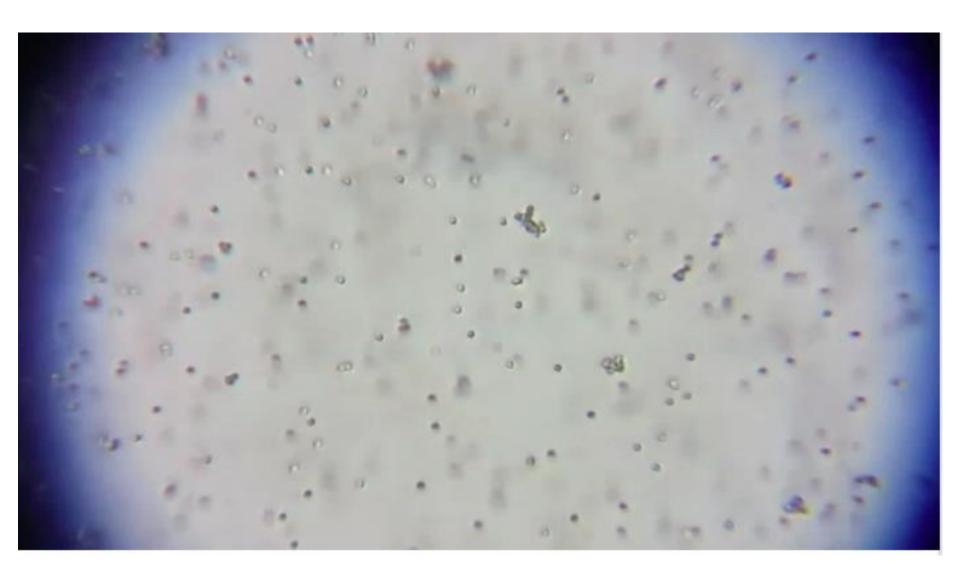
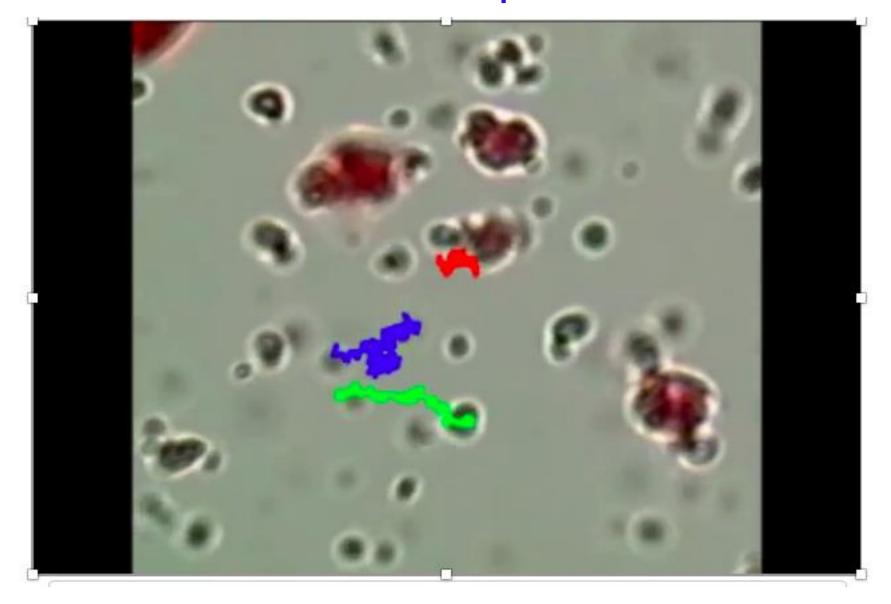
LS2102 Diffusion in Biology

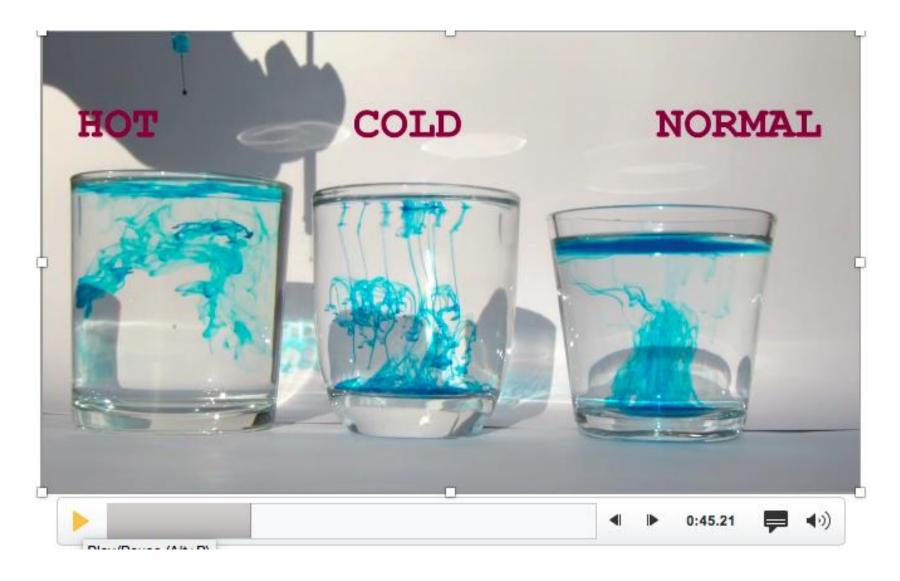
Very small plastic particles in water:



