\frac{d^2\psi}{dx^2} + \frac{2mE}{\hbar^2}\psi = 0$$

$$\psi = A\sin kx + B\cos kx$$

$$A\sin ka = 0$$

$$\sin ka = 0$$

$$ka = n\pi$$

$$k = n\frac{\pi}{a}$$

$$m = 1, 2, 3$$

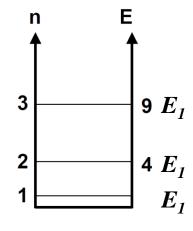
$$\psi(x) = A\sin\frac{n\pi}{a}x \qquad 0 \le x \le a$$

$$\int_0^a |\psi|^2 dx = 1 \qquad A = \sqrt{\frac{2}{a}}$$

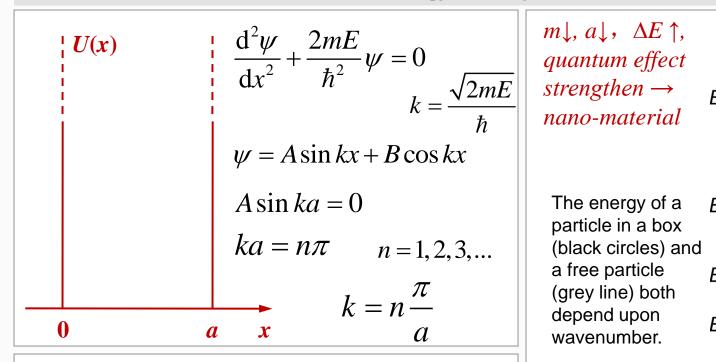
$$\psi(x) = \sqrt{\frac{2}{a}}\sin\frac{n\pi}{a}x$$

$$|\psi_n|^2 = \frac{2}{a}\sin^2\frac{n\pi}{a}x$$

$$n = 1, 2, 3, ...$$



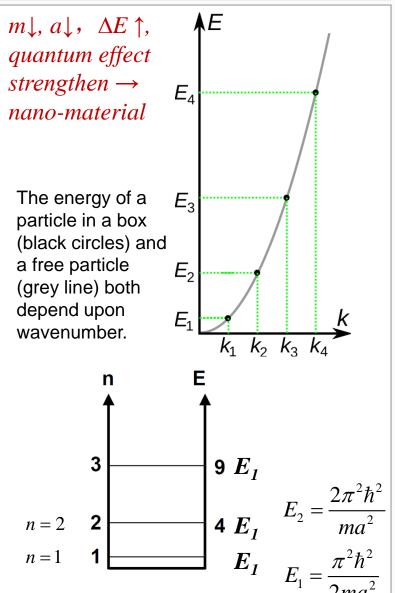
- Potential Well - F-D Distribution - Fermi Energy - Density of State - e⁻ number near E_F - e⁻ gas Energy



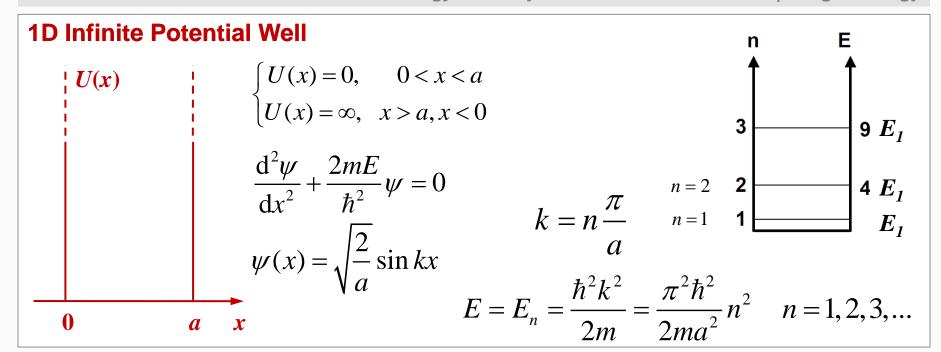
$$E = \frac{(\hbar k)^2}{2m}$$

$$E = E_n = \frac{\pi^2 \hbar^2}{2ma^2} n^2$$

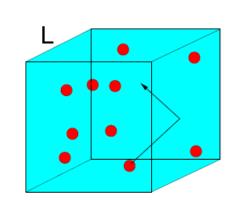
$$n = 1, 2, 3, \dots$$



• Potential Well • F-D Distribution • Fermi Energy • Density of State • e number near E_F • e gas Energy



3D Infinite Potential Well

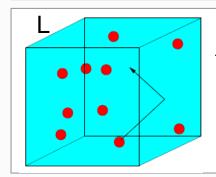


ite Potential Well
$$\begin{cases}
U = 0 & 0 \le x, y, z \le L \\
U = \infty & x, y, z > L, x, y, z < 0
\end{cases} \left(-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + U(x) \right) \psi(x) = E \psi(x)$$

$$-\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) \psi(x, y, z) = E \psi(x, y, z)$$

$$\psi(x, y, z) = \psi(x) \psi(y) \psi(z)$$

- Potential Well - F-D Distribution - Fermi Energy - Density of State - e⁻ number near E_F - e⁻ gas Energy



$$\begin{cases} U = 0 & 0 \le x, y, z \le L \\ U = \infty & x, y, z > L, x, y, z < 0 \end{cases} - \frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) \psi(x, y, z) = E\psi(x, y, z) \\ - \frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) \psi(x) \psi(y) \psi(z) = E\psi(x) \psi(y) \psi(z) \end{cases}$$

$$\left(\frac{1}{\psi(x)}\frac{\partial^{2}\psi(x)}{\partial x^{2}} + \frac{1}{\psi(y)}\frac{\partial^{2}\psi(y)}{\partial y^{2}} + \frac{1}{\psi(z)}\frac{\partial^{2}\psi(z)}{\partial z^{2}}\right) + \frac{2m}{\hbar^{2}}E = 0$$

