



# Material Safety Data Sheet

VALVE REGULATED (VRLA) BATTERIES - ABSORBED ELECTROLYTE (AGM)

## I. Product Identification and Company Identification

Chemical / Trade Name (Identity used on label)

**Absorbed Electrolyte Battery / HGL, DC, HGHL  
Sealed Valve Regulated Lead-Acid Battery**

Manufacturer's Name

**Fullriver Battery Manufacture Co. Ltd.**

Address

**P.O. Box 511475**

**Taishi Industrial Area, Yuwotou Town,  
Panyu Zone, Guangzhou, China**

Chemical Family / Classification

**Electric Storage Battery**

Date Revised

**May 27th, 2009**

Telephone

**86-20-84916671**

Web

**<http://www.fullriverdcbattery.com>**

## II. Hazardous Ingredients / Identify Information

NOTE: Inorganic lead and electrolyte (water and sulfuric acid solution) are the primary components of every battery manufactured by Fullriver Technologies or its subsidiaries. Other ingredients may be present dependent upon battery type.

Materials / Components	% by Wt.	CAS Number	Exposure Limits		
			OSHA	ACGIH	NIOSH
Specific Chemical Identity / Common Name <b>Inorganic Lead / Lead Compounds</b>	65% - 75%	7439-92-1	50 µg/m <sup>3</sup>	150 µg/m <sup>3</sup>	100 µg/m <sup>3</sup>
Specific Chemical Identity / Common Name <b>Tin</b>	<0.5%	7440-31-5	2000 µg/m <sup>3</sup>	2000 µg/m <sup>3</sup>	NA
Specific Chemical Identity / Common Name <b>Calcium</b>	<0.2%	7440-70-2	NA	NA	NA
Specific Chemical Identity / Common Name <b>Sulfuric Acid (40%) / Battery Electrolyte (Acid)</b>	16% - 21%	7664-93-9	1 mg/m <sup>3</sup>	1 mg/m <sup>3</sup>	1 mg/m <sup>3</sup>
Specific Chemical Identity / Common Name <b>Fiberglass Separator</b>	5%	-	NA	NA	NA
Specific Chemical Identity / Common Name <b>Acrylonitrile Butadiene Styrene (ABS)</b>	5% - 10%	9003-56-9	NA	NA	NA

## III. Fire and Explosion Hazard Data

Fire and Explosive Properties

**Hydrogen Flash Point: N/A**

**Hydrogen Auto Ignition Point: 580°C**

**Hydrogen Flammabl Limits in Air (% by Volume): LEL: 4.1**

**UEL: 74.2**

**Low Explosion Limit (LEL), Upper Explosion Limit (UEL)**

Extinguishing Media

**Dry Chemical, Foam, CO<sub>2</sub>**

Special Fire Fighting Hazards

**Use Positive Pressure, Self-Contained breathing apparatus.**

Unusual Fire and Explosion Hazards

If AGM batteries are properly charged they will not release any flammable hydrogen gas. If they are excessively overcharged the safety relief valve can open and release flammable hydrogen gas. They must always be assumed to contain this gas which, if ignited by burning cigarette, naked flame or spark, may cause battery explosion with dispersion of casing fragments and corrosive liquid electrolyte. Carefully follow manufacturer's instruction for installation and service. Keep away all sources of gas ignition and do not allow metallic articles to simultaneously contact the negative and positive terminals of a battery.

## IV. Health Hazard Information

### Routes of Entry

**Sulfuric Acid:** Harmful by all routes of entry.

**Lead Compounds:** Hazardous exposure can occur only when product is heated, oxidized or otherwise processed or damaged to create dust, vapor or fumes.

### Inhalation

**Sulfuric Acid:** Breathing of sulfuric acid vapors or mists may cause severe respiratory irritation.

**Lead Compounds:** Inhalation of lead dust or fumes may cause irritation of upper respiratory tract and lungs.

**Fiberglass Separator:** Fiberglass is an irritant to the upper respiratory tract, skin and eyes. For exposure up to 10F°/use MSA Comfoll with type H filter. Above 10F° use Ultra Twin with type H filter. This product is not considered carcinogenic by NTP or OSHA.