Pond water under the microscope:



Ink droplet in cold, normal and hot water:



Density (ink) > Density (water)

Quantitative Explanation



Robert Brox

Brownian motion was discovered by **Robert Brown** in **1827**

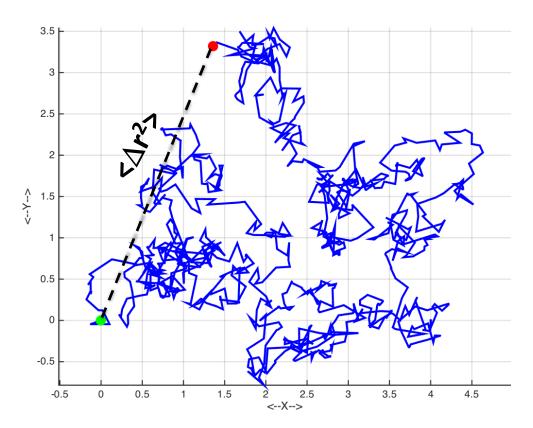


5. Über die von der molekularkinetischen Theorie der Wärme geforderte Bewegung von in ruhenden Flüssigkeiten suspendierten Teilchen; • von A. Einstein.

The explanation was formulated by A. Einstein in 1905

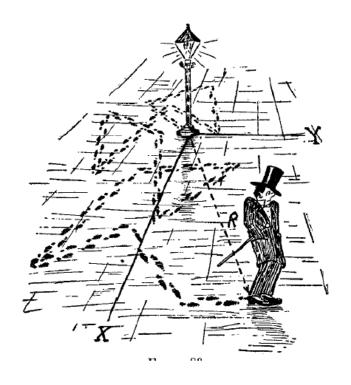
(before the Laws of Thermodynamics were put forward)

Diffusion: movement of particles in unbiased random walks in any dimension

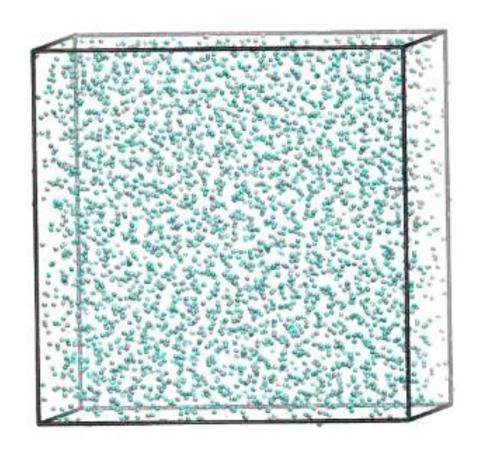


General Relationship:

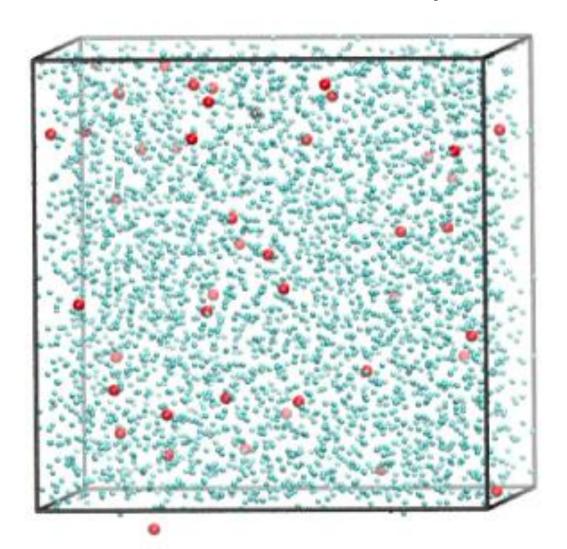
$$\langle r_N^2 \rangle = (2d)Dt$$



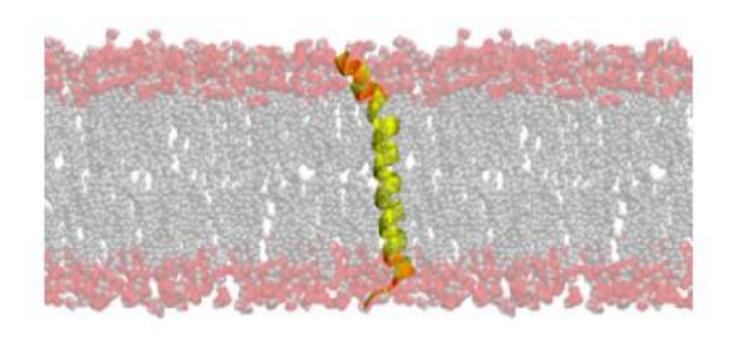
1A. Bulk Liquid - pure(Computer Simulation data)



