

UNIVERSITY OF GHANA

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BSC. MATERIALS SCIENCE & ENGINEERING
SECOND SEMESTER EXAMINATIONS: 2015/2016
SCHOOL OF ENGINEERING SCIENCES
DEPARTMENT OF MATERIALS SCIENCE & ENGINEERING
MTEN 304: COMPUTATIONAL MATERIALS SCIENCE (2 CREDITS)

INSTRUCTIONS: ANSWER ALL QUESTIONS

TIME ALLOWED: TWO (2) HOURS

1.

- a. Briefly describe what you understand by the term basis sets.
- b. Give the major difference between a plane wave basis and gaussian basis sets.
- c. You have been tasked with a materials problem that requires computer simulation in which you are provided with two different basis sets; STO-3G and 6-31++G.
 - i. Give an explanation to the basis sets designation.
 - ii. Which one of the two will you use in a simulation experiment if one requires accurate results with respect to the simulation?
 - iii. Will your choice of basis set also affect the computational time? Explain your answer.
- d. Mention two (2) criteria for selecting basis functions.
- e. As a computational materials engineer, you have been involved in a project that requires the use of computational modelling. How would you approach this project?

30 Marks

- i.' Liquid water
- ii. Protein
- iii. Hard Spheres
- b. Three (3) types of Density Functional Theory (DFT) calculations exist. Name them and comment on their appropriateness in using each for computational simulation.
- c. What does adding correlation do? What is the leading order effect in the wavefunction?
- **d.** State the major difference between an 'ab initio' method and semi-empirical approach.
- c. Differentiate between molecular mechanics and molecular dynamics.

30 Marks

3.

- a. State clearly the differences between Hartree-Fock (HF) and Density Functional Theory (DFT).
- **b.** Mention three (3) challenges of computer simulations.
- c. Explain what you understand by the following;
 - i. Geometry optimization
 - ii. Iteration process
 - iii. Convergence.
- d. Explain what you understand by 'periodic boundary conditions' (PBC).
- e. Briefly describe what a Lennard-Jones potential is used for. Sketch this type of potential using two atomic species.

30 Marks