$$E = \frac{\hbar^{2}k^{2}}{2m}$$

$$\vec{k} = k_{x}\vec{i} + k_{y}\vec{j} + k_{z}\vec{l}$$

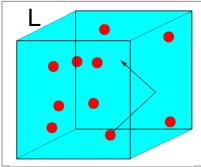
$$E = \frac{\hbar^{2}}{2m}(k_{x}^{2} + k_{y}^{2} + k_{z}^{2})$$

$$\left(\frac{1}{\psi(x)}\frac{\partial^{2}\psi(x)}{\partial x^{2}} + \frac{1}{\psi(y)}\frac{\partial^{2}\psi(y)}{\partial y^{2}} + \frac{1}{\psi(z)}\frac{\partial^{2}\psi(z)}{\partial z^{2}}\right) + (k_{x}^{2} + k_{y}^{2} + k_{z}^{2}) = 0$$

$$\left(\frac{1}{\psi(x)}\frac{\partial^{2}\psi(x)}{\partial x^{2}} + \frac{1}{\psi(y)}\frac{\partial^{2}\psi(y)}{\partial y^{2}} + \frac{1}{\psi(z)}\frac{\partial^{2}\psi(z)}{\partial z^{2}}\right) + (k_{x}^{2} + k_{y}^{2} + k_{z}^{2}) = 0$$

$$\begin{cases} \frac{1}{\psi(x)} \frac{\partial^2 \psi(x)}{\partial x^2} = -k_x^2 \\ \frac{1}{\psi(y)} \frac{\partial^2 \psi(y)}{\partial y^2} = -k_y^2 \\ \frac{1}{\psi(z)} \frac{\partial^2 \psi(z)}{\partial z^2} = -k_z^2 \\ \frac{\partial^2 \psi(x)}{\partial x^2} + k_x^2 \psi(x) = 0 \\ \frac{\partial^2 \psi(y)}{\partial y^2} + k_y^2 \psi(y) = 0 \\ \frac{\partial^2 \psi(z)}{\partial z^2} + k_z^2 \psi(z) = 0 \end{cases}$$

· Potential Well · F-D Distribution · Fermi Energy · Density of State · e⁻ number near E_F · e⁻ gas Energy



$$x = 0, \psi(x) = 0; x = L, \psi(x) = 0$$

$$y = 0, \psi(y) = 0; y = L, \psi(y) = 0$$

$$z = 0, \psi(z) = 0; z = L, \psi(z) = 0$$

$$-\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) \psi(x, y, z) = E\psi(x, y, z)$$

$$\frac{\partial^{2}\psi(x)}{\partial x^{2}} + k_{x}^{2}\psi(x) = 0$$

$$\frac{\partial^{2}\psi(y)}{\partial y^{2}} + k_{y}^{2}\psi(y) = 0$$

$$\frac{\partial^{2}\psi(y)}{\partial y^{2}} + k_{z}^{2}\psi(z) = 0$$

$$\psi(x) = A_{x} \sin k_{x}x$$

$$k_{x} = \frac{n_{x}\pi}{L}$$

$$m_{x} = 1, 2, 3, ...$$

$$\psi(y) = A_{y} \sin k_{y}y$$

$$k_{y} = \frac{n_{y}\pi}{L}$$

$$n_{y} = 1, 2, 3, ...$$

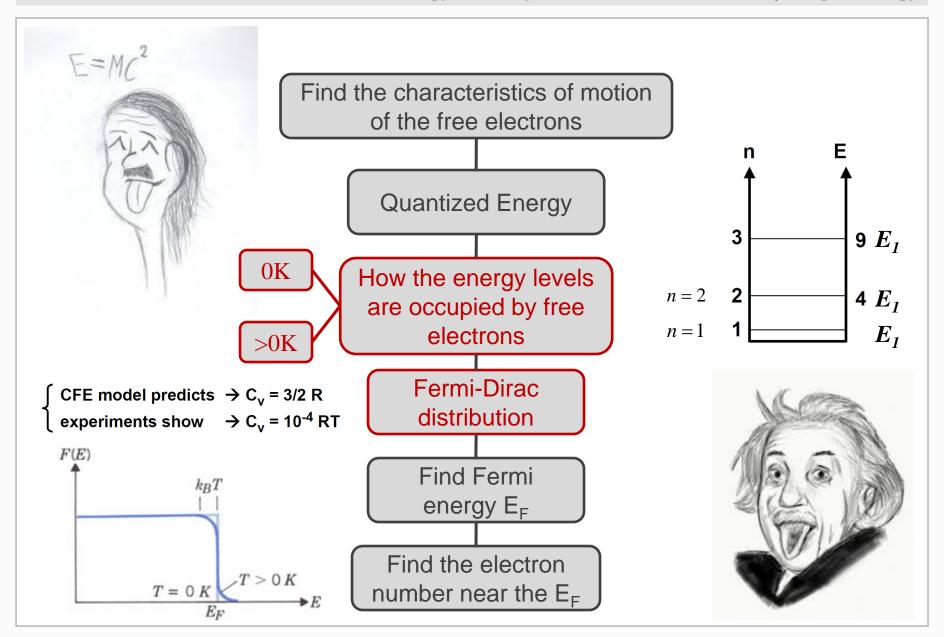
$$\psi(z) = A_{z} \sin k_{z}z$$

$$k_{z} = \frac{n_{z}\pi}{L}$$

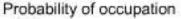
$$n_{z} = 1, 2, 3, ...$$

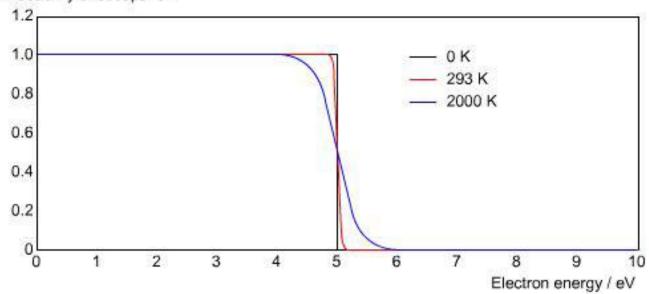
$$E = \frac{\hbar^2 k^2}{2m} = \frac{\hbar^2}{2m} (k_x^2 + k_y^2 + k_z^2) = \frac{\pi^2 \hbar^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2)$$

