

US012391876B2

(12) United States Patent Ji et al.

(54) ORGANIC ELECTROLUMINESCENT MATERIALS AND DEVICES

(71) Applicant: Universal Display Corporation,

Ewing, NJ (US)

(72) Inventors: Zhiqiang Ji, Ewing, NJ (US); Lichang

Zeng, Ewing, NJ (US); Chun Lin,

Ewing, NJ (US)

(73) Assignee: Universal Display Corporation

(*) Notice: Subject to any disclaimer, the term of this

patent is extended or adjusted under 35

U.S.C. 154(b) by 0 days.

(21) Appl. No.: 18/675,910

(22) Filed: May 28, 2024

(65) Prior Publication Data

US 2024/0324445 A1 Sep. 26, 2024

Related U.S. Application Data

- (63) Continuation of application No. 17/382,486, filed on Jul. 22, 2021, now abandoned, which is a continuation of application No. 15/828,678, filed on Dec. 1, 2017, now abandoned.
- (60) Provisional application No. 62/441,653, filed on Jan. 3, 2017.
- (51) Int. Cl.

 C09K 11/06 (2006.01)

 C07F 15/00 (2006.01)

 H10K 50/11 (2023.01)

 H10K 50/81 (2023.01)

 H10K 85/30 (2023.01)

 H10K 101/10 (2023.01)

 H10K 101/20 (2023.01)

(10) Patent No.: US 12,391,876 B2

(45) **Date of Patent:** Aug. 19, 2025

(52) U.S. Cl.

CPC C09K 11/06 (2013.01); C07F 15/0086 (2013.01); H10K 85/346 (2023.02); C09K

2211/1029 (2013.01); C09K 2211/185 (2013.01); H10K 50/11 (2023.02); H10K 50/81

(2023.02); *H10K 2101/10* (2023.02); *H10K 2101/20* (2023.02)

(58) Field of Classification Search

CPC H01L 51/0087; C07F 15/0086; C09K

2211/185; H10K 85/346

See application file for complete search history.

(56) References Cited

U.S. PATENT DOCUMENTS

4,769,292 A	9/1988	Tang	
5,061,569 A	10/1991	Vanslyke	
5,244,919 A	9/1993	Abrams	
5,247,190 A	9/1993	Friend	
5,703,436 A	12/1997	Forrest	
	(Continued)		

FOREIGN PATENT DOCUMENTS

EP	0650955	5/1995	
EP	1238981	9/2002	
	(Continued)		

OTHER PUBLICATIONS

Adachi, Chihaya et al., "High-Efficiency Red Electrophosphorescence Devices," Appl. Phys. Lett., 78(11):1622-1624 (2001).

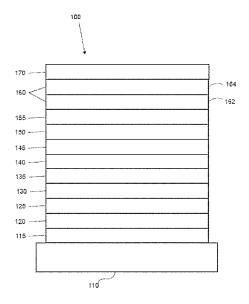
(Continued)

Primary Examiner — Jenna N Chandhok (74) Attorney, Agent, or Firm — Riverside Law LLP

(57) ABSTRACT

The present invention discloses novel Pt(IV) compounds useful for enhancing the performance of organic electroluminescent devices.

12 Claims, 2 Drawing Sheets



US 12,391,876 B2 Page 2

(56)	Referen	nces Cited		2009/011531		5/2009	
т.	I C. DATENIT	DOCUMENTS		2009/011532			Hashimoto Johannes
C	J.S. PATENT	DOCUMENTS		2009/016584 2009/016716		7/2009	
5,707,745	A 1/1008	Forrest		2009/010710		7/2009	
5,834,893		Bulovic		2010/023733		9/2010	
5,844,363				2013/002645		1/2013	
6,013,982		Thompson		2013/011935		5/2013	
6,087,196 A		Sturm		2014/005456		2/2014	
6,091,195		Forrest		2015/031848	37 A1 11	1/2015	Ito
6,097,147 A		Baldo		2016/013386	51 A1 5	5/2016	Li
6,294,398 I 6,303,238 I		Thompson		2017/017498	35 A1 6	5/2017	Royster
6,337,102 H		Forrest		2017/027161	l1 A1 9	9/2017	Li
6,468,819 I							
6,528,187 I		Okada		F	OREIGN	PATE	NT DOCUMENTS
6,687,266 I				ED	150505	10	11/2006
6,835,469 I		Kwong Takiguchi		EP EP	172507 203453		11/2006 3/2009
6,921,915 H 7,087,321 H		Kwong		EP	255193		1/2013
7,090,928 I		Thompson		EP	297737		1/2016
7,154,114 I		Brooks		JP	20051161	.0	1/2005
7,250,226 I		Tokito		JP	200712339		5/2007
7,279,704 I		Walters		JP	200725429		10/2007
7,332,232 I		Ma Thompson		JP JP	200807493 201013546		4/2008 6/2010
7,338,722 I 7,393,599 I		Thompson		WO	013923		5/2001
7,396,598 I		Takeuchi		wo	020271		1/2002
7,431,968 I				WO	021564		2/2002
7,445,855 I		Mackenzie		WO	0304025		5/2003
7,534,505 I				WO	0306095		7/2003
7,968,146 I		Wagner		WO WO	200409320		10/2004
8,409,729 I 2002/0034656 A		Zeng Thompson		WO	200410782 200411106		12/2004 12/2004
2002/0034030 /		Igarashi		wo	200501455		2/2005
2002/0158242				WO	200501937		3/2005
2003/0138657 A				WO	200503090		4/2005
2003/0152802 A		Tsuboyama		WO	200508902		9/2005
2003/0162053		Marks		WO	200512387		12/2005
2003/0175553 A 2003/0230980 A		Thompson Forrest		WO WO	200600902 200605641		1/2006 6/2006
2004/0036077				wo	200607200		7/2006
2004/0137267		Igarashi		WO	200608274	12	8/2006
2004/0137268		Igarashi		WO	200609812		9/2006
2004/0174116				WO	200610029		9/2006
2005/0025993 A 2005/0112407 A		Thompson Ogasawara		WO WO	200610387 200611496		10/2006 11/2006
2005/0112407 2	A1 10/2005	Ogasawara		wo	200613217		12/2006
2005/0244673				WO	200700268		1/2007
2005/0260441		Thompson		WO	200700438		1/2007
2005/0260449 A		Walters		WO	200706375		6/2007
2006/0008670				WO WO	200706379 200804472		6/2007 4/2008
2006/0182992 A 2006/0202194 A	A1 8/2006 A1 9/2006	Jeong			200804472		5/2008
2006/0240279	A1 10/2006	Adamovich		wo	20080577		5/2008
2006/0251923				WO	200810184	12	8/2008
2006/0258043 A	A1* 11/2006	Bold C07F 1		WO	200813208		11/2008
2006/0262625			438/99	WO	200900067		12/2008
2006/0263635 A 2006/0280965 A		Ise Kwong		WO WO	200900389 200900831		1/2009 1/2009
2007/0190359		Knowles		wo	200901800		2/2009
2007/0196687		Oshiyama		WO	200902112		2/2009
2007/0278938		Yabunouchi		WO	200905029		4/2009
2008/0015355	A1 1/2008	Schafer		WO	200906257		5/2009
2008/0018221				WO WO	200906383 200906677		5/2009 5/2009
2008/0106190 A		Yabunouchi Mizuki		WO	200906677		5/2009
2008/0124572 A 2008/0220265 A				wo	200908602		7/2009
2008/0220203 A		Knowles		WO	200910099		8/2009
2009/0008605		Kawamura		WO	201001139	00	1/2010
2009/0009065		Nishimura		WO	201011117		9/2010
2009/0017330		Iwakuma		WO	201012623	4	11/2010
2009/0030202 A 2009/0039776 A		Iwakuma Yamada					
2009/0039776 2		Yamada Nishimura			OTHE	er pui	BLICATIONS
2009/0045731		Nishimura					
2009/0079329		Murakami			•	_	Electroluminescent Device Having
2009/0101870	A1 4/2009	Prakash			tor as an Em	itting La	yer," Appl. Phys. Lett., 55(15):1489-
2009/0108737	A1 4/2009	Kwong		1491 (1989).			

(56) References Cited

OTHER PUBLICATIONS

Adachi, Chihaya et al., "Nearly 100% Internal Phosphorescence Efficiency in an Organic Light Emitting Device," J. Appl. Phys., 90(10):5048-5051 (2001).

Aonuma, Masaki et al., "Material Design of Hole Transport Materials Capable of Thick-Film Formation in Organic Light Emitting Diodes," Appl. Phys. Lett., 90, Apr. 30, 2007, 183503-1-183503-3. Baldo et al., "Highly Efficient Phosphorescent Emission from Organic Electroluminescent Devices," Nature, vol. 395, 151-154, (1998).

Baldo et al., "Very high-efficiency green organic light-emitting devices based on electrophosphorescence," Appl. Phys. Lett., vol. 75, No. 3, 4-6 (1999).

Gao, Zhiqiang et al., "Bright-Blue Electroluminescence From a Silyl-Substituted ter-(phenylene-vinylene) derivative," Appl. Phys. Lett., 74(6):865-867 (1999).

Guo, Tzung-Fang et al., "Highly Efficient Electrophosphorescent Polymer Light-Emitting Devices," Organic Electronics, 1:15-20 (2000).

Hamada, Yuji et al., "High Luminance in Organic Electroluminescent Devices with Bis(10-hydroxybenzo[h]quinolinato) beryllium as an Emitter," Chem. Lett., 905-906 (1993).

Holmes, R.J. et al., "Blue Organic Electrophosphorescence Using Exothermic Host-Guest Energy Transfer," Appl. Phys. Lett., 82(15):2422-2424 (2003).

Hu, Nan-Xing et al., "Novel High Tg Hole-Transport Molecules Based on Indolo[3,2-b]carbazoles for Organic Light-Emitting Devices," Synthetic Metals, 111-112:421-424 (2000).

Huang, Jinsong et al., "Highly Efficient Red-Emission Polymer Phosphorescent Light-Emitting Diodes Based on Two Novel Tris(1-phenylisoquinolinato-C2,N)iridium(III) Derivatives," Adv. Mater., 19:739-743 (2007).

Huang, Wei-Sheng et al., "Highly Phosphorescent Bis-Cyclometalated Iridium Complexes Containing Benzoimidazole-Based Ligands," Chem. Mater., 16(12):2480-2488 (2004).

Hung, L.S. et al., "Anode Modification in Organic Light-Emitting Diodes by Low-Frequency Plasma Polymerization of CHF3,"Appl. Phys. Lett., 78(5):673-675 (2001).

Ikai, Masamichi and Tokito, Shizuo, "Highly Efficient Phosphorescence From Organic Light-Emitting Devices with an Exciton-Block Layer," Appl. Phys. Lett., 79(2):156-158 (2001).

Ikeda, Hisao et al., "P-185: Low-Drive-Voltage OLEDs with a Buffer Layer Having Molybdenum Oxide," SID Symposium Digest, 37:923-926 (2006).

Inada, Hiroshi and Shirota, Yasuhiko, "1,3,5-Tris[4-(diphenylamino)phenyl]benzene and its Methylsubstituted Derivatives as a Novel Class of Amorphous Molecular Materials," J. Mater. Chem., 3(3):319-320 (1993).

Jenkins et al. "Synthesis and Characterization of Luminescent Bis-Cyclometalated PlatinumIV Complexes" Inorg. Chem. 2010, 49, 11297-11308.

Julia et al. "Developing strongly luminescent platinum(IV) complexes: facile synthesis of bis-cyclometalated neutral emitters" Chem. Comm, 2016, 52, 1657-1660.

