

We detail the γ ray decay scheme of the $^{137}_{55}\text{Cs}$ and $^{60}_{27}\text{Co}$, both of which we will use in the experiment.

Decay scheme for $^{137}_{55}\text{Cs}$

Caesium-137 ($^{137}_{55}\text{Cs}$) or radiocaesium, is a radioactive isotope of caesium that is formed as one of the more common fission products by the nuclear fission of uranium-235. The decay scheme for the $^{137}_{55}\text{Cs}$ is given in the schema below.

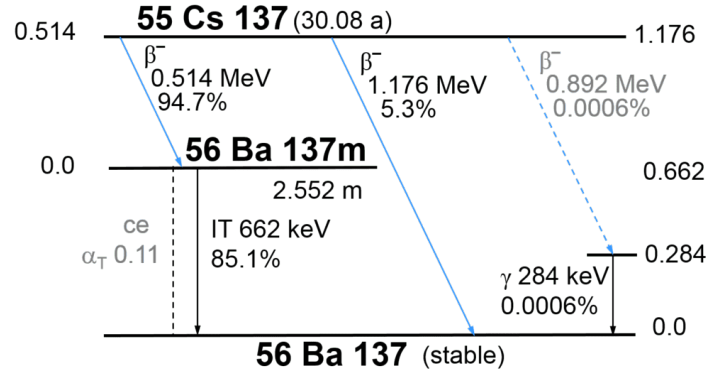
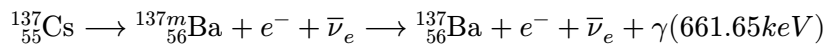


Figure 1: Decay scheme of $^{137}_{55}\text{Cs}$ into $^{137}_{56}\text{Ba}$ (Source: The Internet)

There are three decay schemes for $^{137}_{55}\text{Cs}$ decaying into $^{137}_{56}\text{Ba}$. The most observed ones are the β decay into $^{137m}_{56}\text{Ba}$ and then γ decay into $^{137}_{56}\text{Ba}$, and the direct β decay into $^{137}_{56}\text{Ba}$. We are observing the γ ray spectrum for the γ ray emitted on transition from $^{137m}_{56}\text{Ba}$ to $^{137}_{56}\text{Ba}$. The 'm' in $^{137m}_{56}\text{Ba}$ stands for metastable or unstable. The third decay scheme occurs only 0.0006% of the time, so, we can safely ignore that.

The decay leaves 94.7% of the parent Caesium nuclei into the metastable $^{137m}_{56}\text{Ba}$ nucleus via a β^- ray interaction and the release of an anti-neutrino. Then a γ ray decay of energy 662keV converts the metastable nuclei into the the ground state $^{137}_{56}\text{Ba}$. The nuclear reactions are written below.



Decay scheme for $^{60}_{27}\text{Co}$

Cobalt-60 ($^{60}_{27}\text{Co}$) is a synthetic radioactive isotope of cobalt with a half-life of 5.2714 years. It is produced artificially in nuclear reactors.

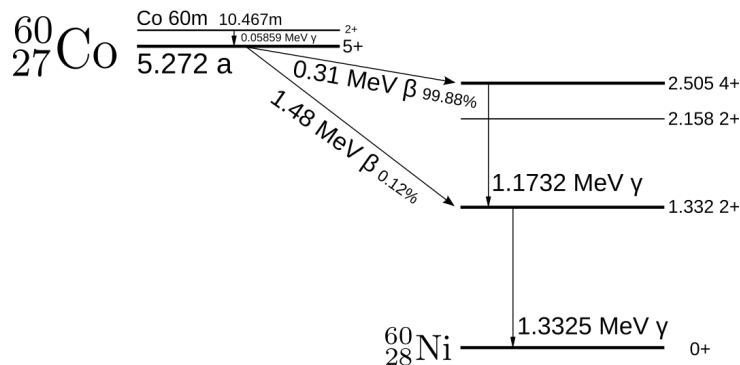


Figure 2: Decay scheme of $^{60}_{27}\text{Co}$ into $^{60}_{28}\text{Ni}$ (Source: The Internet)

This is produced on activation of $^{59}_{27}\text{Co}$ via neutron activation. The dominant interaction here is a β^- decay to a intermediate state and then two consecutive γ ray decays to get to $^{60}_{28}\text{Ni}$. Then we get a

multistep decay from $^{59}_{27}\text{Co}$ to $^{60}_{28}\text{Ni}$. We observe a multistep decay due to some selection rules that do not allow transitions to certain states to take place. We thus get the following decay processes

$$^{59}_{27}\text{Co} + n \longrightarrow ^{60}_{27}\text{Co}$$

$$^{60}_{27}\text{Co} \longrightarrow ^{60}_{28}\text{Ni}^* + e^- + \bar{\nu}_e \longrightarrow ^{60}_{28}\text{Ni}^* + e^- + \bar{\nu}_e + \gamma(1.173\text{MeV}) + \gamma(1.333\text{MeV})$$