AAE 538: Air-Breathing Propulsion

Lecture 27: Supersonic Combustion

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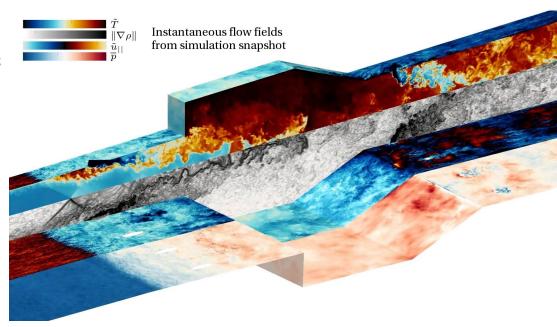


Introduction

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Combustion Processes and Components

- Structural and thermal limitations constrain flight to specific altitudes and free-stream conditions: impacts the airframe as well as the propulsion system (the engine-frame)
- Temperature limits related to combustion product gas dissociation dictate:
 - Combustion system entry Mach number
 - Thermo-physical state
- The practical issues of supersonic combustion are vast:
 - Capture a tube of supersonic air
 - Inject fuel
 - Achieve (fairly) uniform mixing
 - Complete combustion
 - ... All within a minimal length.



https://www.youtube.com/watch?v=86oAlr1EGfc

'We are still confused, but at a higher level' - Enrico Fermi



 The maximum combustion temperature occurs when the injector fuel molecules are sufficiently mixed with the oxidizer to complete the chemical reaction:

$$C_x H_y + z \underbrace{\left[O_2 + \left(\frac{79}{21}\right)N_2\right]}_{qir} \to x(CO_2) + \frac{y}{2}(H_2O) + z\left(\frac{79}{21}\right)N_2$$

- o Fuels:
- o Oxidizer: 21% O_2 and 79% N_2
- Treating nitrogen as an inert diluent that absorbs some of the sensible thermal energy released by combustion due to its specific heat.
- We understand that the stoichiometric fuel-air ratio is that which results in the greatest liberation of sensible energy from stored chemical energy, where:

Actual fuel-air ratio to the stoichiometric fuel-air ratio

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- When we think about combustion in this way, we are making an implicit assumption that the reactants are fully-mixed and burn in a steady, uniform manner
 - This is seldom the case, in practice; particularly in aerospace systems.
- Real combustion is a finite-rate, multi-scale process:

 Scramjets offer extremely low residence times for these processes, so the products look more like:

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- The flame speed changes as a function of the equivalence ratio
 - Therefore, it is absolutely necessary to achieve rapid mixing and ignition in order to enable combustion completion on a useful time-scale:
- Gas phase chemical kinetics proceed as a consequence of an exchange of atoms resulting from molecular collisions.
 - Mixing must occur at the ______ before the reactions can proceed.
- Mixture Characterizations:
 - Very-fine macroscale mixing:

- •
- •
- •

Fully-homogeneous:

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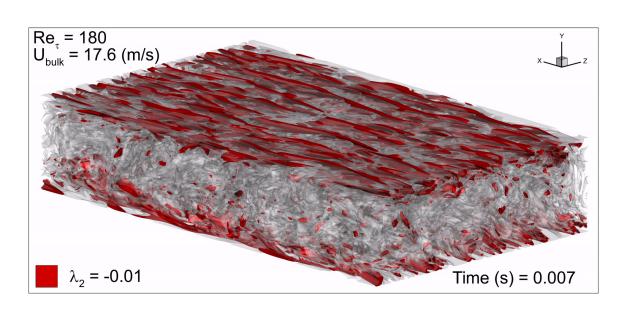
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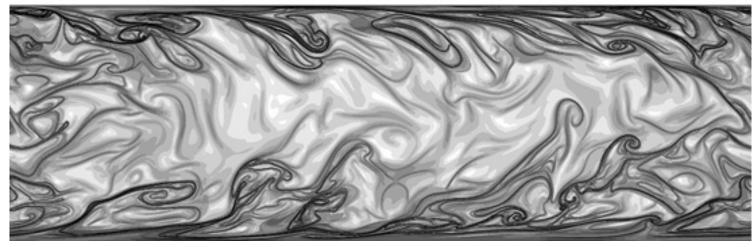
- Transport Physics
 - o The turbulence cascade:
 - The action of large-scale motion is to promote
 - The immediate effect of the large-scale mixing
 - The rate of micro-mixing is
 - Fick's first law:

Fourier's Law:





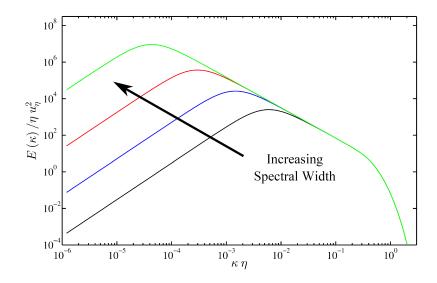


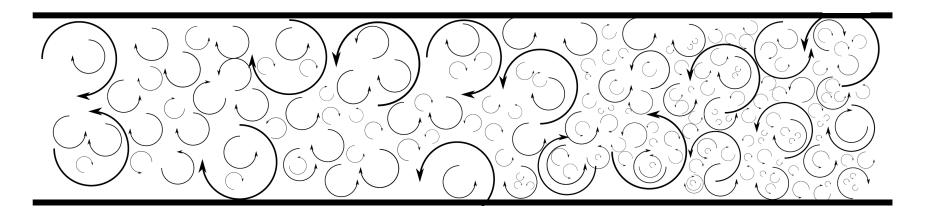




An Interaction of Scales







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- Injection process must induce a very- effective, high-energy cascade! ...where it counts.
 - Compressibility is known to reduce mixing efficiency
 - Devices that enhance mixing generally increase total pressure loss and add drag to the stream
- Maximize the participating area to reduce length.
- Start the cascade with even smaller scales.