$$n_x, n_y, n_z = 1, 2, 3, \dots$$



Fermi-Dirac Distribution





Fermi-Dirac distribution for several temperatures

$$F_{l} = \frac{\alpha_{l}}{\omega_{l}} = \frac{1}{e^{(\alpha + \beta E_{l})} + 1} = \frac{1}{e^{(E_{l} - \mu)/k_{B}T} + 1} = \frac{1}{e^{(E - E_{F})/k_{B}T} + 1}$$

· Potential Well · F-D Distribution · Fermi Energy · Density of State · e⁻ number near E_F · e⁻ gas Energy

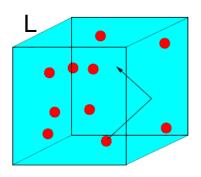
Occupancy Condition at T=0K

- Classical theory

Occupancy Condition at T=0K

Pauli's Exclusion Principle

- Quantum mechanically



$$E = \frac{\pi^2 \hbar^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2)$$

$$= E_0 (n_x^2 + n_y^2 + n_z^2)$$

$$E_0 = \frac{\pi^2 \hbar^2}{2mL^2}$$

 $n_x, n_y, n_z = 1, 2, 3, \dots$

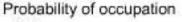
· Potential Well · F-D Distribution · Fermi Energy · Density of State · e⁻ number near E_F · e⁻ gas Energy

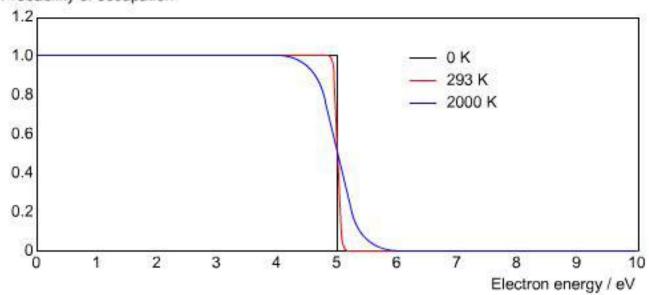
Occupancy Condition at T=0K

- Quantum mechanically

The Fermi energy: the highest energy a fermion can take at absolute zero temperature.

Fermi-Dirac Distribution





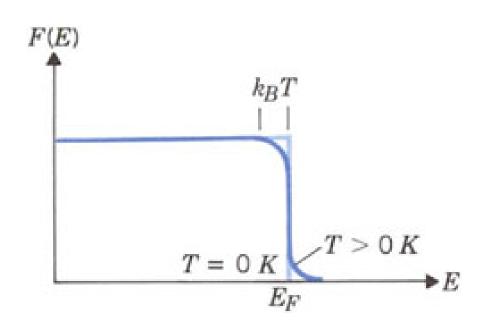
Fermi-Dirac distribution for several temperatures

$$F_{l} = \frac{\alpha_{l}}{\omega_{l}} = \frac{1}{e^{(\alpha + \beta E_{l})} + 1} = \frac{1}{e^{(E_{l} - \mu)/k_{B}T} + 1} = \frac{1}{e^{(E - E_{F})/k_{B}T} + 1}$$

Occupancy Condition at T>0K

- Quantum mechanically

$$F(E) = \frac{1}{e^{(E - E_F)/k_B T} + 1}$$

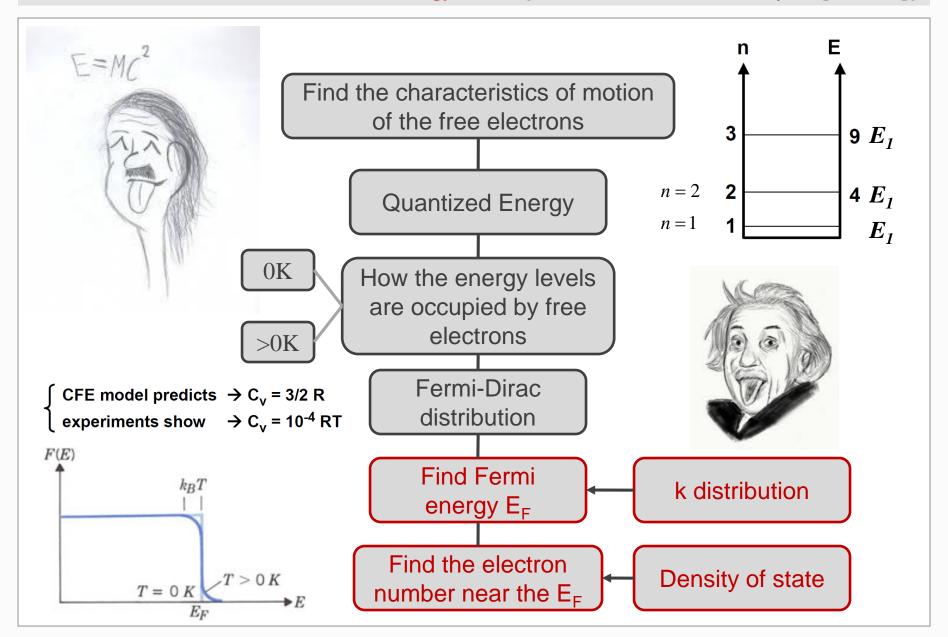


Occupancy Condition at T=0K and T>0K

- Quantum mechanically

Explain
$$\rightarrow$$
 CFE model predicts \rightarrow C_v = 3/2 R experiments show \rightarrow C_v = 10⁻⁴ RT

- 1. Fermi Energy value?
- 2. e⁻ number near E_F?