Julia et al. "Homoleptic tris-cyclometalated platinum(IV) complexes: a new class of long-lived, highly efficient 3LC emitters" Chem. Sci, 2014, 5, 1875-1880.

Kanno, Hiroshi et al., "Highly Efficient and Stable Red Phosphorescent Organic Light-Emitting Device Using bis[2-(2-benzothiazoyl)phenolato]zinc(II) as host material," Appl. Phys. Lett., 90:123509-1-123509-3 (2007).

Kido, Junji et al., "1,2,4-Triazole Derivative as an Electron Transport Layer in Organic Electroluminescent Devices," Jpn. J. Appl. Phys., 32:L917-L920 (1993).

Kuwabara, Yoshiyuki et al., "Thermally Stable Multilayered Organic Electroluminescent Devices Using Novel Starburst Molecules, 4,4',4"-Tri(N-carbazolyl)triphenylamine (TCTA) and 4,4',4"-Tris(3-methylphenyl-amino) triphenylamine (m-MTDATA), as Hole-Transport Materials," Adv. Mater., 6(9):677-679 (1994).

Kwong, Raymond C. et al., "High Operational Stability of Electrophosphorescent Devices," Appl. Phys. Lett., 81 (1):162-164 (2002).

Lamansky, Sergey et al., "Synthesis and Characterization of Phosphorescent Cyclometalated Iridium Complexes," Inorg. Chem., 40(7):1704-1711 (2001).

Lee, Chang-Lyoul et al., "Polymer Phosphorescent Light-Emitting Devices Doped with Tris(2-phenylpyridine) Iridium as a Triplet Emitter," Appl. Phys. Lett., 77(15):2280-2282 (2000).

Lo, Shih-Chun et al., "Blue Phosphorescence from Iridium(III) Complexes at Room Temperature," Chem. Mater., 18 (21):5119-5129 (2006).

Ma, Yuguang et al., "Triplet Luminescent Dinuclear-Gold(I) Complex-Based Light-Emitting Diodes with Low Turn-On voltage," Appl. Phys. Lett., 74(10):1361-1363 (1999).

Mi, Bao-Xiu et al., "Thermally Stable Hole-Transporting Material for Organic Light-Emitting Diode: an Isoindole Derivative," Chem. Mater., 15(16):3148-3151 (2003).

Nishida, Jun-ichi et al., "Preparation, Characterization, and Electroluminescence Characteristics of a-Diimine-type Platinum(II) Complexes with Perfluorinated Phenyl Groups as Ligands," Chem. Lett., 34(4):592-593 (2005).

Niu, Yu-Hua et al., "Highly Efficient Electrophosphorescent Devices with Saturated Red Emission from a Neutral Osmium Complex," Chem. Mater., 17(13):3532-3536 (2005).

Noda, Tetsuya and Shirota, Yasuhiko, "5,6-Bis(dinnesitylboryl)-2,2'-bithiophene and 5,5"-Bis (dimesitylboryl)-2,2':5',2"-terthiophene as a Novel Family of Electron-Transporting Amorphous Molecular Materials," J. Am. Chem. Soc., 120 (37):9714-9715 (1998).

Okumoto, Kenji et al., "Green Fluorescent Organic Light-Emitting Device with External Quantum Efficiency of Nearly 10%," Appl. Phys. Lett., 89:063504-1-063504-3 (2006).

Palilis, Leonidas C., "High Efficiency Molecular Organic Light-Emitting Diodes Based On Silole Derivatives And Their Exciplexes," Organic Electronics, 4:113-121 (2003).

Paulose, Betty Marie Jennifer S, et al., "First Examples of Alkenyl Pyridines as Organic Ligands for Phosphorescent Iridium Complexes," Adv. Mater., 16(22):2003-2007 (2004).

Ranjan, Sudhir et al., "Realizing Green Phosphorescent Light-Emitting Materials from Rhenium(I) Pyrazolato Diimine Complexes," Inorg. Chem., 42(4):1248-1255 (2003).

Sakamoto, Youichi et al., "Synthesis, Characterization, and Electron-Transport Property of Perfluorinated Phenylene Dendrimers," J. Am. Chem. Soc., 122(8):1832-1833 (2000).

Salbeck, J. et al., "Low Molecular Organic Glasses for Blue Electroluminescence," Synthetic Metals, 91:209-215 (1997).

Shirota, Yasuhiko et al., "Starburst Molecules Based on p-Electron Systems as Materials for Organic Electroluminescent Devices," Journal of Luminescence, 72-74:985-991 (1997).

Sotoyama, Wataru et al., "Efficient Organic Light-Emitting Diodes with Phosphorescent Platinum Complexes Containing NCN-Coordinating Tridentate Ligand," Appl. Phys. Lett., 86:153505-1-153505-3 (2005).

Sun, Yiru and Forrest, Stephen R., "High-Efficiency White Organic Light Emitting Devices with Three Separate Phosphorescent Emission Layers," Appl. Phys. Lett., 91:263503-1-263503-3 (2007).

T. Ostergard et al., "Langmuir-Blodgett Light-Emitting Diodes Of Poly(3-Hexylthiophene): Electro-Optical Characteristics Related to Structure," Synthetic Metals, 87:171-177 (1997).

Takizawa, Shin-ya et al., "Phosphorescent Iridium Complexes Based on 2-Phenylimidazo[1,2- a]pyridine Ligands: Tuning of Emission Color toward the Blue Region and Application to Polymer Light-Emitting Devices," Inorg. Chem., 46(10):4308-4319 (2007).

Tang, C.W. and VanSlyke, S.A., "Organic Electroluminescent Diodes," Appl. Phys. Lett., 51(12):913-915 (1987).

Tung, Yung-Liang et al., "Organic Light-Emitting Diodes Based on Charge-Neutral Ru II PHosphorescent Emitters," Adv. Mater., 17(8):1059-1064 (2005).

Van Slyke, S. A. et al., "Organic Electroluminescent Devices with Improved Stability," Appl. Phys. Lett, 69 (15):2160-2162 (1996).

(56)**References Cited**

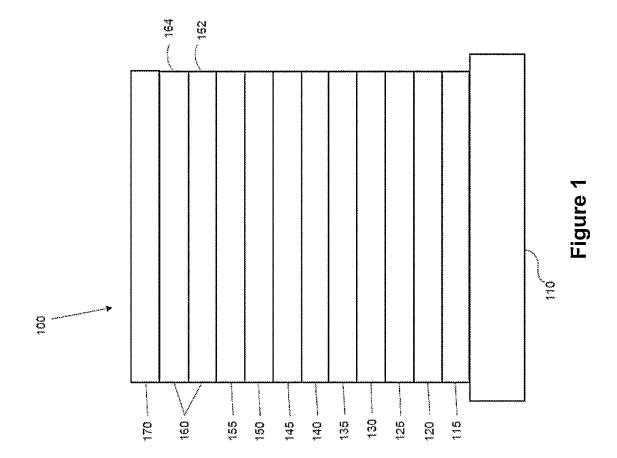
OTHER PUBLICATIONS

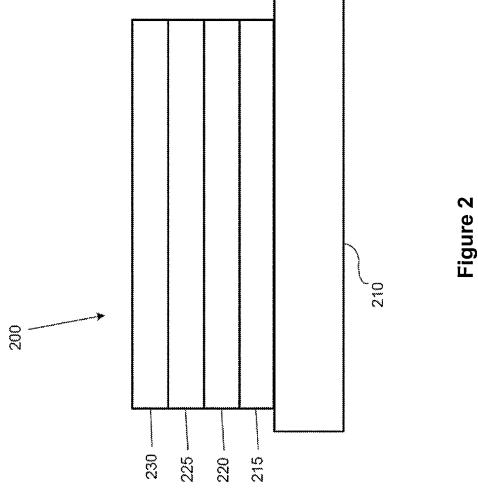
Wang, Y. et al., "Highly Efficient Electroluminescent Materials Based on Fluorinated Organometallic Iridium Compounds," Appl. Phys. Lett., 79(4):449-451 (2001).

Phys. Lett., 79(4):449-451 (2001).
Wong, Keith Man-Chung et al., "A Novel Class of Phosphorescent Gold(III) Alkynyl-Based Organic Light-Emitting Devices with Tunable Colour," Chem. Commun., 2906-2908 (2005).
Wong, Wai-Yeung, "Multifunctional Iridium Complexes Based on Carbazole Modules as Highly Efficient Electrophosphors," Angew.

Chem. Int. Ed., 45:7800-7803 (2006).

^{*} cited by examiner





ORGANIC ELECTROLUMINESCENT MATERIALS AND DEVICES

CROSS-REFERENCE TO RELATED APPLICATIONS

The present application is a continuation of U.S. application Ser. No. 17/382,486, filed Jul. 22, 2021, which is a continuation of U.S. application Ser. No. 15/828,678, filed Dec. 1, 2017, which claims priority to U.S. Provisional Application No. 62/441,653, filed Jan. 3, 2017, all of which applications are incorporated herein in their entireties.

FIELD

The present invention relates to compounds for use as emitters, and devices, such as organic light emitting diodes, including the same.

BACKGROUND

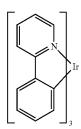
Opto-electronic devices that make use of organic mate- 25 rials are becoming increasingly desirable for a number of reasons. Many of the materials used to make such devices are relatively inexpensive, so organic opto-electronic devices have the potential for cost advantages over inorganic 30 devices. In addition, the inherent properties of organic materials, such as their flexibility, may make them well suited for particular applications such as fabrication on a flexible substrate. Examples of organic opto-electronic 35 devices include organic light emitting diodes/devices (OLEDs), organic phototransistors, organic photovoltaic cells, and organic photodetectors. For OLEDs, the organic materials may have performance advantages over conventional materials. For example, the wavelength at which an $\,^{40}$ organic emissive layer emits light may generally be readily tuned with appropriate dopants.

OLEDs make use of thin organic films that emit light when voltage is applied across the device. OLEDs are becoming an increasingly interesting technology for use in applications such as flat panel displays, illumination, and backlighting. Several OLED materials and configurations are described in U.S. Pat. Nos. 5,844,363, 6,303,238, and 5,707,745, which are incorporated herein by reference in their entirety.

One application for phosphorescent emissive molecules is a full color display. Industry standards for such a display call for pixels adapted to emit particular colors, referred to as "saturated" colors. In particular, these standards call for saturated red, green, and blue pixels. Alternatively the OLED can be designed to emit white light. In conventional liquid crystal displays emission from a white backlight is filtered using absorption filters to produce red, green and blue emission. The same technique can also be used with OLEDs. The white OLED can be either a single EML device or a stack structure. Color may be measured using CIE coordinates, which are well known to the art.

One example of a green emissive molecule is tris(2- 65 phenylpyridine) iridium, denoted Ir(ppy)₃, which has the following structure:

2



In this, and later figures herein, we depict the dative bond from nitrogen to metal (here, Ir) as a straight line.

As used herein, the term "organic" includes polymeric materials as well as small molecule organic materials that may be used to fabricate organic opto-electronic devices. "Small molecule" refers to any organic material that is not a polymer, and "small molecules" may actually be quite large. Small molecules may include repeat units in some circumstances. For example, using a long chain alkyl group as a substituent does not remove a molecule from the "small molecule" class. Small molecules may also be incorporated into polymers, for example as a pendent group on a polymer backbone or as a part of the backbone. Small molecules may also serve as the core moiety of a dendrimer, which consists of a series of chemical shells built on the core moiety. The core moiety of a dendrimer may be a fluorescent or phosphorescent small molecule emitter. A dendrimer may be a "small molecule," and it is believed that all dendrimers currently used in the field of OLEDs are small molecules.

