

## UNIVERSITY OF GHANA

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## BSC. ENGINEERING FIRST SEMESTER EXAMINATIONS: 2017/2018

## DEPARTMENT OF BIOMEDICAL ENGINEERING

BMEN 401: ENGINEERING PRINCIPLES OF HUMAN ANATOMY & PHYSIOLOGY (2 CREDITS)

**INSTRUCTIONS:** 

**ANSWER THREE (3) QUESTIONS** 

TIME ALLOWED: TWO (2) HOURS

1. a. Following a 650 mg intravenous bolus dose of drug to a 65 kg subject, the plasma drug concentration was found to decline bi-exponentially. The equation that best described the drug kinetic was:

$$C = 67e^{-14t} + 33e^{-3t}$$

where t is in hours and C is in µg/ml.

Calculate the following:

The apparent volume. i.

[5 marks]

The plasma level of the drug after 30 minutes of intravenous dose. ii.

The infusion rate if the drug is to be given at constant rate, the desired iii. steady state is 20 µg/ml and the elimination rate is 6.33 /hr.

[5 marks]

b. If the single compartment model (Figure 1) is used to describe a constant infusion rate of drug where Ro is the infusion rate and KE is the elimination rate.

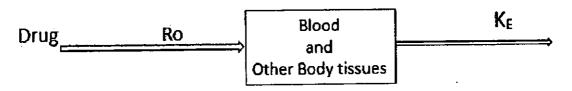


Figure 1: Single Compartment Model

Show by means of diagrams the following:

i.	The plasma drug	concentration	with	infusion	rate	Ro	until	saturation	ı.
								[2 marks]	1

ii. The plasma drug concentration with infusion rate 2Ro until saturation. [2 marks]

iii. The plasma drug concentration when the infusion is stopped at time T. [2 marks]

iv. Explain how you would find the elimination rate constant. [2 marks]

c. State four specific purposes of modelling.

[2 marks]

2. Table 1 gives the plasma drug concentrations (Cp) obtained following an intravenous bolus administration of a 250 mg dose of a drug that exhibited the characteristics of a one-compartment model and was eliminated exclusively by urinary excretion.

Table 1: Plasma Concentrations

Time (h)	Plasma Concentration (ug mL <sup>-1</sup> )				
0.5	68.0				
1.0	54.0				
2.0	30.0				
3.0	18.5				
5.0	6.0				
7.0	1.8				

a. Plot a suitable graph of the data.

[5 marks]

Using the graph, determine the following;

**b.** The elimination half-life  $(t_{1/2})$ . [5 marks] c. The overall elimination rate constant (K). [5 marks] d. The initial plasma concentration, (Cpo). [5 marks] [5 marks]

e. The apparent volume of distribution (V).

f. The drug plasma concentration at 75 min following the administration of a 2.5 mgkg<sup>-1</sup> dose to a subject weighing 70 kg. [5 marks]

**EXAMINER: DR. JOHN KUTOR** Page 2 of 4 3. a. What is Donnan equilibrium and how is it different from thermodynamic equilibrium?

[3 marks]

b. Name two effects of Donnan equilibrium in a cell.

[2 marks]

c. The diagram (Figure 2a and 2b) show two compartments separated by a biological membrane with NaCl solution in one side and Na<sup>+</sup> and A<sup>-</sup> ions on the other side. If the membrane is only permeable to Na<sup>+</sup> and Cl<sup>-</sup>, calculate the concentrations of ions in side I and side II after equilibrium. [10 marks]

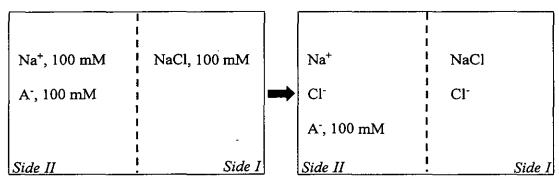


Figure 2a: Initial distribution (time = 0)

Figure 2b: Distribution at equilibrium

d. The deviation g(t) of a subject's blood glucose concentration can be modeled to satisfy the second order differential equation below.

$$\frac{d^2g}{dt^2} + 2\alpha \frac{dg}{dt} + \omega_o^2 g = Q(t)$$

Where  $\alpha$  is the decay constant,  $\omega_0$  is the natural frequency of the subject and Q(t) is the glucose impulse function. Describe briefly the glucose tolerance test and state the condition used to differentiate between a diabetic patient and a non-diabetic patient. [5 marks]

e. Sketch the graphs of blood glucose concentration and plasma insulin concentration for a non-diabetic person and a diabetic patient if both of them are given vanilla ice cream and their blood glucose rose to 250 mg/dL assuming that normally blood glucose is at 64.8 to 104.4 mg/dL.

[5 marks]

4. a. The intracellular total chemical potential of an ion X may be defined as:

$$\mu_i^X = \mu_0^X + RT \ln[X]_i + z_X FV_i$$

where  $z_xFV$  is the electrical component and all other symbols have their usual meanings. Derive the expression for the Nernst potential.

[7 marks]

**b.** A solution of 100 mmol/L KCl is separated from a solution of 10 mmol/L KCl by a membrane that is very permeable to K<sup>+</sup> ions, but impermeable to Cl<sup>-</sup> ions. What are the magnitude and the direction (sign) of the potential difference that will be generated across this membrane? (Assume that 2.3 RT/F = 60 mV).

[8 marks]

c. The drug plasma concentration time profile can be described by the Bateman's formula:

$$C_{p} = \frac{FDk_{a}}{V_{d}(k_{a} - \lambda)} \left(e^{-\lambda t} - e^{-k_{a}t}\right)$$

- i. Define all the symbols in the equation.
- ii. Calculate the maximum plasma concentration Cp if F = 0.9, D = 600 mg,  $k_a = 1$  hr<sup>-1</sup>,  $\lambda = 0.15$  hr<sup>-1</sup> and the  $V_d = 30$  L.

[10 marks]