

# Learn Chemistry - User Manual

## Requirements

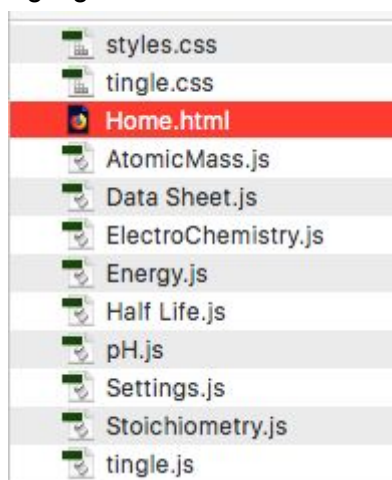
Learn Chemistry simply requires any HTML5 enabled browsers. This includes but is not limited to: Google Chrome, Mozilla Firefox and Microsoft Edge.

## Installation

To use the software, double click to open the file named "Learn Chemistry".



You will then be prompted with multiple files. Double click the file named "Home.html" as highlighted below to enter the software.



The software should now open up in your browser and you are now ready to use the program.

Learn Chemistry			Settings
Atomic Mass	Electrochemistry	Half Life	
13 questions answered	13 questions answered	18 questions answered	
7 questions correct	5 questions correct	4 questions correct	
6 questions incorrect	8 questions incorrect	14 questions incorrect	
54% correct answers	38% correct answers	22% correct answers	
Stoichiometry	Energy	pH	
18 questions answered	1 questions answered	5 questions answered	
3 questions correct	0 questions correct	4 questions correct	
15 questions incorrect	1 questions incorrect	1 questions incorrect	
17% correct answers	0% correct answers	80% correct answers	

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## Use of Learn Chemistry

Once you have loaded up the program. Simply click one of the 6 section buttons to load up practice questions.

# Atomic Mass

A modal should now popup with questions that you can answer by picking one of the four choices.

## Atomic Mass

How much does 83 moles of beryllium weigh?

748.01 grams

705.99 grams

790.03 grams

746.99 grams

If the correct answer was picked, the next question will be automatically loaded. If the wrong answer was to be picked, you have the choice to pick another option or click the message text to see the solution. Then you can click the next step buttons to show the next steps if necessary

## Atomic Mass

How much does 45 moles of carbon weigh?

525.93 grams

498.34 grams

540.5 grams

528.06 grams

That is incorrect. Please try another answer. Alternatively click this text for the solution

There are 45 moles of carbon. Therefore multiply 45 by the atomic mass of carbon.

Next Step  
(1/2)

$12.011\text{g/mol} \times 45\text{ mol} = 540.5\text{g}$

That is all  
(2/2)

If you decide to want to pick a different section of chemistry, simply click the cross in the top right and then click another section button.

×

Atomic Mass

How much does 45 moles of carbon weigh?

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540.5 grams

528.06 grams

That is incorrect. Please try another answer. Alternatively click this text for the solution

There are 45 moles of carbon. Therefore multiply 45 by the atomic mass of carbon.

12.011g/mol x 45 mol = 540.5g

Next Step  
(1/2)

That is all  
(2/2)

## Configuring

Learn Chemistry has the option to adjust font size if necessary. To do this, click the “Settings” button in the top right corner.

# Settings

You will then be prompted with the settings page where you can increase or decrease the font size.

Settings

Progress:

Reset

Font:

Normal

Large