As used herein, "top" means furthest away from the substrate, while "bottom" means closest to the substrate. Where a first layer is described as "disposed over" a second layer, the first layer is disposed further away from substrate. There may be other layers between the first and second layer, unless it is specified that the first layer is "in contact with" the second layer. For example, a cathode may be described as "disposed over" an anode, even though there are various organic layers in between.

As used herein, "solution processible" means capable of being dissolved, dispersed, or transported in and/or deposited from a liquid medium, either in solution or suspension form.

A ligand may be referred to as "photoactive" when it is believed that the ligand directly contributes to the photoactive properties of an emissive material. A ligand may be referred to as "ancillary" when it is believed that the ligand does not contribute to the photoactive properties of an emissive material, although an ancillary ligand may alter the properties of a photoactive ligand.

As used herein, and as would be generally understood by one skilled in the art, a first "Highest Occupied Molecular Orbital" (HOMO) or "Lowest Unoccupied Molecular Orbital" (LUMO) energy level is "greater than" or "higher than" a second HOMO or LUMO energy level if the first energy level is closer to the vacuum energy level. Since ionization potentials (IP) are measured as a negative energy relative to a vacuum level, a higher HOMO energy level corresponds to an IP having a smaller absolute value (an IP that is less negative). Similarly, a higher LUMO energy level corresponds to an electron affinity (EA) having a smaller absolute value (an EA that is less negative). On a conventional energy level diagram, with the vacuum level at the top, the LUMO energy level of a material is higher than the HOMO energy level of the same material. A "higher"

HOMO or LUMO energy level appears closer to the top of such a diagram than a "lower" HOMO or LUMO energy level

As used herein, and as would be generally understood by one skilled in the art, a first work function is "greater than" or "higher than" a second work function if the first work function has a higher absolute value. Because work functions are generally measured as negative numbers relative to vacuum level, this means that a "higher" work function is more negative. On a conventional energy level diagram, with the vacuum level at the top, a "higher" work function is illustrated as further away from the vacuum level in the downward direction. Thus, the definitions of HOMO and LUMO energy levels follow a different convention than work functions.

More details on OLEDs, and the definitions described above, can be found in U.S. Pat. No. 7,279,704, which is incorporated herein by reference in its entirety.

There is a need in the art for novel Pt(IV) emitters that enhance the performance of organic electroluminescent ²⁰ devices. The present invention addresses this need in the art.

SUMMARY

According to an embodiment, a compound is provided 25 that has the formula $Pt(L)_n$:

wherein Pt is Pt(IV) metal, L is a ligand coordinating to Pt, and n is an integer from 1 to 3;

wherein each L can be the same or different, and L is selected from the group consisting of bidentate, tridentate, tetradentate and hexadentate;

wherein the Pt has an octahedral coordinating geometry, and

wherein the compound is neutral.

According to another embodiment, an organic light emitting diode/device (OLED) is also provided. The OLED can include an anode, a cathode, and an organic layer, disposed between the anode and the cathode. The organic layer can include a compound of the formula Pt(L)_n. According to yet another embodiment, the organic light emitting device is incorporated into one or more devices selected from a consumer product, an electronic component module, and/or a lighting panel.

According to yet another embodiment, a formulation containing a compound of the formula Pt(L), is provided. ⁴⁵

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 shows an organic light emitting device.

FIG. 2 shows an inverted organic light emitting device 50 that does not have a separate electron transport layer.

DETAILED DESCRIPTION

Generally, an OLED comprises at least one organic layer 55 disposed between and electrically connected to an anode and a cathode. When a current is applied, the anode injects holes and the cathode injects electrons into the organic layer(s). The injected holes and electrons each migrate toward the oppositely charged electrode. When an electron and hole 60 localize on the same molecule, an "exciton," which is a localized electron-hole pair having an excited energy state, is formed. Light is emitted when the exciton relaxes via a photoemissive mechanism. In some cases, the exciton may be localized on an excimer or an exciplex. Non-radiative 65 mechanisms, such as thermal relaxation, may also occur, but are generally considered undesirable.

4

The initial OLEDs used emissive molecules that emitted light from their singlet states ("fluorescence") as disclosed, for example, in U.S. Pat. No. 4,769,292, which is incorporated by reference in its entirety. Fluorescent emission generally occurs in a time frame of less than 10 nanoseconds.

More recently, OLEDs having emissive materials that emit light from triplet states ("phosphorescence") have been demonstrated. Baldo et al., "Highly Efficient Phosphorescent Emission from Organic Electroluminescent Devices," Nature, vol. 395, 151-154, 1998; ("Baldo-I") and Baldo et al., "Very high-efficiency green organic light-emitting devices based on electrophosphorescence," Appl. Phys. Lett., vol. 75, No. 3, 4-6 (1999) ("Baldo-II"), are incorporated by reference in their entireties. Phosphorescence is described in more detail in U.S. Pat. No. 7,279,704 at cols. 5-6, which are incorporated by reference.

FIG. 1 shows an organic light emitting device 100. The figures are not necessarily drawn to scale. Device 100 may include a substrate 110, an anode 115, a hole injection layer 120, a hole transport layer 125, an electron blocking layer 130, an emissive layer 135, a hole blocking layer 140, an electron transport layer 145, an electron injection layer 150, a protective layer 155, a cathode 160, and a barrier layer 170. Cathode 160 is a compound cathode having a first conductive layer 162 and a second conductive layer 164. Device 100 may be fabricated by depositing the layers described, in order. The properties and functions of these various layers, as well as example materials, are described in more detail in U.S. Pat. No. 7,279,704 at cols. 6-10, which are incorporated by reference.

More examples for each of these layers are available. For example, a flexible and transparent substrate-anode combination is disclosed in U.S. Pat. No. 5,844,363, which is incorporated by reference in its entirety. An example of a p-doped hole transport layer is m-MTDATA doped with F4-TCNQ at a molar ratio of 50:1, as disclosed in U.S. Patent Application Publication No. 2003/0230980, which is incorporated by reference in its entirety. Examples of emissive and host materials are disclosed in U.S. Pat. No. 6,303,238 to Thompson et al., which is incorporated by reference in its entirety. An example of an n-doped electron transport layer is BPhen doped with Li at a molar ratio of 1:1, as disclosed in U.S. Patent Application Publication No. 2003/0230980, which is incorporated by reference in its entirety. U.S. Pat. Nos. 5,703,436 and 5,707,745, which are incorporated by reference in their entireties, disclose examples of cathodes including compound cathodes having a thin layer of metal such as Mg:Ag with an overlying transparent, electrically-conductive, sputter-deposited ITO layer. The theory and use of blocking layers is described in more detail in U.S. Pat. No. 6,097,147 and U.S. Patent Application Publication No. 2003/0230980, which are incorporated by reference in their entireties. Examples of injection layers are provided in U.S. Patent Application Publication No. 2004/0174116, which is incorporated by reference in its entirety. A description of protective layers may be found in U.S. Patent Application Publication No. 2004/0174116, which is incorporated by reference in its entirety.

FIG. 2 shows an inverted OLED 200. The device includes a substrate 210, a cathode 215, an emissive layer 220, a hole transport layer 225, and an anode 230. Device 200 may be fabricated by depositing the layers described, in order. Because the most common OLED configuration has a cathode disposed over the anode, and device 200 has cathode 215 disposed under anode 230, device 200 may be referred

to as an "inverted" OLED. Materials similar to those described with respect to device 100 may be used in the corresponding layers of device 200. FIG. 2 provides one example of how some layers may be omitted from the structure of device 100.

The simple layered structure illustrated in FIGS. 1 and 2 is provided by way of non-limiting example, and it is understood that embodiments of the invention may be used in connection with a wide variety of other structures. The specific materials and structures described are exemplary in 10 nature, and other materials and structures may be used. Functional OLEDs may be achieved by combining the various layers described in different ways, or layers may be omitted entirely, based on design, performance, and cost factors. Other layers not specifically described may also be 15 included. Materials other than those specifically described may be used. Although many of the examples provided herein describe various layers as comprising a single material, it is understood that combinations of materials, such as a mixture of host and dopant, or more generally a mixture, 20 may be used. Also, the layers may have various sublayers. The names given to the various layers herein are not intended to be strictly limiting. For example, in device 200, hole transport layer 225 transports holes and injects holes into emissive layer 220, and may be described as a hole 25 transport layer or a hole injection layer. In one embodiment, an OLED may be described as having an "organic layer" disposed between a cathode and an anode. This organic layer may comprise a single layer, or may further comprise multiple layers of different organic materials as described, 30 for example, with respect to FIGS. 1 and 2.

Structures and materials not specifically described may also be used, such as OLEDs comprised of polymeric materials (PLEDs) such as disclosed in U.S. Pat. No. 5,247, 190 to Friend et al., which is incorporated by reference in its entirety. By way of further example, OLEDs having a single organic layer may be used. OLEDs may be stacked, for example as described in U.S. Pat. No. 5,707,745 to Forrest et al, which is incorporated by reference in its entirety. The OLED structure may deviate from the simple layered structure illustrated in FIGS. 1 and 2. For example, the substrate may include an angled reflective surface to improve outcoupling, such as a mesa structure as described in U.S. Pat. No. 6,091,195 to Forrest et al., and/or a pit structure as described in U.S. Pat. No. 5,834,893 to Bulovic et al., which 45 are incorporated by reference in their entireties.

Unless otherwise specified, any of the layers of the various embodiments may be deposited by any suitable method. For the organic layers, preferred methods include thermal evaporation, ink-jet, such as described in U.S. Pat. 50 Nos. 6,013,982 and 6,087,196, which are incorporated by reference in their entireties, organic vapor phase deposition (OVPD), such as described in U.S. Pat. No. 6,337,102 to Forrest et al., which is incorporated by reference in its entirety, and deposition by organic vapor jet printing 55 (OVJP), such as described in U.S. Pat. No. 7,431,968, which is incorporated by reference in its entirety. Other suitable deposition methods include spin coating and other solution based processes. Solution based processes are preferably carried out in nitrogen or an inert atmosphere. For the other 60 layers, preferred methods include thermal evaporation. Preferred patterning methods include deposition through a mask, cold welding such as described in U.S. Pat. Nos. 6,294,398 and 6,468,819, which are incorporated by reference in their entireties, and patterning associated with some 65 of the deposition methods such as ink-jet and OVJD. Other methods may also be used. The materials to be deposited

6

may be modified to make them compatible with a particular deposition method. For example, substituents such as alkyl and aryl groups, branched or unbranched, and preferably containing at least 3 carbons, may be used in small molecules to enhance their ability to undergo solution processing. Substituents having 20 carbons or more may be used, and 3-20 carbons is a preferred range. Materials with asymmetric structures may have better solution processibility than those having symmetric structures, because asymmetric materials may have a lower tendency to recrystallize. Dendrimer substituents may be used to enhance the ability of small molecules to undergo solution processing.

