Problem 1.1) T=245K Na = 5 × 10 m 7=280K, 300K; 330K What is the electron concentration? What is the whole concentration? 9=1.602 ×10 intrinsie semiconduter pi=ni Na >> 1i Extrince Semiconducter $n_0 p_0 = n_i^2$ $\rho_0 \approx \frac{\kappa_i^2}{N_D}$ Not MA no ~ No Po 2 Wa

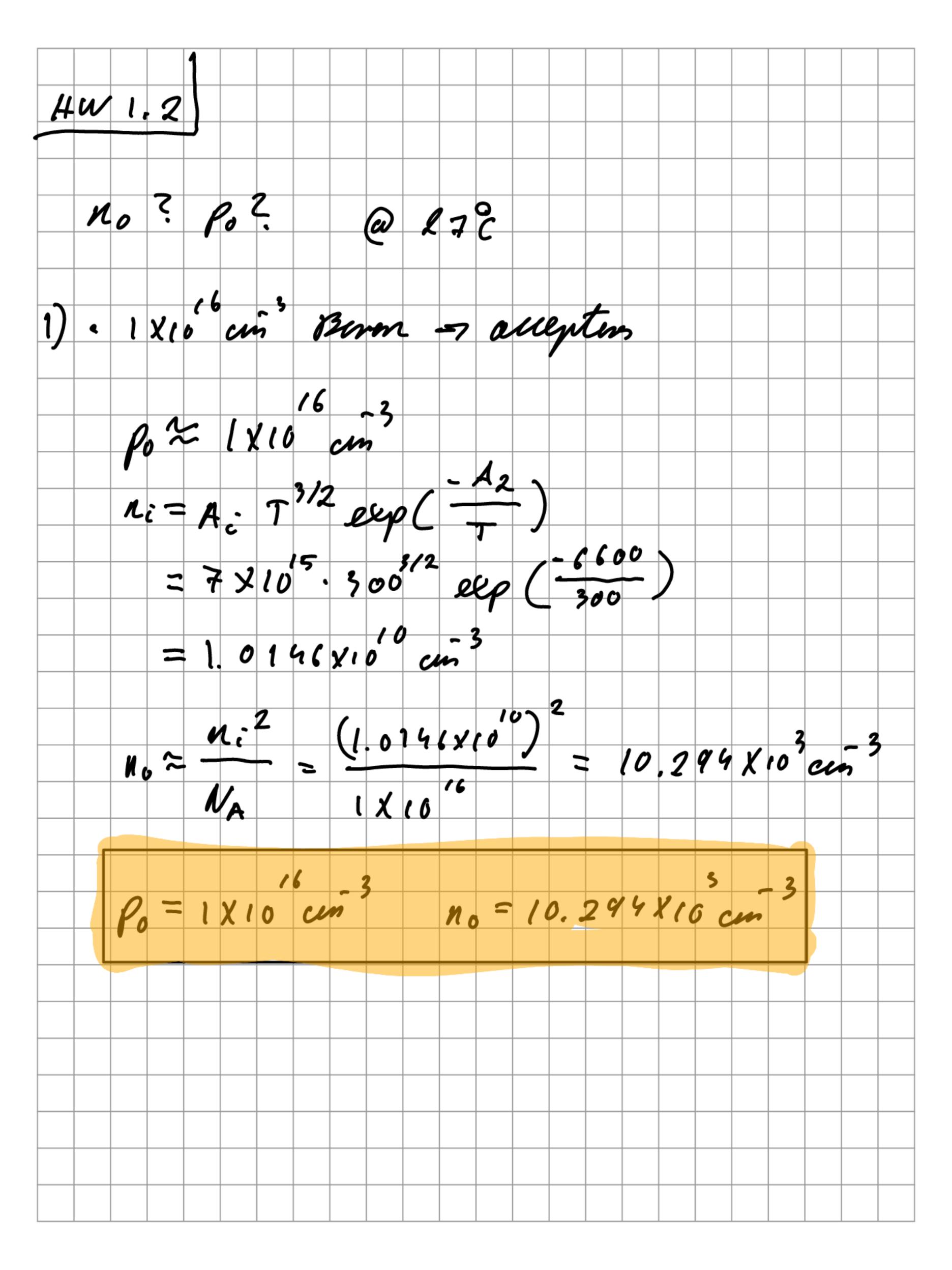
(a) calate
$$n_i$$
 @ different temporatures

 $n_i = A_i T$ elep (- A_2/t)

 $= 7 \times 10^{15} K^{-3/2} - (280 K)$
 $= 94 \times 10^{15} K^{-3/2} - (280 K)$
 $= 1.900 \times 10^{10} Com^3 (295 K)$
 $N_i = 6.814 \times 10^{9} Com^3 (295 K)$
 $N_i = 1.0146 \times 10^{10} Com^3 (295 K)$
 $N_i = 8.649 \times 10^{10} Com^3 (295 K)$

Calculate no using n_i and n_a
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 $N_a \approx \frac{1}{5 \times 10^{17}} = 7.22 (295 K)$
 $N_a \approx \frac{1}{5 \times 10^{17}} = 42.36 (295 K)$
 $Calculate no using n_i and n_a
 $Calculate no using n_i and $n_a$$$