1B. Bulk Liquid – mixture(Computer Simulation data)

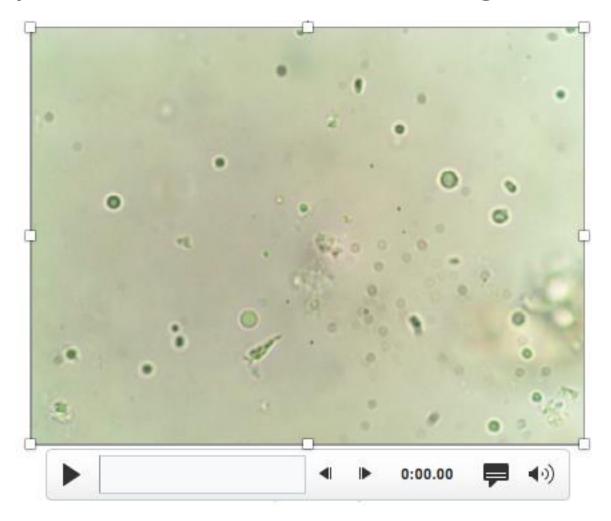


2. Protein fragment within a DPPC Bilayer (Computer simulation data)

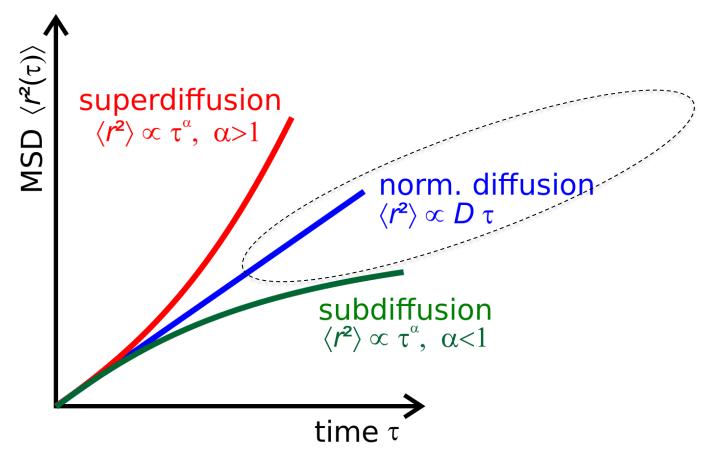


Frames spaced by $\Delta t = 10$ picoseonds

3. Pollen grain in water (Experimental data - Light microscopy)



Frames spaced by $\Delta t = 0.1$ seconds



Linear fits may still be done to extract and compare 'D'

1. Poor statistics

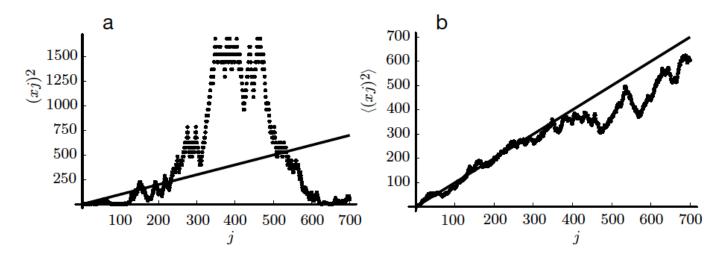
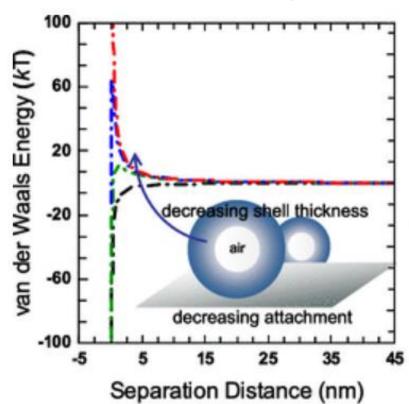


Figure 4.5: (Mathematical functions.) (a) Squared deviation $(x_j)^2$ for a single, one-dimensional random walk of 700 steps. Each step is one unit long. The solid line shows j itself; the graph shows that $(x_j)^2$ is not at all the same as j. (b) As (a), but this time the dots represent the average $\langle (x_j)^2 \rangle$ over thirty such walks. Again the solid line shows j. This time $\langle (x_j)^2 \rangle$ does resemble the idealized diffusion law (Equation 4.4).

2. Particle-Particle Interaction

Movement is not truly 'random'



3. Finite Diffusion Volume

Spatial restriction and interaction with boundaries (walls)