Devices fabricated in accordance with embodiments of the present invention may further optionally comprise a barrier layer. One purpose of the barrier layer is to protect the electrodes and organic layers from damaging exposure to harmful species in the environment including moisture, vapor and/or gases, etc. The barrier layer may be deposited over, under or next to a substrate, an electrode, or over any other parts of a device including an edge. The barrier layer may comprise a single layer, or multiple layers. The barrier layer may be formed by various known chemical vapor deposition techniques and may include compositions having a single phase as well as compositions having multiple phases. Any suitable material or combination of materials may be used for the barrier layer. The barrier layer may incorporate an inorganic or an organic compound or both. The preferred barrier layer comprises a mixture of a polymeric material and a non-polymeric material as described in U.S. Pat. No. 7,968,146, PCT Pat. Application Nos. PCT/ US2007/023098 and PCT/US2009/042829, which are herein incorporated by reference in their entireties. To be considered a "mixture", the aforesaid polymeric and nonpolymeric materials comprising the barrier layer should be deposited under the same reaction conditions and/or at the same time. The weight ratio of polymeric to non-polymeric material may be in the range of 95:5 to 5:95. The polymeric material and the non-polymeric material may be created from the same precursor material. In one example, the mixture of a polymeric material and a non-polymeric material consists essentially of polymeric silicon and inorganic

Devices fabricated in accordance with embodiments of the invention can be incorporated into a wide variety of electronic component modules (or units) that can be incorporated into a variety of electronic products or intermediate components. Examples of such electronic products or intermediate components include display screens, lighting devices such as discrete light source devices or lighting panels, etc. that can be utilized by the end-user product manufacturers. Such electronic component modules can optionally include the driving electronics and/or power source(s). Devices fabricated in accordance with embodiments of the invention can be incorporated into a wide variety of consumer products that have one or more of the electronic component modules (or units) incorporated therein. A consumer product comprising an OLED that includes the compound of the present disclosure in the organic layer in the OLED is disclosed. Such consumer products would include any kind of products that include one or more light source(s) and/or one or more of some type of visual displays. Some examples of such consumer products include flat panel displays, computer monitors, medical monitors, televisions, billboards, lights for interior or exterior illumination and/or signaling, heads-up displays, fully or partially transparent displays, flexible displays, laser printers, telephones, mobile phones, tablets, phablets, per-

sonal digital assistants (PDAs), wearable devices, laptop computers, digital cameras, camcorders, viewfinders, microdisplays (displays that are less than 2 inches diagonal), 3-D displays, virtual reality or augmented reality displays, vehicles, video walls comprising multiple displays tiled together, theater or stadium screen, and a sign. Various control mechanisms may be used to control devices fabricated in accordance with the present invention, including passive matrix and active matrix. Many of the devices are intended for use in a temperature range comfortable to humans, such as 18 degrees C. to 30 degrees C., and more preferably at room temperature (20-25 degrees C.), but could be used outside this temperature range, for example, from -40 degree C. to +80 degree C.

The materials and structures described herein may have 15 applications in devices other than OLEDs. For example, other optoelectronic devices such as organic solar cells and organic photodetectors may employ the materials and structures. More generally, organic devices, such as organic transistors, may employ the materials and structures.

The term "halo," "halogen," or "halide" as used herein includes fluorine, chlorine, bromine, and iodine.

The term "alkyl" as used herein contemplates both straight and branched chain alkyl radicals. Preferred alkyl groups are those containing from one to fifteen carbon atoms 25 and includes methyl, ethyl, propyl, 1-methylethyl, butyl, 1-methylpropyl, 2-methylpropyl, pentyl, 1-methylbutyl, 2-methylbutyl, 3-methylbutyl, 1,1-dimethylpropyl, 1,2-dimethylpropyl, 2,2-dimethylpropyl, and the like. Additionally, the alkyl group may be optionally substituted.

The term "cycloalkyl" as used herein contemplates cyclic alkyl radicals. Preferred cycloalkyl groups are those containing 3 to 10 ring carbon atoms and includes cyclopropyl, cyclopentyl, cyclohexyl, adamantyl, and the like. Additionally, the cycloalkyl group may be optionally substituted.

The term "alkenyl" as used herein contemplates both straight and branched chain alkene radicals. Preferred alkenyl groups are those containing two to fifteen carbon atoms. Additionally, the alkenyl group may be optionally substituted

The term "alkynyl" as used herein contemplates both straight and branched chain alkyne radicals. Preferred alkynyl groups are those containing two to fifteen carbon atoms. Additionally, the alkynyl group may be optionally substituted.

The terms "aralkyl" or "arylalkyl" as used herein are used interchangeably and contemplate an alkyl group that has as a substituent an aromatic group. Additionally, the aralkyl group may be optionally substituted.

The term "heterocyclic group" as used herein contemplates aromatic and non-aromatic cyclic radicals. Heteroaromatic cyclic radicals also means heteroaryl. Preferred hetero-non-aromatic cyclic groups are those containing 3 to 7 ring atoms which includes at least one hetero atom, and includes cyclic amines such as morpholino, piperidino, 55 pyrrolidino, and the like, and cyclic ethers, such as tetrahydrofuran, tetrahydropyran, and the like. Additionally, the heterocyclic group may be optionally substituted.

The term "aryl" or "aromatic group" as used herein contemplates single-ring groups and polycyclic ring systems. The polycyclic rings may have two or more rings in which two carbons are common to two adjoining rings (the rings are "fused") wherein at least one of the rings is aromatic, e.g., the other rings can be cycloalkyls, cycloalkenyls, aryl, heterocycles, and/or heteroaryls. Preferred aryl 65 groups are those containing six to thirty carbon atoms, preferably six to twenty carbon atoms, more preferably six

8

to twelve carbon atoms. Especially preferred is an aryl group having six carbons, ten carbons or twelve carbons. Suitable aryl groups include phenyl, biphenyl, triphenyl, triphenylene, tetraphenylene, naphthalene, anthracene, phenalene, phenanthrene, fluorene, pyrene, chrysene, perylene, and azulene, preferably phenyl, biphenyl, triphenylene, fluorene, and naphthalene. Additionally, the aryl group may be optionally substituted.

The term "heteroaryl" as used herein contemplates singlering hetero-aromatic groups that may include from one to five heteroatoms. The term heteroaryl also includes polycyclic hetero-aromatic systems having two or more rings in which two atoms are common to two adjoining rings (the rings are "fused") wherein at least one of the rings is a heteroaryl, e.g., the other rings can be cycloalkyls, cycloalkenyls, aryl, heterocycles, and/or heteroaryls. Preferred heteroaryl groups are those containing three to thirty carbon atoms, preferably three to twenty carbon atoms, more preferably three to twelve carbon atoms. Suitable heteroaryl 20 groups include dibenzothiophene, dibenzofuran, dibenzoselenophene, furan, thiophene, benzofuran, benzothiophene, benzoselenophene, carbazole, indolocarbazole, pyridylindole, pyrrolodipyridine, pyrazole, imidazole, triazole, oxazole, thiazole, oxadiazole, oxatriazole, dioxazole, thiadiazole, pyridine, pyridazine, pyrimidine, pyrazine, triazine, oxazine, oxathiazine, oxadiazine, indole, benzimidazole, indazole, indoxazine, benzoxazole, benzisoxazole, benzothiazole, quinoline, isoquinoline, cinnoline, quinazoline, quinoxaline, naphthyridine, phthalazine, pteridine, xanthene, acridine, phenazine, phenothiazine, phenoxazine, benzofuropyridine, furodipyridine, benzothienopyridine, thienodipyridine, benzoselenophenopyridine, and selenophenodipyridine, preferably dibenzothiophene, dibenzofuran, dibenzoselenophene, carbazole, indolocarbazole, imi-35 dazole, pyridine, triazine, benzimidazole, 1,2-azaborine, 1,3-azaborine, 1,4-azaborine, borazine, and aza-analogs thereof. Additionally, the heteroaryl group may be optionally

The alkyl, cycloalkyl, alkenyl, alkynyl, aralkyl, heterocyclic group, aryl, and heteroaryl may be unsubstituted or may
be substituted with one or more substituents selected from
the group consisting of deuterium, halogen, alkyl, cycloalkyl, heteroalkyl, arylalkyl, alkoxy, aryloxy, amino, cyclic
amino, silyl, alkenyl, cycloalkenyl, heteroalkenyl, alkynyl,
aryl, heteroaryl, acyl, carbonyl, carboxylic acid, ether, ester,
nitrile, isonitrile, sulfanyl, sulfinyl, sulfonyl, phosphino, and
combinations thereof.

As used herein, "substituted" indicates that a substituent other than H is bonded to the relevant position, such as carbon. Thus, for example, where R^1 is mono-substituted, then one R^1 must be other than H. Similarly, where R^1 is di-substituted, then two of R^1 must be other than H. Similarly, where R^1 is unsubstituted, R^1 is hydrogen for all available positions.

The "aza" designation in the fragments described herein, i.e. aza-dibenzofuran, aza-dibenzothiophene, etc. means that one or more of the C—H groups in the respective fragment can be replaced by a nitrogen atom, for example, and without any limitation, azatriphenylene encompasses both dibenzo [f,h] quinoxaline and dibenzo [f,h] quinoline. One of ordinary skill in the art can readily envision other nitrogen analogs of the aza-derivatives described above, and all such analogs are intended to be encompassed by the terms as set forth herein.

It is to be understood that when a molecular fragment is described as being a substituent or otherwise attached to another moiety, its name may be written as if it were a

fragment (e.g. phenyl, phenylene, naphthyl, dibenzofuryl) or as if it were the whole molecule (e.g. benzene, naphthalene, dibenzofuran). As used herein, these different ways of designating a substituent or attached fragment are considered to be equivalent.

Compounds of the Invention

In one aspect, the present invention includes novel charge-neutral Pt(IV) compounds with multi-dentate ligands. Pt(IV) compounds having octahedral geometry are hypothesized to show better physical and chemical stability compared with square planar Pt(II) compounds. Because of the strong spin-orbital coupling, similar to Ir(III) compounds, cyclometalated Pt(IV) compounds may exhibit tunable phosphorescence with high emission quantum yield. These materials may be useful for applications in organic electroluminescent devices to enhance device performance.

In one aspect, the present invention includes a compound 20 having the formula $Pt(L)_n$;

wherein Pt is Pt(IV) metal, L is a ligand coordinating to Pt, and n is an integer from 1 to 3;

wherein each L can be the same or different, and L is selected from the group consisting of bidentate, triden- 25 tate, tetradentate and hexadentate;

wherein the Pt has an octahedral coordinating geometry,

wherein the compound is neutral.

In one embodiment, the compound is selected from the $^{\,30}$ group consisting of: group consisting of PtL^1L^2 , PtL^3L^4 , and $PtL^1L^5L^6$; wherein L^1 , L^5 , and L^6 are bidentate ligands; L^2 is

tetradentate ligand; L3 and L4 are tridentate ligands; and wherein within each compound, all ligands are optionally joined together to form a tetradentate or a hexadentate ligand.

In one embodiment, each L is independently selected from the group consisting of:

wherein X is a neutral coordinating atom selected from the group consisting of carbene, carbon, phosphorus, and nitrogen and each X can be the same or different; wherein Y is an anionic coordinating atom selected from 65 the group consisting of carbon, nitrogen, and oxygen atom and each Y can be the same or different;

60

10

In one embodiment, the neutral carbene carbon is an N-heterocyclic carbene carbon; wherein the neutral phosphorus is a phosphorus atom of a trisubstituted phosphine; and wherein the neutral nitrogen is an sp² nitrogen atom of an N-heterocyclic ring selected from the group consisting of pyridine, imidazole, pyrazole, and triazole.

In one embodiment, the anionic coordinating carbon is an sp² carbon atom selected from the group consisting of benzene, pyridine, furan, thiophene, and pyrrole.

In one embodiment, the anionic coordinating nitrogen is an sp² nitrogen atom of an N-heterocyclic ring selected from the group consisting of imidazole, pyrazole, pyrrole, and triazole.

As used herein, the term "N-heterocyclic ring" refers to the first ring which coordinates to the metal, but does not encompass additional rings that may be a substituent on or fused to the N-heterocyclic ring. In a non-limiting example, when the nitrogen is an sp² nitrogen atom of an N-heterocyclic ring, wherein the N-heterocyclic ring is pyridine, the pyridine may be fused to or substituted with additional rings such that the sp² nitrogen atom may be an sp² nitrogen atom from other pyridine-containing groups such as, but not limited to, a quinoline or a isoquinoline.

