Topic: Half life

Question: Americium-241 has a half-life of 432 years. Find the decay constant.

Answer choices:

A 0.160

B 0.0160

C 0.00160

D 0.000160



Solution: C

The half life equation is

$$\frac{1}{2} = e^{kt}$$

Solve this for the decay constant k.

$$\ln \frac{1}{2} = \ln e^{kt}$$

$$\ln\frac{1}{2} = kt$$

$$k = \frac{\ln \frac{1}{2}}{t}$$

Use laws of logarithms to rewrite the log.

$$k = \frac{\ln 1 - \ln 2}{t}$$

$$k = \frac{0 - \ln 2}{t}$$

$$k = -\frac{\ln 2}{t}$$

Because k is a constant, we can absorb the negative sign into it.

$$k = \frac{\ln 2}{t}$$

Substitute t = 432.

$$k = \frac{\ln 2}{432}$$

$$k \approx 0.00160$$



Topic: Half life

Question: Carbon-19 has a decay constant of 0.000121. Find its half life.

Answer choices:

A 5,782 years

B 5,728 years

C 5,278 years

D 5,827 years



Solution: B

The half life equation is

$$\frac{1}{2} = e^{kt}$$

Solve this for time t.

$$\ln \frac{1}{2} = \ln e^{kt}$$

$$\ln\frac{1}{2} = kt$$

$$t = \frac{1}{k} \ln \frac{1}{2}$$

Use laws of logarithms to rewrite the log.

$$t = \frac{1}{k}(\ln 1 - \ln 2)$$

$$t = \frac{1}{k}(0 - \ln 2)$$

$$t = -\frac{\ln 2}{k}$$

Because k is a constant, we can absorb the negative sign into it.

$$t = \frac{\ln 2}{k}$$

Substitute the decay constant k = 0.000121.

$$t = \frac{\ln 2}{0.000121}$$

$$t \approx 5,728$$



Topic: Half life

Question: Plutonium-239 has a half-life of 24,110 years. Find the decay constant.

Answer choices:

A 0.0000287

B 0.000287

C 0.000000287

D 0.00000287



Solution: A

The half life equation is

$$\frac{1}{2} = e^{kt}$$

Solve this for the decay constant k.

$$\ln \frac{1}{2} = \ln e^{kt}$$

$$\ln\frac{1}{2} = kt$$

$$k = \frac{\ln \frac{1}{2}}{t}$$

Use laws of logarithms to rewrite the log.

$$k = \frac{\ln 1 - \ln 2}{t}$$

$$k = \frac{0 - \ln 2}{t}$$

$$k = -\frac{\ln 2}{t}$$

Because k is a constant, we can absorb the negative sign into it.

$$k = \frac{\ln 2}{t}$$

Substitute t = 24,110.

$$k = \frac{\ln 2}{24,110}$$

$$k \approx 0.0000287$$

