



UNIVERSITY OF GHANA  
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SCHOOL OF ENGINEERING SCIENCES  
**DEPARTMENT OF FOOD PROCESS ENGINEERING**  
B.Sc FIRST SEMESTER FINAL EXAMINATION, 2014/2015

**FAEN 205: THERMODYNAMICS (3 Credits)**

**TIME ALLOWED: TWO (2) HOURS**

**Answer TWO (2) Questions from SECTION A and ONE (1) Question from SECTION B**

**SECTION A**

1.
  - (a) State the two versions of the Second law of Thermodynamics and indicate the application of each.
  - (b) Define the coefficient of performance of a refrigerator and a heat pump and show the relationship between them.
  - (c) The food compartment of a refrigerator is maintained at  $5^{\circ}\text{C}$  by removing heat at a rate of  $8\text{ kW}$ , if the power requirement is  $2\text{ kW}$ , calculate:
    - (i) the rate at which heat is rejected to the kitchen environment in  $\text{kJ/min}$
    - (ii) the coefficient of performance of the refrigerator.

Hint: draw a schematic diagram of the refrigerator

2.
  - (a) What is a heat reservoir or a thermal energy reservoir?
  - (b) What is a reversible process?
  - (c) A Carnot heat engine receives  $1\text{ MJ}$  of heat per cycle from a high temperature heat source maintained at  $627^{\circ}\text{C}$ , calculate:
    - (i) the thermal efficiency of the heat engine;
    - (ii) the heat rejected to the low temperature sink at  $27^{\circ}\text{C}$ ; and
    - (iii) the percentage of energy turned into useful work.

## SECTION B

3. The combined mechanical and electrical efficiency of a compressor in an air conditioning unit is estimated to be 80%. The air conditioner uses R134a as refrigerant and operates as an ideal vapour cycle. The vapour enters the compressor which operates between 0.3MPa and 1.2MPa as saturated vapour. The liquid from the condenser exits as a saturated liquid. Calculate the power requirement if the cooling load of the room is 9.6 MJ/hr.

NB. Illustrate with T-s diagram and a schematic diagram of the cycle.

4. A quantity of gas is compressed to ten times its initial pressure of 140 kPa according to the law  $PV^{1.30} = \text{constant}$ . If the initial volume and temperature are  $14 \times 10^4 \text{ cm}^3$  and  $27^\circ\text{C}$  respectively, determine:

- (i) the change in entropy,
- (ii) the work transfer to the gas, and
- (iii) the heat transfer to the gas

Take  $C_p = 1.041 \text{ kJ/kg K}$ ,  $C_v = 0.743 \text{ kJ/kg K}$ .

- (b) (i) Given the energy due to volume change,  $U_v$  to be

$$U_v = \frac{(1-2\nu)}{6E} [\sigma_1^2 + \sigma_2^2 + \sigma_3^2 - 2(\sigma_1\sigma_2 + \sigma_2\sigma_3 + \sigma_3\sigma_1)]$$

Show that distortion energy in uniaxial stress state is given by

$$U_{du} = \frac{(1+\nu)}{3E} S_y^2 \quad [10 \text{ marks}]$$

$$\text{Where } S_y = \sqrt{\frac{(\sigma_1 - \sigma_2)^2 + (\sigma_2 - \sigma_3)^2 + (\sigma_3 - \sigma_1)^2}{2}}$$

- (ii) What is the physical meaning of  $S_y$  ?  
(iii) Explain briefly one industrial application of failure theory. [7 marks]