In one embodiment, the anionic oxygen atom is an oxygen atom from a carboxylic acid or an ether.

In one embodiment, the compound is selected from the

In one embodiment, the compound is selected from the group consisting of:

-continued

$$\mathbb{R}^{d}$$

-continued
$$\begin{bmatrix}
R^A & & & \\
R^A & & & \\
R^D & & \\$$

$$\begin{bmatrix} R^{A} & & & & 15 \\ & & & & \\ R^{B} & & & & 20 \\ & & & & 25 \end{bmatrix}$$

$$\begin{array}{c|c}
R^{A} & & & & & & & \\
R^{B} & & & & & & & \\
R^{B} & & & & & & & \\
R^{D}, & & & & & & \\
\end{array}$$
30

$$\begin{array}{c|c}
R^{d} & & & & & & & & & & & & & \\
\hline
 & N & & & & & & & & & & & & \\
N & & & & & & & & & & & & \\
\hline
 & R^{D}, & & & & & & & & & \\
\end{array}$$
50

$$\begin{array}{c|c}
R^{A} & & \\
\hline
N & \\
\end{array}$$

$$\begin{array}{c}
R^{C} & \\
\end{array}$$

$$\begin{array}{c}
60 \\
\end{array}$$

$$\begin{array}{c}
65 \\
\end{array}$$

-continued
$$\mathbb{R}^{\mathcal{C}}$$

$$\mathbb{R}^{\mathcal{A}}$$

$$\mathbb{R}^{\mathcal{D}}$$

$$\mathbb{R}^{\mathcal{D}}$$

$$\begin{bmatrix} R^{A} & & & \\ R^{E} - N & & & \\ R^{B} & & & & \\ \end{bmatrix}_{2} Pt \begin{bmatrix} R^{C} & & & \\ R^{C} & & & \\ R^{D} & & & \\ \end{bmatrix}_{2} R^{D},$$

$$\begin{bmatrix} R^{A} & & & \\ R^{E} - N & & & \\ R^{B} & & & & \\ \end{bmatrix}$$

$$\begin{bmatrix} R^{4} & \\ N & \\ R^{B} & \\ \end{bmatrix}_{2} Pt \begin{bmatrix} R^{C} & \\ R^{D} & \\ R^{D} & \\ \end{bmatrix}$$

$$\begin{bmatrix} R^{A} & N & R^{C} \\ R^{B} & N & R^{D}, \end{bmatrix}$$

$$\begin{bmatrix} R^A \\ N \end{bmatrix} P t \begin{bmatrix} R^C \\ R^D \end{bmatrix}$$

$$\begin{bmatrix} R^{A} & 25 \\ R^{B} & 1 \end{bmatrix}$$

$$R^{C} & 30$$

$$\begin{bmatrix}
R^A \\
N
\end{bmatrix}$$

$$\begin{bmatrix}
R^B \\
\end{bmatrix}$$

-continued
$$\mathbb{R}^d$$
 \mathbb{R}^d \mathbb{R}^D ,

$$\begin{bmatrix} R^A & & \\ R^B & & \\ R^D & & \\ R^$$

$$\begin{bmatrix} R^{A} & & & \\ & N & & \\ & R^{B} & & \end{bmatrix}_{2} Pt_{O} R^{D},$$

$$\mathbb{R}^{A}$$

$$\mathbb{R}^{B}$$

$$\mathbb{R}^{D}$$

$$\mathbb{R}^{D}$$

$$\mathbb{R}^{D}$$

$$\mathbb{R}^{D}$$

$$\mathbb{R}^{D}$$

$$\begin{array}{c|c}
R^{4} & & & \\
R^{P} & & & \\
\end{array}$$

$$\begin{array}{c|c}
R^{P} & & & \\
\end{array}$$

$$\begin{array}{c|c}
\end{array}$$

$$\begin{array}{c|c}
\end{array}$$

$$\begin{array}{c|c}
\end{array}$$

$$\begin{array}{c|c}
\end{array}$$

$$\end{array}$$

$$\begin{array}{c|c}
\end{array}$$

$$\end{array}$$

$$\mathbb{R}^{A}$$
 \mathbb{R}^{B}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}

$$\begin{bmatrix} R^A & & & & & & & & & & \\ & R^A & & & & & & & & \\ & & & & & & & & & \\ & & & & & & & & & \\ & & & & & & & & & \\ & & & & & & & & \\ & & & & & & & & \\ & & & & & & & & \\ & & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & \\ & \\ & & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\$$

$$\begin{array}{c|c}
R^{A} & R^{C} \\
\hline
R^{E} - N & Pt \\
\hline
R^{B} & R^{D},
\end{array}$$
50

$$\mathbb{R}^{d}$$
 \mathbb{R}^{d}
 \mathbb{R}^{d}

-continued
$$\mathbb{R}^{A}$$
 \mathbb{R}^{B} \mathbb{R}^{B} \mathbb{R}^{B}

$$\begin{bmatrix} R^4 & & & \\ & & &$$

$$\begin{bmatrix} R^{A} \\ N \end{bmatrix}_{2} Pt \\ N \\ N \end{bmatrix}_{2}$$

$$\begin{bmatrix} R^{A} \\ N \\ \end{bmatrix}_{2} P_{1} N$$

$$\begin{bmatrix} R^{A} \\ N \end{bmatrix}_{2}$$

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{E}$$
 \mathbb{R}^{E}
 \mathbb{R}^{E}

$$R^{R}$$
 R^{R}
 R^{R

-continued
$$\mathbb{R}^B$$
 \mathbb{R}^C \mathbb{R}^C \mathbb{R}^D , \mathbb{R}^D

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}

-continued
$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{F}$$
 \mathbb{R}^{D}
 \mathbb{R}^{D}
 \mathbb{R}^{D}

60

65

5 Ref. and

20
$$\mathbb{R}^B$$
 \mathbb{R}^A
 \mathbb{R}^E
 \mathbb{R}^A
 \mathbb{R}^E
 \mathbb{R}^B
 \mathbb{R}^B

wherein R^A, R^B, R^C, R^D, R^E, R^F, R^G, R, and R' each independently represent mono to the maximum possible number of substitutions, or no substitution;

wherein Y is selected from the group consisting of BR, NR, PR, O, S, Se, C=O, S=O, SO2, CRR', SIRR', and GeRR';

wherein R^A , R^B , R^C , R^D , R^E , R^F , R^G , R, R' and Z are each independently selected from the group consisting of hydrogen, deuterium, halide, alkyl, cycloalkyl, heteroalkyl, arylalkyl, aryl, heteroaryl, alkoxy, aryloxy, amino, silyl, and combinations thereof; and

wherein any adjacent RA, RB, RC, RD, RE, RF, RG, R, and R' are optionally joined to form a ring.

In one embodiment, the compound is selected from the group consisting of:

Compound 2 15

Compound 3

20

25

30

35

40

45

50

-continued Compound 1

Compound 6
$$Pt$$

$$\begin{bmatrix} & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & \\ & & \\ & & \\ & & \\ & \\ & & \\ & \\ & & \\ & & \\ &$$

Compound 9
$$CD_3$$
 CD_3 Pt

Compound 10
$$P_t$$
, P_t , P_t

Compound 11

-continued

-continued

Compound 16
$$D_3C$$
 P_1 , CD_3 CD_3

25

-continued

$$\begin{bmatrix} \\ \\ \\ \\ \\ \end{bmatrix}_2$$
 Compound 27

-continued

-continued

15

Compound 32

Compound 33

Compound 36

Compound 35

35

40

45

50

55

60

65

$$CD_3$$
 P_t
 P_t

Compound 37

-continued

-continued

15

Compound 45

Compound 41 30

Compound 46

Compound 42

35

40

45

50

60

65

$$\left[\begin{array}{c} N \\ \end{array}\right]_2$$

Compound 48

Compound 47

Compound 43 55

-continued

Compound 50
$$\begin{array}{c} D_3C \\ \hline \\ N \\ \hline \\ \end{array}$$

$$\begin{array}{c} 20 \\ \hline \\ \end{array}$$

$$\begin{array}{c} 25 \\ \hline \\ \end{array}$$

$$\begin{array}{c} 30 \\ \hline \\ \end{array}$$

Compound 51
$$_{35}$$

$$\begin{array}{c} D_3C \\ \\ \\ \\ \\ \\ \\ \end{array}$$

$$\begin{array}{c} \\ \\ \\ \\ \end{array}$$

Compound 52
$$D_3C$$
 CD_3 CD_3

Compound 55
$$Pt$$

10 Compound 60, wherein Y = O Compound 61, wherein Y = S Compound 62, wherein Y = Se 15

20

25

30

35

40

45

50

55

60

65

Compound 63, wherein Y = O Compound 64, wherein Y = S Compound 65, wherein Y = Se

Compound 66, wherein Y = OCompound 67, wherein Y = SCompound 68, wherein Y = Se

Compound 69, wherein Y = OCompound 70, wherein Y = SCompound 71, wherein Y = Se

Compound 72, wherein Y = OCompound 73, wherein Y = SCompound 74, wherein Y = Se

Compound 75 Compound 76

Compound 77

 $Compound \ 78$

35

40

45

50

65

41

-continued

42

Compound 87 wherein Y = O Compound 88, wherein Y = S Compound 89, wherein Y = Se

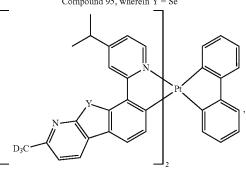
Compound 90, wherein Y = OCompound 91, wherein Y = SCompound 92, wherein Y = Se

Compound 81, wherein Y = OCompound 82, wherein Y = SCompound 83, wherein Y = Se

$$D_3C$$
 D_3C
 D_3C
 D_3C
 D_3C
 D_3C

Compound 93, wherein Y = OCompound 94, wherein Y = SCompound 95, wherein Y = Se

Compound 84, wherein Y = O Compound 85, wherein Y = S Compound 86, wherein Y = Se



Compound 96, wherein Y = OCompound 97, wherein Y = SCompound 98, wherein Y = Se

15

35

40

45

50

55

60

65

$$D_3C$$

Compound 99, wherein Y = OCompound 100, wherein Y = SCompound 101, wherein Y = Se

Compound 111, wherein Y = O Compound 112, wherein Y = S Compound 113, wherein Y = Se

Compound 102, wherein Y = O Compound 103, wherein Y = S Compound 104, wherein Y = Se

Compound 115

Compound 105, wherein Y = O Compound 106, wherein Y = S Compound 107, wherein Y = Se

Compound 117

Compound 116

Compound 108, wherein Y = OCompound 109, wherein Y = SCompound 110, wherein Y = Se

-continued

-continued Compound 122
$$CD_3$$
 CD_3 CD_3 CD_3 CD_3 CD_3 $Compound 123$

-continued

Company 1

Compound 127

-continued

Compound 132

Compound 133

15

20

25

30

Compound 128

Compound 134

Compound 129

 $Compound \ 130$



45

50

N Pri

Compound 135

Compound 131 55

Compound 136

65

15

20

25

30

-continued

Compound 137

Compound 144, wherein Y = O Compound 145, wherein Y = S Compound 146, wherein Y = Se

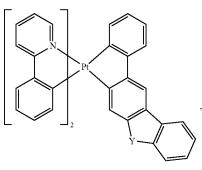
Compound 138

Compound 147, wherein Y = O Compound 148, wherein Y = S Compound 149, wherein Y = Se

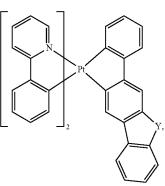
Compound 139

Compound 140

Compound 141, wherein Y = O Compound 142, wherein Y = S Compound 143, wherein Y = Se 65