· Potential Well · F-D Distribution · Fermi Energy · Density of State · e⁻ number near E_F · e⁻ gas Energy

Calculation of Fermi Energy (E_F) at T=0K

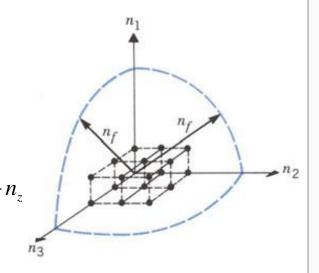
$$E = \frac{\hbar^{2}k^{2}}{2m} = \frac{\hbar^{2}}{2m}(k_{x}^{2} + k_{y}^{2} + k_{z}^{2})$$

$$\vec{k} = k_{x}\vec{i} + k_{y}\vec{j} + k_{z}\vec{l}$$

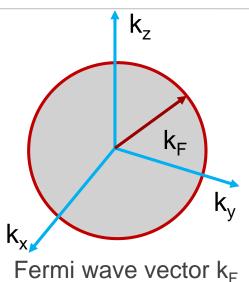
$$k_{x} = \frac{\pi}{L}n_{x}, \quad k_{y} = \frac{\pi}{L}n_{y}, \quad k_{z} = \frac{\pi}{L}n_{z}$$

$$E = (n_x^2 + n_y^2 + n_z^2)E_0 = n^2 E_0$$

$$E_0 = \frac{\pi^2 \hbar^2}{2mL^2}$$
 $n_x, n_y, n_z = 1, 2, 3, ...$



States on the same sphere are degenerate states



 $E_F = \frac{(\hbar k_F)^2}{2m}$

The radius of the Fermi sphere is called the **Fermi wave vector** k_F

Fermi sphere

mi sphere is

ve vector k_F

At T=0K,

Inside Fermi sphere, all orbits are occupied; Outside Fermi sphere, all orbits are empty.

· Potential Well · F-D Distribution · Fermi Energy · Density of State · e⁻ number near E_F · e⁻ gas Energy

Calculation of Fermi Energy (E_F) at T=0K

$$E_F = \frac{(\hbar \mathbf{k_F})^2}{2m}$$

Distribution density of $k : \rho(k)$

2D

$$s_k = \frac{2\pi}{L} \cdot \frac{2\pi}{L} = \frac{(2\pi)^2}{S}$$

$$\rho(k) = \frac{1}{s_k} = \frac{S}{4\pi^2}$$

3D

$$v_k = \frac{2\pi}{L} \cdot \frac{2\pi}{L} \cdot \frac{2\pi}{L} = \frac{(2\pi)^3}{V}$$

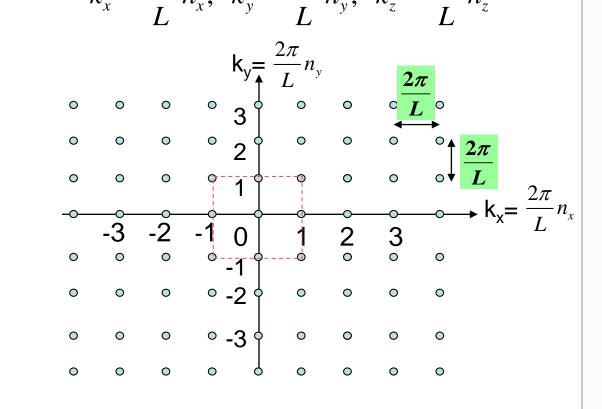
$$\rho(k) = \frac{1}{v_k} = \frac{V}{8\pi^3}$$

$$\rho(k) \cdot V_{\text{Fermi sphere}} = N(k) \qquad (number \ of \ k)$$

$$V_{\text{F}} \qquad \text{Each k} \iff 2 \text{ electrons}$$

number of electrons

$$k_{x} = \frac{2\pi}{I} n_{x}, \quad k_{y} = \frac{2\pi}{I} n_{y}, \quad k_{z} = \frac{2\pi}{I} n_{z}$$



Calculation of Fermi Energy (E_F) at T=0K

$$\rho(k) = \frac{V}{8\pi^3} \longrightarrow E_F = \frac{(\hbar k_F)^2}{2m}$$

Number of k

$$V_F \cdot \rho(k) = \left(\frac{4\pi k_F^3}{3}\right) \left(\frac{V}{8\pi^3}\right) = \frac{k_F^3}{6\pi^2} V$$

Number of e⁻ $N = 2 \cdot \frac{k_F^3}{6\pi^2} V$

Density of e⁻ $n = \frac{N}{V} = \frac{k_F^3}{3\pi^2}$

$$k_F = (3\pi^2 n)^{1/3}$$

$$E_F^0 = \frac{(\hbar k_F)^2}{2m} = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$

 $E_F^0 = 5.84*10^{-38} n^{2/3} J$

E.g.

