



UNIVERSITY OF GHANA

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**BSC. MATERIALS SCIENCE AND ENGINEERING
END OF FIRST SEMESTER EXAMINATIONS: 2016/2017
DEPARTMENT OF MATERIALS SCIENCE AND ENGINEERING**

MTEN 409: GLASSES, CEMENTS AND CONCRETES (2 Credits)

TIME ALLOWED: TWO (2) HOURS

ANSWER ALL QUESTIONS.

1.

- a. Describe the glass transition temperature, T_g using specific volume vs temperature plots for crystalline and glasses.
- b. With the aid of appropriate diagram describe the effect of cooling rate on the density of glasses.
- c. Contrast the crystallite and random-network models glass structure.
- d. Name the Zachariasen rules to be satisfied by an oxide in order to form a glass.
- e. Describe the three types of cations commonly found in oxide glass including their functions. Give two examples of each.
- f. By what mechanism does a network former modifier function?
- g. State two requirements for the formation of noncrystalline ceramic from a melt instead of a crystalline ceramic to be possible.
- h. Using Table 1 (page 2) as a guide, why are SiO_2 and B_2O_3 good glass formers but NaCl and CaSiO_3 are not good glass formers?

30 Marks

2.

- a. For each of the following types of glass, state typical composition, properties and two applications. (i) fused silica glass; (ii) soda lime silica glass and (iii) Pyrex.
- b. For soda-lime-silica glass, describe the function of each of the three components.
- c. For silica melt containing 5 % soda, state the composition of the final liquid to solidify under equilibrium conditions and at what temperature. Use Figure 1 (page 2).
- d. What are glass ceramics? Compare the general characteristics of glass with glass ceramics.
- e. Describe the process for making glass ceramics with the aid of appropriate diagram.

30 Marks

Table 1 Factors Affecting Glass-Forming Ability

Compo- sition	$T_{mp} (^{\circ}\text{C})$	$\Delta H_f/T_{mp}$ (cal/mole/ $^{\circ}\text{K}$)	$(1/\eta)_{mp}$ (poise $^{-1}$)	$(\Delta H_f/T_{mp}) \times$ $(1/\eta)_{mp}$	Comments
B_2O_3	450	7.3	2×10^{-5}	1.5×10^{-4}	Good glass former
SiO_2	1713	1.1	1×10^{-6}	1.1×10^{-6}	Good glass former
$\text{Na}_2\text{Si}_2\text{O}_5$	874	7.4	5×10^{-4}	3.7×10^{-3}	Good glass former
Na_2SiO_3	1088	9.2	5×10^{-3}	4.5×10^{-2}	Poor glass former
CaSiO_3	1544	7.4	10^{-1}	0.74	Very difficult to form as glass
NaCl	800.5	6.9	50	345	Not a glass former

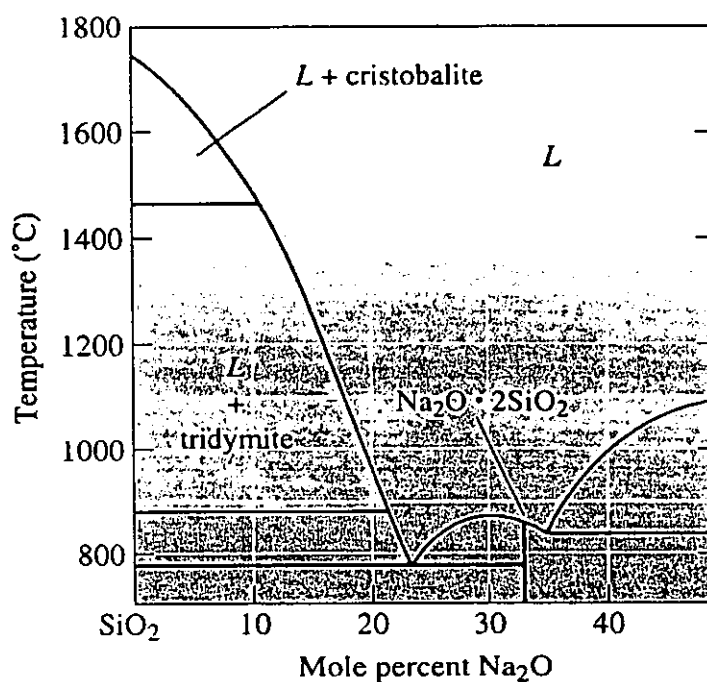


Figure 1 SiO_2 - Na_2O phase diagram. Additions of soda (Na_2O) to silica dramatically reduce the melting temperature of silica by forming eutectics.

3.

- a. What are the components of Portland cement and typical compositions?
- b. Name and describe the functions of the typical phases in Portland cement.
- c. What are the typical phase compositions of and the characteristics of Type I and Type III?
- d. Describe the role of gypsum in Portland cement.

20 Marks

4.

- a. What are the components and typical batch composition of concretes?
- b. Name and describe three (3) pozzolans and their effect on the properties of concrete.
- c. Describe the manufacture of the following concretes
 - i. Autoclaved Aerated Concrete (AAC)
 - ii. Prestressed concrete

20 Marks