Compound 150, wherein Y = O Compound 151, wherein Y = S Compound 152, wherein Y = Se

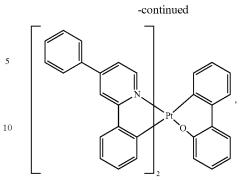


Compound 153, wherein Y = O Compound 154, wherein Y = S Compound 155, wherein Y = Se

-continued

$$\begin{bmatrix} D_3C & & & \\ & & \\ & & & \\ & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$$

Compound 156



15

20

25

Compound 158

Compound 159

Compound 160

60

65

Compound 161 55

32

Compound 162

Compound 164

-continued

Compound 167

35

50

Compound 169

Compound 170, wherein Y = OCompound 171, wherein Y = SCompound 172, wherein Y = Se

Compound 173, wherein Y = O Compound 174, wherein Y = S Compound 175, wherein Y = Se 65 -continued

Compound 177

Compound 176

Compound 178

40

45

50

-continued

Compound 180 10 15

Compound 181 20 25 30

-continued

Compound 184

Compound 186

15

20

25

30

40

45

50

-continued

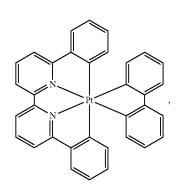
Compound 188

Compound 192

Compound 189

Compound 193

Compound 190



Compound 194

Compound 191

15

30

35

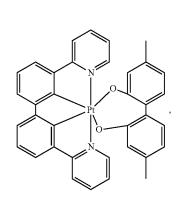
-continued

Compound 196

Compound 200

Compound 201

Compound 198 40 45 50



Compound 202

Compound 199 55 60 65

15

-continued

Compound 204

Compound 208

Compound 205

According to another aspect of the present disclosure, an OLED is also provided. The OLED includes an anode, a cathode, and an organic layer disposed between the anode and the cathode. The organic layer may include a host and a phosphorescent dopant. The organic layer can include a compound of having formula Pt(L),, and its variations as described herein.

In some embodiments, the OLED has one or more characteristics selected from the group consisting of being flexible, being rollable, being foldable, being stretchable, and being curved. In some embodiments, the OLED is transparent or semi-transparent. In some embodiments, the 35 OLED further comprises a layer comprising carbon nano-

Compound 207

and

Compound 206

In some embodiments, the OLED further comprises a layer comprising a delayed fluorescent emitter. In some embodiments, the OLED comprises a RGB pixel arrange-40 ment or white plus color filter pixel arrangement. In some embodiments, the OLED is a mobile device, a hand held device, or a wearable device. In some embodiments, the OLED is a display panel having less than 10 inch diagonal or 50 square inch area. In some embodiments, the OLED is 45 a display panel having at least 10 inch diagonal or 50 square inch area. In some embodiments, the OLED is a lighting panel.

In one embodiment, the OLED is selected from the group consisting of a consumer product, an electronic component 50 module, an organic light-emitting device, and a lighting panel.

In one embodiment, the consumer product is selected from the group consisting of a flat panel display, a computer monitor, a medical monitor, a television, a billboard, a light for interior or exterior illumination and/or signaling, a heads-up display, a fully or partially transparent display, a flexible display, a laser printer, a telephone, a cell phone, tablet, a phablet, a personal digital assistant (PDA), a wearable device, a laptop computer, a digital camera, a 60 camcorder, a viewfinder, a micro-display that is less than 2 inches diagonal, a 3-D display, a virtual reality or augmented reality display, a vehicle, a video walls comprising multiple displays tiled together, a theater or stadium screen, and a sign.

In one embodiment, the organic layer is an emissive layer and the compound is an emissive dopant or a non-emissive dopant.

In some embodiments of the emissive region, the emissive region further comprises a host, wherein the host comprises at least one selected from the group consisting of metal complex, triphenylene, carbazole, dibenzothiophene, dibenzofuran, dibenzoselenophene, aza-triphenylene, aza-carbazole, aza-dibenzothiophene, aza-dibenzofuran, and aza-dibenzoselenophene.

In some embodiment of the emissive region, the emissive region further comprises a host, wherein the host is selected $_{10}$ from the group consisting of:

and combinations thereof.

In some embodiments, the compound can be an emissive dopant. In some embodiments, the compound can produce emissions via phosphorescence, fluorescence, thermally activated delayed fluorescence, i.e., TADF (also referred to as E-type delayed fluorescence), triplet-triplet annihilation, or combinations of these processes.

According to another aspect, a formulation comprising $_{20}$ the compound described herein is also disclosed.

The OLED disclosed herein can be incorporated into one or more of a consumer product, an electronic component module, and a lighting panel. The organic layer can be an emissive layer and the compound can be an emissive dopant ²⁵ in some embodiments, while the compound can be a non-emissive dopant in other embodiments.

The organic layer can also include a host. In some embodiments, two or more hosts are preferred. In some 30 embodiments, the hosts used maybe a) bipolar, b) electron transporting, c) hole transporting or d) wide band gap materials that play little role in charge transport. In some embodiments, the host can include a metal complex. The host can be a triphenylene containing benzo-fused thiophene 35 or benzo-fused furan. Any substituent in the host can be an unfused substituent independently selected from the group $\begin{array}{l} \text{consisting of } C_n H_{2n+1}, \ \ OC_n H_{2n+1}, \ \ OAr_1, \ \ N(C_n H_{2n+1})_2, \\ N(Ar_1)(Ar_2), \ \ CH = CH - C_n H_{2n+1}, \ \ C = C - C_n H_{2n+1}, \ \ Ar_1, \end{array}$ Ar_1 — Ar_2 , and C_nH_{2n} — Ar_1 , or the host has no substitutions. 40 In the preceding substituents n can range from 1 to 10; and Ar₁ and Ar₂ can be independently selected from the group consisting of benzene, biphenyl, naphthalene, triphenylene, carbazole, and heteroaromatic analogs thereof. The host can be an inorganic compound. For example, a Zn containing 45 inorganic material e.g. ZnS.

The host can be a compound comprising at least one chemical group selected from the group consisting of triphenylene, carbazole, dibenzothiophene, dibenzofuran, dibenzoselenophene, azatriphenylene, azacarbazole, aza-dibenzothiophene, aza-dibenzofuran, and aza-dibenzoselenophene. The host can include a metal complex. The host can be, but is not limited to, a specific compound selected from the group consisting of:

35

40

45

50

and combinations thereof.

Additional information on possible hosts is provided below. In yet another aspect of the present disclosure, a formulation that comprises the novel compound disclosed herein is described. The formulation can include one or more components selected from the group consisting of a solvent, a host, a hole injection material, hole transport material, and 60 an electron transport layer material, disclosed herein. Combination with Other Materials

The materials described herein as useful for a particular layer in an organic light emitting device may be used in combination with a wide variety of other materials present 65 in the device. For example, emissive dopants disclosed herein may be used in conjunction with a wide variety of

hosts, transport layers, blocking layers, injection layers, electrodes and other layers that may be present. The materials described or referred to below are non-limiting examples of materials that may be useful in combination with the compounds disclosed herein, and one of skill in the art can readily consult the literature to identify other materials that may be useful in combination. Conductivity Dopants:

A charge transport layer can be doped with conductivity dopants to substantially alter its density of charge carriers, which will in turn alter its conductivity. The conductivity is increased by generating charge carriers in the matrix material, and depending on the type of dopant, a change in the Fermi level of the semiconductor may also be achieved. Hole-transporting layer can be doped by p-type conductivity dopants and n-type conductivity dopants are used in the electron-transporting layer.

Non-limiting examples of the conductivity dopants that may be used in an OLED in combination with materials disclosed herein are exemplified below together with references that disclose those materials: EP01617493, EP01968131, EP2020694, EP2684932, US20050139810, US20070160905, US20090167167, US2010288362, WO06081780, WO2009003455, WO2009008277, WO2009011327, WO2014009310, US2007252140, US2015060804 and US201246012.

A hole injecting/transporting material to be used in the present invention is not particularly limited, and any compound may be used as long as the compound is typically used as a hole injecting/transporting material. Examples of the material include, but are not limited to: a phthalocyanine or porphyrin derivative; an aromatic amine derivative; an indolocarbazole derivative; a polymer containing fluorohydrocarbon; a polymer with conductivity dopants; a conducting polymer, such as PEDOT/PSS; a self-assembly monomer derived from compounds such as phosphonic acid and silane derivatives; a metal oxide derivative, such as MoO_x; a p-type semiconducting organic compound, such as 1,4,5, 8,9,12-Hexaazatriphenylenehexacarbonitrile; a metal complex, and a cross-linkable compounds.

HIL/HTL

Examples of aromatic amine derivatives used in HIL or HTL include, but not limit to the following general structures:

$$Ar^{2}$$
 Ar^{3}
 Ar^{3}
 Ar^{3}
 Ar^{3}
 Ar^{4}
 Ar^{5}
 Ar^{5}
 Ar^{4}
 Ar^{5}
 A

-continued

$$Ar^{4} \qquad Ar^{5}$$

$$Ar^{1} \qquad Ar^{2}$$

$$Ar^{3} \qquad Ar^{9}$$

$$Ar^{7} \qquad Ar^{8}$$

Each of Ar¹ to Ar⁹ is selected from the group consisting of aromatic hydrocarbon cyclic compounds such as benzene, biphenyl, triphenyl, triphenylene, naphthalene, anthracene, phenalene, phenanthrene, fluorene, pyrene, chrysene, perylene, and azulene; the group consisting of aromatic heterocyclic compounds such as dibenzothiophene, dibenzofuran, dibenzoselenophene, furan, thiophene, benzofuran, benzothiophene, benzoselenophene, carbazole, indolocarbazole, pyridylindole, pyrrolodipyridine, pyrazole, imidazole, triazole, oxazole, thiazole, oxadiazole, oxatriazole, dioxazole, thiadiazole, pyridine, pyridazine, pyrimidine, pyrazine, triazine, oxazine, oxathiazine, oxadiazine, indole, benzimidazole, indazole, indoxazine, benzoxazole, benzisoxazole, 25 benzothiazole, quinoline, isoquinoline, cinnoline, quinazoline, quinoxaline, naphthyridine, phthalazine, pteridine, xanthene, acridine, phenazine, phenothiazine, phenoxazine, benzofuropyridine, furodipyridine, benzothienopyridine, thienodipyridine, benzoselenophenopyridine, and seleno- 30 phenodipyridine; and the group consisting of 2 to 10 cyclic structural units which are groups of the same type or different types selected from the aromatic hydrocarbon cyclic group and the aromatic heterocyclic group and are bonded to each other directly or via at least one of oxygen atom, nitrogen atom, sulfur atom, silicon atom, phosphorus atom, boron atom, chain structural unit and the aliphatic cyclic group. Each Ar may be unsubstituted or may be substituted by a substituent selected from the group consisting of deuterium, halide, alkyl, cycloalkyl, heteroalkyl, arylalkyl, alkoxy, aryloxy, amino, silyl, alkenyl, cycloalkenyl, heteroalkenyl, alkynyl, aryl, heteroaryl, acyl, carbonyl, carboxylic acids, ester, nitrile, isonitrile, sulfanyl, sulfinyl, sulfonyl, phosphino, and combinations thereof.

In one aspect, Ar¹ to Ar⁹ is independently selected from the group consisting of:

wherein k is an integer from 1 to 20; X^{101} to X^{108} is C (including CH) or N; Z^{101} is NAr^1 , O, or S; Ar^1 has the same group defined above.