The valance electron density n=N/V is $1.402 \times 10^{28} \text{ m}^{-3}$

Fermi-energy E_F : $E_F = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3} = 2.12 eV$

Comparison:

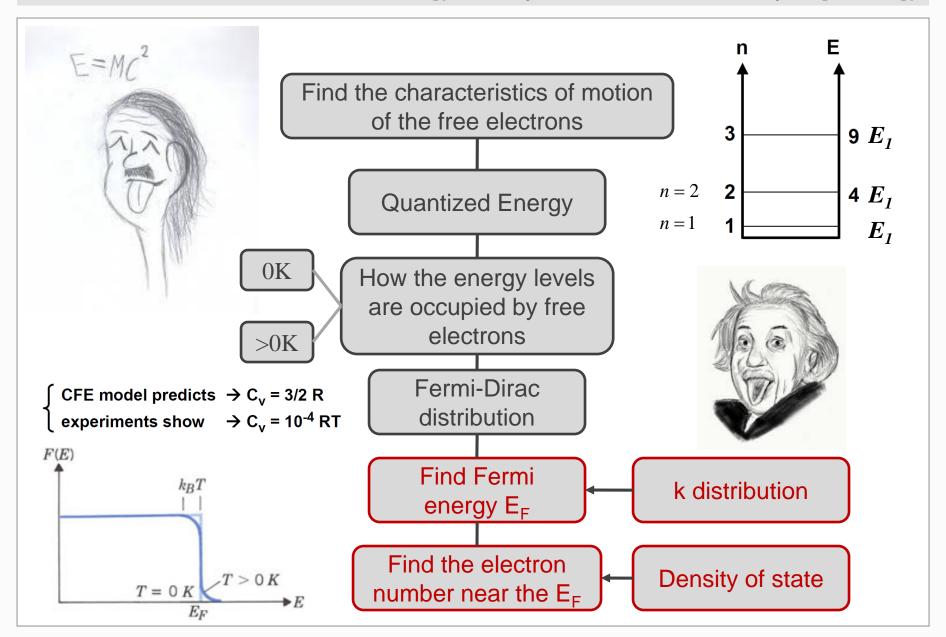
For a conductor of a=4cm

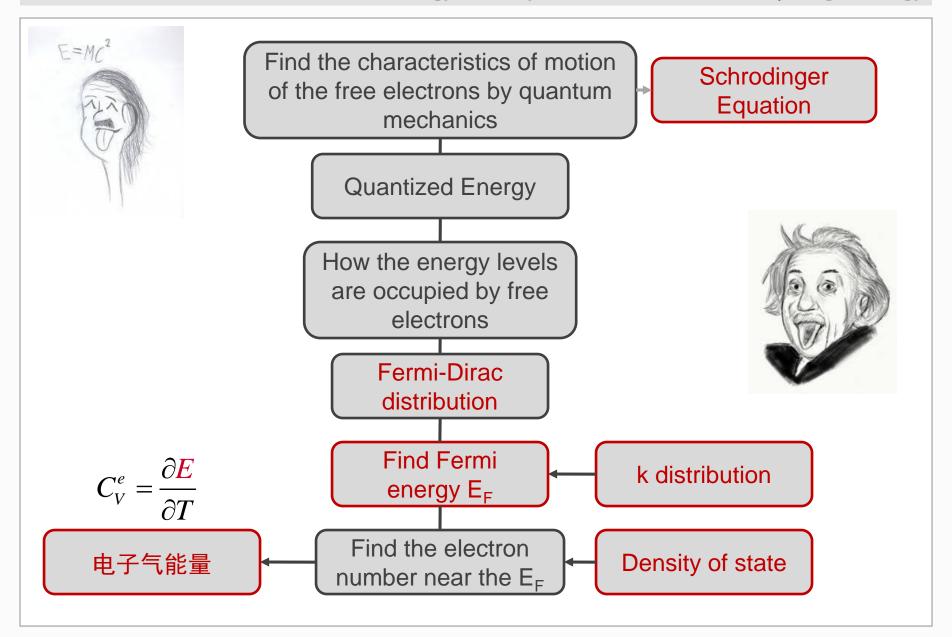
$$E_1 = 2.3 \times 10^{-41} \text{eV}$$

Fermi wave vector \mathbf{k}_{F} : $k_F = \left(\frac{3\pi^2 N}{V}\right)^{1/3} = 0.746 \,\mathrm{A}^{-1}$

Fermi temperature T_F : $T_F = \frac{E_F}{k_B} = 2.46 \times 10^4 K$

· Potential Well · F-D Distribution · Fermi Energy · Density of State · e⁻ number near E_F · e⁻ gas Energy





· Potential Well · F-D Distribution · Fermi Energy · Density of State · e · number near E_F · e · gas Energy

Density of state (DOS)

DOS: the number of electronic energy states (orbits) per unit energy. --- g(E)

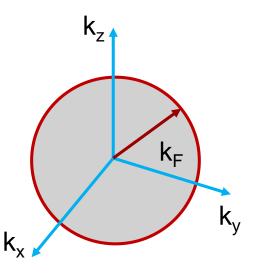
$$g(E) = \frac{\mathrm{d}Z}{\mathrm{d}E}$$

Remember: In crystal dynamics, the density of states $g(\omega)$ is defined as the number of oscillators (or k) per unit frequency interval.

Each k state represents two possible electron states: one for spin up, the other for spin down, thus the total number of electronic states in a sphere of diameter k is:

$$Z(E) = \frac{2}{3} \cdot \rho(k) \cdot \frac{4}{3} \pi k^3 = \frac{2}{3} \cdot \frac{V}{8\pi^3} \cdot \frac{4}{3} \pi k^3 = \frac{Vk^3}{3\pi^2}$$

