



— $\text{n_H} + \text{O}_2 \rightleftharpoons \text{O} + \text{OH}$ [-0.37830382] — ('H + HO2 \rightleftharpoons H2O + OX', 'H + HO2 \rightleftharpoons H2O + O', 'H + HO2 \rightleftharpoons 2 OH', '2 OH \rightleftharpoons H2O + OX', '2 OH \rightleftharpoons H2O + O', 'H2O + OX \rightleftharpoons H2O + O', '2 OH \rightleftharpoons H2O2')_P_38_[1.30918007] — ('H + HO2 \rightleftharpoons H2O + OX', 'H + HO2 \rightleftharpoons H2O + O', 'H + HO2 \rightleftharpoons 2 OH', '2 OH \rightleftharpoons H2O + OX', '2 OH \rightleftharpoons H2O + O', 'H2O + OX \rightleftharpoons H2O + O', '2 OH \rightleftharpoons H2O2')_P_37_[-0.40989445] — X_0 _experiment_0_[-0.00565344] — ('H + HO2 \rightleftharpoons H2O + OX', 'H + HO2 \rightleftharpoons H2O + O', 'H + HO2 \rightleftharpoons 2 OH', '2 OH \rightleftharpoons H2O + OX', '2 OH \rightleftharpoons H2O + O', 'H2O + OX \rightleftharpoons H2O + O', '2 OH \rightleftharpoons H2O2')_P_36_[1.26338395]

— $\text{A_CO} + \text{OH} \rightleftharpoons \text{CO}_2 + \text{H}$ [-0.05751714] — $\text{Ea_H} + \text{O}_2 \rightleftharpoons \text{O} + \text{OH}$ [-924.58364173] — $\text{A_H} + \text{O}_2 \rightleftharpoons \text{O} + \text{OH}$ [-0.23413575] — ('HO2 + OH \rightleftharpoons H2O + O2', 'HO2 + OH \rightleftharpoons H2O + O2X')_P_1_[-1.26659148] — ('H + HO2 \rightleftharpoons H2O + OX', 'H + HO2 \rightleftharpoons H2O + O', 'H + HO2 \rightleftharpoons 2 OH', '2 OH \rightleftharpoons H2O + OX', '2 OH \rightleftharpoons H2O + O', 'H2O + OX \rightleftharpoons H2O + O', '2 OH \rightleftharpoons H2O2')_P_21_[1.08272897]