Examples of metal complexes used in HIL or HTL include, but are not limited to the following general formula:

$$\begin{bmatrix} \begin{pmatrix} Y^{101} \\ Y^{102} \end{pmatrix}_{k'} \text{Met} & (L^{101})k'' \end{bmatrix}$$

wherein Met is a metal, which can have an atomic weight greater than 40; $(Y^{101}_Y^{102})$ is a bidentate ligand, Y^{101} and Y^{102} are independently selected from C, N, O, P, and S; L^{101} is an ancillary ligand; k' is an integer value from 1 to the maximum number of ligands that may be attached to the metal; and k'+k" is the maximum number of ligands that may be attached to the metal.

In one aspect, $(Y^{101}_Y^{102})$ is a 2-phenylpyridine derivative. In another aspect, $(Y^{101}_Y^{102})$ is a carbene ligand. In another aspect, Met is selected from Ir, Pt, Os, and Zn. In a further aspect, the metal complex has a smallest oxidation potential in solution vs. Fc⁺/Fc couple less than about 0.6 V.

Non-limiting examples of the HIL and HTL materials that may be used in an OLED in combination with materials disclosed herein are exemplified below together with refer-50 ences that disclose those materials: CN102702075, DE102012005215, EP01624500. EP01698613. EP01806334, EP01930964, EP01972613, EP01997799, EP02011790, EP02055700, EP02055701, EP1725079, EP2660300, EP650955, JP07-073529, EP2085382. 55 JP2005112765, JP2007091719, JP2008021687, JP2014-009196, KR20110088898, KR20130077473, TW201139402, U.S. Pat. No. 6,517,957, US20020158242, US20030162053, US20050123751, US20060182993, US20060240279, US20070145888, US20070181874, US20070278938. US20080014464. US20080091025. US20080106190, US20080124572, US20080145707, US20080220265, US20080233434, US20080303417, US2008107919, US20090115320, US20090167161, US2009066235, US2011007385, US20110163302, US2011240968, US2011278551, US2012205642, US2013241401, US20140117329, US2014183517, U.S. Nos. 5,061,569, 5,639,914, WO05075451,

WO07125714, WO2009145016, WO2012177006, WO2013087142, WO08023550, WO2010061824, WO2013018530, WO2013118812,

WO08023759, WO2011075644, WO2013039073, WO2013120577, WO2013157367, WO2014015935, WO2014030921, WO2014157018. WO2013175747, WO2014015937, WO2014034791,

WO2014002873, WO2014030872, WO2014104514,

EBL

An electron blocking layer (EBL) may be used to reduce the number of electrons and/or excitons that leave the emissive layer. The presence of such a blocking layer in a device may result in substantially higher efficiencies, and or longer lifetime, as compared to a similar device lacking a blocking layer. Also, a blocking layer may be used to confine 55 emission to a desired region of an OLED. In some embodiments, the EBL material has a higher LUMO (closer to the vacuum level) and/or higher triplet energy than the emitter closest to the EBL interface. In some embodiments, the EBL material has a higher LUMO (closer to the vacuum level) and or higher triplet energy than one or more of the hosts closest to the EBL interface. In one aspect, the compound used in EBL contains the same molecule or the same functional groups used as one of the hosts described below.

The light emitting layer of the organic EL device of the present invention preferably contains at least a metal complex as light emitting material, and may contain a host material using the metal complex as a dopant material. Examples of the host material are not particularly limited, and any metal complexes or organic compounds may be used as long as the triplet energy of the host is larger than that of the dopant. Any host material may be used with any dopant so long as the triplet criteria is satisfied.

Examples of metal complexes used as host are preferred to have the following general formula:

$$\left[\left(\begin{matrix} Y^{103} \\ Y^{104} \end{matrix} \right]_{\mathcal{V}} \text{Met} \longrightarrow (L^{101})k''$$

wherein Met is a metal; $(Y^{103}_Y^{104})$ is a bidentate ligand, Y^{103} and Y^{104} are independently selected from C, N, O, P, and S; L^{101} is an another ligand; k' is an integer value from 1 to the maximum number of ligands that may be attached

to the metal; and k'+k" is the maximum number of ligands that may be attached to the metal.

In one aspect, the metal complexes are:

$$\begin{bmatrix} O \\ N \end{bmatrix}_{l'}^{l} Al - (L^{101})_{3-l'} \begin{bmatrix} O \\ N \end{bmatrix}_{l'}^{l} Zn - (L^{101})_{2-l'}$$

wherein (O—N) is a bidentate ligand, having metal coordinated to atoms O and N.

In another aspect, Met is selected from Ir and Pt. In a further aspect, $(Y^{103}_Y^{104})$ is a carbene ligand.

Examples of other organic compounds used as host are 15 selected from the group consisting of aromatic hydrocarbon cyclic compounds such as benzene, biphenyl, triphenyl, triphenylene, tetraphenylene, naphthalene, anthracene, phenalene, phenanthrene, fluorene, pyrene, chrysene, perylene, and azulene; the group consisting of aromatic heterocyclic 20 compounds such as dibenzothiophene, dibenzofuran, dibenzoselenophene, furan, thiophene, benzofuran, benzothiophene, benzoselenophene, carbazole, indolocarbazole, pyridylindole, pyrrolodipyridine, pyrazole, imidazole, triazole, oxazole, thiazole, oxadiazole, oxatriazole, dioxazole, thiadiazole, pyridine, pyridazine, pyrimidine, pyrazine, triazine, oxazine, oxathiazine, oxadiazine, indole, benzimidazole, indazole, indoxazine, benzoxazole, benzisoxazole, benzothiazole, quinoline, isoquinoline, cinnoline, quinazo-30 line, quinoxaline, naphthyridine, phthalazine, pteridine, xanthene, acridine, phenazine, phenothiazine, phenoxazine, benzofuropyridine, furodipyridine, benzothienopyridine, thienodipyridine, benzoselenophenopyridine, and selenophenodipyridine; and the group consisting of 2 to 10 cyclic 35 structural units which are groups of the same type or different types selected from the aromatic hydrocarbon cyclic group and the aromatic heterocyclic group and are bonded to each other directly or via at least one of oxygen atom, nitrogen atom, sulfur atom, silicon atom, phosphorus 40 atom, boron atom, chain structural unit and the aliphatic cyclic group. Each option within each group may be unsubstituted or may be substituted by a substituent selected from the group consisting of deuterium, halide, alkyl, cycloalkyl, heteroalkyl, arylalkyl, alkoxy, aryloxy, amino, silyl, alkenyl, 45 cycloalkenyl, heteroalkenyl, alkynyl, aryl, heteroaryl, acyl, carbonyl, carboxylic acids, ester, nitrile, isonitrile, sulfanyl, sulfinyl, sulfonyl, phosphino, and combinations thereof.

In one aspect, the host compound contains at least one of the following groups in the molecule:

-continued
$$Z^{101}$$
 Z^{102}
 Z^{102}
 Z^{102}
 Z^{102}
 Z^{102}
 Z^{103}
 Z^{104}

wherein each of R^{101} to R^{107} is independently selected from the group consisting of hydrogen, deuterium, halide, alkyl, cycloalkyl, heteroalkyl, arylalkyl, alkoxy, aryloxy, amino, silyl, alkenyl, cycloalkenyl, heteroalkenyl, alkynyl, aryl, $_{15}$ heteroaryl, acyl, carbonyl, carboxylic acids, ester, nitrile, isonitrile, sulfanyl, sulfinyl, sulfonyl, phosphino, and combinations thereof, and when it is aryl or heteroaryl, it has the similar definition as Ar's mentioned above. k is an integer from 0 to 20 or 1 to 20; k''" is an integer from 0 to 20. X^{101} 20 to X^{108} is selected from C (including CH) or N. Z^{101} and Z^{102} is selected from NR 101 , O, or S.

and

Additional Emitters

One or more additional emitter dopants may be used in conjunction with the compound of the present disclosure.

Examples of the additional emitter dopants are not particularly limited, and any compounds may be used as long as the compounds are typically used as emitter materials.

Examples of suitable emitter materials include, but are not limited to, compounds which can produce emissions via phosphorescence, fluorescence, thermally activated delayed fluorescence, i.e., TADF (also referred to as E-type delayed fluorescence), triplet-triplet annihilation, or combinations of the present disclosure.

US20080297033, US2008210930, US20090115322, US20090115322, US20090104472, US20100244004, US20110057559, fluorescence, i.e., TADF (also referred to as E-type delayed fluorescence), triplet-triplet annihilation, or combinations of US2012292601, US2013165653,

Non-limiting examples of the emitter materials that may be used in an OLED in combination with materials disclosed herein are exemplified below together with references that disclose those materials: CN103694277, CN1696137, EB01238981, EP01239526, EP01961743, EP1239526, 55 EP1244155, EP1642951, EP1647554, EP1841834, EP1841834B, EP2062907, EP2730583, JP2012074444, JP2013110263, JP4478555, KR1020090133652, KR20120032054, KR20130043460, TW201332980, U.S. 6,699,599, 6,916,554, US20010019782, 60 Pat. Nos. US20020034656, US20030068526, US20030072964, US20030138657, US20050123788, US20050244673, US2005123791, US2005260449, US20060008670, US20060065890, US20060127696, US20060134459, US20060134462, US20060202194, US20060251923, 65 US20070087321, US20070034863, US20070103060, US20070111026, US20070190359, US20070231600,

US2007034863, US2007104979, US2007104980, US2007138437, US2007224450, US2007278936, US20080020237, US20080233410, US20080261076, US200805851, US2008161567, US2008210930, US20090039776, US20090108737, US20090115322, US20090179555, US2009085476, US2009104472, US20100090591, US20100148663, US20100244004, US20100295032, US2010102716, US2010105902, US2010244004, US2010270916, US20110057559, US20110108822, US20110204333, US2011215710, US2011227049, US2011285275, US2012292601, US20130146848, US2013033172, US2013165653, US2013181190, US2013334521, US20140246656, US2014103305, U.S. Pat. Nos. 6,303,238, $6,413,656, \ 6,653,654, \ 6,670,645, \ 6,687,266, \ 6,835,469,$ 6,921,915, 7,279,704, 7,332,232, 7,378,162, 7,534,505, 7,675,228, 7,728,137, 7,740,957, 7,759,489, 7,951,947, 8.067,099, 8,592,586, 8,871,361, WO06081973, WO06121811, WO07018067, WO07108362, WO07115970, WO07115981, WO08035571, WO2002015645, WO2003040257, WO2005019373, WO2006056418, WO2008078800, WO2008054584, WO2008096609, WO2008101842, WO2009000673. WO2009050281, WO2009100991, WO2010028151, WO2010054731, WO2010086089, WO2010118029, WO2011044988. WO2011051404, WO2011107491, WO2012020327, WO2013094620, WO2012163471, WO2013107487, WO2013174471, WO2014007565, WO2014008982, WO2014023377, WO2014024131, WO2014031977, WO2014038456, WO2014112450.

-continued

35

HBL

A hole blocking layer (HBL) may be used to reduce the number of holes and/or excitons that leave the emissive layer. The presence of such a blocking layer in a device may result in substantially higher efficiencies and/or longer lifetime as compared to a similar device lacking a blocking layer. Also, a blocking layer may be used to confine emission to a desired region of an OLED. In some embodiments, the HBL material has a lower HOMO (further from the vacuum level) and or higher triplet energy than the emitter closest to the HBL interface. In some embodiments, the HBL material has a lower HOMO (further from the vacuum level) and or bigher triplet energy than one or more of the hosts closest to the HBL interface.

In one aspect, compound used in HBL contains the same molecule or the same functional groups used as host described above.