Density of state (DOS)



$$g(E) = \frac{dZ}{dE} = \frac{dZ}{dk} \cdot \frac{dk}{dE}$$

$$g(E) = \frac{\mathrm{d}Z}{\mathrm{d}k} \cdot \frac{\mathrm{d}k}{\mathrm{d}E} = \frac{V}{2\pi^2} \cdot \left(\frac{2m}{\hbar^2}\right)^{3/2} \cdot E^{1/2}$$

$$g(E) = C \cdot E^{1/2}$$

$$C = \frac{V}{2} \left(\frac{2m}{L^2}\right)^{3/2}$$

$$\frac{\mathrm{d}Z}{\mathrm{d}k} = \frac{Vk^2}{\pi^2}$$

$$E = \frac{\hbar^2 k^2}{2m}$$

$$\frac{\mathrm{d}E}{\mathrm{d}k} = \frac{\hbar^2 k}{m}$$

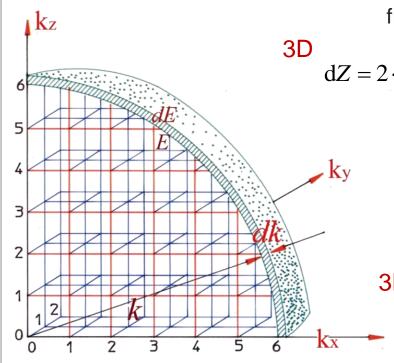
$$k = \frac{\sqrt{2mE}}{\hbar}$$

$$C = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} \quad \frac{dk}{dE} = \sqrt{\frac{m}{2}} \frac{E^{-1/2}}{\hbar}$$

· Potential Well · F-D Distribution · Fermi Energy · Density of State · e · number near E_F · e · gas Energy

Another Way to Calculate DOS

For a free electron gas in 3D, the density of states for free electrons increases with the increase of energy E.



$$dZ = 2 \cdot \rho(k) \cdot 4\pi k^2 \cdot dk = 2 \cdot \frac{V}{(2\pi)^3} \cdot 4\pi \cdot \frac{2mE}{\hbar^2} \cdot \sqrt{\frac{m}{2}} \frac{E^{-1/2}}{\hbar} \cdot dE$$

$$g(E) = \frac{dZ}{dE} = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} E^{1/2}$$

$$g(E) = C \cdot E^{1/2} \qquad C = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2}$$

3D
$$dZ = 2 \cdot \rho(k) \cdot 4\pi k^2 \cdot dk = 2 \cdot \frac{V}{8\pi^3} \cdot 4\pi k^2 \cdot dk = \frac{Vk^2}{\pi^2} dk$$

$$g(E) = \frac{dZ}{dE} = \frac{Vk^{2}}{\pi^{2}} \frac{dk}{dE} = \frac{Vk^{2}}{\pi^{2}} \frac{m}{\hbar^{2}k} = \frac{mVk}{\pi^{2}\hbar^{2}} = \frac{mV}{\pi^{2}\hbar^{2}} \sqrt{\frac{2mE}{\hbar^{2}}}$$

1D
$$dZ = 2 \cdot \rho(k) \cdot 2 \cdot dk = 4 \frac{L}{2\pi} dk$$
 2D $dZ = 2 \cdot \rho(k) \cdot 2\pi k \cdot dk = 4\pi k \frac{S}{4\pi^2} dk$ $g(E) = \frac{dZ}{dE} = \frac{2L}{\pi} \frac{dk}{dE} = \frac{2L}{\pi} \frac{m}{\hbar^2 k} = \frac{2L}{\pi} \frac{m}{\hbar^2 \sqrt{\frac{2mE}{\hbar^2}}}$ $g(E) = C \cdot E^{-1/2}$ $g(E)$ has no relationship with E of $g(E)$ and $g(E)$ and $g(E)$ and $g(E)$ are $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ are $g(E)$ and $g(E)$ are $g(E)$ ar

2D
$$dZ = 2 \cdot \rho(k) \cdot 2\pi k \cdot dk = 4\pi k \frac{S}{4\pi^2} dk$$

$$g(E) = \frac{dZ}{dE} = \frac{kS}{\pi} \frac{dk}{dE} = \frac{kS}{\pi} \frac{m}{\hbar^2 k} = \frac{mS}{\pi \hbar^2}$$

g(E) has no relationship with E or k



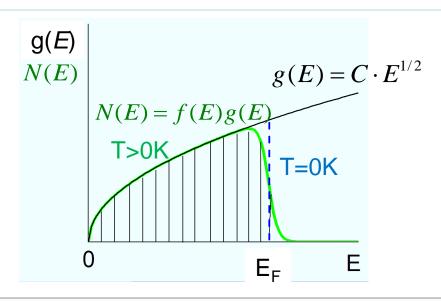
Situation of Orbits Occupied by Electrons

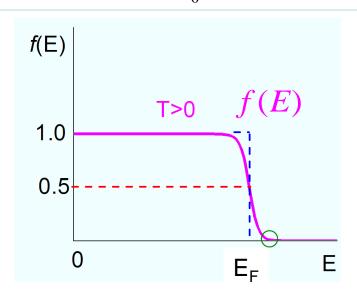
DOS g(E): the number of electronic energy states (orbits) per unit energy. f(E): probability that an electronic energy state be occupied.

Define N(E) as Electron energy distribution function $N(E)dE = \text{number of e}^{-}$ with energies between E and E+dE $g(E)f(E)dE = \text{number of e}^{-}$ with energies between E and E+dE

