5(a)(i)	**M1** ionic compound **M2** molten and / or aqueous	**2**
5(a)(ii)	oxidation number (of copper)	**1**
5(a)(iii)	fades / (becomes) colourless	**1**
5(a)(iv)	$Cu^{2+} + 2e \rightarrow Cu$ **M1** Cu^{2+} and (any number of) e on left hand side **M2** equation correct	**2**
5(a)(v)	OH^-	**1**
5(b)	anode dissolves	**1**

Question	Answer	Marks
5(c)(i)	**M1** silver **M2** spoon **M3** (aqueous or solution) of silver nitrate	**3**
5(c)(ii)	**M1** prevent corrosion **M2** improve appearance	**2**
5(d)(i)	**M1** carbon dioxide: (increased) global warming **M2** carbon monoxide: toxic	**2**
5(d)(ii)	needs high pressure to store hydrogen	**1**

Question	Answer	Marks