In another aspect, compound used in HBL contains at least one of the following groups in the molecule:

$$\begin{array}{c|c}
F & F \\
F & N & N
\end{array}$$

$$\begin{array}{c|c}
N & N & N
\end{array}$$

$$\begin{array}{c|c}
N & Al - (L^{101})_{3-k'}
\end{array}$$

wherein k is an integer from 1 to 20; L^{101} is an another ligand, k^\prime is an integer from 1 to 3. ETL

Electron transport layer (ETL) may include a material capable of transporting electrons. Electron transport layer 60 may be intrinsic (undoped), or doped. Doping may be used to enhance conductivity. Examples of the ETL material are not particularly limited, and any metal complexes or organic compounds may be used as long as they are typically used to transport electrons.

In one aspect, compound used in ETL contains at least one of the following groups in the molecule:

wherein R¹⁰¹ is selected from the group consisting of hydrogen, deuterium, halide, alkyl, cycloalkyl, heteroalkyl, arylalkyl, alkoxy, aryloxy, amino, silyl, alkenyl, cycloalkenyl, heteroalkenyl, alkynyl, aryl, heteroaryl, acyl, carbonyl, carboxylic acids, ester, nitrile, isonitrile, sulfanyl, sulfinyl, sulfonyl, phosphino, and combinations thereof, when it is aryl or heteroaryl, it has the similar definition as Ar's mentioned above. Ar¹ to Ar³ has the similar definition as Ar's mentioned above. k is an integer from 1 to 20. X¹⁰¹ to X¹⁰⁸ is selected from C (including CH) or N.

In another aspect, the metal complexes used in ETL 50 contains, but not limit to the following general formula:

$$\begin{bmatrix} O \\ N \end{bmatrix}_{k} Al - (L^{101})_{3-k'} \begin{bmatrix} O \\ N \end{bmatrix}_{k} Be - (L^{101})_{2-k'} \\ \begin{bmatrix} O \\ N \end{bmatrix}_{\nu} Zn - (L^{101})_{2-k'} \begin{bmatrix} O \\ N \end{bmatrix}_{\nu} Zn - (L^{101})_{2-k'} \end{bmatrix}$$

wherein (O—N) or (N—N) is a bidentate ligand, having metal coordinated to atoms O, N or N, N; L¹⁰¹ is another ligand; k' is an integer value from 1 to the maximum number of ligands that may be attached to the metal.

Non-limiting examples of the ETL materials that may be used in an OLED in combination with materials disclosed herein are exemplified below together with references that

20

25

30

35

40

45

50

55

60

65

disclose those materials: CN103508940, EP01602648, EP01734038, EP01956007, JP2004-022334, JP2005149918, JP2005-268199, KR0117693, KR20130108183, US20040036077, US20070104977, US2007018155, US20090101870, US20090115316, 5 US20090140637, US2009079554, US2009218940, US2010108990, US2011156017, US2011210320, US2012193612, US2012214993, US2014014925, US2014014927, US20140284580, U.S. Pat. Nos. 6,656,612, WO2007111263, 8,415,031, WO2003060956, WO2010067894, WO2010072300, WO2009148269, WO2011105373, WO2013079217, WO2011074770, WO2013145667, WO2013180376, WO2014104499, WO2014104535,

Charge Generation Layer (CGL)

In tandem or stacked OLEDs, the CGL plays an essential role in the performance, which is composed of an n-doped layer and a p-doped layer for injection of electrons and holes, respectively. Electrons and holes are supplied from the CGL and electrodes. The consumed electrons and holes in the CGL are refilled by the electrons and holes injected from the cathode and anode, respectively; then, the bipolar currents reach a steady state gradually. Typical CGL materials include n and p conductivity dopants used in the transport layers.

In any above-mentioned compounds used in each layer of the OLED device, the hydrogen atoms can be partially or fully deuterated. Thus, any specifically listed substituent, such as, without limitation, methyl, phenyl, pyridyl, etc. may be undeuterated, partially deuterated, and fully deuterated versions thereof. Similarly, classes of substituents such as, without limitation, alkyl, aryl, cycloalkyl, heteroaryl, etc. also may be undeuterated, partially deuterated, and fully deuterated versions thereof.

EXPERIMENTAL

An example of the inventive compounds (compound 1) can be synthesized by the procedure shown in the following scheme.

15

$$K_2$$
PtCl₄

tert-butanol/water

25

 P_1
 P_1
 P_2
 P_3
 P_4
 P_4

The intermediate materials of Pt(ppy)ppyHCl and Pt(ppy)₂Cl₂ may be prepared using a previously reported procedure (*Inorg. Chem.* 2010, 49, 11297-11308, which is incorporated by reference herein in its entirety) Briefly, a mixture of 2-phenylpyridine and potassium tetrachloroplatinate in tert-butanol and water is heated to reflux to provide Pt(ppy) ppyHCl, which is then reacted with iodobenzene dichloride in dichloromethane at room temperature to provide Pt(ppy)₂Cl₂. Then, a solution of 2,2'-dibromo-1,1'-biphenyl is treated with BuLi at low temperature, and then added to a solution of Pt(ppy)₂Cl₂ and silver triflate in CH₃CN. The mixture is heated to reflux to provide compound 1.

Compound 1

It is understood that the various embodiments described herein are by way of example only, and are not intended to limit the scope of the invention. For example, many of the materials and structures described herein may be substituted with other materials and structures without deviating from 5 the spirit of the invention. The present invention as claimed may therefore include variations from the particular examples and preferred embodiments described herein, as will be apparent to one of skill in the art. It is understood that various theories as to why the invention works are not 10 intended to be limiting.

We claim:

1. A compound having the formula Pt(L)n;

wherein $\bar{P}t$ is Pt(IV) metal, L is a ligand coordinating to $_{15}$ Pt, and n is 2;

wherein each L is tridentate and wherein each L can be the same or different;

wherein the Pt has an octahedral coordinating geometry, and

wherein the compound is neutral;

wherein each \boldsymbol{L} is independently selected from the group consisting of:

wherein X is a neutral coordinating atom selected from the group consisting of carbene carbon, phosphorus, and nitrogen, and each X can be the same or different; 40 and

wherein Y is an anionic coordinating atom selected from the group consisting of carbon, nitrogen, and oxygen atom, and each Y can be the same or different;

provided that at least one Y is an oxygen atom.

2. The compound of claim 1, wherein the compound is selected from the group consisting of:

3. The compound of claim 1, wherein the compound has a formula selected from the group consisting of:

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}

$$\mathbb{R}^{A}$$
 \mathbb{R}^{B}
 \mathbb{R}^{B}

$$\mathbb{R}^{R}$$
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{C}
 \mathbb{R}^{R}
 \mathbb{R}^{C}
 \mathbb{R}^{R}
 \mathbb{R}^{C}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{R}
 \mathbb{R}^{C}
 \mathbb{R}^{R}
 \mathbb{R}^{R}

wherein R⁴, R^B, R^C, R^D, R^E, R^F, R, and R' each independently represent mono to the maximum possible number of substitutions, or no substitution;

wherein R^A, R^B, R^C, R^D, R^E, and R^F, R, and R' are each independently selected from the group consisting of hydrogen, deuterium, halide, alkyl, cycloalkyl, heteroalkyl, arylalkyl, alkoxy, aryloxy, amino, cyclic amino, silyl, alkenyl, cycloalkenyl, heteroalkenyl, 25 alkynyl, aryl, heteroaryl, acyl, carbonyl, carboxylic acid, ether, ester, nitrile, isonitrile, sulfanyl, sulfinyl, sulfonyl, phosphino and combinations thereof;

wherein any adjacent R^A, R^B, R^C, R^D, R^E, R^F, R, and R' are optionally joined to form a ring; and

wherein the compound is neutral.

- **4**. The compound of claim **3**, wherein each of \mathbb{R}^A , \mathbb{R}^B , \mathbb{R}^C , \mathbb{R}^D , \mathbb{R}^E , and \mathbb{R}^F , represents hydrogen or deuterium.
- 5. The compound of claim 1, wherein the compound is selected from the group consisting of:

6. An organic light emitting device (OLED) comprising: an anode;

a cathode; and

an organic layer, disposed between the anode and the cathode, comprising a compound having the formula Pt(L)n;

wherein Pt is Pt(IV) metal, L is a ligand coordinating to Pt, and n is 2;

wherein each L is tridentate and wherein each L can be the same or different;

wherein the Pt has an octahedral coordinating geometry, and

wherein the compound is neutral;

wherein each L is independently selected from the group $_{\ 15}$ consisting of:

$$\begin{array}{c} X \\ X \\ Y \end{array}$$

$$\begin{array}{c} Y \\ Y \end{array}$$

wherein X is a neutral coordinating atom selected from the group consisting of carbene carbon, phosphorus, and nitrogen, and each X can be the same or different;

wherein Y is an anionic coordinating atom selected from the group consisting of carbon, nitrogen, and oxygen atom, and each Y can be the same or different;

provided that at least one Y is an oxygen atom.

7. The OLED of claim 6, wherein the organic layer is an emissive layer and the compound is an emissive dopant or a non-emissive dopant.

8. The OLED of claim 6, wherein the organic layer further comprises a host, wherein the host comprises a triphenylene containing benzo-fused thiophene or benzo-fused furan;

wherein any substituent in the host is an unfused substituent independently selected from the group consisting of C_nH_{2n+1} , OC_nH_{2n+1} , OAr_1 , NC_nH_{2n+1}), $N(Ar_1)$ (Ar₂), $CH = CH - C_nH_{2n+1}$, $C = CC_nH_{2n+1}$, Ar_1 , $Ar_1 - Ar_2$, $C_nH_{2n} - Ar_1$;

wherein n is from 1 to 10; and

wherein Ar₁ and Ar₂ are independently selected from the group consisting of benzene, biphenyl, naphthalene, triphenylene, carbazole, and heteroaromatic analogs thereof.

9. The OLED of claim 6, wherein the organic layer further comprises a host, wherein host comprises at least one chemical group selected from the group consisting of triphenylene, carbazole, dibenzothiophene, dibenzofuran, dibenzoselenophene, azatriphenylene, azacarbazole, aza-dibenzothiophene, aza-dibenzofuran, and aza-dibenzoselenophene; or

wherein the host comprises a metal complex.

10. The OLED of claim 6, wherein the organic layer 65 further comprises a host, wherein the host is selected from the group consisting of:

-continued

and combinations thereof.

11. A consumer product comprising an organic light emitting device (OLED) wherein the OLED comprises:

5 an anode;

20

25

30

35

40

45

50

a cathode; and

an organic layer, disposed between the anode and the cathode, comprising a compound having the formula Pt(L)n;

wherein Pt is Pt(IV) metal, L is a ligand coordinating to Pt, and n is 2;

wherein each L is tridentate and wherein each L can be the same or different;

wherein the Pt has an octahedral coordinating geometry, and

wherein the compound is neutral;

wherein each L is independently selected from the group consisting of:

wherein X is a neutral coordinating atom selected from the group consisting of carbene carbon, phosphorus, and nitrogen, and each X can be the same or different;

wherein Y is an anionic coordinating atom selected from the group consisting of carbon, nitrogen, and oxygen atom, each Y can be the same or different, and at least one Y is an oxygen atom.

12. The consumer product of claim 11, wherein the consumer product is selected from the group consisting of a flat panel display, a computer monitor, a medical monitors television, a billboard, a light for interior or exterior illumination and/or signaling, a heads-up display, a fully or partially transparent display, a flexible display, a laser printer, a telephone, a cell phone, tablet, a phablet, a personal digital assistant (PDA), a wearable device, a laptop computer, a digital camera, a camcorder, a viewfinder, a microdisplay that is less than 2 inches diagonal, a 3-D display, a virtual reality or augmented reality display, a vehicle, a video walls comprising multiple displays tiled together, a theater or stadium screen, and a sign.

* * * *