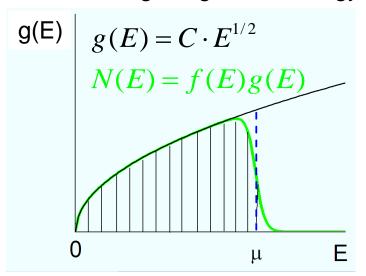
$$\rightarrow N(E) = g(E)f(E)$$

Total number of
$$e^{-} N = \int_{0}^{\infty} g(E) f(E) dE$$





Another method of getting Fermi energy



Total number of e^{-} $N = \int_{0}^{\infty} g(E) f(E) dE$

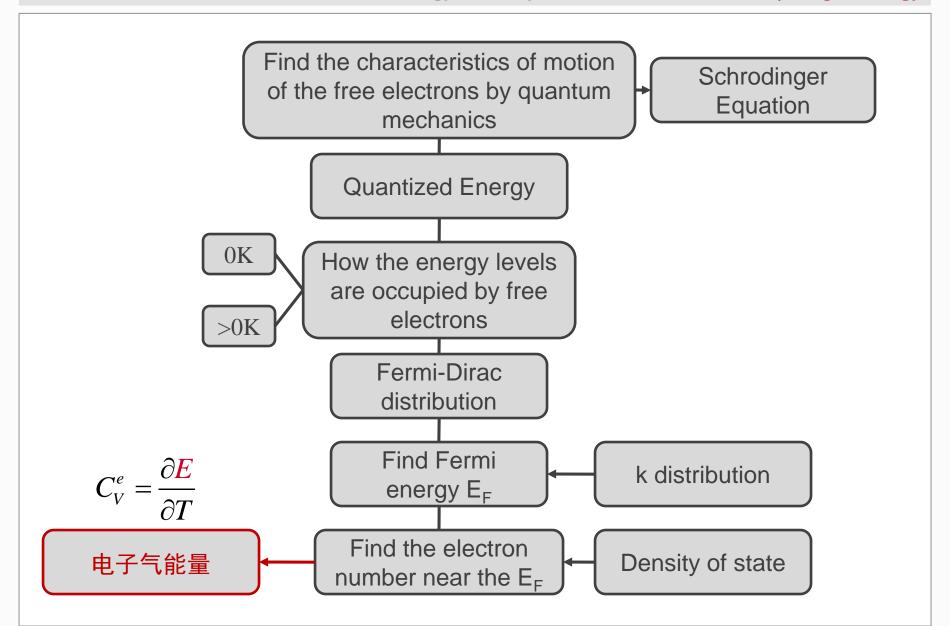
$$g(E) = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} E^{1/2}$$

T=0K

$$N = \int_{0}^{E^{0}_{F}} f(E)g(E)dE = \int_{0}^{E^{0}_{F}} \frac{V}{2\pi^{2}\hbar^{3}} (2m)^{3/2} E^{1/2}dE = \frac{V}{3\pi^{2}\hbar^{3}} (2mE_{F}^{0})^{3/2}$$

$$\downarrow E_{F}^{0} = \frac{\hbar^{2}}{2m} \left(3\pi^{2}\frac{N}{V}\right)^{\frac{2}{3}} = \frac{\hbar^{2}}{2m} \left(3\pi^{2}n\right)^{\frac{2}{3}} \longrightarrow k_{F} = \left(\frac{3\pi^{2}N}{V}\right)^{1/3} = \left(3\pi^{2}n\right)^{1/3}$$

$$E_{F} = \frac{\hbar^{2}k_{F}^{2}}{2m_{2}}$$



Total energy for electron gas

T=0K

$$E_{t} = \int E dN = \int E \cdot g(E) \cdot f(E) \cdot dE$$

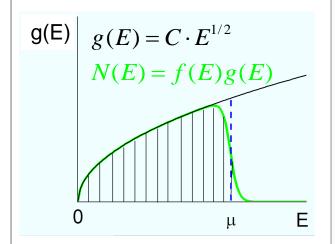
$$= \int_{0}^{E_{F}^{0}} C \cdot E^{1/2} E dE = \int_{0}^{E_{F}^{0}} C E^{3/2} dE = \frac{2C}{5} \left(E_{F}^{0} \right)^{5/2}$$

Average energy for free electrons

$$\overline{E} = \frac{\frac{2C}{5} (E_F^0)^{5/2}}{N} = \dots = \frac{3}{5} E_F^0 \qquad E_t = \frac{3}{5} N E_F^0$$

At T=0K, the average energy of a free electron is 60% of the Femi energy.

Q: Is the result similar to that of classical theory?



$$N = \int_0^\infty g(E) f(E) dE$$
$$= \int_0^{E_F^0} g(E) dE$$
$$= \frac{V}{3\pi^2 \hbar^3} (2mE_F^0)^{3/2}$$

$$E_F^0 = \frac{\hbar^2}{2m} \left(3\pi^2 \, \frac{N}{V} \right)^{2/3}$$

$$C = \frac{V\left(2m\right)^{\frac{3}{2}}}{2\pi^2\hbar^3}$$

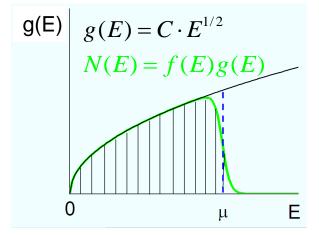
Total energy for electron gas

T>0K

$$E_{t} = \int E \cdot g(E) \cdot f(E) dE$$

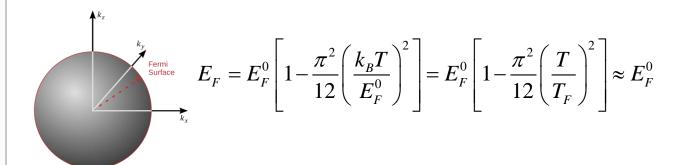
$$= C \int_{0}^{\infty} E^{3/2} \frac{1}{e^{(E-E_{F})/k_{B}T} + 1} dE$$