### Ingestion

**Sulfuric Acid:** May cause severe irritation of mouth, throat, esophagus and stomach.

**Lead Compounds:** Acute ingestion may cause abdominal pain, nausea, vomiting, diarrhea and severe cramping. This may lead rapidly to system toxicity and must be treated by a physician.

### Skin Contact

**Sulfuric Acid:** Severe irritation, burns and ulceration.

**Lead Compounds:** Not absorbed through the skin.

### Eye Contact

**Sulfuric Acid:** Severe irritation, burns, cornea damage, and blindness.

**Lead Compounds:** May cause eye irritation.

### Effects of Overexposure - Acute

**Sulfuric Acid:** Severe skin irritation, damage to cornea, upper respiratory irritation.

**Lead Compounds:** Symptoms of toxicity include headache, fatigue, abdominal pain, loss of appetite, muscular aches and weakness, sleep disturbances and irritability.

### Effects of Overexposure - Chronic

**Sulfuric Acid:** Possible erosion of tooth enamel, inflammation of nose, throat and bronchial tubes.

**Lead Compounds:** Anemia; neuropathy, particularly of the motor nerves, with wrist drop; kidney damage; reproductive changes in males and females.

### Carcinogenicity

**Sulfuric Acid:** The International Agency for Research on Cancer (IARC) has classified "strong inorganic acid mist containing sulfuric acid" as a Category I carcinogen, a substance that is carcinogenic to humans. This classification does not apply to liquid forms of sulfuric acid or sulfuric acid solutions contained within a battery. Inorganic acid mist (sulfuric acid mist) is not generated under normal use of this product. Misuse of the product, such as overcharging, may result in the generation of sulfuric acid mist.

**Lead Compounds:** Lead is listed as a 2B carcinogen, likely in animals at extreme doses. Proof of carcinogenicity in humans is lacking at present.

### Medical Conditions Generally Aggravated by Exposure

Overexposure to sulfuric acid mist may cause lung damage and aggravate pulmonary conditions. Contact of sulfuric acid with skin may aggravate diseases such as eczema and contact dermatitis. Lead and its compounds can aggravate some forms of kidney, liver and neurologic diseases.

## V. Emergency and First Aid Procedures

### Inhalation

**Sulfuric Acid:** Remove to fresh air immediately. If breathing is difficult, give oxygen.

**Lead Compounds:** Remove from exposure, gargle, wash nose and lips; consult physician.

### Ingestion

**Sulfuric Acid:** Give large quantities of water; do not include vomiting; consult physician.

**Lead Compounds:** Consult physician immediately.

### Skin

**Sulfuric Acid:** Flush with large amounts of water for at least 15 minutes; remove contaminated clothing completely, including shoes.

**Lead Compounds:** Wash immediately with soap and water.

### Eyes

**Sulfuric Acid and Lead:** Flush immediately with large amounts of water for at least 15 minutes; consult physician.

## VI. Precautions For Safe Handling and Use

### Handling and Storage

Store batteries in cool, dry, well-ventilated areas with impervious surfaces and adequate containment in the event of spills. Batteries should also be stored under roof for protection against adverse weather conditions. Separate from incompatible materials. Store and handle only in areas with adequate water supply and spill control. Avoid damage to containers. Keep away from fire, sparks and heat.

### Precautionary Labeling

**POISON - Causes severe burns DANGER - Contains Sulfuric Acid**

### Charging

There is a possible risk of electric shock from charging equipment and from strings of series connected batteries, whether or not being charged. Shut-off power to chargers whenever not in use and before detachment of any circuit connections. If batteries are properly charged they will not release any flammable hydrogen gas. If they are excessively overcharged the safety relief valve can open and release flammable hydrogen gas. Charging space should be ventilated. Keep battery vent caps in position. Prohibit smoking and avoid creation of flames and sparks nearby. Wear face and eye protection when near batteries being charged.

### Spill or Leak Procedures

Stop flow of material; contain/absorb small spills with dry sand, earth and vermiculite. Do not use combustible materials. If possible, carefully neutralize spilled electrolyte with soda ash, sodium bicarbonate, lime, etc. Wear acid-resistant clothing, boots, gloves, and face shield. Do not allow discharge of un-neutralized acid to sewer.

