

----- indtast data -----

$$\text{hældning}_1 = -0,00335 \cdot 1 \quad [\text{C/s}] \quad \text{hældning fra regression i excel på flaske 1 (øverst)}$$

$$t_{;1} = 20,03 \quad [\text{C}] \quad \text{middeltemperatur fundet i excel}$$

$$\text{hældning}_2 = -0,003662 \cdot 1 \quad [\text{C/s}] \quad \text{hældning fra regression i excel på flaske 2 (nederst)}$$

$$t_{;2} = 20,15 \quad [\text{C}] \quad \text{middeltemperatur fundet i excel}$$

$$\text{hældning}_3 = -0,003255 \cdot 1 \quad [\text{C/s}] \quad \text{hældning fra regression i excel på flaske 3 (øverst)}$$

$$t_{;3} = 21,51 \quad [\text{C}] \quad \text{middeltemperatur fundet i excel}$$

$$p_0 = 1 \quad [\text{bar}] \quad \text{tryk i flaskerne}$$

$$m_{\text{vand};0} = 12 \quad [\text{kg}] \quad \text{samlet masse af vandet}$$

-----1-----

----- Energibalance for nedkøling -----

$$\Phi_{\text{tot};1} = -q_{h;1} \cdot m_{\text{vand};0} \quad \text{Samlet varmetilførsel fra flasker baseret på flaske 1}$$

$$\Phi_{\text{flask};1} = \frac{\Phi_{\text{tot};1}}{40} \quad \text{Varmetilførsel fra flaske 1}$$

----- udregninger hertil -----

$$q_{h;1} = q_{t;1} \cdot cp_{;1} \quad \text{Entalpiændring i flaske 1}$$

$$q_{t;1} = \text{hældning}_1 \quad \text{Temperaturænding i flaske 1}$$

$$cp_{;1} = \mathbf{Cp} \left(\text{water} ; T = t_{;1}; P = p_0 \right) \quad \text{middel cp-værdi for vandet i flasken}$$

-----2-----

----- Energibalance for nedkøling -----

$$\Phi_{\text{tot};2} = -q_{h;2} \cdot m_{\text{vand};0}$$

$$\Phi_{\text{flask};2} = \frac{\Phi_{\text{tot};2}}{40}$$

----- udregninger hertil -----

$$q_{h;2} = q_{t;2} \cdot cp_{;2}$$

$$q_{t;2} = \text{hældning}_2$$

$$cp_{;2} = \mathbf{Cp} \left(\text{water} ; T = t_{;2}; P = p_0 \right)$$

-----3-----

----- Energibalance for nedkøling -----

$$\Phi_{\text{tot};3} = -q_{h;3} \cdot m_{\text{vand};0}$$

$$\Phi_{\text{flask};3} = \frac{\Phi_{\text{tot};3}}{40}$$

----- udregninger hertil -----

$$q_{h,3} = q_{t,3} \cdot cp_{,3}$$

$$q_{t,3} = h_{aldning_3}$$

$$cp_{,3} = Cp \text{ (water ; } T = t_{,3}; P = p_0 \text{)}$$

For di der står to flasker på øverste hylde, skal disse ikke tæller mere i gennemsnittet end den på nederste

$$\Phi_{tot,0} = \frac{\frac{\Phi_{tot,1} + \Phi_{tot,3}}{2} + \Phi_{tot,2}}{2}$$

$$\Phi_{flask,0} = \frac{\frac{\Phi_{flask,1} + \Phi_{flask,3}}{2} + \Phi_{flask,2}}{2}$$

$$\Delta t = 11 \text{ [C]}$$

$$A_{tot} = 296 \cdot \left| 0,0001 \cdot \frac{m^2}{cm^2} \right|$$

$$\Phi_{flask,0} = A_{tot} \cdot U_u \cdot \Delta t$$

$$U_t = 1,4$$

Unit Settings: SI C bar J mass deg

Variable±Uncertainty

$$\Phi_{flask,0} = 4,371 \pm 0,0006183 \text{ [W]}$$

$$\bar{t}_1 = 20,03 \pm 1,4 \text{ [C]}$$

$$\bar{t}_2 = 20,15 \pm 1,4 \text{ [C]}$$

$$\bar{t}_3 = 21,51 \pm 1,4 \text{ [C]}$$

Partial derivative

$$\partial \Phi_{flask,0} / \partial \bar{t}_1 = -0,0001749$$

$$\partial \Phi_{flask,0} / \partial \bar{t}_2 = -0,0003779$$

$$\partial \Phi_{flask,0} / \partial \bar{t}_3 = -0,000147$$

% of uncertainty

$$15,68 \%$$

$$73,24 \%$$

$$11,08 \%$$

$$\Phi_{tot,0} = 174,8 \pm 0,02473 \text{ [W]}$$

$$\bar{t}_1 = 20,03 \pm 1,4 \text{ [C]}$$

$$\bar{t}_2 = 20,15 \pm 1,4 \text{ [C]}$$

$$\bar{t}_3 = 21,51 \pm 1,4 \text{ [C]}$$

$$\partial \Phi_{tot,0} / \partial \bar{t}_1 = -0,006995$$

$$\partial \Phi_{tot,0} / \partial \bar{t}_2 = -0,01512$$

$$\partial \Phi_{tot,0} / \partial \bar{t}_3 = -0,00588$$

$$15,68 \%$$

$$73,24 \%$$

$$11,08 \%$$

$$U_u = 13,42 \pm 0,001899 \text{ [W/m}^2\text{°C]}$$

$$\bar{t}_1 = 20,03 \pm 1,4 \text{ [C]}$$

$$\bar{t}_2 = 20,15 \pm 1,4 \text{ [C]}$$

$$\bar{t}_3 = 21,51 \pm 1,4 \text{ [C]}$$

$$\partial U_u / \partial \bar{t}_1 = -0,000537$$

$$\partial U_u / \partial \bar{t}_2 = -0,001161$$

$$\partial U_u / \partial \bar{t}_3 = -0,0004515$$

$$15,68 \%$$

$$73,24 \%$$

$$11,08 \%$$

No unit problems were detected.

$$A_{tot} = 0,0296 \text{ [m}^2\text{]}$$

$$U_u = 13,42 \text{ [W/m}^2\text{°C]}$$

$$\Delta t = 11 \text{ [C]}$$

$$U_t = 1,4 \text{ [C]}$$

No unit problems were detected.