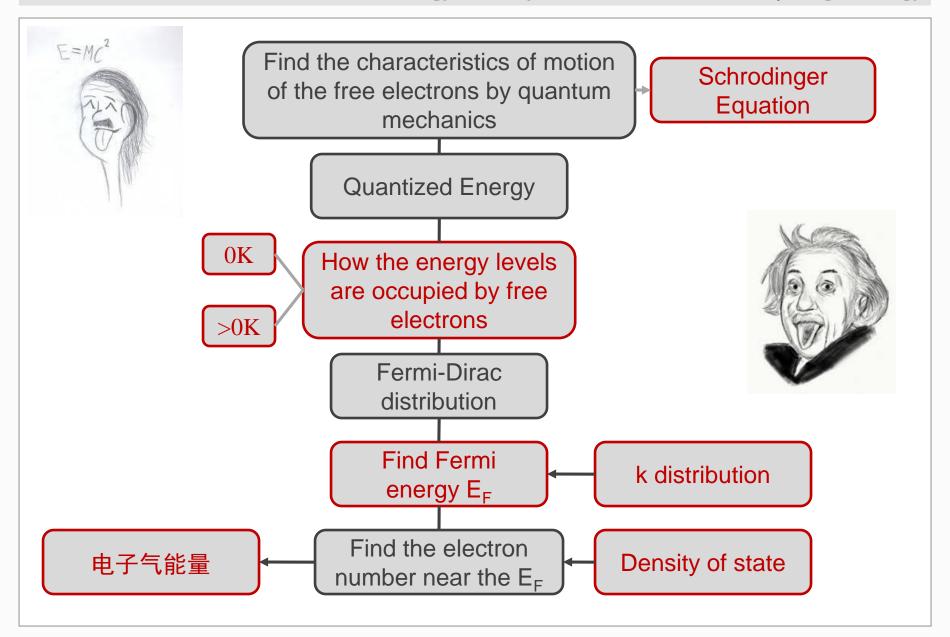
$$= \dots = \frac{3}{5} N E_{F}^{0} [1 + \frac{5}{12} \pi^{2} (\frac{k_{B}T}{E_{F}^{0}})^{2}]$$



$$N = \int_0^\infty g(E) f(E) dE$$
$$= \int_0^\infty \frac{1}{e^{(E - E_F)/k_B T} + 1} g(E) dE$$

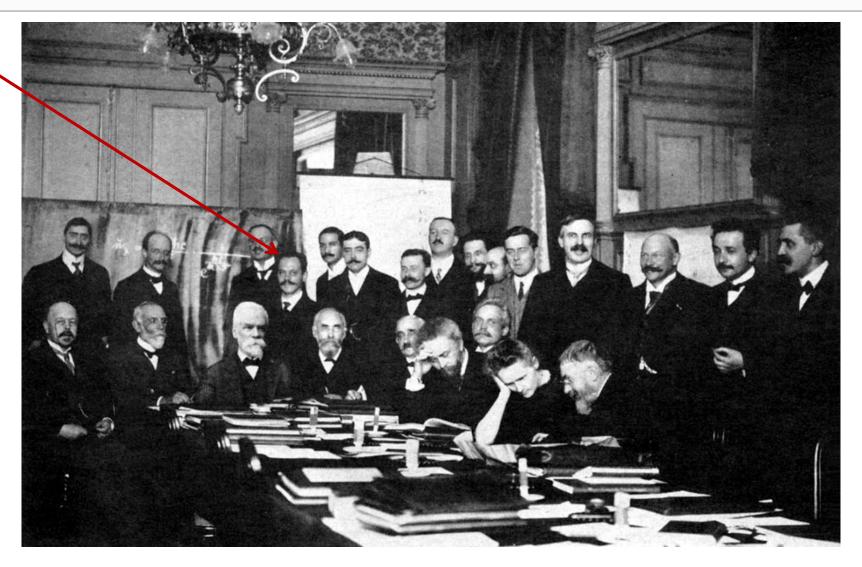
Fermi energy







A. Piccard, E. Henriot, P. Ehrenfest, E. Herzen, Th. De Donder, E. Schrödinger, J.E. Verschaffelt, W. Pauli, W. Heisenberg, R.H. Fowler, L. Brillouin; B. P. Debye, M. Knudsen, W.L. Bragg, H.A. Kramers, P.A.M. Dirac, A.H. Compton, L. de Broglie, M. Born, N. Bohr; I. Langmuir, M. Planck, M. Skłodowska-Curie, H.A. Lorentz, A. Einstein, P. Langevin, Ch. E. Guye, C.T.R. Wilson, O.W. Richardson



Photograph of the first conference in 1911 at the Hotel Metropole. Seated (L-R): W. Nernst, M. Brillouin, E. Solvay, H. Lorentz, E. Warburg, J. Perrin, W. Wien, M. Skłodowska-Curie, and H. Poincaré. Standing (L-R): R. Goldschmidt, M. Planck, H. Rubens, A. Sommerfeld, F. Lindemann, M. de Broglie, M. Knudsen, F. Hasenöhrl, G. Hostelet, E. Herzen, J.H. Jeans, E. Rutherford, H. Kamerlingh Onnes, A. Einstein and P. Langevin.



Arnold Sommerfeld (1868 - 1951), a German physicist who is one of the founders of the quantum mechanics

Doctoral students

Werner Heisenberg Wolfgang Pauli Peter Debye Paul Sophus Epstein Hans Bethe **Ernst Guillemin** Karl Bechert Paul Peter Ewald Herbert Fröhlich **Erwin Fues** Helmut Hönl Ludwig Hopf Walther Kossel Adolf Kratzer Alfred Landé Otto Laporte Wilhelm Lenz **Rudolf Peierls** Walter Rogowski Rudolf Seeliger Heinrich Welker **Gregor Wentzel**



1935

Chapter 3 Free electrons in solids

- 3.1 Free electron model
 - 3.1.1 Drude Model
 - Classical Free Electron Model
 - 3.1.2 Sommerfeld Model
 - Quantum Mechanical Free Electron Model
- 3.2 Heat capacity of free electron gas
- 3.3 Transport properties of conductive electrons
- 3.4 Hall effect and thermal-electric effect