



UNIVERSITY OF GHANA  
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B.Sc ENGINEERING  
FIRST SEMESTER EXAMINATION: 2015/2016  
DEPARTMENT OF FOOD PROCESS ENGINEERING  
FAEN 205: THERMODYNAMICS (3 Credits)

Answer FOUR questions

TIME: TWO AND HALF (2 1/2) HRS

Question 1

- a) A steam turbine, operating reversibly and adiabatically, takes in superheated steam at 500 kPa and 300°C and discharges at 50 kPa. What is the power output of the turbine if it operates under these conditions and the steam rate is 5 kg/s?
- b) One kilogram of steam is contained in a piston/cylinder device at 800 kPa and 200°C. If it undergoes a mechanically reversible, isothermal expansion to 150 kPa, how much heat does it absorb?

(Use the steam tables provided.)

Question 2

One mole of an ideal gas is compressed isothermally but irreversibly at 450K from 3 bar to 8 bar in a piston/cylinder device. The work required is 40% greater than work of reversible isothermal compression. The heat transferred from the gas during compression flows to a heat reservoir at 30°C.

- a) Calculate the entropy change of the gas
- b) Calculate the entropy change of the heat reservoir
- c) Calculate the total entropy of both system and surroundings
- d) Is the second law of thermodynamics violated in (c) or not. Comment.

Question 3

- a) Calculate the amount of heat Q required to raise the temperature of 1 mol of ethane from 250 to 500°C in a steady-flow process at a pressure sufficiently low that ethane may be considered an ideal gas. Temperature dependence of specific heat capacity is given by:

$$\frac{c_p^{ig}}{R} = A + BT + CT^2 + DT^{-2}$$

$$R=8.314 \text{ J/mol.K; } A=1.131; 10^3B=19.225; 10^6C=-5.561; 10^5D=0$$

b) Steam at 250 (psia) and 600 (°F) [state 1] enters a turbine through a 3-inch diameter pipe with a velocity of 12 (ft)(s)<sup>-1</sup>. The exhaust from the turbine is carried through a 10-inch diameter pipe and is at 5 (psia) and 200 (°F) [state 2]. What is the power output of the turbine?

$$H_1 = 1322.6 \text{ Btu/lb}_m$$

$$V_1 = 3.058 \text{ ft}^3/\text{lb}_m$$

$$H_2 = 1148.6 \text{ Btu/lb}_m$$

$$V_2 = 78.14 \text{ ft}^3/\text{lb}_m$$

$$g_c = 32.174 \text{ lb}_m \text{ ft/lb}_f \cdot \text{s}^2$$

$$[1 \text{ ft-lb}_f = 0.01285 \text{ Btu}]$$

#### Question 4

a) Determine the molar volume of n-propane at 362.4K and 21.24 bar by each of the following methods:

- (i) The ideal-gas equation
- (ii) The generalized compressibility-factor correlation

$$T_c = 369.8 \text{ K}, P_c = 42.48 \text{ bar}, \omega = 0.152, R = 83.14 \text{ cm}^3 \cdot \text{bar/mol} \cdot \text{K}$$

(b) A particular power plant operates with a heat-source reservoir at 350°C and a heat sink reservoir at 30°C. It has a thermal efficiency equal to 45 % of the Carnot-engine thermal efficiency for the same temperatures. What is the thermal efficiency of the plant? And to what temperature must the heat-source reservoir be raised to increase the thermal efficiency of the plant to 35%? Again the  $\eta$  is 45% of the Carnot-engine value.

#### Question 5

a) Given that  $U$  is a function of  $T$  and  $V$ , i.e.  $U = U(T, V)$ , derive the following expression

$$dU = C_v dT + \left[ T \left( \frac{\partial P}{\partial T} \right)_V - P \right] dV$$

where  $U$  is the internal energy of the system, and all other variables have the usual meanings.

b) Steam at 2100 kPa and 250°C expands at constant enthalpy (as in a throttling process) to 125 kPa. What is the temperature of the steam in its final state?

(Use the steam tables provided)