*time!	usuer	•		
Temp		no		
280K	5×10 ¹⁷	7.22 cm ⁻³		
2 45 K	5 × 10 3	92.86 cm-3		
300 K	5 × 10 3	205.88 cm ⁻³		
330K	5 × 10 7 3	14.961×10 ³		



$$p_{0} = \text{Mi} \ \text{lesp}\left(\frac{E_{i} - E_{f}}{2T}\right)$$

$$\frac{P_{0}}{N_{i}} = \text{lesp}\left(\frac{E_{i} - E_{f}}{2T}\right)$$

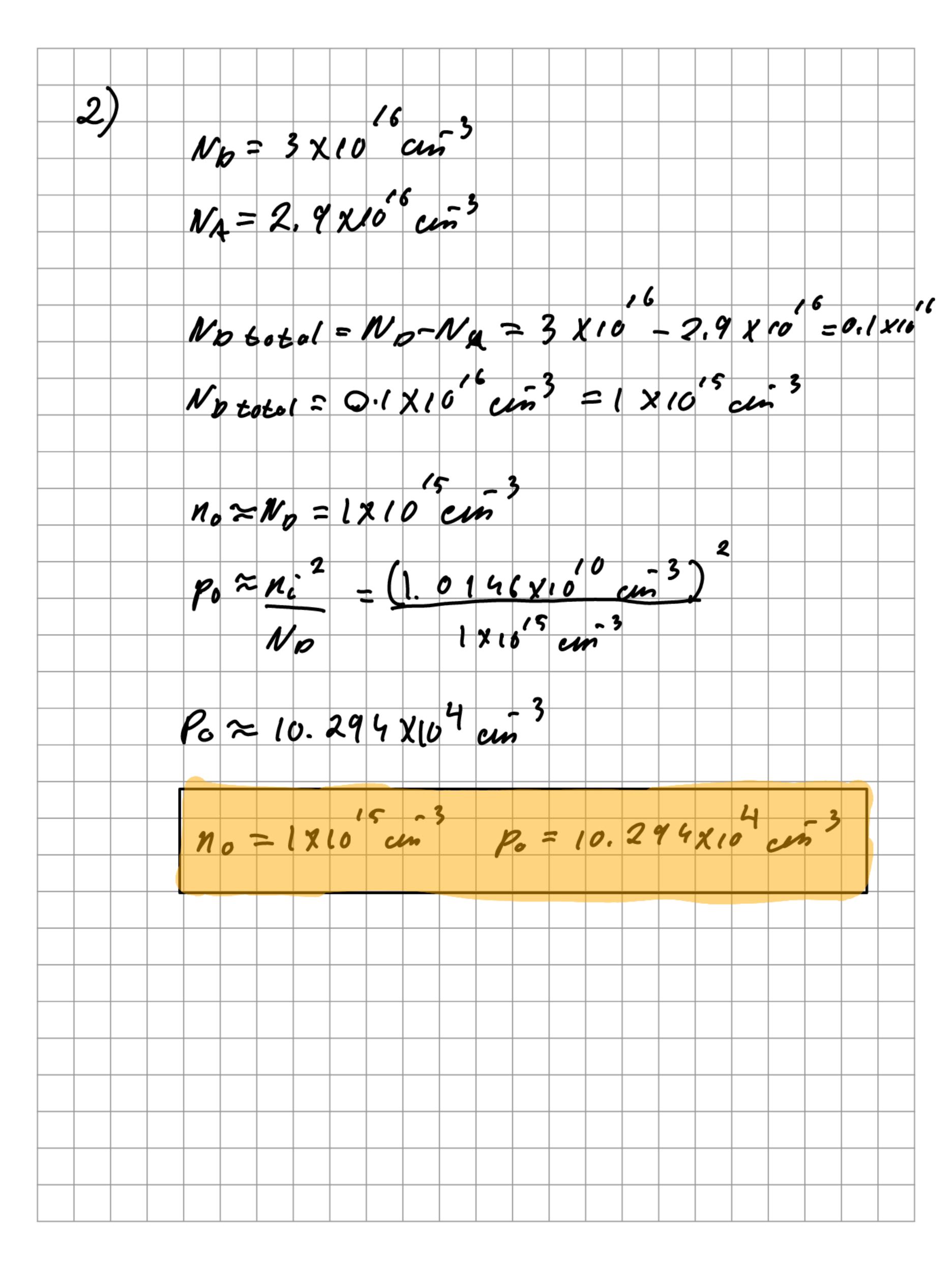
$$\ln\left(\frac{P_{0}}{N_{i}}\right) = \frac{E_{i} - E_{f}}{2T}$$

$$2T \cdot \ln\left(\frac{P_{0}}{N_{i}}\right) = E_{i} - E_{f}$$

$$E_{f} = E_{i} - 2T \cdot \ln\left(\frac{P_{0}}{N_{i}}\right)$$

$$E_{f} = 0.56 \text{ eV} - 3.62 \times 10^{5} \ln\left(\frac{1 \times 10^{6}}{1.014 \times 10^{6}}\right)$$

$$\frac{E_{f}}{N_{i}} = 0.203106$$



$$n_{0} = \text{Mic exp}\left(\frac{E_{5} - E_{c}}{2T}\right)$$

$$n_{0} = \text{exp}\left(\frac{E_{5} - E_{c}}{2T}\right)$$

$$n_{1} = \text{exp}\left(\frac{E_{5} - E_{c}}{2T}\right)$$

$$ln\left(\frac{n_{0}}{n_{c}}\right) = \frac{E_{5} - E_{c}}{2T}$$

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$$E_{7} = l_{2}T ln\left(\frac{n_{0}}{n_{c}}\right) + E_{6}$$

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$$E_{7} = 300 \times 8.62 \times 10^{5} \text{ eV} ln\left(\frac{1 \times 10^{5}}{1.01 \times 10^{5}}\right) + 0.56 \text{ eV}$$

$$E_{7} = 0.957 \text{ eV}$$

