Tutorial IV (Organometallics): Aut-17-18

- Q1. "CO" is a π acid ligand, explain the statement using simple orbital overlap diagram.
- **Q2**. For each of the given complexes (a) determine the number of electrons; (b) Find out the oxidation state of the metal, (c) count the total electron present in each complex.
- **Q3.** Write down the products of the following reactions (A and B) and calculate the EAN for the complexes (I and II).

$$Et_{3}P CO + CH_{3}Br \longrightarrow A$$

$$(I)$$

$$Ph_{3}P Ph_{3} + H_{2} \longrightarrow B$$

$$CI PPh_{3}$$

Draw the structure of pre-catalyst used in Monsanto acetic acid process.

Q4. Identify the missing products with the structures A, C and D; and also count the EAN (effective atomic number) on the compound B.

D

$$CI$$
 E
 CI
 E
 A
 $H_2C = CHR$
 $S = Solvent$
 $S = Solven$
 $S = Solven$

Q5. Show the 'ligand association' and 'migration' catalytic steps of Wilkinson's catalyst. Calculate EAN for both the reactants and products.