### Waste Disposal Method

Spent batteries: Send to secondary lead smelter for recycling.



**LEAD - RETURN - RECYCLE**

## VII. Control Measures

### Engineering Controls

Store and handle in well-ventilated area. If mechanical ventilation is used, components must be acid-resistant.

### Work Practices

Handle batteries cautiously to avoid spills. Make certain vent caps are on securely. Avoid contact with internal components. Wear protective clothing when filling or handling batteries.

### Respiratory Protection

None required under normal conditions. When concentrations of sulfuric acid mist are known to exceed the PEL, use NIOSH or MSHA-approved respiratory protection.

### Protective Gloves

Rubber or plastic acid-resistant gloves with elbow-length gauntlet.

### Eye Protection

Chemical goggles or face shield.

### Other Protection

Acid-resistant apron. Under severe exposure emergency conditions, wear acid-resistant clothing and boots.

### Emergency Flushing

In areas where sulfuric acid is handled in concentrations greater than 1%, emergency eyewash stations and showers should be provided, with unlimited water supply.

## VIII. Physical Data

<b>Electrolyte:</b>			
<b>Boiling Point:</b>	203-240°F	<b>Specific Gravity (H<sub>2</sub>O=1):</b>	1.300-1.330
<b>Melting Point:</b>	N/A	<b>Vapor Pressure (mm Hg):</b>	10
<b>Solubility in Water:</b>	100%	<b>Vapor Density (AIR = 1):</b>	3.4
<b>Evaporation Rate: (Butyl Acetate = 1)</b>	Less than 1	<b>% Volatile by Weight:</b>	N/A
<b>Appearance and Odor:</b>	Manufactured article; no apparent odor. Electrolyte is a clear liquid with a sharp, penetrating, pungent odor.		

## IV. Reactivity Data

**Stability:** Stable

**Conditions to Avoid:** High temperature, Sparks and other sources of ignition

### Incompatibility (Materials to avoid)

**Electrolyte (Water and Sulfuric Acid Solution):** Contact with combustibles and organic materials may cause fire and explosion. Also reacts violently with strong reducing agents, metals, sulfuric trioxide gas, strong oxidizers, and water. Contact with metals may produce toxic sulfur dioxide fumes and may release flammable hydrogen gas.

Reactivity Data Cont...

## IV. Reactivity Data (Continued)

### Incompatibility (Materials to avoid) Continued...

**Lead Compounds:** Avoid contact with strong acids, bases, halides, halogenated, potassium nitrate, permanganate, peroxides, nascent hydrogen, and reducing agents.

### Hazardous Byproducts

**Sulfuric Acid:** Sulfur trioxide, carbon monoxide, sulfuric acid mist, sulfur dioxide, and hydrogen.

**Lead Compounds:** High temperatures likely to produce toxic metal fume, vapor, or dust; contact with strong acid or base or presence of nascent hydrogen may generate highly toxic arsine gas.

## X. Ecological Information

**Lead and its compounds can pose a threat if released into the environment.**

## XI. Transport Information

All **Fullriver** AGM batteries, when transported by air, surface or by vessel are identified as **"Battery, Electric Storage, Wet, Nonspillable, Not Regulated"**.

The battery(s) must be identified as above on the Bill of Lading and properly packaged with their terminals protected from short circuit. NA or UN numbers do not apply.

**Fullriver** AGM battery(s) warning label identifies each battery as NONSPILLABLE.

**Fullriver** AGM battery(s) preprinted cartons identify each battery as NONSPILLABLE.

**Fullriver** AGM battery(s) shipped without Fullriver cartons (bulk packed) need to be identified as NONSPILLABLE or NONSPILLABLE BATTERY on the outer packaging.

**Air: Fullriver** AGM batteries meet the conditions in IATA/ICAO Special Provision A67.

**Surface: Fullriver** AGM batteries meet the conditions for DOT Haz Mat Regulations CFR-Title 49 parts 171-189.

**Vessel: Fullriver** Batteries meet the conditions of IMDG.

## XII. Regulatory Information

**See 29 CFR 1910.268(b)(2)**

## XIII. Other Information

The information herein is given in good faith, but no warranty, expressed or implied, is made.